

Supporting Information

Local Symmetric Distortion Boosted Photon Up-conversion and Thermometric Sensitivity in Lanthanum Oxide Nanospheres

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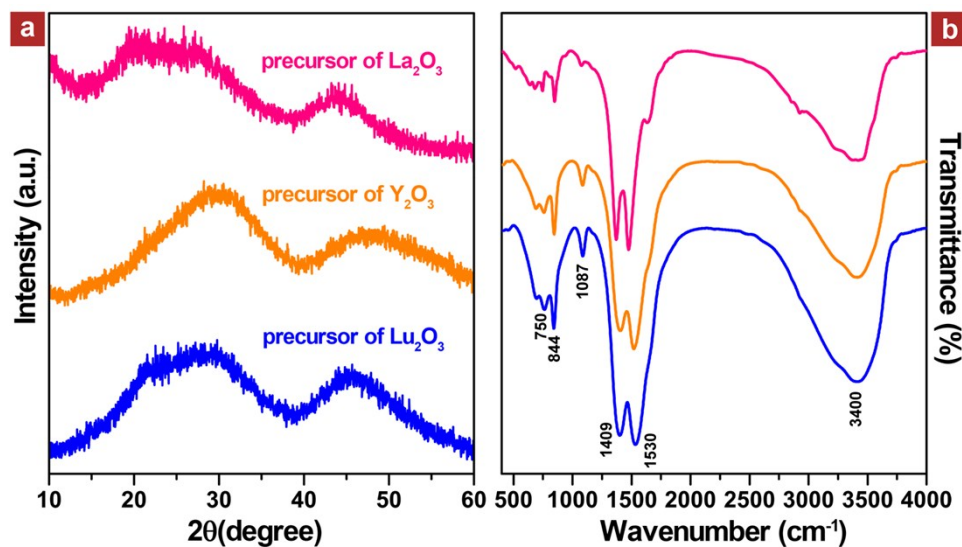


Figure S1. (a) XRD patterns and (b) FT-IR spectra of 2% Er^{3+} /3% Yb^{3+} co-doped three precursors. All precursor samples exhibited two broad bands without obvious diffraction peaks, which could be inferred as amorphous carbonate hydroxides $\text{RE}(\text{OH})\text{CO}_3$ ($\text{RE} = \text{Lu}, \text{Y}$ and La). Three precursor samples exhibited similar FT-IR spectra, in which the broad absorption band at around 3400 cm^{-1} were assigned to the stretching vibrations of hydroxyl groups. The absorption peaks centered at 750, 844 and 1087 cm^{-1} in the precursors were attributed to carbonate while the absorption at around 1409 and 1530 cm^{-1} corresponded to COO^- group, which further confirmed the component of the precursors.

