

Electrochemically induced oxidative S-O coupling: synthesis of sulfonates from sulfonyl hydrazides and *N*-hydroxyimides or *N*-hydroxybenzotriazoles

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SUPPORTING INFORMATION

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General

NMR spectra were registered on Bruker Avance II 300 MHz instrumental. Chemical shifts were measured relative to residual solvent peaks as an internal standard set to δ 7.25 and δ 77.0 (CDCl_3), δ 2.50 and δ 39.51 (DMSO-d_6). High resolution mass spectra (HRMS) were measured on a Bruker maXis instrument equipped with an electrospray ionization (ESI) ion source.^[1] The measurements were done in a positive ion mode (interface capillary voltage 4500 V); the mass ratio was from m/z 50 to 3000 Da; external/internal calibration was done with Electrospray Calibrant Solution. A syringe injection was used for solutions in CH_3CN (flow rate 3 $\mu\text{L}/\text{min}$). Nitrogen was applied as a dry gas; interface temperature was set at 180°C. The TLC analyses were carried out on standard silica-gel chromatography plates. The melting points were determined on a Kofler hot-stage apparatus. Chromatography was performed on silica gel (60-200 mesh).

p-Toluenesulfonohydrazide (**1a**), benzenesulfonohydrazide (**1j**), methanesulfonohydrazide (**1m**), *N*-hydroxysuccinimide (**2a**), 2-hydroxyisoindoline-1,3-dione (**2b**), 4,5,6,7-tetrachloro-2-hydroxyisoindoline-1,3-dione (**2c**), 2-hydroxy-3a,4,7,7a-tetrahydro-1*H*-4,7-methanoisoindole-1,3-dione (**2d**), 1*H*-benzo[d][1,2,3]triazol-1-ol (**2e**), 7-chloro-1*H*-benzo[d][1,2,3]triazol-1-ol (**2f**), sodium *p*-toluenesulfinate (**4a**), sodium benzenesulfinate (**4j**), 4-methoxybenzenesulfonyl chloride, 4-fluorobenzenesulfonyl chloride, 4-chlorobenzenesulfonyl chloride, 4-bromobenzenesulfonyl chloride, 4-iodobenzenesulfonyl chloride, 4-acetamidobenzenesulfonyl chloride, naphthalene-2-sulfonyl chloride, 2,4,6-trimethylbenzenesulfonyl chloride, isothiazole-5-sulfonyl chloride, thiophene-2-sulfonyl chloride, tetrabutylammonium perchlorate, NH_4I , NH_4Br , NH_4Cl , KBr , NaBr , LiClO_4 , Na_2SO_4 , Br_2 , KOH , THF, MeCN, MeOH, EtOH, CHCl_3 , petroleum ether (PE, 40/70), ethyl acetate (EA) were purchased from commercial sources and were used as is.

4-Methoxybenzenesulfonohydrazide (**1b**), 4-fluorobenzenesulfonohydrazide (**1c**), 4-chlorobenzenesulfonohydrazide (**1d**), 4-bromobenzenesulfonohydrazide (**1e**), 4-iodobenzenesulfonohydrazide (**1f**), 4-acetamidobenzenesulfonohydrazide (**1g**), naphthalene-2-sulfonohydrazide (**1h**), 2,4,6-trimethylbenzenesulfonohydrazide (**1i**), isothiazole-5-sulfonohydrazide (**1k**), thiophene-2-sulfonohydrazide (**1l**) were synthesized according to the literature through the reaction between corresponding sulfonyl chlorides and hydrazine hydrate.^[2] Sodium *p*-chlorobenzene sulfinate (**4d**) was synthesized according to the literature through the reduction of *p*-chlorobenzenesulfonyl chloride.^[3]

Voltammetric studies were carried out using potentiostat Elins P-30JM with the scan 100 $\text{mV}\cdot\text{s}^{-1}$ in a temperature-controlled (25 °C) glass cell ($V = 10 \text{ mL}$) under a nitrogen atmosphere. A glassy carbon disk ($d = 2.9 \text{ mm}$) was used as the working electrode (polished before each measurement). Software iR compensation using ferrocene ($R = 700 \Omega$) was used in all experiments. A saturated calomel electrode (SCE) separated from the solution being studied by a salt bridge filled with the supporting electrolyte (0.1 M Bu_4NClO_4 in H_2O -THF (1:1)) was used as the reference electrode. A platinum plate (3 cm^2) was used as the counter electrode. All experiments were performed with the concentration of a studied compound of 3 mM in H_2O -THF (1:1).

Electrochemical Study of Redox Properties of the Reagents

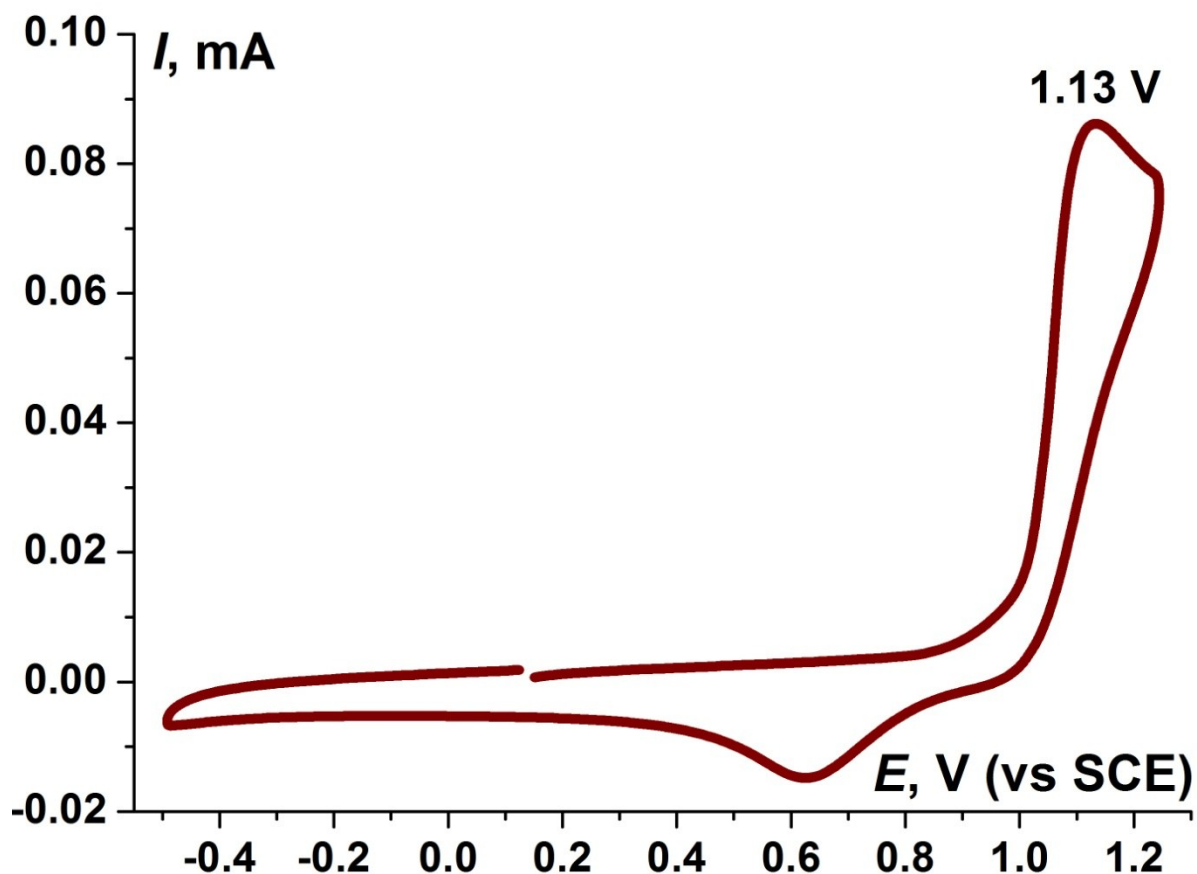


Figure 1a. The CV curve obtained for 3.0 mmol·L⁻¹ solution of NH₄Br in 0.1 M Bu₄NClO₄ in H₂O-THF (1:1) on a working glassy-carbon electrode (d = 2.9 mm) at a scan rate of 100 mV·s⁻¹.

Table 1a. CV curve parameters in V vs SCE for NH₄Br solution in 0.1 M Bu₄NClO₄ in H₂O-THF (1:1) on a working glassy-carbon electrode (d = 2.9 mm) under a scan rate of 100 mV·s⁻¹

<i>Substance</i>	<i>E_{onset}</i>	<i>Scan direction</i>	<i>E_{peak forward}</i>	<i>E_{peak reverse}</i>	<i>E^{1/2}</i>
Br ⁻	0.85	+	1.13	0.63	1.05

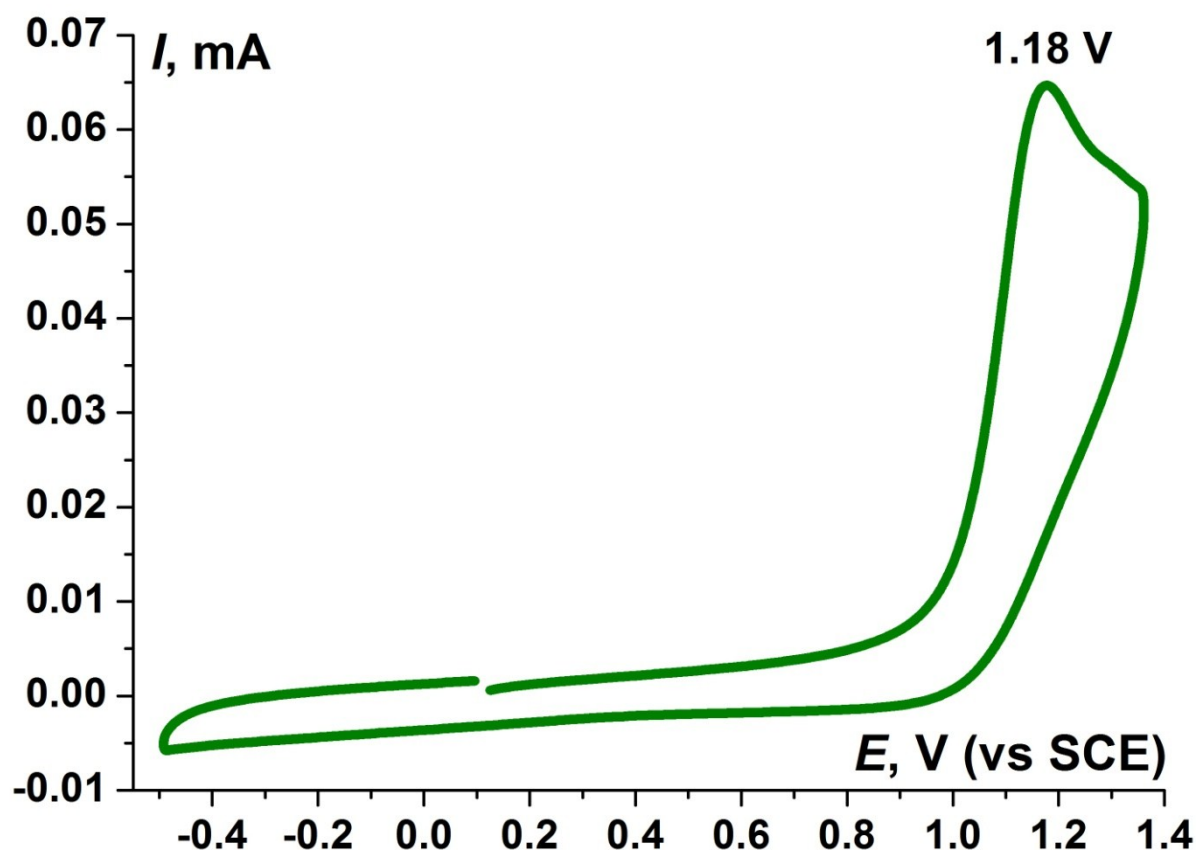


Figure 2a. The CV curve obtained for 3.0 mmol·L⁻¹ solution of *p*-toluenesulfonyl hydrazide **1a** in 0.1 M Bu₄NClO₄ in H₂O-THF (1:1) on a working glassy-carbon electrode (d = 2.9 mm) at a scan rate of 100 mV·s⁻¹.

Table 2a. CV curve parameters in V vs SCE for *p*-toluenesulfonyl hydrazide **1a** solution in 0.1 M Bu₄NClO₄ in H₂O-THF (1:1) on a working glassy-carbon electrode (d = 2.9 mm) under a scan rate of 100 mV·s⁻¹

<i>Substance</i>	E_{onset}	<i>Scan direction</i>	$E_{forward}^{peak}$	$E_{reverse}^{peak}$	$E^{1/2}$
TsNHNH ₂	0.92	+	1.18	-	1.08

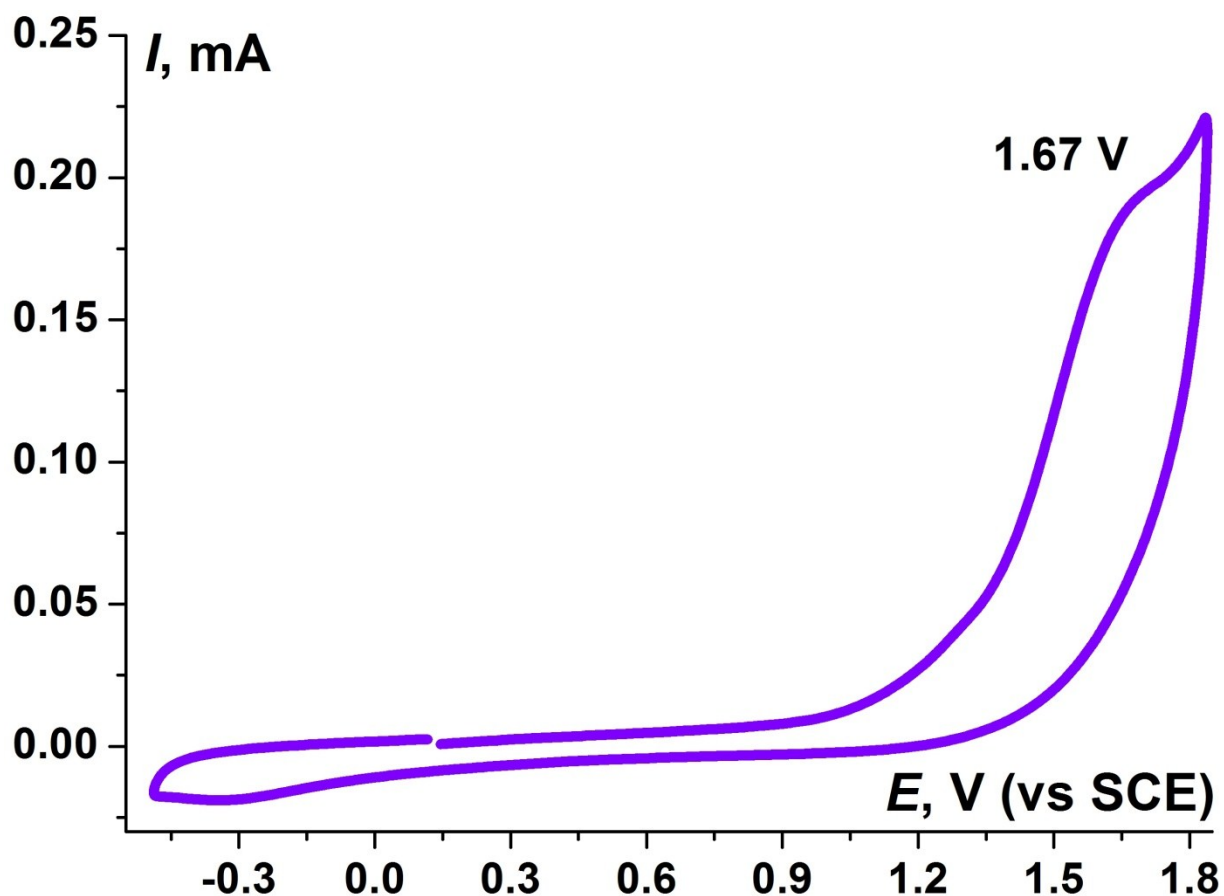


Figure 3a. The CV curve obtained for $3.0 \text{ mmol}\cdot\text{L}^{-1}$ solution of *N*-hydroxysuccinimide **2a** in $0.1 \text{ M Bu}_4\text{NClO}_4$ in H_2O -THF (1:1) on a working glassy-carbon electrode ($d = 2.9 \text{ mm}$) at a scan rate of $100 \text{ mV}\cdot\text{s}^{-1}$.

