

## Supporting Information

### **Cyano-Substituted                      Benzochalcogenadiazole-Based                      Polymer Semiconductors for Balanced Ambipolar Organic Thin-Film Transistors†**

Shengbin Shi,<sup>‡a</sup> Hang Wang,<sup>‡a,b</sup> Peng Chen,<sup>a</sup> Mohammad Afsar Uddin,<sup>c</sup> Yuxi Wang,<sup>a</sup> Yumin Tang,<sup>a</sup> Han Guo,<sup>\*a</sup> Xing Cheng,<sup>a</sup> Shiming Zhang,<sup>b</sup> Han Young Woo,<sup>\*c</sup> and Xugang Guo<sup>\*a</sup>

<sup>a</sup> Department of Materials Science and Engineering and The Shenzhen Key Laboratory for Printed Organic Electronics, Southern University of Science and Technology (SUSTech), No. 1088, Xueyuan Road, Shenzhen, Guangdong 518055, China

<sup>b</sup> Key Laboratory of Flexible Electronics (KLOFE) & Institute of Advanced Materials (IAM), Jiangsu National Synergetic Innovation Center for Advanced Materials (SICAM), Nanjing Tech University, 30 South Puzhu Road, Nanjing 211816, Jiangsu, P. R. China

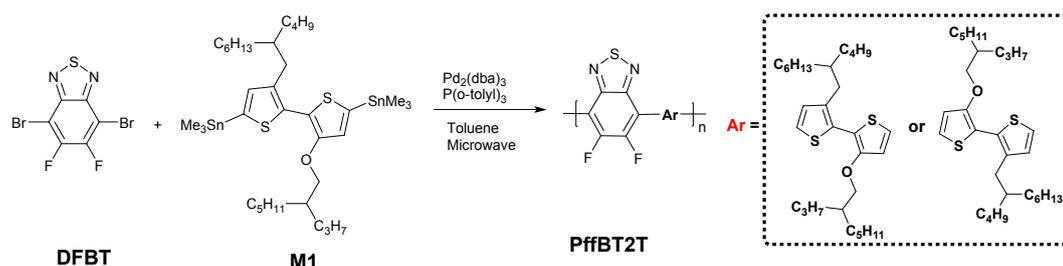
<sup>c</sup> Research Institute for Natural Sciences, Department of Chemistry, Korea University, Seoul 02841, South Korea

Email: guoh3@sustc.edu.cn; hywoo@korea.ac.kr; guoxg@sustc.edu.cn;

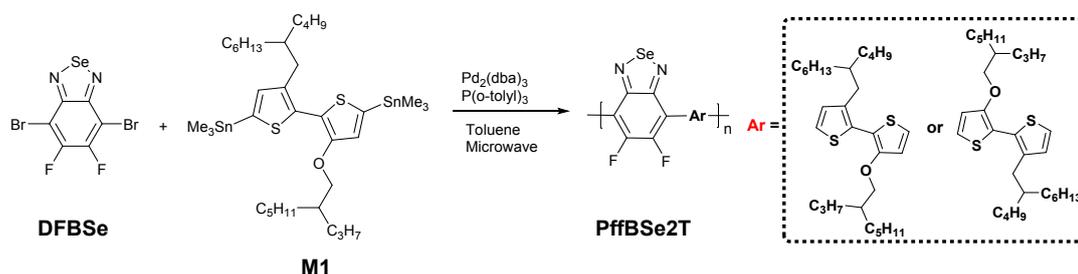
† Electronic supplementary information (ESI) available.

‡ These authors contributed equally to this work.

## 1. Synthesis of PffBT2T and PffBSe2T polymers.



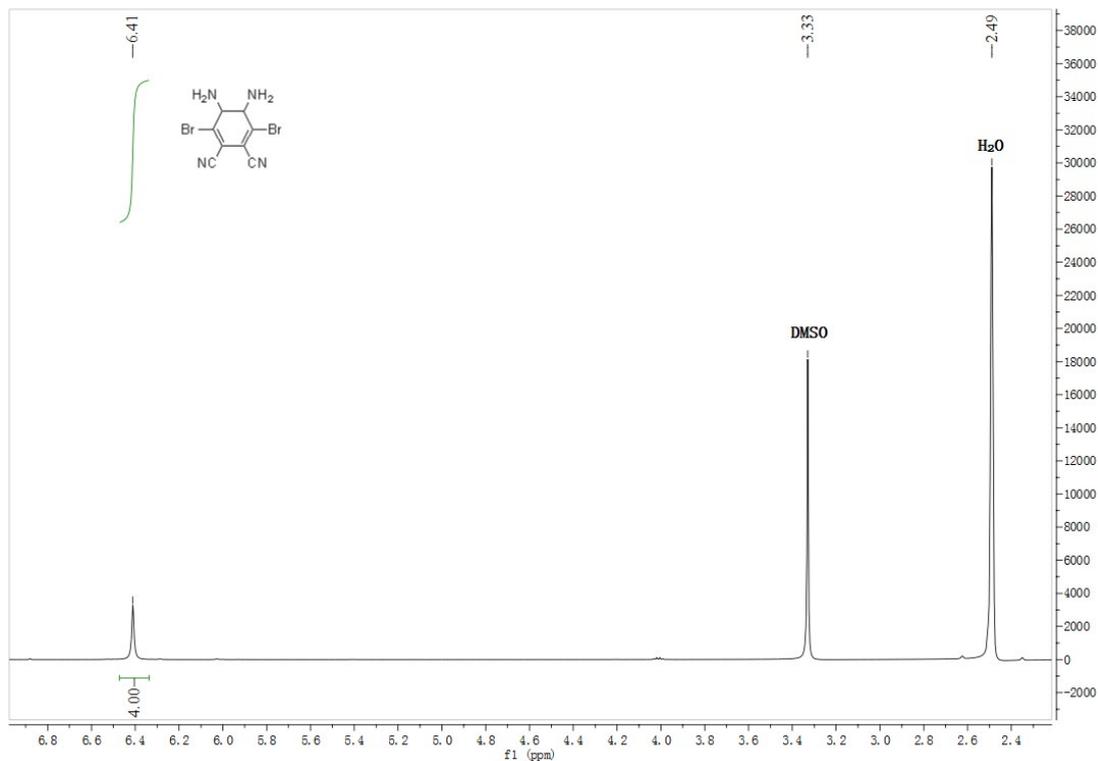
**PffBT2T:** The polymer was synthesized according to the literature report.<sup>1</sup>  $M_n = 35.5$  kDa,  $M_w = 58.0$  kDa, PDI = 1.63.