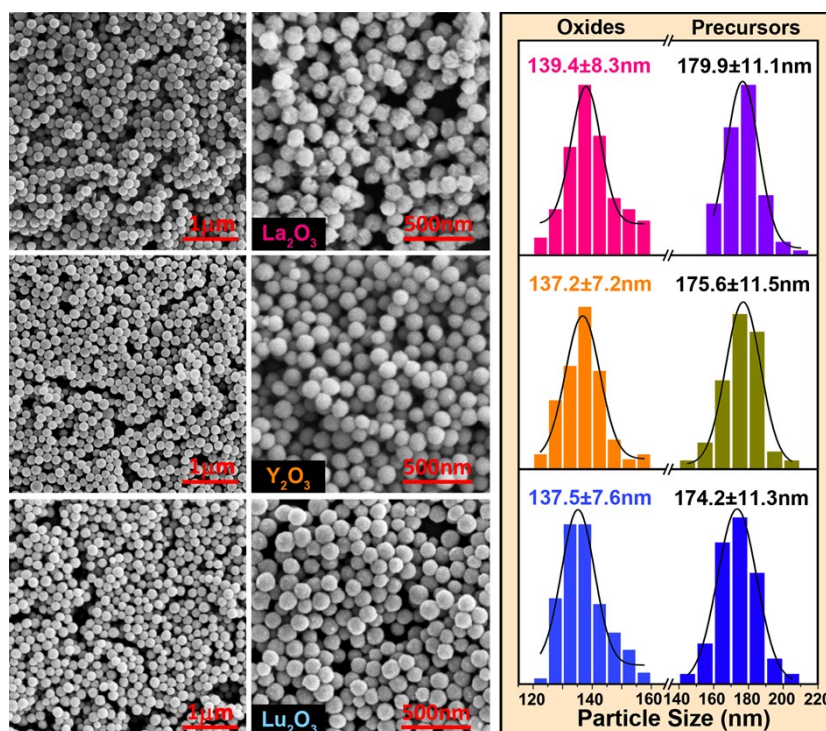


Figure S2. SEM images and size distributions of 2%Er³⁺/3%Yb³⁺ co-doped precursor and final RE₂O₃ (RE = Lu, Y and La) samples. The precursor samples of Lu₂O₃, Y₂O₃ and La₂O₃ exhibited highly monodispersed and regular spherical shape with uniform size of about 174.2 ± 11.3, 175.6 ± 11.5 and 179.9 ± 11.1 nm, respectively. After heating treatment at 800 °C for 3 h, resulting RE₂O₃ particles still maintained highly dispersive and original spherical shape with rough surfaces and decreased size distributions of 137.2 ± 7.2, 137.5 ± 7.6 and 139.4 ± 8.3 nm.

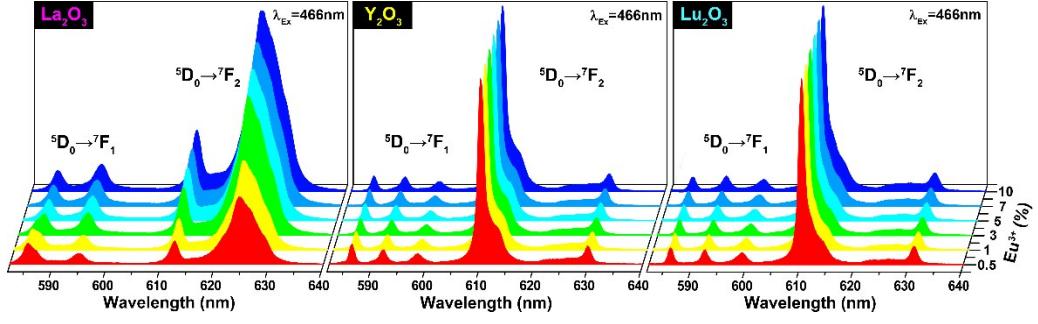


Figure S3. Emission spectra of $\text{RE}_2\text{O}_3: x\text{Eu}^{3+}$ (RE = Lu, Y and La; $x = 0.5\% \sim 10\%$) nanospheres ranging from 580 to 640 nm under 466 nm excitation. All the spectra were normalized at magnetic-dipole (MD) $^5\text{D}_0 \rightarrow ^7\text{F}_1$ transition for better comparison. With increasing Eu^{3+} contents from 0.5% to 10%, the relative emission intensity from $^5\text{D}_0 \rightarrow ^7\text{F}_2$ transition monotonously increased in La_2O_3 while maintained nearly unchanged in Lu_2O_3 and Y_2O_3 samples.

