

Supplementary Information

PEG/copper(I) halide cluster as an eco-friendly catalytic system for C-N bond formation

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Table S1 Crystal data and refinement details for **L3** and **L4**.

	L3	L4
Empirical formula	C ₂₃ H ₂₄ Fe ₂ Te ₂	C ₂₈ H ₂₆ Fe ₂ Te ₂
Formula weight	667.32	729.39
Crystal system	Monoclinic	Monoclinic
Space group	<i>P</i> 2 ₁ / <i>n</i>	<i>C</i> 2/ <i>c</i>
<i>T</i> /K	293	293
Wavelength /Å	0.71073	0.71073
<i>a</i> /Å	5.9310(12)	20.7280(14)
<i>b</i> /Å	10.639(2)	7.4755(5)
<i>c</i> /Å	35.440(7)	17.1801(10)
$\alpha/^\circ$	90.00	90
$\beta/^\circ$	94.06(3)	102.440(2)
$\gamma/^\circ$	90.00	90
V /Å ³	2230.6(8)	2599.6(3)
Z	4	4
D _{calcd} /Mg·m ⁻³	1.987	1.864
μ /mm ⁻¹	3.870	3.329
<i>F</i> (000)	1272	1400
Reflections/Unique	4471/4051	14137/2951
Parameters/Restraints	244/115	145/0
θ range /°	1.15-25.27	3.472-27.504
R _{int}	0.0436	0.0189
R ₁ [<i>I</i> > 2σ(<i>I</i>)]	0.0755	0.0205
wR ₂ [<i>I</i> > 2σ(<i>I</i>)]	0.1706	0.0598
GOF	1.020	1.167
Residuals /e·Å ⁻³	1.227/-0.834	0.427/-0.592
CCDC 1823302 (L3) CCDC 1823303 (L4) contain the supplementary crystallographic data for this paper.		

Table S2 Selected bond lengths and bond angles for **L3**, **L4** and **4-8**

		Bond length(Å)		Bond angles (°)				
L3	Te1-C7	2.114(13)		C7-Te1-C11	97.4(5)			
	Te1-C11	2.191(16)		C14-Te2-C13	95.6(5)			
	Te2-C14	2.124(12)		C12-C13-Te2	113.5(11)			
	Te2-C13	2.197(15)		Fe1-C7-Te1	124.4(6)			
	C12-C13	1.466(19)						
L4	Te1-C1	2.088(2)		C1-Te1-C11	94.56(10)			
	Te1-C11	2.172(3)		Fe1-C1-Te1	124.49(12)			
	C11-C12	1.492(4)						
		Bond length (Å)		Bond angles(°)				
4	Te1-C1	2.108(8)	Cu1-Br1 ⁱ	2.4466(14)	C1-Te1-C21	98.3(3)	Br1 ⁱ -Cu1-Te2 ⁱ	106.82(5)
	Te1-C21	2.176(9)	Cu1-Te2 ⁱ	2.5798(12)	C1-Te1-Cu1	113.1(3)	Te1-Cu1-Br1	90.58(4)
	Te1-Cu1	2.5579(13)	Cu1-Cu1 ⁱ	2.696(2)	C21-Te1-Cu1	105.9(2)	Te2 ⁱ -Cu1-Br1	99.79(4)
	Te2-C11	2.108(8)			C11-Te2-C23	93.6(3)	Br1 ⁱ -Cu1-Cu1 ⁱ	62.72(5)
	Te2-C23	2.177(8)			C11-Te2-Cu1 ⁱ	103.1(2)	Te1-Cu1-Cu1 ⁱ	112.72(6)
	Te2-Cu1 ⁱ	2.5798(12)			C23-Te2-Cu1 ⁱ	97.8(2)	Te2 ⁱ -Cu1-Cu1 ⁱ	115.59(6)
	Br1-Cu1 ⁱ	2.4467(14)			Cu1 ⁱ -Br1-Cu1	63.19(4)	Br1-Cu1-Cu1 ⁱ	54.09(4)
	Br1-Cu1	2.6847(14)			Br1 ⁱ -Cu1-Te1	114.49(5)	Te1-Cu1-Te2	126.64(4)
							Br1 ⁱ -Cu1-Br1	116.81(4)
(i)-x, 1-y, -z								
		Bond length(Å)		Bond angles(°)				
5	Te1-C1	2.094(3)		C1-Te1-C21	94.99(14)	I1-Cu1-Te2	117.39(2)	
	Te1-C21	2.158(4)		C1-Te1-Cu1	109.77(10)	Te2-Cu1-I1 ⁱ	86.597(16)	
	Te1-Cu1	2.5774(6)		C21-Te1-Cu1	92.18(9)	I1-Cu1-Cu1 ⁱ	62.585(18)	
	Te2-Cu1	2.6801(6)		Cu1-I1-Cu1 ⁱ	62.402(18)	Te1-Cu1-Cu1 ⁱ	136.88(3)	
	I1-Cu1	2.5486(5)		I1-Cu1-Te1	125.35(2)	Te2-Cu1-Cu1 ⁱ	111.46(3)	
	I1-Cu1 ⁱ	2.7614(6)		I1-Cu1-I1 ⁱ	117.599(2)	Cu1 ⁱ -Cu1-I1 ⁱ	55.014(18)	
	Cu1-Cu1 ⁱ	2.7568(10)		Te1-Cu1-I1 ⁱ	101.224(19)	Te1-Cu1-Te2	100.881(17)	
(i)1-x, -y, 1-z								