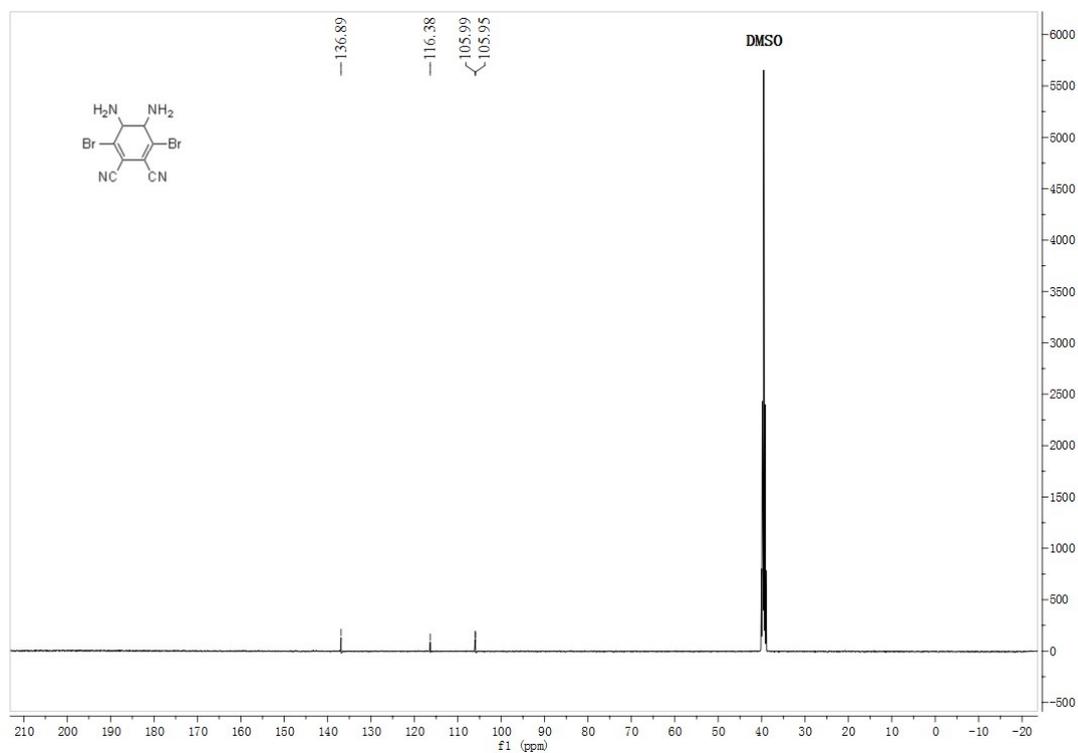
**PffBSe2T:** To a 5 mL microwave vial was added 4,7-dibromo-5,6-fluorobenzo[c][1,2,5]selenadiazole<sup>2</sup> (54.0 mg, 0.14 mmol), the distannylated monomer M1 (118.2 mg, 0.14 mmol), tris(dibenzylideneacetone)dipalladium(0) (1.5 mol%), tri(*o*-tolyl)phosphine (12 mol%), and 2.5 mL anhydrous toluene. The microwave vial and its contents were subjected to 3 pump/purge cycles with argon, then sealed under argon flow. It was then stirred at 80 °C for 10 min, 100 °C for 10 min, and 140 °C for 4.5 h under microwave irradiation. Then, 50  $\mu$ L of 2-(tributylstanny)thiophene was added and the reaction mixture was stirred under microwave irradiation at 140 °C for 20 min. Finally, 100  $\mu$ L of 2-bromothiophene was added and the reaction mixture was stirred at 140 °C for another 20 min. After cooling to room temperature, the reaction mixture was dripped into 150 mL methanol containing 1 mL 12 N HCl under vigorous stirring. After stirring for 1 h, the polymer precipitate was transferred to a Soxhlet thimble. The crude product was subjected to sequential Soxhlet extraction with methanol, acetone, hexane, dichloromethane, and chloroform. The chloroform fraction was concentrated to  $\sim$ 5 mL then dripped into 150 mL methanol under vigorous stirring. The precipitate was collected by filtration and dried under reduced pressure to afford a deep colored solid as the final product polymer (yield: 70.8%). <sup>1</sup>H NMR (400 MHz, 80 °C, Toluene-*d*<sub>8</sub>, ppm)  $\delta$  8.37-8.30 (m, 2H), 4.22 (s, 2H),

3.20 (s, 2H), 1.97-1.95 (m, 2H), 1.80-1.38 (m, 28H), 1.05-0.91 (m, 12H). Elem. Anal.: Calcd. for  $C_{36}H_{48}F_2N_2OS_2Se$  (%): C, 61.26; H, 6.85; N, 3.97; S, 9.08. Found (%): C, 60.64; H, 6.61; N, 3.84; S, 8.86. Molecular weight:  $M_n = 17.2$  kDa,  $M_w = 26.8$  kDa, PDI = 1.56.