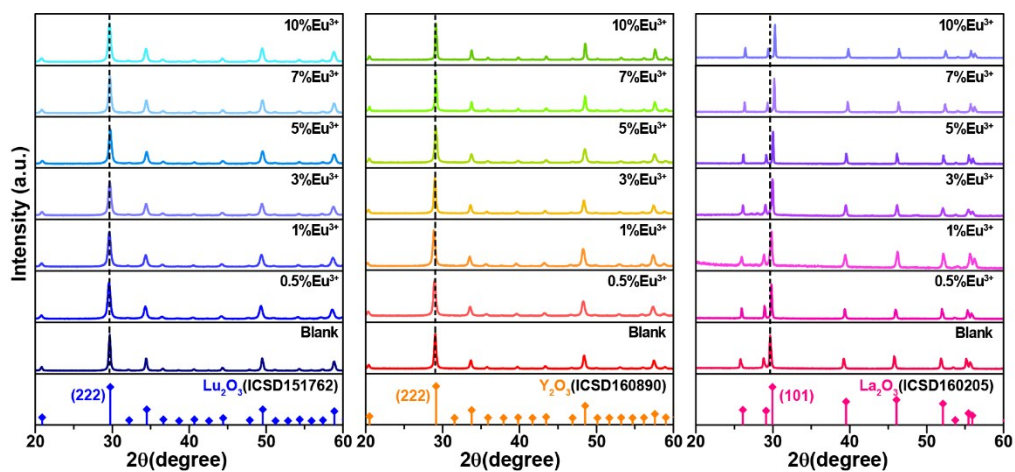


Figure S4. XRD patterns of $\text{RE}_2\text{O}_3: x\text{Eu}^{3+}$ ($\text{RE} = \text{Lu}, \text{Y}$ and La ; $x = 0\% \sim 10\%$) nanospheres along with their standard profiles. All the prepared samples are purity and well-crystallized without secondary phases.

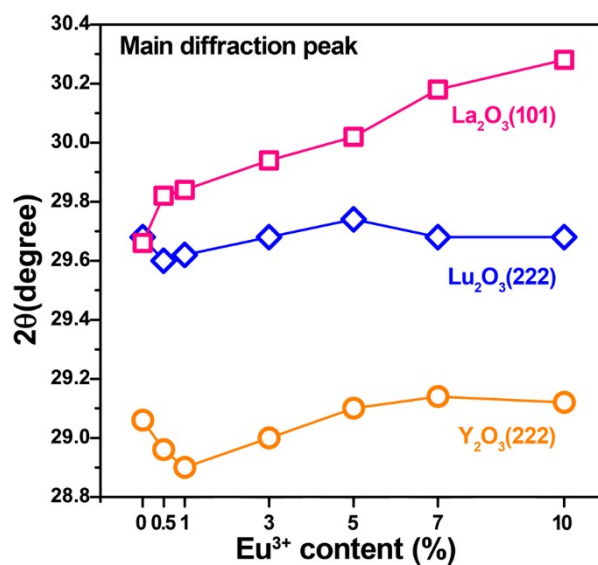


Figure S5. The shift of strongest diffraction peaks in XRD patterns of $\text{RE}_2\text{O}_3: x\text{Eu}^{3+}$ ($\text{RE} = \text{Lu}, \text{Y}$ and La ; $x = 0\% \sim 10\%$) nanospheres. Continuously shift of diffraction peaks (101) from low towards the higher angles was found with increasing Eu^{3+} contents in La_2O_3 due to the substitution of La^{3+} by Eu^{3+} ions with much smaller radius. However, no obvious shift was found in diffraction peaks (222) of Lu_2O_3 and Y_2O_3 samples as increasing Eu^{3+} contents. Above results indicated that the introduction of Eu^{3+} would result the lattice shrinkage of La_2O_3 , yet barely affect the cell volumes of Lu_2O_3 and Y_2O_3 samples.