Bond length(Å)		Bond angles(°)		
6	Te1-C1	2.104(5)	C1-Te1-C21	96.0(2)
	Te1-C21	2.180(5)	C1-Te1-Cu1	108.50(14)
	Te1-Cu1	2.5509(7)	C21-Te1-Cu1	97.54(13)
	Br1-Cu1 ⁱ	2.5133(9)	Cu1 ⁱ -Br1-Cu1	70.57(3)
	Br1-Cu1	2.5187(8)	Br1 ⁱ -Cu1-Br1	109.43(3)
	Cu1-Br1 ⁱ	2.5133(9)	Br1 ⁱ -Cu1-Te1	116.52(3)
	Cu1-Te2	2.5892(7)	Br1-Cu1-Te1	105.20(3)
	Cu1-Cu1 ⁱ	2.9069(14)	Br1 ⁱ -Cu1-Te2	97.77(3)
			Br1-Cu1-Te2	99.57(3)

(i)2-x, -y, -z

Bond length (Å)		Bond angles(°)		
7	Te1-C1	2.102(5)	C1-Te1-C21 ⁱ	95.4(2)
	Te1-C21 ⁱ	2.184(5)	C1-Te1-Cu1	110.65(16)
	Te1-Cu1	2.5777(6)	C21 ⁱ -Te1-Cu1	98.05(13)
	I1-Cu1 ⁱⁱ	2.6634(7)	Cu1 ⁱⁱ -I1-Cu1	66.08(2)
	I1-Cu1	2.6700(7)	Te1-Cu1-I1 ⁱⁱ	115.60(2)
	Cu1-I1 ⁱⁱ	2.6635(7)	Te1-Cu1-Te2	125.18(3)
	Cu1-Te2	2.6195(6)	Te2-Cu1-I1 ⁱⁱ	97.73(2)
	Cu1-Cu1 ⁱⁱ	2.9079(13)	Te1-Cu1-I1	104.22(2)
			Te2-Cu1-I1	99.56(2)

(i)-0.5+x, 1.5-y, 0.5+z

(ii)2-x, 1-y, -1-z

Bond length (Å)		Bond angles(°)		
8	I1-Cu1	2.6408(18)	Cu1-I1-Cu1 ⁱ	79.90(3)
	I1-Cu ⁱ	2.765(3)	C1-Te1-C21	95.03(16)
	Te1-C1	2.106(4)	C1-Te1-Cu1	120.38(10)
	Te1-C21	2.183(4)	C21-Te1-Cu1	102.89(12)
	Te1-Cu1	2.586(2)	Te2-Cu1-Te1	117.13(3)
	Cu1-Te2	2.567(2)	Te2-Cu1-I1	116.97(6)
	Cu1-I1 ⁱ	2.765(3)	Te1-Cu1-I1	111.61(6)

(i)1-x, -y, 1-z

Table S3 Cyclic voltammetric data (mV).

	E _{1/2} ^a	ΔE _{1/2} ^a
L3	401/299	-
4	604	203
5	352	-49

[a] E_{1/2} values are the half-wave potential of the ferrocenyl group and are quoted relative to FcH/[FcH]⁺.

