

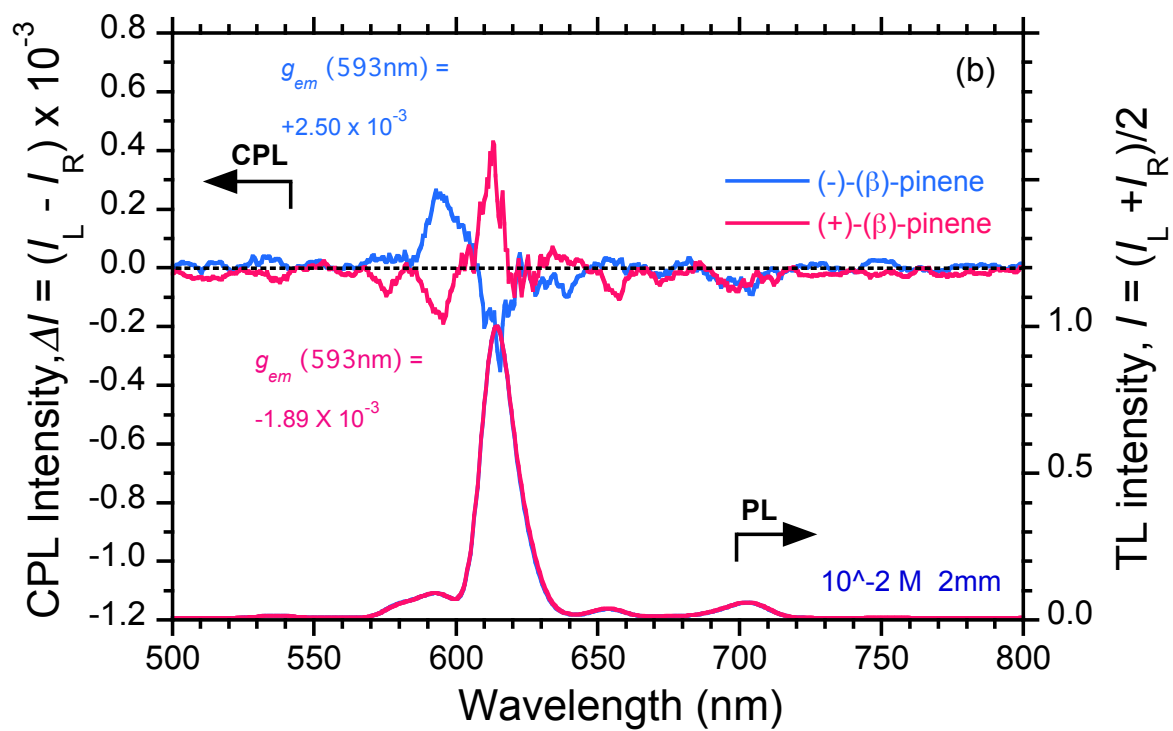
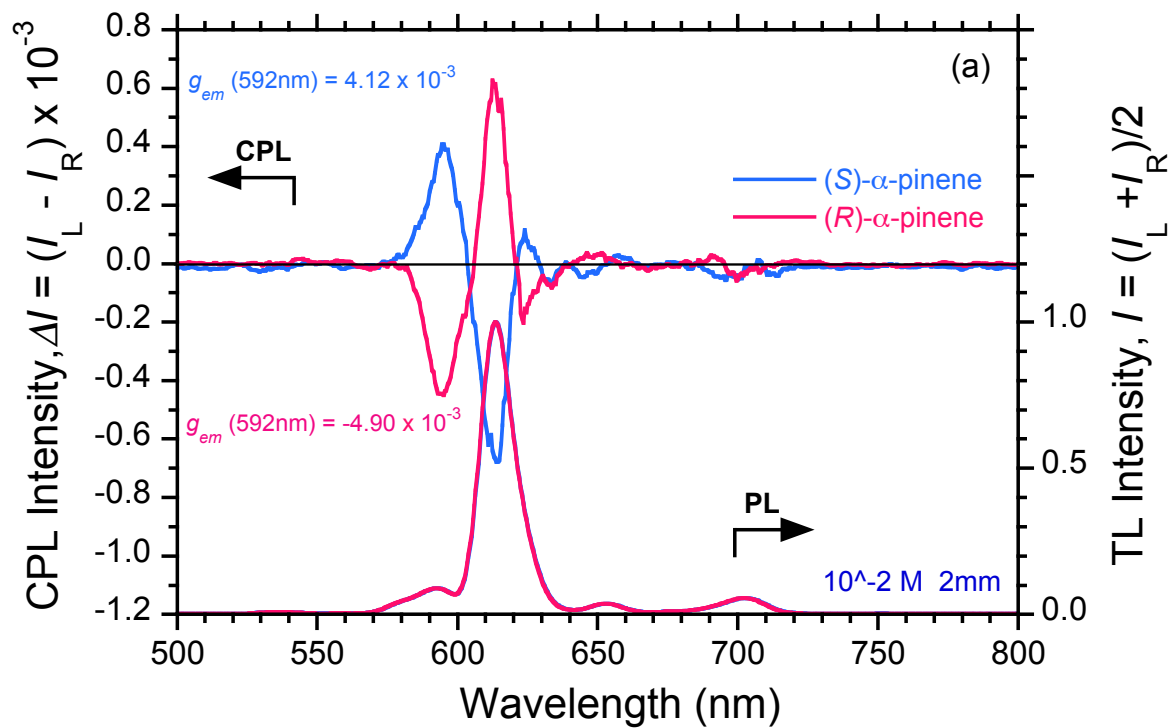
Electronic Supporting Information (ESI) for

Unveiling controlled breaking of the mirror symmetry of Eu(fod)₃ with α -/ β -pinene and BINAP by circularly polarised luminescence (CPL), CPL excitation, and ¹⁹F-/³¹P{¹H}-NMR spectra and Mulliken charges

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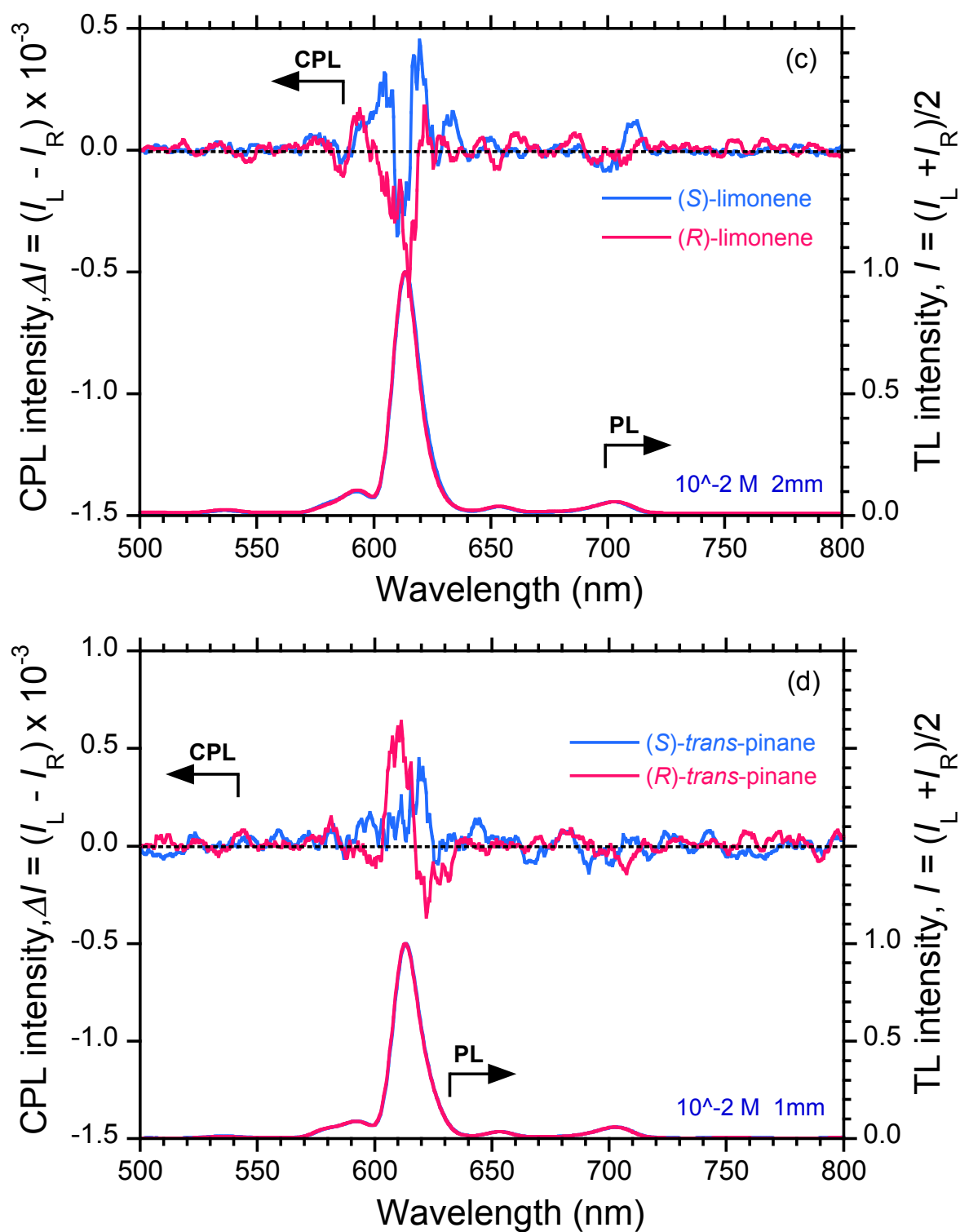


Fig. S1. CPL/PL spectra of $\text{Eu}(\text{fod})_3$ dissolved in neat (a) (S) -/ (R) - α -pinene excited at 345 nm, (b) (S) -/ (R) - β -pinene excited at 335 nm, (c) (S) -/ (R) -limonene excited at 335 nm and (d) (S) -/ (R) -trans-pinane excited at 330 nm.

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Dipole Moment = 11.7577 Debye
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Fig. S2. Brief log data of (top) *fac*- and (bottom) *mer*-Sc(fod)₃ as models of *fac*- and *mer*-Eu(fod)₃ in a vacuum, obtained with Gaussian 09 (rev.D.01)^{S1}

Ref. S1) Gaussian 09, Rev. D.01, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, T. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, O. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, and D. J. Fox, Gaussian, Inc., Wallingford CT, 2013.

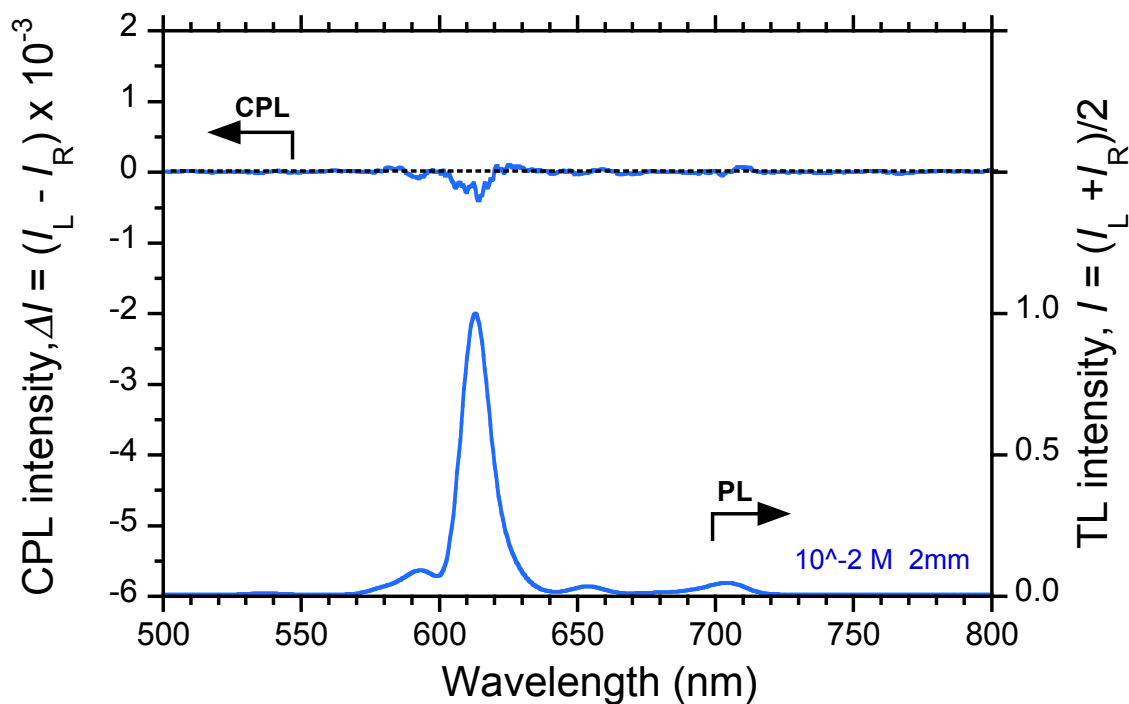


Fig. S3. CPL/PL spectra of Eu(fod)₃ in EtOH-free anhydrous chloroform excited at 335 nm.

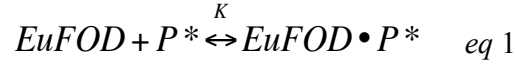
Table S1. PL quantum yield, ϕ_{PL} , of Eu(fod)₃ in EtOH-free dehydrated chloroform, in neat (*S*)-/(*R*)- α -pinene, (*R*)-BINAP (1:1 nominal molar ratio in the chloroform), (*R*)-BINAPO (1:1 nominal molar ratio in the chloroform), (*S*)-BINAPO (1:1 nominal molar ratio in the chloroform).

Solvents	Quantum yield, ϕ_{PL} (%) ^{S2,S3}	
	⁵ D ₀ → ⁷ F ₁ transition	⁵ D ₀ → ⁷ F ₂ transition
chloroform	0.3	3.9
(1 <i>S</i>)- α -pinene	0.3	3.4
(1 <i>R</i>)- α -pinene	0.2	2.8
(<i>R</i>)-BINAP in chloroform	0.1	1.9
(<i>R</i>)-BINAPO in chloroform	0.6	7.5
(<i>S</i>)-BINAPO in chloroform	0.3	3.9

^{S2)} A. Wakamiya, K. Mori and S. Yamaguchi, *Angew. Chem., Int. Ed.*, 2007, **46**, 4273–4276.

^{S3)} J. R. Lakowicz, *Principles of Fluorescence Spectroscopy*, Springer, Heidelberg, 3rd edn, 2006.

