

Electronic Supplementary Information

Deep oxidative desulfurization of dibenzothiophene with {Mo₁₃₂} nanoball supported on activated carbon as efficient catalyst at room temperature

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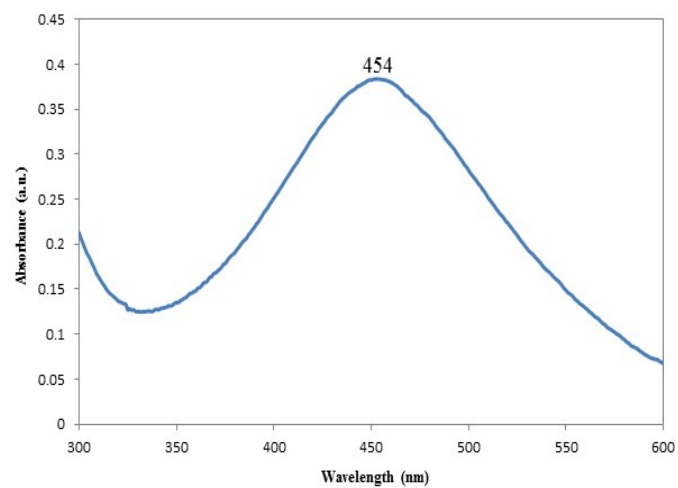


