

Supporting Information

Recent Advances and Strategies Applied to Tailor Energy levels, Active Sites and Electron Mobility in Titania and its Doped/Composite Analogues for Hydrogen Evolution in Sunlight

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Table S1 Different methods of hydrogen generation.¹

H ₂ production process	Advantages	Drawback
Steam reforming of methane	Least expensive (48 % of world hydrogen produced by this route)	Generation of greenhouse gases.
Gasification of coal	Only competitive with methane reforming where natural gas is expensive.	Generation of CO ₂ ; Less efficient
Gasoline and methanol reforming	Not mentioned. Requirement of pure oxygen;	Generation of more CO ₂ than steam reforming.
From biomass	Less expensive raw materials.	Little contribution (4%) towards world H ₂ production.
Electrolysis	Cost effective for production of extremely pure hydrogen in small amount.	Electrolysis is very much expensive at large scale.
Solar and wind power based electrolysis	Less expensive than conventional electrolysis	Still in developing stage

Table S2 Crystal phase/structure dependent H₂ evolution of TiO₂ photocatalyst

Photocatalyst	Crystal structure	Band gap (eV)	Co-catalyst	Light source	Reactant solution	Sacrificial agents	H ₂ evolution	Year	Ref
TiO ₂	Anatase: Rutile (87:13)	3.24	No co-catalyst used	415W Philips CLEO fluorescent tubes	Water	Methanol	10.74 μmol/h	2013	[2]
Cu-doped TiO ₂ film	Anatase	Not mentioned	No co-catalyst used	300-W Xe lamp	Water	Methanol	810 μmol/h/g	2014	[3]
TiO ₂ nanofiber	Anatase and Rutile	3.0	Pt	350 W Xe arc lamp	Water	Methanol	324 μmol/g/h	2014	[4]
Pt/TiO ₂	Anatase/Rutile	3.1	No co-catalyst used	100 Watt ultraviolet lamp (H-144GC-100, Sylvania par 38)	Water	Ethanol	2000 μmol/m ²	2015	[5]
Ti ³⁺ -doped TiO ₂ (Sheets)	Anatase	3	Pt	300 W xenon lamp	Water	Methanol	46 μmol/g	2013	[6]
Au/HTiNbO ₅	Orthorhombic	2.1	No co-catalyst used	400 W halide lamp	Water	Methanol	100 μmol/h	2014	[7]

Table S3 H₂ evolution efficiency of the porous structured TiO₂ composites

Photocatalyst	Porosity	Crystal structure	Band gap (eV)	Co-catalyst	Light source	Reactant solution	Sacrificial agents	H₂ evolution	Year	Ref
N-doped mesoporous TiO ₂	Mesoporous	Anatase	2.4	No co-catalyst used	450W Xe lamp	Water	Methanol	14.9 μmol/g/h	2012	[8]
Hierarchical and fibrous meso-macroporous N-doped TiO ₂	Meso-macroporous	Anatase	2.3	Pt	125 W Hg visible lamp	Water	Methanol	380 μmol/g/h	2013	[9]
Mesoporous TiO ₂ -SiO ₂	Mesoporous	Anatase	3.2	No co-catalyst used	300 W Xenon arc lamp	Water	Methanol	0.08805 mmol/h	2014	[10]
Three-dimensionally ordered macroporous N-doped TiO ₂	Macroporous	Anatase	~2.4	No co-catalyst used	300 W xenon lamp	Water	0.25 M Na ₂ S and 0.35 M Na ₂ SO ₃ ,	29 μmol/g/h	2014	[11]
g-C ₃ N ₄ nanosheets hybridized N-TiO ₂ nanofibers	Mesoporous	Anatase	2.8	Pt	300 W xenon arc lamp	Water	Methanol	8931 μmol/g/h	2014	[12]

Table S4 Particle shape/morphology dependent H₂ evolution efficiency of TiO₂ photocatalyst.

