

Supporting Information

Relative Orientation of the Carbonyl Groups Determines the Nature of Orbital Interactions in Carbonyl-Carbonyl Short Contacts

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Computational Methods

To investigate the nature of carbonyl–carbonyl interactions in the different structural motifs we carried out theoretical calculations using Gaussian 09 suite of quantum chemical program.¹ We used Minnesota functional (M05-2X)² and Møller-Plesset second-order perturbation theory (MP2) exchange correlational functional in conjunction with 6-311+G(2d,p) basis set for the calculations. We carried out NBO second-order perturbation analysis to evaluate the contribution of electron delocalization to the stabilization of carbonyl–carbonyl interactions. AIM analyses were done with Multiwfn³ software. To visualize structure and determine the crystallographic parameters, we used Mercury 3.6⁴ and Chemcraft⁵ graphical visualization software.

CSD analysis

The CSD searches were carried out using CCDC ConQuest⁴ software. Molecules for motif I-V were extracted from the CSD version 5.37 updates (Nov 2015). For motif VI a separate search was done using ConQuest CSD version 5.39 updates (Feb 2018). Only single crystals of organic molecules with R-factor ≤ 0.05 that are error free and without any disorder and polymeric structures were considered in the search. In addition, “only one matching fragment” option was chosen in the CSD to avoid repetition of the same structure. For motif VI, the terminal H positions were normalised from the advanced options.

Criteria for different structural motifs:

The motifs I-VI were categorized based on the $\text{C}=\text{O}\cdots\text{C}=\text{O}$ dihedral angle T, the two $\text{C}=\text{O}\cdots\text{C}$ angles θ_3 and θ_4 and the two molecular planes having the X, C and O atoms. The different parameters for an ideal geometry of a particular motif listed in Table S2. The various structural parameters are shown in Figure S1.

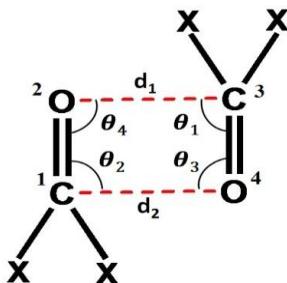


Figure S1: Fragment having $\text{C}=\text{O}\cdots\text{C}=\text{O}$ short contact used for the CSD search. Torsion Angle (T) is defined as $\text{C}^1=\text{O}^2\cdots\text{C}^3=\text{O}^4$.

Table S1: Statistical data for inter- and intramolecular CO…CO interaction. Molecules were divided based on C=O…C=O torsion angle.

Total	Torsion Angle, T(°)			Torsion Angle, T(°)					
	One-sided	0°±45°	90°±45°	180°±45°	Reciprocal	0°±45°	90°±45°	180°±45°	
Inter-molecular	4237	2897 (68.4%)	1458 (50.3%)	1105 (38.2%)	334 (11.5%)	1340 (31.6%)	1311 (97.8%)	29 (2.2%)	0.0 (0.0%)
Intra-molecular	6221	4749 (76.3%)	112 (2.3%)	2674 (56.3%)	1963 (31.6%)	1472 (23.7%)	110 (7.5%)	1352 (91.8%)	10 (0.7%)

Table S2: Structural parameters for an ideal geometry of each motif type.

Motif	d ₁ Å	d ₂ Å	T (°)	θ ₁ (°)	θ ₂ (°)	θ ₃ (°)	θ ₄ (°)
Motif I	≤3.20	>3.20	ND	90	15	75	180
Motif II	≤3.20	≤3.20	0	90	90	90	90
Motif III	≤3.20	>3.20	180	90	50	50	90
Motif IV	≤3.20	>3.20	90	90	70	70	90
Motif V	≤3.20	≤3.20	70	90	90	75	75

A deviation of ±30° of T from the ideal value was considered to club molecules into various structural motif types. ND- not defined as one of the angle is linear.

Table S3. Number of Molecules of each motifs that strictly satisfy the criteria mentioned in Table S2 in intramolecular and intermolecular cases.

	Intramolecular		Intermolecular	
	Overall	≤ 3.0 Å ^a	Overall	≤ 3.0 Å ^a
Motif I	15	5	621	220
Motif II	27	4	1243	133
Motif III	1060	656	200	10
Motif IV	1762	1221	400	14
Motif V	1300	194	21	6
Motif VI	-	-	48	15

^aO…C distances, (d_{O…C}).

Motif I (*T-Shaped*)

For the motif I structures, we considered the molecules with one nonbonded O···C short contacts $\leq 3.20 \text{ \AA}$ and one of the C=O···C angles (θ_3 or θ_4) between 150° - 180° . For motif I, the C=O···C=O dihedral angle is not meaningful as one of the C=O···C angle is linear. For an ideal motif I, the structure will have (θ_3, θ_4) or $(\theta_4, \theta_3) = 180^\circ, 75^\circ$ or $75^\circ, 180^\circ$, which is reflected in the θ_3 versus θ_4 shown in Figure S2.

Motif II (*Antiparallel*)

For the antiparallel motif II, we considered $T = 0^\circ \pm 20^\circ$, θ_3 and $\theta_4 \sim 90^\circ$ and both O···C contacts $\leq 3.20 \text{ \AA}$. For molecules with $T = 0^\circ \pm 20^\circ$, the plot of θ_3 versus θ_4 is shown in Figure 3. The deviation of the θ_3, θ_4 from the ideal $90^\circ, 90^\circ$ values indicate sheared antiparallel structures.

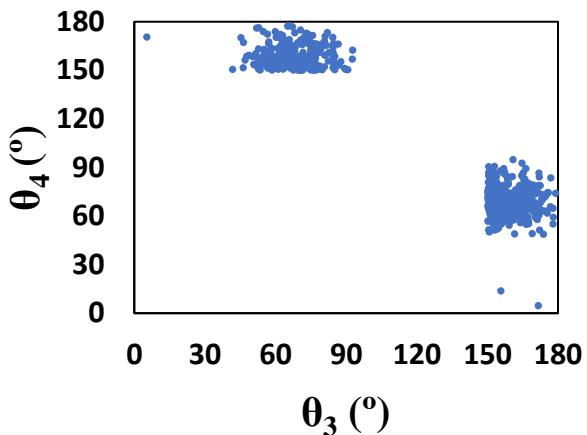


Figure S2: Plot of θ_3 versus θ_4 for molecules that fall under structural motif I.

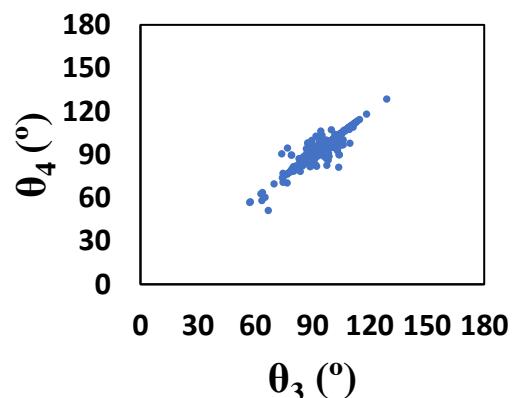


Figure S3: Plot of θ_3 versus θ_4 for molecules with $T = 0^\circ \pm 20^\circ$.