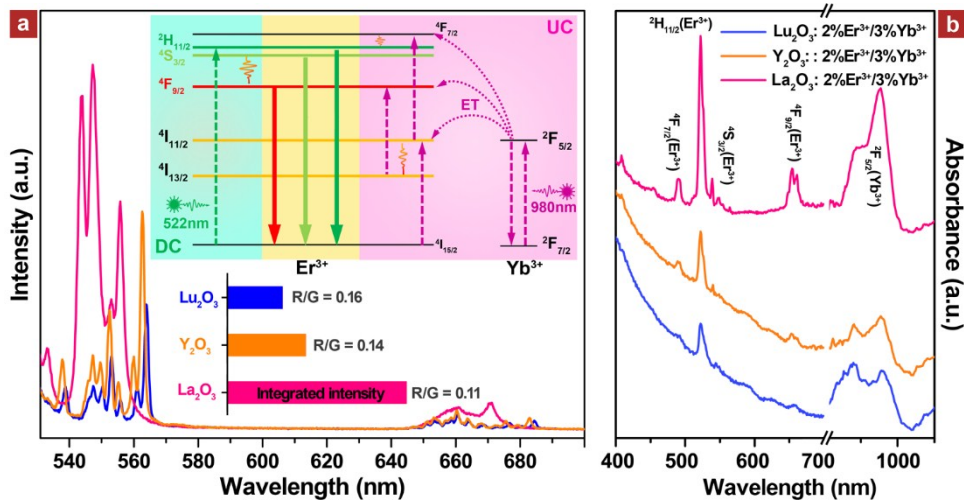


Figure S6. (a) DC emission spectra of RE_2O_3 : 2% Er^{3+} /3% Yb^{3+} (RE = Lu, Y and La) nanospheres with corresponding integrated intensity and R/G ratio under 522 nm excitation, and the inset shows the simplified energy level diagram and possible energy transfer mechanisms in the DC and UC process; (b) UV-vis absorption spectra of three samples. Under 522 nm excitation, trigonal-phased La_2O_3 nanospheres exhibited about twice stronger visible emission than that of cubic-phased samples, and the DC emission intensity is mainly associated with the absorption ability around 522 nm and the radiative transition rates of emitting levels of Er^{3+} in three samples. This indicated that the lifetime of metastable states $^4\text{I}_{11/2}$ and $^4\text{I}_{13/2}$ plays a decisive role in determining UCL efficiency, as depicted in schematic energy level diagram. Local symmetric distortion enhanced f-f absorption and transition probabilities in La_2O_3 crystal could be further confirmed *via* UV-vis absorption spectra, from which the characteristic absorption bands at 490, 522, 539, 654 and 977 nm^{-1} corresponding to $^4\text{F}_{7/2}$, $^2\text{H}_{11/2}$, $^4\text{S}_{3/2}$, $^4\text{F}_{9/2} \rightarrow ^4\text{I}_{15/2}$ of Er^{3+} and $^2\text{F}_{5/2} \rightarrow ^2\text{F}_{7/2}$ of Yb^{3+} in La_2O_3 sample were all stronger than that in cubic samples.

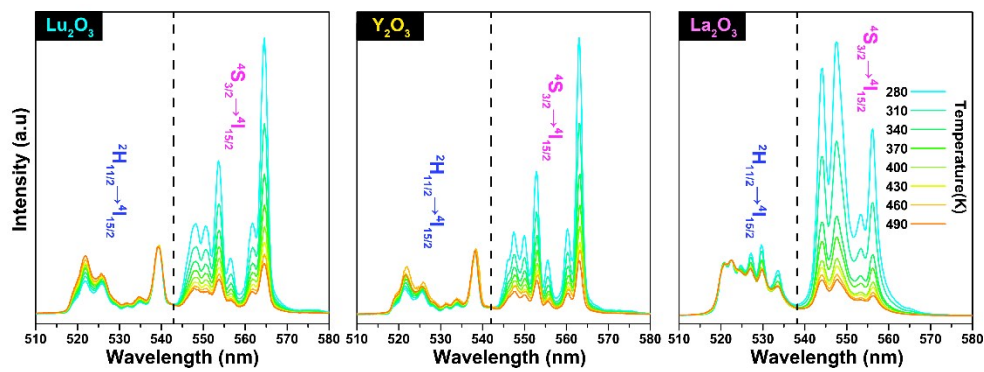


Figure S7. Temperature-dependent UC emission spectra of $\text{RE}_2\text{O}_3: 2\%\text{Er}^{3+}/3\%\text{Yb}^{3+}$ (RE = Lu, Y and La) nanospheres within green region under 980 nm excitation (normalized at $^2\text{H}_{11/2} \rightarrow ^4\text{I}_{15/2}$). With the increase of temperature from 280 to 490 K, the relative emission intensity of $^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$ monotonously declined in three samples without obvious change in band position.