Photocatalyst	Morphology	Crystal structure	Band gap (eV)	Co-catalyst	Light source	Reactant solution	Sacrificial agents	H ₂ evolution	Year	Ref
Pt@CuO/TiO ₂	Nanorods and Nanosheets	Anatase	2.9	Pt	500 W high-pressure Xe lamp	Water	Methanol	200 µL/h	2014	[13]
Pt-TiO ₂ nanotube arrays	Nanotube arrays	Not mentioned	3.2	No co-catalyst used	300 W Xe lamp	Water	Na ₂ SO ₄ & ethylene Glycol	135 µmol/h	2014	[14]
Titanium phosphate	Layered hexagonal shape	Anatase	3.2	No co-catalyst used	300 W Xenon lamp	Water	Methanol	286 µmol/h	2014	[15]
Hierarchical nanstructured TiO ₂	3D urchin	Anatase	3.17	Pt	300 W Xe lamp	Water	Methanol	1310 µmol/h/g	2014	[16]

Table S5 H₂ evolution efficiency of the doped-TiO₂ photocatalyst

Photocatalyst	Dopant	Crystal structure	Band gap (eV)	Co-catalyst	Light source	Reactant solution	Sacrificial agents	H ₂ evolution	Year	Ref
Nitrogen doped TiO ₂	N	Anatase	3.08	-	400 W medium pressure halide lamp UV / 500 W halogen Lamp-Visible	Water	Methanol	4386 µmol/g/h 185 µmol/g/h	2012	[17]
C,N-TiO ₂	C, N	Anatase	2.92	-	Xe-lamp	Water	Methanol	81.8 µmol/h	2013	[18]
Carbon coated N-TiO ₂ nanotube films	C, N	Rutile and Anatase	2	-	500 W high-pressure ball-shaped Xe lamp	Water	Na ₂ S (0.1 M) and Na ₂ SO ₃ (0.02 M)	400 µmol/g/h	2012	[19]
TiO ₂ functionalized N-graphene (NGR)	N and graphene	Anatase	-	-	150 W GY-10 xenon lamp	Water	triethanol amine	13.3 µmol/h	2014	[20]
N-TiO ₂ /graphene oxide	N and graphene	Anatase	2.69	-	500W high-pressure Hg lamp	Water	Methanol	250 µmol/h	2012	[21]
Self doped TiO ₂ -graphene nanoplatelets	Ti ³⁺ and graphene	Anatase	-	Pd	day-light fluorescent lamps	Water	Methanol	288 µmol/h	2014	[22]
N-TiO ₂	N	Anatase	2.9	Pd	day-light fluorescent lamps (Wipro, 36 W each)	Water	Methanol	12 µmol/2 h/50 mg	2012	[23]
N-TiO ₂	N	Anatase	2.4	Pt	Xe lamp (300 W)	Water	Na ₂ SO ₃	310 µmol/h	2006	[24]

S- and C-codoped TiO ₂	S, C	Anatase	3.1	Pt	400 W high pressure Hg lamp	Water	Methanol	70000 μmol/h	2013	[25]
TiO ₂ was co-doped with P and N	P,N	Rutile	2.0	-	300-W Xe lamp	Water	-	11.2 μmol/g/h	2014	[26]
Pt-N La ₂ Ti ₂ O ₇	Pt, N	-	1.56	-	Luzchem LZC-UVA, centered at 360 nm	Deionized water	-	200 μmol/h	2013	[27]
Indium and nitrogen co-doped TiO ₂	In, N	Anatase	3.0	-	Ordinary day light fluorescent lamps (Wipro, 36W)	Water	Methanol	13.9 μmol/h	2010	[28]
Ga-N co-doped TiO ₂	Ga, N	Anatase	3	-	125 W mercury lamp with a 400 nm cut-off filter	Water	Methanol	5 μmol/g/h	2012	[29]
Ce/N-Codoped TiO ₂	Ce, N	Anatase	2.52	-	500W medium pressure mercury lamp	Water	Methanol	100 μmol/h	2010	[30]