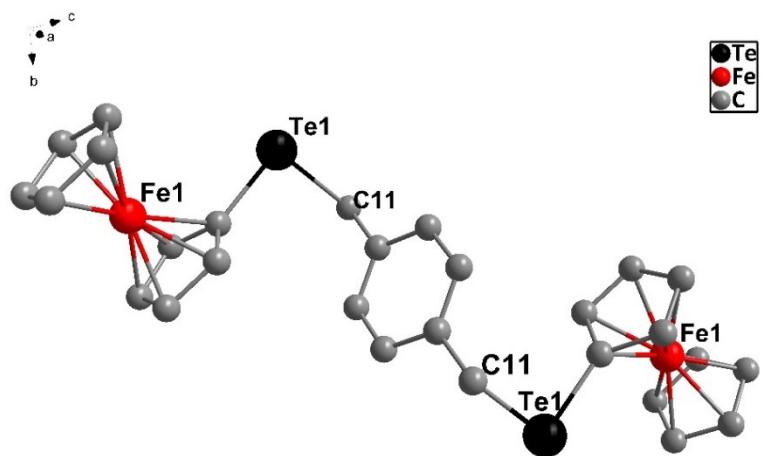
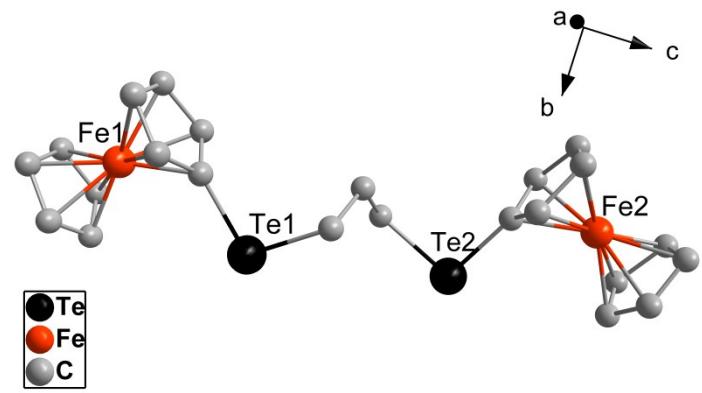


Fig. S1 The molecular structures of **L3** and **L4**.

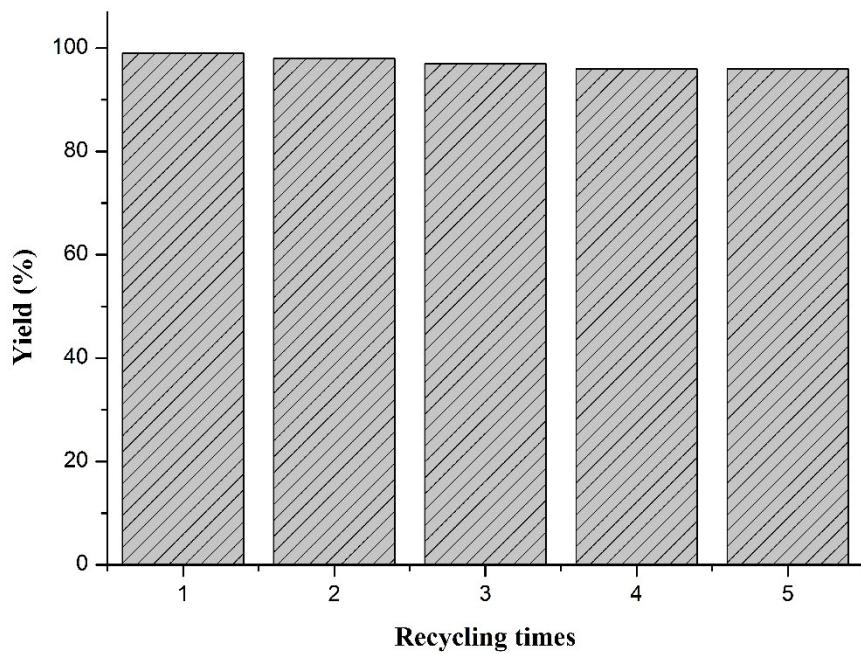


Fig. S2. Recycle of PEG-400/**5** catalytic system.