Table 3a. CV curve parameters in V vs SCE for *N*-hydroxysuccinimide **2a** solution in $0.1 \text{ M Bu}_4\text{NClO}_4$ in H_2O -THF (1:1) on a working glassy-carbon electrode ($d = 2.9 \text{ mm}$) under a scan rate of $100 \text{ mV}\cdot\text{s}^{-1}$

<i>Substance</i>	E_{onset}	<i>Scan direction</i>	$E_{\text{peak forward}}$	$E_{\text{peak reverse}}$	$E^{1/2}$
<i>N</i> -hydroxysuccinimide	1.13	+	1.67	-	1.47

Electrochemical reaction of sulfonyl hydrazides with *N*-hydroxy compounds

General procedure 1. Optimization of the reaction conditions for synthesis of 3aa from sulfonyl hydrazide 1a and *N*-hydroxysuccinimide 2a (Table 1): An undivided cell was equipped with a carbon plate anode (5 cm²) and a stainless steel plate cathode (5 cm²) and connected to a DC regulated power supply. The solution of *p*-toluenesulfonyl hydrazide **1a** (1 mmol, 186 mg), *N*-hydroxysuccinimide **2a** (1 mmol, 115 mg) and supporting electrolyte (0.2-3 mmol, 20-435 mg) in H₂O-THF (1:1, 30 mL), H₂O-MeCN (1:1), MeOH-THF (1:1) or MeOH was electrolyzed under constant current conditions (60 mA/cm² at 25-60 °C) under magnetic stirring. Electrodes were washed with EA (10 mL), after that reaction mixture was diluted with this EA and organic and water layers were separated. Water layer was extracted with EA (3×10 mL). Combined organic phase was washed with brine (2×8 mL), dried over Na₂SO₄ and concentrated under reduced pressure. The residue was recrystallized from EtOH to give pure desired **3aa** product.

Calculation of the time, which is necessary for passing of exact amount of electricity (representative example Table 1, entry 1).

It is necessary to pass 4 F/mol **1a** of electricity.

$$Q = N \cdot F \cdot n_r$$

Q — amount of passed electric current, C (Coulomb)

N — number of electrons passed in the cell per 1 molecule of sulfonyl hydrazide **1a**, F/mol

F — Faraday constant, $F = 96485 \text{ C} \cdot \text{mol}^{-1}$

n_r — amount of sulfonyl hydrazide **1a**, mol

$$Q = 4 \cdot 96485 \cdot 1 \cdot 10^{-3} = 386 \text{ C}$$

$$t = \frac{Q}{I}$$

t — time, sec

Q — amount of passed electric current, C (Coulomb)

I — electric current, A

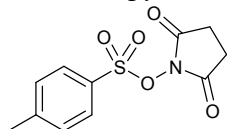
$$t = \frac{386}{0.3} = 1286 \text{ sec} = 21.5 \text{ min}$$

General procedure 2. Synthesis of products 3aa-3ma, 3ab-3af (Table 2). An undivided cell was equipped with a carbon plate anode (5 cm²) and a stainless steel plate cathode (5 cm²) and connected to a DC regulated power supply. The solution of sulfonyl hydrazide **1a-1m** (1 mmol), *N*-hydroxy compound **2a-2f** (1 mmol) and supporting electrolyte NH₄Br (3 mmol) in 30 ml of THF-H₂O (1:1) was electrolyzed using constant current conditions ($I = 300 \text{ mA}$, $j = 60 \text{ mA/cm}^2$, $\tau = 1 \text{ hour}$, $T = 40 \text{ }^\circ\text{C}$) under magnetic stirring. Electrodes were washed with EA (10 mL), after that reaction mixture was diluted with this EA and organic and water layers were separated. Water layer was extracted with EA (3×10 mL). Combined organic phase was washed with brine (2×8 mL), dried over Na₂SO₄ and concentrated under reduced pressure. The desired products **3aa-3ma**, **3ab-3af** were isolated by chromatography on SiO₂ with elution using PE-EA in a gradient of the latter from 10 to 75 vol %.

General procedure 3. Synthesis of products 3aa, 3da, 3ja (Scheme 2) from sodium sulfonates 4a, 4d, 4j and *N*-hydroxysuccinimide 2a (Scheme 2). An undivided cell was equipped with a carbon plate anode (5 cm²) and a stainless steel plate cathode (5 cm²) and connected to a DC regulated power supply. The solution of sodium sulfonate **4a, 4d, 4j** (1

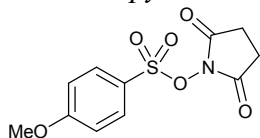
mmol), *N*-hydroxysuccinimide **2a** (1 mmol) and supporting electrolyte NH₄Br (3 mmol) in THF-H₂O (1:1, 30 mL) was electrolyzed using constant current conditions (*I* = 300 mA, *j* = 60 mA/cm², τ = 1 hour, *T* = 40 °C) under magnetic stirring. Electrodes were washed with EA (10 mL), after that reaction mixture was diluted with this EA and organic and water layers were separated. Water layer was washed with EA (3×10 mL). Combined organic phase was washed with brine (2×8 mL), dried over Na₂SO₄ and concentrated under reduced pressure. The yields of desired products **3aa**, **3da**, **3ja** were determined with NMR using 1, 4-dinitrobenzene as internal standard.

2,5-Dioxopyrrolidin-1-yl 4-methylbenzenesulfonate (3aa).^[4]



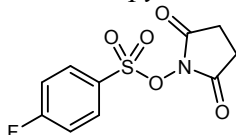
White solid, m.p. = 143-144 °C. Yield 95%. *R*_f = 0.49 (TLC, PE:EA, 1:1). ¹H NMR (CDCl₃), δ : 2.45 (s, 3H), 2.78 (s, 4H), 7.37 (d, *J* = 8.1 Hz, 2H), 7.89 (*J* = 8.1 Hz, 2H). ¹³C NMR (CDCl₃), δ : 21.8, 25.3, 129.3, 130.0, 131.0, 147.0, 168.6.

2,5-Dioxopyrrolidin-1-yl 4-methoxybenzenesulfonate (3ba).



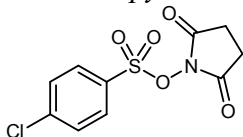
White solid, m.p. = 172-173 °C. Yield 67%. *R*_f = 0.25 (TLC, PE:EA, 1:1). ¹H NMR (DMSO-*d*₆), δ : 2.70 (s, 4H), 3.89 (s, 3H), 7.18 (d, *J* = 8.9 Hz, 2H), 7.94 (*J* = 8.9 Hz, 2H). ¹³C NMR (DMSO-*d*₆), δ : 25.4, 56.1, 115.1, 124.1, 131.7, 165.0, 169.7. HRMS (ESI) *m/z* (*M*+Na⁺) calculated for [C₁₁H₁₁NNaO₆S]⁺: 308.0199. Found: 308.0204.

2,5-Dioxopyrrolidin-1-yl 4-fluorobenzenesulfonate (3ca).



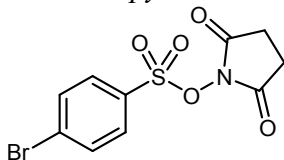
White solid, m.p. = 113-114 °C. Yield 73%. *R*_f = 0.55 (TLC, PE:EA, 1:1). ¹H NMR (DMSO-*d*₆), δ : 2.69 (s, 4H), 7.51-7.57 (m, 2H), 8.10-8.14 (m, 2H). ¹³C NMR (DMSO-*d*₆), δ : 25.4, 117.3 (d, ²*J* = 11.6 Hz), 129.5 (d, ⁴*J* = 1.1 Hz), 132.7 (d, ³*J* = 5.5 Hz), 166.3 (d, ¹*J* = 128.3 Hz), 169.7. HRMS (ESI) *m/z* (*M*+Na⁺) calculated for [C₁₀H₈FNNaO₅S]⁺: 295.9999. Found: 296.0005.

2,5-Dioxopyrrolidin-1-yl 4-chlorobenzenesulfonate (3da).



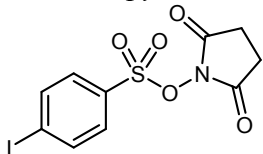
White solid, m.p. = 171-172 °C. Yield 75%. *R*_f = 0.56 (TLC, PE:EA, 1:1). ¹H NMR (DMSO-*d*₆), δ : 2.70 (s, 4H), 7.77 (d, *J* = 8.8 Hz, 2H), 8.04 (*J* = 8.8 Hz, 2H). ¹³C NMR (DMSO-*d*₆), δ : 25.5, 130.1, 131.0, 132.2, 141.2, 169.7. HRMS (ESI) *m/z* (*M*+Na⁺) calculated for [C₁₀H₈ClNNaO₅S]⁺: 311.9704. Found: 311.9713.

2,5-Dioxopyrrolidin-1-yl 4-bromobenzenesulfonate (3ea).



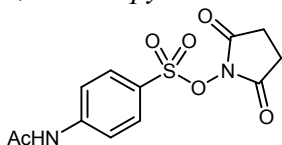
White solid, m.p. = 197-198 °C. Yield 65%. R_f = 0.56 (TLC, PE:EA, 1:1). ^1H NMR (DMSO- d_6), δ : 2.70 (s, 4H), 7.90-7.97 (m, 4H). ^{13}C NMR (DMSO- d_6), δ : 25.4, 130.5, 130.9, 132.6, 133.0, 169.6. HRMS (ESI) m/z ($M+\text{Na}^+$) calculated for $[\text{C}_{10}\text{H}_8\text{BrNNaO}_5\text{S}]^+$: 357.9178. Found: 357.9187.

2,5-Dioxopyrrolidin-1-yl 4-iodobenzenesulfonate (3fa).



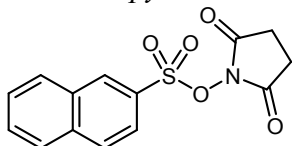
White solid, m.p. = 209-210 °C. Yield 19%. R_f = 0.56 (TLC, PE:EA, 1:1). ^1H NMR (DMSO- d_6), δ : 2.70 (s, 4H), 7.75 (d, J = 7.3 Hz, 2H), 8.09 (J = 7.3 Hz, 2H). ^{13}C NMR (DMSO- d_6), δ : 25.4, 105.6, 130.3, 132.9, 138.8, 169.6. HRMS (ESI) m/z ($M+\text{Na}^+$) calculated for $[\text{C}_{10}\text{H}_8\text{INNaO}_5\text{S}]^+$: 403.9060. Found: 403.9054.

2,5-Dioxopyrrolidin-1-yl 4-acetamidobenzenesulfonate (3ga).



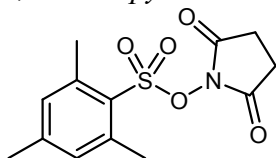
White solid, m.p. = 232-232.5 °C. Yield 45%. R_f = 0.24 (TLC, PE:EA, 1:5). ^1H NMR (DMSO- d_6), δ : 2.11 (s, 3H), 2.68 (s, 4H), 7.84 (d, J = 8.8 Hz, 2H), 7.92 (d, J = 8.8 Hz, 2H), 10.54 (s, 1H). ^{13}C NMR (DMSO- d_6), δ : 24.2, 25.4, 118.6, 125.7, 130.7, 145.8, 169.4, 169.6. HRMS (ESI) m/z ($M+\text{H}^+$) calculated for $[\text{C}_{12}\text{H}_{12}\text{N}_2\text{NaO}_6\text{S}]^+$: 335.0308. Found: 335.0308.

2,5-Dioxopyrrolidin-1-yl naphthalene-2-sulfonate (3ha).



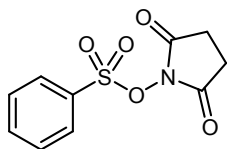
White solid, m.p. = 164-165 °C. Yield 87%. R_f = 0.46 (TLC, PE:EA, 1:1). ^1H NMR (DMSO- d_6), δ : 2.69 (s, 4H), 7.73 (dd, J = 8.1, 7.2 Hz, 1H), 7.81 (dd, J = 8.0, 7.2 Hz, 1H), 7.97 (d, J = 9.5 Hz, 1H), 8.12 (d, J = 8.0 Hz, 1H), 8.22 (d, J = 9.5 Hz, 1H), 8.25 (d, J = 8.1 Hz, 1H), 8.79 (s, 1H). ^{13}C NMR (DMSO- d_6), δ : 25.4, 122.9, 128.0, 128.1, 129.8, 129.9, 130.3, 130.4, 131.5, 131.6, 135.6, 169.7. HRMS (ESI) m/z ($M+\text{Na}^+$) calculated for $[\text{C}_{14}\text{H}_{11}\text{NNaO}_5\text{S}]^+$: 328.0250. Found: 328.0249.

2,5-Dioxopyrrolidin-1-yl 2,4,6-trimethylbenzenesulfonate (3ia).^[5]



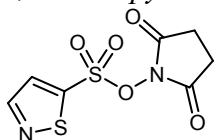
White solid, m.p. = 147-148 °C. Yield 17%. R_f = 0.56 (TLC, PE:EA, 1:1). ^1H NMR (CDCl_3), δ : 2.32 (s, 3H), 2.65 (s, 6H), 2.76 (s, 4H), 7.00 (s, 2H). ^{13}C NMR (CDCl_3), δ : 21.2, 22.9, 26.3, 129.9, 131.9, 141.1, 145.1, 168.6.

2,5-Dioxopyrrolidin-1-yl benzenesulfonate (3ja)^[6]



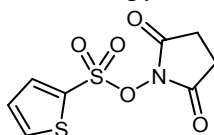
White solid, m.p. = 96-98 °C. Yield 62%. R_f = 0.59 (TLC, PE:EA, 1:1). ^1H NMR (CDCl_3), δ : 2.79 (s, 4H), 7.59 (t, J = 7.7 Hz, 2H), 7.74 (t, J = 7.7 Hz, 1H), 8.04 (d, J = 7.7 Hz, 2H). ^{13}C NMR (CDCl_3), δ : 25.4, 129.3, 129.4, 134.2, 135.5, 168.4.

