Electronic Supplementary Information for

Syntheses and structural aspects of six-membered palladacyclic complexes derived from N,N',N''-triarylguanidines with N- or S-thiocyanate ligand

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Materials and methods

Cyclopalladated *N*,*N*'',*N*''-triarylguanidines, $I-V^{1-3}$ and 1,3,5-triaza-7-phosphaadamantane (PTA)⁴ were prepared following the literature procedures. KSCN, 3,5-Me₂C₅H₃N, 4-MeC₅H₄N, P(OCH₂)₃CEt, PMe₃ and P(NMe₂)₃ were procured from commercial vendors and used as received. Solvents were distilled following the standard procedures and stored at ambient condition prior to use. Elemental analyses were performed on an Elementar Analysensysteme GmbH VarioEL V3.00. The IR spectral data were obtained on a Shimadzu IR435 spectrometer using KBr pellet and CHCl₃ solution of the analyte in the frequency range 400–4000 cm⁻¹. ¹H NMR and ¹³C{¹H} NMR spectra were recorded on a Jeol ECX 400 MHz spectrometer operating at field strengths of 400 and 100.5 MHz,

respectively. ¹³C{¹H} NMR spectra were recorded for the saturated CDCl₃ solution of the analyte with at least 5000 scans in each case. Fewer number of carbon signals than anticipated were observed in the ¹³C NMR spectra of **4**, **11**, **13**, **14**, **16** and **17**, presumably due to overlapping peaks as well as low intensity of some of the peaks of the minor isomers of **4** and **16**. ³¹P{¹H} NMR spectra were recorded on a Jeol ECX 400 NMR spectrometer operating at 161.8 MHz with 85% H₃PO₄ as the external standard. The electrospray ionization mass spectra (ESI–MS) were recorded on Waters Q-TOF Premier Micromass, Bruker Micro TOF-Q II and MicroMass Quattro II instruments. MALDI-TOF mass spectra were recorded on Applied Biosystems 4700 Proteomics Analyser 34800031 using α -cyano-4-hydroxycinnamic acid (CHCA) as matrix with MeOH or MeCN as carrier solvent. Melting points were measured on a Perkin Elmer DSC instrument (Model: Pyrex 6).

X-ray crystallography

Intensity data of suitably sized crystals of **2**, **4**, **5**·CH₂Cl₂, **6**·PhMe, **9**·MeOH, **10**·PhMe, **11**·1/3CH₂Cl₂·1/3H₂O, **13**·PhMe, **15**·2PhMe, **16**·CHCl₃, **17** and **18**·PhMe were collected on an Oxford Xcalibur S diffractometer (4-circle κ goniometer, Sapphire-3 CCD detector, ω scans, graphite monochromator, and a single wavelength enhanced X-ray source with MoK α radiation).⁵ Pre-experiment, data collection, data reduction, and absorption corrections were performed with the CrysAlisPro software suite.⁶ Intensity data of suitably sized crystals of **8**·2/3PhMe was collected on a Bruker AXS SMART-APEX diffractometer with a CCD area detector, graphite monochromator. The frames were collected by ω , ϕ , and 20 rotation at 10 s per frame with SMART.⁷ The measured intensities were reduced to F^2 and corrected for absorption with SADABS.⁸ The structures of complexes were solved by direct methods using SIR 92,⁹ which revealed the atomic positions, and refined using the SHELX-97 program package¹⁰ and SHELXL97 (within the WinGX program package¹¹). Non-hydrogen atoms were refined anisotropically. C–H/N–H hydrogen atoms were placed in geometrically calculated positions by using a riding model. The molecular structures were created with the Olex2 program.¹² Images of Hydrogen-bond interactions in the cystal lattice of **8**·MeOH was created with the Diamond program.¹³

Computational details

The geometries of the momomeric and dimeric NCS and SCN linkage isomers were optimized using the density functional theory (B3LYP).^{14–16} N, P, O, S, C and H atoms were described using the 6-31G* basis set¹⁷ and the metal atom Pd was described using the LANL2DZ basis set.^{18–21} The solvent correction (for methanol) was carried out using the polarized continuum mode.^{22–24} The computational analysis was done using the Gaussian 09 software.²⁵

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	2	4	$5 \cdot CH_2Cl_2$	6.PhMe	8 ·2/3PhMe
Formula	$C_{46}H_{44}N_8S_2Pd_2$	$C_{52}H_{56}N_8S_2Pd_2$	C ₃₁ H ₃₃ Cl ₂ N ₅ O ₃ SPd	C37H39N5SPd	C33.67H39.33N7O3SPPd
Fw	985.81	1069.97	732.98	692.19	759.48
Temperature (K)	298(2)	298(2)	298(2)	298(2)	298(2)
Wavelength (λ)	0.71073	0.71073	0.71073	0.71073	0.71073
Crystal system	Orthorhombic	Monoclinic	Monoclinic	Triclinic	Triclinic
Space group	Pbca	$P2_{1}/c$	$P2_{1}/n$	$P\overline{I}$	ΡĪ
<i>a</i> (Å)	14.0567(6)	11.6247(5)	14.0233(5)	10.1830(7)	13.502(5)
<i>b</i> (Å)	16.0971(7)	14.8522(5)	13.5565(5)	11.6661(5)	14.691(5)
<i>c</i> (Å)	18.918(1)	14.4971(6)	16.9737(8)	14.8821(5)	29.759(5)
α (deg)	90	90	90	69.549(3)	97.088(5)
β (deg)	90	91.424(4)	89.870(4)	78.925(4)	92.815(5)
γ (deg)	90	90	90	74.569(5)	112.468(5)
Volume ($Å^3$)	4280.6(3)	2502.2(2)	3226.8(2)	1587.1(1)	5383(3)
Ζ	4	2	4	2	6
ρ_{calcd} (g cm ⁻³)	1.530	1.420	1.509	1.448	1.406
F(000)	2000	1096	1496	716	2348
μ (Mo K α) (mm ⁻¹)	0.981	0.845	0.845	0.685	0.663
θ range (deg)	2.90-26.37	3.08-26.37	2.91-26.37	2.94-26.37	1.39-26.37
No. of reflns collected	19693	19401	14116	12051	78394
No. of reflns used	4371	5119	6421	6488	21700
Parameters	265	291	393	353	1261
$\mathbf{R}_1 \left[\mathbf{I} > 2\sigma(\mathbf{I}) \right]^a$	0.0438	0.0603	0.0430	0.0483	0.0578
wR_2 (all reflns) ^b	0.1214	0.1497	0.1249	0.1335	0.2205
GooF on F^{2c}	1.184	1.212	0.986	1.036	1.065
Largest diff peak/hole ($e \cdot Å^{-3}$)	0.590/- 0.445	1.058/-0.345	1.366/-1.264	1.045/-0.599	1.666/-1.098

Table S1 Crystallographic data for 2, 4, 5·CH₂Cl₂, 6·PhMe and 8·2/3PhMe

 ${}^{a}R_{1} = \Sigma ||F_{o}| - |F_{c}||/\Sigma |F_{o}|; {}^{b}wR_{2} = \{\Sigma [w(F_{o}^{2} - F_{c}^{2})^{2}]/\Sigma [w(F_{o}^{2})^{2}]\}^{1/2}; {}^{c}S = \{\Sigma [w(F_{o}^{2} - F_{c}^{2})^{2}]/(n-p)\}^{1/2}$