Table S6 H₂ evolution of the TiO₂ composite photocatalyst

Photocatalyst	Composite material	Crystal structure	Band gap (eV)	Co-catalyst	Light source	Reactant solution	Sacrificial agents	H₂ evolution	Ref
Eosin -sensitized CuO incorporated TiO ₂	Eosin dye and CuO	-	-	-	200 W halogen lamp	Water	diethanolamine	10.56 μmol/h	[31]
CuO/TiO ₂	CuO	Anatase: Rutile	2.61	-	SB-100P/F lamp (100 W, 365 nm)	Water	Ethanol	21300 μmol/h	[32]
Cu(OH) ₂ /TiO ₂ nanotube arrays	Cu(OH) ₂	Anatase	3.2	-	PLS-SXE300UV Xe lamp	Water	ethylene glycol	6.5 μmol/h/cm ⁻²	[33]
TiO ₂ /CuO composite nanofibers	CuO	Anatase	2.6	-	400 W Hg lamp and 360 W Na lamp as the UV-vis light source	Water	Methanol	5500 μmol/h	[34]
ZnO/TiO ₂ composite	ZnO	Anatase/Crystalline /Amorphous	-	-	300 W xenon lamp	Water	Methanol	865 μmol/h	[35]
Ta ₂ O ₅ -TiO ₂ / S doped TiO ₂ / S doped Ta ₂ O ₅	Ta ₂ O ₅ , S	Anatase	3.09 / 2.9 / 2.9	Pt	visible and UV irradiation (>300 nm) applying four 300W Vitalux bulbs	Water	Methanol	0.14 mmol / 4.55 mmol / 0.24mmol	[36]
Core-shell CuS@TiO ₂	CuS	Anatase	CuS 1.35 / TiO ₂ 3.19	-	UV-lamps (18Wcm ⁻²)	Water	Methanol	1.2 mmol/h	[37]
Copper oxide incorporated indium Titanium oxide	Cu ₂ O, In ₂ O ₃	cubic In ₂ O ₃ and Anatase TiO ₂ structure	2.01	-	500 W tungsten halogen lamp	Water	Methanol	2149 μmol/g/h	[38]

Pt-Sr(Zr _{1-x} Y _x)O _{3-δ} -TiO ₂ (Pt-SZYT)	Pt,Sr, Zr, Y	Anatase, Rutile	-	-	250 W xenon lamp	Water	oxalic acid	1.68 mmol/h	[39]
Ni/NiO/N-TiO _{2-x}	Ni, NiO,N	Anatase	~3.0-2.4	-	110-W high-pressure sodium lamp with UV cutoff filter ($\lambda > 420$ nm)	Water	Methanol	1210 μmol/g/h	[40]
Ti ³⁺ -doped TiO ₂ (Sheets)	Ti ³⁺	Anatase	3	Pt	300 W xenon lamp	Water	Methanol	46 μmol/g	[41]
CdS decorated Au@TiO ₂ core-shell	CdS, Au	Anatase	-	-	300-W Xenon lamp	Water	Na ₂ S and Na ₂ SO ₃	3.94 μmol/h	[42]
CdS/TiO ₂	CdS	Anatase	2.2	-	Hg-Arc lamp (500 W)	Water	Na ₂ S and Na ₂ SO ₃	422.4 μmol/h	[43]
CdS-tritanate nanotube	CdS	Hexagonal	2.4	-	300 W Xenon lamp	Water	Ethanol	10 μmol/h	[44]
TiO ₂ /Fe ₂ O ₃	Fe ₂ O ₃	-	-	Pt	300 W xenon lamp	Water	Methanol	1.5 mmol/h	[45]
Nanocomposite of MWCNTs and TiO ₂ nanoparticles	MWCNTs	-	~ 3	-	500 W Hg-lamp	Water	triethanolamine	8 mmol/g/h	[46]
Carbon/TiO ₂ /carbon nanotubes	Carbon	Anatase	~ 0.88	-	Solar simulator (Newport, 94023 A)	Water	Ethanol	37.6 mmol/h/g	[47]
TiO ₂ -RGO	RGO	Anatase	2.9	-	300 W xenon/mercury lamp	Water	Ethanol	1.125 mmol/g/h	[48]

