

Electronic Supplementary Information (ESI) for

Ultrafine Mo-doped SnO₂ Nanostructure and Derivative Mo-doped Sn/C

Nanofibers for High-Performance Lithium-ion Batteries

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Supporting Materials

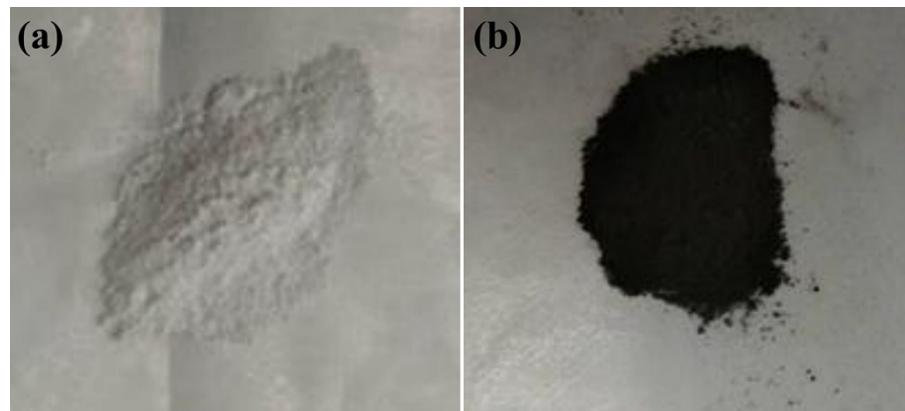


Fig. S1 Digital pictures of the SnO₂ (a) and Mo-doped SnO₂ NPs (b). Mo-doping endue SnO₂ present a black color, which is much different from standard white SnO₂.

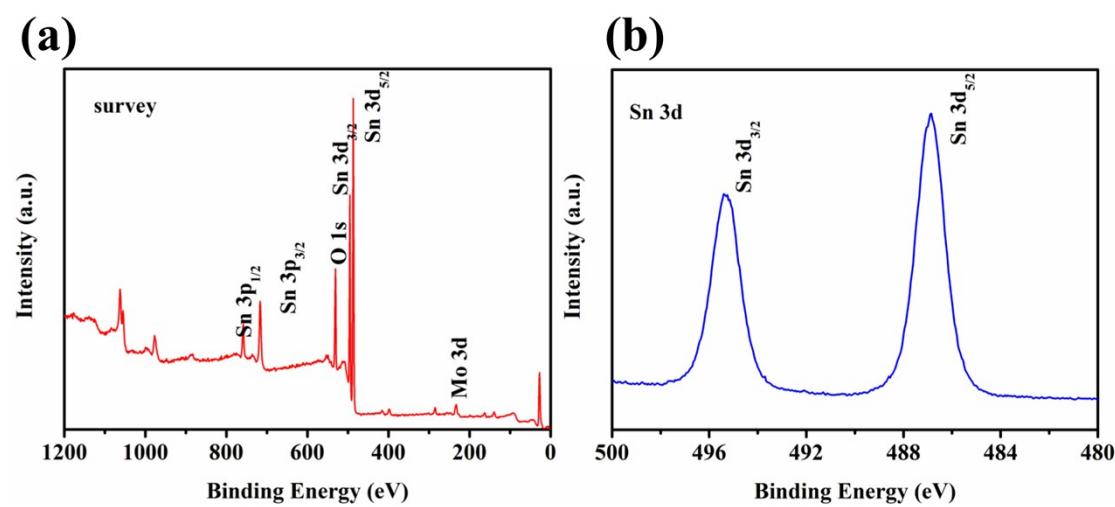


Fig. S2 XPS spectra of Mo-doped SnO₂ sample: (a) survey spectrum and (b) Sn 3d.