	9.MeOH	10-PhMe	$11 \cdot 1/3 CH_2 Cl_2 \cdot 1/3 H_2 O$	13-PhMe
Formula	$C_{30}H_{38}N_7O_4SP$	C ₃₆ H ₄₂ N ₇ SPPd	$C_{29.33}H_{28.33}N_4O_{6.33}SPCl_{0.67}Pd$	C33H39N4O3PSPd
	Pd			
Fw	730.10	742.20	731.29	709.11
Temperature (K)	298(2)	298(2)	298(2)	298(2)
Wavelength (λ)	0.71073	0.71073	0.71073	0.71073
Crystal system	Monoclinic	Triclinic	Triclinic	Triclinic
Space group	C2/c	$P\overline{I}$	$P\overline{I}$	$P\overline{I}$
a (Å)	28.988(2)	10.9066(5)	10.7843(4)	9.2297(3)
$b(\text{\AA})$	11.4580(7)	12.8213(6)	19.7675(6)	11.1576(4)
<i>c</i> (Å)	20.238(2)	12.8595(6)	23.7030(7)	16.1202(5)
α (deg)	90	72.737(4)	101.376(3)	72.627(3)
β (deg)	104.786(8)	73.456(4)	99.035(2)	76.087(3)
$\gamma(\text{deg})$	90	69.959(4)	95.764(3)	85.936(2)
Volume ($Å^3$)	6499.1(8)	1579.3(1)	4847.0(3)	1537.87(9)
Ζ	8	2	2	2
ρ_{calcd} (g cm ⁻³)	1.492	1.561	1.503	1.531
F(000)	3008	768	2228	732
μ (Mo K α) (mm ⁻¹)	0.731	0.745	0.791	0.765
θ range (deg)	2.83-26.37	3.27-26.37	3.03–26.37	2.97-26.37
No. of reflns collected	26060	22288	74879	22504
No. of reflns used	6638	6435	19822	6282
Parameters	401	372	1172	344
$R_1 [I > 2\sigma(I)]^a$	0.0258	0.0537	0.0577	0.0337
wR_2 (all reflns) ^b	0.0632	0.1434	0.1544	0.0977
GooF on F^{2c}	1.071	1.104	1.062	1.078
Largest diff peak/hole (e·Å ⁻³)	0.579/-0.401	1.130/-1.484	1.510/-1.222	0.746/-0.716
$^{a}\mathbf{R}_{1} = \Sigma F_{o} - F_{c} \Sigma F_{o} ; ^{b}\mathbf{w}$	$\mathbf{R}_2 = \{ \Sigma [w(F_o^2 - I)] \}$	$F_c^{2})^2] / \Sigma [w(F_o^{2})^2] \}$	^{1/2} ; ^c S = { $\Sigma[w(F_o^2 - F_c^2)^2]/(n-p)$	$\}^{1/2}$

 $Table \ S2 \ Crystallographic \ data \ for \ 9 \cdot MeOH, \ 10 \cdot PhMe, \ 11 \cdot 1/3 CH_2 Cl_2 \cdot 1/3 H_2 O \ and \ 13 \cdot PhMe$

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	15.2PhMe	16 ·CHCl ₃	17	18-PhMe
Formula	C43H56N7O3PSPd	C ₃₀ H ₄₁ N ₇ PSCl ₃ Pd	$C_{29}H_{40}N_4O_3P_2SPd$	$C_{36}H_{48}N_4P_2SPd$
Fw	888.38	775.48	693.05	737.18
Temperature (K)	298(2)	298(2)	298(2)	298(2)
Wavelength (λ)	0.71073	0.71073	0.71073	0.71073
Crystal system	Triclinic	Triclinic	Orthorhombic	Monoclinic
Space group	$P\overline{\mathrm{I}}$	$P\overline{I}$	Pbca	Cc
<i>a</i> (Å)	10.9588(5)	11.6013(5)	11.3614(3)	28.592(2)
<i>b</i> (Å)	13.9322(8)	12.8550(6)	13.7572(3)	9.0068(2)
<i>c</i> (Å)	15.0765(6)	13.2434(7)	41.720(1)	19.446(1)
α (deg)	112.290(4)	110.543(4)	90	90
β (deg)	97.668(4)	98.363(4)	90	129.889(10)
$\gamma(\text{deg})$	108.324(5)	96.758(3)	90	90
Volume (Å ³)	1936.4(2)	1799.2(2)	6520.8(3)	3842.4(4)
Ζ	2	2	8	4
$\rho_{\text{calcd}} (\text{g cm}^{-3})$	1.524	1.431	1.412	1.274
F(000)	928	796	2864	1536
μ (Mo K α) (mm ⁻¹)	0.627	0.872	0.766	0.649
θ range (deg)	2.88-26.37	3.22-26.37	2.93-26.37	3.08-26.37
No. of reflns collected	24363	26544	92009	25542
No. of reflns used	7924	7346	6661	7854
Parameters	419	396	361	407
$R_1 [I > 2\sigma(I)]^a$	0.0476	0.0473	0.0490	0.0250
wR ₂ (all reflns) ^b	0.1394	0.1348	0.1048	0.0654
GooF on $F^{2 c}$	1.077	1.034	1.225	1.083
Largest diff peak/hole (e·Å ⁻³)	1.050/-0.664	0.800/-0.685	0.743/-1.631	0.401/-0.514
${}^{a}\mathbf{R}_{1} = \Sigma F_{o} - F_{c} / \Sigma F_{o} ; {}^{b}\mathbf{w}$	$\mathbf{R}_{2} = \{ \Sigma [w (F_{o}^{2} - F_{c}^{2})]$	$(2^{2})^{2} [w(F_{o}^{2})^{2}]^{1/2}; ^{c}S =$	$= \{ \Sigma [w(F_o^2 - F_c^2)^2] / (w(F_o^2 - F_c^2)^2) \} $	(n-p) ^{1/2}

Table S3 Crystallographic data for 15·2PhMe, 16·CHCl3, 17 and 18·PhMe

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	2	4
N(4)–Pd(1)	2.112(4)	2.105(4)
N(1)–Pd(1)	2.055(3)	2.052(4)
C(17)-Pd(1)/C(19)-Pd(1)	1.989(4)	1.999(5)
Pd(1)–S(1)	2.324(1)	2.336(1)
C(23)-S(1)/C(26)-S(1)	1.660(4)	1.655(5)
C(23)-N(4)/C(26)-N(4)	1.138(5)	1.149(6)
N(1)–C(1)	1.287(5)	1.315(6)
N(2)–C(1)	1.360(5)	1.350(7)
N(3)–C(1)	1.360(5)	1.345(7)
N(4) - Pd(1) - S(1)	91.79(9)	91.2(1)
C(17)-Pd(1)-S(1)/C(19)-Pd(1)-S(1)	87.5(1)	88.0(1)
C(17) - Pd(1) - N(1)	87.2(1)	88.8(2)
N(1) - Pd(1) - N(4)	93.5(1)	92.3(2)
N(1)-Pd(1)-S(1)	174.44(9)	173.7(1)
C(17)-Pd(1)-N(4)/C(19)-Pd(1)-N(4)	179.2(2)	176.8(2)
N(4)-C(23)-S(1)/N(4)-C(26)-S(1)	179.2(4)	178.2(5)