The binding constant, K , can be obtained from the solution equilibria illustrated below:



$$K = \frac{[EuFOD \cdot P^*]}{[EuFOD][P^*]} \quad eq\ 2$$

where

$$\begin{aligned} [P^*] &= [P^*]_0 - [EuFOD \cdot P^*] \\ [EuFOD] &= [EuFOD]_0 - [EuFOD \cdot P^*] \end{aligned} \quad eq\ 3$$

considering that

$$[P^*]_0 \gg [EuFOD]_0 \quad eq\ 4$$

thus,

$$[P^*] = [P^*]_0 \quad eq\ 5$$

and K can be written as *eq 6*,

$$K = \frac{[EuFOD \cdot P^*]}{([EuFOD]_0 - [EuFOD \cdot P^*])[P^*]} \quad eq\ 6$$

The definition of g_{em} in concentration is given in *eq 7*,

$$g_{em}(apparent) = \frac{[EuFOD \cdot P^*]}{[EuFOD] + [EuFOD \cdot P^*]} \times g_{em}(EuFOD100\%) [EuFOD \cdot P^*] \quad eq\ 7$$

since

$$[EuFOD]_0 = [EuFOD] + [EuFOD \cdot P^*] \quad eq\ 8$$

the equation can be rewritten as *eqs 9*,

$$\frac{g_{em}(apparent)}{g_{em}(EuFOD100\%)} [EuFOD \cdot P^*]^{-1} = \frac{[EuFOD \cdot P^*]}{[EuFOD]_0} \quad eq\ 9$$

From *eqs 6 and 9*, it can be arranged to *eq10*,

$$\frac{g_{em}(apparent)}{g_{em}(EuFOD100\%)} [EuFOD \cdot P^*]^{-1} = \frac{1}{K[P^*]} + 1 \quad eq\ 10$$

The fitting of *eq 10* to the plots in Fig. 3 gives the K values when $1/[EuFOD \cdot P^*]$ versus $1/[P^*]$ is plotted. The calculated binding constants, K values, given by the fitting of the plots of $Eu(fod)_3$ in both (S)- and (R)- α -pinene are as low as $\approx 10^{-3}$.

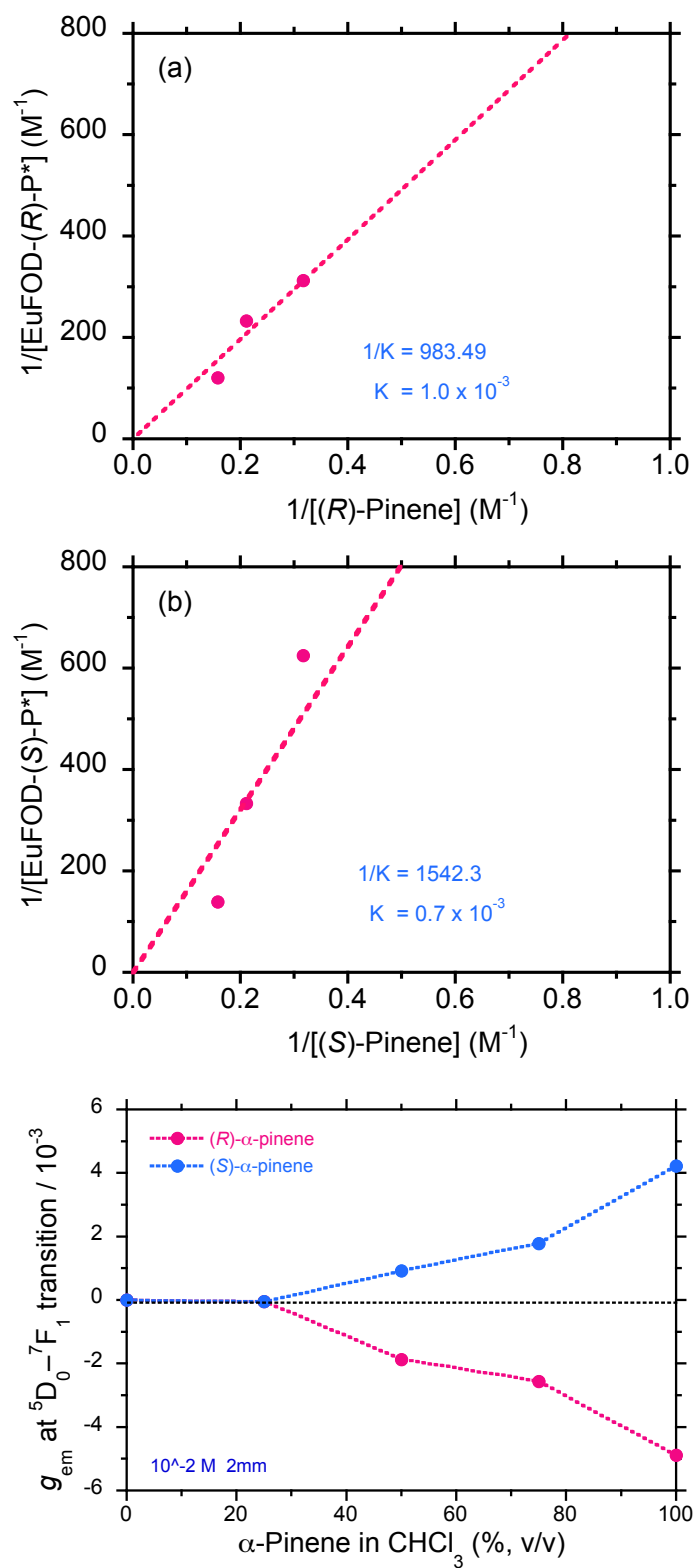


Fig. S4. The fitting plots of $1/[\text{Eu}(\text{fod})_3//\alpha\text{-pinene}] \text{ M}^{-1}$ versus (a) $1/[(R)\text{-}\alpha\text{-pinene}]$ in M^{-1} and (b) $1/[(S)\text{-}\alpha\text{-pinene}]$ in M^{-1} associated with (c) the g_{em} value at 593 nm as a function of volume fraction of α -pinene in EtOH-free anhydrous chloroform.

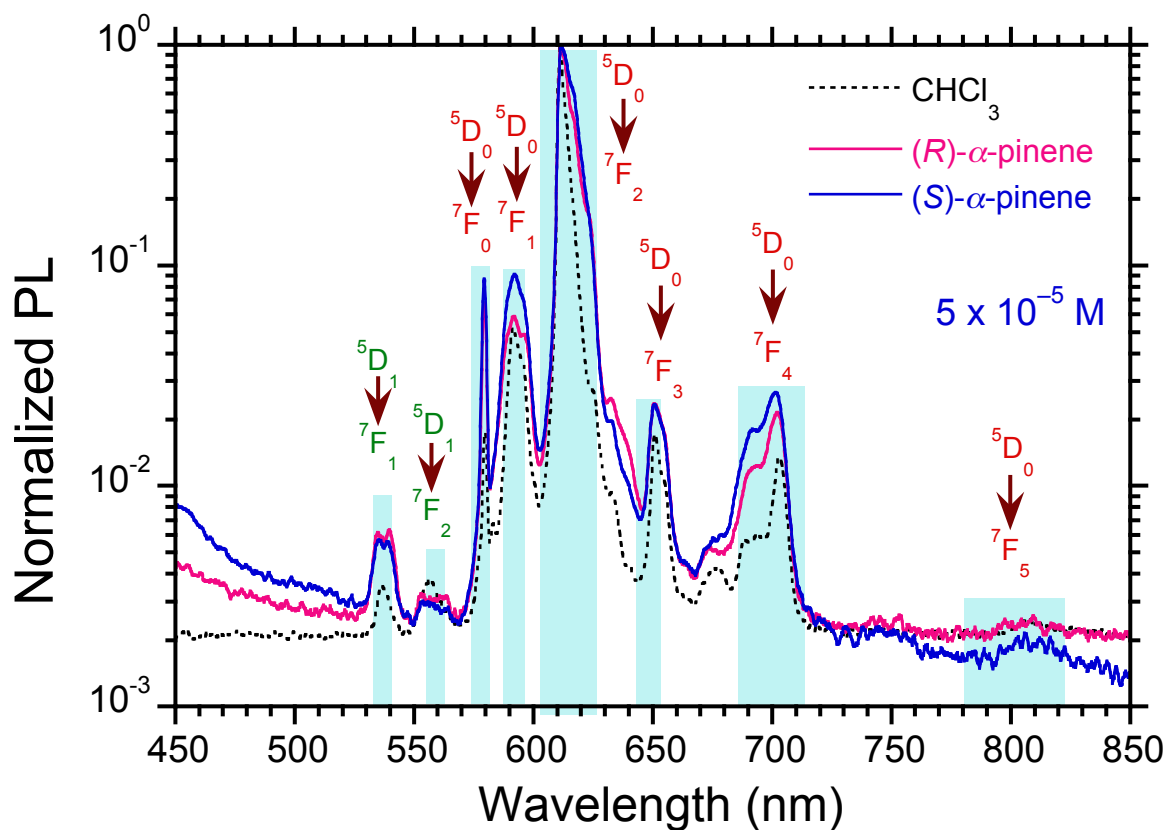


Fig. S5. Semilog plot of the PL spectra of $\text{Eu}(\text{fod})_3$ in chloroform (λ_{ex} 278 nm; 5.0×10^{-6} M) and in (*S*)-/(*R*)- α -pinene (λ_{ex} 317 nm; 5.0×10^{-5} M). Measurement conditions: 10 nm bandwidth for excitation; 1 nm bandwidth for emission; PMT in high-sensitivity mode; 1 second response time; 0.2 nm data interval; and one scan collected at a rate of 100 nm min^{-1} . Under these conditions, no stray light (556 nm and 634 nm) from the two gratings of the spectrometer and Raman band Stokes line ($\approx 3000 \text{ cm}^{-1}$) by the chloroform solvent was observed.

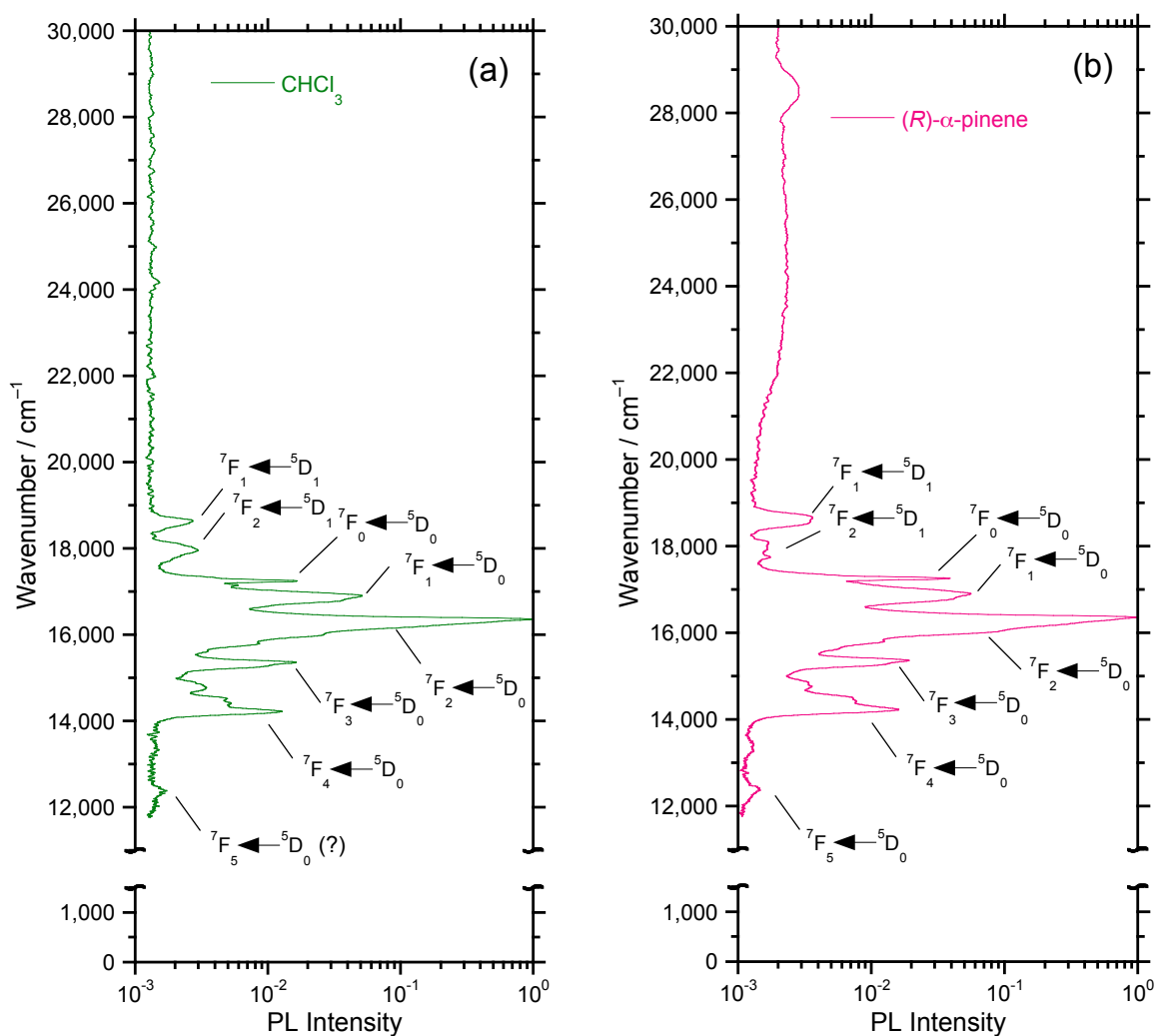


Fig. S6. PL spectra (in cm^{-1}) of $\text{Eu}(\text{fod})_3$ in (a) chloroform (λ_{ex} 278 nm, $35,970 \text{ cm}^{-1}$) and (b) in neat (*R*)- α -pinene (λ_{ex} 317 nm, $31,450 \text{ cm}^{-1}$). In (b), the small bump at $28,400 \text{ cm}^{-1}$ is attributed to the Raman C–H vibration band of α -pinene. All data were taken from Fig. S5.

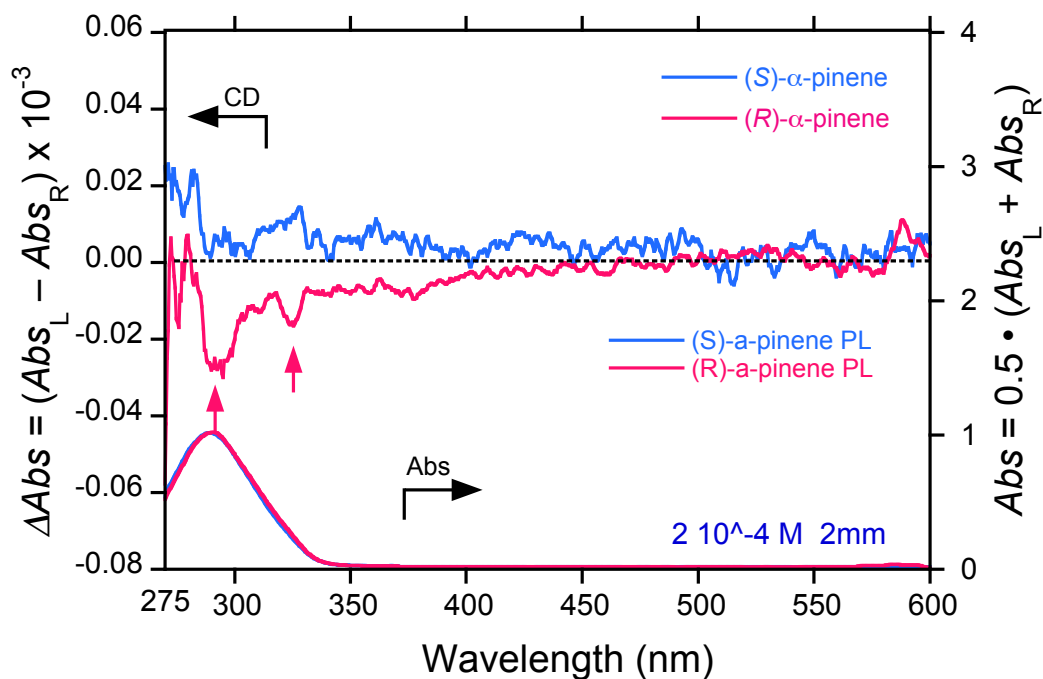


Fig. S7. CD and UV-vis spectra of $\text{Eu}(\text{fod})_3$ in neat (*S*)-/(*R*)- α -pinene at 2.0×10^{-4} M (path length: 2.0 mm) obtained by subtracting the spectra of (*S*)-/(*R*)- α -pinene.