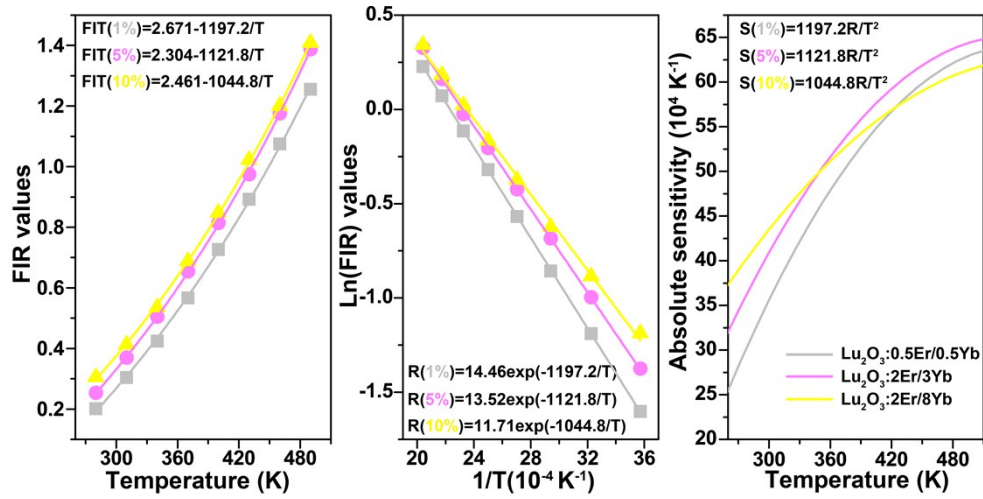


Figure S8. Temperature-dependent FIR and absolute sensitivity based on green emission in Lu_2O_3 doping with different $\text{Er}^{3+}/\text{Yb}^{3+}$ contents under 980 nm excitation. According to the optimal fitting exponential curves of experimental points, the energy separation and coefficient B were found to be 825 cm^{-1} and 14.46 ($0.5\%\text{Er}^{3+}/0.5\%\text{Yb}^{3+}$), 775 cm^{-1} and 13.52 ($2\%\text{Er}^{3+}/3\%\text{Yb}^{3+}$), 721 cm^{-1} and 11.71 ($2\%\text{Er}^{3+}/8\%\text{Yb}^{3+}$), respectively. The obtained absolute sensitivity of samples gradually increased in our experimental range and reached to the maximum about 0.0063, 0.0064 and 0.0061 K^{-1} at the upper limit temperature 490 K.