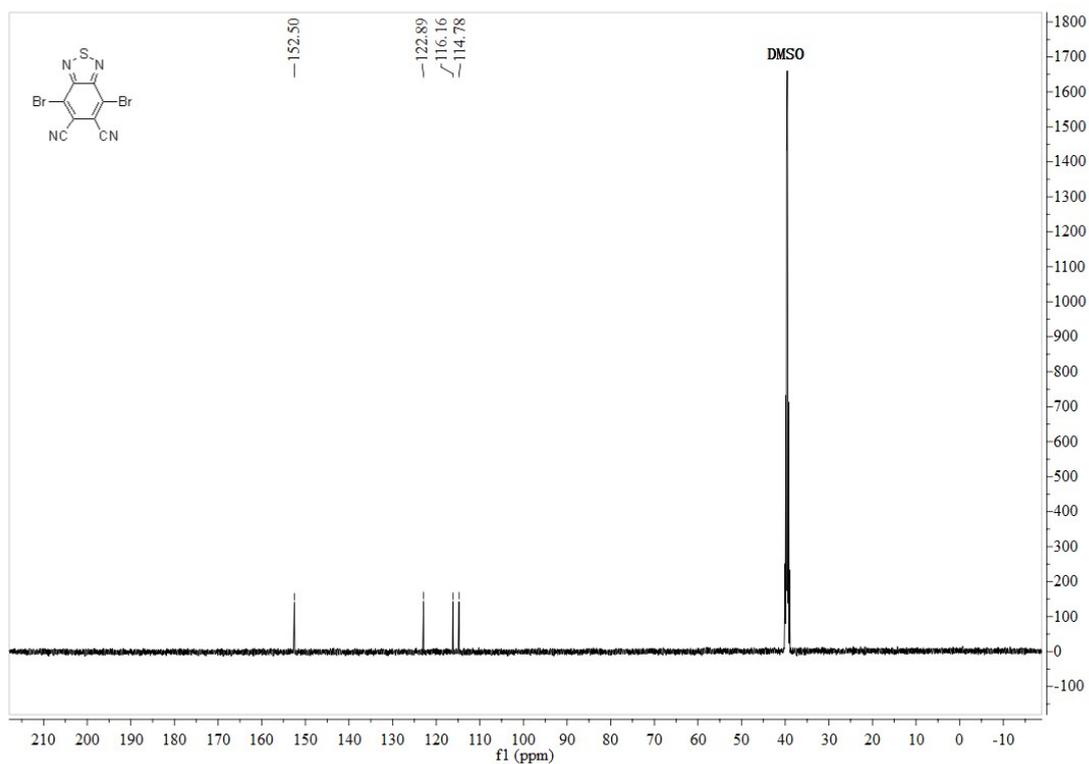
## 2. NMR and HRMS Spectra.



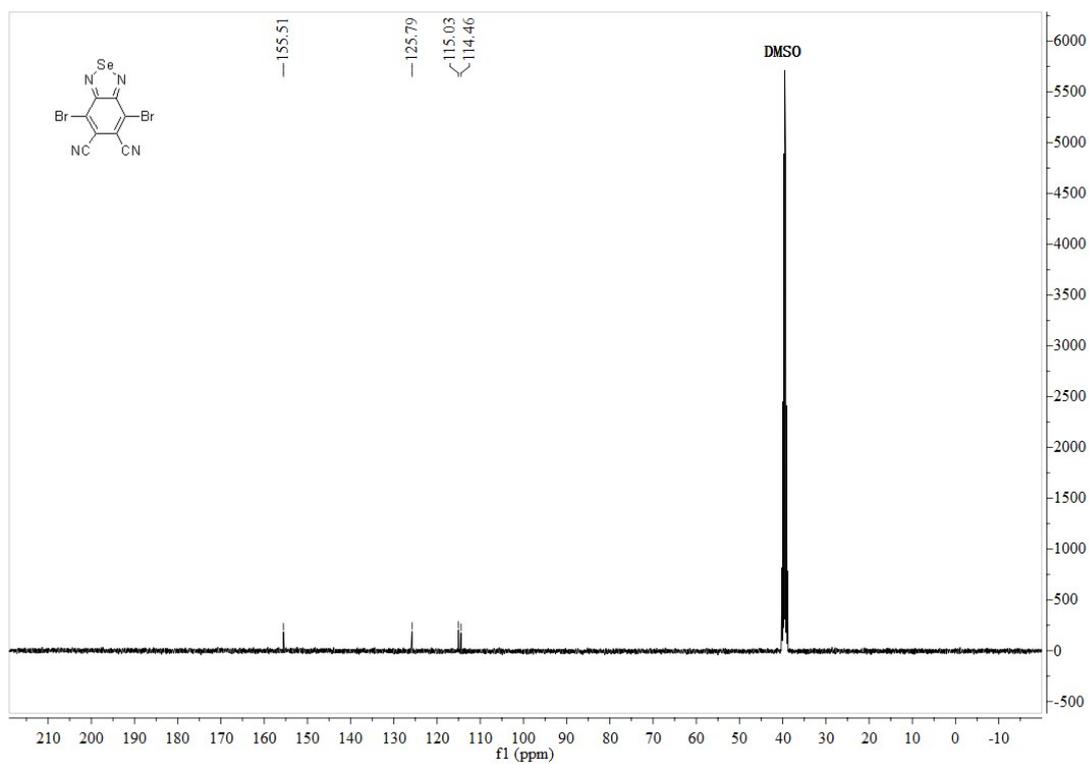
**Figure S1.**  $^1\text{H}$  NMR spectrum of compound DCNBr (r.t., in  $\text{DMSO-d}_6$ ).