CuO/TiO ₂ -GR	CuO, GR	Anatase: Rutile	2.8	-	500 W Xe arc lamp	Water	Methanol	2905.60 μmol/h	[49]
Pt/TiO ₂ /activated carbon	Pt, activated carbon	Anatase	3.1	-	250 W Hg lamp	Water	Methanol	650 μmol/h	[50]
Three-dimensionally ordered macroporous (3DOM) N-doped TiO ₂	N	Anatase	~ 2.4	-	300 W xenon lamp	Water	0.25 M Na ₂ S and 0.35 M Na ₂ SO ₃	29 μmol/g/h	[51]
CuO/Carbon Fiber/TiO ₂	CuO, Carbon Fiber	80:20 A:R	2.85	-	Xenon lamp	Water	Ethanol	2000 μmol/h	[52]

Table S7 Noble metal supported TiO₂ photocatalyst and their hydrogen evolution

Photocatalyst	Crystal structure	Band gap (eV)	Co-catalyst	Light source	Reactant solution	Sacrificial agents	H₂ evolution	Year	Ref
Au/TiO ₂	-	3.2	-	UV-PC mercury lamp	Water	Methanol	1866 μmol/h	2013	[53]
Au/TiO ₂	Anatase	2.29	-	Solar simulator (Thermo Oriel 1000 W)	Water	Methanol	45 μmol/h	2013	[54]
Pt/TiO ₂	Anatase and Rutile	3.19	-	150 W mercury lamp	Water	Methanol	220 μmol/h	2014	[55]
Pt/TiO ₂	Anatase/Rutile	3.1	-	350 W high-pressure mercury lamp	Water	Methanol	2100 μmol/h	2012	[56]
Pt/TiO ₂	Anatase	-	-	300W Xe lamp	Water	Methanol	739.94 μmol/h	2010	[57]
Pt/TiO ₂	Rutile	3.0	-	300 W xenon lamp	Water	-	33 μmol/h	2014	[58]

References

- [1] M. Momirlan, T. N. V. Current status of hydrogen energy. *Renewable and Sustainable Energy Reviews* **2002**, *6*, 141–179.
- [2] Pulido Melián, E.; González Díaz, O.; Ortega Méndez, A.; López, C. R.; Nereida Suárez, M.; Doña Rodríguez, J. M.; Navío, J. A.; Fernández Hevia, D.; Pérez Peña, J. Efficient and affordable hydrogen production by water photo-splitting using TiO₂-based photocatalysts. *International Journal of Hydrogen Energy* **2013**, *38*, 2144-2155.
- [3] Wang, C.; Hu, Q.; Huang, J.; Zhu, C.; Deng, Z.; Shi, H.; Wu, L.; Liu, Z.; Cao, Y. Enhanced hydrogen production by water splitting using Cu-doped TiO₂ film with preferred (001) orientation. *Applied Surface Science* **2014**, *292*, 161-164.
- [4] Xu, F.; Xiao, W.; Cheng, B.; Yu, J. Direct Z-scheme anatase/rutile bi-phase nanocomposite TiO₂ nanofiber photocatalyst with enhanced photocatalytic H₂-production activity. *International Journal of Hydrogen Energy* **2014**, *39*, 15394-15402.
- [5] Bashir, S.; Wahab, A. K.; Idriss, H. Synergism and photocatalytic water splitting to hydrogen over M/TiO₂ catalysts: Effect of initial particle size of TiO₂. *Catalysis Today* **2015**, *240*, 242-247.
- [6] Li, Y.; Yu, Z.; Meng, J.; Li, Y. Enhancing the activity of a SiC-TiO₂ composite catalyst for photo-stimulated catalytic water splitting. *International Journal of Hydrogen Energy* **2013**, *38*, 3898-3904.
- [7] Lin, H.-Y.; Chang, Y.-S. Photocatalytic water splitting on Au/HTiNbO₅ nanosheets. *International Journal of Hydrogen Energy* **2014**, *39*, 3118-3126.
- [8] Liu, S.-H.; Syu, H.-R. One-step fabrication of N-doped mesoporous TiO₂ nanoparticles by self-assembly for photocatalytic water splitting under visible light. *Applied Energy* **2012**, *100*, 148-154.
- [9] Parida, K. M.; Pany, S.; Naik, B. Green synthesis of fibrous hierarchical meso-macroporous N doped TiO₂ nanophotocatalyst with enhanced photocatalytic H₂ production. *International Journal of Hydrogen Energy* **2013**, *38*, 3545-3553.
- [10] Niphadkar, P. S.; Chitale, S. K.; Sonar, S. K.; Deshpande, S. S.; Joshi, P. N.; Awate, S. V. Synthesis, characterization and photocatalytic behavior of TiO₂-SiO₂ mesoporous

