

*Electronic Supporting Information for*

## **Ruthenium(II)-polypyridyl doped zirconium(IV) metal-organic frameworks as solid-state electrochemiluminescence detectors**

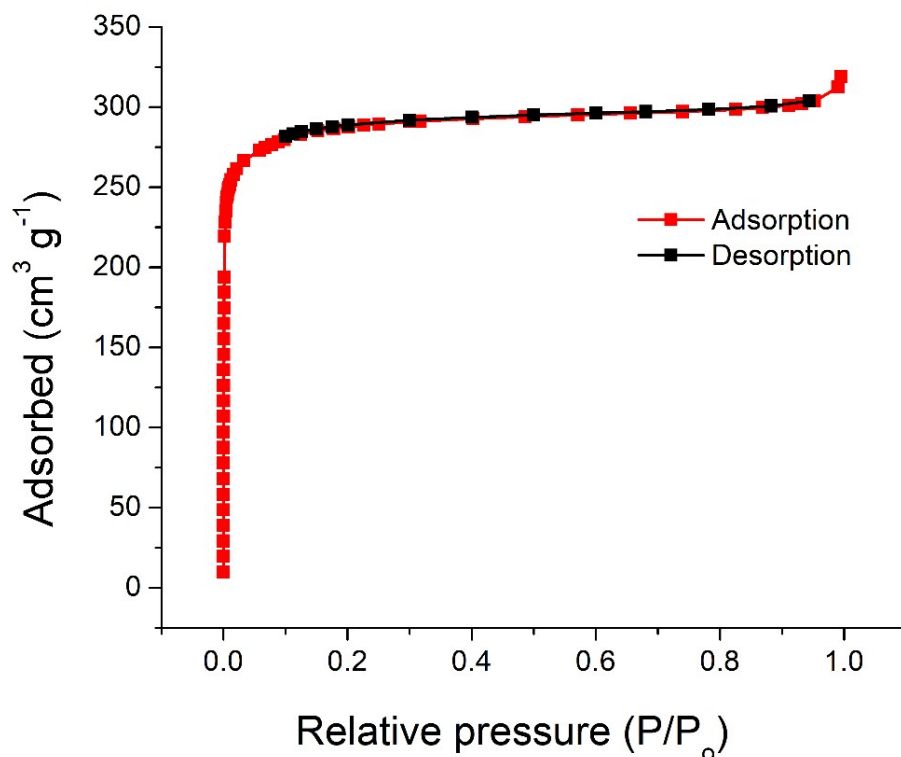
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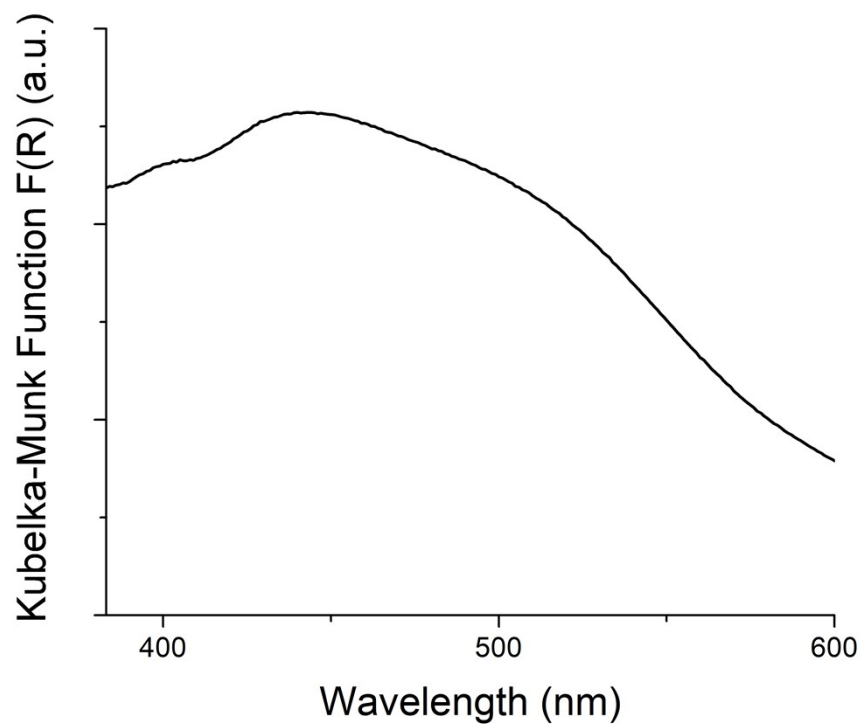
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E-mail: [ajmorris@vt.edu](mailto:ajmorris@vt.edu).