Table S4 Selected bond distances (Å) and angles (deg) for ${\bf 2}$ and ${\bf 4}$

Table S5 Selected bond distances (Å) and angles (deg) for $5 \cdot CH_2Cl_2$ and $6 \cdot PhMe$

	$5 \cdot CH_2Cl_2$	6-PhMe
Pd(1)–C(17)	1.986(3)	1.995(4)
Pd(1) - N(1)	2.024(3)	2.014(3)
Pd(1) - N(5)	2.159(3)	2.154(3)
Pd(1) - N(4)	2.018(3)	2.007(4)
N(4)–C(23)	1.117(5)	1.152(6)
S(1)–C(23)	1.621(5)	1.617(5)
N(1)-C(1)	1.308(4)	1.306(5)
N(2)-C(1)	1.351(4)	1.354(5)
N(3)-C(1)	1.344(4)	1.358(4)
C(17) - Pd(1) - N(1)	88.7(1)	88.6(1)
N(1)-Pd(1)-N(5)	91.1(1)	92.8(1)
N(5)-Pd(1)-N(4)	88.7(1)	86.0(1)
N(4)-Pd(1)-C(17)	91.5(1)	92.7(2)
C(17) - Pd(1) - N(5)	177.1(1)	178.3(1)
N(1)-Pd(1)-N(4)	178.7(1)	178.0(1)
C(23)-N(4)-Pd(1)	162.8(4)	173.9(4)

	8 ·2/3PhMe	9-MeOH	10-PhMe
Pd(1)-C(17)	1.990(6)	2.004(2)	1.987(4)
Pd(1)–N(1)	2.082(5)	2.099(2)	2.095(3)
Pd(1)-S(1)/Pd(1)-N(7)	2.083(6)	2.4330(6)	2.096(4)
Pd(1)–P(1)	2.237(2)	2.2387(5)	2.2481(9)
N(7)–C(29)	1.119(8)	1.146(4)	1.154(6)
S(1)-C(29)	1.657(8)	1.636(3)	1.625(4)
N(1)–C(1)	1.300(8)	1.314(2)	1.311(5)
N(2)-C(1)	1.360(8)	1.370(2)	1.363(5)
N(3)-C(1)	1.353(8)	1.343(2)	1.365(5)
C(17) - Pd(1) - N(1)	87.3(2)	85.73(7)	85.9(1)
N(1)-Pd(1)-S(1)/N(1)-Pd(1)-N(7)	93.3(2)	88.39(5)	94.3(1)
S(1)-Pd(1)-P(1)/N(7)-Pd(1)-P(1)	87.5(8)	94.93(2)	86.08(9)
C(17) - Pd(1) - P(1)	92.4(2)	91.20(5)	93.5(1)
C(17)-Pd(1)-S(1)/C(17)-Pd(1)-N(7)	175.9(3)	173.71(5)	179.1(1)
N(1)-Pd(1)-P(1)	173.0(2)	171.35(5)	179.2(2)
C(29)-S(1)-Pd(1)/C(29)-N(7)-Pd(1)	160.5(6)	108.10(9)	163.3(3)

Table S6 Selected bond distances (Å) and angles (deg) for 8.2/3PhMe, 9.MeOH and 10.PhMe

	$11 \cdot 1/3 CH_2 Cl_2$	13-PhMe	15.2PhMe	16-CHCl ₃
	$\cdot 1/3H_2O$			
Pd(1)-C(17)	2.011(5)	1.991(3)	2.009(4)	2.014(3)
Pd(1)–N(1)	2.077(4)	2.086(2)	2.113(2)	2.101(3)
Pd(1)-N(4)/Pd(1)-N(7)	2.077(4)	2.093(3)	2.076(4)	2.087(3)
Pd(1) - P(1)	2.208(1)	2.2501(7)	2.2597(8)	2.2601(9)
N(4)-C(29)/N(4)-C(26)/N(7)-C(29)	1.144(6)	1.156(4)	1.146(5)	1.146(5)
S(1)-C(29)/S(1)-C(26)	1.633(5)	1.623(3)	1.636(4)	1.636(4)
N(1)–C(1)	1.310(6)	1.300(3)	1.299(4)	1.297(4)
N(2)–C(1)	1.356(7)	1.365(3)	1.365(4)	1.357(4)
N(3)–C(1)	1.347(6)	1.357(3)	1.360(5)	1.357(4)
C(17) - Pd(1) - N(1)	88.0(2)	86.3(1)	86.5(1)	87.2(1)
N(1)-Pd(1)-N(4)/N(1)-Pd(1)-N(7)	89.3(2)	92.25(9)	89.6(1)	89.6(1)
N(4)-Pd(1)-P(1)/N(7)-Pd(1)-P(1)	90.6(1)	87.19(8)	90.66(8)	90.52(9)
C(17) - Pd(1) - P(1)	92.0(1)	95.09(8)	93.37(9)	92.8(1)
C(17)-Pd(1)-N(4)/C(17)-Pd(1)-N(7)	176.4(2)	172.8(1)	176.0(1)	174.5(1)
N(1) - Pd(1) - P(1)	179.9(1)	172.96(7)	175.44(7)	176.95(8)
C(29)-N(4)-Pd(1)/C(26)-N(4)-Pd(1)/	172.4(4)	158.6(3)	176.6(3)	170.2(4)
C(29)–N(7)–Pd(1)				

Table S7 Selected bond distances (Å) and angles (deg) for $11 \cdot 1/3 CH_2 Cl_2 \cdot 1/3 H_2 O$, $13 \cdot PhMe$, $15 \cdot 2PhMe$ and $16 \cdot CHCl_3$

	17	18-PhMe
Pd(1)–P(1)	2.327(1)	2.3163(7)
Pd(1) - P(2)	2.308(1)	2.3260(7)
Pd(1)-C(3)	2.014(3)	2.028(2)
Pd(1) - N(4)	2.080(4)	2.085(3)
N(4)-C(26)	1.149(5)	1.153(4)
S(1)–C(26)	1.634(4)	1.629(3)
N(1)-C(1)	1.390(4)	1.380(3)
N(2)-C(1)	1.278(4)	1.286(3)
N(3)-C(1)	1.373(4)	1.388(3)
P(1)-Pd(1)-C(3)	88.2(1)	89.35(7)
C(3) - Pd(1) - P(2)	88.2(1)	87.54(8)
P(2)-Pd(1)-N(4)	93.2(1)	93.87(8)
N(4) - Pd(1) - P(1)	90.3(1)	89.23(8)
P(1)-Pd(1)-P(2)	175.51(4)	176.33(3)
C(3) - Pd(1) - N(4)	177.7(1)	178.5(1)
C(26)-N(4)-Pd(1)	168.4(3)	177.1(3)

Table S8 Selected bond distances (Å) and angles (deg) for 17 and 18 \cdot PhMe

Table S9 Selected bond distances (Å) and angles (deg) of optimized NCS coordinated complex ${\bf 8}$

Pd1-P1	2.3134	P1-Pd1-C1	97.03204
Pd1-C1	2.01547	C1-Pd1-N1	86.65862
Pd1-N1	2.12233	N1-Pd1-N2	92.80097
Pd1-N2	2.12492	N2-Pd1-P1	83.29273
N2-C2	1.18391	Pd1-N2-C2	164.09074
C2-S1	1.63508	N2-C2-S1	179.62968