composites in hydrogen generation from water splitting. *Journal of Materials Science* **2014**, *49*, 6383-6391.

- [11] Wang, T.; Yan, X.; Zhao, S.; Lin, B.; Xue, C.; Yang, G.; Ding, S.; Yang, B.; Ma, C.; Yang, G.; Yang, G. A facile one-step synthesis of three-dimensionally ordered macroporous N-doped TiO₂ with ethanediamine as the nitrogen source. *Journal of Materials Chemistry A* **2014**, *2*, 15611-15619.
- [12] Han, C.; Wang, Y.; Lei, Y.; Wang, B.; Wu, N.; Shi, Q.; Li, Q. In situ synthesis of graphitic-C₃N₄ nanosheet hybridized N-doped TiO₂ nanofibers for efficient photocatalytic H₂ production and degradation. *Nano Research* **2015**, *8*, 1199-1209.
- [13] Teng, F.; Chen, M.; Li, N.; Hua, X.; Wang, K.; Xu, T. Effect of TiO₂ Surface Structure on the Hydrogen Production Activity of the Pt@CuO/TiO₂ Photocatalysts for Water Splitting. *ChemCatChem* **2014**, *6*, 842-847.
- [14] Zhang, L.; Pan, N.; Lin, S. Influence of Pt deposition on water-splitting hydrogen generation by highly-ordered TiO₂ nanotube arrays. *International Journal of Hydrogen Energy* **2014**, *39*, 13474-13480.
- [15] Guo, S.-y.; Han, S.; Chi, B.; Pu, J.; Li, J. Synthesis of shape-controlled mesoporous titanium phosphate nanocrystals: The hexagonal titanium phosphate with enhanced hydrogen generation from water splitting. *International Journal of Hydrogen Energy* **2014**, *39*, 2446-2453.
- [16] Haider, Z.; Kang, Y. S. Facile Preparation of Hierarchical TiO₂ Nano Structures: Growth Mechanism and Enhanced Photocatalytic H₂ Production from Water Splitting Using Methanol as a Sacrificial Reagent. *ACS Applied Materials & Interfaces* **2014**, *6*, 10342-10352.
- [17] Lin, H.-y.; Shih, C.-y. Efficient One-Pot Microwave-Assisted Hydrothermal Synthesis of Nitrogen-Doped TiO₂ for Hydrogen Production by Photocatalytic Water Splitting. *Catalysis Surveys from Asia* **2012**, *16*, 231-239.
- [18] Liu, S.-H.; Syu, H.-R. High visible-light photocatalytic hydrogen evolution of C,N-codoped mesoporous TiO₂ nanoparticles prepared via an ionic-liquid-template approach. *International Journal of Hydrogen Energy* **2013**, *38*, 13856-13865.

