

A Novel Fluorescent Sensor for Water in Organic Solvents Based on Dynamic Quenching of Carbon Quantum Dots

Jianfei Wei^{a,b,†}, Ye Yuan^{b,†}, Haikuo Li^b, Dan Hao^b, Chenying Sun^b, Guo Zheng^{a,*},
Rui Wang^{b,*}

a. School of Material Science and Engineering, Tianjin Polytechnic University, No. 399 BinShuiXi Road, Xiqing District, Tianjin, P.R. of China

b. School of Material Science and Engineering, Beijing Institute of Fashion Technology, No. A2, East Yinghua Street, Chaoyang District, Beijing, P.R. China,
E-mail: clywangrui@bift.edu.cn , Tel: 86-10-64288279

† Jianfei Wei and Ye Yuan contribute equally to this work.

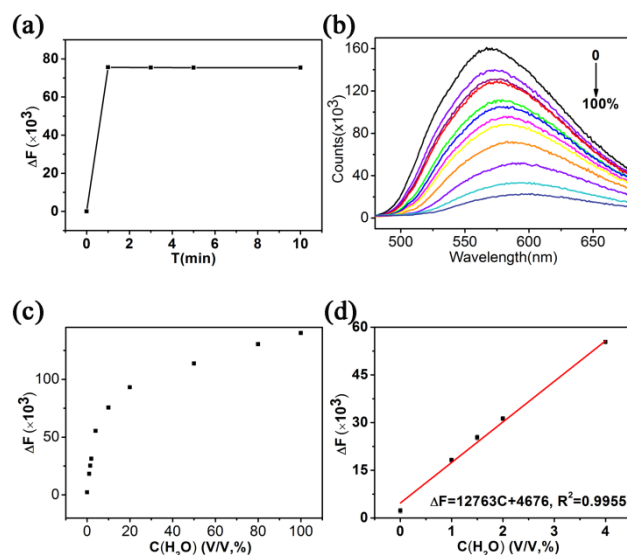


Fig. S1 (a) time-dependent fluorescence quenching of CQDs dispersed in 1,4-Dioxane in the presence of H₂O (10%,V/V); (b) the fluorescence emission spectra of CQDs dispersed in 1,4-Dioxane with different concentration of H₂O (0, 2, 4, 6, 8, 10, 20, 50, 80, 100 V/V, %); (c) the change of fluorescence intensity of CQDs solution versus the concentration of H₂O; (d) a linear relationship between ΔF and concentration of H₂O from 0.001 % to 4 %. Error bars in (c) and (d) represent the standard deviations of five independent measurements

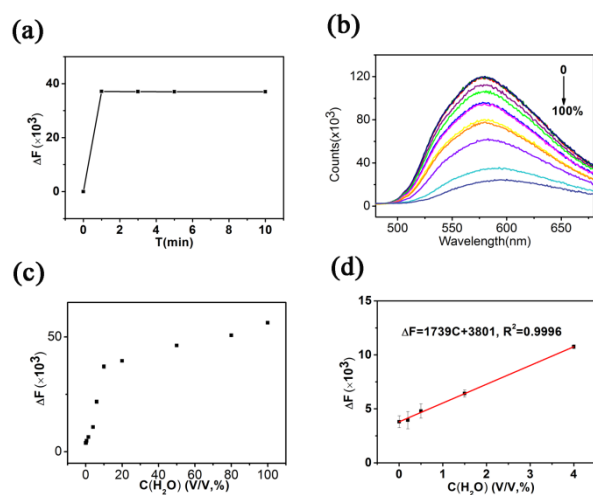


Fig. s2 (a) time-dependent fluorescence quenching of CQDs dispersed in ethanol in the presence of H₂O (10%,V/V); (b) the fluorescence emission spectra of CQDs dispersed in ethanol with different concentration of H₂O (0, 2, 4, 6, 8, 10, 20, 50, 80, 100 V/V, %); (c) the change of fluorescence intensity of CQDs solution versus the concentration of H₂O; (d) a linear relationship between ΔF and concentration of H₂O from 0.001 % to 4 %. Error bars in (c) and (d) represent the standard deviations of five independent measurements.

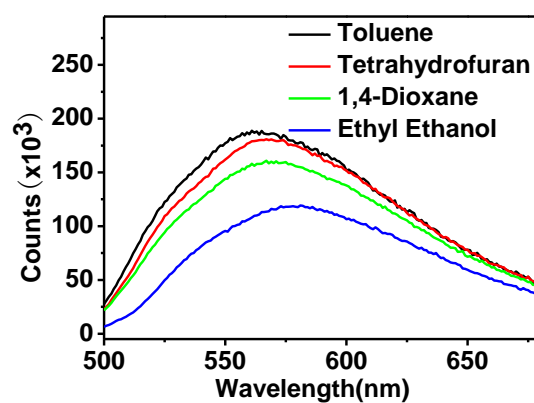


Fig. S3 The fluorescence emission spectra (excited by the light of 440 nm) of CQDs dispersed in different organ solvents (toluene, tetrahydrofuran, 1,4-Dioxane, ethyl ethanol). (The polarity: toluene < tetrahydrofuran < 1,4-Dioxane < ethyl ethanol)