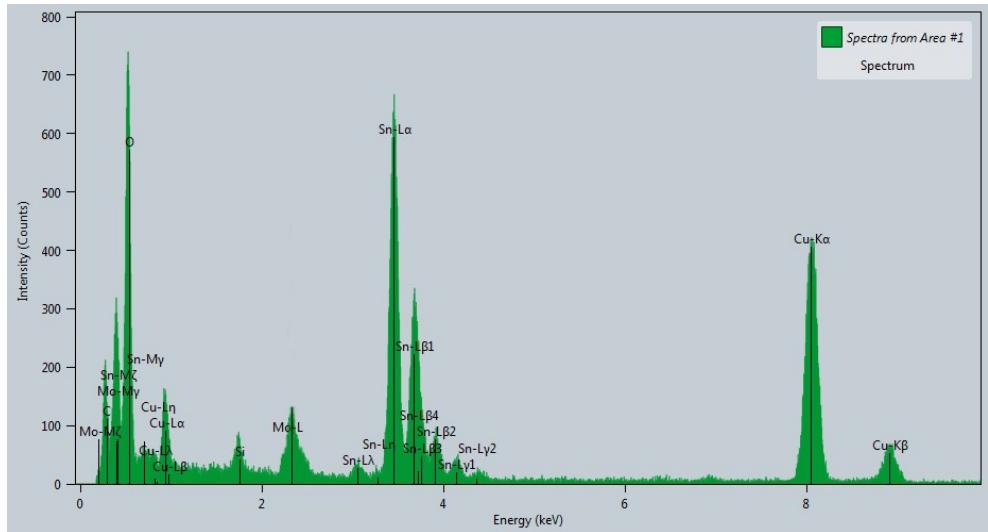


Fig. S3 EDX pattern of the Mo-doped SnO₂ NPs, implying a 3.6 wt% contnt of Mo.

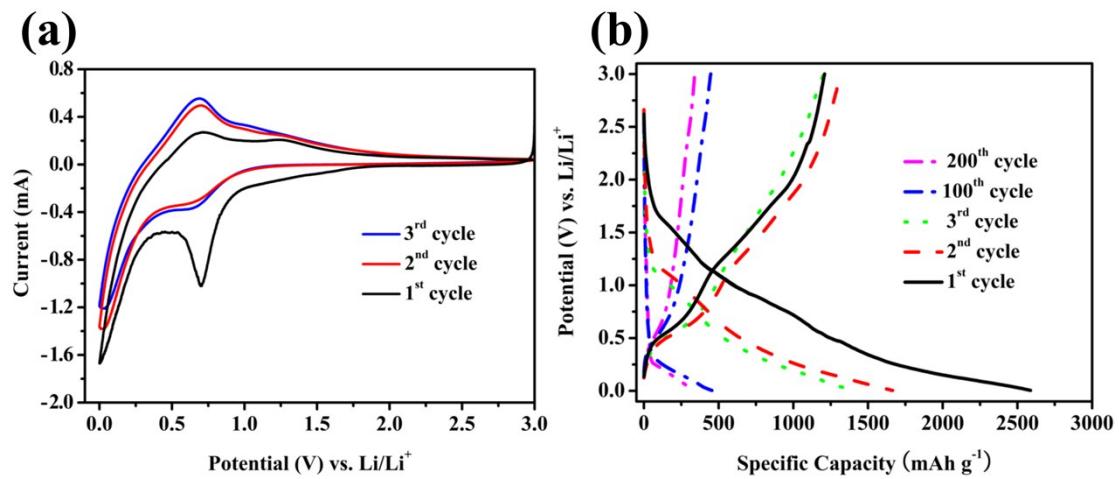


Fig. S4 (a) CV curves and (b) charge-discharge profiles of cells with pure SnO₂ anode.

Table S1 Summary of capacities of various doping SnO₂ anodes and this work.

Doping SnO ₂	Current density (mA/g)	Specific Capacity (mAh/g)	Potential range	Ref.
Sb doped SnO ₂ microspheres	156	194 (after 50 cycles)	0.005-3.0	1
Bowl-like SnO ₂ @Carbon Particles	400	963 (after 100 cycles)	0.005-3.0	2
Multilayer Zn-doped SnO ₂ nanospheres	1000	446 (after 1000 cycles)	0.01-2.5	3
C/Cr-SnO ₂ /G	100	611 (after 200 cycles)	0.01-3.0	4
Ti-doped SnO ₂ /RGO	100	470 (after 200 cycles)	0.01-3.0	5
Sb-doped SnO ₂ / graphene- CNT aerogels	1000	685 (after 1000 cycles)	0.005-3.0	6
Antimony- Doped SnO ₂ Nanopowders	156	637 (after 100 cycles)	0.005-3.0	7
Mo-doped SnO ₂ Hollow Spheres	100	801 (after 60 cycles)	0.01-3.0	8
In-doped SnO ₂ Nanocrystalline	500	500 (after 10 cycles)	0.005-3.0	9
Mo-doped SnO ₂ nanoparticle	500	670 (after 700 cycles)	0.005-3.0	This work

