

Supplementary Material

Photo-mediated co-loading of highly dispersed MnO_x-Pt on g-C₃N₄ boosts the ambient catalytic oxidation of formaldehyde

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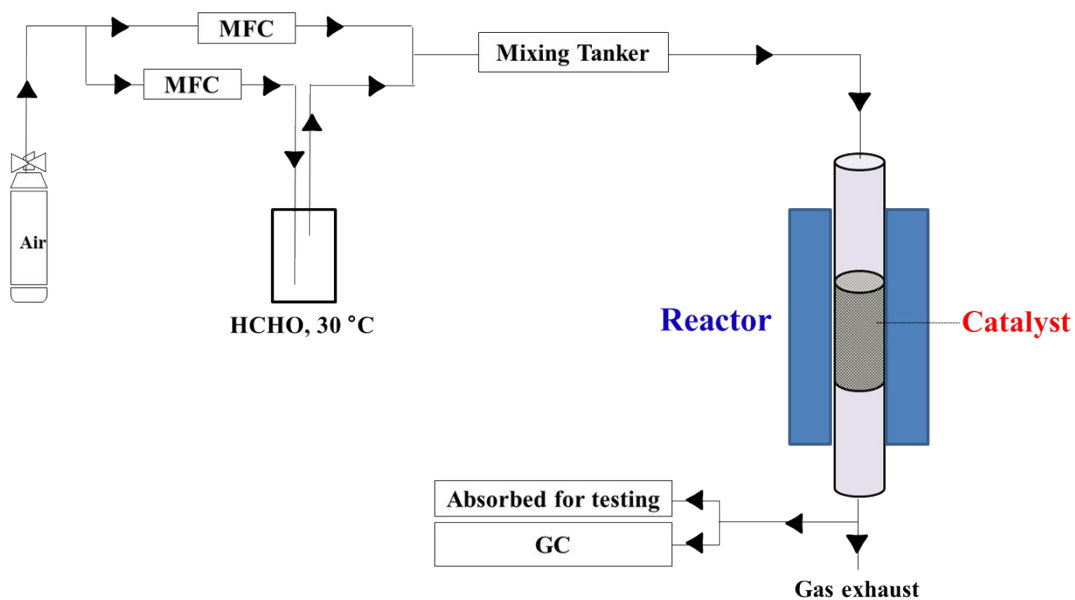


Fig. S1 Schematic representation of experimental setup for catalytic oxidation of HCHO.

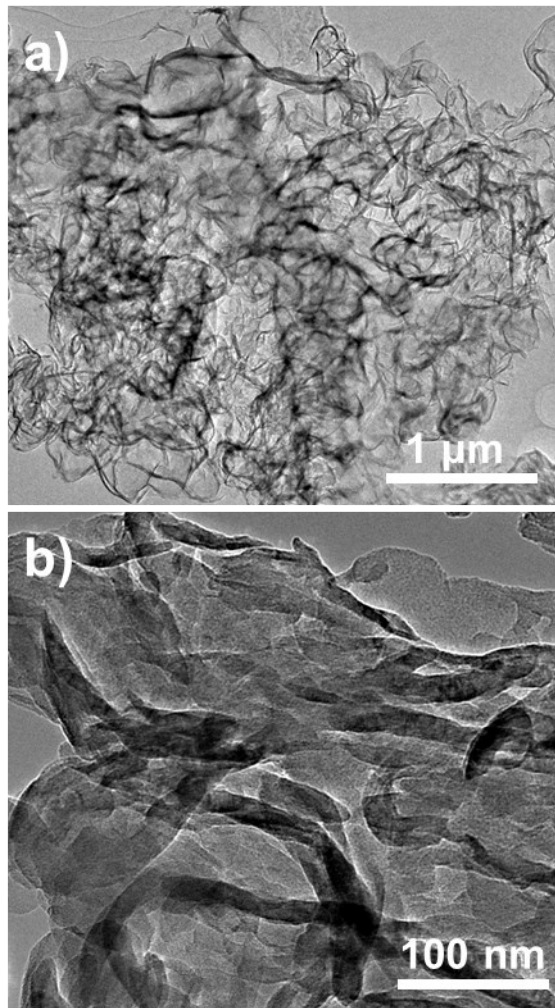


Fig. S2 Field-emission TEM images of g-C₃N₄.