Table S10 Selected bond distances (Å) and angles (deg) of optimized SCN coordinated complex 9

Pd1-P1	2.31475	P1-Pd1-C1	92.30552
Pd1-C1	2.03535	C1-Pd1-N1	85.1815
Pd1-N1	2.14768	N1-Pd1-S1	90.22251
Pd1-S1	2.51584	S1-Pd1-P1	92.28301
N2-C2	1.17458	Pd1-S1-C2	103.7699
C2-S1	1.68358	S1-C2-N2	177.7553

Pd1-P1	2.321	P1-Pd1-C1	95.71
Pd1-C1	2.030	C1-Pd1-N1	88.95
Pd1-N1	2.030	N1-Pd1-N2	86.29
Pd1-N2	2.149	N2-Pd1-P1	89.19
N2-C2	1.183	Pd1-N2-C2	144.82
C2-S1	1.642	N2-C2-S1	178.99

Table S11 Selected bond distances (Å) and angles (deg) of optimized NCS coordinated complex 8'

Table S12 Selected bond distances (Å) and angles (deg) of optimized SCN coordinated complex 9'

Pd1-P1	2.309	P1-Pd1-C1	97.02
Pd1-C1	2.043	C1-Pd1-N1	88.71
Pd1-N1	2.045	N1-Pd1-S1	89.12
Pd1-S1	2.540	S1-Pd1-P1	85.44
N2-C2	1.172	Pd1-S1-C2	99.47
C2-S1	1.682	S1-C2-N2	179.02

Table S13 Selected bond lengths (Å) and bond angles (deg) of the methanol mediated NCSbound modified Pd dimeric complex 8''

Pd1-P1	2.321	P1-Pd1-C1	95.71
Pd1-C1	2.030	C1-Pd1-N1	88.95
Pd1-N1	2.030	N1-Pd1-N2	86.29
Pd1-N2	2.149	N2-Pd1-P1	89.19
N2-C2	1.183	N2-C2-S1	178.99
C2-S1	1.642	Pd1-N2-C2	144.82
S1-H2'	2.403	P1'-Pd1'-C1'	95.32
O1'-H1'	2.603	C1'-Pd1'-N1'	88.94
Pd1'-P1'	2.326	N1'-Pd1'-N2'	88.10
Pd1'-C1'	2.030	N2'-Pd1'-P1'	87.79
Pd1'-N1'	2.050	N2'-C2'-S1'	179.08
Pd1'-N2'	2.118	Pd1'-N2'-C2'	163.32
N2'-C2'	1.183		
C2'-S1'	1.637		
S1'-H2	2.392		
01-H1	2.485		

Pd1-P1	2.296	P1-Pd1-C1	94.40
Pd1-C1	2.037	C1-Pd1-N1	85.98
Pd1-N1	2.076	N1-Pd1-S1	88.79
Pd1-S1	2.557	S1-Pd1-P1	89.97
S1-C2	1.681	N2-C2-S1	178.00
C2-N2	1.174	Pd1-S1-C2	92.81
N2-H2'	1.871	P1'-Pd1'-C1'	97.02
O1'-H1'	2.482	C1'-Pd1'-N1'	88.71
Pd1'-P1'	2.309	N1'-Pd1'-S1'	89.12
Pd1'-C1'	2.043	S1'-Pd1'-P1'	85.44
Pd1'-N1'	2.045	N2'-C2'-S1'	179.02
Pd1'-S1'	2.540	Pd1'-S1'-C2'	99.47
S1'-C2'	1.682		
C2'-N2'	1.172		
N2'-H2	1.914		
01-H1	2.693		

Table S14 Selected bond lengths (Å) and bond angles (deg) of the methanol mediated SCN bound modified Pd dimeric complex 9''

Cartesian coordinates of the optimized NCS coordinated complex 8

Pd	1.00294100	0.07230800	0.56489500
Р	3.17079300	0.47178800	-0.13700000
0	-1.75858000	-2.76419200	-0.38929100
0	-4.72215800	-1.53545900	-1.84498700
0	-2.43807100	4.01779000	-0.10461900
Ν	-1.03544800	-0.41047600	0.90571200
Ν	-3.17387300	-0.34311600	-0.07267300
Н	-3.10182700	-1.32324900	-0.33988800
Ν	-1.73453100	1.50447800	-0.20456800
Н	-2.48161400	1.93841600	-0.73018100
Ν	5.40575300	1.96994800	-0.84729500
Ν	4.89403700	0.02150900	-2.27631300
Ν	5.83346600	-0.31821100	-0.01650500
Ν	1.53883800	-1.95903400	0.24592600

S	2.34450900	-4.40316000	-0.90460900
С	-1.96016800	0.23348600	0.23482100
С	-1.29371000	-1.61737100	1.61600600
С	-1.13413000	-1.61522300	3.00460600
Н	-0.88616300	-0.67403300	3.48647800
С	-1.27036900	-2.78666100	3.74938700
Н	-1.14401700	-2.76168400	4.82774100
С	-1.55238600	-3.98569100	3.09711100
Н	-1.64688400	-4.90939100	3.66120800
С	-1.71133100	-4.01288500	1.71030700
Н	-1.92080100	-4.95209300	1.21145000
С	-1.58641700	-2.83552300	0.96845000
С	-1.71963500	-3.97215100	-1.14756000
Н	-1.75672900	-3.66244800	-2.19321500
Н	-2.58449100	-4.61146000	-0.92665800
Н	-0.78698400	-4.51493500	-0.96157400
С	-4.37087300	0.30423900	-0.43210600
С	-4.80312100	1.50568300	0.13752400
Н	-4.18495900	1.99544600	0.88169100
С	-6.02307100	2.07462300	-0.24015200
Н	-6.33771000	3.01038700	0.21232300
С	-6.83580700	1.43161500	-1.16830300
Н	-7.78896000	1.86294300	-1.45973100
С	-6.43321900	0.21319300	-1.72726200
Н	-7.07440700	-0.28645000	-2.44439400
С	-5.20955300	-0.34985000	-1.36778600
С	-5.52737300	-2.28560600	-2.74148600
Н	-4.95791400	-3.18733700	-2.97213300
Н	-5.71821100	-1.72980300	-3.66846400
Н	-6.48419700	-2.56475100	-2.28186200

С	-0.79137300	2.40295500	0.35288300
С	0.46684800	1.99852400	0.81867700
С	1.26390700	2.97096800	1.45079500
Н	2.21814100	2.68512600	1.88065000
С	0.86057300	4.30141500	1.53956600
Н	1.50406400	5.03486700	2.01904000
С	-0.36385500	4.71138400	1.00318800
Н	-0.66548300	5.75074500	1.06216700
С	-1.19521000	3.75667300	0.42349400
С	-2.88957500	5.36244700	-0.13455000
Н	-3.87087300	5.33816100	-0.61206000
Н	-2.21345800	5.99862800	-0.72003500
Н	-2.98687600	5.77496100	0.87799000
С	4.06571500	2.12664700	-0.26239500
Н	3.47180700	2.81685700	-0.87190800
Н	4.16053900	2.57585800	0.73177100
С	3.48099700	-0.10443900	-1.90118000
Н	3.16345400	-1.14840500	-1.98361500
Н	2.86283700	0.48747900	-2.58618700
С	4.55793900	-0.49610200	0.68833600
Н	4.66264100	-0.16703600	1.72884700
Н	4.27707300	-1.55337700	0.70095500
С	5.34801400	1.41760200	-2.21089400
Н	4.68650500	2.04089600	-2.82245500
Н	6.35731500	1.46682800	-2.63506400
С	5.75652000	-0.79878300	-1.40772700
Н	6.76934000	-0.78270300	-1.82568600
Н	5.39295200	-1.83076900	-1.41271800
С	6.25649700	1.08886600	-0.02805300
Н	7.27328700	1.13444400	-0.43421100