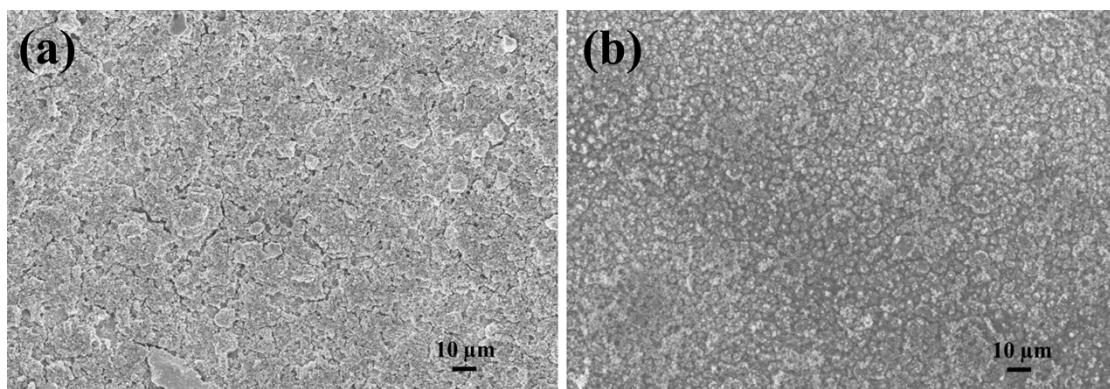


Fig. S5 SEM images of Mo-doped SnO₂ electrode: (a) before and (b) after 50 cycles.

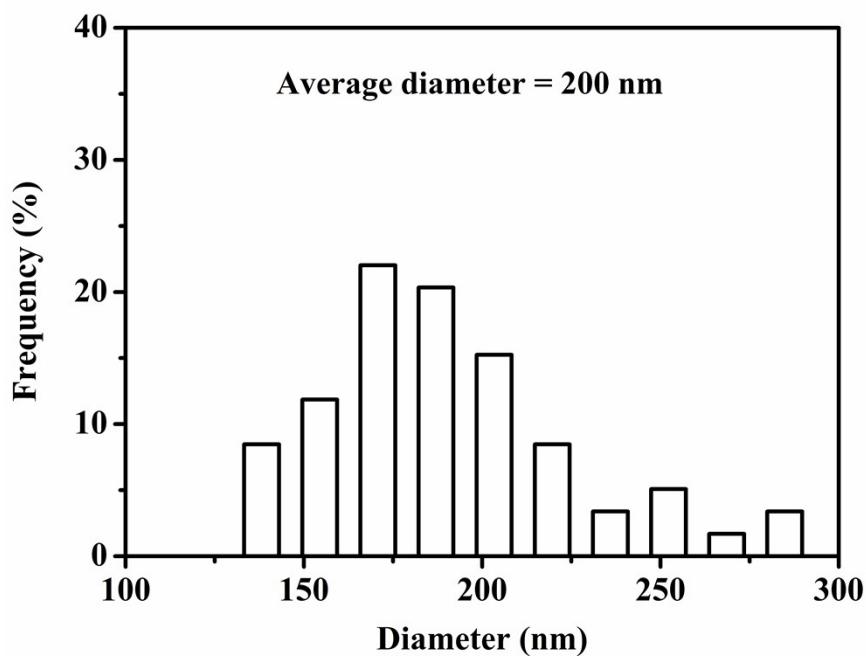


Fig. S6 Diameter distribution chart of the SMCNFs; the average diameter is about 200 nm.