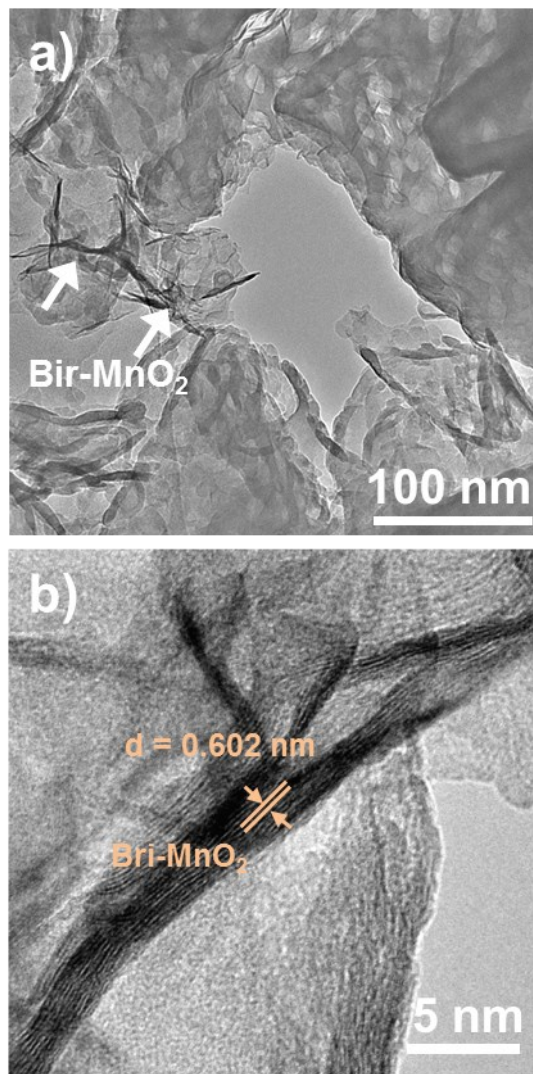


Fig. S3 Field-emission TEM and HRTEM images of MnO_x/g-C₃N₄.

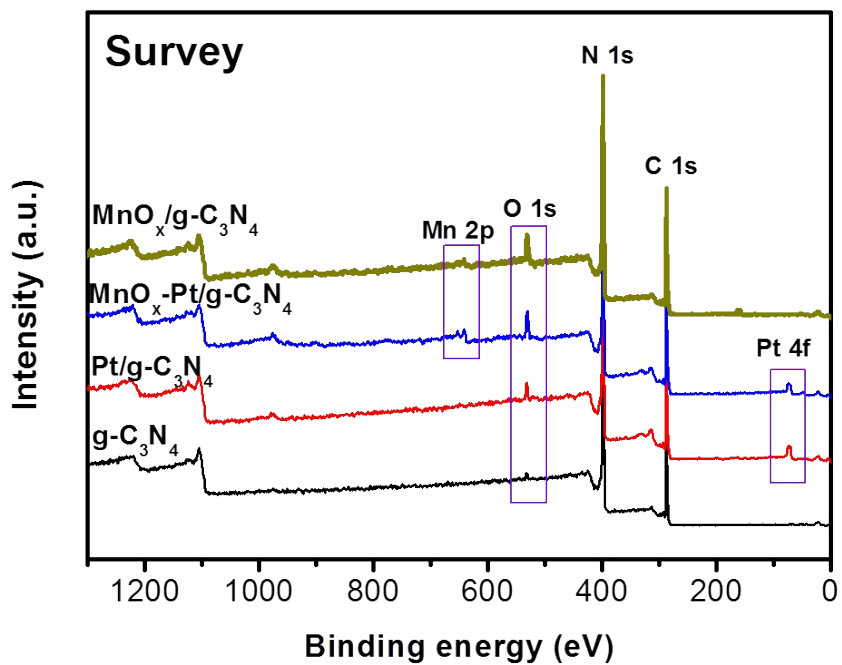


Fig. S4 XPS spectra of survey for various catalysts.