Fig. S1. UV-Vis spectrum of {Mo₁₃₂} aqueous solution

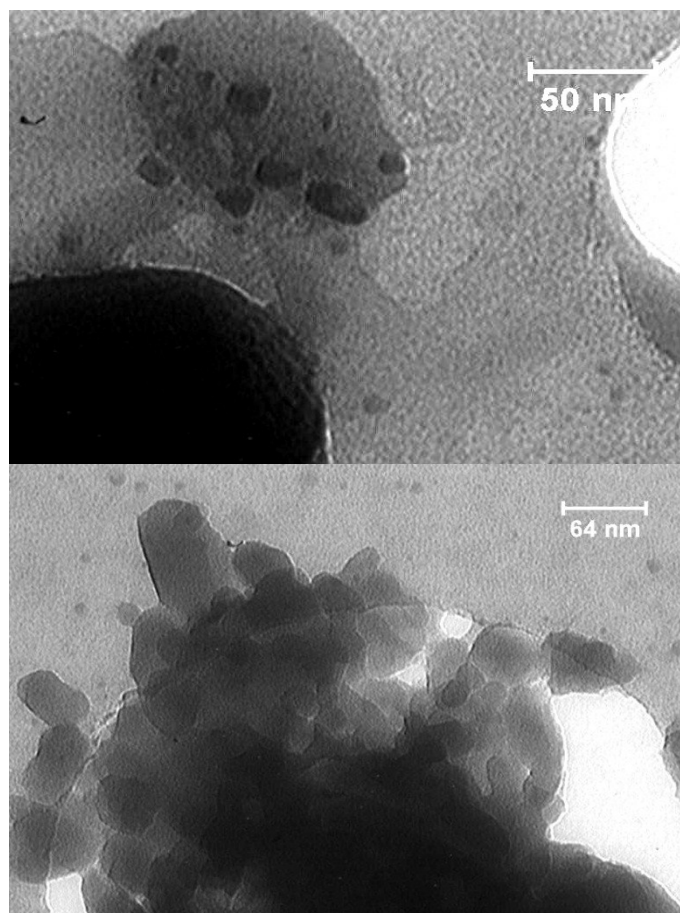


Fig. S2. TEM images of prepared {Mo₁₃₂} in ethanol

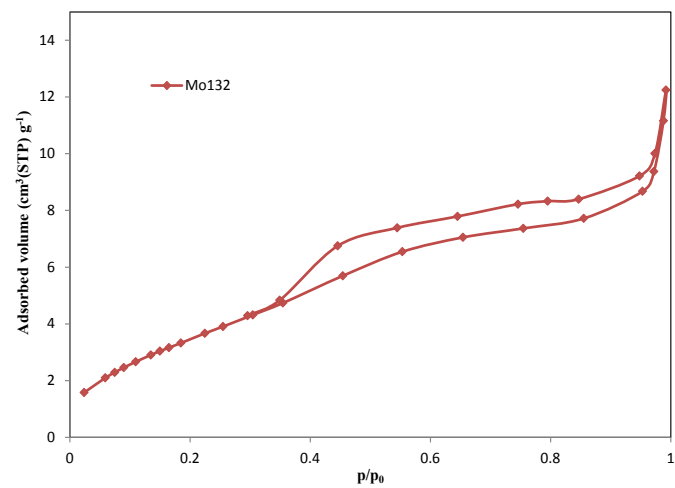


Fig. S3. Nitrogen adsorption/desorption isotherms of {Mo₁₃₂}

H₂-TPR analysis:

The H₂-temperature programmed reduction (H₂-TPR) experiment was performed on 50 mg of the catalyst ($\{\text{Mo}_{132}\}/\text{AC-20}$) placed in a U-shaped quartz reactor. Prior to the TPR run, the catalyst was degassed in flow of 10 sccm Ar at 110°C for 1 hour, and cooled down to 40°C. Then, the sample was heated from 40 to 900°C at a heating rate of 10°C/min in a flow of 5% H₂/Ar mixture (10 sccm).

The H₂-TPR was employed to study the reducibility of the catalyst. As shown in Fig. S4, $\{\text{Mo}_{132}\}/\text{AC-20}$ exhibited two reduction peaks, a sharp peak at 560°C and a weak peak at higher temperature of 710 °C. These two peaks indicate reduction of Mo^{VI} and Mo^{VI} in $\{\text{Mo}_{132}\}$ cluster in the catalyst.