Η	6.27916200	1.46701000	0.99995200
С	1.87365300	-2.98542300	-0.23993700

Cartesian coordinates of the optimized SCN coordinated complex 9

Pd	-1.00519900	-0.34179700	0.33645300
Р	-3.15531100	0.15521600	-0.36217800
S	-1.22625200	-2.74827400	-0.36316800
0	2.20803200	-2.78477400	-0.64773800
0	4.90267000	-0.95460000	-2.06048300
0	2.00011500	4.00351500	0.37528400
Ν	1.08712800	-0.63695900	0.72061700
Ν	3.21163700	-0.18361500	-0.18212100
Н	3.27650900	-1.13863500	-0.52834800
Ν	1.54601700	1.46944400	-0.14911300
Н	2.23568500	2.04913600	-0.60879600
Ν	-5.00884400	1.99532900	-1.33701000
Ν	-5.87983800	0.23709400	0.16861600
Ν	-5.23393800	-0.32258200	-2.14841100
Ν	-3.84840200	-3.46938300	0.51454000
С	1.93013700	0.19441800	0.15248200
С	1.52496600	-1.81636600	1.38922900
С	1.34267800	-1.90001700	2.77257000
Н	0.91895900	-1.03973000	3.28192800
С	1.67628700	-3.05661900	3.47722400

Н	1.52618400	-3.09837300	4.55193200
С	2.18847300	-4.15631600	2.79138200
Н	2.44368600	-5.06753400	3.32506200
С	2.37298700	-4.09893700	1.40844900
Н	2.76504600	-4.96214600	0.88244500
С	2.04240100	-2.93654600	0.70822600
С	2.42840300	-3.94633700	-1.43877800
Н	3.40583000	-4.39930200	-1.22737400
Н	2.40256400	-3.60765800	-2.47581300
Н	1.63354500	-4.68168600	-1.27344400
С	4.32191600	0.64297900	-0.44143700
С	4.59976800	1.80967900	0.27668800
Н	3.92311100	2.12441100	1.06309000
С	5.73912600	2.56723400	-0.00962000
Н	5.93452300	3.47242000	0.55762500
С	6.62672700	2.14724200	-0.99526000
Н	7.51908200	2.72534500	-1.21661500
С	6.38072900	0.96569800	-1.70367100
Н	7.08116600	0.64142100	-2.46485500
С	5.23623900	0.21576500	-1.43493000
С	5.78954000	-1.47004900	-3.04156000
Н	5.89935100	-0.77628900	-3.88478800
Н	5.33841500	-2.39868900	-3.39500300
Н	6.77840800	-1.68332500	-2.61564500

С	0.53071900	2.17492800	0.54865900
С	-0.67022100	1.57319200	0.93914900
С	-1.54255800	2.32703400	1.74528600
Н	-2.45858900	1.87664600	2.11717900
С	-1.26630600	3.65348300	2.07235900
Н	-1.96366500	4.22071100	2.68393400
С	-0.10228500	4.27384600	1.60762300
Н	0.09783500	5.31067400	1.85184800
С	0.80343100	3.52748600	0.85720700
С	2.31236900	5.37271700	0.57969600
Н	1.56063900	6.02719200	0.12026300
Н	3.27873800	5.53456500	0.09852700
Н	2.39296700	5.61028800	1.64827500
С	-3.60058600	1.89754100	-0.92929000
Н	-3.39719300	2.60987300	-0.12483400
Н	-2.95595900	2.17017700	-1.77336500
С	-4.60583700	-0.11447700	0.80897000
Н	-4.61442600	-1.16801000	1.10438100
Н	-4.46696100	0.49500900	1.70968800
С	-3.85965900	-0.74970400	-1.86123100
Н	-3.21839000	-0.54941400	-2.72688100
Н	-3.84062000	-1.82359600	-1.66826700
С	-5.92055700	1.64987400	-0.23206200
Н	-5.68651200	2.27716900	0.63527700

Н	-6.94154000	1.88052800	-0.55736900
С	-6.13486400	-0.60243600	-1.01576100
Н	-7.16236200	-0.41497200	-1.34796900
Н	-6.04013400	-1.65477900	-0.73366500
С	-5.30706300	1.10669800	-2.47218000
Н	-6.32514500	1.32737000	-2.81337800
Н	-4.61381900	1.32699900	-3.29163700
С	-2.76045100	-3.18324400	0.17671200

Cartesian coordinates of the optimized NCS coordinated modified complex 8'

Pd	0.90840000	-0.63599700	0.00792500
Р	1.26148000	-2.88129700	0.47711200
0	-4.02763000	1.42066700	0.21637000
Ν	0.75113700	1.37590400	-0.21449900
Ν	-0.07610800	3.50980200	0.25933600
Н	0.89407000	3.80209300	0.39030500
Ν	-1.48051700	1.66238900	0.36465100
Н	-2.23264200	2.33826100	0.42354000
Ν	3.03483200	-0.32544500	0.04337500
S	5.08487000	1.58087300	-0.33576500
С	-0.23501600	2.15444100	0.13915100
С	-1.93489700	0.35434100	0.08330600
С	-1.11420900	-0.76936200	-0.10779700
С	-1.75798400	-1.97830000	-0.44193300

Н	-1.17193300	-2.86650300	-0.65779600
С	-3.14234300	-2.08236700	-0.53124500
Н	-3.60022100	-3.03590500	-0.78087600
С	-3.95238200	-0.96847400	-0.29747200
Н	-5.03130900	-1.05006600	-0.35432000
С	-3.34820500	0.24732600	-0.00101800
С	-5.42875000	1.46246900	-0.07080100
Н	-5.71666600	2.50924900	0.02770400
Н	-5.99533600	0.86549100	0.65089300
Н	-5.62211000	1.11622800	-1.09302200
С	3.89971200	0.46512800	-0.12212900
Н	1.64673300	1.84950000	-0.26651700
Н	-0.74030000	4.00272300	0.84527600
Н	1.27099100	-3.83756200	-0.56479000
Н	2.51280500	-3.17620100	1.05374400
Н	0.41947700	-3.57105500	1.37551700

Cartesian coordinates of the optimized SCN coordinated modified complex 9'

Pd	1.07459500	-0.22249400	-0.31505600
Р	1.99315400	-2.24290600	0.32142800
0	-4.11337400	0.74689700	0.51869200
Ν	0.44786800	1.68505500	-0.70443700
Ν	-0.86512600	3.61270800	-0.38318900
Н	-0.04285100	4.19984900	-0.34507300

