

Supplementary materials

Easy sensing of lead and zinc in water using smart glass based on cationic porphyrin layers

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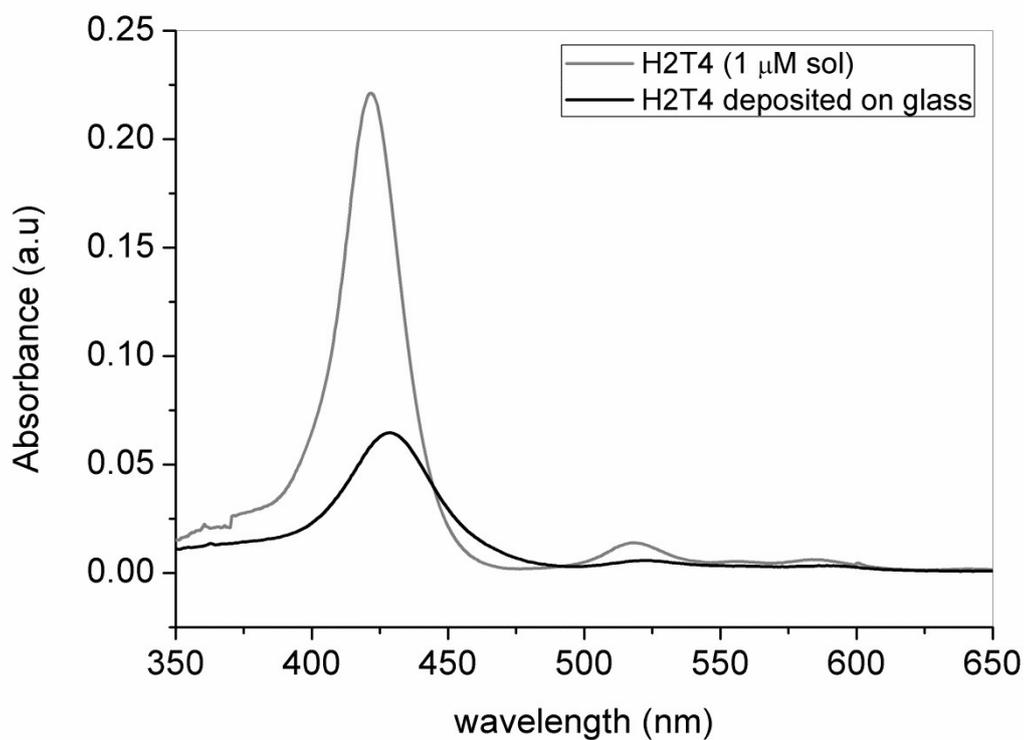


Figure S1: UV-Vis spectra of [H2T4] 1 μM (black line) in ultrapure water pH= 7 (grey line) and of H2T4 deposited on glass (black line).