Motif III (*Parallel*)

For the parallel structure (motif III), one nonbonded O···C $\leq 3.20 \text{ \AA}$ and $|T|$ lie within the range of $180^\circ \pm 20^\circ$. The molecules that contain C=O···C angle within 150° - 180° were excluded (motif I). For $|T| = 180^\circ \pm 20^\circ$, $(\theta_3, \theta_4) \sim (50^\circ, 90^\circ)$ or $(\theta_3, \theta_4) \sim (90^\circ, 50^\circ)$ [Figure S4, points within the circles]. A few of the slipped parallel structures were also observed wherein θ_3 and θ_4 deviated slightly from the perfect geometry may be due steric hindrance or packing forces.

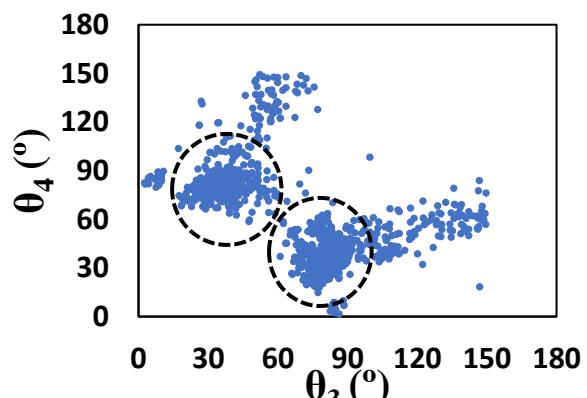


Figure S4: Plot of θ_3 versus θ_4 for molecules with $T = 180^\circ \pm 20^\circ$.

Motif IV (*L-Shaped*)

For ‘*L-shaped*’ motif IV, one of the nonbonded O···C is $\leq 3.20 \text{ \AA}$ and $|T|$ lies within the range of $90^\circ \pm 20^\circ$. The molecules with C=O···C angle 150° - 180° were excluded from this motif as they fall in the motif I category. The θ_3 and θ_4 values of molecules that fall in the structural motif IV category are shown in Figure S5.

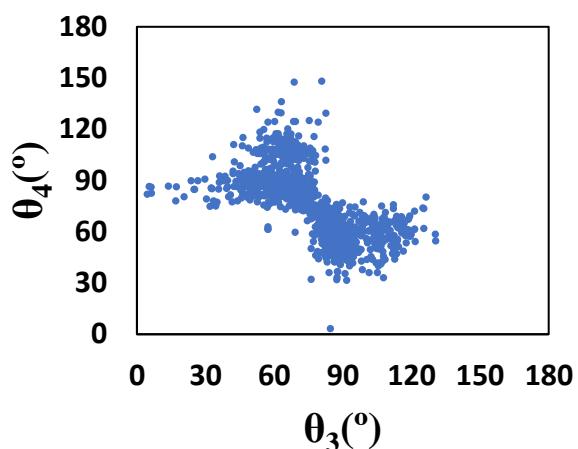


Figure S5: Plot of θ_3 versus θ_4 for molecules with $T = 90^\circ \pm 20^\circ$.

Motif V (*Reciprocal*)

For motif V, both the non-bonded O···C distances are $\leq 3.20 \text{ \AA}$ and $|T|$ that lies within the range of 50° - 100° . For $|T| = 50^\circ$ - 100° , the θ_3 and θ_4 values will both be 75° in an ideal motif V geometry. The plot of θ_3 and θ_4 values for the molecules having motif V geometry is shown in Figure S6.

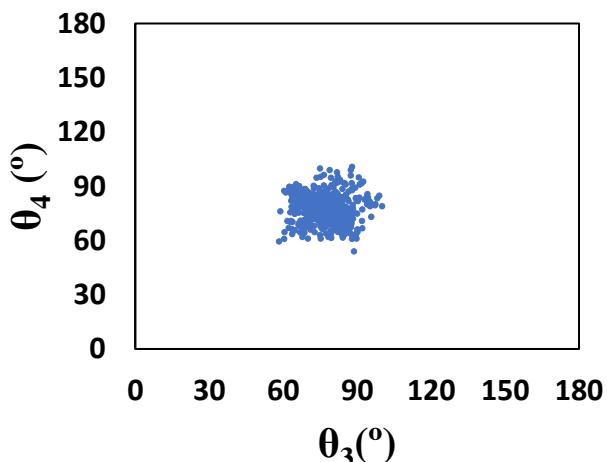


Figure S6: Plot of θ_3 versus θ_4 for molecules with $T = 50^\circ$ - 100°

Motif VI: An Unprecedented structural motif

To find unique geometries that resemble the formaldehyde dimer, we carried out three independent CSD search using motifs shown in Figure S7. For the aldehydic, carboxylic and amidic motifs, two planes were considered. Planes having atoms H-C¹=O² [motif VI(a)], O-C¹=O² [motif VI(b)], N-C¹=O² [motif VI(c)] were taken as plane 1 in each case and the plane having atoms X-C³=O⁴ is taken as plane 2. The angle between plane 1 and plane 2 is restricted to 75° - 90° during the CSD search (for ideal motif VI angle between plane 1 and 2 should be 90°). We also restricted the distances d_1 and d_3 to $d_1 \leq 3.2 \text{ \AA}$ and $d_3 \leq 3.0 \text{ \AA}$ during the search (Figure S7).

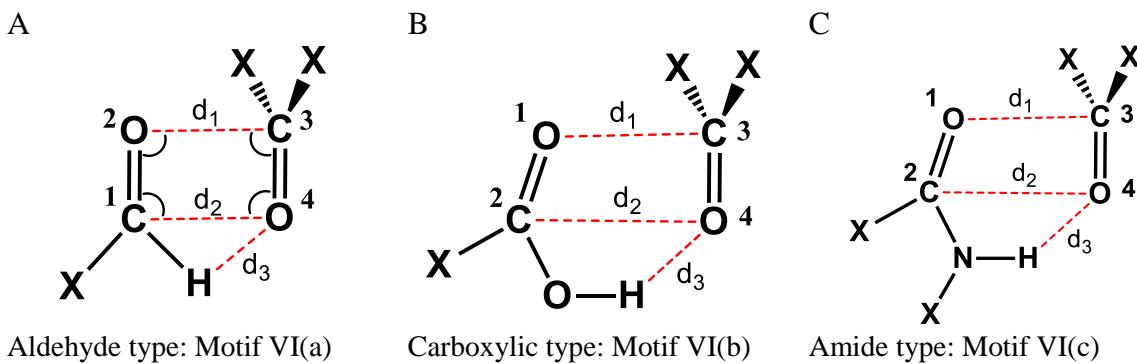


Figure S7: Fragment used search for motif VI in the CSD. Torsion Angle, T is defined as $C^1=O^2\cdots C^3=O^4$.