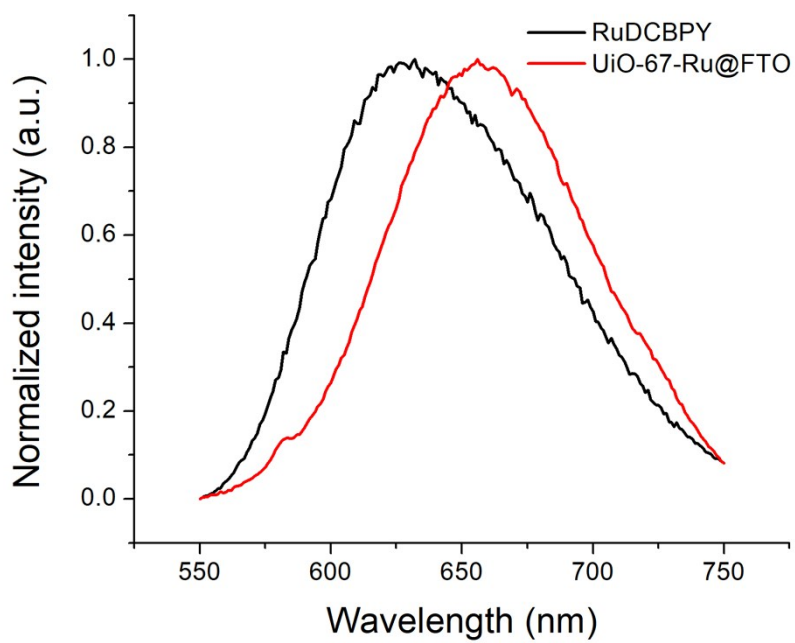
### *Supporting figures & analysis*



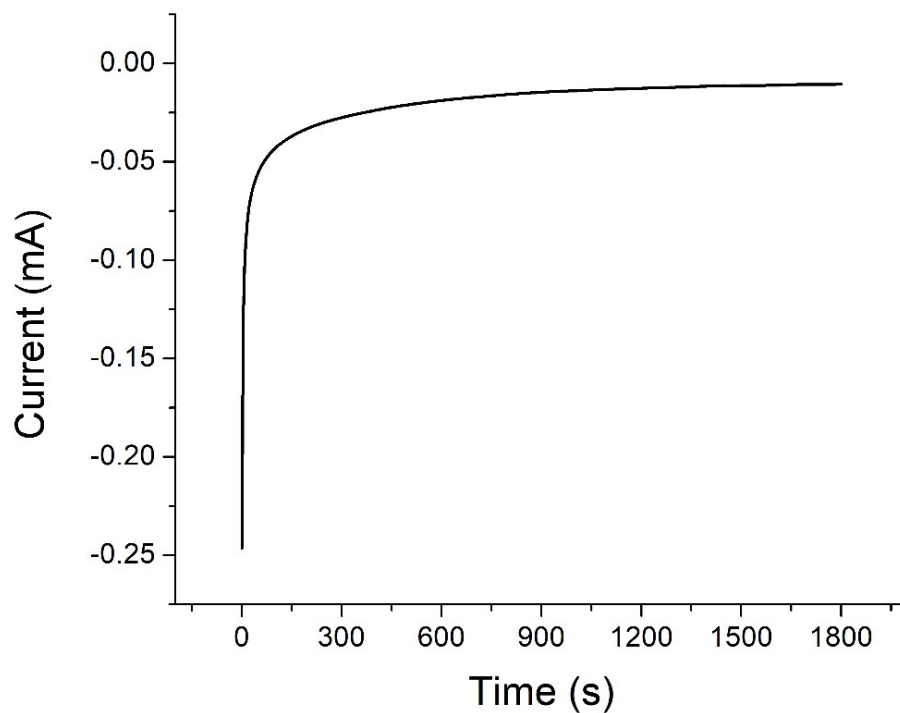
**Fig. S1**  $N_2$  adsorption-desorption isotherm of UiO-67-Ru powders at 77 K, 1 atm.



**Fig. S2** UV-Visible spectrum of UiO-67-Ru@FTO in MeCN



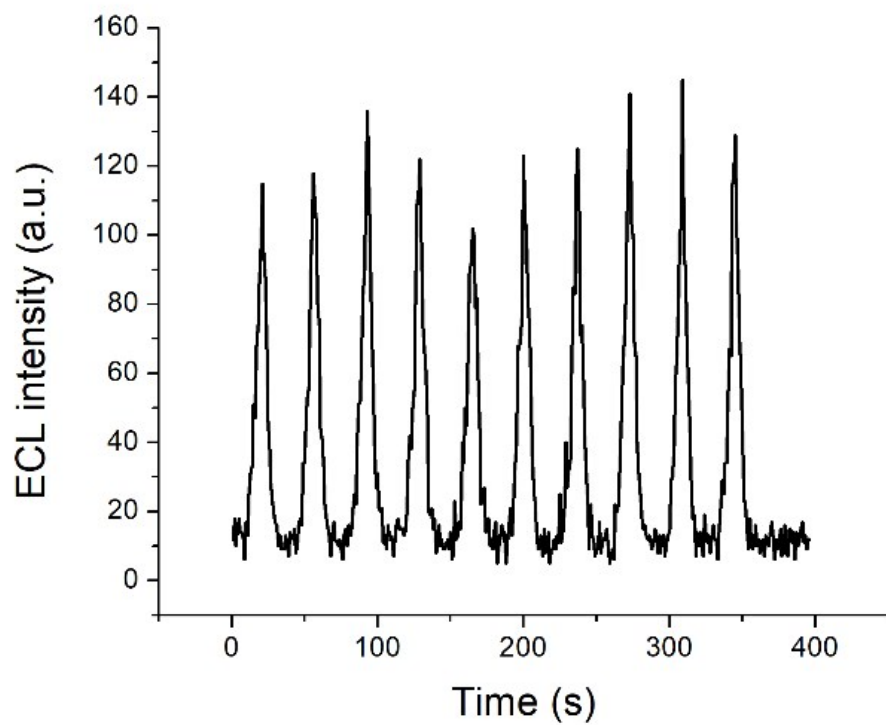
**Fig. S3** Steady-state emission spectra of UiO-67-Ru@FTO and RuDCBPY in MeCN