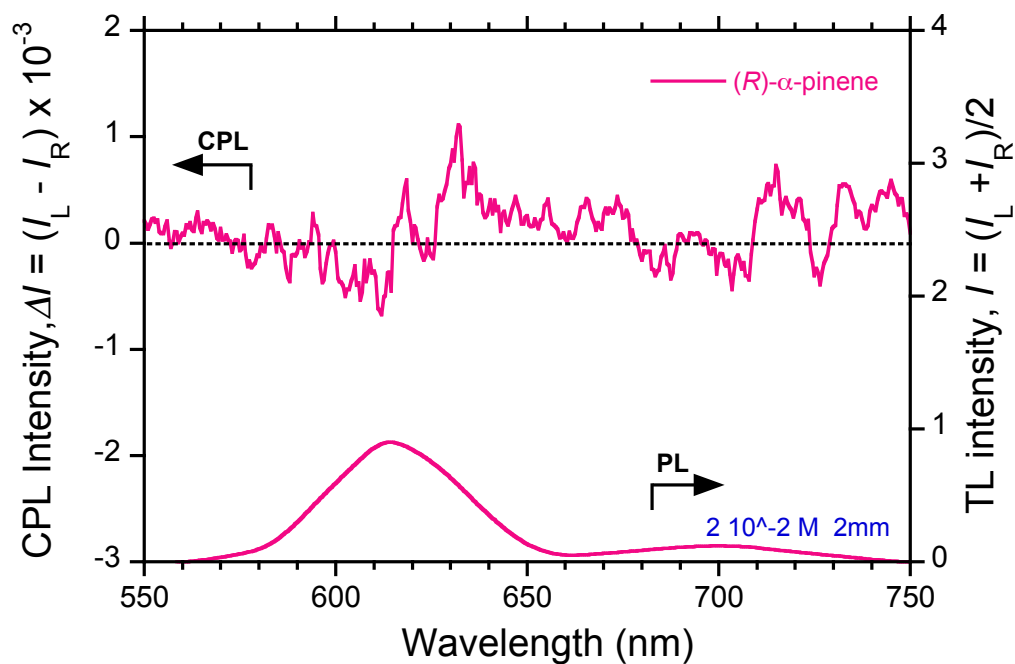


Fig. S8. CPL/PL spectra of $\text{Eu}(\text{dpm})_3$ dissolved in neat (*S*)-/(*R*)- α -pinene excited at 345 nm (2.0×10^{-2} M, path length: 2.0 mm).

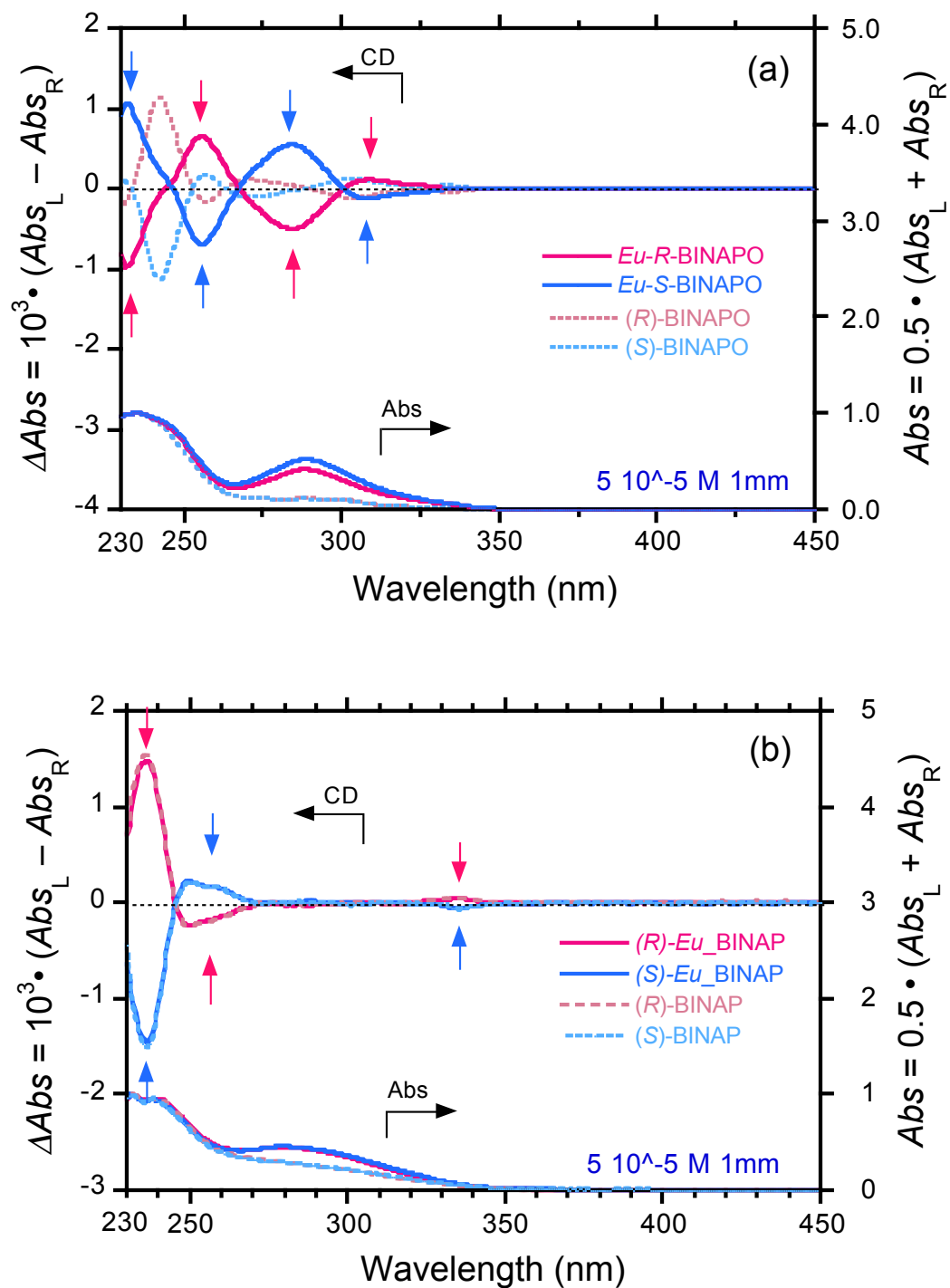


Fig. S9. CD and UV-vis spectra of (a) Eu(fod)_3 with (S) -/ (R) -BINAPO (1:1) and (b) (S) -/ (R) -BINAP (1:1) at $5.0 \times 10^{-5} \text{ M}$ (path length: 1.0 mm).

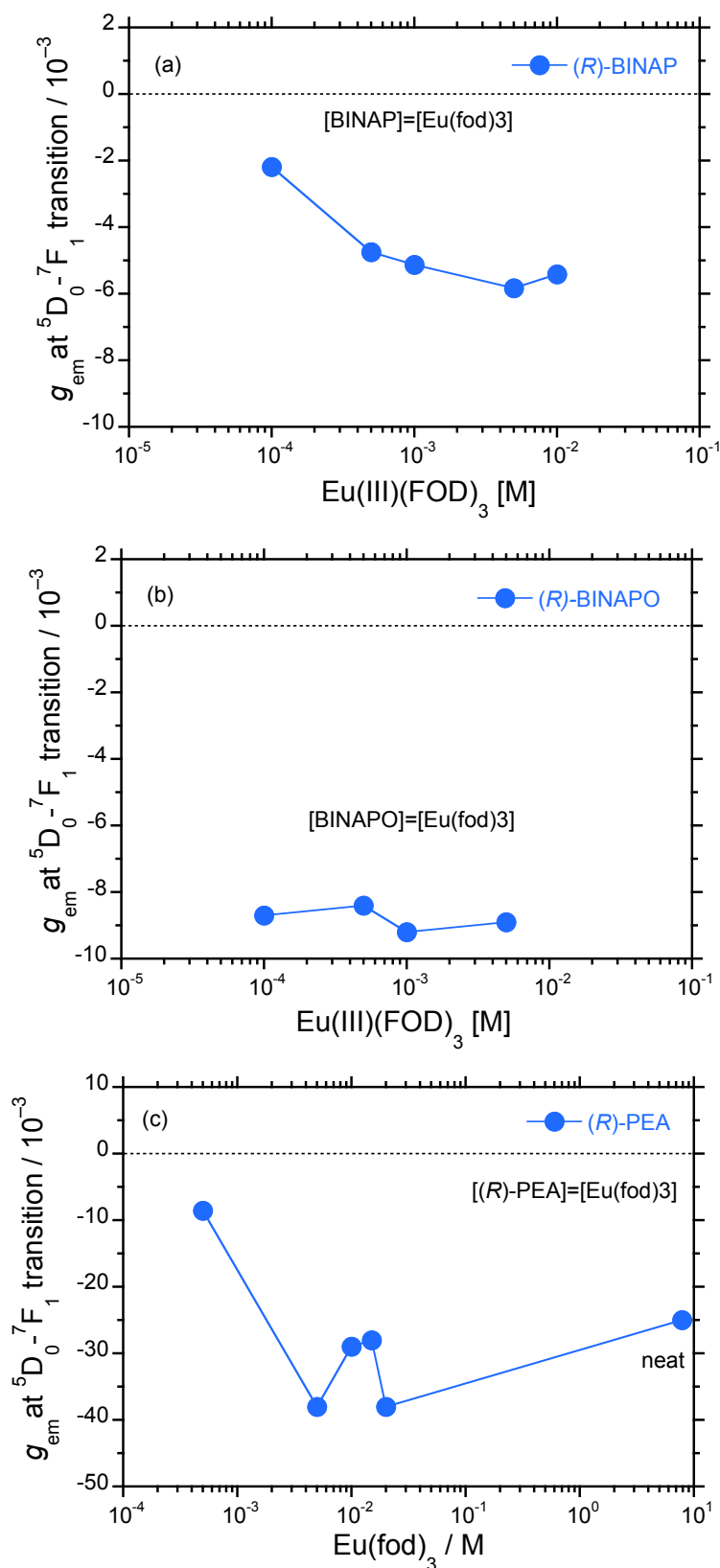


Fig. S10. Variation in the g_{em} values at the $^5D_0 \rightarrow ^7F_1$ transition of $Eu(fod)_3$ as a function of (a) (R) -BINAP (1:1) and (b) (R) -BINAPO (1:1) ranging from $[Eu(fod)_3]_0 = 10^{-4}$ and 10^{-2} M. (c) $Eu(fod)_3$ in (R) -PEA at concentrations ranging from $[Eu(fod)_3]_0 = 5 \times 10^{-4}$ and 7.86 (neat) M.

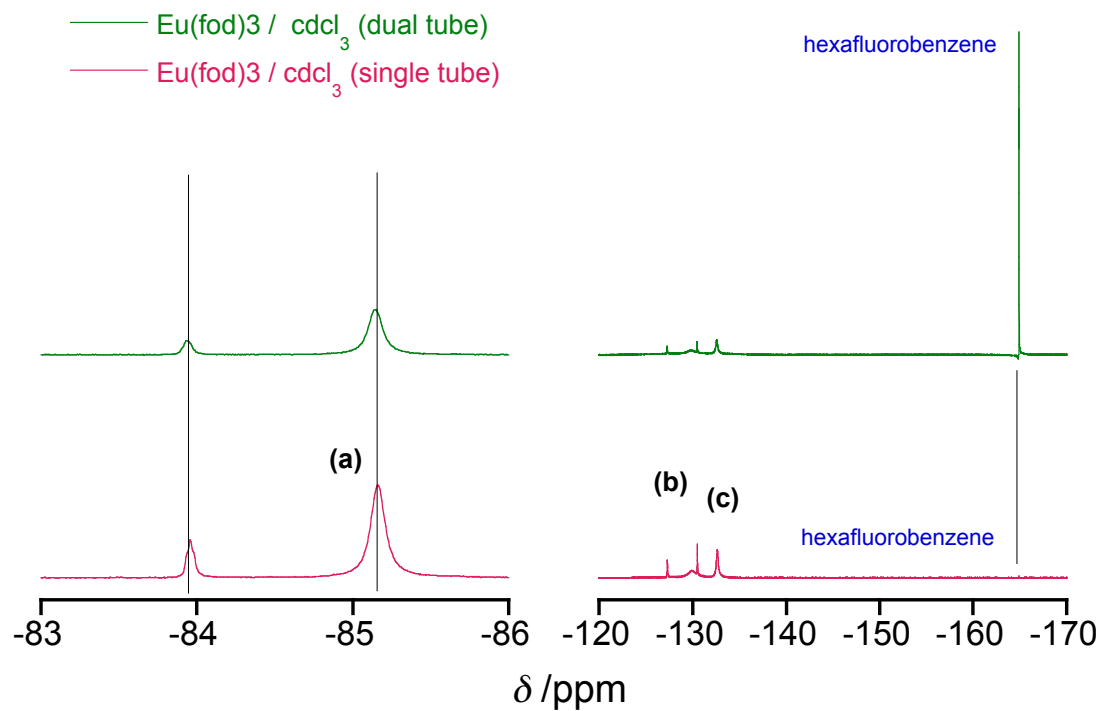


Fig. S11. Comparison of the ^{19}F -NMR spectra of $\text{Eu}(\text{fod})_3$ in CDCl_3 and hexafluorobenzene (HFB) as a reference by using a single NMR tube and using a reference insert tube.