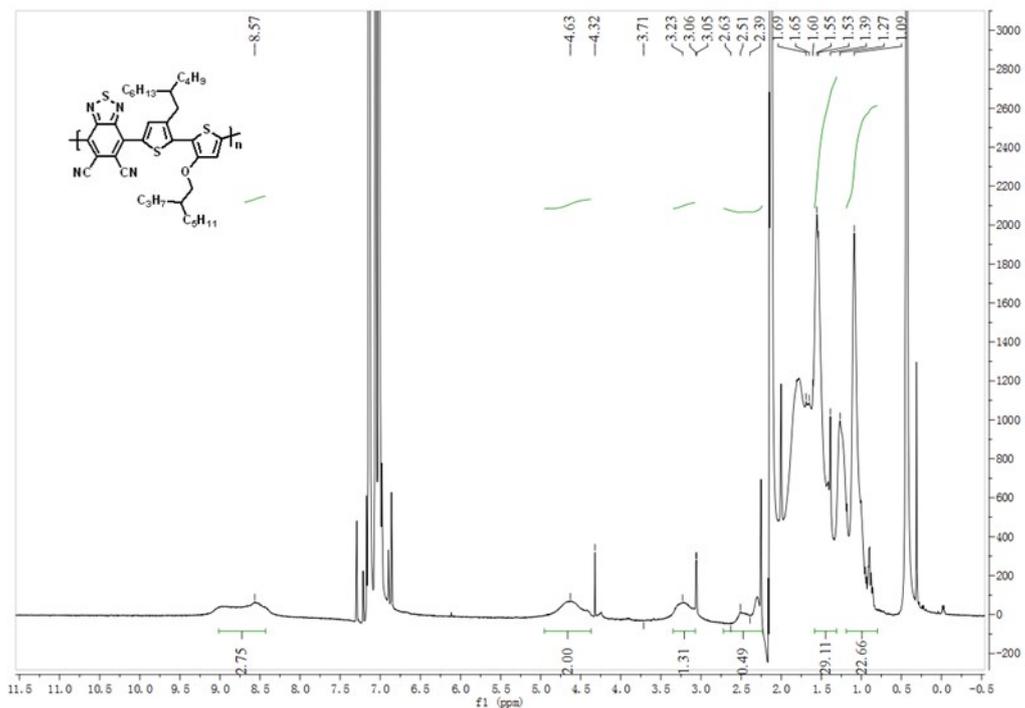
**Figure S2.**  $^{13}\text{C}$  NMR spectrum of compound DCNBr (r.t., in  $\text{DMSO-d}_6$ ).



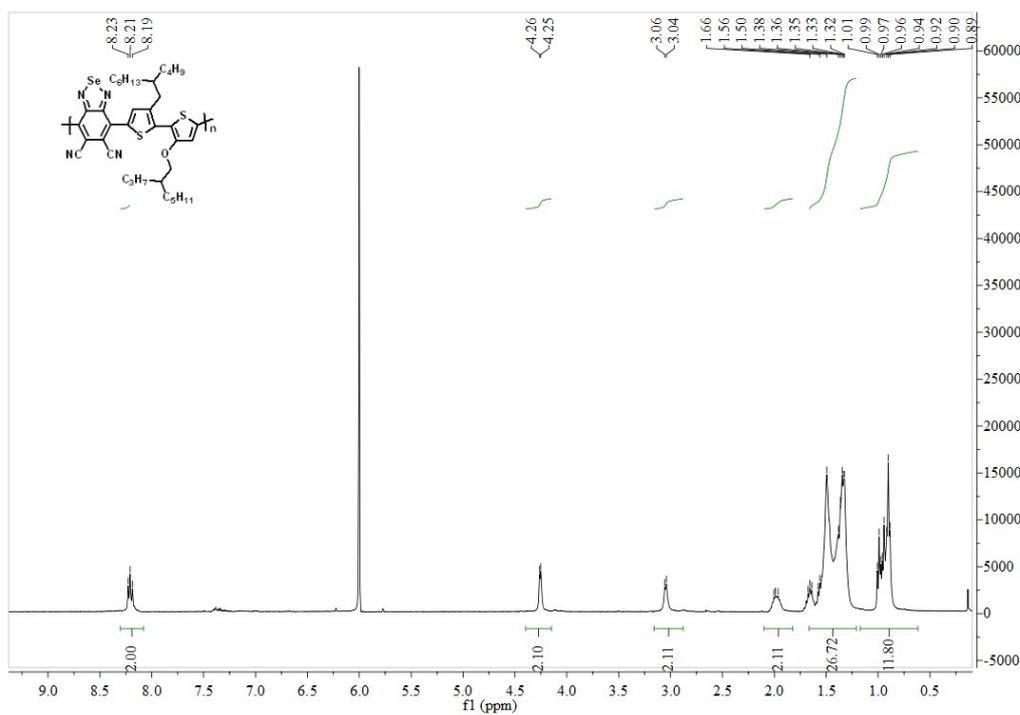
**Figure S3.**  $^{13}\text{C}$  NMR spectrum of monomer DCNBT (r.t., in  $\text{DMSO-d}_6$ ).