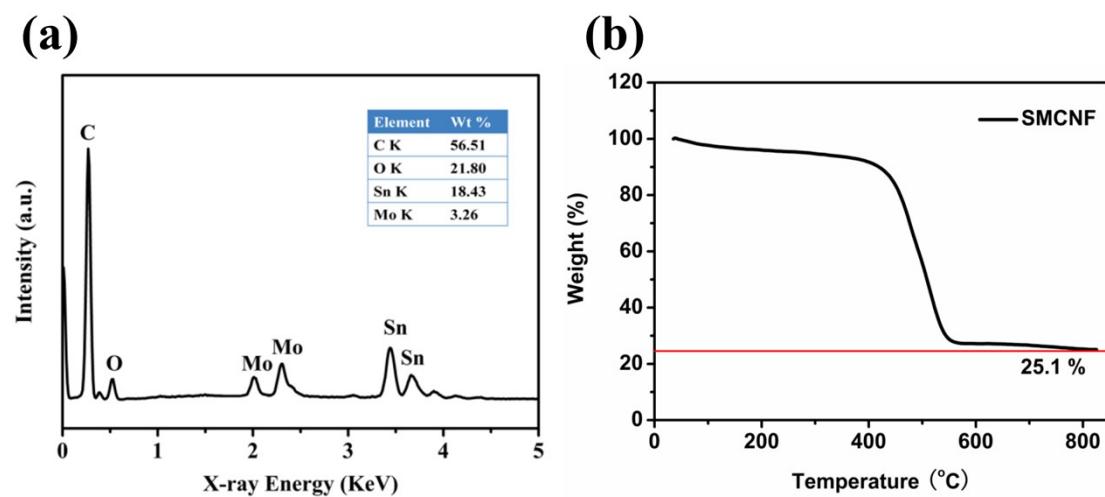


Fig. S7 (a) EDS spectrum and (b) TGA curves of SCNFs sample.