Ν	-1.69448300	1.53393900	0.18281300
Н	-2.54493500	2.04633100	0.38588600
С	-0.65391000	2.25791000	-0.30419700
С	-1.89276500	0.14181600	0.04440600
С	-0.88364200	-0.79617700	-0.21553100
С	-1.28615100	-2.12979100	-0.42647200
Н	-0.55246900	-2.88595200	-0.68577000
С	-2.61696500	-2.52235200	-0.32504200
Н	-2.89110100	-3.56048700	-0.49321700
С	-3.60764100	-1.59352900	0.00678900
Н	-4.64132900	-1.90443800	0.10169300
С	-3.24438400	-0.26535100	0.19303500
С	-5.48803200	0.43522900	0.72264500
Н	-5.94916400	1.35718900	1.07836100
Н	-5.60367600	-0.34564300	1.48460400
Н	-5.95907400	0.10237000	-0.21090300
С	3.88164700	0.78231100	1.02534400
Н	1.17059600	2.31286500	-1.03941000
Н	-1.67055600	3.97541500	0.12249000
Н	2.98999100	-2.14836800	1.31261900
Н	2.70457500	-2.98808600	-0.64331500
Н	1.21388600	-3.28415400	0.87367200
Ν	4.17799800	0.92929900	2.14937900
S	3.47760600	0.55242900	-0.59111300

Cartesian coordinates of the optimized methanol mediated NCS coordinated modified dimeric complex 8"

Pd	4.07910500	-1.52388700	-0.16920100
Р	5.43470000	-3.21021800	0.67033800
0	-1.14897500	-2.38588100	-0.91876800
Ν	2.96017500	0.08274100	-0.70649900
Ν	1.08523700	1.47754000	-0.66750200
Н	1.72380800	2.24625100	-0.45503200
Ν	0.83360600	-0.82373500	-0.47461000
Н	-0.15363400	-0.64676300	-0.61604000
Ν	5.69537000	-0.13275800	0.09887000
S	6.50985800	2.54468300	-0.28631700
С	1.66834800	0.23948700	-0.60500800
С	1.17550400	-2.18745800	-0.61592700
С	2.47442200	-2.71466200	-0.52860700
С	2.61692600	-4.09784200	-0.76126400
Н	3.60343400	-4.55144000	-0.76400000
С	1.52707400	-4.92425800	-1.01526600
Н	1.67864000	-5.98800600	-1.17857700
С	0.23399900	-4.39727700	-1.05790700
Н	-0.61757200	-5.04104100	-1.24329000
С	0.06195500	-3.03141600	-0.86840600
С	-2.28994700	-3.10903500	-1.39055500
Н	-3.08548500	-2.37003200	-1.48721000

Н	-2.59628300	-3.87310200	-0.66902600
Н	-2.08078800	-3.56288500	-2.36650500
С	6.04704000	0.98492500	-0.06704600
Pd	-3.45386600	0.88627400	0.28864300
Р	-2.83636200	-0.68148700	1.89165100
0	-0.31009900	5.12125900	-0.57435300
N	-3.99492500	2.10859200	-1.26623600
N	-3.52482500	3.45107100	-3.15183600
Н	-4.41849700	3.17994100	-3.53906500
N	-2.10798100	3.44796000	-1.32693900
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Cartesian coordinates of the optimized methanol mediated SCN coordinated modified dimeric complex 9"

Pd	-4.80751800	0.69491200	-0.48818200
Р	-6.32916300	1.85069600	0.78414500
0	0.18523000	1.15590400	1.10940300
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Н	-6.10231900	3.18243600	1.19454900
Н	-7.64812600	1.97943700	0.29865800
Н	-6.61966100	1.28047400	2.04101400

Ν	-4.97583600	-3.17401200	0.06889100
Ν	3.74581500	3.94159900	-0.17647900
S	-6.58240900	-1.07290000	-1.00296800
S	5.02589000	2.06008200	-1.89815700



Fig. S1 C–H··· π , π ··· π and C–H···S interactions present in the crystal lattice of **9**·MeOH. The sulfur atom of thiocyanate is involved in intermolecular C–H····S hydrogen bond with PTA in the neighboring molecule related by 2-fold screw axis. The 2-anisyl ring of the N(H)Ar unit of the reference molecule forms a π -stacking interaction (π ··· π = 3.587 Å) with the 2-anisyl ring of the N(H)Ar unit of the inversion related adjacent molecule. Further, the same 2-anisyl ring of the reference molecule also forms C–H··· π interactions with the C-H protons of PTA from a two-fold screw axis related adjacent molecule.



Fig. S2 ¹H NMR spectrum of **2** (CDCl₃, 400.0 MHz).



Fig. S3 Expansion of ¹H NMR spectrum of 2 (CDCl₃, 400.0 MHz).



Fig. S4 Expansion of ¹H NMR spectrum of 2 (CDCl₃, 400.0 MHz).



Fig. S5 ¹H NMR spectrum of 3 (CDCl₃, 400.0 MHz).



Fig. S6 Expansion of ¹H NMR spectrum of 3 (CDCl₃, 400.0 MHz).



Fig. S7 Expansion of ¹H NMR spectrum of 3 (CDCl₃, 400.0 MHz).


Fig. S8 ¹H NMR spectrum of 4 (CDCl₃, 400.0 MHz).



Fig. S9 Expansion of ¹H NMR spectrum of 4 (CDCl₃, 400.0 MHz).



Fig. S10 Expansion of ¹H NMR spectrum of 4 (CDCl₃, 400.0 MHz).



Fig. S11 ¹³C NMR spectrum of 4 (CDCl₃, 100.5 MHz).



Fig. S12 Expansion of ¹³C NMR spectrum of 4 (CDCl₃, 100.5 MHz).



Fig. S13 Expansion of ¹³C NMR spectrum of 4 (CDCl₃, 100.5 MHz).



Fig. S14 DEPT 90 NMR spectrum of 4 (CDCl₃, 100.5 MHz).



Fig. S15 Expansion of DEPT 90 NMR spectrum of 4.



Fig. S16 ¹H NMR spectrum of **5** (CDCl₃, 400.0 MHz).



Fig. S17 Expansion of ¹H NMR spectrum of 5 (CDCl₃, 400.0 MHz).



Fig. S18 Expansion of ¹H NMR spectrum of 5 (CDCl₃, 400.0 MHz).



Fig. S19 Expansion of ¹H NMR spectrum of 5 (CDCl₃, 400.0 MHz).



Fig. S20 ¹H NMR spectrum of 6 (CDCl₃, 400.0 MHz).



Fig. S21 Expansion of ¹H NMR spectrum of 6 (CDCl₃, 400.0 MHz).



Fig. S22 Expansion of ¹H NMR spectrum of 6 (CDCl₃, 400.0 MHz).



Fig. S23 ¹H NMR spectrum of **7** (CDCl₃, 400.0 MHz).



Fig. S24 Expansion of ¹H NMR spectrum of 7 (CDCl₃, 400.0 MHz).



Fig. S25 Expansion of ¹H NMR spectrum of 7 (CDCl₃, 400.0 MHz).



Fig. S26 Expansion of ¹H NMR spectrum of 7 (CDCl₃, 400.0 MHz).



Fig. S27 ¹H NMR spectrum of 8/9 (CDCl₃, 400.0 MHz).