2,5-Dioxopyrrolidin-1-yl isothiazole-5-sulfonate (3ka)



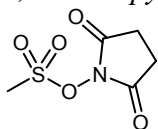
Yellow solid, m.p. = 83-85 °C. Yield 41%. R_f = 0.34 (TLC, PE:EA, 1:1). ^1H NMR (DMSO-d_6), δ : 2.74 (s, 4H), 8.36 (d, J = 2.0 Hz, 1H), 8.86 (d, J = 2.0 Hz, 1H). ^{13}C NMR (DMSO-d_6), δ : 25.5, 132.0, 155.5, 159.5, 169.5. HRMS (ESI) m/z ($\text{M}+\text{Na}^+$) calculated for $[\text{C}_7\text{H}_6\text{N}_2\text{NaO}_5\text{S}_2]^+$: 284.9610. Found: 284.9613.

2,5-Dioxopyrrolidin-1-yl thiophene-2-sulfonate (3la)



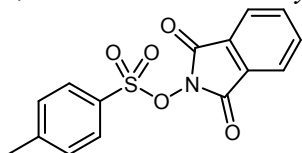
Yellow solid, m.p. = 109-111 °C. Yield 52%. R_f = 0.19 (TLC, PE:EA, 1:1). ^1H NMR (CDCl_3), δ : 2.80 (s, 4H), 7.19-7.22 (m, 1H), 7.86 (d, J = 4.4 Hz, 1H), 7.90 (d, J = 2.9 Hz, 1H). ^{13}C NMR (CDCl_3), δ : 25.4, 128.2, 132.7, 137.1, 137.6, 168.3. HRMS (ESI) m/z ($\text{M}+\text{Na}^+$) calculated for $[\text{C}_8\text{H}_7\text{NNaO}_5\text{S}_2]^+$: 283.9658. Found: 283.9654.

2,5-Dioxopyrrolidin-1-yl methanesulfonate (3ma).^[7]



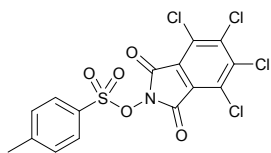
White solid, m.p. = 151-153 °C. Yield 67%. R_f = 0.27 (TLC, PE:EA, 1:5). ^1H NMR (DMSO-d_6), δ : 2.76 (s, 4H), 3.55 (s, 3H). ^{13}C NMR (DMSO-d_6), δ : 25.5, 29.5, 170.2.

1,3-Dioxoisindolin-2-yl 4-methylbenzenesulfonate (3ab).



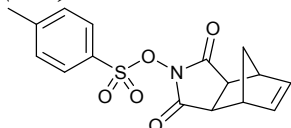
White solid, m.p. = 161-162 °C. Yield 53%. R_f = 0.73 (TLC, PE:EA, 1:1). ^1H NMR (CDCl_3), δ : 2.48 (s, 3H), 7.39 (d, J = 8.1 Hz, 2H), 7.78-7.86 (m, 4H), 7.93 (d, J = 8.1 Hz, 2H). ^{13}C NMR (CDCl_3), δ : 21.9, 124.2, 128.4, 129.5, 130.0, 130.7, 135.1, 147.0, 161.3. HRMS (ESI) m/z ($\text{M}+\text{Na}^+$) calculated for $[\text{C}_{15}\text{H}_{11}\text{NNaO}_5\text{S}]^+$: 340.0250. Found: 340.0246.

4,5,6,7-Tetrachloro-2-[(4-methylphenyl)sulfonyl]oxy-1H-isindole-1,3(2H)-dione (3ac)



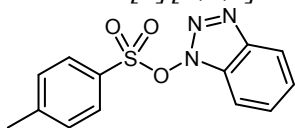
Yellow solid, m.p. = 203.5-205 °C. Yield 38%. R_f = 0.48 (TLC, PE:EA, 5:1). ^1H NMR (DMSO- d_6), δ : 2.46 (s, 3H), 7.52 (d, J = 8.1 Hz, 2H), 7.95 (d, J = 8.1 Hz, 2H). ^{13}C NMR (DMSO- d_6), δ : 21.3, 125.5, 128.8, 129.3, 129.5, 130.4, 139.2, 147.4, 157.5. HRMS (ESI) m/z ($M+\text{Na}^+$) calculated for $[\text{C}_{15}\text{H}_7\text{Cl}_4\text{NNaO}_5\text{S}]^+$: 475.8691. Found: 475.8688.

2-[(4-Methylphenyl)sulfonyl]oxy-3a,4,7,7a-tetrahydro-1H-4,7-methanoisoindole-1,3-dione (3ad)



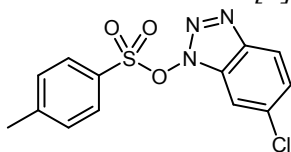
White solid, m.p. = 128.5-130.5 °C. Yield 55%. R_f = 0.32 (TLC, PE:EA, 2:1). ^1H NMR (CDCl_3), δ : 1.47 (d, J = 9.0 Hz, 1H), 1.74 (d, J = 9.0 Hz, 1H), 2.45 (s, 3H), 3.22-3.26 (m, 2H), 3.40-3.42 (m, 2H), 6.13-6.14 (m, 2H), 7.36 (d, J = 8.1 Hz, 2H), 7.86 (d, J = 8.1 Hz, 2H). ^{13}C NMR (CDCl_3), δ : 21.8, 42.8, 44.9, 51.1, 129.3, 129.9, 131.2, 134.8, 146.8, 169.5. HRMS (ESI) m/z ($M+\text{Na}^+$) calculated for $[\text{C}_{16}\text{H}_{15}\text{NNaO}_5\text{S}]^+$: 356.0563. Found: 356.0558.

1H-Benzo[d][1,2,3]triazol-1-yl 4-methylbenzenesulfonate (3ae).



White solid, m.p. = 108-110 °C. Yield 25%. R_f = 0.72 (TLC, PE:EA, 2:1). ^1H NMR (CDCl_3), δ : 2.42 (s, 3H), 7.33-7.40 (m, 3H), 7.49-7.57 (m, 2H), 7.72 (d, J = 8.3 Hz, 2H), 7.94 (d, J = 8.5 Hz, 1H). ^{13}C NMR (CDCl_3), δ : 21.7, 109.2, 120.0, 125.1, 128.5, 128.8, 129.1, 129.6, 130.3, 142.7, 147.9. HRMS (ESI) m/z ($M+\text{H}^+$) calculated for $[\text{C}_{13}\text{H}_{12}\text{N}_3\text{O}_3\text{S}]^+$: 290.0594. Found: 290.0599.

6-Chloro-1H-benzo[d][1,2,3]triazol-1-yl 4-methylbenzenesulfonate (3af).



White solid, m.p. = 159-161 °C. Yield 53%. R_f = 0.24 (TLC, PE:EA, 1:1). ^1H NMR (CDCl_3), δ : 2.49 (s, 3H), 7.37 (dd, J = 8.8, 1.8 Hz, 1H), 7.39 (d, J = 8.5 Hz, 2H), 7.55 (d, J = 1.8 Hz, 1H), 7.76 (d, J = 8.5 Hz, 2H), 7.90 (d, J = 8.8 Hz, 1H). ^{13}C NMR (CDCl_3), δ : 22.0, 109.3, 121.2, 126.6, 128.9, 129.3, 129.8, 130.6, 136.0, 141.4, 148.3. HRMS (ESI) m/z ($M+\text{Na}^+$) calculated for $[\text{C}_{13}\text{H}_{10}\text{ClN}_3\text{NaO}_3\text{S}]^+$: 346.0024. Found: 346.0017.

Control experiment (Scheme 4)

1. Synthesis of *p*-toluenesulfonyl bromide A. *p*-Toluenesulfonyl hydrazide **1a** (2 mmol) was dissolved in CHCl_3 (10 ml) and cooled to 0°C. At this temperature molecular bromine (5 mL) dissolved in CHCl_3 (5mL) was added while vigorous stirring until reaction mixture stops to discolor. Then 3 ml of 1M Na_2SO_3 was added to neutralize remaining bromine. The reaction mixture was washed with water (3×5 mL), dried over Na_2SO_4 and concentrated under reduced

pressure. Target *p*-toluenesulfonyl bromide was isolated by chromatography on SiO₂ with elution using PE-EA in a linear gradient of the latter from 0 to 15 vol %. Yield 55%, m.p. = 95-97 °C. *R*_f = 0.78 (TLC, PE:EA, 10:1). ¹H NMR (CDCl₃), δ: 2.48 (s, 3H), 7.38 (d, *J* = 8.2 Hz, 2H), 7.87 (d, *J* = 8.2 Hz, 2H). ¹³C NMR (CDCl₃), δ: 21.8, 126.5, 130.1, 144.6, 146.7. HRMS (ESI) *m/z* (*M*+*H*⁺) calculated for [C₇H₈BrO₂S]⁺: 234.9428. Found: 234.9423.

2. Synthesis of potassium salt of *N*-hydroxysuccinimide C'. Potassium hydroxide (3.5 mmol) was added to the solution of *N*-hydroxysuccinimide **2a** (3.5 mmol) in EtOH (10 mL). Reaction mixture was refluxed for 5 minutes, crystals of target salt precipitated from the solution. After that the reaction mixture was filtered under reduced pressure, precipitate was washed with EtOH (3×5 mL) and dried. Yield 78%. ¹H NMR (D₂O), δ: 2.65 (s, 4H). ¹³C NMR (D₂O), δ: 26.2, 179.9.

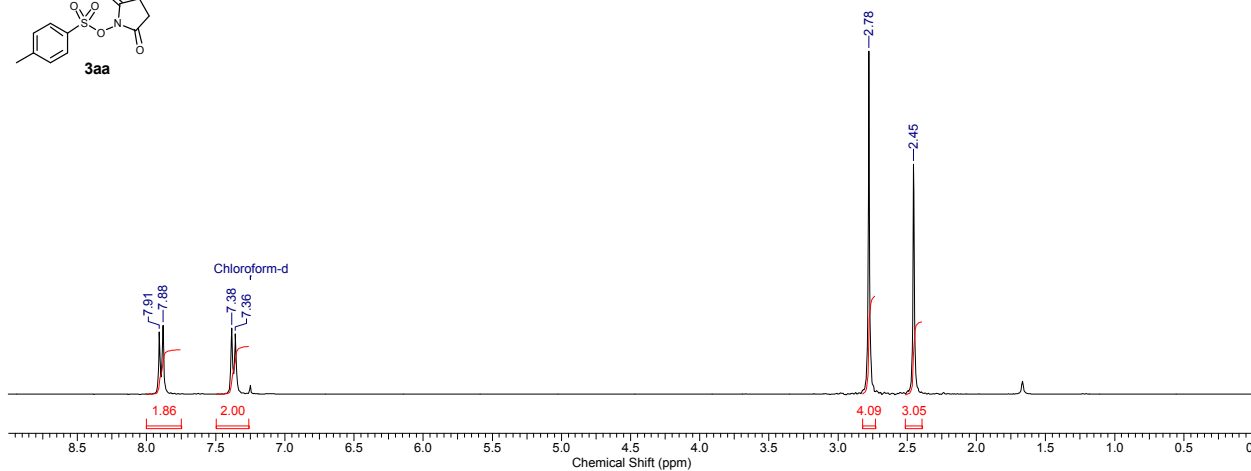
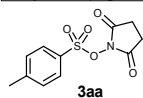
3. Synthesis of coupling product 3aa from *p*-toluenesulfonyl bromide A and potassium salt of *N*-hydroxysuccinimide C'. The solution of potassium salt of *N*-hydroxysuccinimide C' (1 mmol) in H₂O (1 mL) was added to the solution of *p*-toluenesulfonyl bromide **A** (1 mmol) in THF (5 mL). The reaction mixture was vigorously stirred at 40 °C for an hour. After that it was washed with EA (3×5mL). Combined organic layers were dried over Na₂SO₄, solvent was removed under reduced pressure. The yield of desired product **3aa** was determined by NMR using 1, 4-dinitrobenzene as internal standard.

References

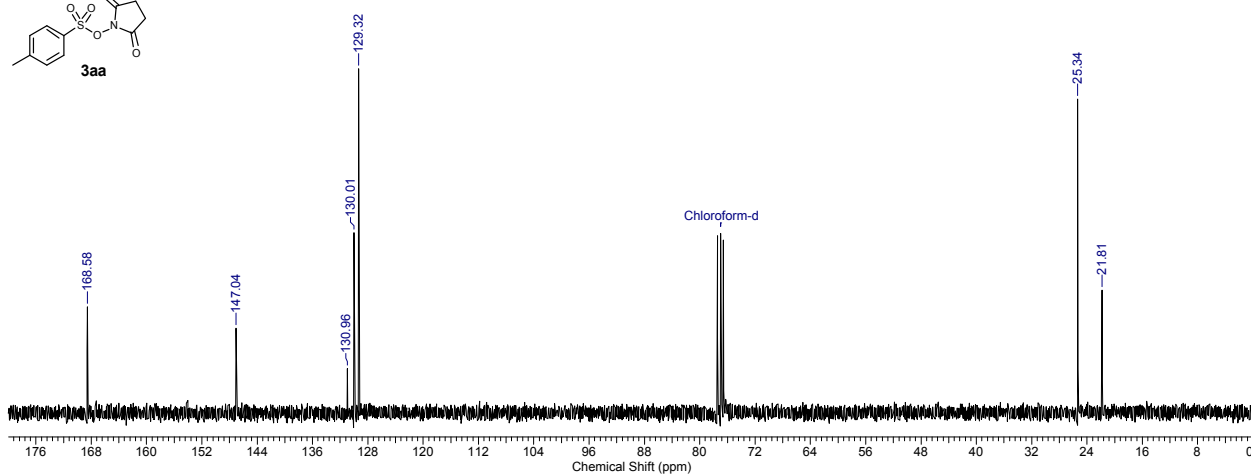
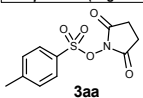
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NMR spectra of synthesized compounds