**Figure S4.**  $^{13}\text{C}$  NMR spectrum of monomer DCNBSe (r.t., in  $\text{DMSO-d}_6$ ).



**Figure S5.**  $^1\text{H}$  NMR spectrum of polymer PDCNBT2T (80 °C in Toluene- $d_8$ ).



**Figure S6.**  $^1\text{H}$  NMR spectrum of polymer PDCNBS2T (80 °C in  $\text{C}_2\text{D}_2\text{Cl}_4$ ).

T: FTMS - p ESI Full ms [100.00-600.00]

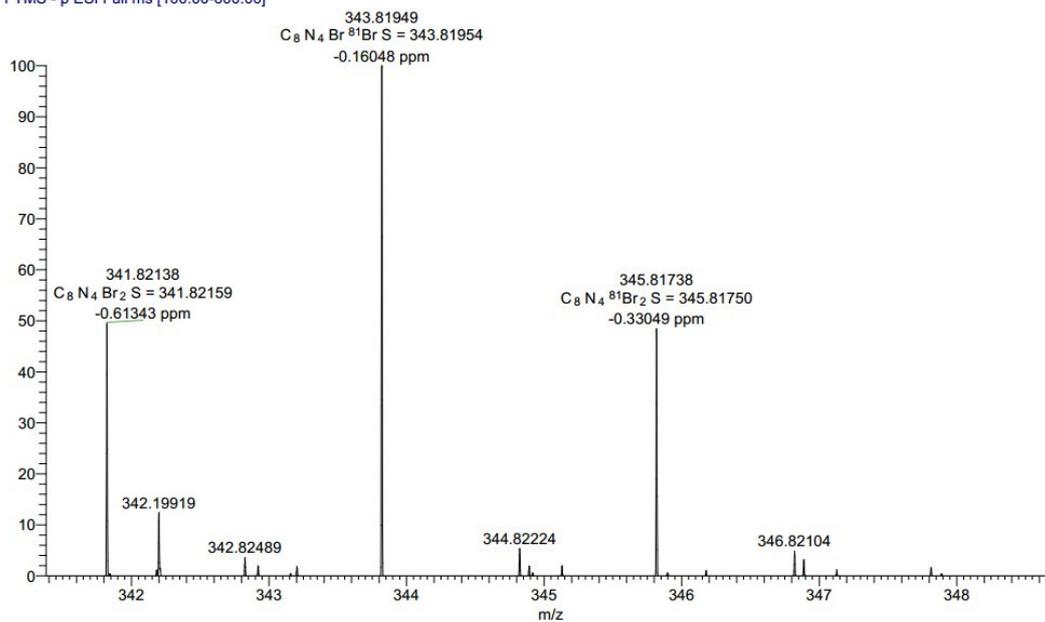


Figure S7. HRMS spectrum of monomer DCNBT.

T: FTMS - p ESI Full ms [100.00-800.00]

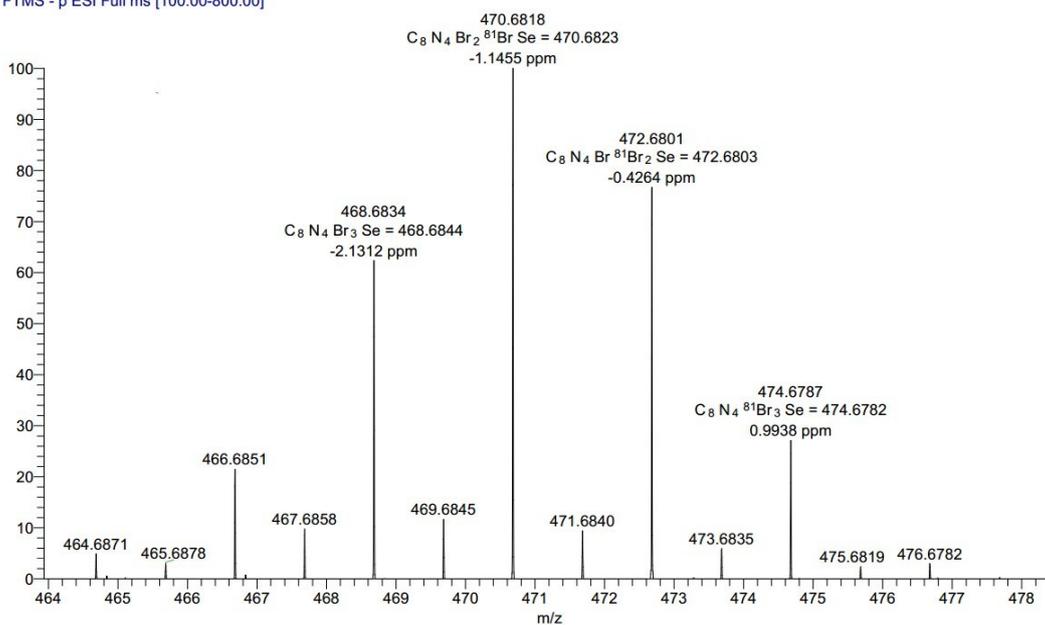
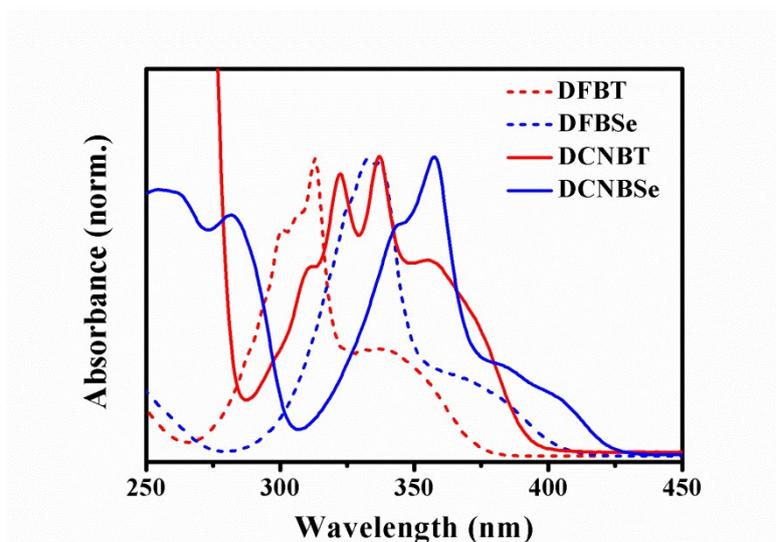
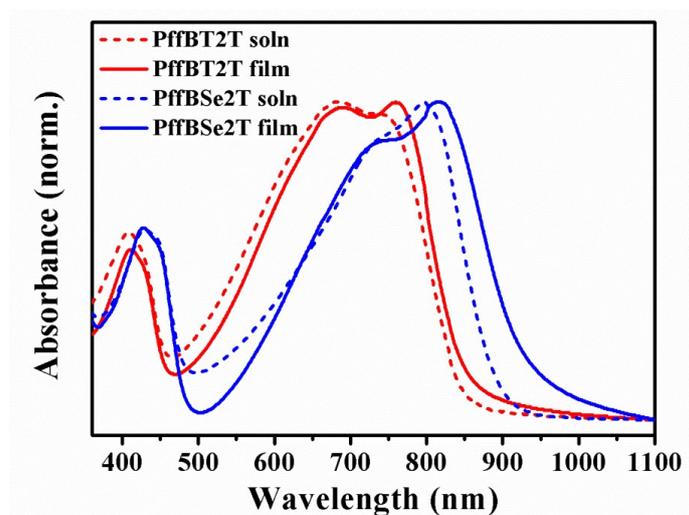