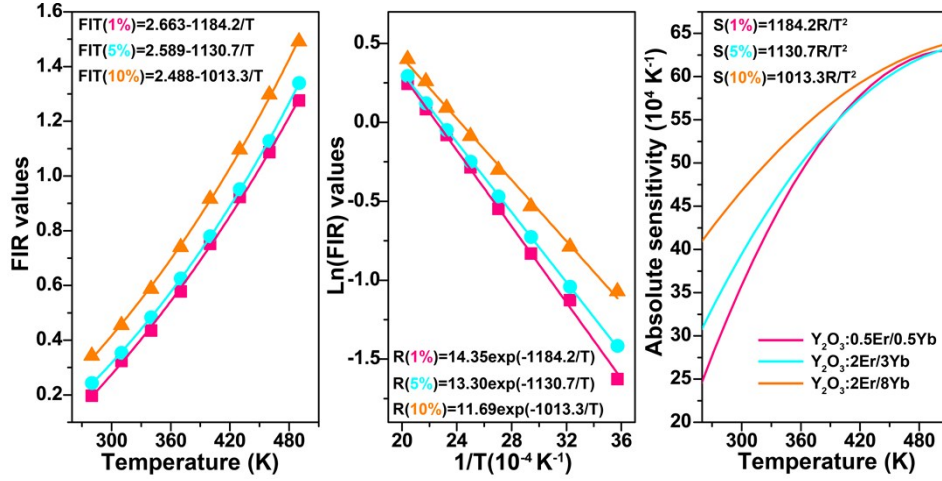


Figure S9. Temperature-dependent FIR and absolute sensitivity based on green emission in Y_2O_3 doping with different $\text{Er}^{3+}/\text{Yb}^{3+}$ contents under 980 nm excitation. According to the optimal fitting exponential curves of experimental points, the energy separation and coefficient B were found to be 817 cm^{-1} and 14.35 ($0.5\%\text{Er}^{3+}/0.5\%\text{Yb}^{3+}$), 780 cm^{-1} and 13.3 ($2\%\text{Er}^{3+}/3\%\text{Yb}^{3+}$), 700 cm^{-1} and 11.69 ($2\%\text{Er}^{3+}/8\%\text{Yb}^{3+}$), respectively. The obtained absolute sensitivity of all samples gradually increased in our experimental range and reached to the maximum about 0.0063 K^{-1} at the upper limit temperature 490 K.

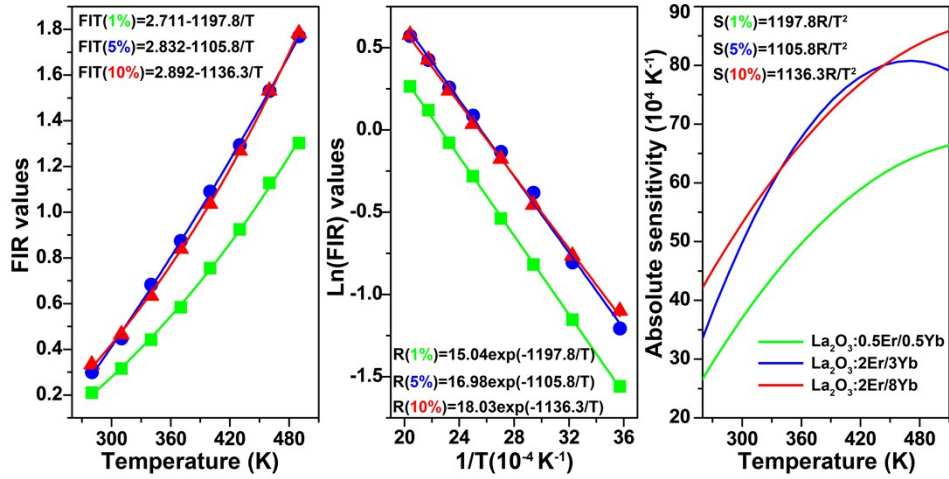


Figure S10. Temperature-dependent FIR and absolute sensitivity based on green emission in La_2O_3 doping with different $\text{Er}^{3+}/\text{Yb}^{3+}$ contents under 980 nm excitation. According to the optimal fitting exponential curves of experimental points, the energy separation and coefficient B were found to be 825 cm^{-1} and 15.04 ($0.5\%\text{Er}^{3+}/0.5\%\text{Yb}^{3+}$), 763 cm^{-1} and 16.98 ($2\%\text{Er}^{3+}/3\%\text{Yb}^{3+}$), 784 cm^{-1} and 18.03 ($2\%\text{Er}^{3+}/8\%\text{Yb}^{3+}$), respectively. The obtained absolute sensitivity of three samples gradually increased in our experimental range and reached to the maximum about 0.0065, 0.0081 and 0.0085 K^{-1} at the upper limit temperature 490 K.