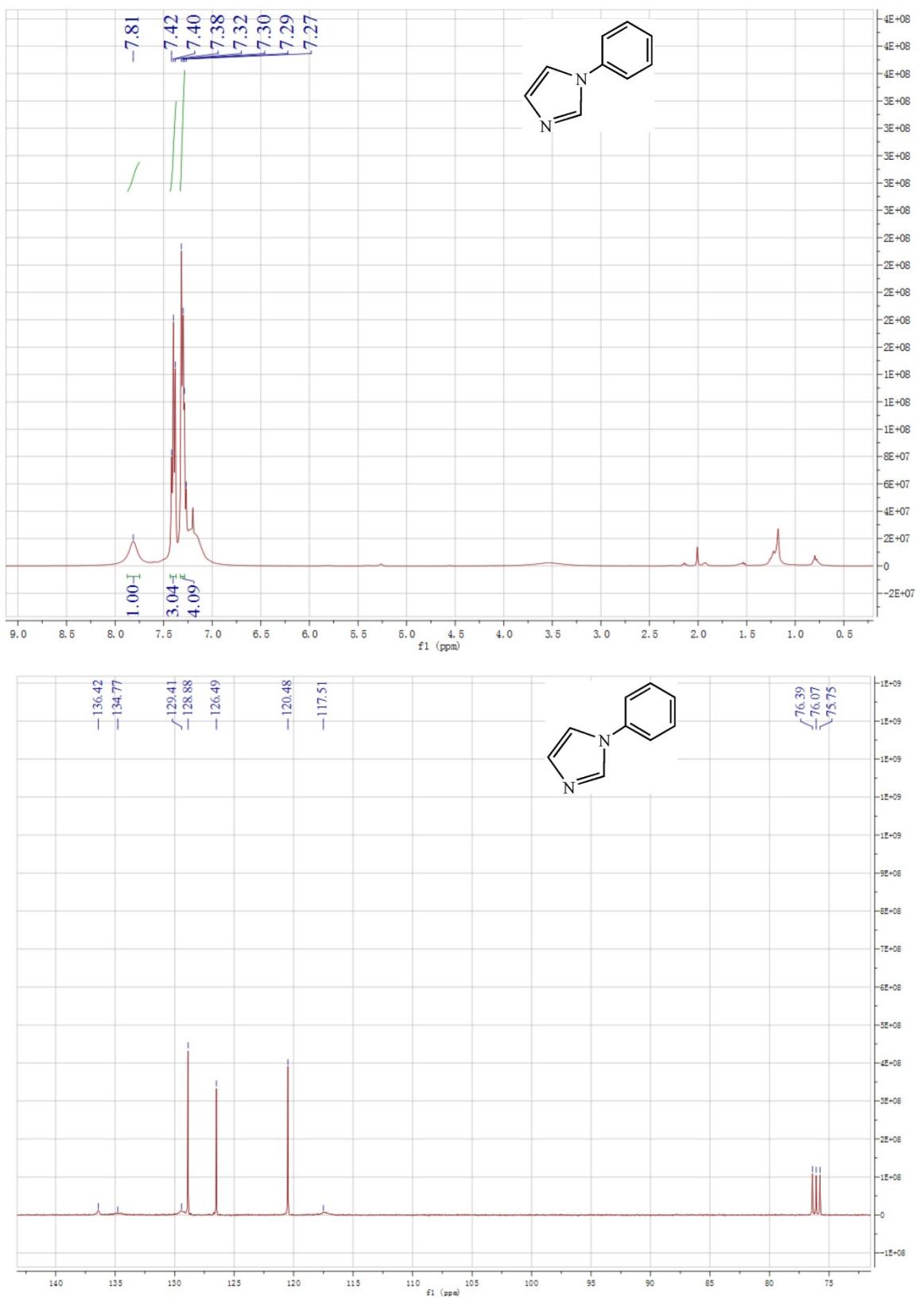


Fig. S3 ^1H and ^{13}C NMR spectra of N-(4-phenyl)imidazole.