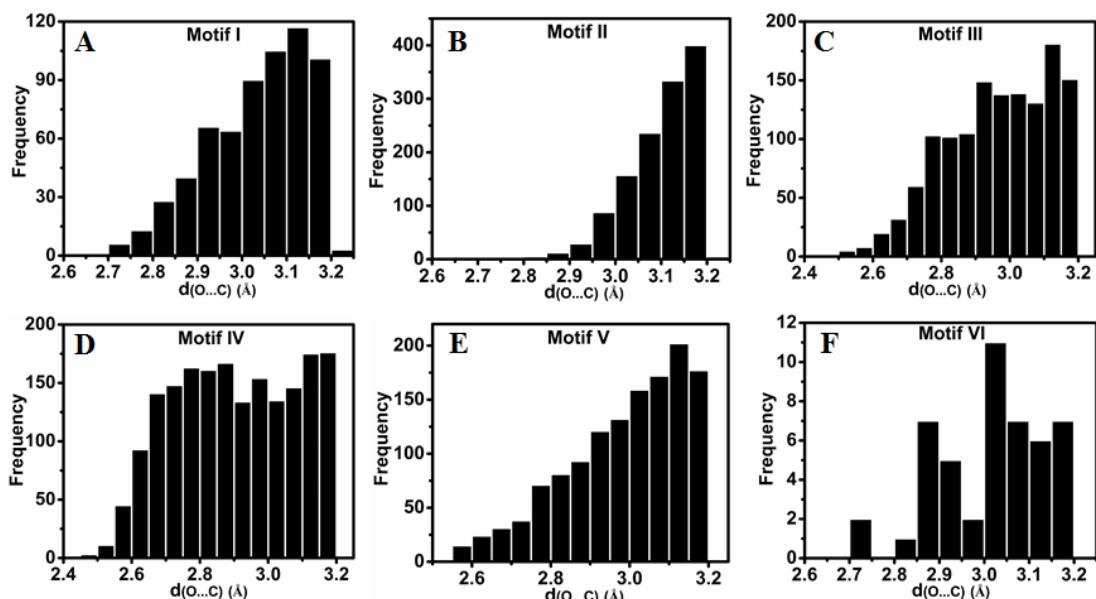


Figure S8: Frequency distribution of O...C distances ($d_{O\cdots C}$) for different structural motifs (I-VI).

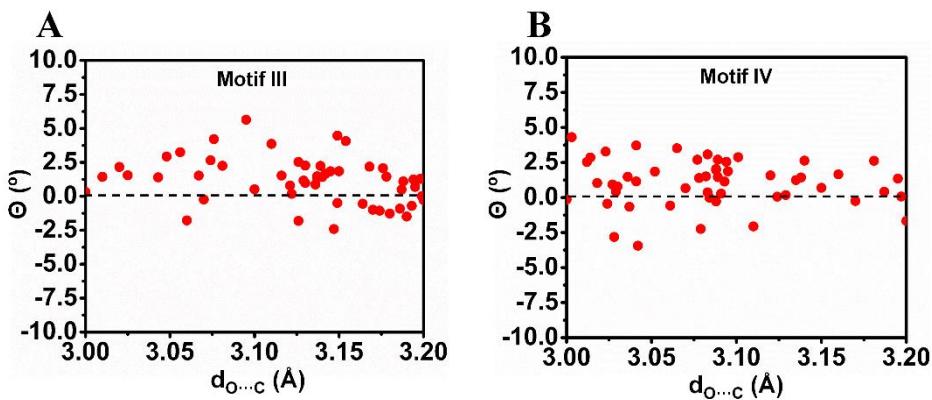


Figure S9: Pyramidality (Θ) of compounds from motif III and motif IV are plotted against the O...C distances [$d_{O\cdots C} = 3.0\text{--}3.2 \text{ \AA}$].

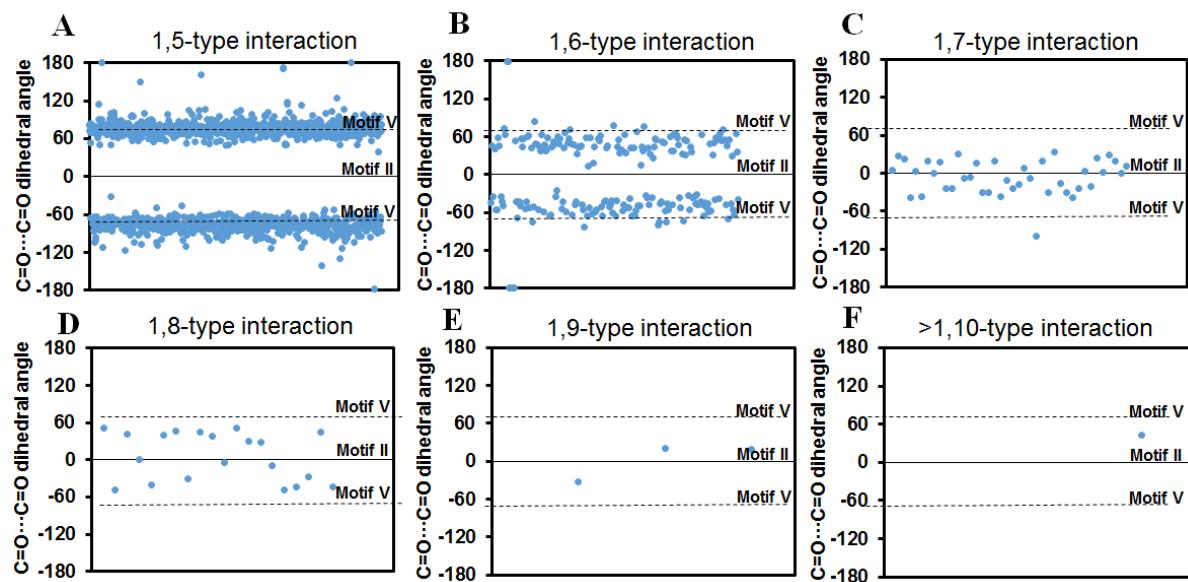


Figure S10. Change in $\text{C}=\text{O}\cdots\text{C}=\text{O}$ dihedral angle from intramolecular 1,x-type interactions. Motif V slowly converts to antiparallel motif II with increase in the number of bonds between the two carbonyl groups.

Table S4: Structural parameters and NBO data for different motifs that were retrieved from the CSD. NBO calculations were done in M052X/6-311+G(2d,p) level of theory.