Figure S8. HRMS spectrum of monomer DCNBSe.

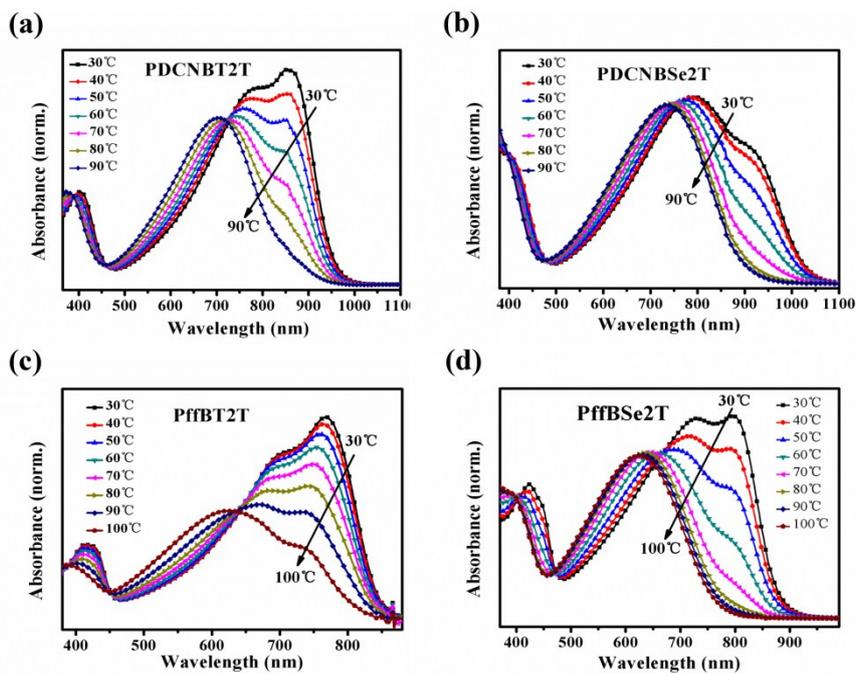
### 3. Other supporting experiment data.



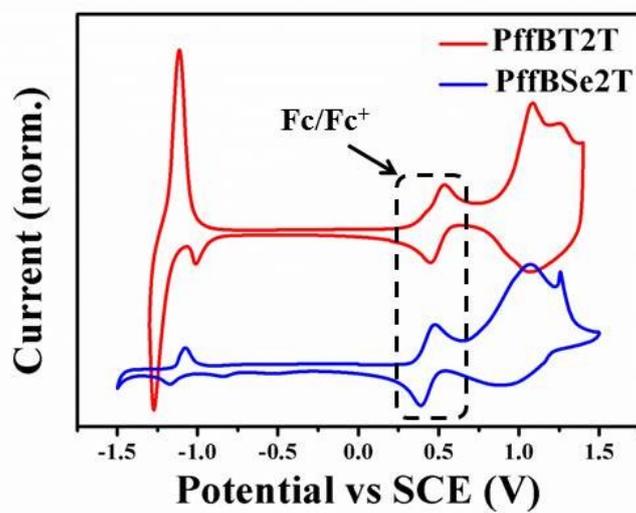
**Figure S9.** UV-Vis absorption spectra of the monomers (DFBT, DFBSi, DCNBT, and DCNBSi) in diluted solutions ( $10^{-5}$  M in dichloromethane).



**Figure S10.** UV-Vis absorption spectra of polymers PffBT2T and PffBSe2T in diluted solutions ( $10^{-5}$  M in *o*-DCB) and at thin-film state (spin-coated on glass from  $5 \text{ mg mL}^{-1}$  *o*-DCB solutions).



**Figure S11.** Temperature-dependent UV-Vis absorption spectra of polymers PDCNBT2T, PDCNBSe2T, PffBT2T, and PffBSe2T in diluted solutions ( $10^{-5}$  M in *o*-DCB).