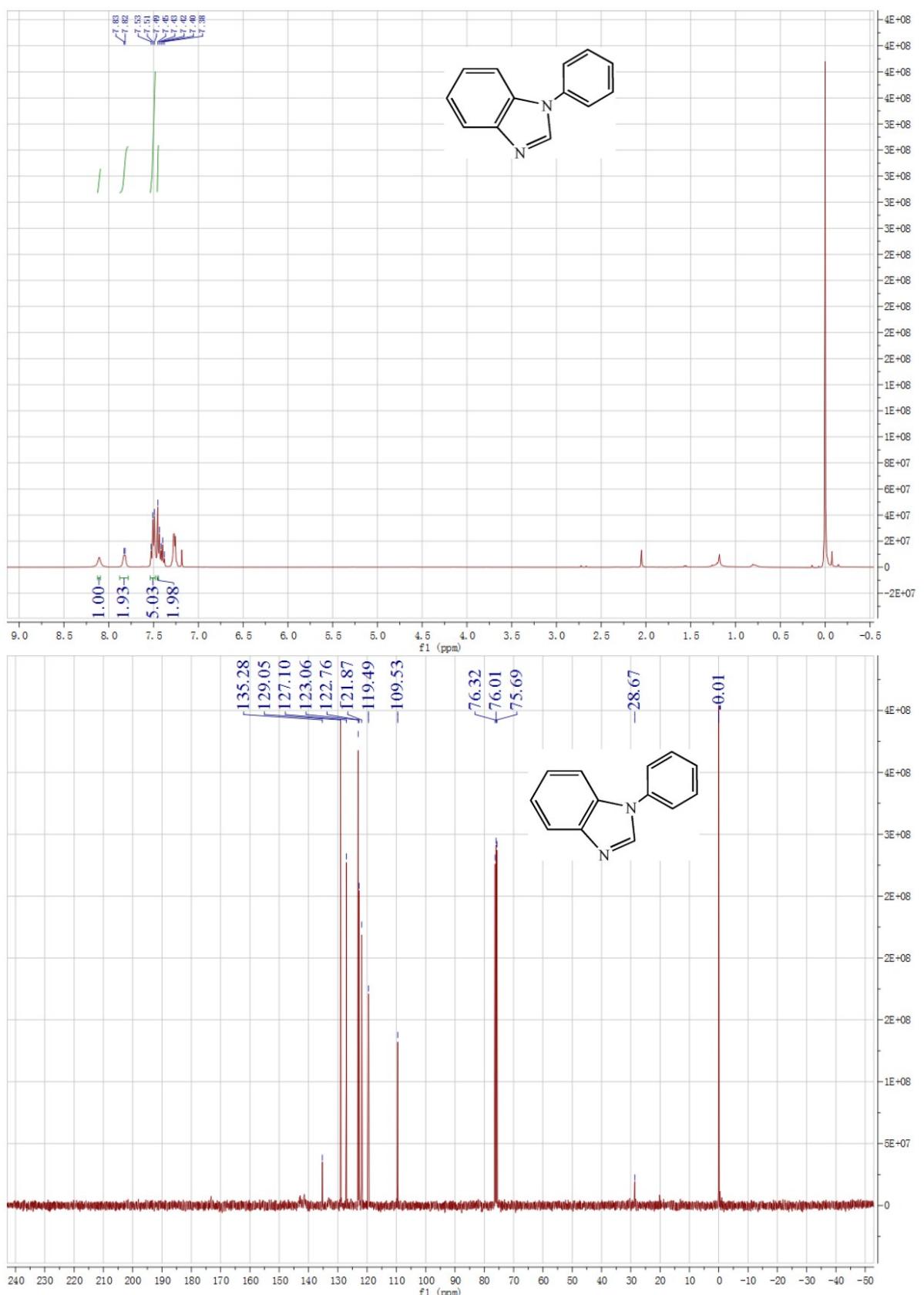


Fig. S4 ^1H and ^{13}C NMR spectra of N-(4-phenyl)benzimidazole.