Fig. S28 Expansion of ¹H NMR spectrum of 8/9 (CDCl₃, 400.0 MHz).



Fig. S29 Expansion of ¹H NMR spectrum of 8/9 (CDCl₃, 400.0 MHz).



Fig. S30 ¹³C NMR spectrum of 8/9 (CDCl₃, 100.5 MHz).



Fig. S31 Expansion of ¹³C NMR spectrum of 8/9 (CDCl₃, 100.5 MHz).



Fig. S32 Expansion of ¹³C NMR spectrum of 8/9 (CDCl₃, 100.5 MHz).



Fig. S33 DEPT 90 NMR spectrum of **8/9** (CDCl₃, 100.5 MHz). The peaks around δ 50–56 and 73 correspond to residual peaks of CH₂ and OCH₃ carbons.



Fig. S34 Expansion of DEPT 90 NMR spectrum of 8/9.



Fig. S35 The ¹³C-¹H HETCOR NMR spectrum of 8/9 (X-axis 100.5 MHz, Y-axis 400 MHz, CDCl₃).



Fig. S36 The ¹³C-¹H HETCOR NMR spectrum of 8/9 illustrated for the ArCH carbons (X-axis 100.5 MHz, Y-axis 400 MHz, CDCl₃).



Fig. S37 ³¹P{¹H} NMR spectrum of 8/9 (CDCl₃, 161.8 MHz) at 35.8 μ M concentration.



Fig. S38 ${}^{31}P{}^{1}H$ NMR spectrum of 8/9 (CDCl₃, 161.8 MHz) at 7.16 μ M concentration.



Fig. S39 Expansion of ${}^{31}P{}^{1}H$ NMR spectrum of **8/9** (CDCl₃, 161.8 MHz) at 7.16 μ M concentration.



Fig. S40 ¹H NMR spectrum of **10** (CDCl₃, 400.0 MHz).



Fig. S41 Expansion of ¹H NMR spectrum of **10** (CDCl₃, 400.0 MHz).



Fig. S42 Expansion of ¹H NMR spectrum of 10 (CDCl₃, 400.0 MHz).



Fig. S43 ¹³C NMR spectrum of 10 (CDCl₃, 100.5 MHz).


Fig. S44 Expansion of ¹³C NMR spectrum of **10** (CDCl₃, 100.5 MHz).



Fig. S45 Expansion of ¹³C NMR spectrum of 10 (CDCl₃, 100.5 MHz).



Fig. S46 DEPT 90 NMR spectrum of 10 (CDCl₃, 100.5 MHz). The peaks around δ 15, 50 and 73 correspond to residual peaks of CH₃ and CH₂ carbons.



Fig. S47 Expansion of DEPT 90 NMR spectrum of 10.



Fig. S48 The ¹³C-¹H HETCOR NMR spectrum of 10 (X-axis 100.5 MHz, Y-axis 400 MHz, CDCl₃).



Fig. S49 The ¹³C-¹H HETCOR NMR spectrum of 10 illustrated for the ArCH carbons (X-axis 100.5 MHz, Y-axis 400 MHz, CDCl₃).



Fig. S50 ³¹P{¹H} NMR spectrum of **10** (CDCl₃, 161.8 MHz).



Fig. S51 ¹H NMR spectrum of **11** (CDCl₃, 400.0 MHz).



Fig. S52 Expansion of ¹H NMR spectrum of 11 (CDCl₃, 400.0 MHz).



Fig. S53 Expansion of ¹H NMR spectrum of 11 (CDCl₃, 400.0 MHz).



Fig. S54 Expansion of ¹H NMR spectrum of 11 (CDCl₃, 400.0 MHz).



Fig. S55 ¹³C NMR spectrum of **11** (CDCl₃, 100.5 MHz).



Fig. S56 Expansion of ¹³C NMR spectrum of **11** (CDCl₃, 100.5 MHz).



Fig. S57 Expansion of ¹³C NMR spectrum of **11** (CDCl₃, 100.5 MHz).



Fig. S58 Expansion of ¹³C NMR spectrum of **11** (CDCl₃, 100.5 MHz).



Fig. S59 Expansion of ¹³C NMR spectrum of 11 (CDCl₃, 100.5 MHz).



Fig. S60 DEPT 90 NMR spectrum of **11** (CDCl₃, 100.5 MHz). The peaks around δ 7–75 correspond to residual peaks of C, CH₂, CH₃, OCH₂ or OCH₃ carbons.



Fig. S61 Expansion of DEPT 90 NMR spectrum of 11.



Fig. S62 The ¹³C-¹H HETCOR NMR spectrum of 11 (X-axis 100.5 MHz, Y-axis 400 MHz, CDCl₃).



Fig. S63 The ¹³C-¹H HETCOR NMR spectrum of 11 illustrated for the ArCH carbons (X-axis 100.5 MHz, Y-axis 400 MHz, CDCl₃).



Fig. S64 ³¹P{¹H} NMR spectrum of **11** (CDCl₃, 161.8 MHz).



Fig. S65 Expansion of ³¹P{¹H} NMR spectrum of 11 (CDCl₃, 161.8 MHz).



Fig. S66 ¹H NMR spectrum of **12** (CDCl₃, 400.0 MHz).



Fig. S67 Expansion of ¹H NMR spectrum of **12** (CDCl₃, 400.0 MHz).



Fig. S68 Expansion of ¹H NMR spectrum of **12** (CDCl₃, 400.0 MHz).



Fig. S69 ¹³C NMR spectrum of 12 (CDCl₃, 100.5 MHz).



Fig. S70 Expansion of ¹³C NMR spectrum of **12** (CDCl₃, 100.5 MHz).



Fig. S71 Expansion of ¹³C NMR spectrum of 12 (CDCl₃, 100.5 MHz).



Fig. S72 Expansion of ¹³C NMR spectrum of **12** (CDCl₃, 100.5 MHz).



Fig. S73 Expansion of ¹³C NMR spectrum of **12** (CDCl₃, 100.5 MHz).



Fig. S74 DEPT 90 NMR spectrum of **12** (CDCl₃, 100.5 MHz). The peaks around δ 7–75 correspond to residual peaks of C, CH₂, OCH₂ or CH₃ carbons.



Fig. S75 Expansion of DEPT 90 NMR spectrum of 12.



Fig. S76 The ¹³C-¹H HETCOR NMR spectrum of **12** (X-axis 100.5 MHz, Y-axis 400 MHz, CDCl₃).



Fig. S77 The ¹³C-¹H HETCOR NMR spectrum of 12 illustrated for the ArCH carbons (X-axis 100.5 MHz, Y-axis 400 MHz, CDCl₃).



Fig. S78 ³¹P{¹H} NMR spectrum of **12** (CDCl₃, 161.8 MHz).



Fig. S79 Expansion of ${}^{31}P{}^{1}H$ NMR spectrum of 12 (CDCl₃, 161.8 MHz).


Fig. S80 ¹H NMR spectrum of **13** (CDCl₃, 400.0 MHz).



Fig. S81 Expansion of ¹H NMR spectrum of 13 (CDCl₃, 400.0 MHz).



Fig. S82 Expansion of ¹H NMR spectrum of 13 (CDCl₃, 400.0 MHz).