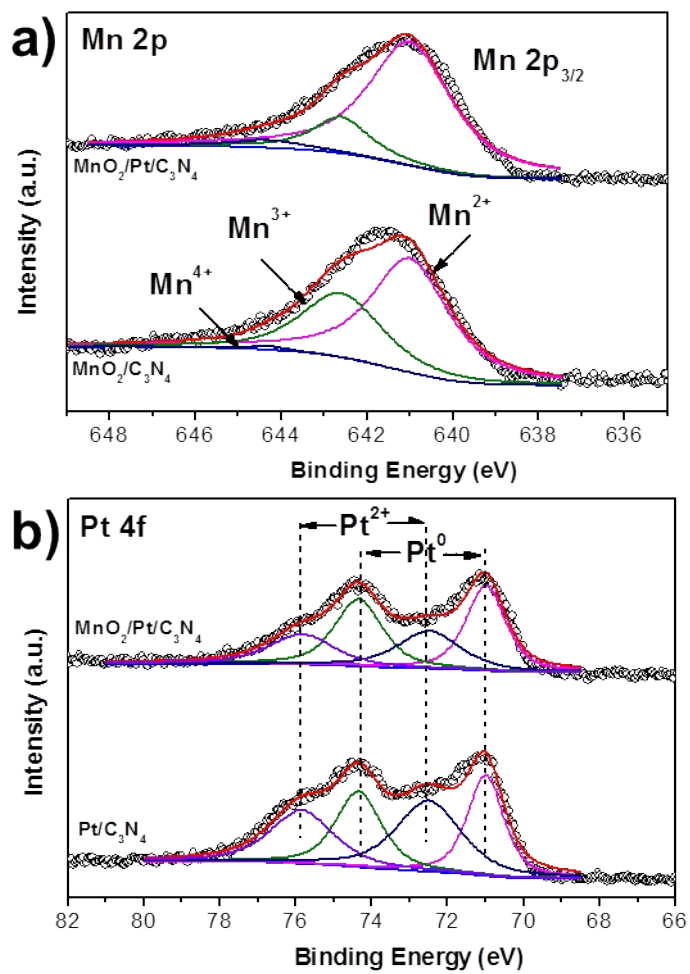


Fig. S. XPS spectra of Mn 2p (a) and Pt 4f (b) of various catalysts.

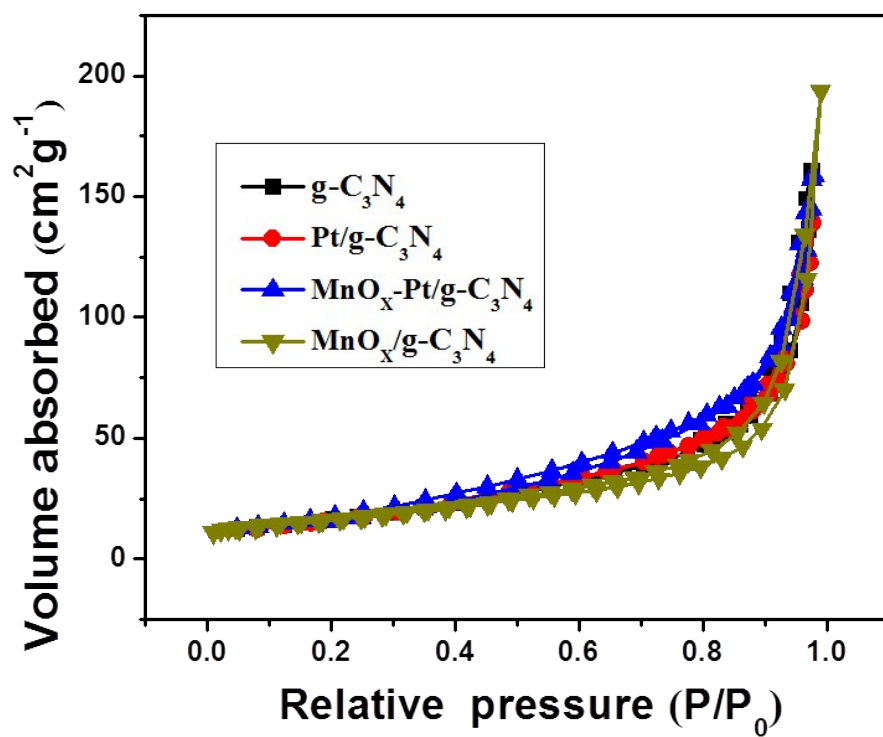


Fig. S6 N₂ adsorption-desorption isotherms of g-C₃N₄, Pt/g-C₃N₄, MnO_x-Pt/g-C₃N₄ and MnO_x/g-C₃N₄.