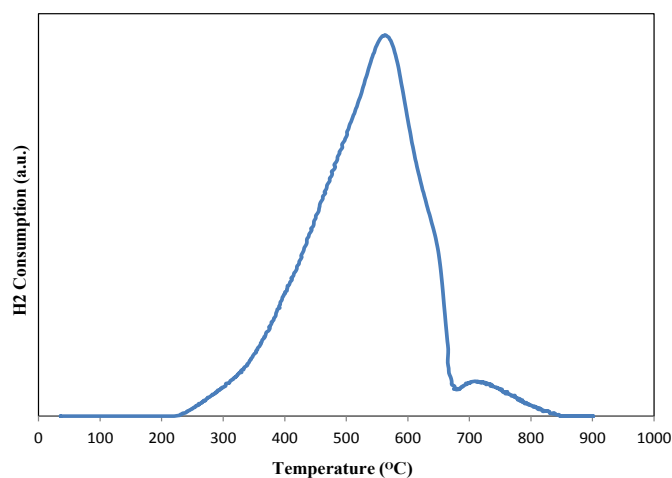
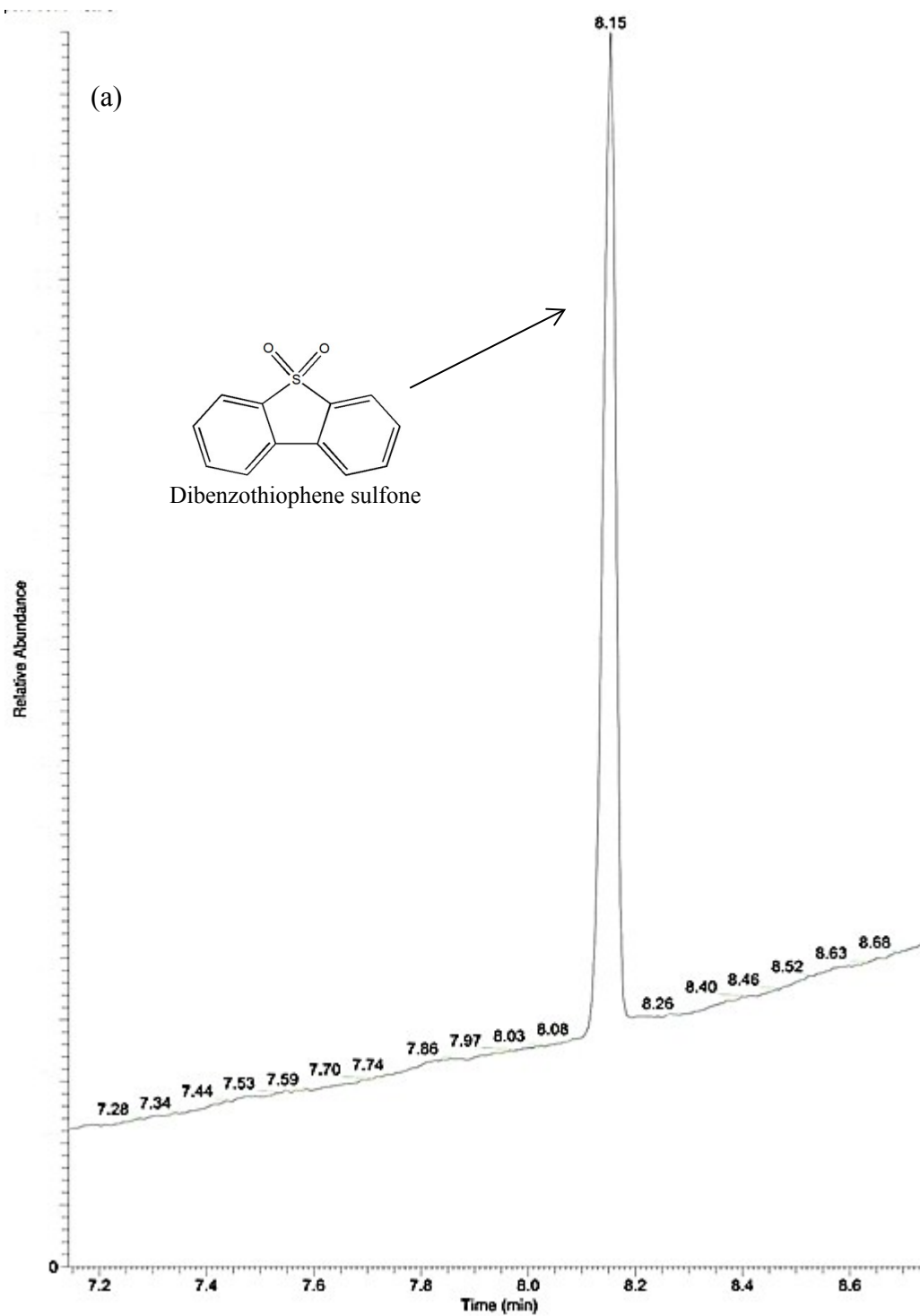


Fig. S4. H₂-TPR profile of $\{\text{Mo}_{132}\}/\text{AC-20}$

Table S1*Reusability of {Mo₁₃₂}/AC-20 and {Mo₁₃₂} content of the catalyst in each step*

Catalyst samples	DBT removal (%)	Actual {Mo ₁₃₂ } contents (%)
Fresh catalyst	99.5	9.04
1 st reuse	98.1	8.59
2 nd reuse	97.8	8.38
3 rd reuse	97.7	8.35