- [19] Jia, F. Z.; Yao, Z. P.; Jiang, Z. H. Carbon Coated N-Doping of TiO₂ Nanotube Films with Enhanced Visible-Light Photocatalytic Activity. *Advanced Materials Research* **2012**, *512-515*, 1564-1567.
- [20] Mou, Z.; Wu, Y.; Sun, J.; Yang, P.; Du, Y.; Lu, C. TiO₂ Nanoparticles-Functionalized N-Doped Graphene with Superior Interfacial Contact and Enhanced Charge Separation for Photocatalytic Hydrogen Generation. *ACS Applied Materials & Interfaces* **2014**, *6*, 13798-13806.
- [21] Pei, F.; Liu, Y.; Xu, S.; Lü, J.; Wang, C.; Cao, S. Nanocomposite of graphene oxide with nitrogen-doped TiO₂ exhibiting enhanced photocatalytic efficiency for hydrogen evolution. *International Journal of Hydrogen Energy* **2013**, *38*, 2670-2677.
- [22] Sayed, F. N.; Sasikala, R.; Jayakumar, O. D.; Rao, R.; Betty, C. A.; Chokkalingam, A.; Kadam, R. M.; Jagannath; Bharadwaj, S. R.; Vinu, A.; Tyagi, A. K. Photocatalytic hydrogen generation from water using a hybrid of graphene nanoplatelets and self doped TiO₂-Pd. *RSC Advances* **2014**, *4*, 13469-13476.
- [23] Sayed, F. N.; Jayakumar, O. D.; Sasikala, R.; Kadam, R. M.; Bharadwaj, S. R.; Kienle, L.; Schürmann, U.; Kaps, S.; Adelung, R.; Mittal, J. P.; Tyagi, A. K. Photochemical Hydrogen Generation Using Nitrogen-Doped TiO₂-Pd Nanoparticles: Facile Synthesis and Effect of Ti³⁺ Incorporation. *The Journal of Physical Chemistry C* **2012**, *116*, 12462-12467.
- [24] J. Yuan, M. C., J. Shi, and W. Shangguan, . Preparations and photocatalytic hydrogen evolution of N-doped TiO₂ from urea and titanium tetrachloride. *International Journal of Hydrogen Energy*, **2006**, *31*, 1326–1331.
- [25] Bai, H.; Kwan, K. S. Y.; Liu, Z.; Song, X.; Lee, S. S.; Sun, D. D. Facile synthesis of hierarchically meso/nanoporous S- and C-codoped TiO₂ and its high photocatalytic efficiency in H₂ generation. *Applied Catalysis B: Environmental* **2013**, *129*, 294-300.
- [26] Zheng, P.; Wu, H.; Guo, J.; Dong, J.; Jia, S.; Zhu, Z. P-N co-doping induced structural recovery of TiO₂ for overall water splitting under visible light irradiation. *Journal of Alloys and Compounds* **2014**, *615*, 79-83.

- [27] Meng, F.; Li, J.; Hong, Z.; Zhi, M.; Sakla, A.; Xiang, C.; Wu, N. Photocatalytic generation of hydrogen with visible-light nitrogen-doped lanthanum titanium oxides. *Catalysis Today* **2013**, *199*, 48-52.
- [28] Sasikala, R.; Shirole, A. R.; Sudarsan, V.; Jagannath; Sudakar, C.; Naik, R.; Rao, R.; Bharadwaj, S. R. Enhanced photocatalytic activity of indium and nitrogen co-doped TiO_2 -Pd nanocomposites for hydrogen generation. *Applied Catalysis A: General* **2010**, *377*, 47-54.
- [29] XiaoBo Li, Q. L., XiaoYing Jiang, Jianhua Huang. Enhanced Photocatalytic Activity of Ga-N Co-doped Anatase TiO_2 for Water Decomposition to Hydrogen. *International Journal of Electrochemical Science* **2012**, *7*, 11519 - 11527.
- [30] Sun, X.; Liu, H.; Dong, J.; Wei, J.; Zhang, Y. Preparation and Characterization of Ce/N-Codoped TiO_2 Particles for Production of H₂ by Photocatalytic Splitting Water Under Visible Light. *Catalysis Letters* **2010**, *135*, 219-225.
- [31] Jin, Z.; Zhang, X.; Li, Y.; Li, S.; Lu, G. 5.1% Apparent quantum efficiency for stable hydrogen generation over eosin-sensitized CuO/TiO_2 photocatalyst under visible light irradiation. *Catalysis Communications* **2007**, *8*, 1267-1273.
- [32] Chen, W.-T.; Jovic, V.; Sun-Waterhouse, D.; Idriss, H.; Waterhouse, G. I. N. The role of CuO in promoting photocatalytic hydrogen production over TiO_2 . *International Journal of Hydrogen Energy* **2013**, *38*, 15036-15048.
- [33] Zhang, S.; Wang, H.; Yeung, M.; Fang, Y.; Yu, H.; Peng, F. Cu(OH)₂-modified TiO_2 nanotube arrays for efficient photocatalytic hydrogen production. *International Journal of Hydrogen Energy* **2013**, *38*, 7241-7245.
- [34] Lee, S. S.; Bai, H.; Liu, Z.; Sun, D. D. Optimization and an insightful properties-Activity study of electrospun TiO_2/CuO composite nanofibers for efficient photocatalytic H₂ generation. *Applied Catalysis B: Environmental* **2013**, *140-141*, 68-81.
- [35] Guo, S.; Han, S.; Mao, H.; Dong, S.; Wu, C.; Jia, L.; Chi, B.; Pu, J.; Li, J. Structurally controlled ZnO/TiO_2 heterostructures as efficient photocatalysts for hydrogen generation from water without noble metals: The role of microporous