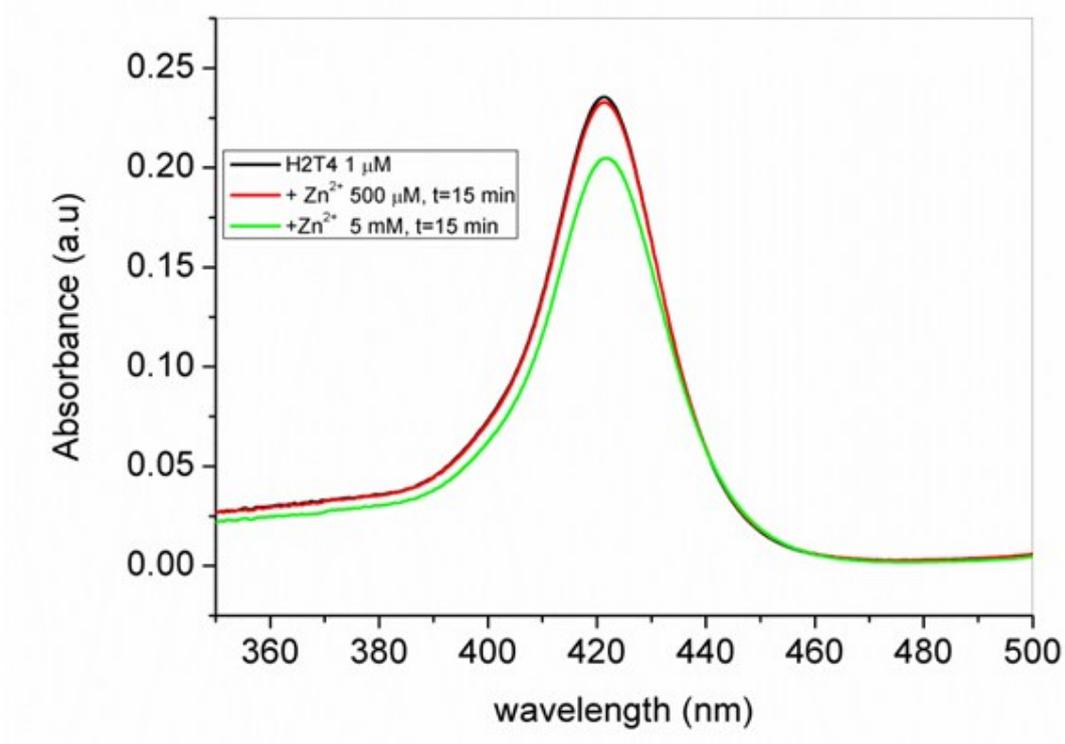


Fig. S2: UV-Vis spectra of [H2T4] 1 μM (black line) in ultrapure water pH= 7 in presence of [Zn²⁺] 500 μM (red line) or 5 mM (green line).

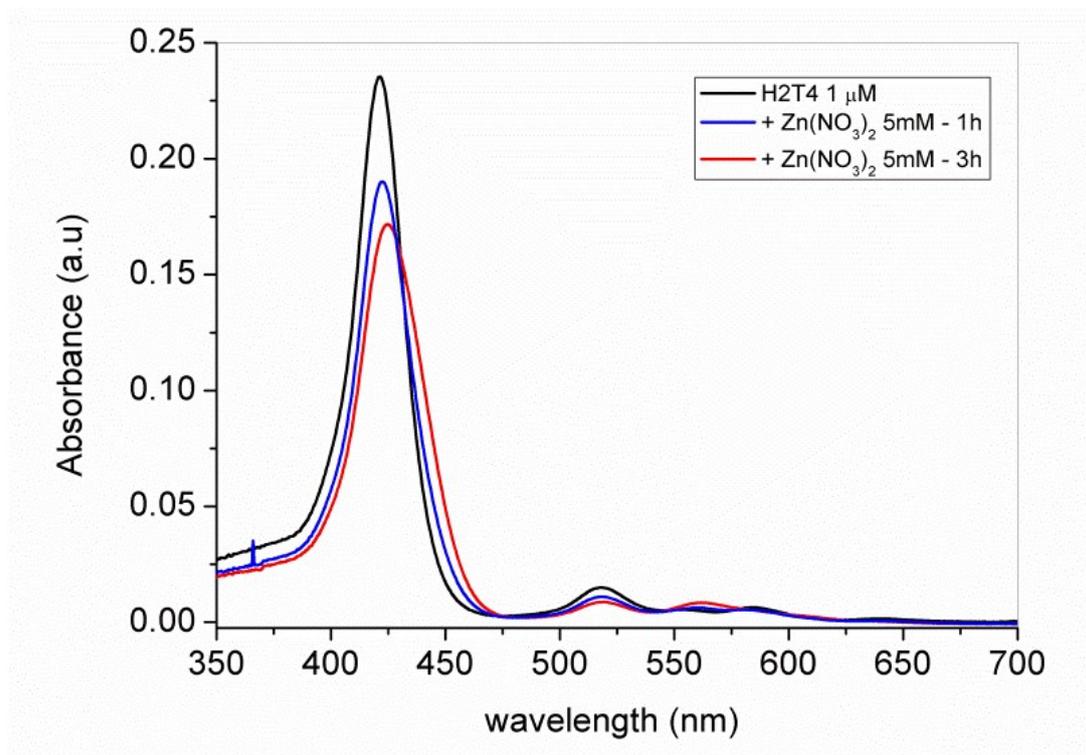


Fig. S3: UV-Vis spectra of [H2T4] 1 μM (black line) in ultrapure water pH= 7; in presence of [Zn²⁺] 5 mM after 1 hour (blue line) and 2 hours (red line).