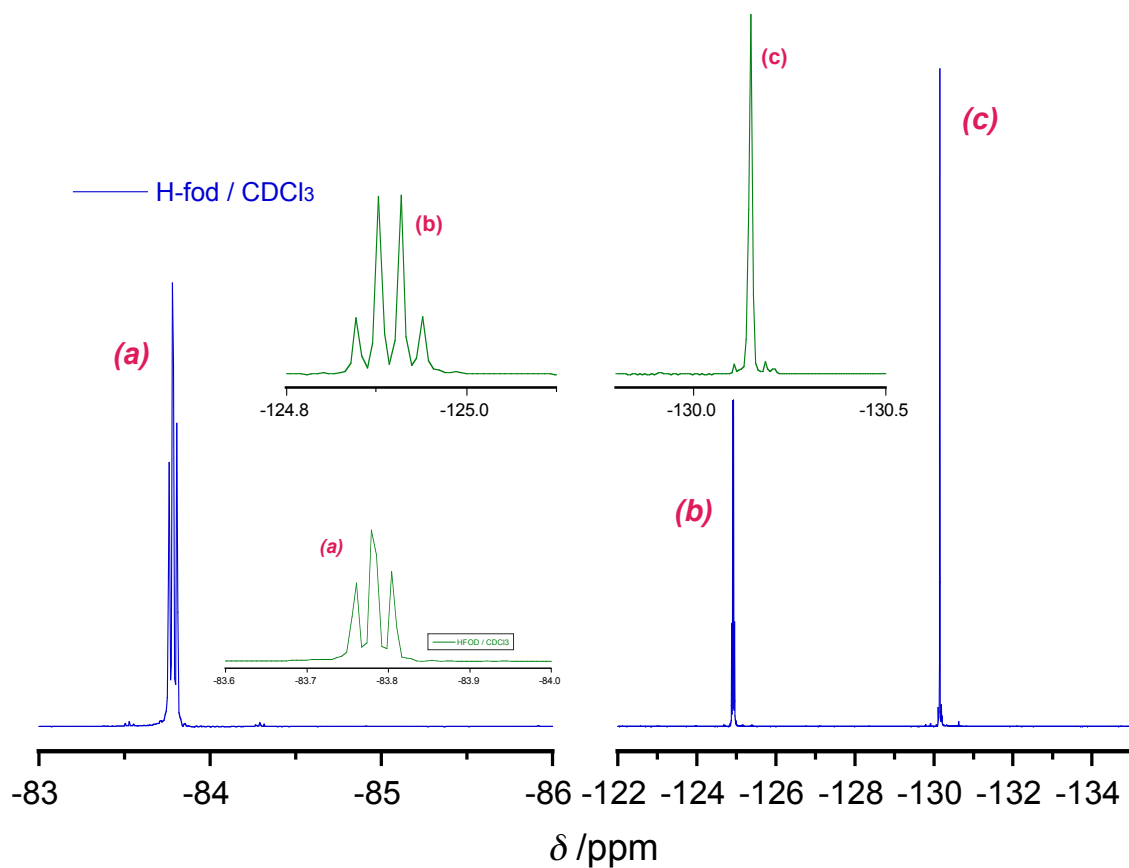


Fig. S12. ^{19}F -NMR spectra and magnified regions of H-fod in CDCl_3 .

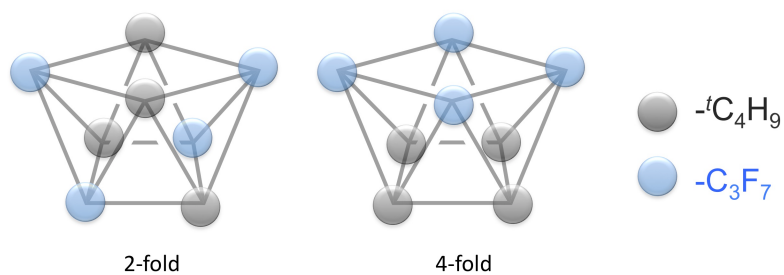
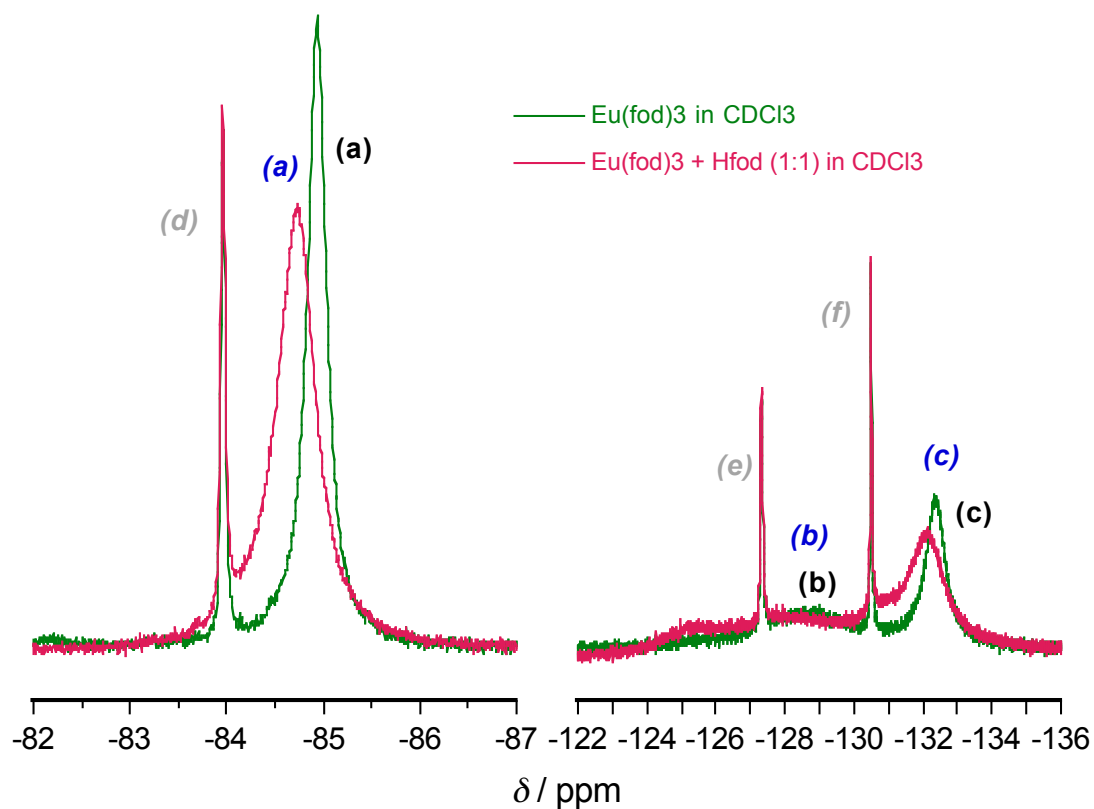


Fig. S13. ^{19}F -NMR spectra of $\text{Eu}(\text{fod})_3$ with H-fod (1:1 molar ratio) in CDCl_3 and proposed square anti-prism (SAP)-like geometries (bottom left; pseudo-2-fold and bottom right; pseudo-4-fold symmetry) of $\text{Eu}^{3+}(\text{fod}^-)_4$ with H^+ .

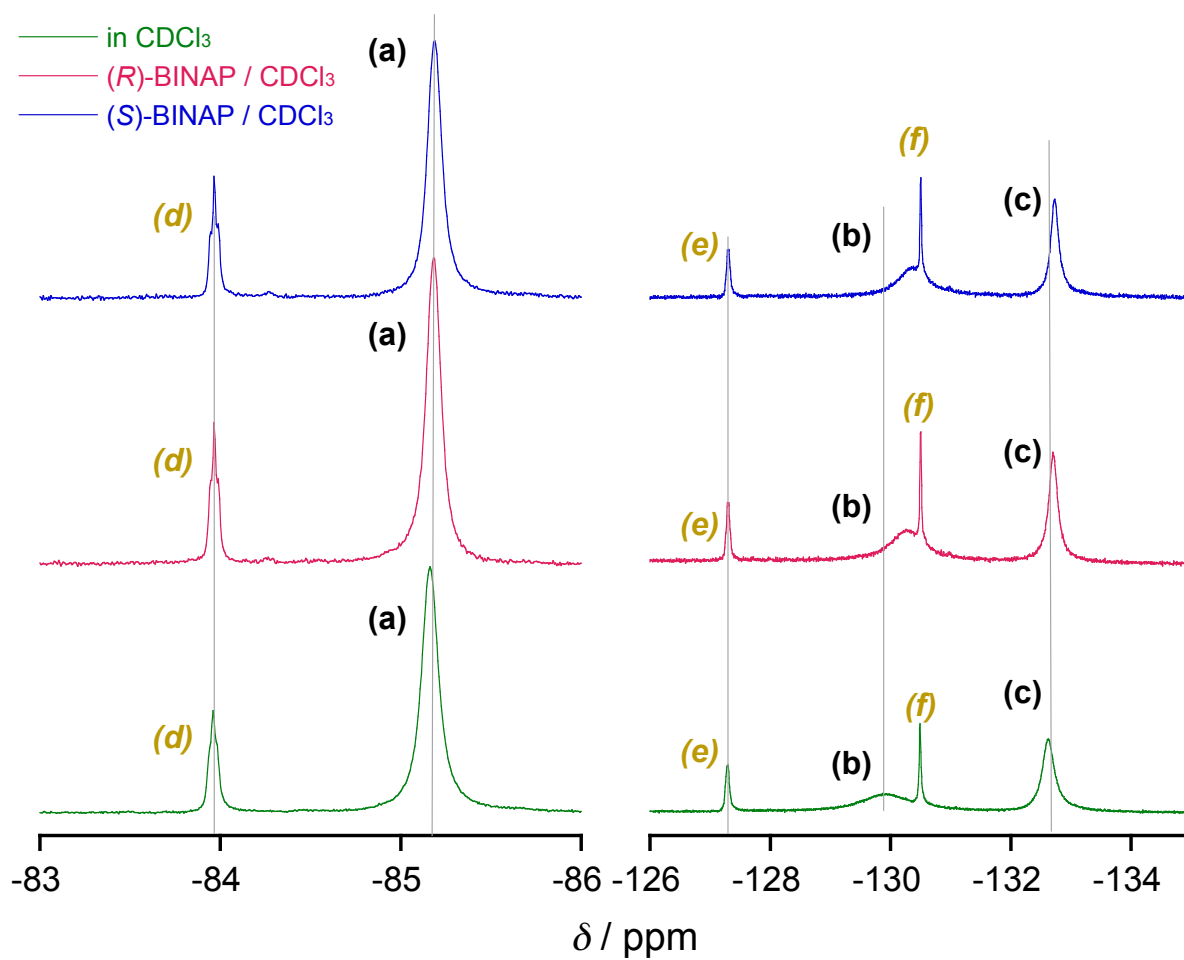


Fig. S14. A comparison of the ^{19}F -NMR spectra of $\text{Eu}(\text{fod})_3$ with (*S*)-BINAP (1:1, green line) and (*R*)-BINAP (1:1, red line) and without BINAP (green line) in CDCl_3 . HFB was used as an internal standard in the same NMR tube.

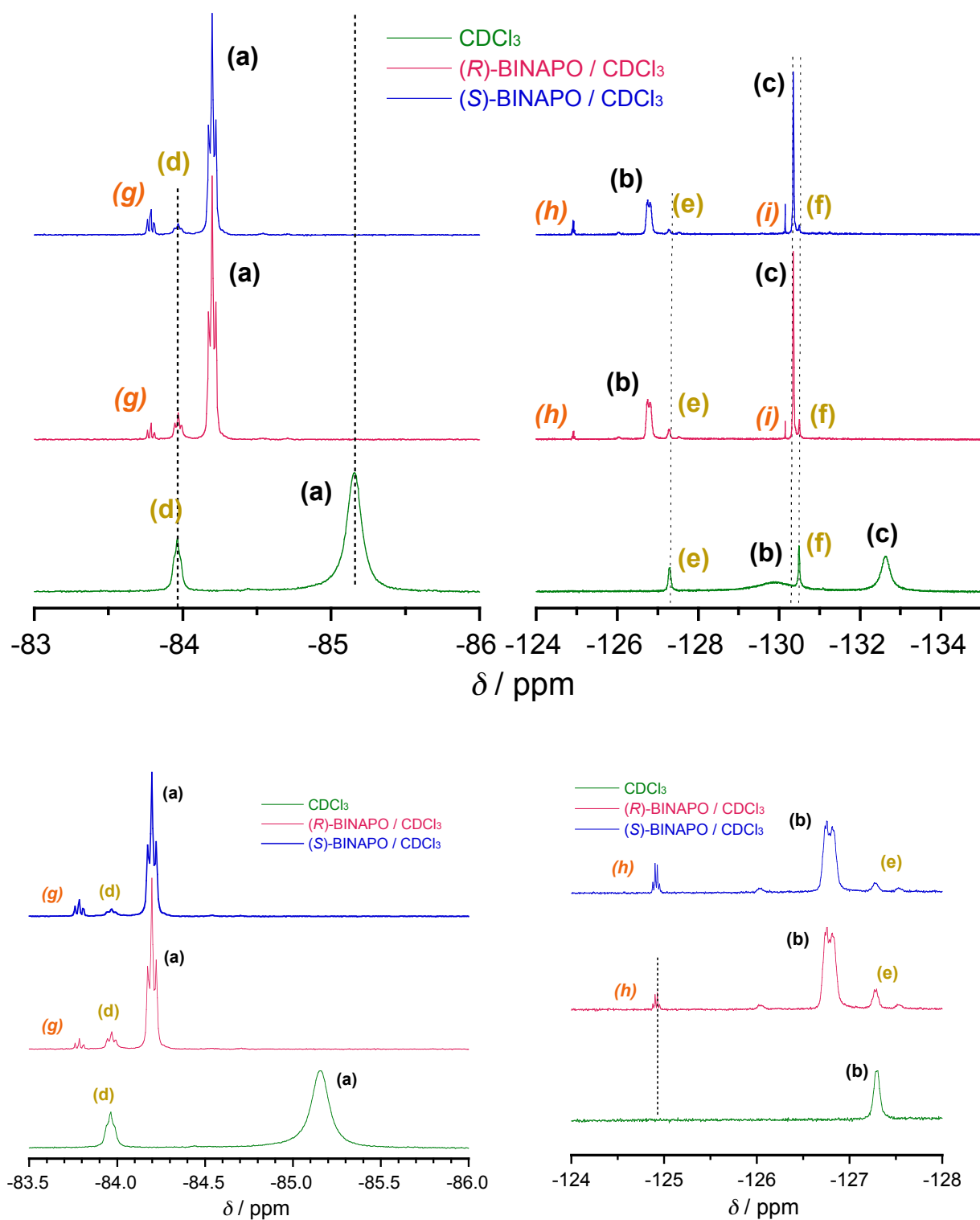


Fig. S15. A comparison of the ^{19}F -NMR spectra and magnified regions of the spectra of $\text{Eu}(\text{fod})_3$ with (S) -BINAPO (1:1, green line) and (R) -BINAP (1:1, red line) and without BINAPO (green line) in CDCl_3 . HFB was used as an internal standard in the same NMR tube.