amorphous/crystalline composite structure. *Journal of Power Sources* **2014**, *245*, 979-985.

- [36] Stodolny, M.; Laniecki, M. Synthesis and characterization of mesoporous $Ta_2O_5-TiO_2$ photocatalysts for water splitting. *Catalysis Today* **2009**, *142*, 314-319.
- [37] Im, Y.; Kang, S.; Kim, K. M.; Ju, T.; Han, G. B.; Park, N.-K.; Lee, T. J.; Kang, M. Dynamic Hydrogen Production from Methanol/Water Photo-Splitting Using Core@Shell-Structured $CuS@TiO_2$ Catalyst Wrapped by High Concentrated TiO_2 Particles. *International Journal of Photoenergy* **2013**, *2013*, 1-10.
- [38] Vinothkumar, N.; De, M. Hydrogen production from water-methanol solution over visible light active indium-titanium oxide photocatalysts modified with copper oxide. *International Journal of Hydrogen Energy* **2014**, *39*, 11494-11500.
- [39] Yan, J.; Liu, Q.; Guan, L.; Liang, F.; Gu, H. Photocatalytic hydrogen generation of $Pt-Sr(Zr_{1-x}Y_x)O_{3-\delta}-TiO_2$ heterojunction under the irradiation of simulated sunlight. *Frontiers of Chemistry in China* **2009**, *4*, 121-126.
- [40] Hu, S.; Li, F.; Fan, Z.; Gui, J. Improved photocatalytic hydrogen production property over $Ni/NiO/N-TiO_{2-x}$ heterojunction nanocomposite prepared by NH_3 plasma treatment. *Journal of Power Sources* **2014**, *250*, 30-39.
- [41] Wang, J.; Zhang, P.; Li, X.; Zhu, J.; Li, H. Synchronical pollutant degradation and H_2 production on a Ti^{3+} -doped TiO_2 visible photocatalyst with dominant (001) facets. *Applied Catalysis B: Environmental* **2013**, *134-135*, 198-204.
- [42] Fang, J.; Xu, L.; Zhang, Z.; Yuan, Y.; Cao, S.; Wang, Z.; Yin, L.; Liao, Y.; Xue, C. $Au@TiO_2-CdS$ Ternary Nanostructures for Efficient Visible-Light-Driven Hydrogen Generation. *ACS Applied Materials & Interfaces* **2013**, *5*, 8088-8092.
- [43] Jang, J. S.; Li, W.; Oh, S. H.; Lee, J. S. Fabrication of CdS/TiO_2 nano-bulk composite photocatalysts for hydrogen production from aqueous H_2S solution under visible light. *Chemical Physics Letters* **2006**, *425*, 278-282.
- [44] Parayil, S. K.; Baltrusaitis, J.; Wu, C.-M.; Koodali, R. T. Synthesis and characterization of ligand stabilized CdS -Trititanate composite materials for visible light-induced

photocatalytic water splitting. *International Journal of Hydrogen Energy* **2013**, *38*, 2656-2669.