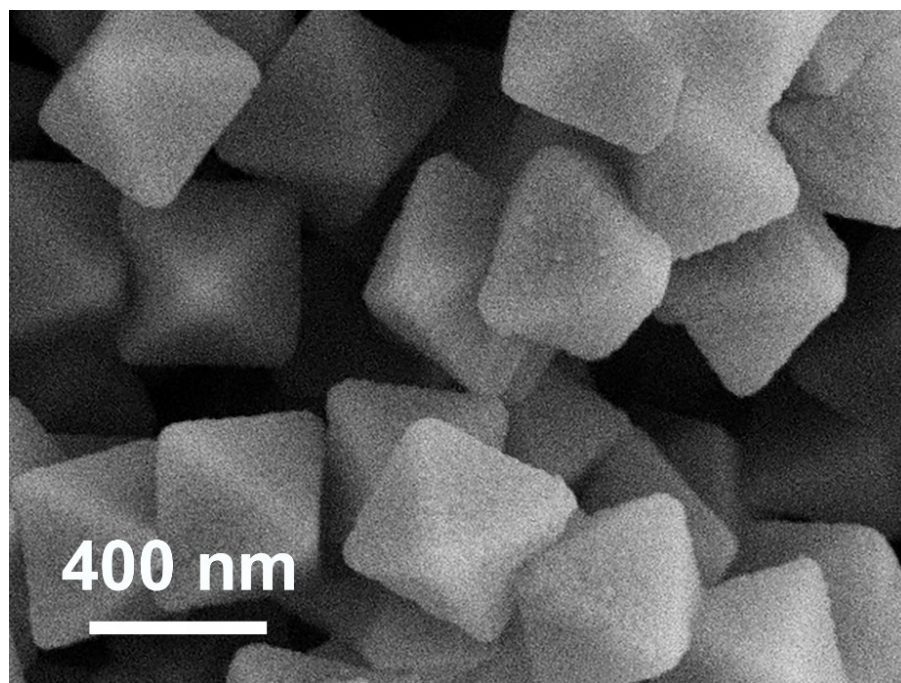
**Fig. S4** Amperometry of UiO-67-Ru@FTO in MeCN with a controlled potential at 1.38 V vs. Fc/Fc<sup>+</sup> for 30 min

To calculate electroactive Ru<sup>2+</sup> coverage, the charge Q can be obtained by integration. Based on the assumption that the charge that passed during electrolysis (0-1800 s) was only due to the one-electron transfer process of Ru<sup>2+</sup>/Ru<sup>3+</sup> redox couple, and all electroactive Ru contribute to the

charge, the amount of Ru can be calculated using equation  $n = \frac{Q}{F}$ , where n is the number of moles of Ru<sup>2+</sup>, Q is the charge, and F is Faraday constant. The coverage of electroactive Ru<sup>2+</sup> is thereby obtained by n over the film area (2.75 cm<sup>2</sup>). The total amount of Ru<sup>2+</sup> can be obtained from ICP-MS of digested films. Finally, the ratio of electroactive coverage calculated from potential-step electrolysis to the total Ru content determined by ICP-MS.



**Fig. S5** ECL of solution-based  $\text{Ru}(\text{bpy})_3^{2+}/\text{TPA}$  system in the presence of 20 mM TPA in MeCN containing 0.1 M  $\text{LiClO}_4$



**Fig. S6** SEM of UiO-67-Ru@FTO after ECL measurement

**Associated with Fig. 4C**

The linear relationship between ECL intensity and TPA concentration ranging from 0.04 mM to 20 Mm can be represented by equation  $I_{\text{ECL}} \text{ (a.u.)} = 129.08857 + 413.62138C_{\text{TPA}} \text{ (mM)}$ , with a correlation coefficient of 0.99747.

**Associated with Fig. 5**

The curve can be fitted to the equation  $I_{\text{ECL}}/I_0 = 1.87112 - 0.74028 \log(C_{\text{DA}}) \text{ (}\mu\text{M)}$  with a correlation coefficient of 0.96036.