Motifs	Ref. Code	$d_{1(O \cdots C)}$	$d_{2(O \cdots C)}$	$\theta_{1(O \cdots C=O)}$	$\theta_{2(O \cdots C=O)}$	$n \rightarrow \pi^*$ (kcal·mol ⁻¹)		$\pi \rightarrow \pi^*$ (kcal·mol ⁻¹)	
		(Å)	(Å)	(deg)	(deg)	$E^1_{(n \rightarrow \pi^*)}$	$E^2_{(n \rightarrow \pi^*)}$	$E^1_{(\pi \rightarrow \pi^*)}$	$E^2_{(\pi \rightarrow \pi^*)}$
Motif I (Intra-molecular)	NALJAL	2.67	3.97	90.9	27.1	2.89	NP	NP	0.11
	DIFSAM	2.73	4.21	93.9	12.8	1.84	NP	0.18	NP
	GIZCOI	2.94	4.22	85.1	18.6	0.83	0.12	0.07	NP
	OBEMUF	3.00	4.27	81.0	8.27	0.45	NP	0.09	NP
	HOFNUM	3.08	4.33	85.3	22.1	0.09	NP	0.09	NP
Motif I (Inter-molecular)	TCBENQ07	2.78	4.43	99.6	2.6	1.62	NP	0.08	NP
	GIHXUS	2.85	4.33	100.6	24.7	1.59	NP	NP	NP
	VAFPID	2.90	4.79	114.5	5.5	0.77	NP	NP	NP
	UKEQEHEH	2.94	4.34	95.6	25.7	1.00	NP	NP	0.05
	QOPWEY	2.99	4.44	92.8	15.5	1.77	NP	0.13	NP
Motif II (Intra-molecular)	PUXPAB	2.85	2.85	88.0	88.0	0.39	0.50	6.58	2.53
	GOCWIG	2.87	2.87	91.9	91.9	0.33	0.33	0.90	0.90
	NENXOT	2.89	2.93	101.4	98.7	0.15	0.11	1.34	1.3
	ROJTER	3.05	3.05	78.6	78.6	0.29	0.29	0.14	0.14
	ZILQIU	3.10	3.10	117.0	117.0	NP	NP	NP	NP
Motif II (Inter-molecular)	JIFRUL	2.81	2.816	88.5	88.58	0.56	0.97	0.62	0.70
	BADHUL	2.85	2.848	91.8	91.81	NP	NP	NP	2.61
	RIWGAH	2.90	2.90	89.1	89.10	0.31	0.29	0.81	0.66
	ROXRIH	2.95	2.95	84.9	84.91	0.40	0.39	0.58	0.70
	BIVTUV	2.99	2.99	85.9	85.95	0.21	0.21	0.45	0.45
Motif III (Intra-molecular)	IJUTUC	2.56	3.72	103.3	49.1	4.92	NP	0.71	NP
	OSOCOP	2.61	3.78	108.3	52.9	3.77	NP	0.58	NP
	FIKFUA	2.71	3.77	104.9	56.1	2.58	NP	0.61	NP
	LOSNAI	2.80	4.03	101.1	43.2	1.90	NP	0.92	0.06
	VEHYUC	2.89	3.97	109.6	58.3	1.57	NP	0.17	NP
Motif III (Inter-molecular)	NEPTUY	2.82	4.19	94.8	28.82	1.05	NP	0.92	0.11
	NAWLAA	2.95	4.28	95.6	29.56	1.44	NP	0.21	0.02
	OXACDH46	3.03	4.25	101.6	44.21	0.15	NP	0.62	0.05
	REZJAJ	3.09	4.56	111.5	39.13	0.20	NP	0.37	NP
	FOMHAZ13	3.15	4.17	95.0	48.74	0.07	NP	0.28	NP
Motif IV (Intra-molecular)	OROWIC	2.60	3.70	95.0	45.02	4.32	NP	0.16	NP
	YAGXAH	2.82	3.54	106.0	70.89	2.45	NP	0.06	NP
	LIZBIF	2.90	3.55	101.1	70.65	1.74	NP	0.07	NP
	WIMHAC	3.02	3.59	96.2	70.30	0.86	NP	0.07	NP
	WOLGIP	3.10	3.56	94.6	73.65	0.71	NP	NP	NP
Motif V (Intra-molecular)	JIWCNU	2.77	2.75	89.3	89.98	0.86	0.78	1.11	1.45
	XASFUT	2.87	2.86	84.2	84.42	0.73	1.55	0.35	0.12
	KOSZAT	2.90	2.90	92.0	92.06	0.74	0.74	0.40	0.40
	JODGUG	2.92	2.94	82.6	82.16	0.48	0.62	0.10	0.09
	YAHYOX	3.01	3.07	85.0	81.93	0.55	0.20	NP	NP
Motif VI (Inter-molecular)	OJAJEQ	2.87	3.32	101.3	79.3	2.58	NP	NP	NP
	XODSIU	2.88	3.34	102.2	79.7	1.84	NP	NP	NP
	CEGFIF	3.01	3.43	97.4	77.7	1.56	NP	NP	NP
	GESNUP	3.04	3.17	84.7	78.5	1.13	NP	NP	NP
	SAYZIC	3.07	3.31	91.5	79.9	1.16	NP	NP	NP

d_1 = Distance of first donor O to acceptor carbonyl C in Angstrom (Å); d_2 = Distance of second donor O to acceptor carbonyl C in Angstrom (Å); θ_1 = Angle from the first donor O to acceptor C=O in degree (°); θ_2 = Angle from the second donor O to acceptor C=O in degree (°); $E^1_{(n \rightarrow \pi^*)}$ = $n \rightarrow \pi^*$ second order perturbation energy from the first donor carbonyl O lone pair to acceptor carbonyl π^* orbital; $E^2_{(n \rightarrow \pi^*)}$ = $n \rightarrow \pi^*$ second order perturbation energy from the second donor carbonyl O lone pair to acceptor carbonyl π^* orbital; $E^1_{(\pi \rightarrow \pi^*)}$ = $\pi \rightarrow \pi^*$ second order perturbation energy from the first donor carbonyl π orbital to acceptor carbonyl π^* orbital; $E^2_{(\pi \rightarrow \pi^*)}$ = $\pi \rightarrow \pi^*$ second order perturbation energy from the second donor carbonyl π orbital to acceptor carbonyl π^* orbital; NP – not present at 0.05 threshold value.

Table S5. X-ray Structural parameters and NBO analysis for compounds of motif VI retrieved from CSD. NBO calculations were done in M052X/6-311+G(2d,p) level of theories.

Motif	Ref. Code	$d_{1(O\cdots C)}$ (Å)	$d_{3(O\cdots H)}$ (Å)	$\theta_{1(O\cdots C=O)}$ (°)	$\theta_{5(O\cdots H-C)}$ (°)	$n\rightarrow\pi^*$ (kcal·mol ⁻¹)	$\pi\rightarrow\pi^*$ (kcal·mol ⁻¹)	$n\rightarrow\sigma^*$ (kcal·mol ⁻¹)
Motif VI (Inter-molecular)	OJAJEQ	2.87	2.65	101.3	128.1	2.58	NP	NP
	XODSIU	2.89	2.68	102.2	128.0	1.84	NP	NP
	CEGFIF	3.01	2.80	97.4	126.6	1.56	NP	NP
	GESNUP	3.04	2.53	84.7	126.7	1.13	NP	NP
	SAYZIC	3.07	2.74	91.5	119.2	1.16	NP	NP

d_3 = Distance between O and H in Angstrom (Å); θ_5 = Angle formed by O···H-C. $E^1_{(n\rightarrow\pi^*)}$ = $n\rightarrow\pi^*$ second order perturbation energy from the first donor carbonyl O lone pair to acceptor carbonyl π^* orbital; $E^2_{(n\rightarrow\pi^*)}$ = $n\rightarrow\pi^*$ second order perturbation energy from the second donor carbonyl O lone pair to acceptor carbonyl π^* orbital; ; $E_{(\pi\rightarrow\pi^*)}$ = $\pi\rightarrow\pi^*$ second order perturbation energy from the donor carbonyl π orbital to acceptor carbonyl π^* orbital; $E_{(n\rightarrow\sigma^*)}$ = $n\rightarrow\sigma^*$ second order perturbation energy from the O lone pair to CH σ^* orbital. NP – not present at 0.05 threshold value.

Table S6: Structural data and NBO analysis of molecules from the CSD. Participation of sp-character and p-character oxygen lone pairs in $n\rightarrow\pi^*$ interaction is shown.