Fig. S83 ¹³C NMR spectrum of 13 (CDCl₃, 100.5 MHz).



Fig. S84 Expansion of ¹³C NMR spectrum of **13** (CDCl₃, 100.5 MHz).



Fig. S85 Expansion of ¹³C NMR spectrum of **13** (CDCl₃, 100.5 MHz).



Fig. S86 DEPT 90 NMR spectrum of **13** (CDCl₃, 100.5 MHz). The peaks around δ 15 and 56 correspond to residual peaks of CH₃ and OCH₃ carbons.



Fig. S87 Expansion of DEPT 90 NMR spectrum of 13.



Fig. S88 The ¹³C-¹H HETCOR NMR spectrum of 13 (X-axis 100.5 MHz, Y-axis 400 MHz, CDCl₃).



Fig. S89 The ¹³C-¹H HETCOR NMR spectrum of 13 illustrated for the ArCH carbons (X-axis 100.5 MHz, Y-axis 400 MHz, CDCl₃).



Fig. S90 ³¹P{¹H} NMR spectrum of **13** (CDCl₃, 161.8 MHz).



Fig. S91 ¹H NMR spectrum of **14** (CDCl₃, 400.0 MHz).



Fig. S92 Expansion of ¹H NMR spectrum of 14 (CDCl₃, 400.0 MHz).



Fig. S93 Expansion of ¹H NMR spectrum of 14 (CDCl₃, 400.0 MHz).



Fig. S94 ¹³C NMR spectrum of **14** (CDCl₃, 100.5 MHz).



Fig. S95 Expansion of ¹³C NMR spectrum of 14 (CDCl₃, 100.5 MHz).



Fig. S96 Expansion of ¹³C NMR spectrum of 14 (CDCl₃, 100.5 MHz).



Fig. S97 DEPT 90 NMR spectrum of 14 (CDCl₃, 100.5 MHz). The peaks around δ 15 correspond to residual peaks of CH₃ carbons.



Fig. S98 Expansion of DEPT 90 NMR spectrum of 14.



Fig. S99 The ¹³C-¹H HETCOR NMR spectrum of 14 (X-axis 100.5 MHz, Y-axis 400 MHz, CDCl₃).



Fig. S100 The ¹³C-¹H HETCOR NMR spectrum of 14 illustrated for the ArCH carbons (X-axis 100.5 MHz, Y-axis 400 MHz, CDCl₃).



Fig. S101 ³¹P{¹H} NMR spectrum of **14** (CDCl₃, 161.8 MHz).



Fig. S102 ¹H NMR spectrum of **15** (CDCl₃, 400.0 MHz).



Fig. S103 Expansion of ¹H NMR spectrum of 15 (CDCl₃, 400.0 MHz).



Fig. S104 Expansion of ¹H NMR spectrum of 15 (CDCl₃, 400.0 MHz).



Fig. S105 ¹³C NMR spectrum of **15** (CDCl₃, 100.5 MHz).



Fig. S106 Expansion of ¹³C NMR spectrum of 15 (CDCl₃, 100.5 MHz).



Fig. S107 DEPT 90 NMR spectrum of 15 (CDCl₃, 100.5 MHz). The peaks around δ 38 and 56 correspond to residual peaks of CH₃ and OCH₃ carbons.



Fig. S108 Expansion of DEPT 90 NMR spectrum of 15.



Fig. S109 The ¹³C-¹H HETCOR NMR spectrum of 15 (X-axis 100.5 MHz, Y-axis 400 MHz, CDCl₃).



Fig. S110 The ¹³C-¹H HETCOR NMR spectrum of 15 illustrated for the ArCH carbons (X-axis 100.5 MHz, Y-axis 400 MHz, CDCl₃).



Fig. S111 ³¹P{¹H} NMR spectrum of **15** (CDCl₃, 161.8 MHz).



Fig. S112 ¹H NMR spectrum of **16** (CDCl₃, 400.0 MHz).



Fig. S113 Expansion of ¹H NMR spectrum of 16 (CDCl₃, 400.0 MHz).



Fig. S114 Expansion of ¹H NMR spectrum of 16 (CDCl₃, 400.0 MHz).



Fig. S115 ¹³C NMR spectrum of **16** (CDCl₃, 100.5 MHz).


Fig. S116 Expansion of ¹³C NMR spectrum of 16 (CDCl₃, 100.5 MHz).



Fig. S117 Expansion of ¹³C NMR spectrum of 16 (CDCl₃, 100.5 MHz).



Fig. S118 DEPT 90 NMR spectrum of 16 (CDCl₃, 100.5 MHz). The peaks around δ 17 and 38 correspond to residual peaks of CH₃ carbons.



Fig. S119 Expansion of DEPT 90 NMR spectrum of 16.



Fig. S120 The ¹³C-¹H HETCOR NMR spectrum of 16 (X-axis 100.5 MHz, Y-axis 400 MHz, CDCl₃).



Fig. S121 The ¹³C-¹H HETCOR NMR spectrum of 16 illustrated for the ArCH carbons (X-axis 100.5 MHz, Y-axis 400 MHz, CDCl₃).



Fig. S122 ³¹P{¹H} NMR spectrum of **16** (CDCl₃, 161.8 MHz).



Fig. S123 Expansion of ${}^{31}P{}^{1}H$ NMR spectrum of 16 (CDCl₃, 161.8 MHz).



Fig. S124 ¹H NMR spectrum of **17** (CDCl₃, 400.0 MHz).



Fig. S125 Expansion of ¹H NMR spectrum of 17 (CDCl₃, 400.0 MHz).



Fig. S126 ¹³C NMR spectrum of **17** (CDCl₃, 100.5 MHz).



Fig. S127 Expansion of ¹³C NMR spectrum of 17 (CDCl₃, 100.5 MHz).



Fig. S128 Expansion of ¹³C NMR spectrum of 17 (CDCl₃, 100.5 MHz).



Fig. S129 DEPT 90 NMR spectrum of 17 (CDCl₃, 100.5 MHz). The peaks around δ 14 and 55 correspond to residual peaks of CH₃ and OCH₃ carbons.



Fig. S130 Expansion of DEPT 90 NMR spectrum of 17.



Fig. S131 ³¹P{¹H} NMR spectrum of **17** (CDCl₃, 161.8 MHz).



Fig. S132 Expansion of ${}^{31}P{}^{1}H$ NMR spectrum of 17 (CDCl₃, 161.8 MHz).



Fig. S133 ¹H NMR spectrum of 18 (CDCl₃, 400.0 MHz).



Fig. S134 Expansion of ¹H NMR spectrum of 18 (CDCl₃, 400.0 MHz).



Fig. S135 ¹³C NMR spectrum of 18 (CDCl₃, 400.0 MHz).



Fig. S136 Expansion of ¹³C NMR spectrum of 18 (CDCl₃, 400.0 MHz).



Fig. S137 Expansion of ¹³C NMR spectrum of 18 (CDCl₃, 400.0 MHz).



Fig. S138 DEPT 90 NMR spectrum of 18 (CDCl₃, 100.5 MHz). Peak around δ 13 corresponds to residual peaks of CH₃ carbons.



Fig. S139 Expansion of DEPT 90 NMR spectrum of 18.



Fig. S140 ³¹P{¹H} NMR spectrum of **18** (CDCl₃, 161.8 MHz).