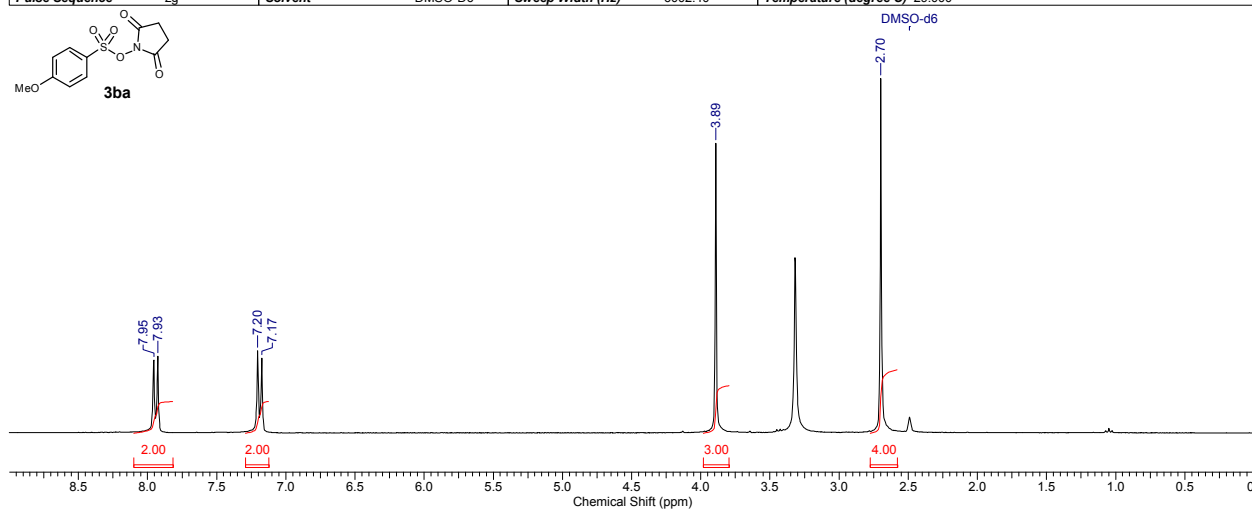
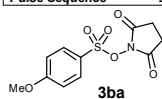
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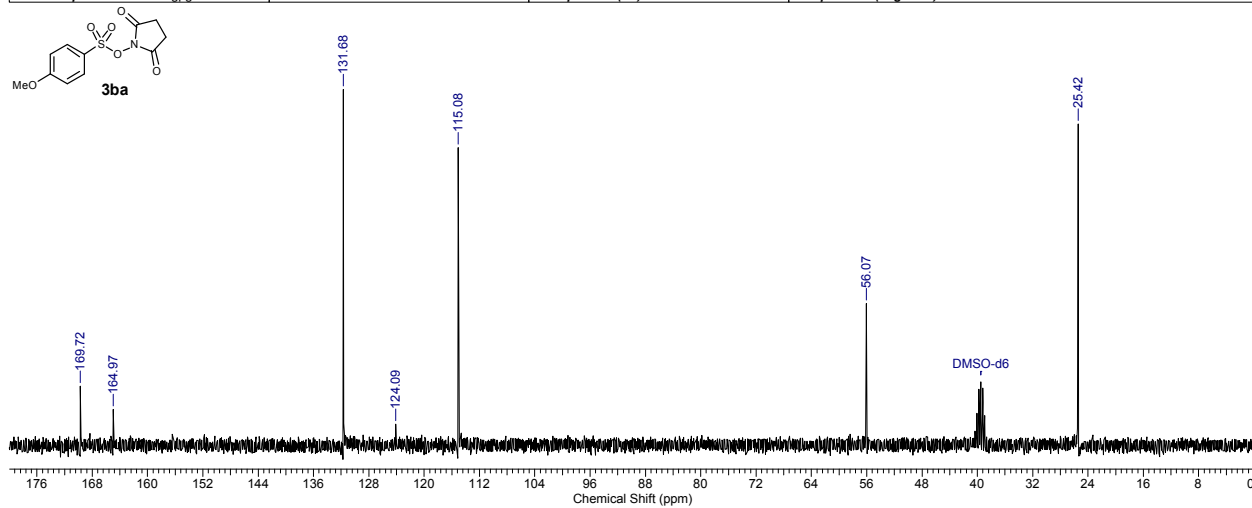
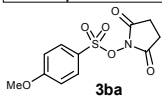
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Temperature (degree C)	26.200				