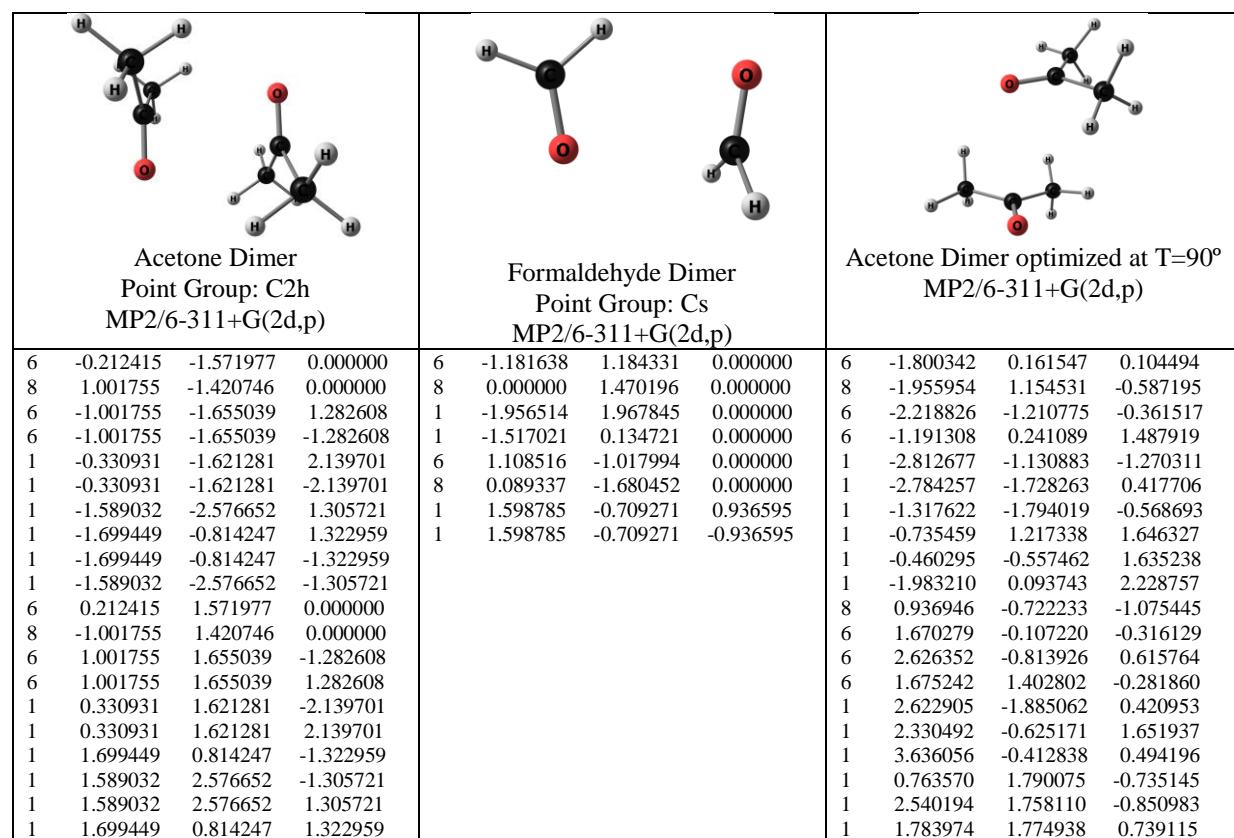
Motifs	Ref. Code	$d_{1(O\cdots C)}$ (Å)	$\theta_{1(O\cdots C=O)}$ (deg)	$\theta_{4(C=O\cdots C)}$ (deg)	$n\rightarrow\pi^*$ (kcal·mol ⁻¹)	Θ	Δ
					$n(sp)\rightarrow\pi^*$	$n(p)\rightarrow\pi^*$	
Motif I (Intra-molecular)	NALJAL	2.67	90.9	153.7	2.27	0.62	0.31
	DIFSAM	2.73	93.9	154.6	1.70	0.14	2.3
	GIZCOI	2.94	85.1	166.1	0.78	0.05	-0.93
	OBEMUF	3.00	81.0	167.4	0.42	0.03	-0.98
	HOFNUM	3.08	85.3	163.3	0.02	0.07	0.45
Motif I (Inter-molecular)	TCBENQ07	2.78	99.6	160.8	1.62	0.01	0.64
	GIHXUS	2.85	100.6	166.9	1.51	0.08	0.34
	VAFPID	2.90	114.5	169.0	0.77	0.01	1.18
	UKEQEHE	2.94	95.6	163.6	0.93	0.07	0.43
	QOPWEY	2.99	92.8	163.2	1.77	NP	0.68
Motif II (Intra-molecular)	PUXPAB	2.85	88.0	91.8	0.80	0.09	0.6(2)
	GOCWIG	2.87	91.9	88.0	0.66	NP	2.8(2)
	NENXOT	2.89	101.4	80.2	0.26	NP	1.81(18)
	ROJTER	3.05	78.6	101.0	0.46	0.12	0.7(3)
	ZILQIU	3.10	117.0	62.9	0.03	0.04	0.66(15)
Motif II (Inter-molecular)	JIFRUL	2.81	88.5	91.4	1.12	NP	2.48
	BADHUL	2.85	91.8	88.1	NP	NP	0.65
	RIWGAH	2.90	89.1	90.9	0.51	0.09	1.47
	ROXRIH	2.95	84.9	95.0	0.70	0.09	0.88
	BIVTUV	2.99	85.9	94.0	0.38	0.04	1.43
Motif III (Intra-molecular)	IJUTUC	2.56	103.3	92.1	1.01	3.91	4.6(6)
	OSOCOP	2.61	108.3	89.1	0.24	3.53	3.22(15)
	FIKFUA	2.71	104.9	85.0	0.40	2.18	4.5(4)
	LOSNAI	2.80	101.1	100.8	0.79	1.11	3.7(4)
	VEHYUC	2.89	109.6	86.2	0.25	1.32	4.1(8)
Motif III (Inter-molecular)	NEPTUY	2.82	94.8	118.1	0.96	0.09	2.83
	NAWLAA	2.95	95.6	118.7	0.59	0.85	2.92
	OXACDH46	3.03	101.6	101.5	0.15	NP	0.27
	REZJAJ	3.09	111.5	111.9	0.17	0.03	-0.06
	FOMHAZ13	3.15	95.0	95.0	0.04	0.03	-1.27

	OROWIC	2.60	95.0	124.6	1.85	2.47	2.0(2)	0.017
Motif IV (Intra-molecular)	YAGXAH	2.82	106.0	90.4	0.44	2.01	5.28(18)	0.044
	LIZBIF	2.90	101.1	95.5	0.40	1.34	-1.2(2)	-0.001
	WIMHAC	3.02	96.2	94.1	0.18	0.68	0.41(16)	0.004
	WOLGIP	3.10	94.6	92.1	0.12	0.59	0.3(2)	0.002
Motif V (Intra-molecular)	JIWCUO	2.77	89.3	77.7	0.79	0.85	0.7(3)	0.006
	XASFUT	2.87	84.2	83.6	1.11	1.17	1.18(18)	0.01
	KOSZAT	2.90	92.0	77.1	0.32	1.16	1.9(3)	0.015
	JODGUG	2.92	82.6	87.1	0.29	0.81	0.78(16)	0.007
Motif VI (Inter-molecular)	YAHYOX	3.01	85.0	79.4	0.11	0.64	2.08(19)	0.017
	OJAJEQ	2.87	101.3	99.1	0.30	2.28	0.02	0
	XODSIU	2.88	102.2	98.7	0.16	1.68	0.01	0
	CEGFIF	3.01	97.4	101.0	0.26	1.30	-0.04	0
GESNUP	3.04	84.7	101.6	0.20	0.93	0.08	0	
	SAYZIC	3.07	91.5	99.9	0.17	0.99	-0.04	-0.032

$\theta_{4(C=O \cdots C)}$ = Angle between the C=O \cdots C; Θ = pyramidalization of the carbonyl group in degree ($^{\circ}$); Δ =

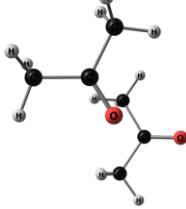
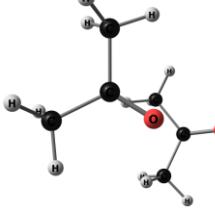
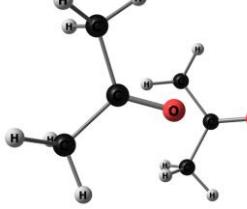
pyramidalization of the carbonyl group in angstrom (\AA); NP – not present at 0.01 threshold value.