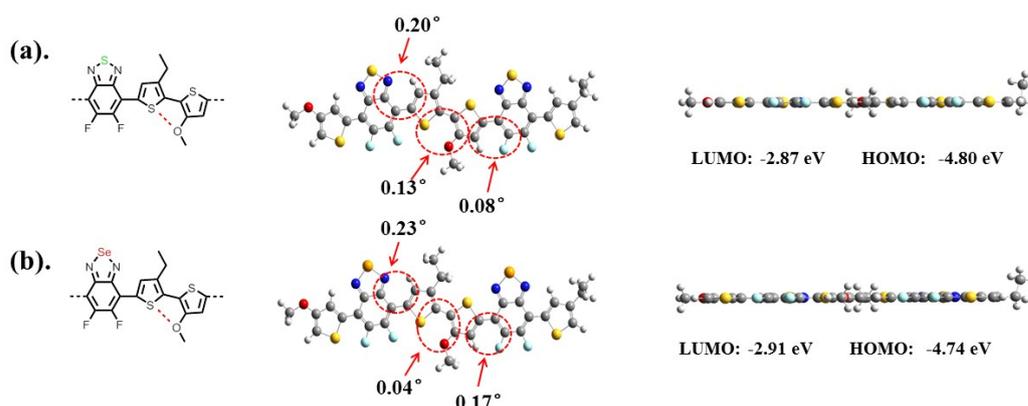


**Figure S12.** Cyclic voltammograms of PffBT2T and PffBSe2T thin films measured in 0.1 M  $(n\text{-Bu})_4\text{N}^+\text{PF}_6^-$  acetonitrile solutions at scan rate of  $50\text{ mV s}^{-1}$ .

**Table S1.** Molecular weights, electrochemical and optical properties of polymer PffBT2T and PffBSe2T.

Polymer	$M_n$ [kDa] <sup>a)</sup>	PDI	$\lambda_{\max}$ (soln) [nm] <sup>b)</sup>	$\lambda_{\max}$ (film) [nm] <sup>c)</sup>	$E_g^{\text{opt}}$ [eV] <sup>e)</sup>	$E_{\text{HOMO}}$ [eV] <sup>d)</sup>	$E_{\text{LUMO}}$ [eV] <sup>d)</sup>	$E_{\text{LUMO}}$ [eV] <sup>e)</sup>
<b>PffBT2T</b>	35.5	1.63	748	759	1.46	-5.20	-3.44	-3.74
<b>PffBSe2T</b>	17.2	1.56	799	820	1.33	-5.12	-3.61	-3.79

- a) High-temperature GPC at 150 °C, versus polystyrene standards using trichlorobenzene as the eluent; b) Diluted *o*-DCB solution ( $10^{-5}$  M); c) As-cast thin-film from *o*-DCB solution,  $E_g^{\text{opt}} = 1240/\lambda_{\text{onset}}$  eV. d)  $E_{\text{HOMO}} = -e(E_{\text{ox}}^{\text{onset}} + 4.80)$  eV,  $E_{\text{LUMO}} = -e(E_{\text{red}}^{\text{onset}} + 4.8)$  eV,  $E_{\text{ox}}^{\text{onset}}$  and  $E_{\text{red}}^{\text{onset}}$  determined electrochemically using Fc/Fc<sup>+</sup> internal standard. e)  $E_{\text{LUMO}}^{\text{cal}} = E_{\text{HOMO}} + E_g^{\text{opt}}$ .

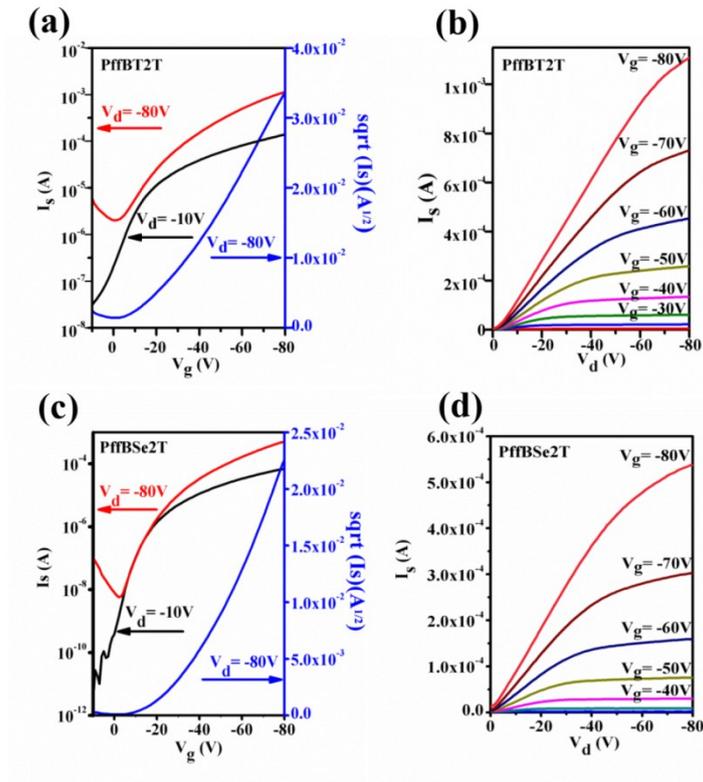


**Figure S13.** DFT optimized geometries of the dimers for (a) PffBT2T and (b) PffBSe2T at the B3LYP/6-31G (d, p) level. The dihedral angles are indicated in the red circles. The alkyl substituents are truncated for simplicity.