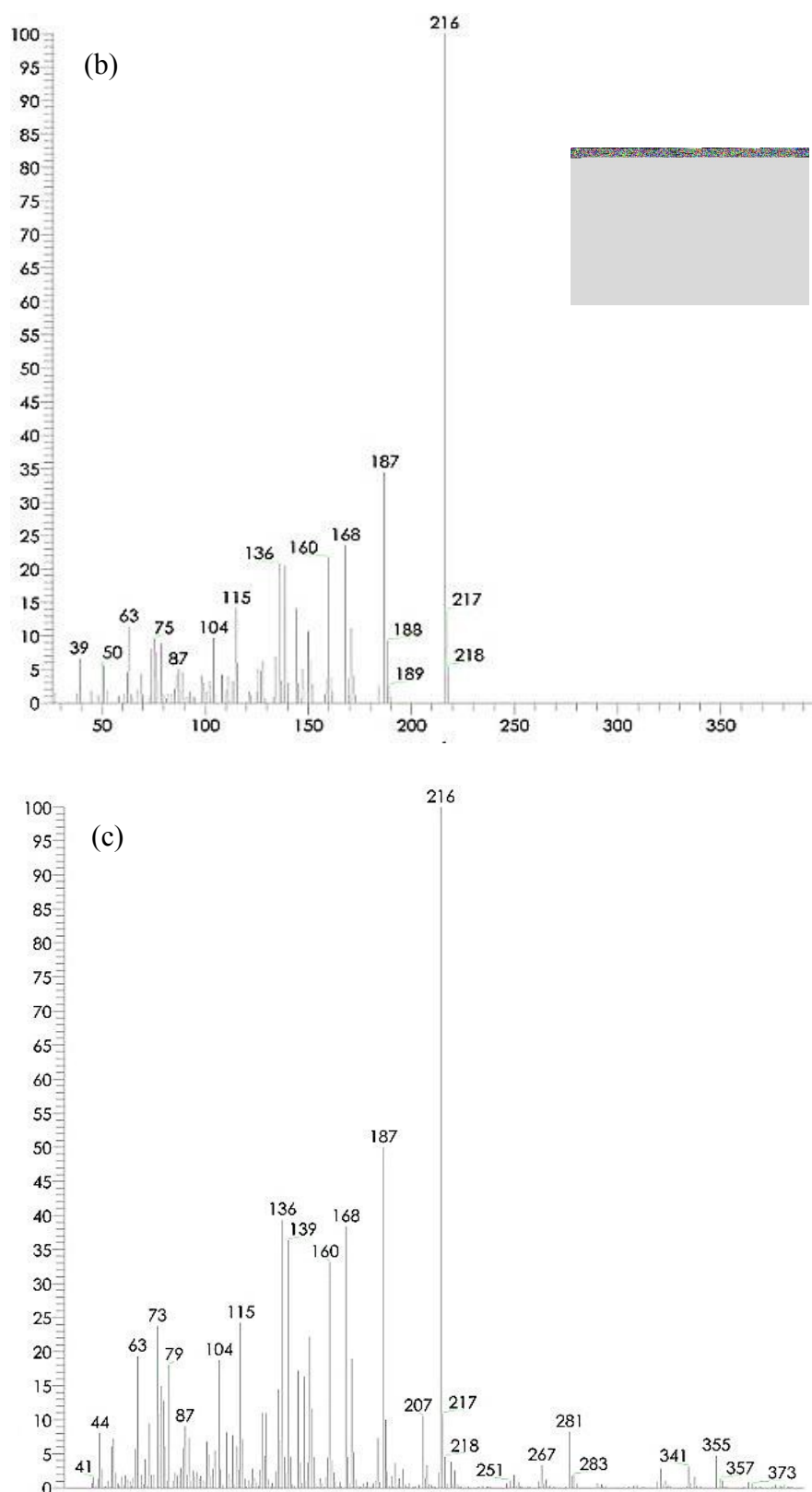


Fig. S5. GC-MS analysis of the acetonitrile phase at the end of the process. (a) GC-MS chromatogram, (b) The standard mass spectrum of DBT sulfone, (c) The mass spectrum of the oxidation product of DBT. Reaction condition: $T = 25\text{ }^{\circ}\text{C}$, $\{\text{Mo}_{132}\}/\text{AC-20}$ catalyst dosage = 0.0025 g cat/ g F, O/S = 10, $t = 30$ min, initial sulfur content = 500 ppm.