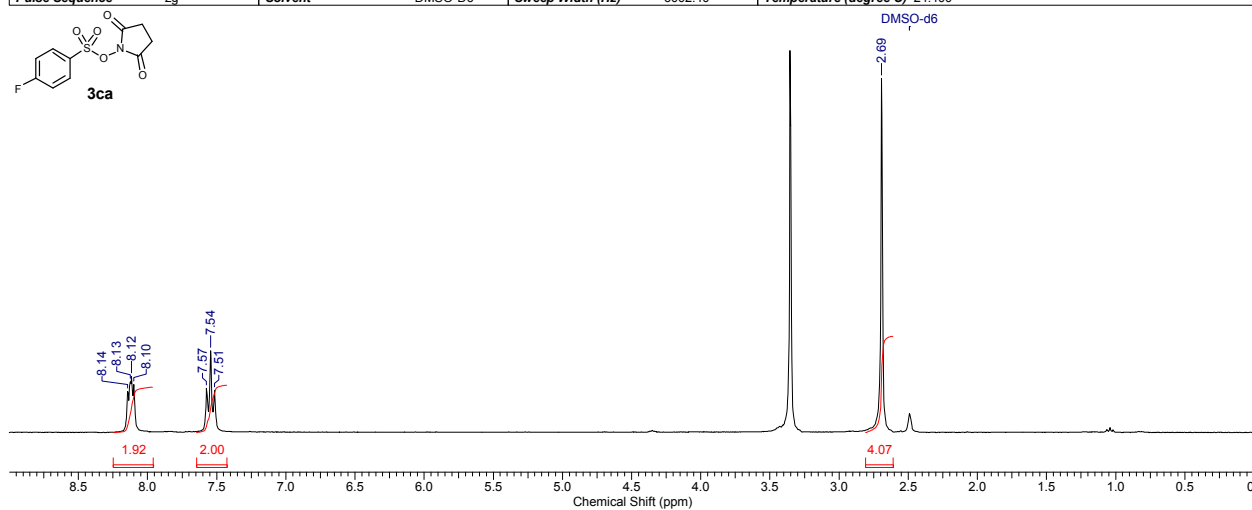
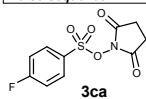
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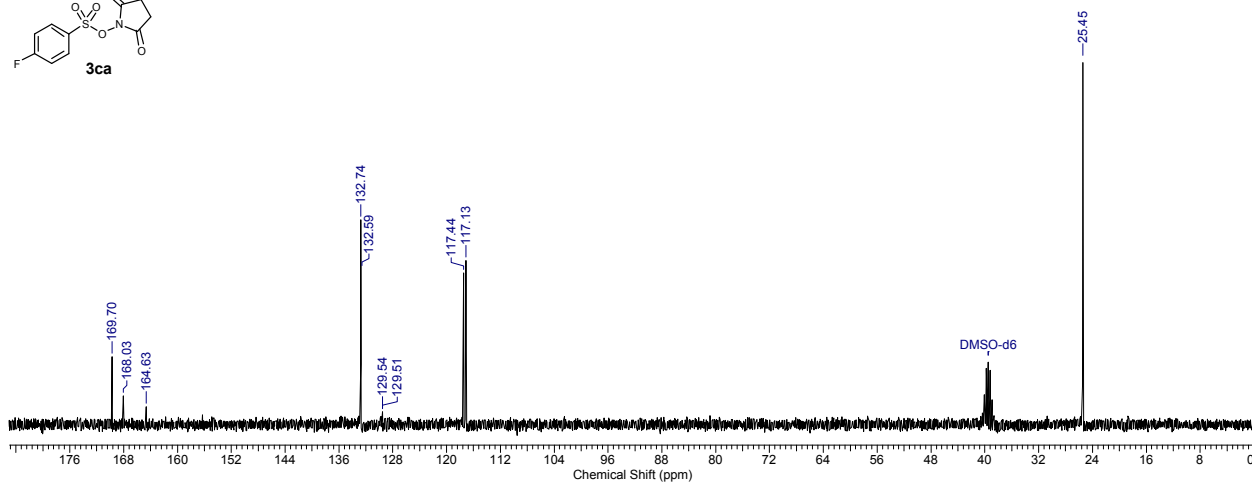
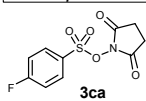
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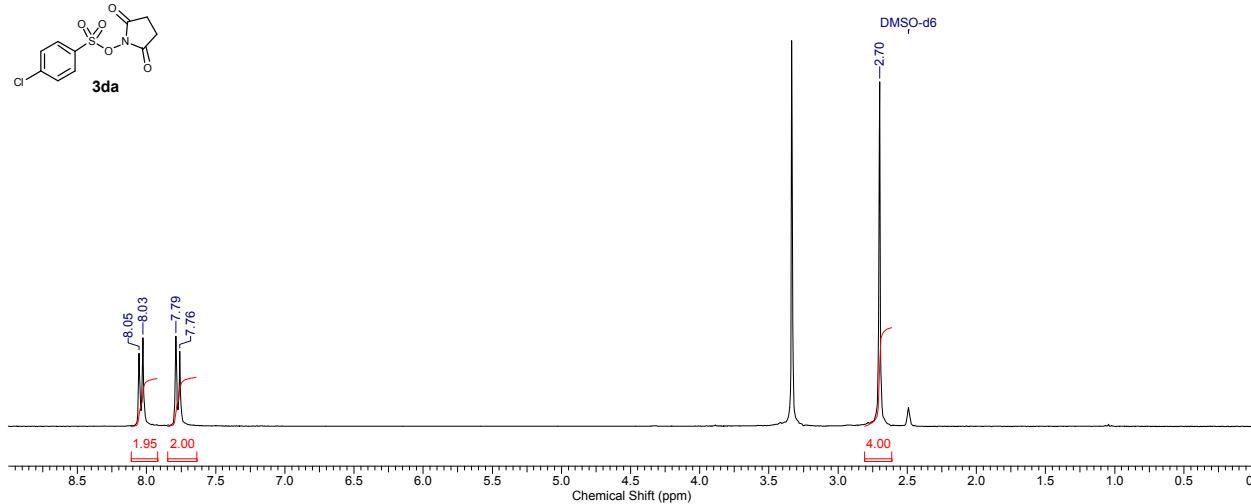
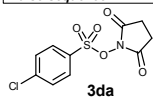
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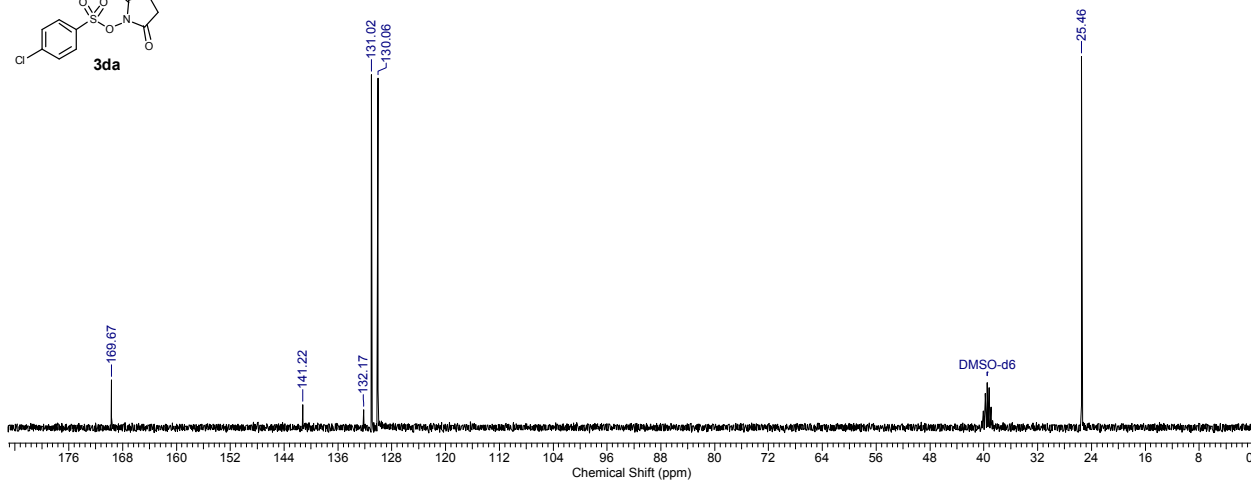
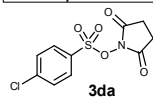
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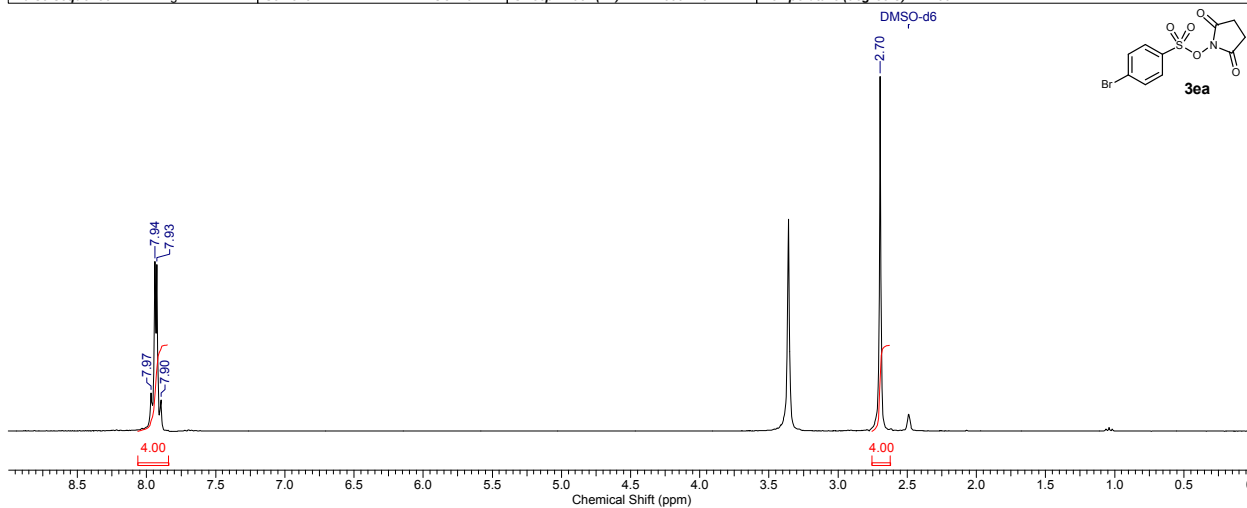
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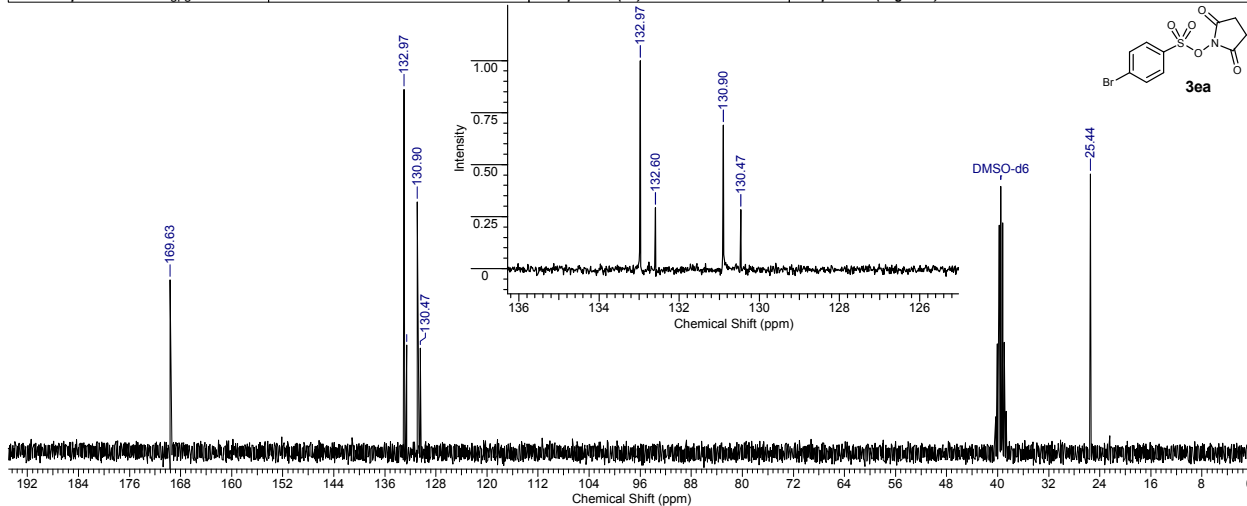
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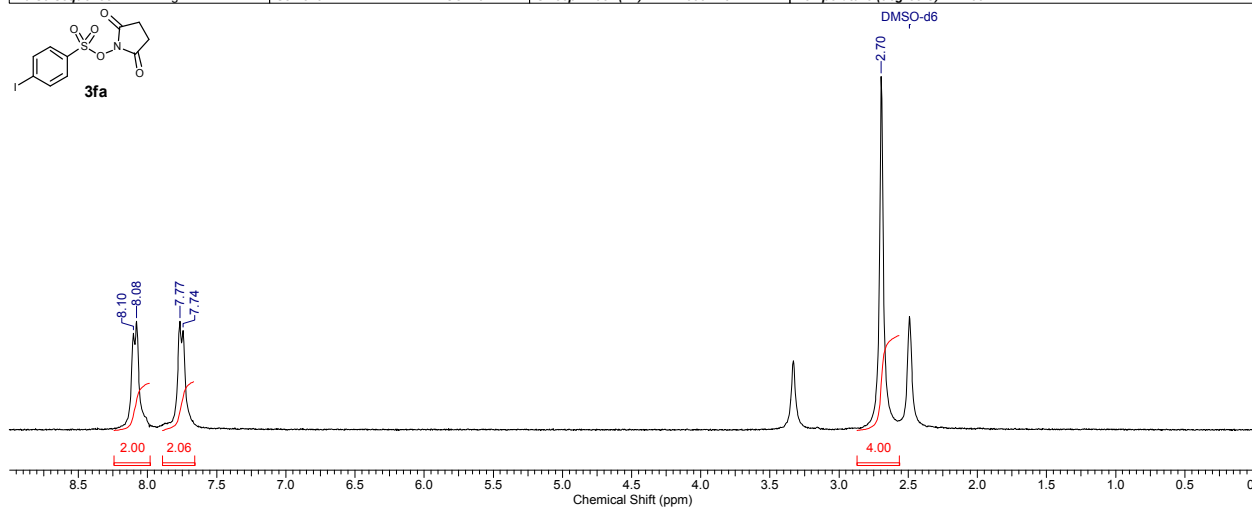
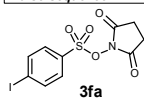
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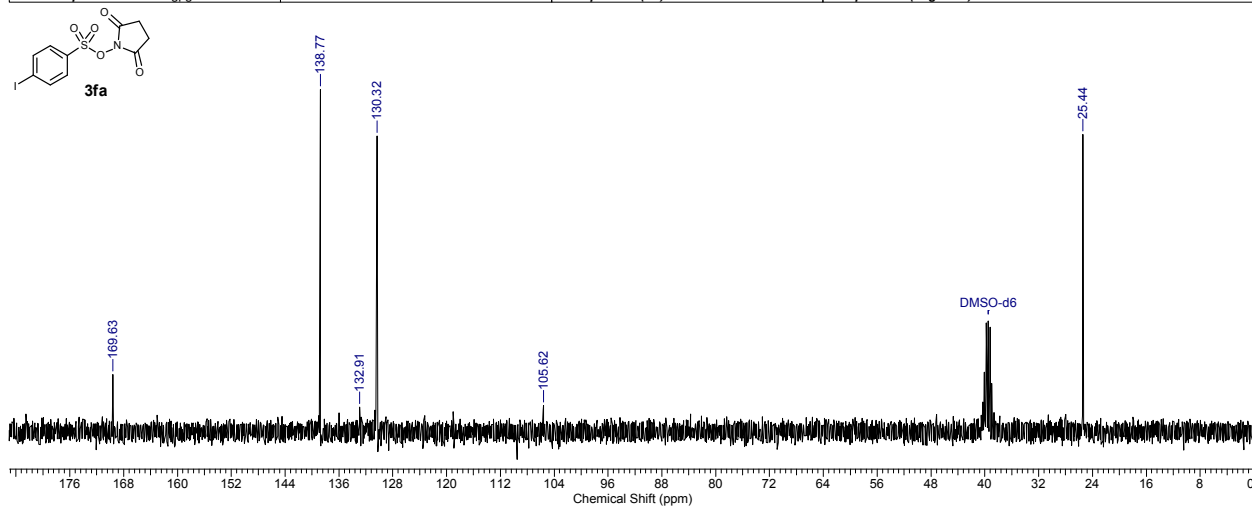
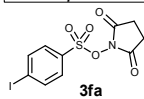
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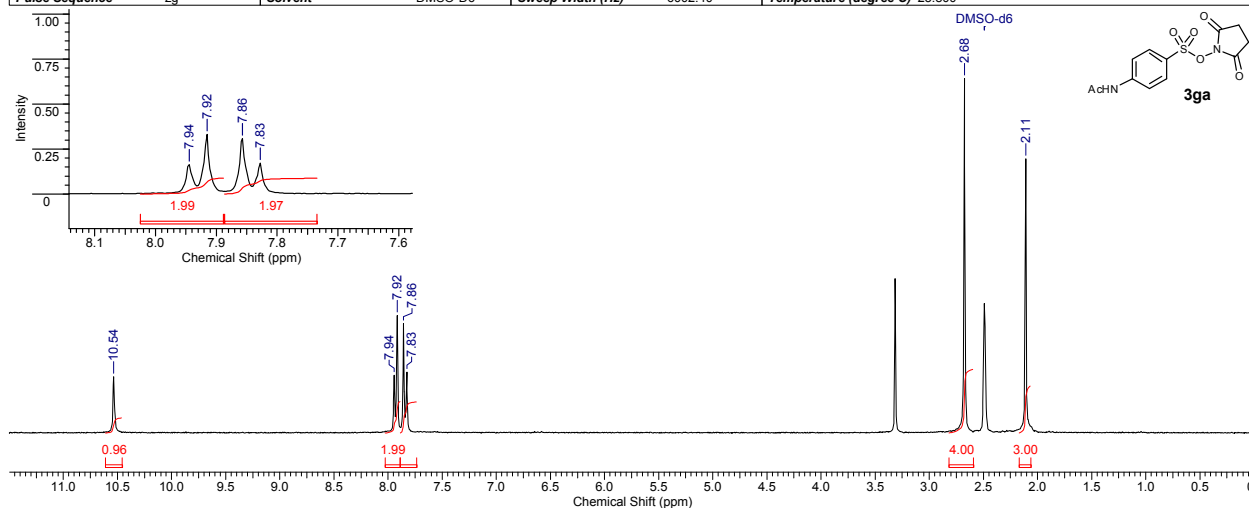
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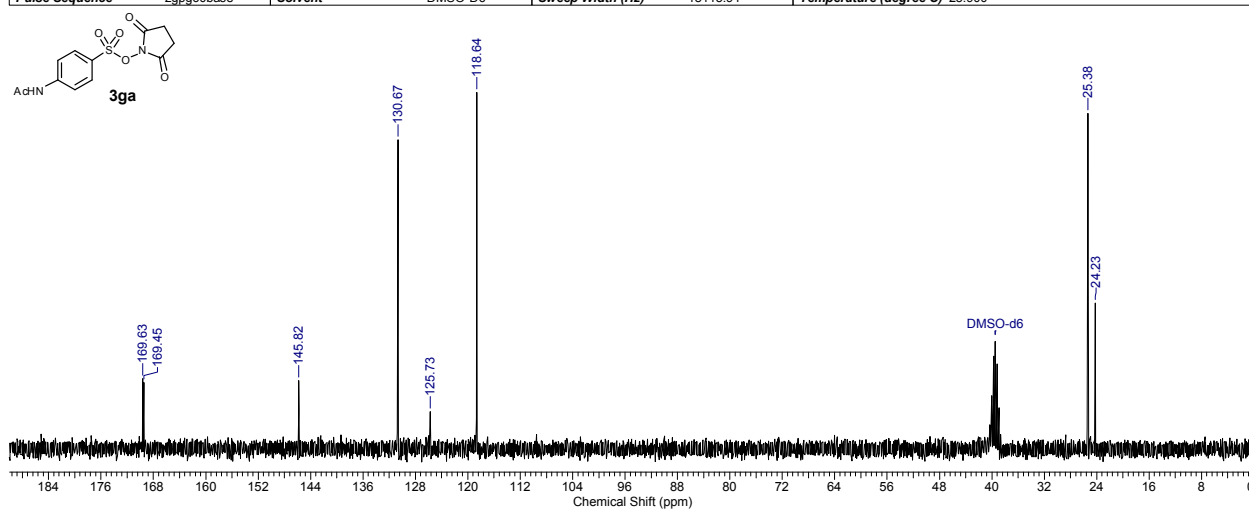
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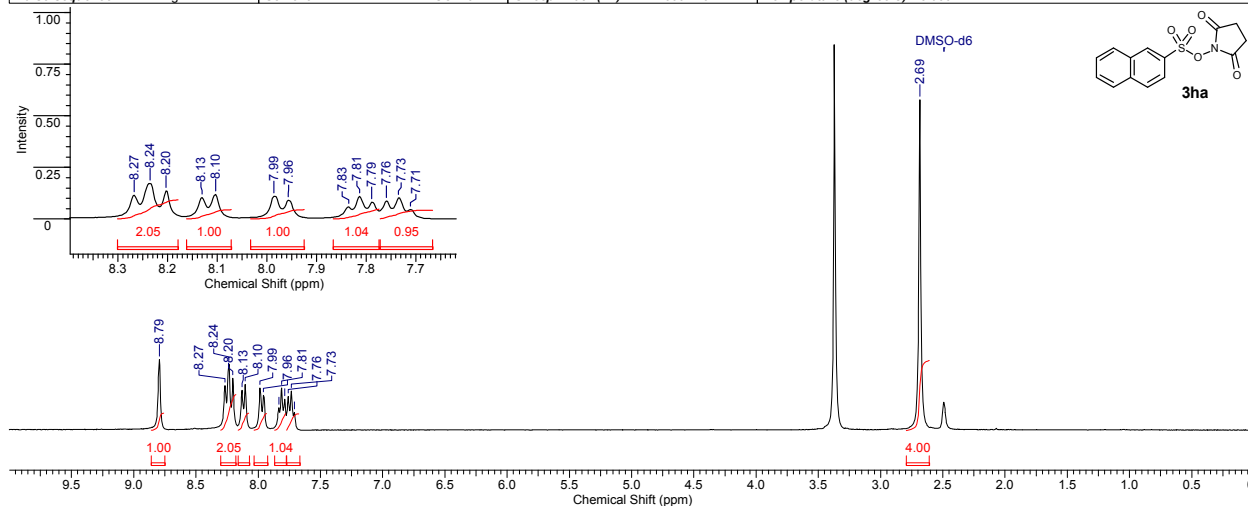
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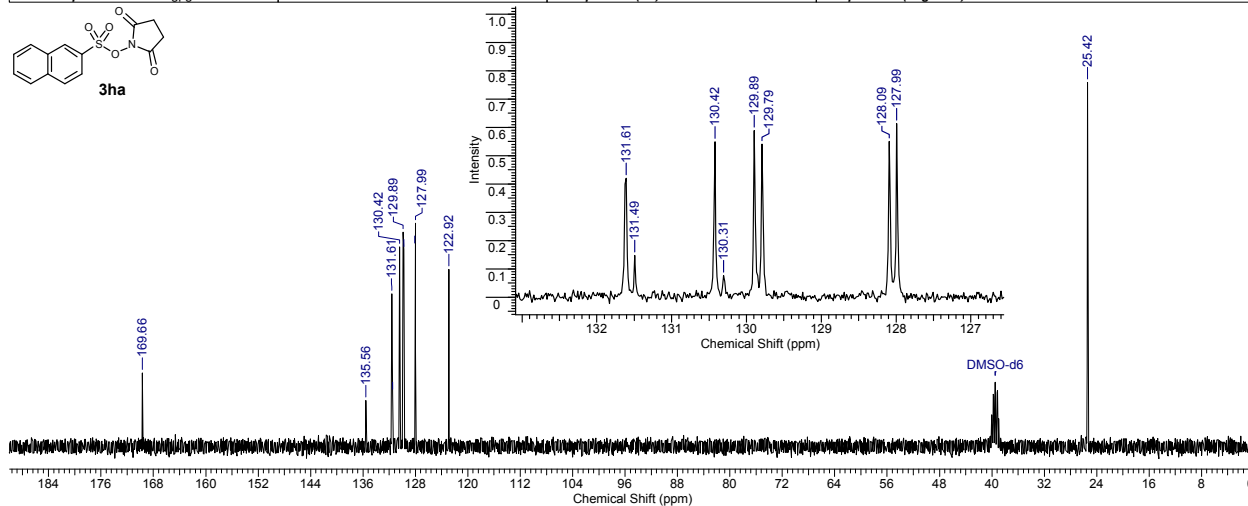
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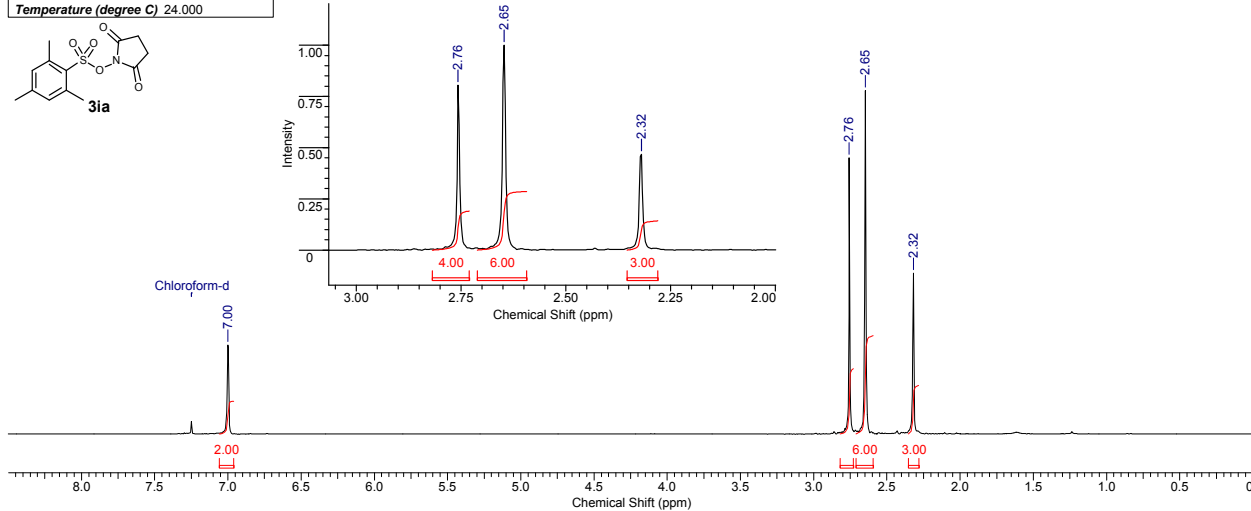
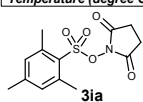
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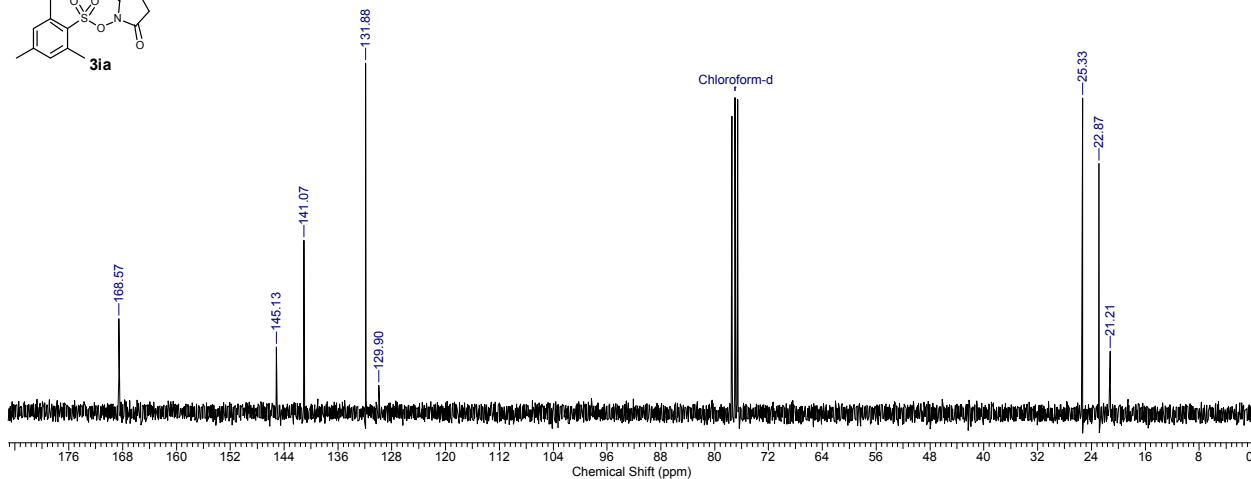
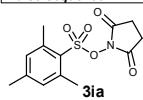
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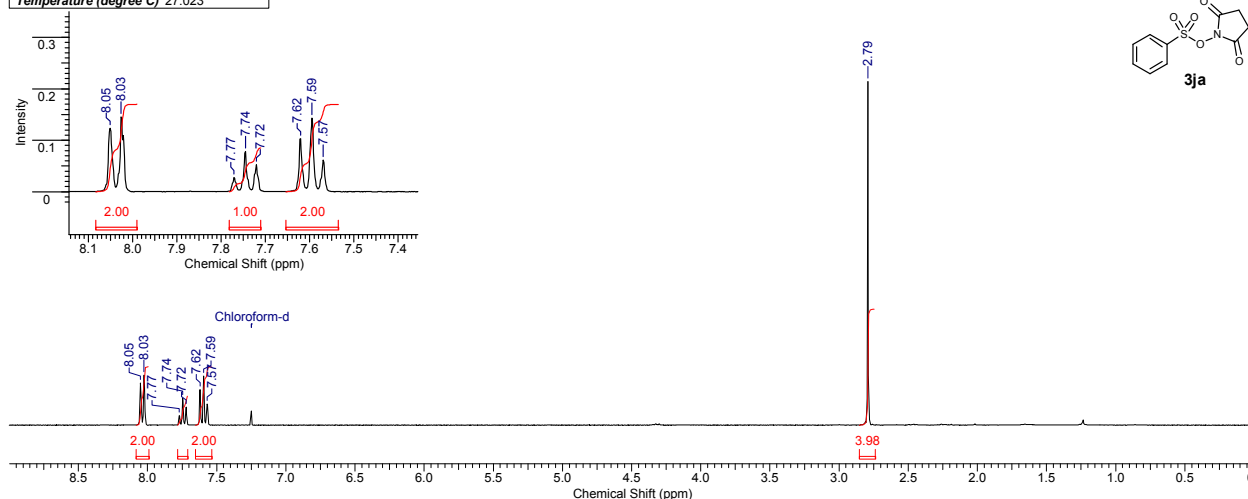
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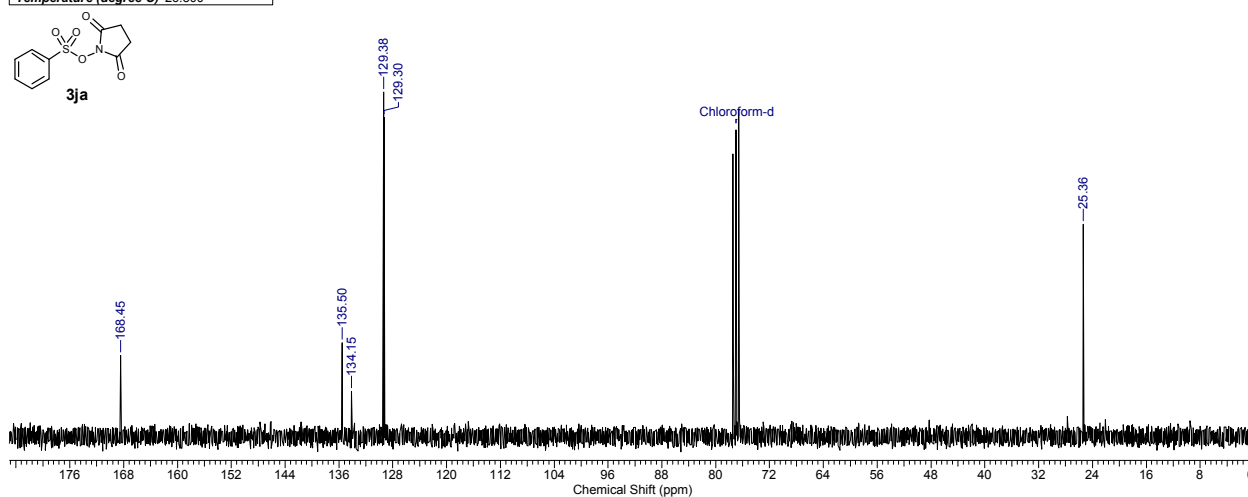
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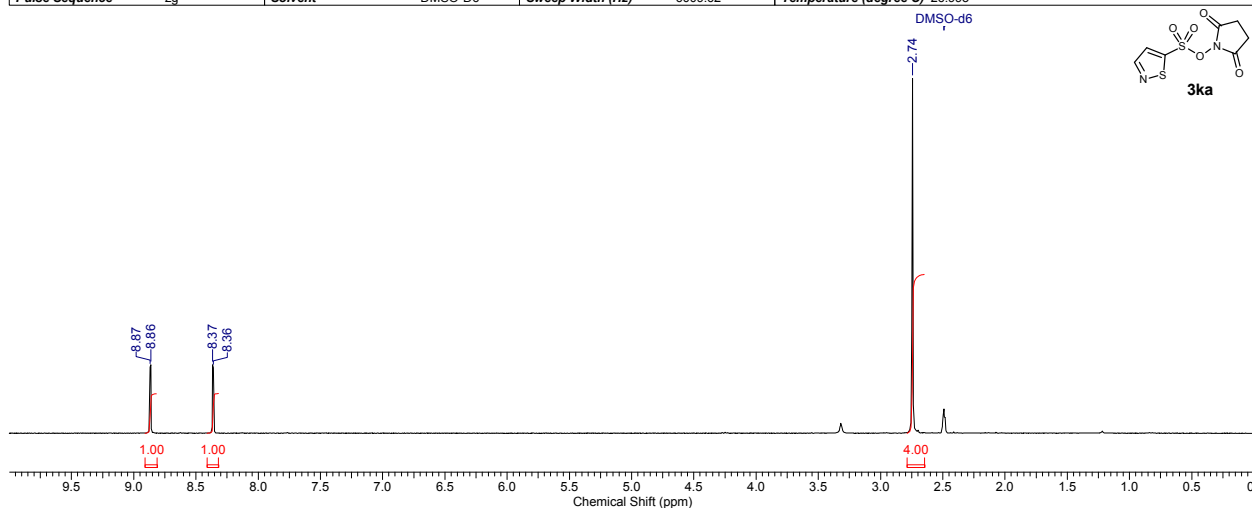
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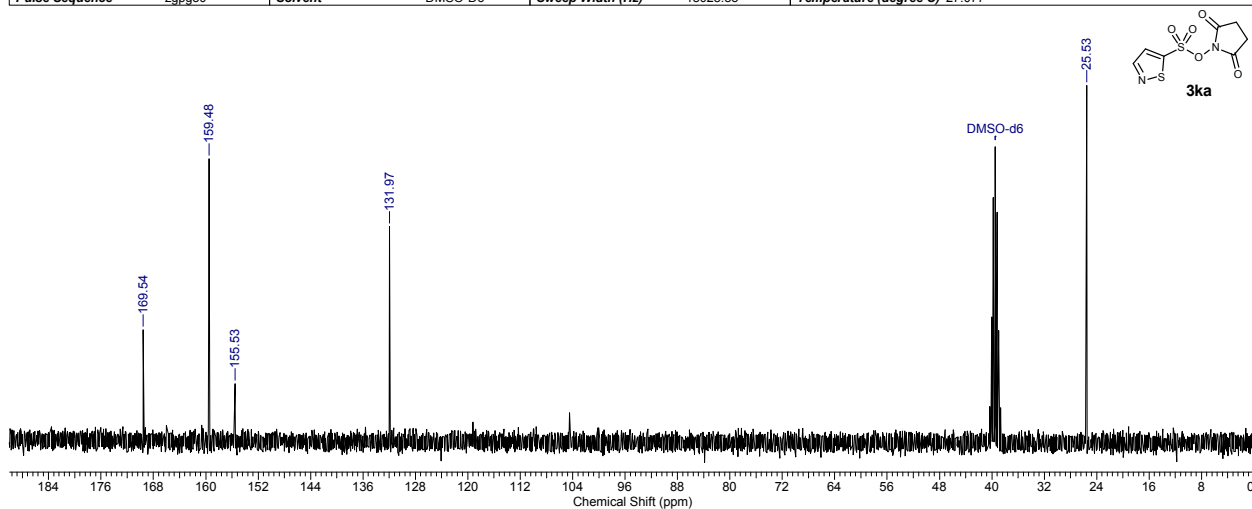
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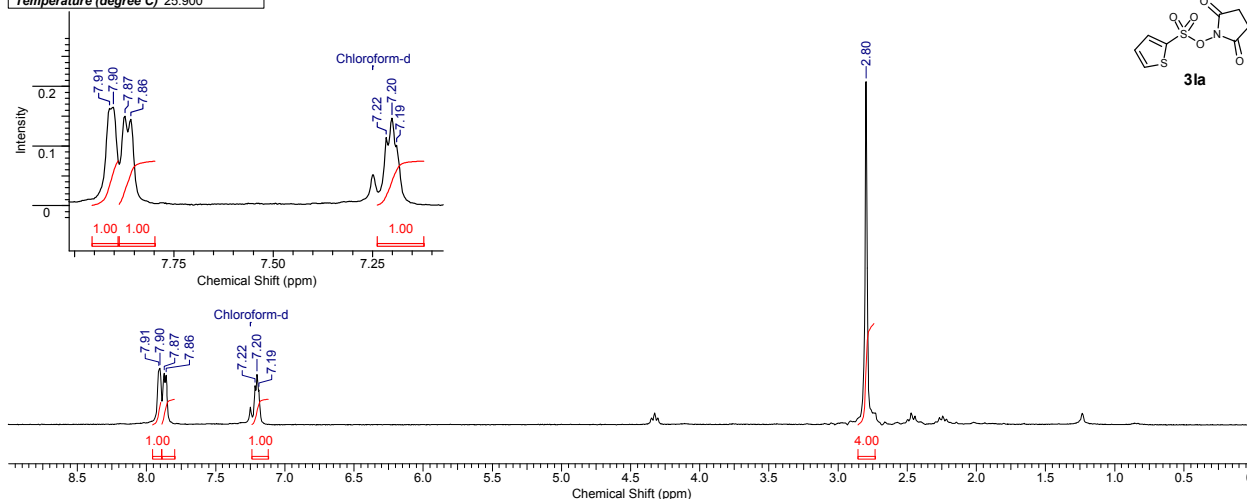
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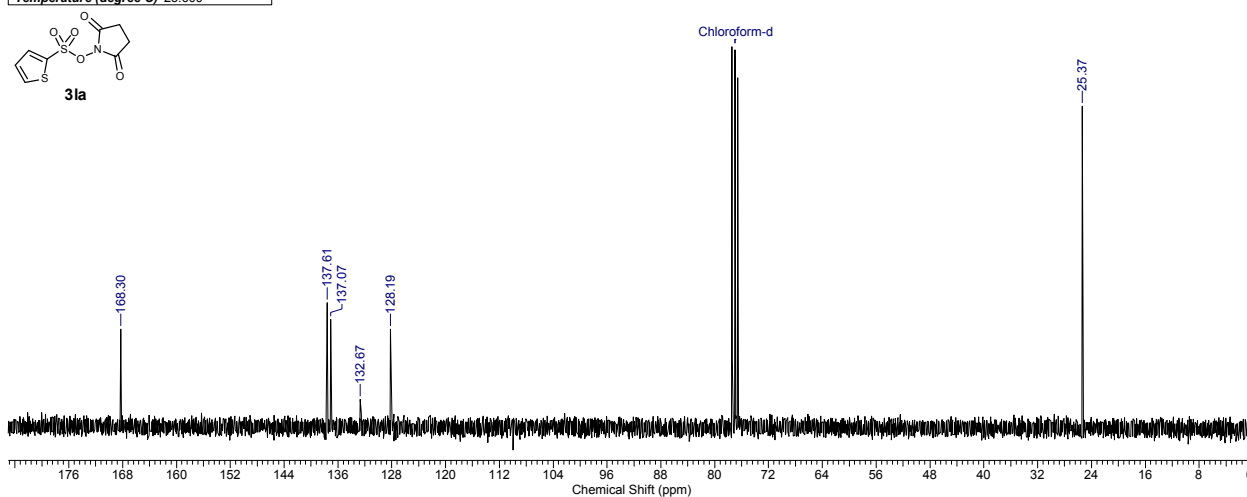
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				Points Count	16384
				Temperature (degree C)	27.077