- [45] Luan, P.; Xie, M.; Fu, X.; Qu, Y.; Sun, X.; Jing, L. Improved photoactivity of TiO₂-Fe₂O₃ nanocomposites for visible-light water splitting after phosphate bridging and its mechanism. *Physical Chemistry Chemical Physics* **2015**, *17*, 5043-5050.
- [46] Ke, D.; Tianyou, P.; Dingning, K.; Bingqing, W. Photocatalytic hydrogen generation using a nanocomposite of multi-walled carbon nanotubes and TiO₂ nanoparticles under visible light irradiation. *Nanotechnology* **2009**, *20*, 125603.
- [47] Zhao, C.; Luo, H.; Chen, F.; Zhang, P.; Yi, L.; You, K. A novel composite of TiO₂ nanotubes with remarkably high efficiency for hydrogen production in solar-driven water splitting. *Energy & Environmental Science* **2014**, *7*, 1700-1707.
- [48] Nagaraju, G.; Manjunath, K.; Sarkar, S.; Gunter, E.; Teixeira, S. R.; Dupont, J. TiO₂-RGO hybrid nanomaterials for enhanced water splitting reaction. *International Journal of Hydrogen Energy* **2015**, *40*, 12209-12216.
- [49] Wang, B.; Sun, Q.; Liu, S.; Li, Y. Synergetic catalysis of CuO and graphene additives on TiO₂ for photocatalytic water splitting. *International Journal of Hydrogen Energy* **2013**, *38*, 7232-7240.
- [50] Hakamizadeh, M.; Afshar, S.; Tadjarodi, A.; Khajavian, R.; Fadaie, M. R.; Bozorgi, B. Improving hydrogen production via water splitting over Pt/TiO₂/activated carbon nanocomposite. *International Journal of Hydrogen Energy* **2014**, *39*, 7262-7269.
- [51] Zang, Y.; Li, L.; Xu, Y.; Zuo, Y.; Li, G. Hybridization of brookite TiO₂ with g-C₃N₄: A visible-light-driven photocatalyst for As 3+ oxidation, MO degradation and water splitting for hydrogen evolution. *Journal of Materials Chemistry A* **2014**, *2*, 15774-15780.
- [52] Yu, Z.; Meng, J.; Li, Y.; Li, Y. Efficient photocatalytic hydrogen production from water over a CuO and carbon fiber comodified TiO₂ nanocomposite photocatalyst. *International Journal of Hydrogen Energy* **2013**, *38*, 16649-16655.
- [53] Oros-Ruiz, S.; Zanella, R.; Lopez, R.; Hernandez-Gordillo, A.; Gomez, R. Photocatalytic hydrogen production by water/methanol decomposition using Au/TiO₂

prepared by deposition-precipitation with urea. *Journal of hazardous materials* **2013**, *263 Pt 1*, 2-10.

- [54] Buaki-Sogo, M.; Serra, M.; Primo, A.; Alvaro, M.; Garcia, H. Alginate as Template in the Preparation of Active Titania Photocatalysts. *ChemCatChem* **2013**, *5*, 513-518.
- [55] Serrano, D. P.; Calleja, G.; Pizarro, P.; Gálvez, P. Enhanced photocatalytic hydrogen production by improving the Pt dispersion over mesostructured TiO₂. *International Journal of Hydrogen Energy* **2014**, *39*, 4812-4819.
- [56] Wei, P.; Liu, J.; Li, Z. Effect of Pt loading and calcination temperature on the photocatalytic hydrogen production activity of TiO₂ microspheres. *Ceramics International* **2013**, *39*, 5387-5391.
- [57] Fang, J.; Shi, F.; Bu, J.; Ding, J.; Xu, S.; Bao, J.; Ma, Y.; Jiang, Z.; Zhang, W.; Gao, C.; Huang, W. One-Step Synthesis of Bifunctional TiO₂ Catalysts and Their Photocatalytic Activity. *The Journal of Physical Chemistry C* **2010**, *114*, 7940-7948.
- [58] Maeda, K. Photocatalytic properties of rutile TiO₂ powder for overall water splitting. *Catalysis Science & Technology* **2014**, *4*, 1949-1953.