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AIE+ESIPT based red fluorescent aggregates for visualization of latent fingerprints

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<u>1.</u> EXPERIMENTAL SECTION



Scheme S1: Synthetic route to 2.







Fig. S2. ¹³C NMR spectrum of DPSA.



Fig. S3. Mass spectrum of DPSA.





Fig. S4. Energy minimized structures of **DPSA** in Gaussian 09 software using DFT method at B3LYP 6-31G basis set.



Fig. S5. Relative total energy of **DPSA** in S_0 state by varying bond length of O-H using TD-DFT calculations at B3LYP /6-31G level.



Fig. S6. Relative total energy of **DPSA** in S_1 state by varying bond length of O-H using TD-DFT calculations at B3LYP /6-31G level.



Fig. S7. (Top) Fluorescence spectrum of DPSA after incremental addition of 10 vol% of H₂O in CH₃CN at a concentration of 10 μ M (λ_{ex} = 390 nm, slit width Ex/Em = 10/20 nm). (Bottom) Fluorescent images of DPSA (0.5 mM) on addition of 0-95% H₂O in CH₃CN under 365 nm UV light.



Fig. S8 (a) UV-Vis Spectra and (b) Fluorescence spectra of DPSA (1 x10⁻⁵ M) in different polarity solvents ($\lambda_{ex} = 390$ nm, slit width Ex/Em = 7/5 nm).



Fig. S9. SEM micrograph of DPSA (10⁻³ M) showing self-assembled spherical morphology in 70% $H_2O:CH_3CN$



Fig. S10. LFP developed with DPSA (1 mM, 90% H₂O-CH₃CN) on aluminium surface.



Fig. S11. LFP developed with DPSA (1 mM, 90% H₂O-CH₃CN) on stainless steel surface.



Fig. S12. LFP developed with DPSA (1 mM, 90% H₂O-CH₃CN) on metal coin surface.



Fig. S13. LFP developed with DPSA (1 mM, 90% H₂O-CH₃CN) on glass.