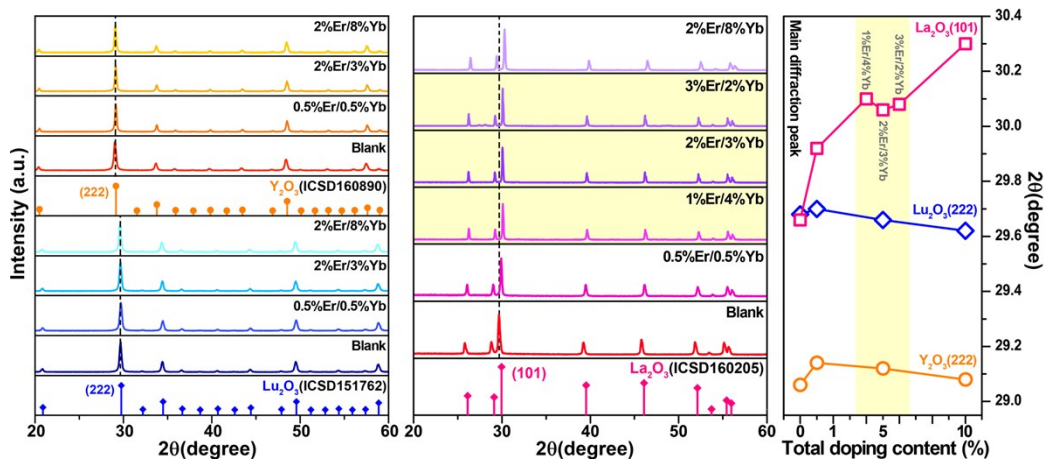


Figure S11. XRD patterns of RE_2O_3 (RE = Lu, Y and La) nanospheres doping with different $\text{Er}^{3+}/\text{Yb}^{3+}$ contents and the corresponding shift of the strongest diffraction peaks. All the prepared samples are purity and well-crystallized without secondary phases. It is found that continuously shift of diffraction peaks (101) from low towards the higher angles was observed with increasing total contents of $\text{Er}^{3+}/\text{Yb}^{3+}$ in La_2O_3 due to the substitution of La^{3+} by $\text{Er}^{3+}/\text{Yb}^{3+}$ ions with much smaller radius, and the positions of (101) peak remained almost unchanged in La_2O_3 with 5% total doping content ($1\%\text{Er}^{3+}/4\%\text{Yb}^{3+}$, $2\%\text{Er}^{3+}/3\%\text{Yb}^{3+}$ and $3\%\text{Er}^{3+}/2\%\text{Yb}^{3+}$). However, no obvious shift were observed in diffraction peaks (222) of Lu_2O_3 and Y_2O_3 samples as increasing $\text{Er}^{3+}/\text{Yb}^{3+}$ contents. Above results indicated that the introduction of $\text{Er}^{3+}/\text{Yb}^{3+}$ would result drastic lattice shrinkage of La_2O_3 crystal, yet barely affect the cell volumes of Lu_2O_3 and Y_2O_3 samples.