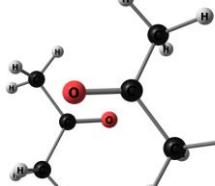
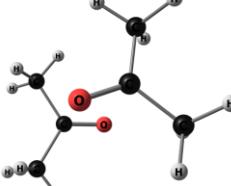
Cartesian Coordinates

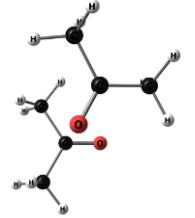
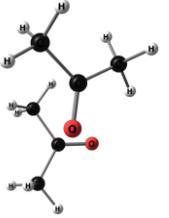


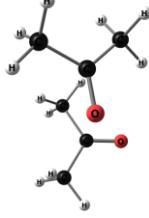
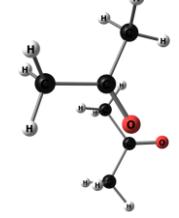
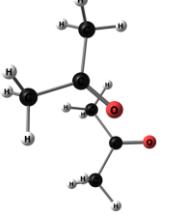
<p>Acetone Dimer $O \cdots C = 3.09 \text{ \AA}$ $C=O \cdots C=O = 0^\circ$ MP2/6-311+G(2d,p)</p>	<p>Acetone Dimer $O \cdots C = 3.09 \text{ \AA}$ $C=O \cdots C=O = 20^\circ$ MP2/6-311+G(2d,p)</p>	<p>Acetone Dimer $O \cdots C = 3.09 \text{ \AA}$ $C=O \cdots C=O = 40^\circ$ MP2/6-311+G(2d,p)</p>
<p>6 0.632267 -1.454810 0.000000 8 1.592674 -0.696722 0.000000 6 0.000000 -1.934598 1.282608 6 0.000000 -1.934598 -1.282608 1 0.556407 -1.558358 2.139701 1 -0.025191 -3.027131 1.305721 1 -1.032245 -1.576577 1.322958 1 0.556407 -1.558358 -2.139701 1 -1.032245 -1.576577 -1.322958 1 -0.025191 -3.027131 -1.305721 8 -1.592674 0.696722 0.000000 6 -0.632267 1.454810 0.000000 6 0.000000 1.934598 -1.282608 6 0.000000 1.934598 1.282608 1 -0.556407 1.558358 -2.139701 1 1.032245 1.576577 -1.322958 1 0.025191 3.027131 -1.305721 1 -0.556407 1.558358 2.139701 1 0.025191 3.027131 1.305721 1 1.032245 1.576577 1.322958</p>	<p>6 1.559064 -0.030840 0.352998 8 0.979915 -0.239156 1.410169 6 2.058681 -1.159384 -0.513120 6 1.788705 1.366555 -0.168627 1 1.873369 -2.115393 -0.026407 1 3.127533 -1.039886 -0.708643 1 1.535884 -1.126735 -1.472593 1 1.425221 2.102982 0.546269 1 1.264317 1.478564 -1.121547 1 2.852997 1.528001 -0.358367 8 -0.977679 -0.244627 -1.407310 6 -1.560238 -0.029632 -0.353355 6 -1.782843 1.370409 0.164198 6 -2.066634 -1.153311 0.515111 1 -1.385162 2.102348 -0.536887 1 -1.290685 1.473861 1.135032 1 -2.851055 1.546551 0.315657 1 -1.945479 -2.106350 0.002895 1 -3.116308 -0.995763 0.774841 1 -1.488937 -1.160690 1.443644</p>	<p>6 -1.639428 -0.025446 -0.342186 8 -1.194907 -0.712122 -1.250657 6 -2.364255 -0.636122 0.830587 6 -1.463135 1.473989 -0.311600 1 -2.596743 -1.679701 0.625615 1 -3.276585 -0.080667 1.059875 1 -1.706066 -0.575299 1.702275 1 -0.849409 1.801629 -1.149248 1 -1.004112 1.768895 0.634998 1 -2.442558 1.957676 -0.367484 8 0.931831 -0.205615 1.370169 6 1.601613 -0.091428 0.353903 6 2.111215 1.249690 -0.118870 6 1.948232 -1.281684 -0.503651 1 1.700288 2.051676 0.491849 1 1.852599 1.400679 -1.170010 1 3.202833 1.262368 -0.049834 1 1.739160 -2.204643 0.034399 1 2.994649 -1.248282 -0.815388 1 1.325103 -1.246430 -1.402238</p>

<p>Acetone Dimer $O \cdots C = 3.09 \text{ \AA}$ $C=O \cdots C=O = 60^\circ$ MP2/6-311+G(2d,p)</p>	<p>Acetone Dimer $O \cdots C = 3.09 \text{ \AA}$ $C=O \cdots C=O = 80^\circ$ MP2/6-311+G(2d,p)</p>	<p>Acetone Dimer $O \cdots C = 3.09 \text{ \AA}$ $C=O \cdots C=O = 100^\circ$ MP2/6-311+G(2d,p)</p>
<p>6 1.695718 0.080433 -0.291622 8 1.509686 1.189754 -0.768542 6 2.418560 -0.113433 1.017347 6 1.197960 -1.174551 -0.972290 1 2.902331 0.813841 1.319212 1 3.152751 -0.919222 0.942208 1 1.680471 -0.400779 1.771649 1 0.537966 -0.920357 -1.800971 1 0.684002 -1.816277 -0.253588 1 2.054816 -1.734621 -1.358949 8 -0.917577 -0.046588 1.360815 6 -1.618301 0.068689 0.366657 6 -2.358789 -1.108732 -0.224189 6 -1.778763 1.391717 -0.338880 1 -2.044810 -2.034779 0.254778 1 -2.193653 -1.162754 -1.303059 1 -3.432530 -0.967481 -0.069320 1 -1.409138 2.198959 0.290662 1 -2.821849 1.562798 -0.614934 1 -1.185551 1.370607 -1.258017</p>	<p>6 -1.797959 0.167821 0.161985 8 -1.861588 1.307681 -0.271134 6 -2.379130 -1.004756 -0.587164 6 -1.123374 -0.148459 1.479022 1 -3.014490 -0.653593 -1.398327 1 -2.943407 -1.656627 0.084314 1 -1.549587 -1.585227 -1.001047 1 -0.537709 0.704906 1.818705 1 -0.496851 -1.038087 1.386119 1 -1.893499 -0.368620 2.224949 8 0.905728 -0.612962 -1.125044 6 1.670697 -0.078612 -0.337507 6 2.643698 -0.879971 0.494838 6 1.696889 1.421032 -0.161414 1 2.633647 -1.924765 0.188788 1 2.364775 -0.801813 1.549925 1 3.651905 -0.468543 0.399263 1 0.770220 1.857923 -0.531094 1 2.539628 1.820552 -0.734613 1 1.857318 1.693814 0.883881</p>	<p>6 0.000000 0.000000 0.000000 8 0.000000 0.000000 1.220248 6 1.279209 0.000000 -0.799835 6 -1.287085 -0.002989 -0.797032 1 2.125816 0.222082 -0.152532 1 1.227282 0.720445 -1.620025 1 1.406305 -0.995902 -1.233877 1 -2.124909 -0.281252 -0.159459 1 -1.210882 -0.673558 -1.655412 1 -1.459058 1.005343 -1.186935 8 -0.028750 -3.064046 -0.432183 6 -1.184729 -3.364986 -0.681492 6 -1.609983 -3.836255 -2.052561 6 -2.260523 -3.291895 0.376410 1 -0.740654 -3.958789 -2.696425 1 -2.295179 -3.105311 -2.492018 1 -2.156052 -4.780069 -1.975489 1 -1.900810 -2.723876 1.233246 1 -2.510957 -4.308491 0.695030 1 -3.173179 -2.844484 -0.023995</p>