N-(4-phenyl)4-methyimidazole

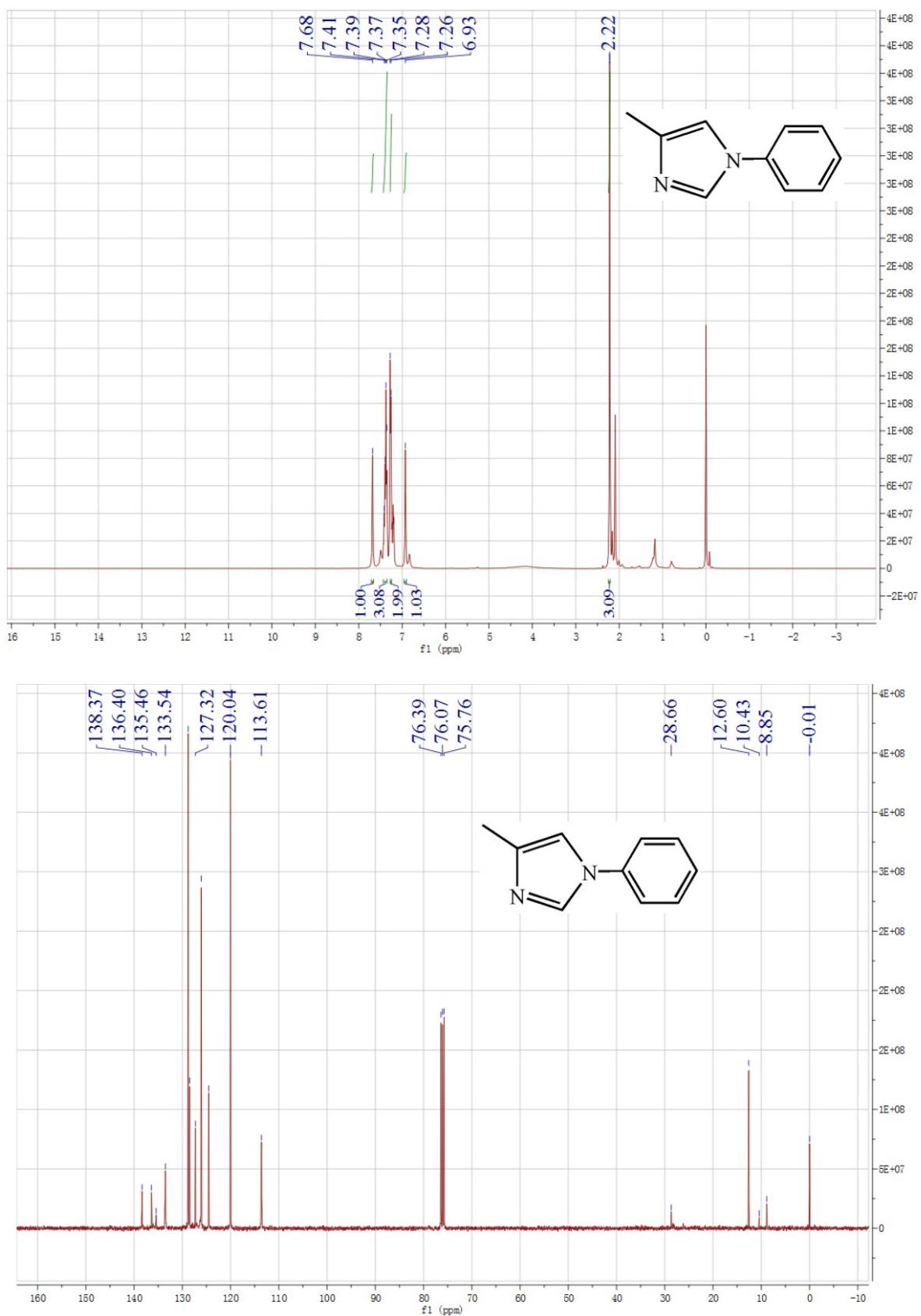


Fig. S5 ¹H and ¹³C NMR spectra of N-(4-phenyl)-4-methyimidazole.

