Chemistry of black leaf films synthesised with rail steels and its influence on low friction mechanism

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Supplementary Information

Figure S1 XRF results: (a) GLP, (b) GBP, (c) BLP, (d) BBP

Measured 2θ	2θ in reference	Assignment
14.7	14.6208	CaSO ₄
20.7	-	-
25.51	25.4645	CaSO ₄
28.31	28.3453	KCl
29.09	29.1521	CaSO ₄
29.51	29.4867	$CaSO_4$
31.05	-	-
31.69	31.8395	$CaSO_4$
40.48	40.5074	KCl
49.02	49.0628	$CaSO_4$
50.15	50.1688	KCl
53.8	53.8096	$CaSO_4$
58.58	58.6402	KCl
66.35	66.3809	KCl
73.67	73.7331	KCl

Table S1 Assignments of XRD peaks measured in BBP

KCl: PDF number 00-041-1476 CaSO₄: PDF number 04-016-3271



Figure S2 peak fittings of the acquired Raman spectra (a) GBP, (b) BBP, (c) Activated charcoal



Figure S3 XPS results of GBP: (a) Thick GBP film, (b) Thin GBP film



Analysis	Main findings		
XRF	• Fe ions are dissolved into LEs and cause the chemical reaction		
RS	 Graphitic carbon covers the surface of BP This structure is often seen in RS spectra of a sugar-derived carbon Iron oxides are underneath the carbon surface 		
XPS	 BP mainly consists of carbon, oxygen, iron and nitrogen Oxidised graphite might exist according to the chemical shifts of C 1s and 0 The thick organic layer is formed on the surface of BBP, covering iron oxid The relatively high concentration of phosphate is detected only in heated B 	O 1s des BP	
FT-IR	• Chelate structure of iron-carboxylate seems to be formed		

Table S2 Main findings in the material analyses