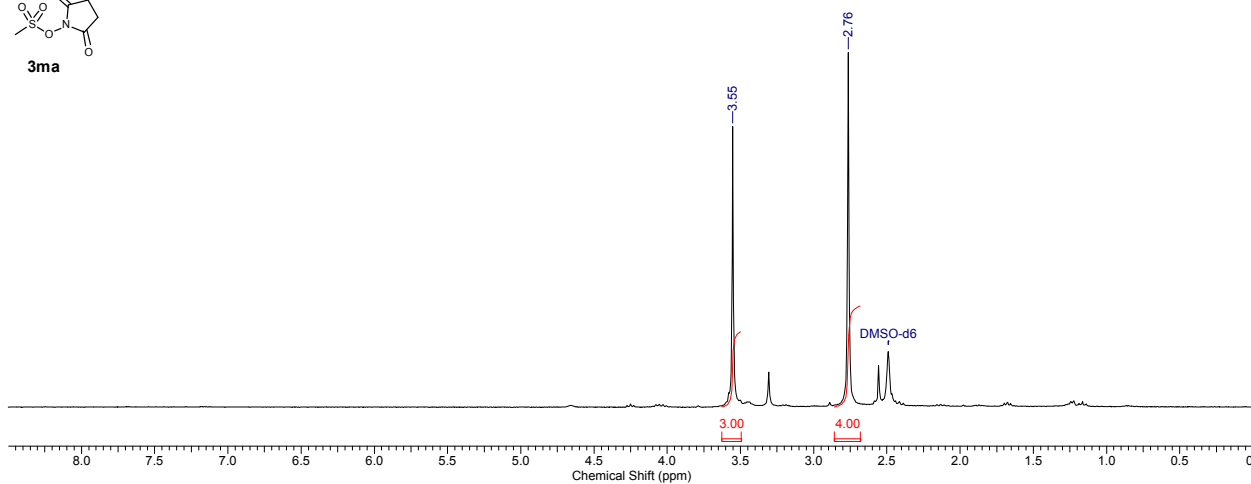
Acquisition Time (sec)	1.3518	Date	14 Nov 2018 09:55:12	Frequency (MHz)	300.13
File Name		Number of Transients	1	Original Points Count	8124
Nucleus	¹ H			Points Count	8192
Pulse Sequence	zg	Solvent	CHLOROFORM-D	Sweep Width (Hz)	6009.62
Temperature (degree C)	25.900				