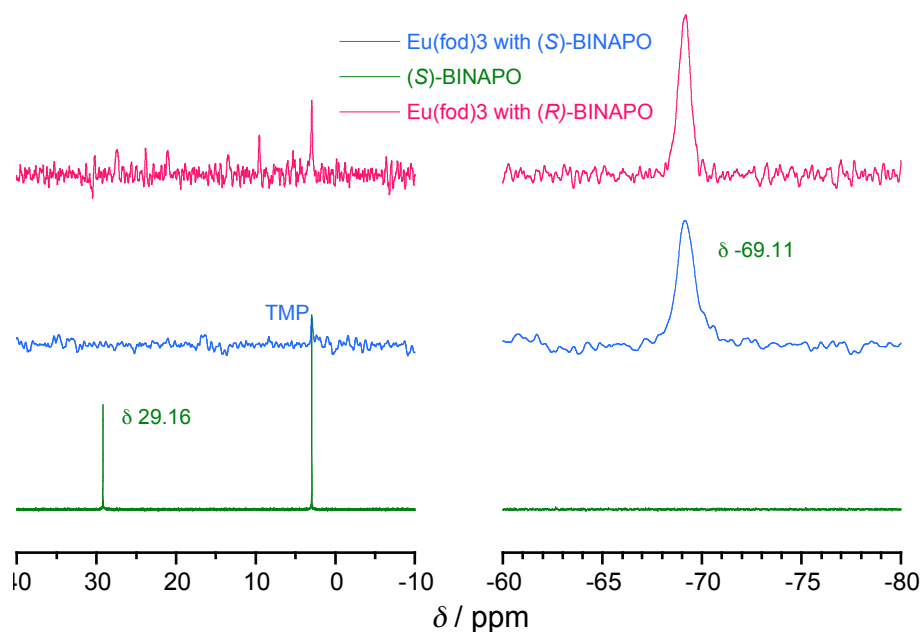


Fig. S16. $^{31}\text{P}\{^1\text{H}\}$ -NMR spectra of (S)-BINAPO (green line), Eu(fod)₃ with (S)-BINAPO (1:1, blue line) and Eu(fod)₃ with (R)-BINAPO (1:1, red line) in CDCl₃. TMP was used as an internal standard in the same NMR tube for (S)-BINAPO and a reference insert was used for samples containing Eu(fod)₃.

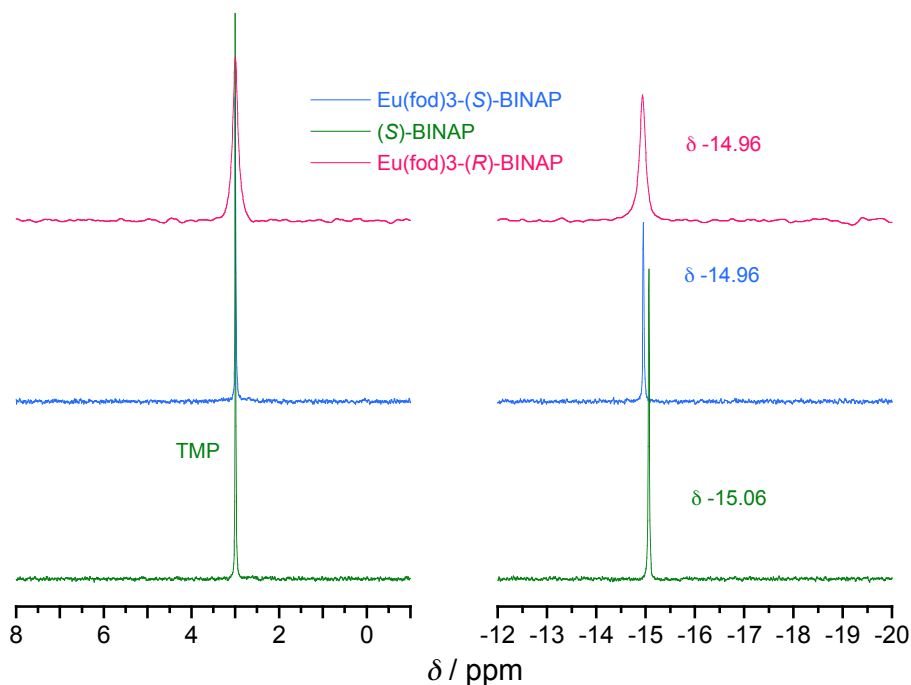


Fig. S17. $^{31}\text{P}\{^1\text{H}\}$ -NMR spectra of (S)-BINAP (green line), Eu(fod)₃ with (S)-BINAP (1:1) in CDCl₃ (blue line) and (R)-BINAP (1:1) in CDCl₃ (red line). TMP was used as an internal standard in the same NMR tube for (S)-BINAP and a reference insert was used for samples containing Eu(fod)₃.

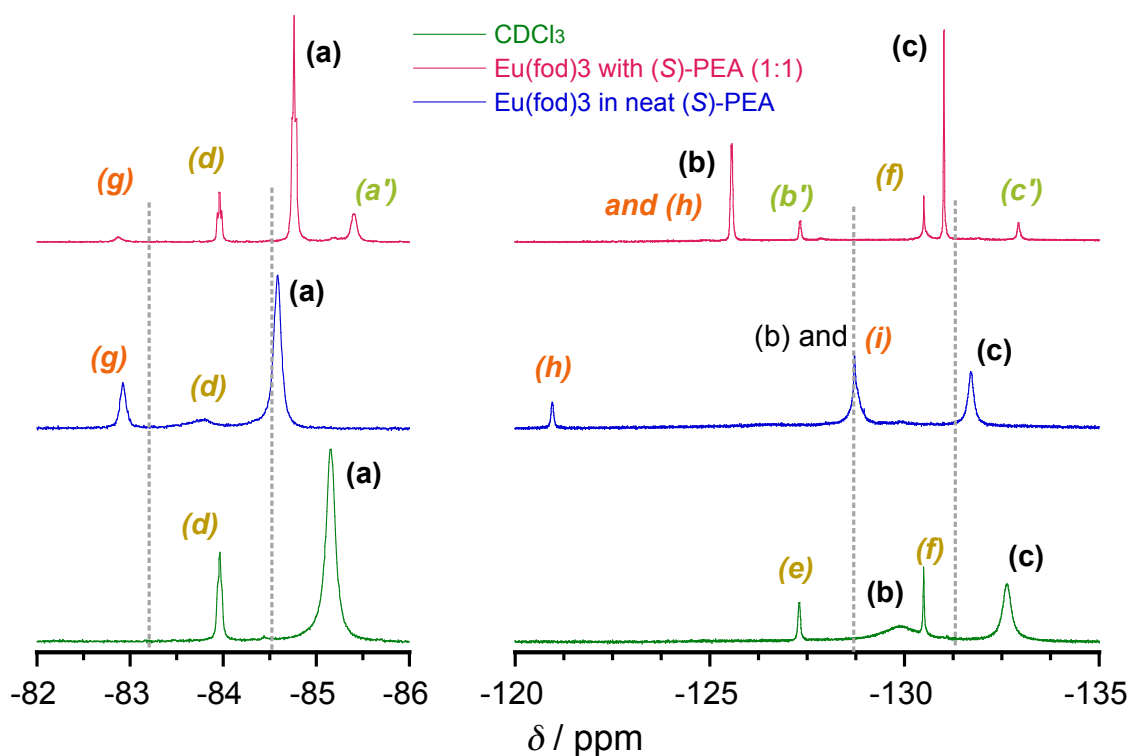


Fig. S18. A comparison of the ^{19}F -NMR spectra and magnified regions of the spectra of $\text{Eu}(\text{fod})_3$ in neat (*S*)-PEA (blue line), with (*R*)-PEA (1:1, red line) in CDCl_3 and in pure CDCl_3 (green line). HFB was used as an external standard in a reference insert for neat solvent spectra, and as an internal standard for the sample in CDCl_3 .

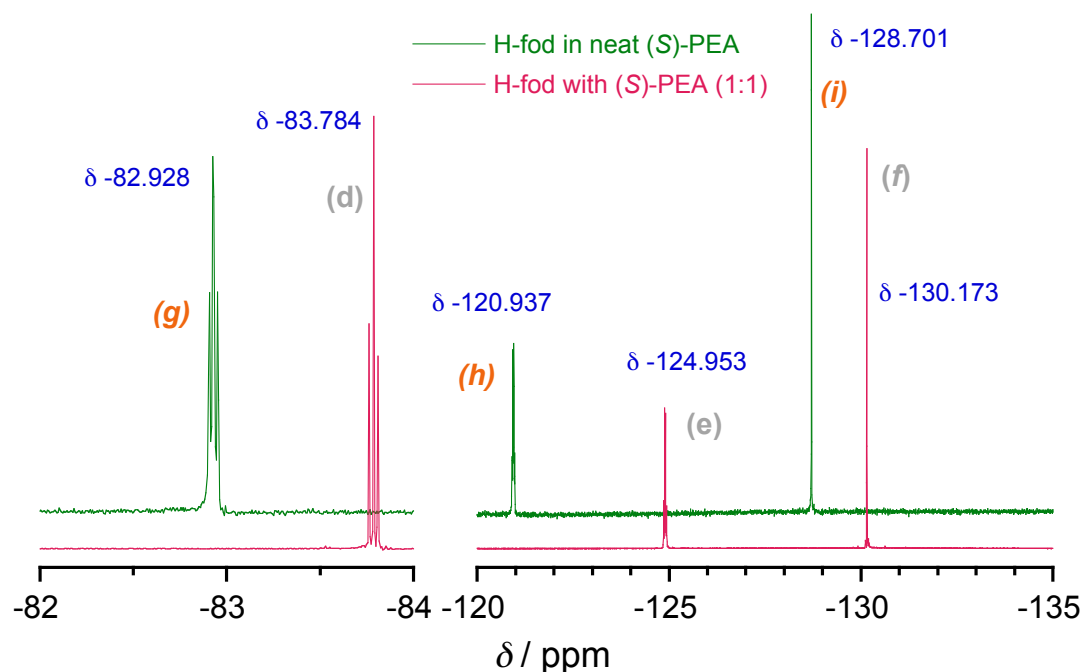


Fig. S19. A comparison of the ^{19}F -NMR spectra of H-fod in neat (*S*)-PEA (green line) and with (*S*)-PEA (1:1, red line) in CDCl_3 . HFB was used as an external standard in a reference insert for neat solvent spectra, and as an internal standard for the sample in CDCl_3 .

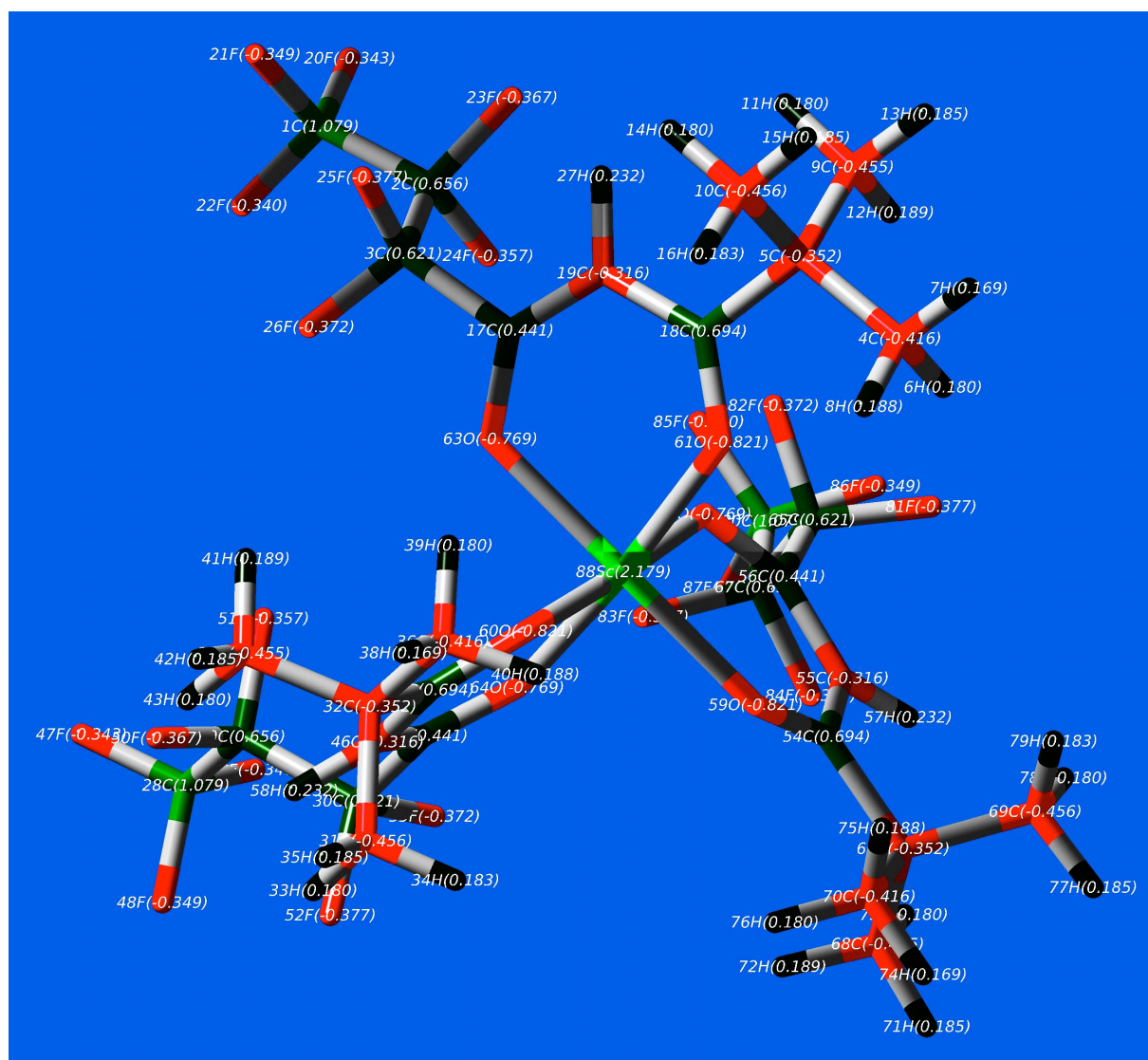


Fig. S20. Mulliken charge distribution of *fac*-Sc(fod)₃ obtained with MP2 (6-311G).