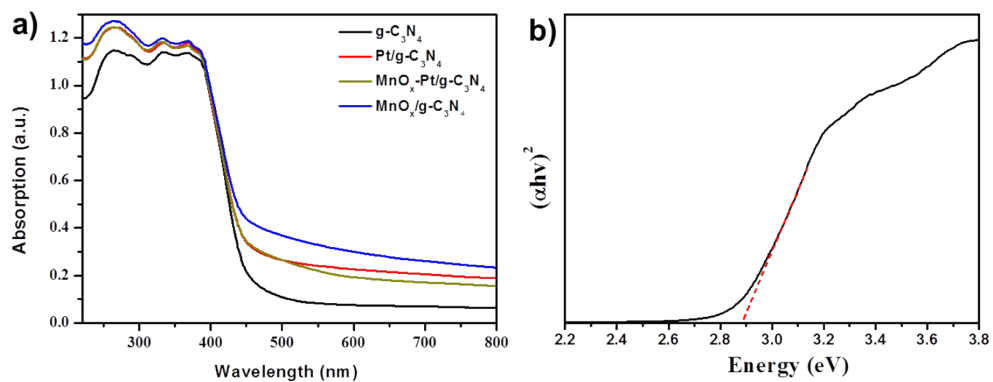


Fig. S7 (a) is UV-vis diffusion absorption spectra of $g\text{-C}_3\text{N}_4$, $\text{Pt}/g\text{-C}_3\text{N}_4$, $\text{MnO}_x\text{-Pt}/g\text{-C}_3\text{N}_4$ and $\text{MnO}_x/g\text{-C}_3\text{N}_4$. (b) is the plots of the $(\alpha h\nu)^2$ versus photon energy ($h\nu$).

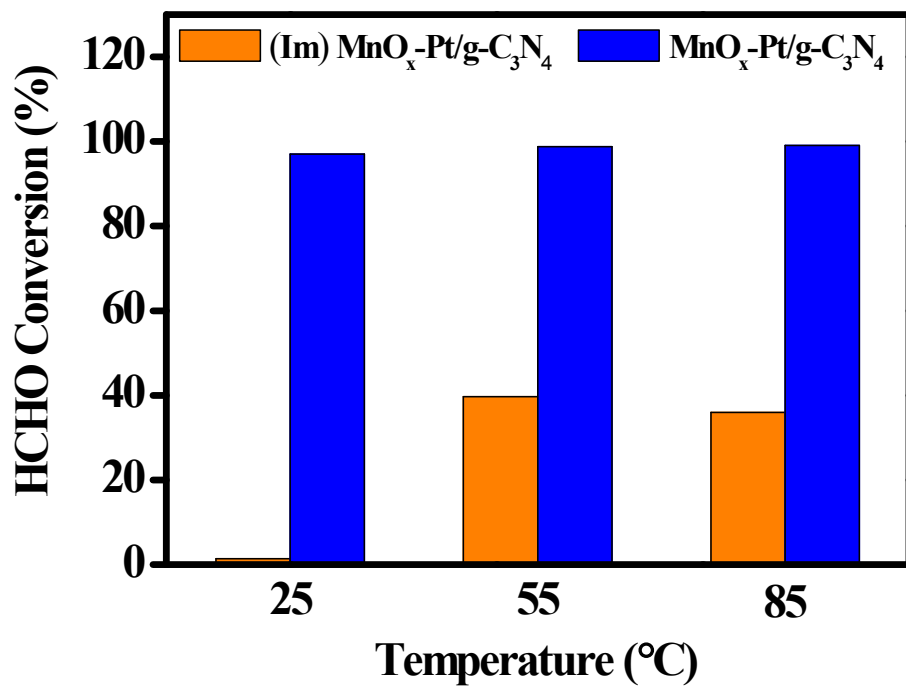


Fig. S8 HCHO conversion over MnO_x-Pt/g-C₃N₄ and (Im)MnO_x-Pt/g-C₃N₄.