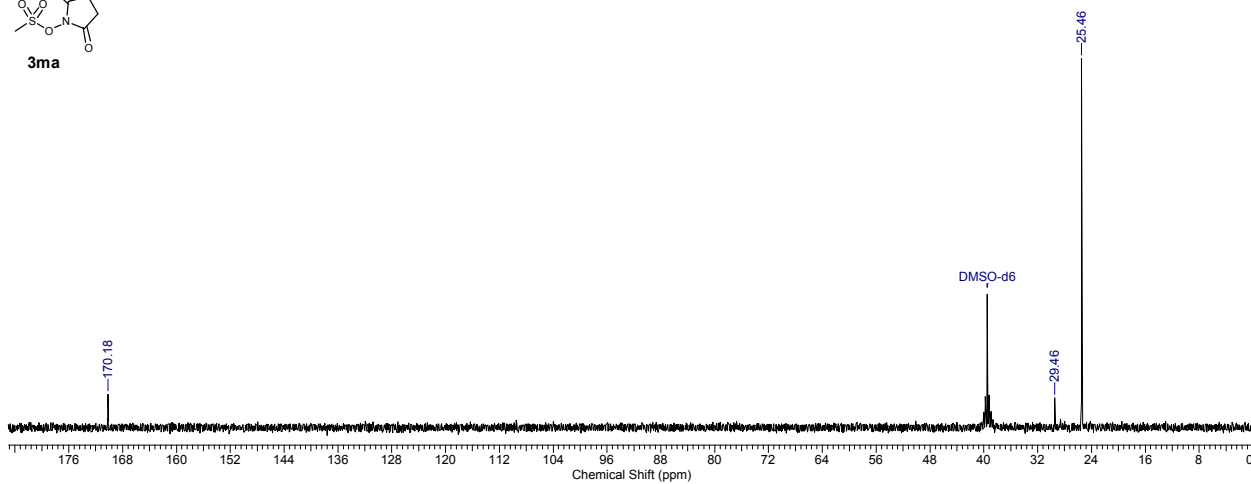
Acquisition Time (sec)	0.9006	Date	14 Nov 2018 11:35:28	Frequency (MHz)	75.48
File Name		Number of Transients	269	Original Points Count	16316
Nucleus	¹³ C			Points Count	16384
Pulse Sequence	zgpg30	Solvent	CHLOROFORM-D	Sweep Width (Hz)	18115.94
Temperature (degree C)	25.600				



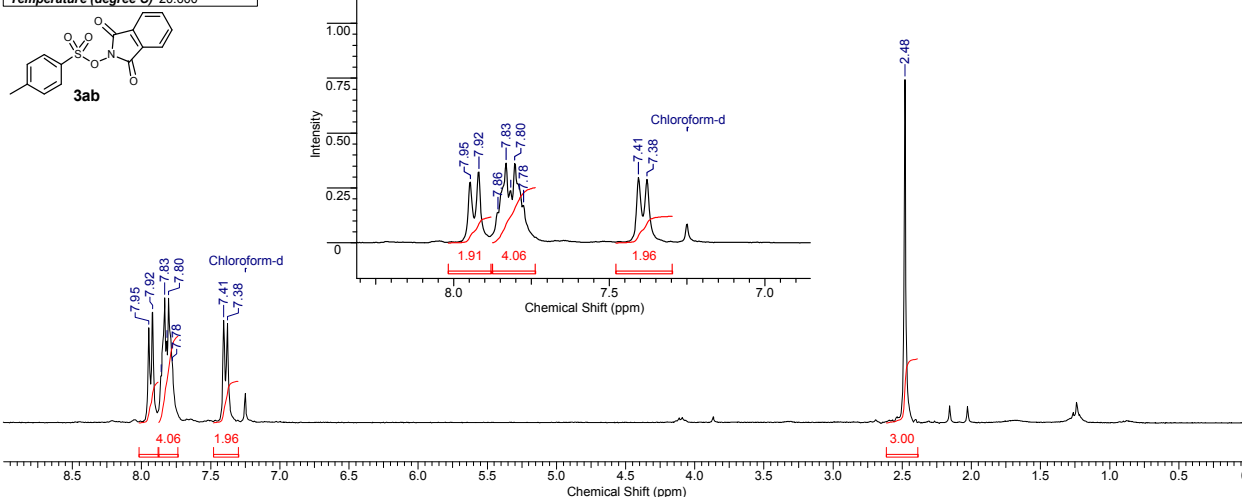
Acquisition Time (sec)		1.3521		Date		20 Sep 2017 15:04:32					
File Name								Frequency (MHz)		300.13	
Nucleus		1H		Number of Transients		1		Original Points Count		8116	
Pulse Sequence		zg		Solvent		DMSO-D6		Sweep Width (Hz)		6002.40	
								Points Count		8192	
								Temperature (degree C)		26.200	



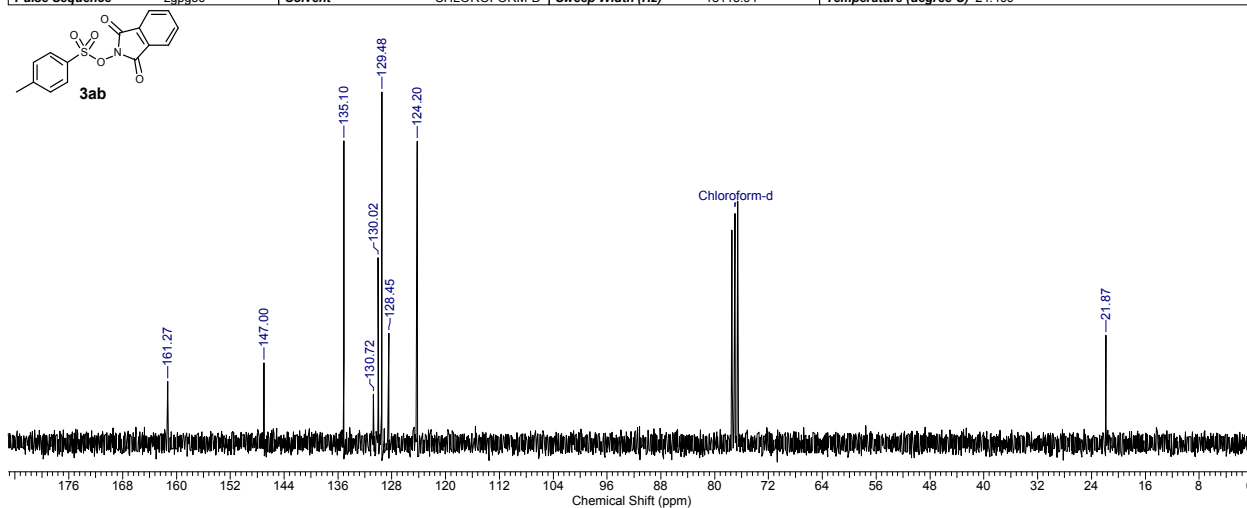
Acquisition Time (sec)	0.9002	Date	21 Sep 2017 10:48:32				
File Name						Frequency (MHz)	75.48
Nucleus	¹³ C	Number of Transients	202	Original Points Count	16308	Points Count	16384
Pulse Sequence	zgpg60base	Solvent	DMSO-D6	Sweep Width (Hz)	18115.94	Temperature (degree C)	26.000



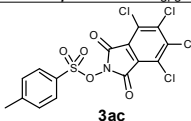
Acquisition Time (sec)	1.3518	Date	21 Apr 2016 15:15:12		Frequency (MHz)	300.13
File Name					Points Count	8192
Nucleus	¹ H	Number of Transients	1	Original Points Count	8124	
Pulse Sequence	zg	Solvent	CHLOROFORM-D		Sweep Width (Hz)	6009.62
Temperature (degree C)	20.600					



Acquisition Time (sec)	0.9006	Date	20 Mar 2017 15:06:40		Frequency (MHz)	75.48
File Name					Points Count	16384
Nucleus	¹³ C	Number of Transients	87	Original Points Count	16316	
Pulse Sequence	zgpg30	Solvent	CHLOROFORM-D		Sweep Width (Hz)	18115.94
					Temperature (degree C)	24.400

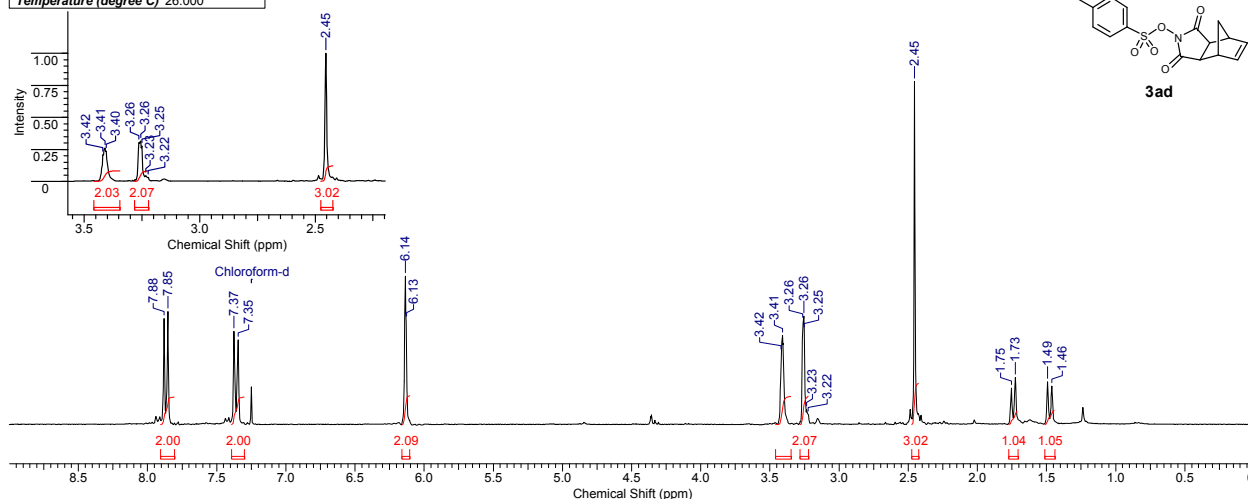


3ac

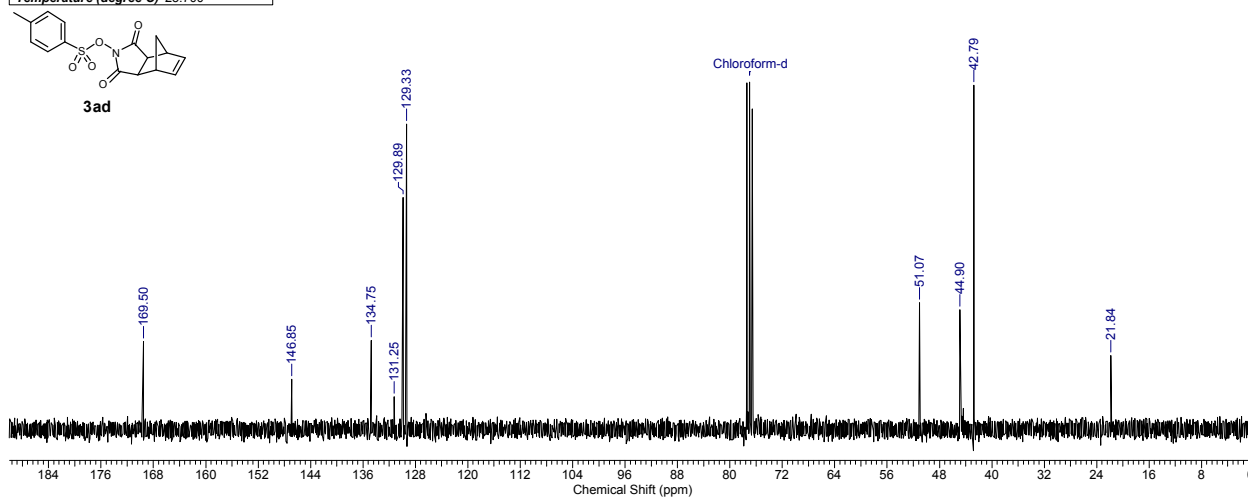


3ac

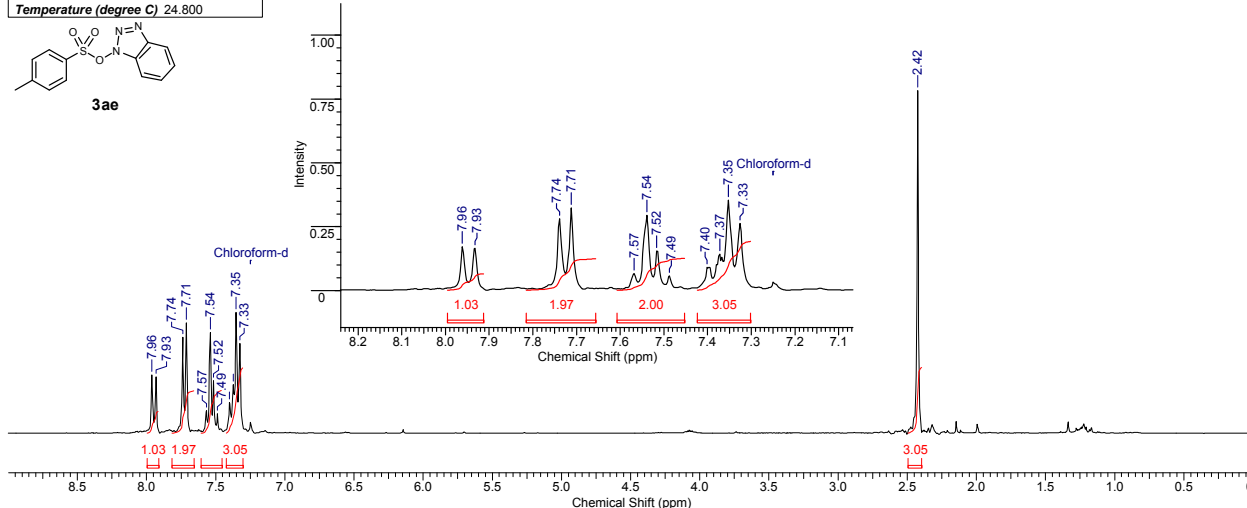
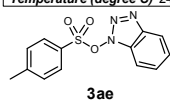
Acquisition Time (sec)	1.3518	Date	20 Nov 2018 10:44:16				
File Name					Frequency (MHz)	300.13	
Nucleus	1H	Number of Transients	1	Original Points Count	8124	Points Count	8192
Pulse Sequence	zg	Solvent	CHLOROFORM-D			Sweep Width (Hz)	6009.62
Temperature (degree C)	26.000						