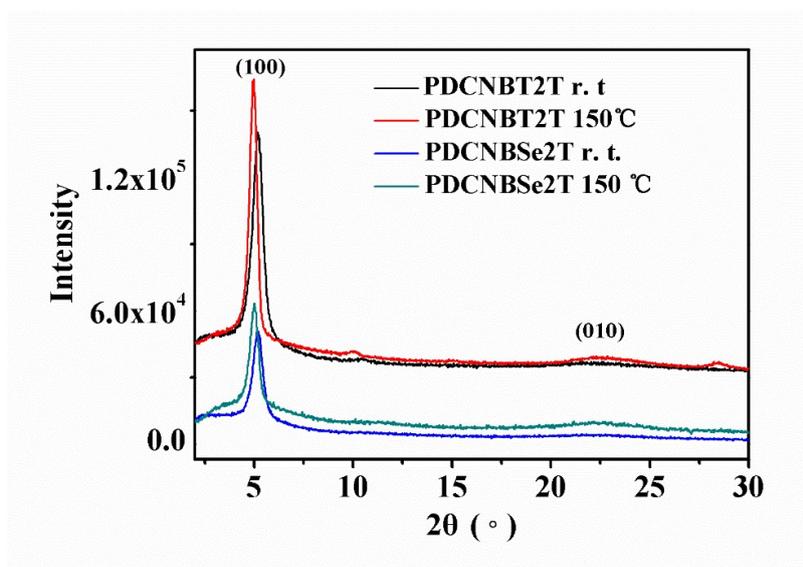
**Table S2.** TG/BC OTFT performance parameters of PffBT2T and PffBSe2T.

Polymer	$T_{\text{anneal}}$ (°C)	$\mu_{\text{lin}}$ ( $\text{cm}^2 \text{V}^{-1} \text{s}^{-1}$ ) <sup>a)</sup>	$\mu_{\text{sat}}$ ( $\text{cm}^2 \text{V}^{-1} \text{s}^{-1}$ ) <sup>a)</sup>	$V_{\text{th}}$ (V)	$I_{\text{on}}/I_{\text{off}}$
PffBT2T <sup>1</sup>	r. t	0.19 (0.18)	0.53 (0.32)	-20	$10^3$
	250	0.36 (0.32)	0.69 (0.53)	-19	$10^4$
PffBSe2T	250	0.13 (0.09)	0.28 (0.21)	-28	$10^5$

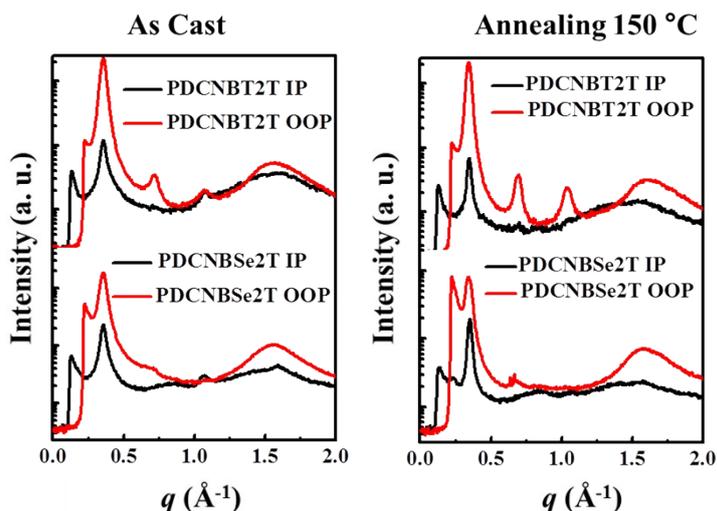
- <sup>a)</sup> Maximum mobility with average value from at least 5 devices shown in the parentheses.



**Figure S14.** Transfer and output characteristics of (a, b) PffBT2T and (c, d) PffBSe2T-based OTFT devices ( $L = 10 \mu\text{m}$ ,  $W = 5 \text{mm}$ ) fabricated under the optimal condition. The fabrication and characterization of polymer PffBT2T and PffBSe2T-based OTFTs were according to the literature report.<sup>1</sup>



**Figure S15.** Out-of-plane  $2\theta$  XRD measurement of PDCNBT2T and PDCNBSe2T films annealed at room temperature (r. t.) and  $150 \text{ }^\circ\text{C}$ .



**Figure S16.** In-plane (IP) and out-of-plane (OOP) line-cut profiles of 2D-GIXD measurements.

**Table S3.** Summary of the lamellar distances, the  $\pi$ - $\pi$ -stacking distances, and CCLs of as-cast polymer films and thermal annealed polymer films.

Polymer	Crystallographic parameters		As-cast	Thermal Annealed
PDCNBT2T	$q_z$ profile (100)	$q(\text{\AA}^{-1})$	0.36	0.36
		d-spacing ( $\text{\AA}$ )	17.44	17.44
		Correlation length (nm)	12.67	14.76
	$q_{xy}$ profile (100)	$q(\text{\AA}^{-1})$	0.36	0.36
		d-spacing ( $\text{\AA}$ )	17.44	17.44
		Correlation length (nm)	11.97	19.55
$q_z$ profile (010)	$q(\text{\AA}^{-1})$	1.56	1.60	
	d-spacing ( $\text{\AA}$ )	4.03	4.03	
PDCNBSe2T	$q_z$ profile (100)	$q(\text{\AA}^{-1})$	0.35	0.35
		d-spacing ( $\text{\AA}$ )	17.94	17.94
		Correlation length (nm)	11.41	11.90
	$q_{xy}$ profile (100)	$q(\text{\AA}^{-1})$	0.35	0.35
		d-spacing ( $\text{\AA}$ )	17.94	17.94
		Correlation length (nm)	12.61	18.7
	$q_z$ profile (010)	$q(\text{\AA}^{-1})$	1.56	1.56
d-spacing ( $\text{\AA}$ )		4.03	4.03	

## Reference

1. S. Shi, Q. Liao, Y. Tang, H. Guo, X. Zhou, Y. Wang, T. Yang, Y. Liang, X. Cheng, F. Liu and X. Guo, *Adv. Mater.*, 2016, **28**, 9969-9977.
2. X. He, B. Cao, T. C. Hauger, M. Kang, S. Gusarov, E. J. Lubber and J. M. Buriak, *ACS. Appl. Mater. Interfaces.*, 2015, **7**, 8188-8199.