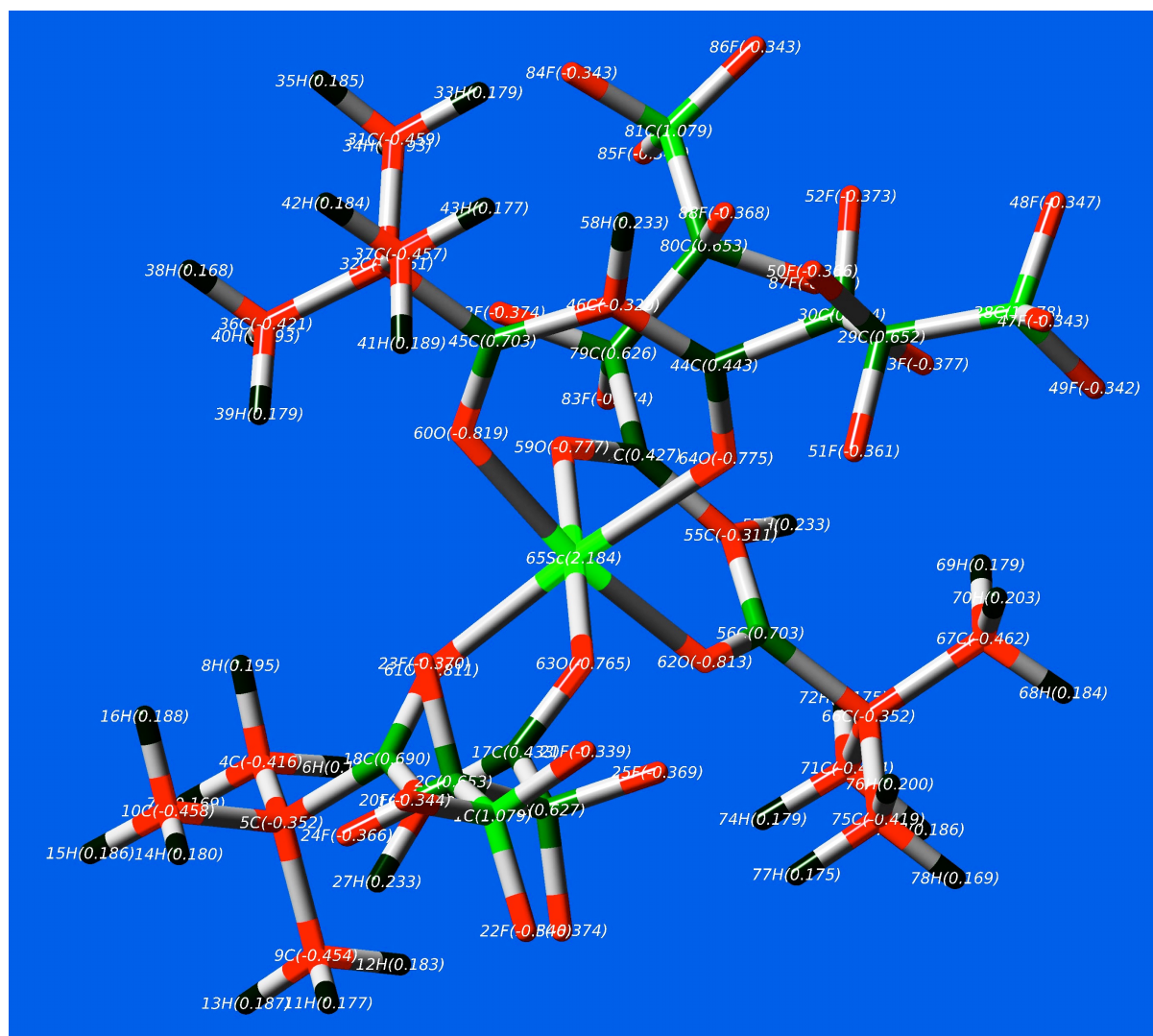


Fig. S21. Mulliken charge distribution of *mer*-Sc(fod)₃ obtained with MP2 (6-311G).

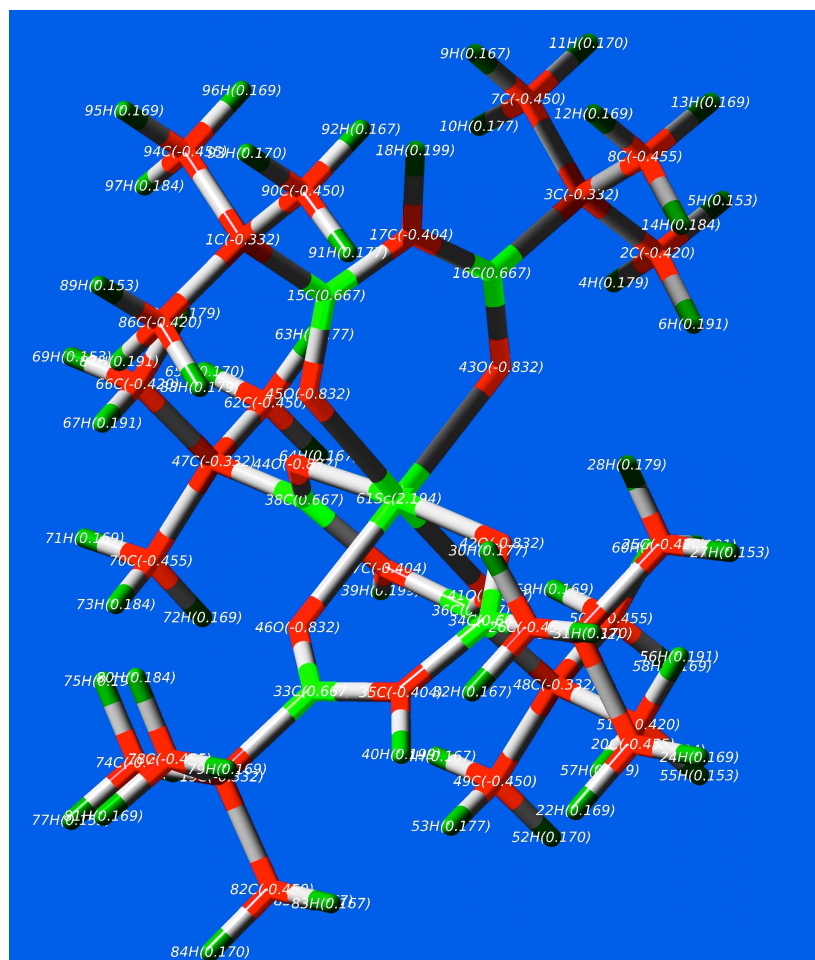


Fig. S22. Mulliken charge distribution of Sc(dpm)₃ obtained with MP2 (6-311G).

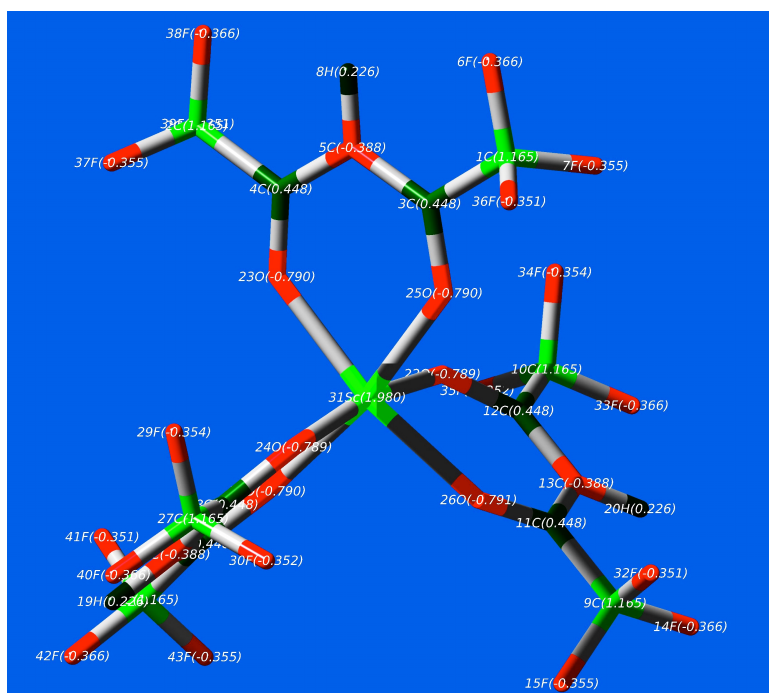


Fig. S23. Mulliken charge distribution of Sc(hfa)₃ obtained with MP2 (6-311G).

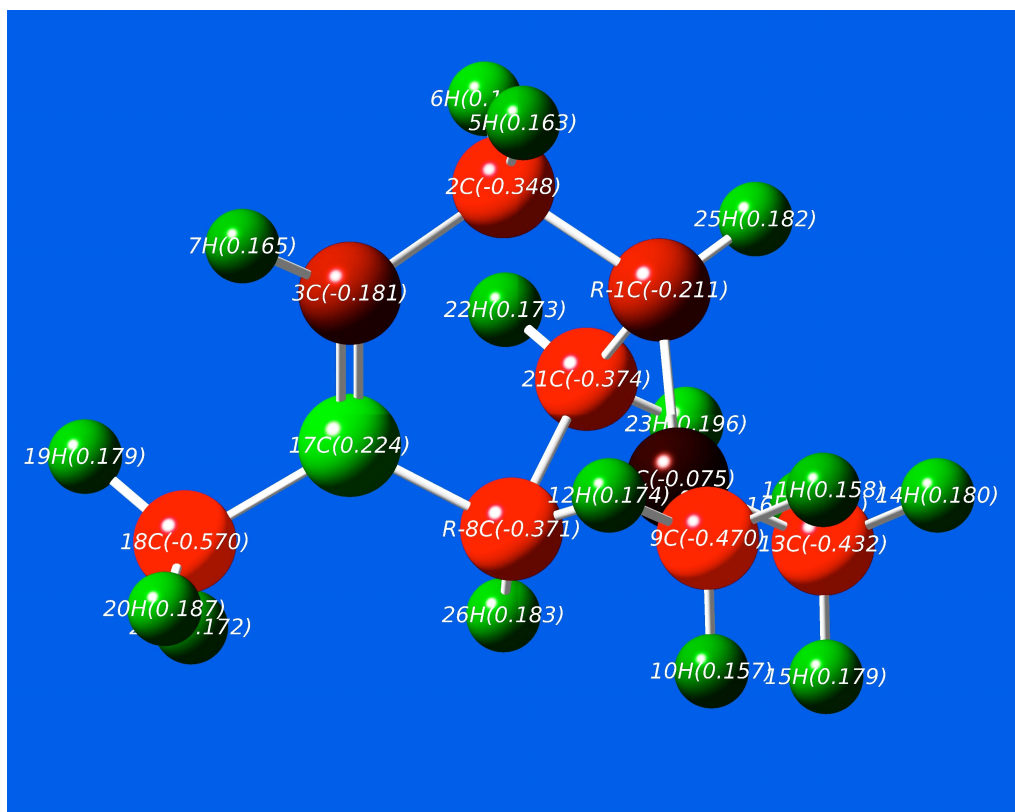


Fig. S24. Mulliken charge distribution of (*R*)-α-pinene obtained with MP2 (6-311G).

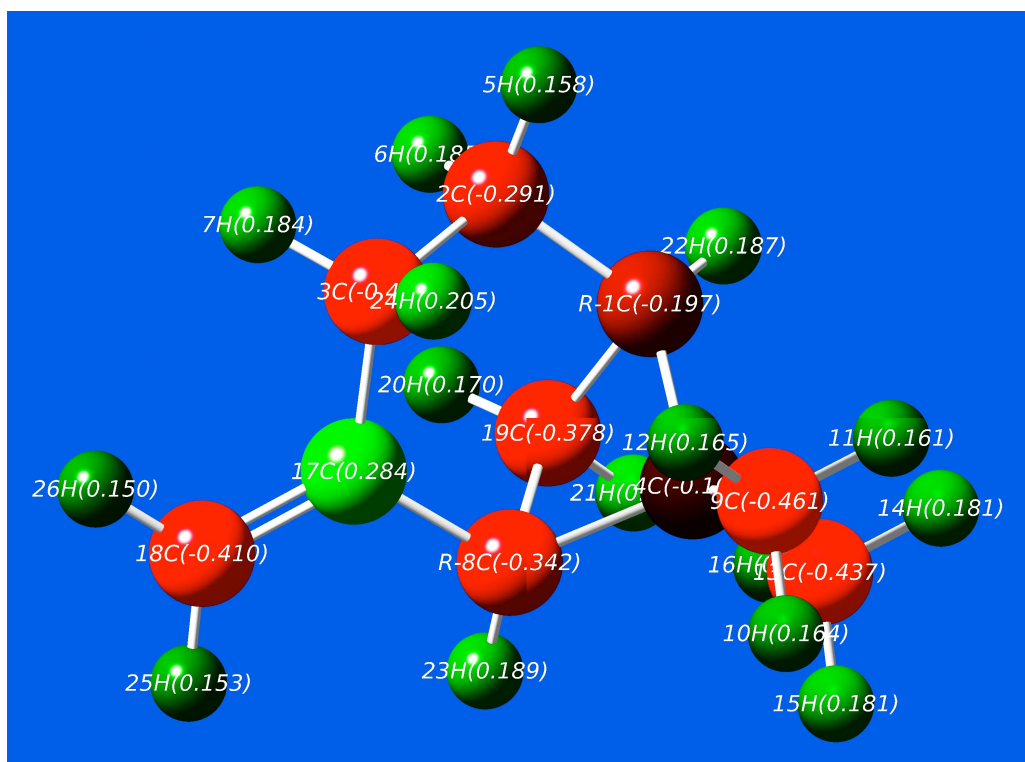


Fig. S25. Mulliken charge distribution of (*R*)-β-pinene obtained with MP2 (6-311G).

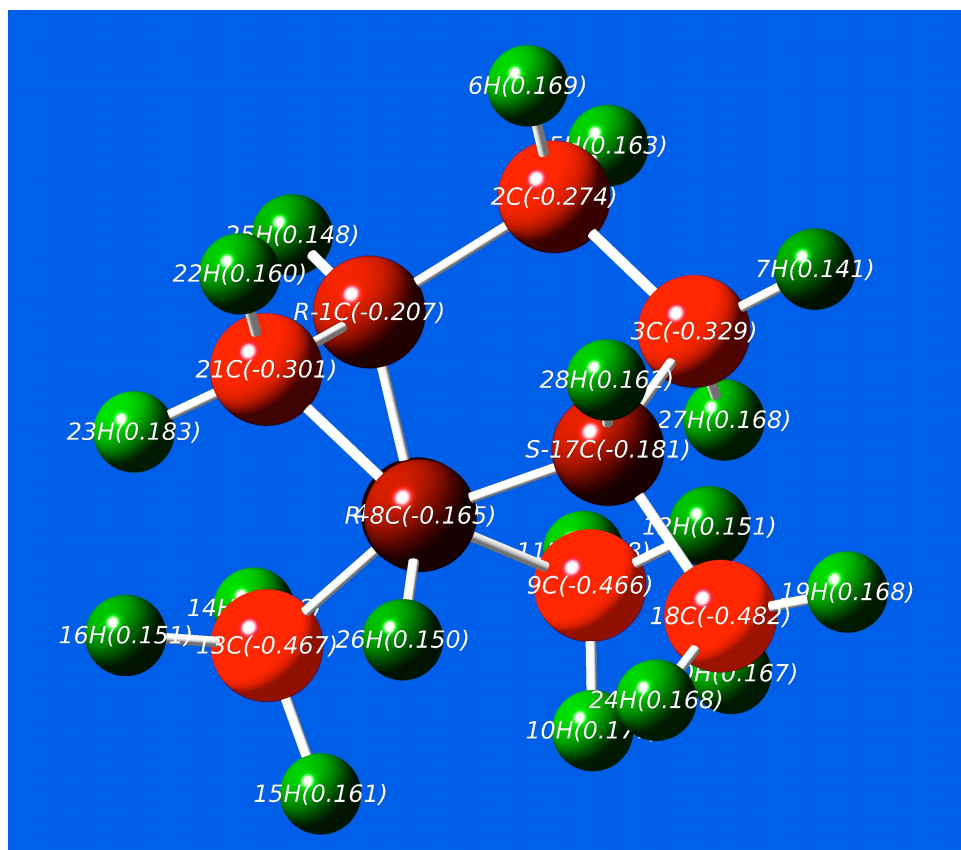


Fig. S26. Mulliken charge distribution of (*R*)-*trans*-pinane obtained with MP2 (6-311G).