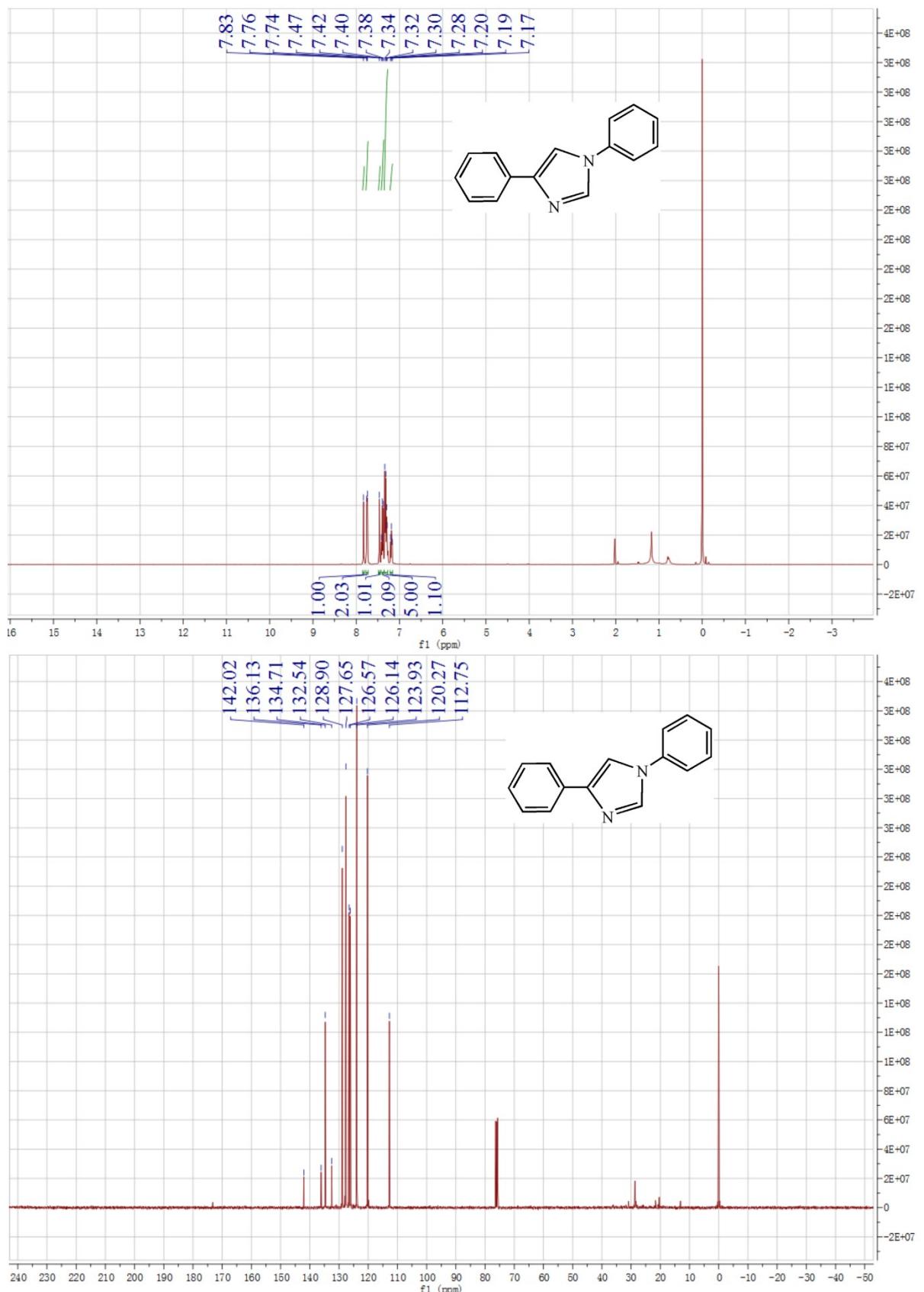


Fig. S6 ^1H and ^{13}C NMR spectra of N-(4-phenyl)-4-phenylimidazole.

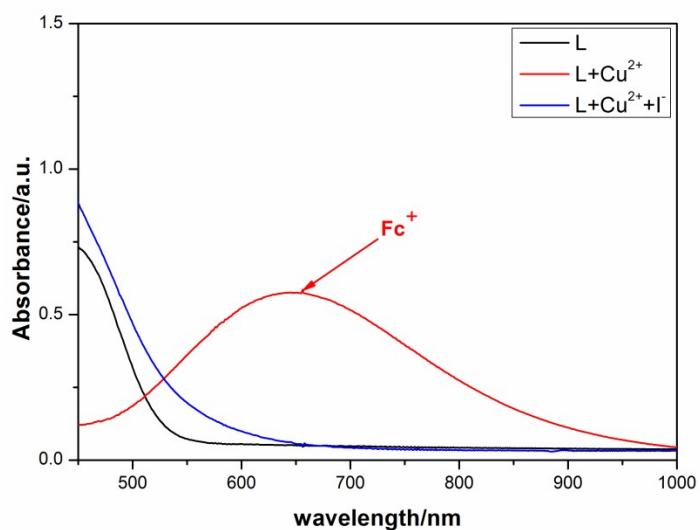


Fig. S7 UV-Vis spectra of free **L3** (1×10^{-4} mol/L), and successive addition of Cu(ClO₄)₂·6H₂O (red line) and KI (blue line) in methanol solution.