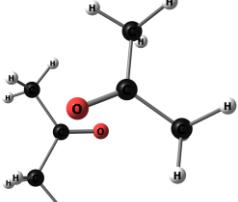
 <p>Acetone Dimer $O \cdots C = 3.09 \text{ \AA}$ $C=O \cdots C=O = 120^\circ$ MP2/6-311+G(2d,p)</p>				 <p>Acetone Dimer $O \cdots C = 3.09 \text{ \AA}$ $C=O \cdots C=O = 140^\circ$ MP2/6-311+G(2d,p)</p>				 <p>Acetone Dimer $O \cdots C = 3.09 \text{ \AA}$ $C=O \cdots C=O = 160^\circ$ MP2/6-311+G(2d,p)</p>			
6	1.972203	-0.148139	0.075704	6	0.000000	0.000000	0.000000	6	2.006752	-0.072656	0.033507
8	2.427827	-0.964292	-0.708351	8	0.000000	0.000000	1.219897	8	2.651418	-0.415189	-0.943698
6	2.163541	1.336890	-0.111357	8	3.018303	0.000000	-0.682523	8	-0.866601	0.079979	-1.105147
6	1.201644	-0.578368	1.306611	6	3.380270	-0.696985	-1.615980	6	-1.787434	0.033495	-0.306551
1	2.898761	1.524655	-0.891981	6	-0.043601	1.278892	-0.800851	6	1.830145	1.380360	0.406141
1	2.474774	1.807525	0.824990	6	0.006081	-1.292418	-0.789542	6	1.376185	-1.090511	0.959980
1	1.203122	1.770053	-0.403626	1	-0.263492	2.122187	-0.148358	1	2.481020	2.005228	-0.203074
1	0.805149	-1.583128	1.167105	1	-0.789643	1.210017	-1.597064	1	2.042866	1.533140	1.467721
1	0.400860	0.125928	1.539026	1	0.934773	1.424990	-1.265864	1	0.788822	1.659472	0.227750
1	1.889315	-0.586546	2.158399	1	0.433211	-2.094118	-0.188915	1	1.209943	-2.025073	0.426652
8	-0.859687	0.343007	-1.071102	1	0.549624	-1.184132	-1.729756	1	0.441139	-0.721566	1.384734
6	-1.775806	0.072308	-0.312228	1	-1.029417	-1.549164	-1.035031	1	2.066642	-1.273708	1.789563
6	-2.509046	1.140505	0.466080	6	3.528981	-0.142240	-3.014348	6	-2.368581	1.284294	0.312905
6	-2.229793	-1.352253	-0.097789	6	3.689954	-2.164520	-1.435348	6	-2.400927	-1.279072	0.122534
1	-2.144101	2.127516	0.186905	1	3.370494	0.934890	-3.011310	1	-1.957438	2.169736	-0.169160
1	-2.361145	0.980550	1.538094	1	2.800227	-0.620025	-3.676037	1	-2.129570	1.303816	1.380716
1	-3.583757	1.073980	0.277935	1	4.520919	-0.372188	-3.411289	1	-3.458044	1.283359	0.228189
1	-1.574619	-2.036603	-0.634094	1	3.474424	-2.467972	-0.412391	1	-1.824022	-2.110559	-0.278403
1	-3.256124	-1.467068	-0.457984	1	4.744636	-2.347361	-1.659301	1	-3.428589	-1.335692	-0.247885
1	-2.233812	-1.592237	0.968725	1	3.103211	-2.763869	-2.136738	1	-2.448145	-1.341936	1.212857

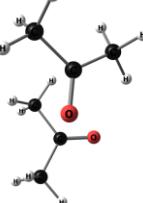
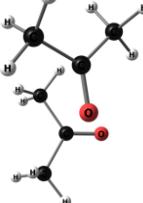
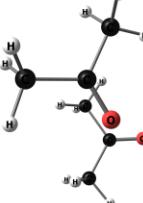
 <p>Acetone Dimer $O \cdots C = 3.09 \text{ \AA}$ $C=O \cdots C=O = 180^\circ$ MP2/6-311+G(2d,p)</p>				 <p>Acetone Dimer $O \cdots C = 3.0 \text{ \AA}$ $C=O \cdots C=O = 0^\circ$ MP2/6-311+G(2d,p)</p>				 <p>Acetone Dimer $O \cdots C = 3.0 \text{ \AA}$ $C=O \cdots C=O = 20^\circ$ MP2/6-311+G(2d,p)</p>			
6	2.008982	-0.020093	0.023915	6	0.553205	-1.462634	0.000000	6	1.560758	-0.035492	0.351779
8	2.673433	-0.077234	-0.997422	8	1.519709	-0.712138	0.000000	8	0.978399	-0.246357	1.406662
8	-0.871541	-0.041879	-1.106616	6	-0.079259	-1.941325	1.282623	6	2.061023	-1.162520	-0.515840
6	-1.787763	0.007598	-0.302887	6	-0.079259	-1.941325	-1.282623	6	1.791731	1.362990	-0.166213
6	1.692400	1.295573	0.696880	1	0.476894	-1.564237	2.139487	1	1.942108	-2.113790	0.000191
6	1.498075	-1.270081	0.706487	1	-0.103715	-3.033927	1.306558	1	3.109134	-1.006780	-0.782842
1	2.257297	2.100837	0.230260	1	-1.110492	-1.583349	1.322313	1	1.477240	-1.172352	-1.440531
1	1.919807	1.243384	1.765170	1	0.476894	-1.564237	-2.139487	1	1.382469	2.096713	0.526286
1	0.623233	1.494999	0.592890	1	-1.110492	-1.583349	-1.322313	1	1.317850	1.464780	-1.146142
1	1.423149	-2.081469	-0.015381	1	-0.103715	-3.033927	-1.306558	1	2.862859	1.537836	-0.298073
1	0.533315	-1.099270	1.186320	8	-1.519709	0.712138	0.000000	8	-0.978238	-0.237379	-1.406686
1	2.212707	-1.551526	1.486695	6	-0.553205	1.462634	0.000000	6	-1.560824	-0.024650	-0.352277
6	-2.356724	1.324049	0.176071	6	0.079259	1.941325	-1.282623	6	-1.785663	1.374501	0.166630
6	-2.411347	-1.244375	0.270384	6	0.079259	1.941325	1.282623	6	-2.067037	-1.150021	0.514014
1	-1.895414	2.150576	-0.361898	1	-0.476894	1.564237	-2.139487	1	-1.375503	2.106990	-0.526647
1	-2.175828	1.434209	1.249476	1	1.110492	1.583349	-1.322313	1	-1.308677	1.474121	1.145310
1	-3.439818	1.343133	0.030387	1	0.103715	3.033927	-1.306558	1	-2.855731	1.553244	0.301657
1	-1.874424	-2.124319	-0.079480	1	-0.476894	1.564237	2.139487	1	-1.949582	-2.101704	-0.001581
1	-3.458087	-1.306165	-0.040358	1	0.103715	3.033927	1.306558	1	-3.115487	-0.990980	0.777753
1	-2.402818	-1.207511	1.363121	1	1.110492	1.583349	1.322313	1	-1.486163	-1.161644	1.440509