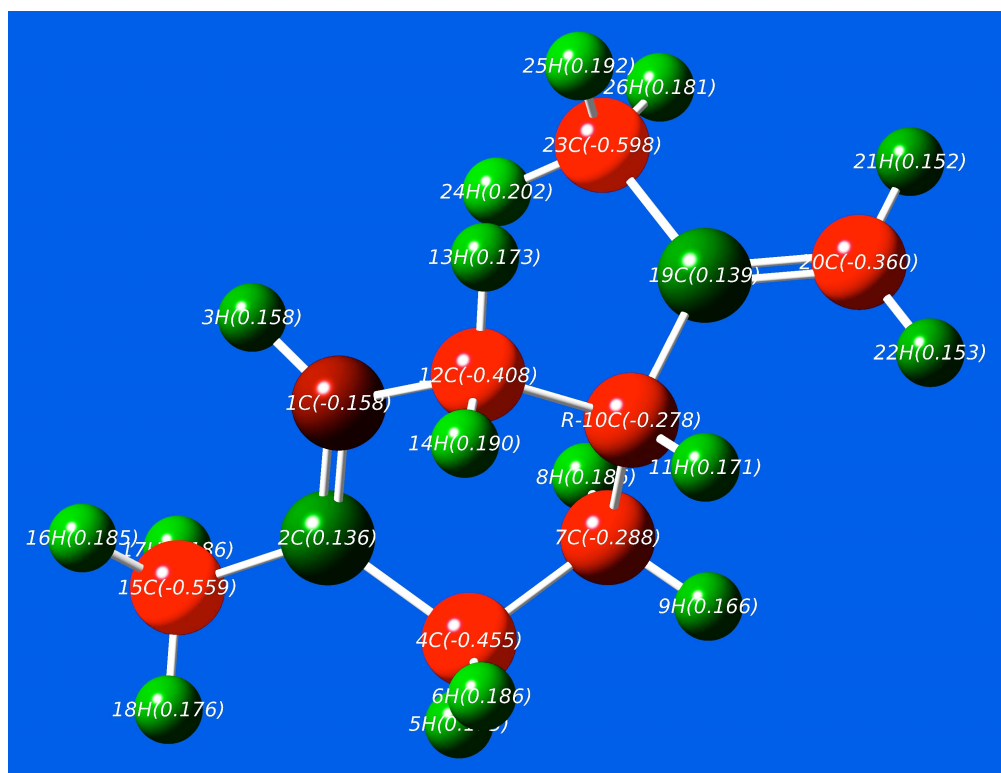


Fig. S27. Mulliken charge distribution of (*R*)-limonene obtained with MP2 (6-311G).

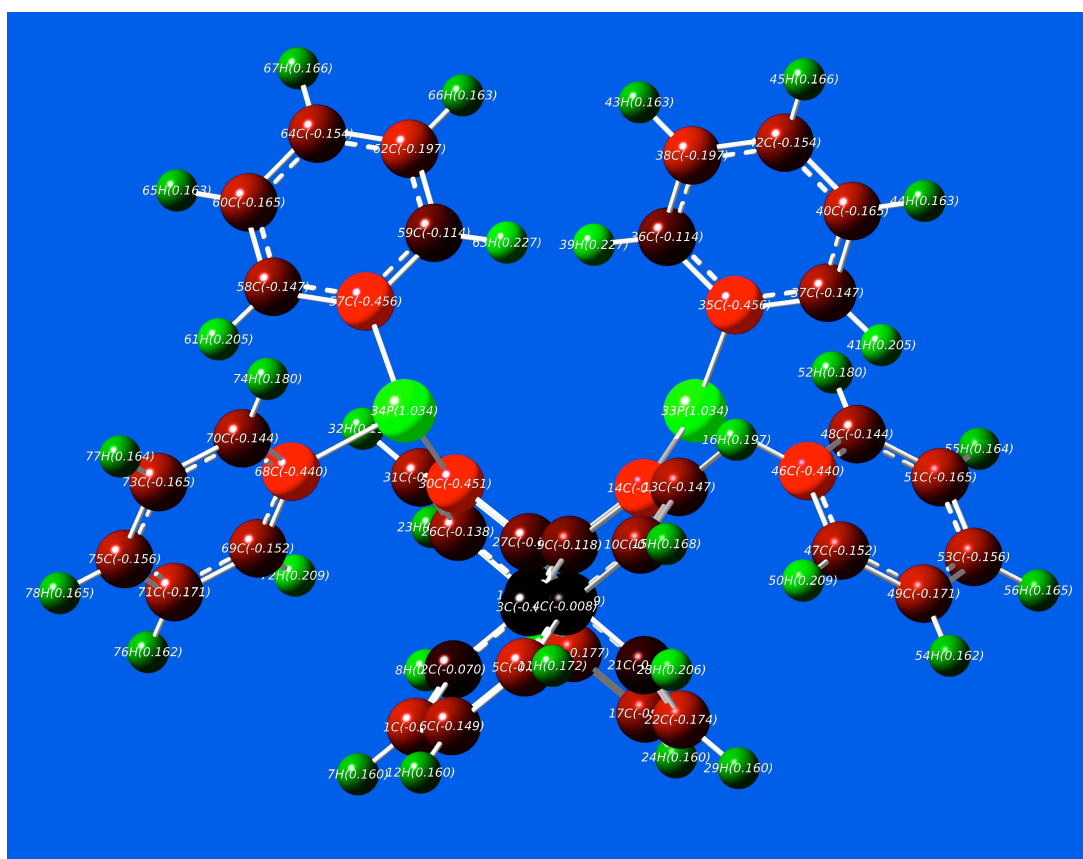


Fig. S28. Mulliken charge distribution of (*R*)-BINAP obtained with MP2 (6-311G).

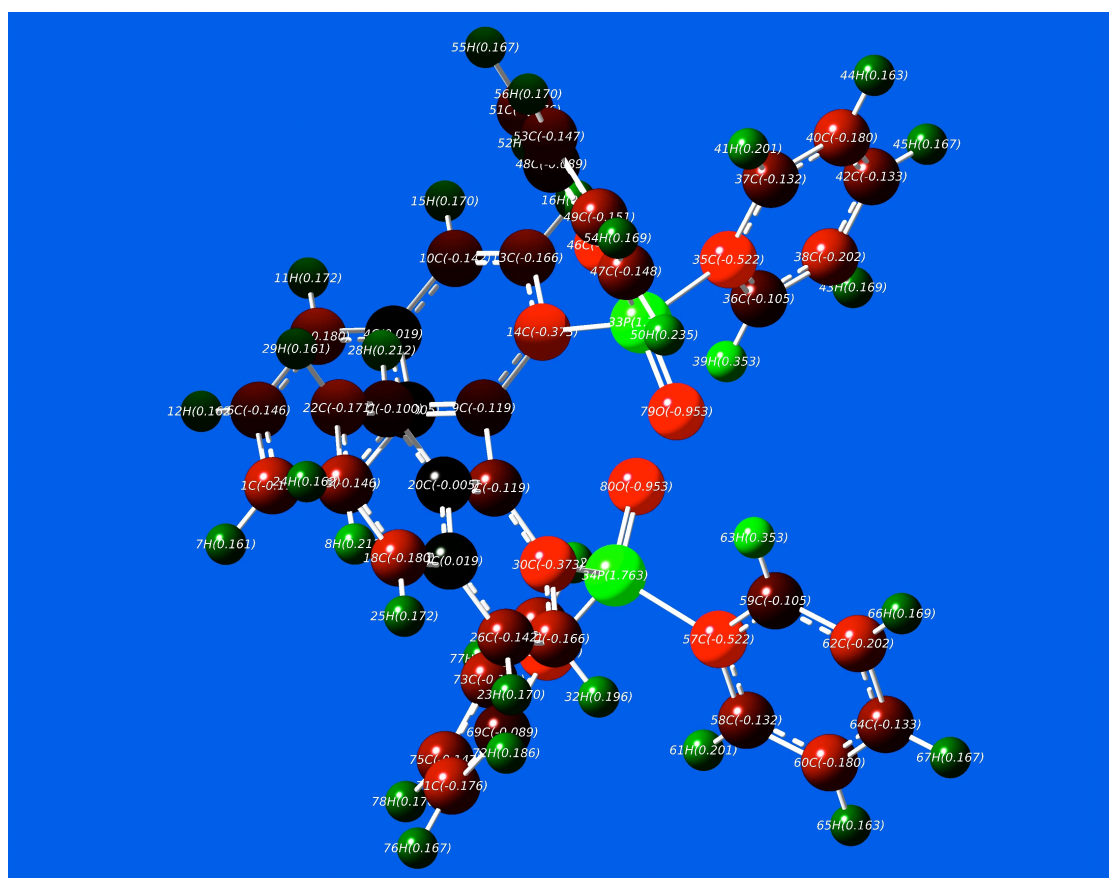


Fig. S29. Mulliken charge distribution of (*R*)-BINAPO obtained with MP2 (6-311G).

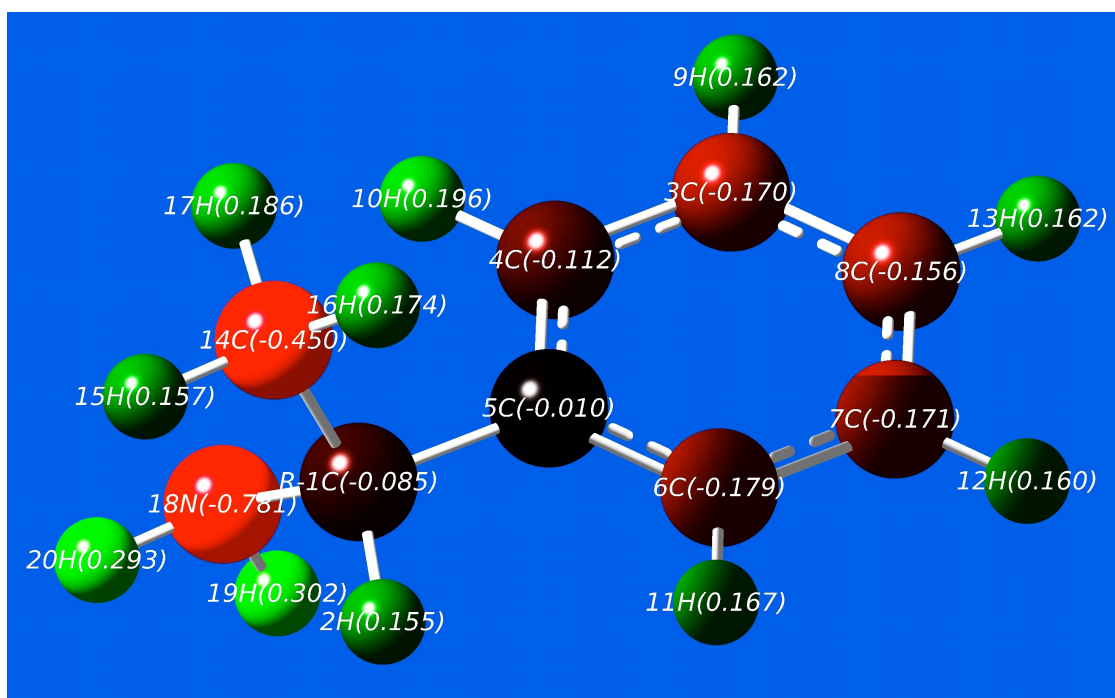


Fig. S30. Mulliken charge distribution of (R)-α-PEA obtained with MP2 (6-311G).

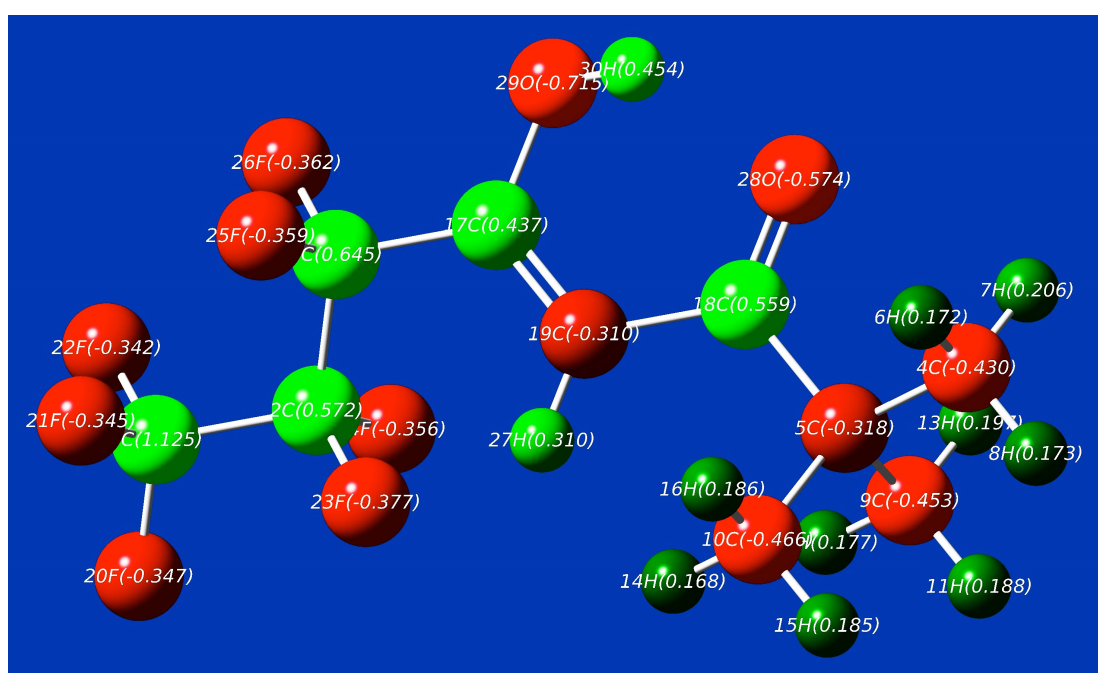


Fig. S31. Mulliken charge distribution of H-fod obtained with MP2 (6-311G).