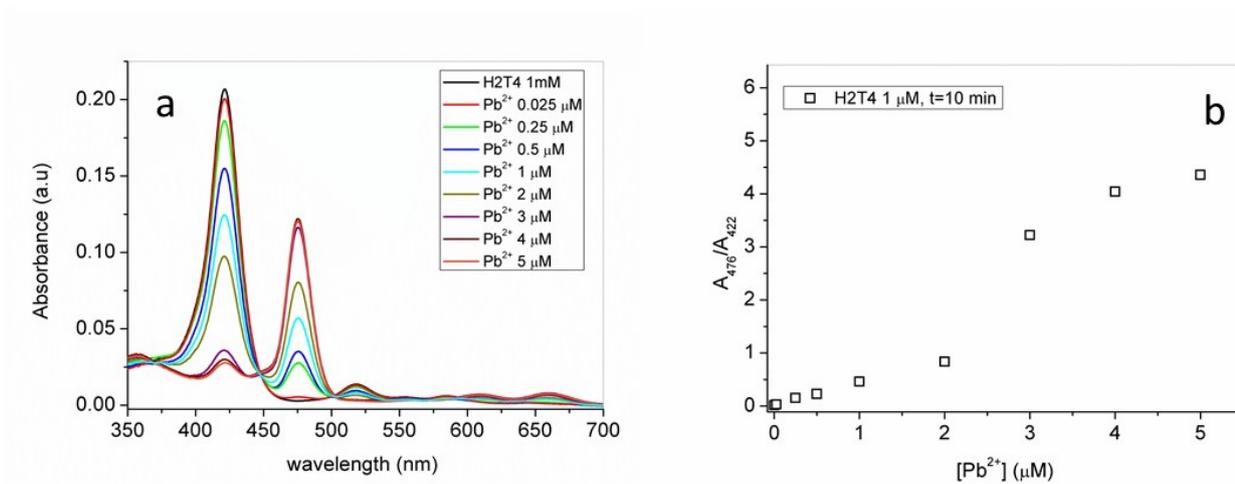


Fig. S4: UV-Vis spectra of H2T4 1 μM in water pH=7 (a) and related plot of A_{476}/A_{422} ratio (b), upon increasing [Pb²⁺]

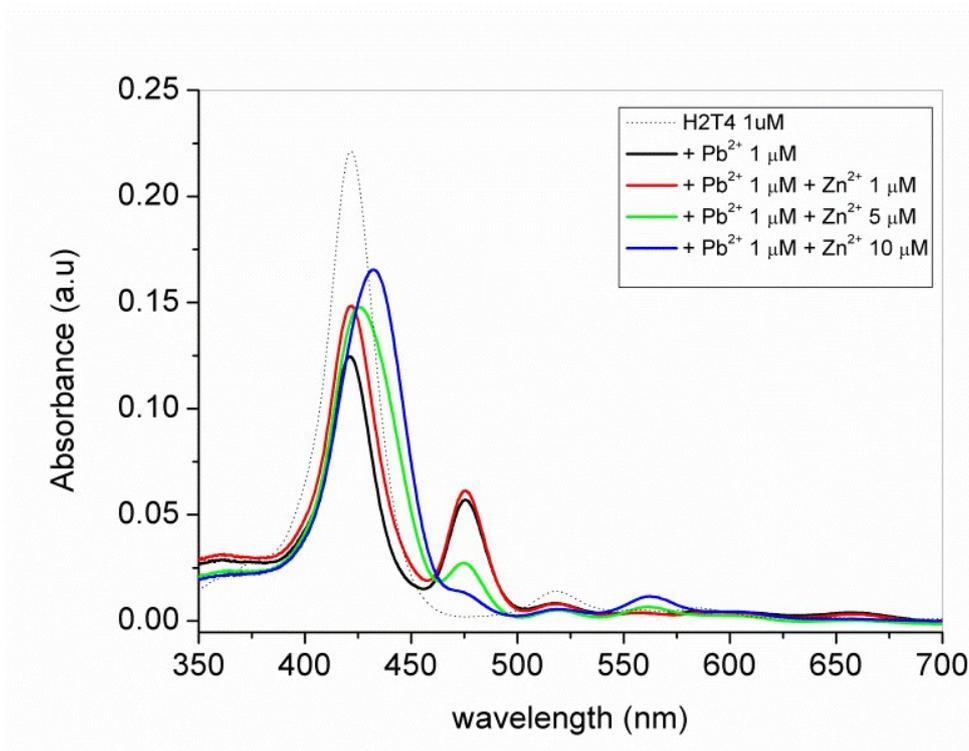


Fig. S5: UV-Vis spectra of bare H2T4 1 μM in water pH=7 (dotted grey line), in presence of $[\text{Pb}^{2+}] = 1 \mu\text{M}$ (black line) and after addition of $[\text{Zn}^{2+}] = 1 \mu\text{M}$ (red line); 5 μM (green line) and 10 μM (blue line).

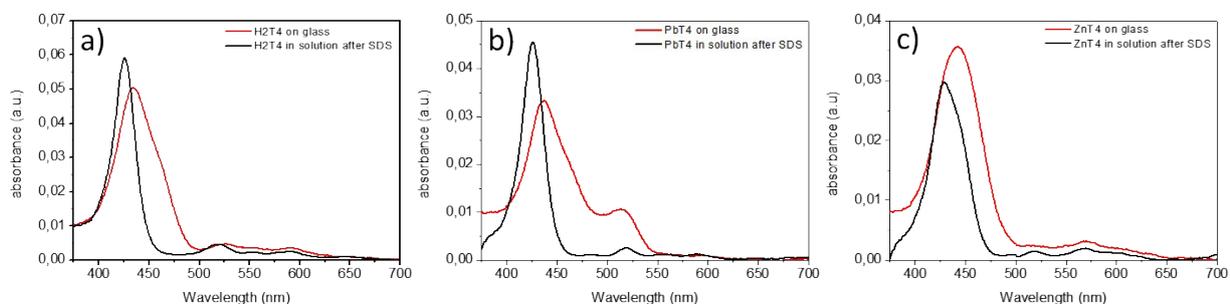


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