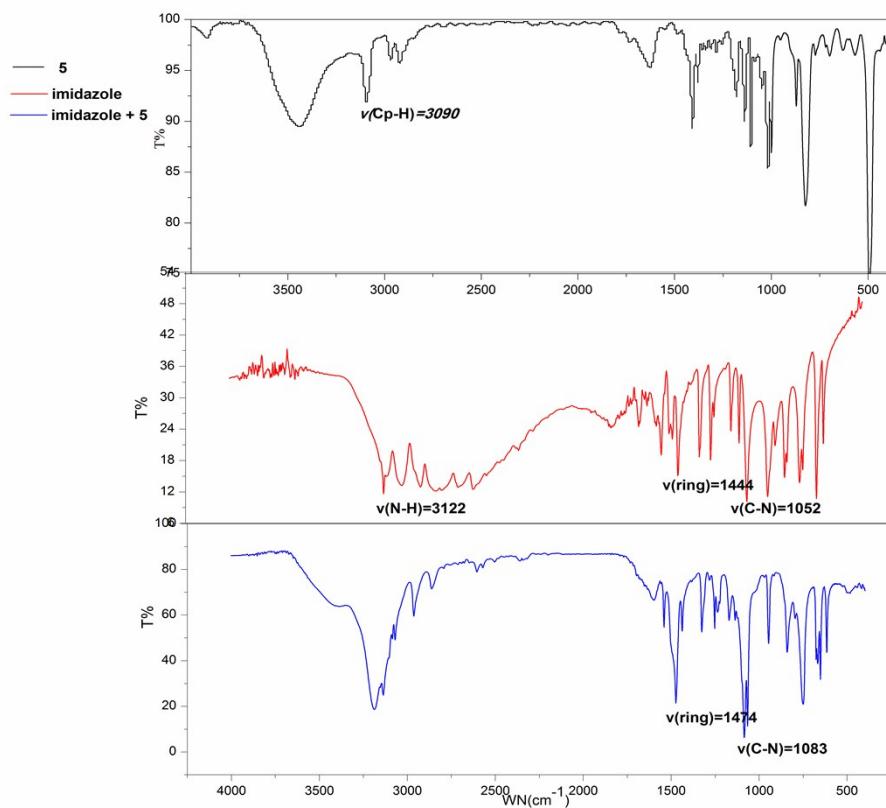


Fig. S8 Infrared spectrum of **5** (black line), imidazole (red line) and intermediates (**5i**, blue line)

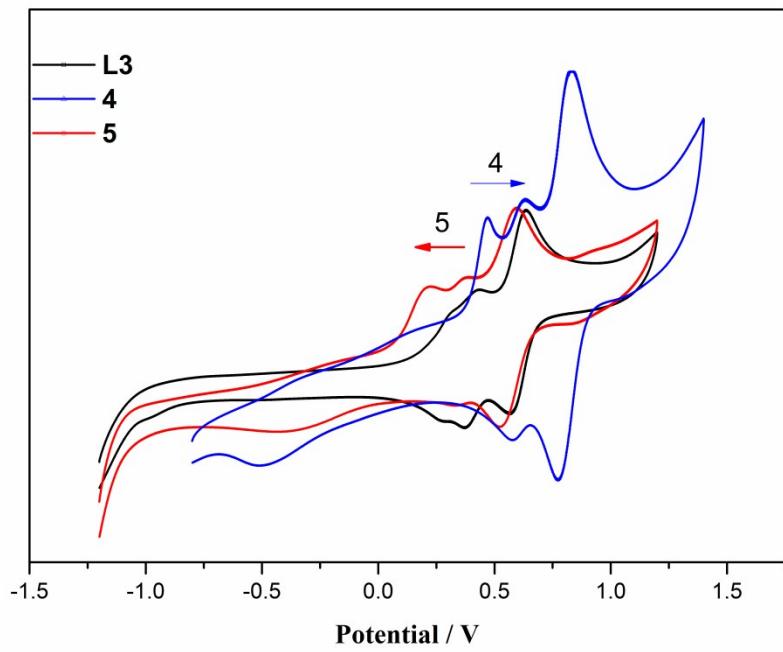


Fig. S9 Cyclic voltammetric of **L3**, **4** and **5**