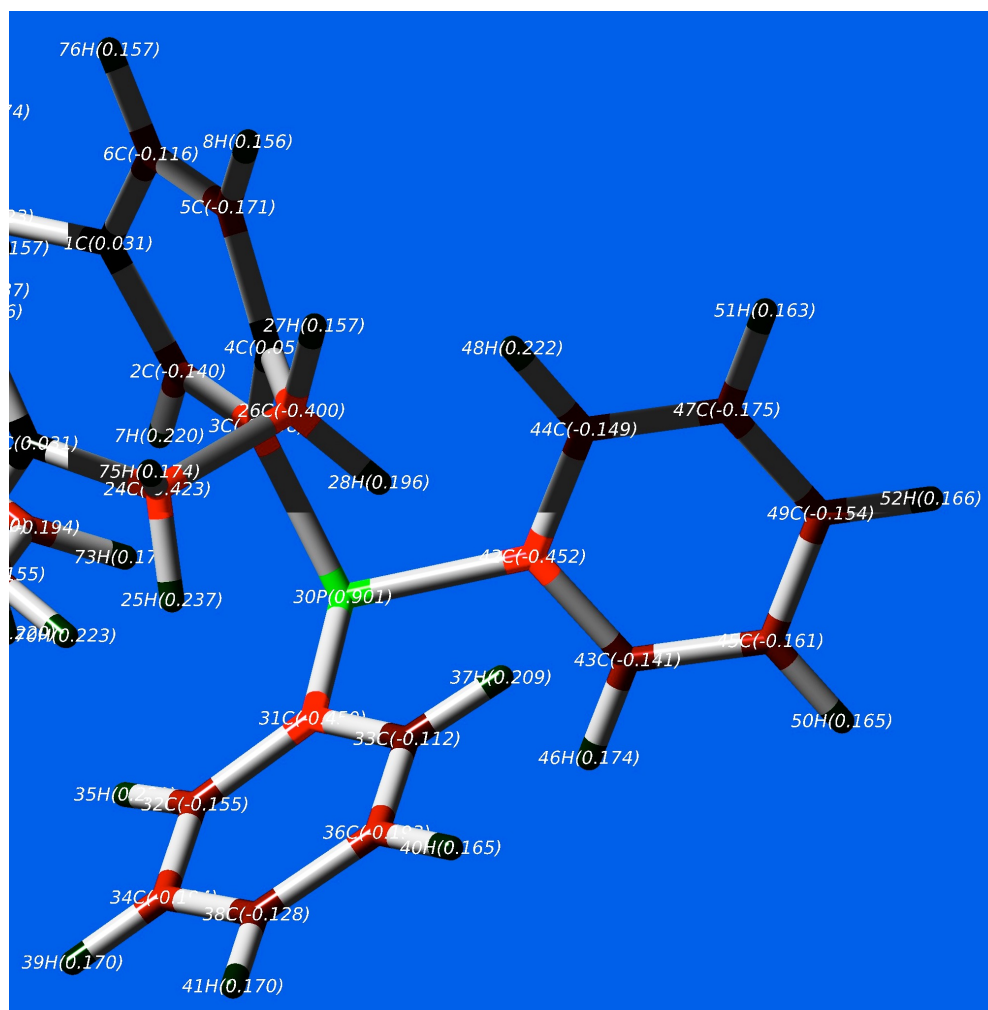
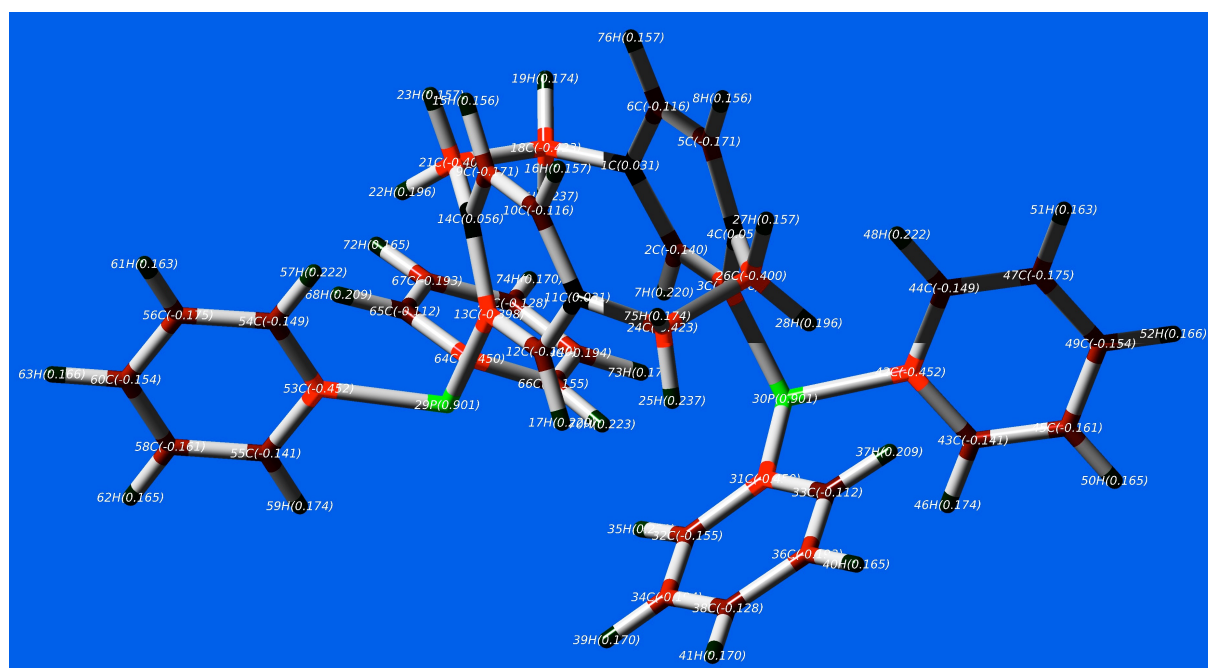


Fig. S32. Mulliken charge distribution of (*R*)-Phanephos obtained with MP2 (6-311G).

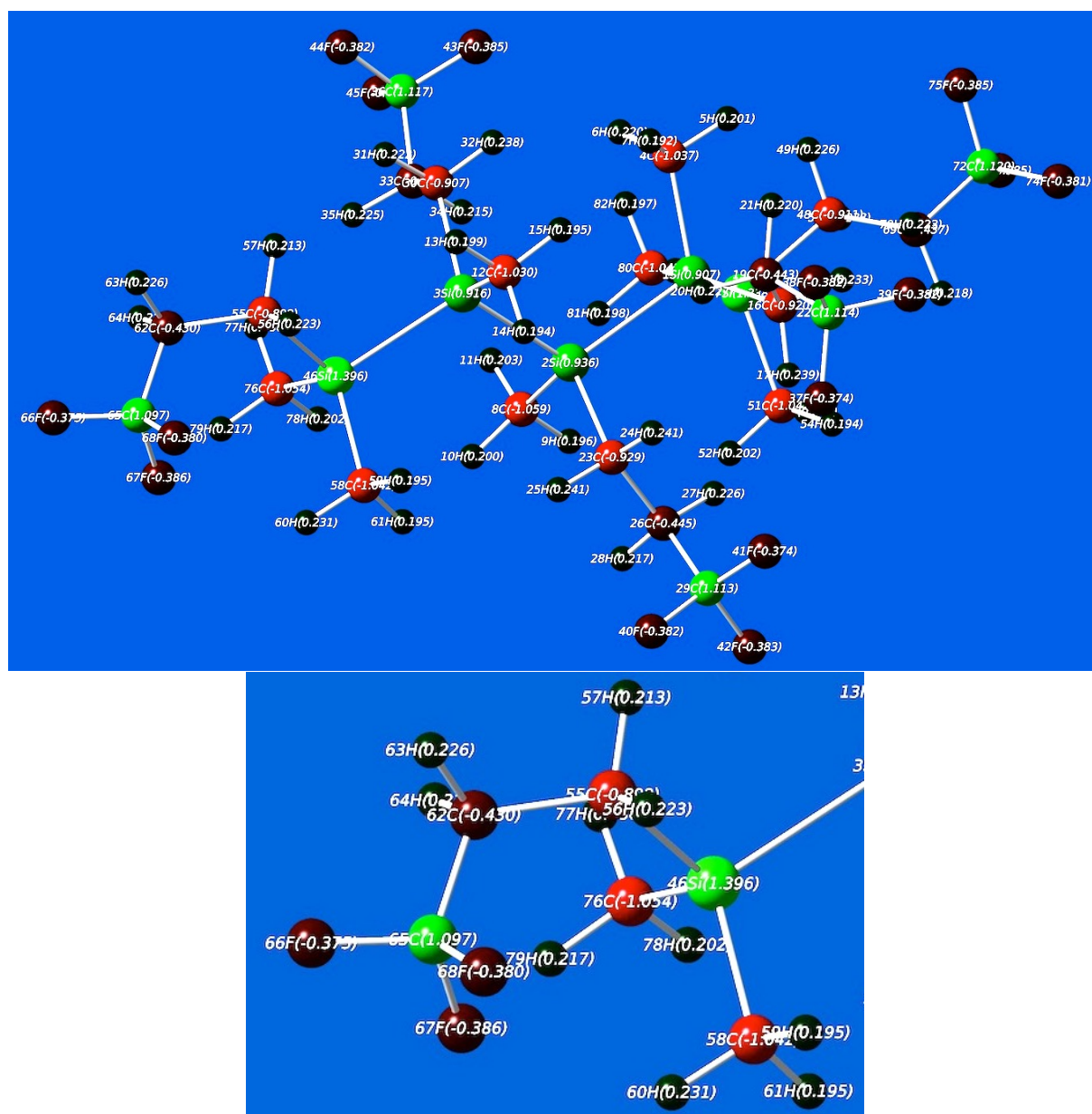


Fig. S33. Mulliken charge distribution of $\text{CH}_3-(\text{CH}_2-\text{Si}-\text{CH}_2\text{CH}_2\text{CF}_3)_5-\text{CH}_3$ obtained with MP2 (6-311G).

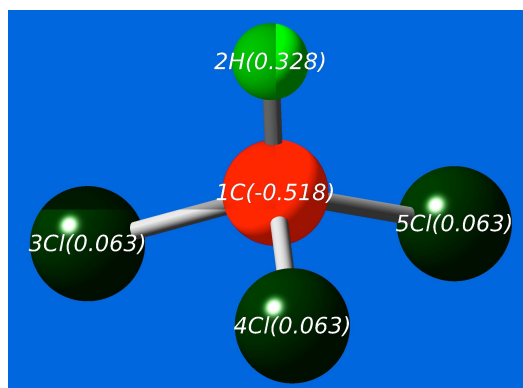


Fig. S34. Mulliken charge distribution of CHCl_3 obtained with MP2 (6-311G).

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Clipped	= FALSE	X_Sweep_Clipped	= 49.01961[kHz]
Scans	= 64	Irr_Domain	= Proton
Total_Scans	= 64	Irr_Freq	= 600.1723[MHz]
Relaxation_Delay	= 4[s]	Irr_Offset	= 5[ppm]
Recvr_Gain	= 17	Clipped	= FALSE
Temp_Get	= 20.6[dC]	Scans	= 64
X_90_Width	= 14[us]	Total_Scans	= 64
X_Acq_Time	= 0.43254[s]	Relaxation_Delay	= 2[s]
X_Angle	= 45[deg]	Recvr_Gain	= 58
X_Pulse	= 7[us]	Temp_Get	= 19.6[dC]
Initial_Wait	= 1[s]	X_90_Width	= 10.9[us]
Phase_Preset	= 3[us]	X_Acq_Time	= 0.53477[s]
Unblank_Time	= 2[us]	X_Angle	= 30[deg]
		X_Atn	= 9[dB]
		X_Pulse	= 3.63333[us]

Fig. S35. Detailed measurement conditions of ^{19}F - and $^{31}\text{P}\{\text{H}\}$ -NMR.

元素分析依頼書				依頼月日: 2月 1日		
研究科名: 物質創成		指導教員: 藤木 道也		印		
講座名: 高分子創成科学		氏名: Jalilah Jalil				
□職員 ■学生(学年: D1) 内線: 6043		E-mail: jalilah.jalil.je8@ms.naist.jp				
試料名・略号: Eu(FOD) ₃ · 1 <i>S</i> -(+)- α -pinene						
分子式: C ₄₀ H ₄₆ EuF ₂₁ O ₆		分子量: 1174.22				
性状	融点: °C	構造式:				
	沸点: °C	Eu				
	分解点: °C	Eu				
	状態: ☐ 液体 ☐ 固体	Eu				
	☐ 吸湿性 ☐ 有毒性 ☐ 昇華性 ☐ 揮発性 ☐ 爆発性 ☐ Arガス雰囲気希望	Eu				
試料全重量: 10 mg		取扱いに関するコメント(測定値の許容範囲、測定回数等):				
理論値 (重量%)		C: 40.93	H: 3.95	N: 0.00		
下欄には記入しないで下さい。						
実測値 (重量%)	C: 39.11	H: 3.56	N: 0.07	重量 (mg)	1.795	
K-factor	C: 14.491	H: 31.677	N: 5.133			
分析月日: 2月 4日 No: 9						
依頼先: 奈良先端科学技術大学院大学 物質創成科学研究科 技官室(場所: E202, 内線: 6174) 渡野間(gasanoma@ms.naist.jp), 片尾(katao@ms.naist.jp)まで						

元素分析依頼書				依頼月日: 2月 1日		
研究科名: 物質創成		指導教員: 藤木 道也		印		
講座名: 高分子創成科学		氏名: Jalilah Jalil				
□職員 ■学生(学年: D1) 内線: 6043		E-mail: jalilah.jalil.je8@ms.naist.jp				
試料名・略号: Eu(FOD) ₃ · 1 <i>R</i> -(+)- α -pinene						
分子式: C ₄₀ H ₄₆ EuF ₂₁ O ₆		分子量: 1174.22				
性状	融点: °C	構造式:				
	沸点: °C	Eu				
	分解点: °C	Eu				
	状態: ☐ 液体 ☐ 固体	Eu				
	☐ 吸湿性 ☐ 有毒性 ☐ 昇華性 ☐ 揮発性 ☐ 爆発性 ☐ Arガス雰囲気希望	Eu				
試料全重量: 10 mg		取扱いに関するコメント(測定値の許容範囲、測定回数等):				
理論値 (重量%)		C: 40.93	H: 3.95	N: 0.00		
下欄には記入しないで下さい。						
実測値 (重量%)	C: 42.05	H: 4.13	N: 0.05	重量 (mg)	2.159	
K-factor	C: 14.491	H: 31.677	N: 5.133			
分析月日: 2月 4日 No: 10						
依頼先: 奈良先端科学技術大学院大学 物質創成科学研究科 技官室(場所: E202, 内線: 6174) 渡野間(gasanoma@ms.naist.jp), 片尾(katao@ms.naist.jp)まで						

Fig. S36. Raw elemental analysis data of the viscous residual oils after removal of volatile α -pinene from Eu(fod)₃ dissolved in (left) (*S*)- α -pinene and (right) (*R*)- α -pinene (samples were placed in a vacuum oven at 60 °C overnight (in Japanese)). The oils suggested that Eu(fod)₃ and α -pinene formed 1:1 adducts.