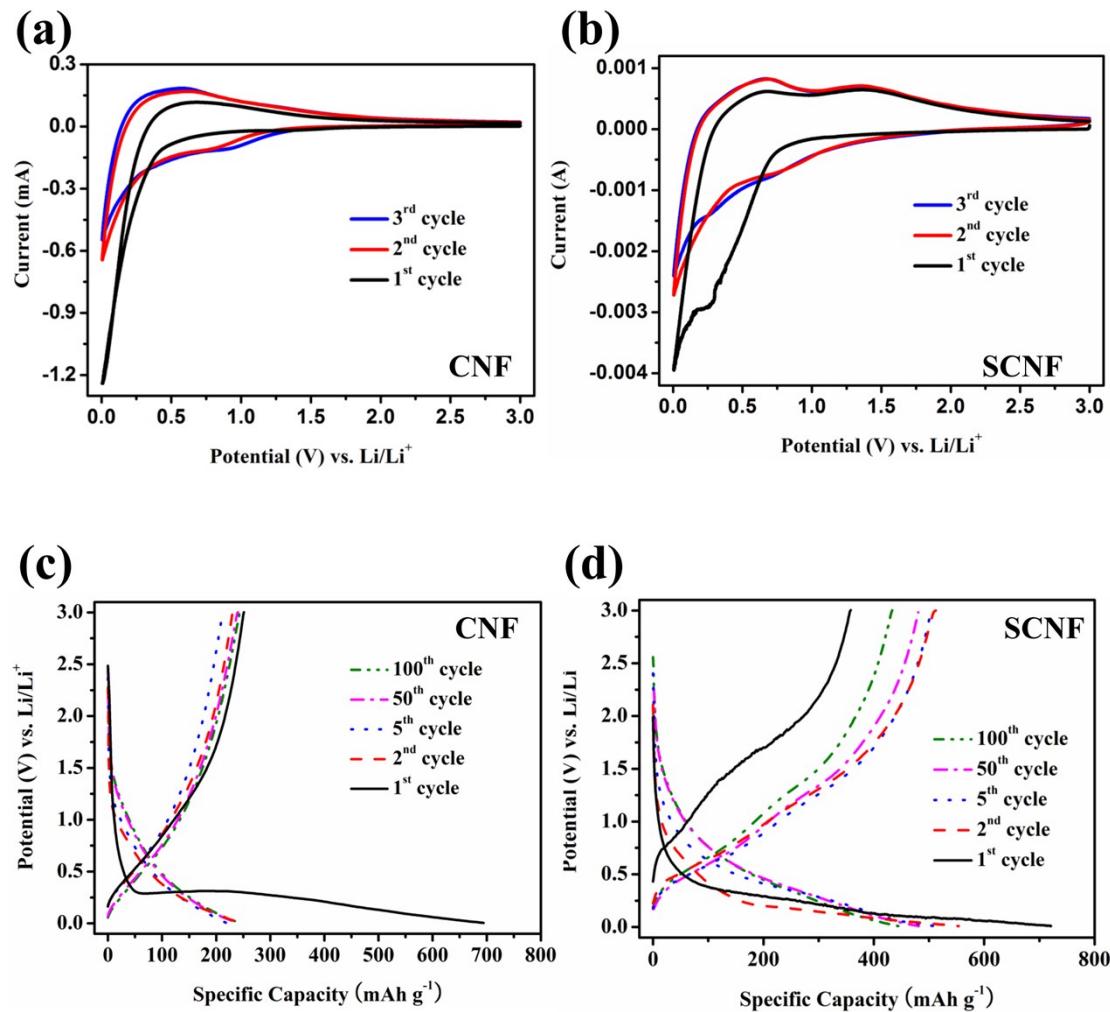


Fig. S8 CV curves of the cells with CNF anode (a) and cells with SCNF anode (b); charge-discharge profiles of cells with CNF anode (c) and cells with SCNF anode (d).

Table S2 Comparison of electrochemical performance of SMCNF anode in this work and other Sn-based/carbon nanofiber anode

Material	Current density (mA/g)	Specific Capacity (mAh/g)	Cycles	Ref.
Sn/SnO _x /CNF	30	510	40	10
Sn/C nanofibers	50	457	50	11
Co–Sn alloy/carbon nanofibers	161	560	80	12
Ni-Sn/Carbon Nanofibers	500	641	100	13
Sn/carbon nanofiber	75	741.1	40	14
Sn NP-dispersed in carbon nanofibers	25	450	30	15
Sn NP/Carbon nanofiber	200	460	200	16
Mo-doped Sn/carbon nanofiber	500	617	200	This work

References:

1. Y. Wang and T. Chen, *Electrochim. Acta*, 2009, **54**, 3510-3515.
2. J. Liang, X.-Y. Yu, H. Zhou, H. B. Wu, S. Ding and X. W. Lou, *Angew. Chem. Int. Edit.* 2014, **53**, 12803-12807.
3. P. Dou, Z. Cao, C. Wang, J. Zheng and X. Xu, *Chem. Eng. J.* 2017, **320**, 405-415.
4. T. Yang, Y. Liu and M. Zhang, *Solid State Ionics*, 2017, **308**, 1-7.
5. C. Ba, L. Shi, Z. Wang, G. Chen, S. Wang, Y. Zhao, M. Zhang and S. Yuan, *Res. Chem. Intermediat.* 2017, **43**, 5857-5869.
6. J. Cui, S. Yao, J.-Q. Huang, L. Qin, W. G. Chong, Z. Sadighi, J. Huang, Z. Wang and J.-K. Kim, *Energy Storage Mater.* 2017, **9**, 85-95.
7. Y. Wang, I. Djerdj, B. Smarsly and M. Antonietti, *Chem. Mater.* 2009, **21**, 3202-3209.
8. X. Wang, Z. Li, Z. Zhang, Q. Li, E. Guo, C. Wang and L. Yin, *Nanoscale*, 2015, **7**, 3604-3613.

9. V. Subramanian, Solid State Ionics, 2004, **175**, 181-184.
10. L. Zou, L. Gan, R. Lv, M. Wang, Z.-h. Huang, F. Kang and W. Shen, Carbon, 2011, **49**, 89-95.
11. X. Xia, X. Wang, H. Zhou, X. Niu, L. Xue, X. Zhang and Q. Wei, Electrochim. Acta, 2014, **121**, 345-351.
12. B.-O. Jang, S.-H. Park and W.-J. Lee, J. Alloy. Compd. 2013, **574**, 325-330.
13. H.-R. Jung and W.-J. Lee, J. Electrochem. Soc. 2011, **158**, A644.
14. H. Wang, P. Gao, S. Lu, H. Liu, G. Yang, J. Pinto and X. Jiang, Electrochim. Acta, 2011, **58**, 44-51.
15. I. Meschini, F. Nobili, M. Mancini, R. Marassi, R. Tossici, A. Savoini, M. L. Focarete and F. Croce, J. Power Sources, 2013, **226**, 241-248.
16. Y. Yu, Q. Yang, D. Teng, X. Yang and S. Ryu, Electrochim. Commun. 2010, **12**, 1187-1190.