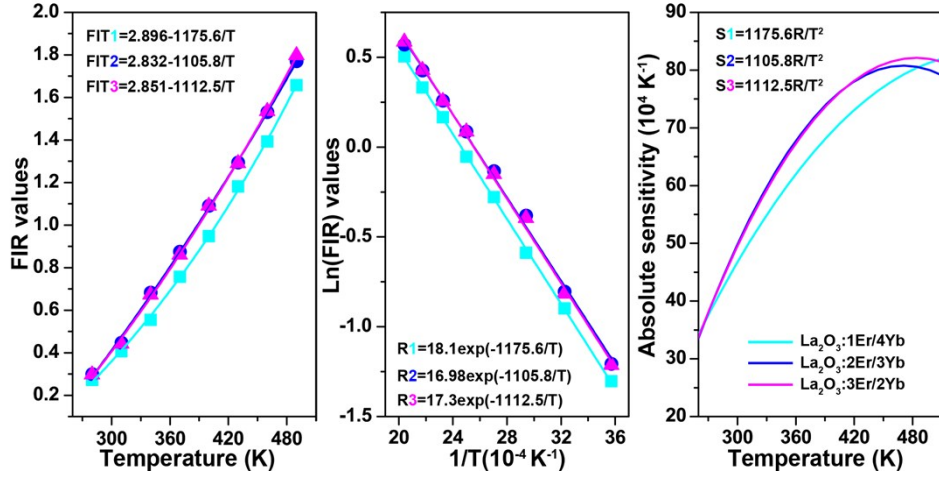


Figure S12. Temperature-dependent FIR and absolute sensitivity based on green emission in La_2O_3 samples doping with 5% total doping content under 980 nm excitation. According to the optimal fitting exponential curves of experimental points, the energy separation and coefficient B were found to be 810 cm^{-1} and 18.1 ($1\%\text{Er}^{3+}/4\%\text{Yb}^{3+}$), 763 cm^{-1} and 16.98 ($2\%\text{Er}^{3+}/3\%\text{Yb}^{3+}$), 767 cm^{-1} and 17.3 ($3\%\text{Er}^{3+}/2\%\text{Yb}^{3+}$), respectively. The obtained absolute sensitivity of three samples gradually increased in our experimental range and reached to the maximum about 0.0081, 0.0081 and 0.0083 K^{-1} at the upper limit temperature 490 K, indicating that the concentration-dependent energy transfer process hardly influence the sensitivity in present systems.

Table S1. Rietveld refinement and calculated crystallographic results for RE₂O₃: 2%Er³⁺/3%Yb³⁺ (RE = Lu, Y and La) samples.

	Lu ₂ O ₃ : 2%Er ³⁺ /3%Yb ³⁺	Y ₂ O ₃ : 2%Er ³⁺ /3%Yb ³⁺	La ₂ O ₃ : 2%Er ³⁺ /3%Yb ³⁺
Space group	<i>Ia</i> - $\bar{3}$, cubic	<i>Ia</i> - $\bar{3}$, cubic	<i>p</i> - $\bar{3}m1$, trigonal
<i>a</i> (Å)	10.3968	10.5971	3.9237
<i>b</i> (Å)	10.3968	10.5971	3.9237
<i>c</i> (Å)	10.3968	10.5971	6.1170
α, β, γ (deg)	90, 90, 90	90, 90, 90	90, 90, 120
<i>V</i> (Å ³)	1123.83	1190.04	81.56
<i>Z</i>	16	16	1
<i>R</i> _{wp} (%)	4.806	4.778	4.473
<i>R</i> _p (%)	3.158	3.142	2.879

Table S2. The indexation of SAED patterns of RE₂O₃: 2%Er³⁺/3%Yb³⁺ (RE = Lu, Y and La) nanospheres.

Samples	Crystal facets
Lu₂O₃: 2%Er³⁺/3%Yb³⁺	(211), (222), (321) and (400) of cubic-phased Lu ₂ O ₃
Y₂O₃: 2%Er³⁺/3%Yb³⁺	(211), (222), (321) and (400) of cubic-phased Y ₂ O ₃
La₂O₃: 2%Er³⁺/3%Yb³⁺	(100), (002), (102) and (110) of trigonal-phased La ₂ O ₃

Table S3. Atomic distance of RE-RE in RE₂O₃: 2Er³⁺/3Yb³⁺ (RE = Lu, Y and La) crystals.

Samples	RE-RE	Distance (Å)
Lu₂O₃: 2%Er³⁺/3%Yb³⁺	Lu2-Lu2	3.4563
Y₂O₃: 2%Er³⁺/3%Yb³⁺	Y2-Y2	3.5266
La₂O₃: 2%Er³⁺/3%Yb³⁺	La-La	3.7213

Table S4. The ionic radius of rare earth (RE³⁺) ions with different coordination numbers

RE ³⁺ ions	Ionic Radius (Å, CN = 6)	RE ³⁺ ions	Ionic Radius (Å, CN = 7)
Lu ³⁺	0.86	La ³⁺	1.1
Y ³⁺	0.9	Er ³⁺	0.94
Er ³⁺	0.89	Yb ³⁺	0.92
Yb ³⁺	0.87	Eu ³⁺	1.01
Eu ³⁺	0.94		