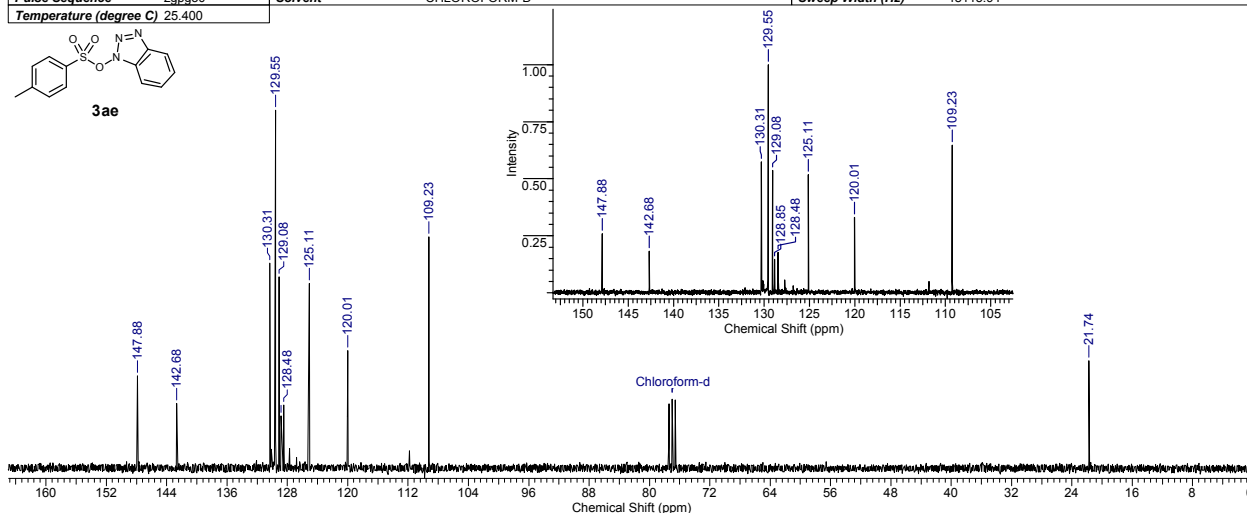
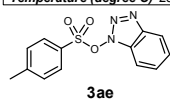
Acquisition Time (sec) 0.9006		Date 20 Nov 2018 14:15:28			
File Name				Frequency (MHz)	75.48
Nucleus	13C	Number of Transients	148	Original Points Count	16316
				Points Count	16384
Pulse Sequence	zgpg30	Solvent	CHLOROFORM-D		Sweep Width (Hz)
Temperature (degree C) 25.700					



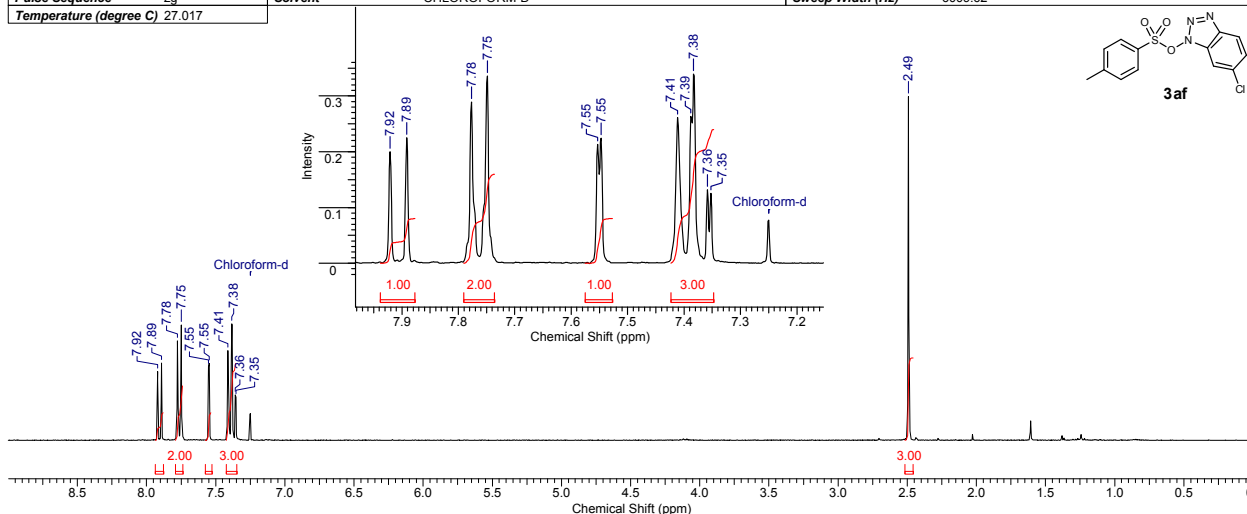
Acquisition Time (sec)	1.3518	Date	01 Feb 2017 16:04:16	Frequency (MHz)	300.13
File Name		Number of Transients	1	Original Points Count	8124
Nucleus	¹ H			Points Count	8192
Pulse Sequence	zg	Solvent	CHLOROFORM-D	Sweep Width (Hz)	6009.62
Temperature (degree C)	24.800				



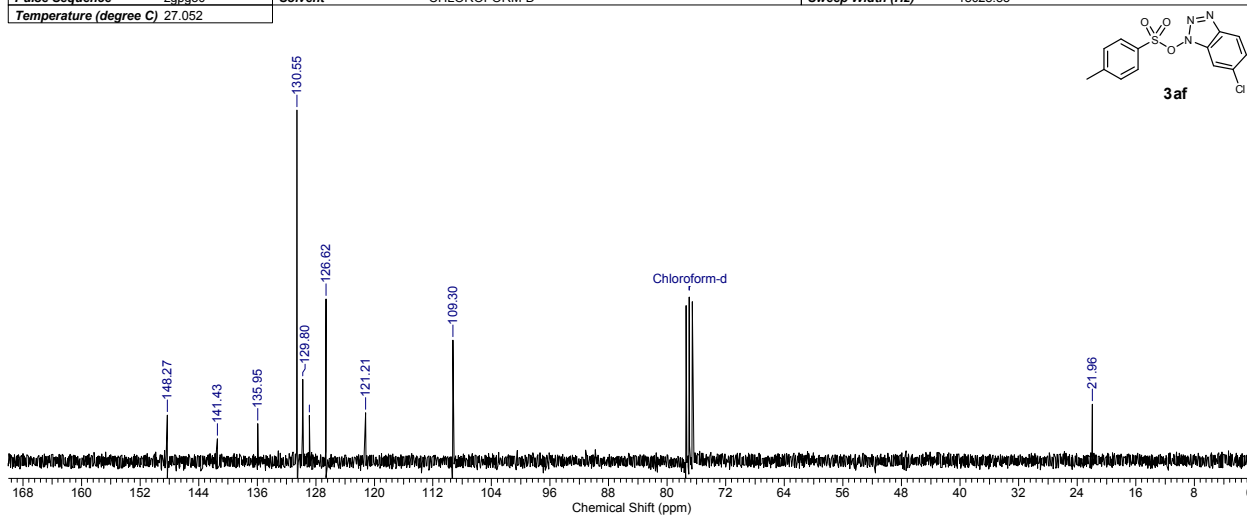
Acquisition Time (sec)	0.9006	Date	02 Feb 2017 11:58:56	Frequency (MHz)	75.48
File Name		Number of Transients	97	Original Points Count	16316
Nucleus	¹³ C			Points Count	16384
Pulse Sequence	zgpg30	Solvent	CHLOROFORM-D	Sweep Width (Hz)	18115.94
Temperature (degree C)	25.400				



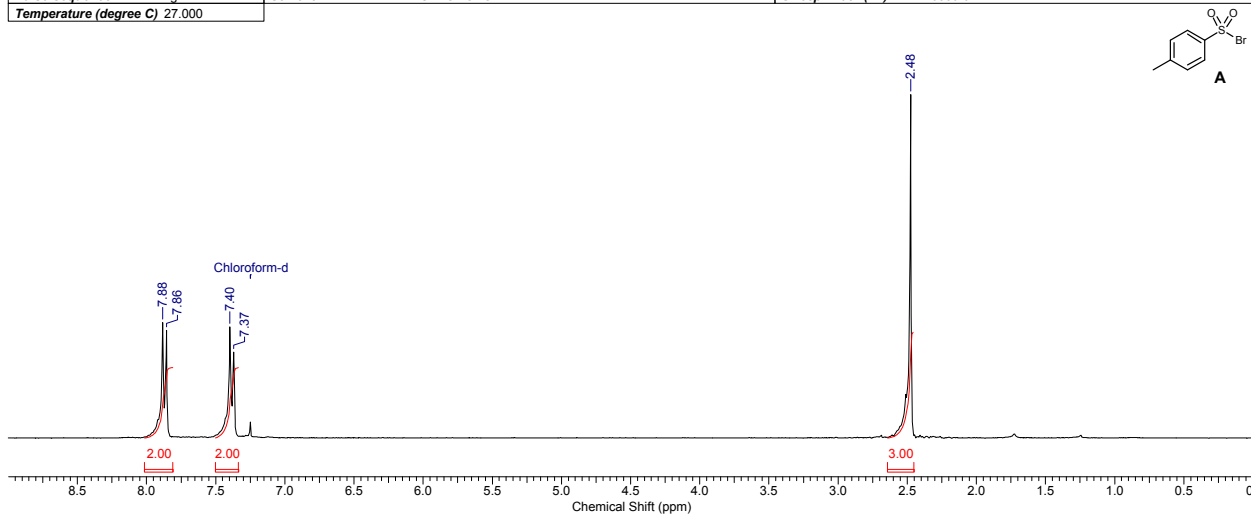
Acquisition Time (sec)	2.7150	Comment		Date	23 Nov 2018 13:39:12
File Name				Frequency (MHz)	300.13
Nucleus	¹ H	Number of Transients	1	Points Count	16384
Pulse Sequence	zg	Solvent	CHLOROFORM-D	Sweep Width (Hz)	6009.62
Temperature (degree C)	27.017				



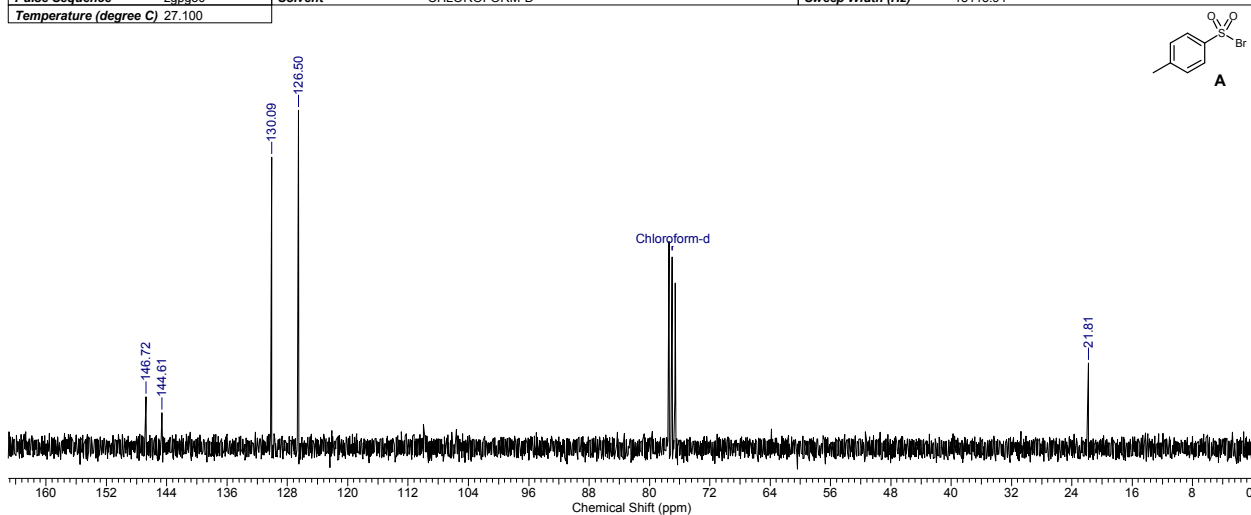
Acquisition Time (sec)	0.9050	Comment		Date	23 Nov 2018 13:41:20
File Name				Frequency (MHz)	75.48
Nucleus	¹³ C	Number of Transients	58	Points Count	16384
Pulse Sequence	zgpg30	Solvent	CHLOROFORM-D	Sweep Width (Hz)	18028.85
Temperature (degree C)	27.052				



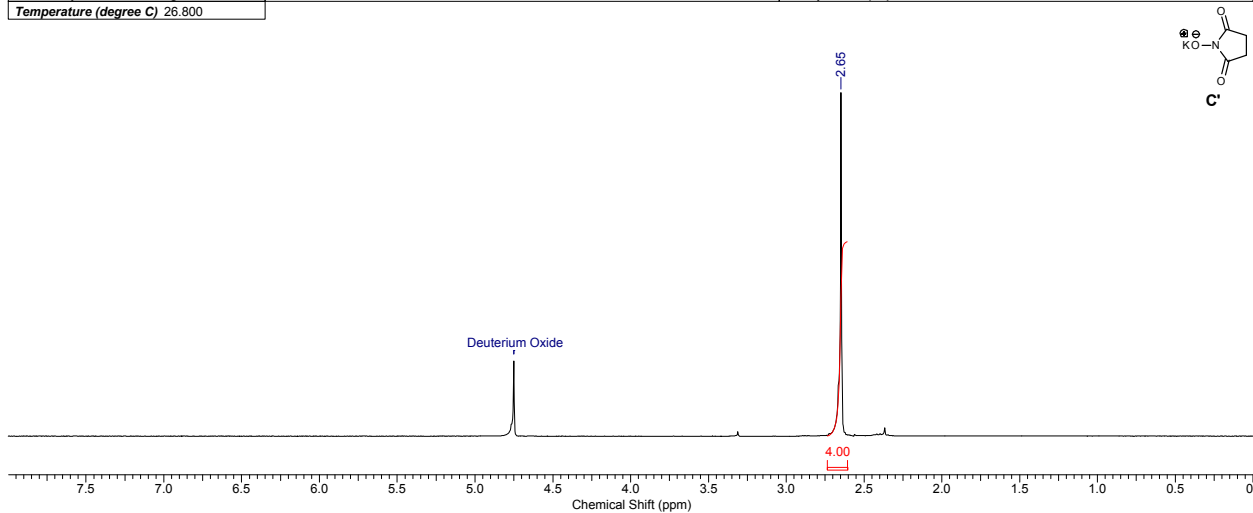
Acquisition Time (sec)	1.3518	Date	29 Nov 2018 09:04:00		Frequency (MHz)	300.13
File Name					Points Count	8192
Nucleus	¹ H	Number of Transients	1	Original Points Count	8124	
Pulse Sequence	zg	Solvent	CHLOROFORM-D		Sweep Width (Hz)	6009.62
Temperature (degree C)	27.000					



Acquisition Time (sec)	0.9006	Date	29 Nov 2018 09:04:00		Frequency (MHz)	75.48
File Name					Points Count	16384
Nucleus	¹³ C	Number of Transients	110	Original Points Count	16316	
Pulse Sequence	zgpg30	Solvent	CHLOROFORM-D		Sweep Width (Hz)	18115.94
Temperature (degree C)	27.100					



Acquisition Time (sec)	1.3518	Date	29 Nov 2018 10:08:00		Frequency (MHz)	300.13
File Name					Points Count	8192
Nucleus	¹ H	Number of Transients	1	Original Points Count	8124	
Pulse Sequence	zg	Solvent	DEUTERIUM OXIDE		Sweep Width (Hz)	6009.62
Temperature (degree C)	26.800					



Acquisition Time (sec)	0.9002	Date	29 Nov 2018 10:10:08		Frequency (MHz)	75.48
File Name					Points Count	16384
Nucleus	¹³ C	Number of Transients	117	Original Points Count	16308	
Pulse Sequence	zgpg30base	Solvent	DEUTERIUM OXIDE		Sweep Width (Hz)	18115.94
Temperature (degree C)	26.900					

