

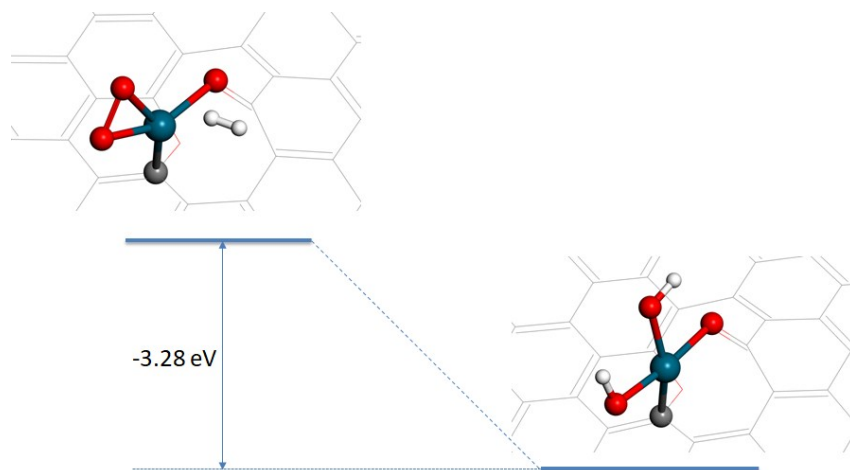
## Supporting Information

**Selective Hydrogenation of 1, 3-Butadiene Catalyzed by A Single Pd Atom**

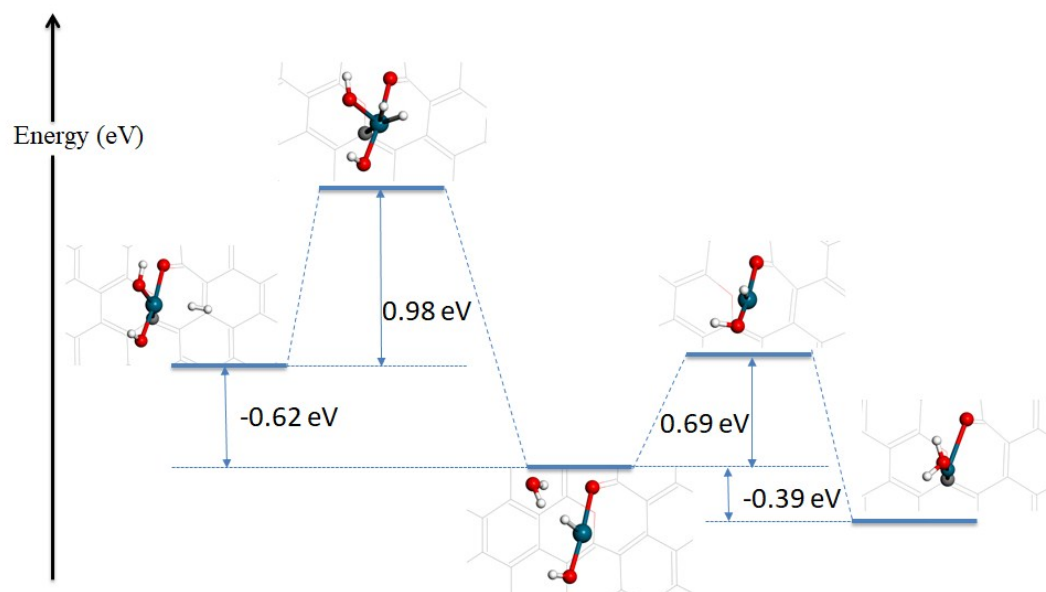
**Anchored on Graphene: The Importance of Dynamics**

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**Figure S1.** Energetics and geometry for the hydrogenation of the oxygen atoms on Pd.



**Figure S2.** Energetics and geometry for the reaction of water formation on Pd.

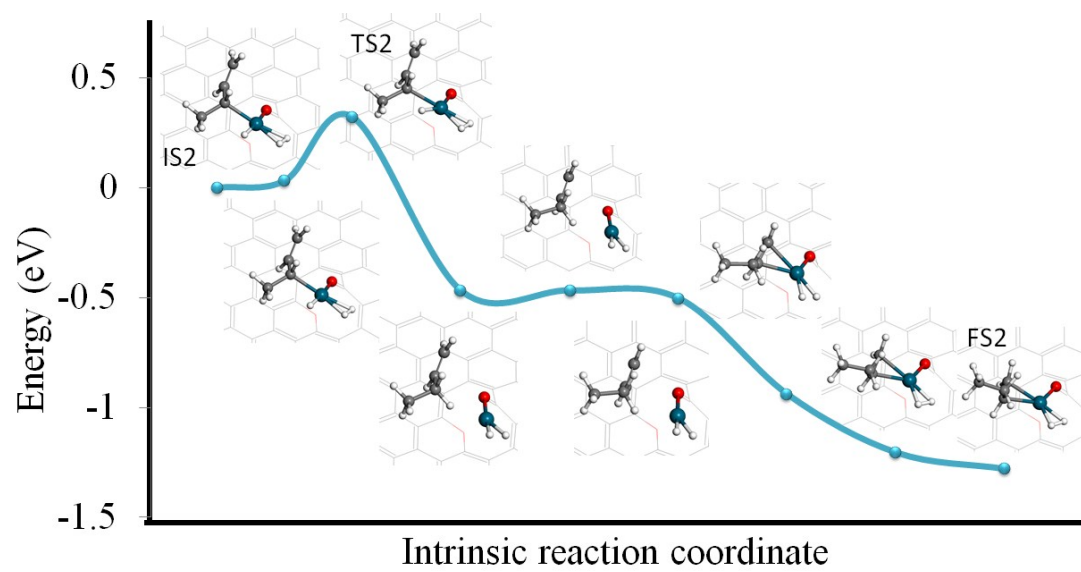


Figure S3. Minimum energy path from TS2 to 1B\* obtained from NEB calculations.

Table S1 The adsorption energies for Pd, 1,3-butadiene and butane with and without Grimme' s semi-empirical scheme. From the test results, it is clear that the Grimme' s semi-empirical scheme is important for the dispersion interaction.

	$E_{\text{Pd}}$	$E_{\text{1,3-butadiene}}$	$E_{\text{butane}}$
With Grimme' s scheme	-1.87	-1.66	-0.56
Without Grimme' s scheme	-1.44	-1.30	-0.08