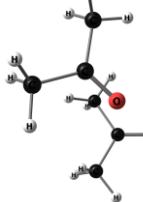
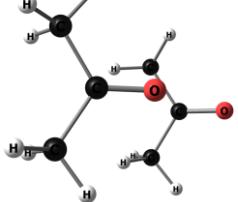
 Acetone Dimer $O \cdots C = 3.0 \text{ \AA}$ $C=O \cdots C=O = 40^\circ$ MP2/6-311+G(2d,p)			 Acetone Dimer $O \cdots C = 3.0 \text{ \AA}$ $C=O \cdots C=O = 60^\circ$ MP2/6-311+G(2d,p)			 Acetone Dimer $O \cdots C = 3.0 \text{ \AA}$ $C=O \cdots C=O = 80^\circ$ MP2/6-311+G(2d,p)					
6	-1.607716	0.012180	-0.313685	6	1.663815	0.053291	-0.276474	6	-1.760818	0.149784	0.160226
8	-1.161604	-0.658921	-1.232767	8	1.478517	1.157397	-0.764220	8	-1.830807	1.291877	-0.263907
6	-2.300711	-0.620185	0.866546	6	2.356166	-0.127185	1.050805	6	-2.308351	-1.025024	-0.610559
6	-1.489544	1.517559	-0.291478	6	1.220411	-1.211664	-0.976465	6	-1.140112	-0.163025	1.504760
1	-2.474089	-1.677868	0.675991	1	2.779720	0.818982	1.382948	1	-2.882938	-0.675078	-1.466325
1	-3.245166	-0.112966	1.078376	1	3.137944	-0.887490	0.976896	1	-2.930984	-1.651204	0.033880
1	-1.654956	-0.509555	1.741812	1	1.618533	-0.472361	1.779865	1	-1.468004	-1.630907	-0.959045
1	-0.889282	1.864594	-1.131067	1	0.578008	-0.971567	-1.823051	1	-0.568360	0.691417	1.865325
1	-1.045160	1.839076	0.653362	1	0.704310	-1.872956	-0.277328	1	-0.511238	-1.053791	1.444174
1	-2.488312	1.959428	-0.352350	1	2.104279	-1.743947	-1.340870	1	-1.941825	-0.377389	2.218340
8	0.890101	-0.259452	1.325560	8	-0.869511	-0.007492	1.329317	8	0.855003	-0.582990	-1.112771
6	1.577561	-0.130763	0.322790	6	-1.590267	0.099505	0.347845	6	1.637325	-0.056776	-0.335460
6	2.074339	1.219457	-0.136729	6	-2.332353	-1.085186	-0.225914	6	2.613739	-0.869081	0.481957
6	1.954248	-1.311911	-0.534178	6	-1.772125	1.418638	-0.359027	6	1.680258	1.441869	-0.157583
1	1.651766	2.011987	0.478327	1	-2.007657	-2.006682	0.254632	1	2.586228	-1.913907	0.177061
1	1.816001	1.374864	-1.187418	1	-2.180077	-1.145031	-1.306429	1	2.353230	-0.786676	1.541388
1	3.165497	1.244583	-0.064946	1	-3.404824	-0.948329	-0.058903	1	3.625196	-0.470093	0.369376
1	1.747448	-2.240167	-0.004501	1	-1.396879	2.230524	0.261080	1	0.758784	1.889330	-0.527246
1	3.005022	-1.264282	-0.828880	1	-2.820824	1.581722	-0.618081	1	2.528743	1.833196	-0.727928
1	1.344882	-1.278494	-1.442263	1	-1.193887	1.397053	-1.287659	1	1.841844	1.710162	0.888782

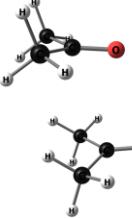
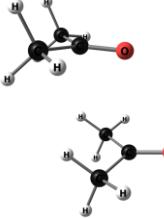
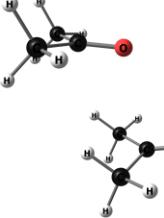
 Acetone Dimer $O \cdots C = 3.0 \text{ \AA}$ $C=O \cdots C=O = 100^\circ$ MP2/6-311+G(2d,p)			 Acetone Dimer $O \cdots C = 3.0 \text{ \AA}$ $C=O \cdots C=O = 120^\circ$ MP2/6-311+G(2d,p)			 Acetone Dimer $O \cdots C = 3.0 \text{ \AA}$ $C=O \cdots C=O = 140^\circ$ MP2/6-311+G(2d,p)					
6	-1.778489	0.183035	0.102201	6	1.916931	-0.142898	0.056435	6	1.922287	-0.101522	0.087492
8	-1.979852	1.157305	-0.603127	8	2.335079	-0.953867	-0.751730	8	2.421973	-0.793200	-0.782666
6	-2.179324	-1.208467	-0.322194	6	2.119817	1.342386	-0.119497	6	2.032278	1.404327	0.076331
6	-1.145717	0.307776	1.472197	6	1.201242	-0.579302	1.318212	6	1.197950	-0.717656	1.266580
1	-2.766247	-1.163219	-1.237843	1	2.808502	1.530567	-0.941391	1	2.729311	1.723016	-0.696840
1	-2.750980	-1.700027	0.469674	1	2.499626	1.789677	0.803084	1	2.356092	1.772276	1.053782
1	-1.275028	-1.795524	-0.500290	1	1.152526	1.799000	-0.343391	1	1.043981	1.821696	-0.131287
1	-0.693148	1.290851	1.592670	1	0.799695	-1.583879	1.192338	1	0.820360	-1.703107	0.997621
1	-0.408756	-0.481870	1.635929	1	0.410842	0.122966	1.590033	1	0.387749	-0.077055	1.619838
1	-1.926352	0.181792	2.228935	1	1.926101	-0.590408	2.138481	1	1.912622	-0.827263	2.088570
8	0.851475	-0.563868	-1.132882	8	-0.825057	0.341968	-1.059992	8	-0.824118	0.228308	-1.072468
6	1.666766	-0.116056	-0.341327	6	-1.746753	0.069962	-0.306654	6	-1.737795	0.172266	-0.264540
6	2.566678	-1.014857	0.473877	6	-2.494005	1.138258	0.457749	6	-2.274133	1.408933	0.420045
6	1.830618	1.372677	-0.145207	6	-2.192664	-1.356294	-0.087887	6	-2.385178	-1.140314	0.109258
1	2.441213	-2.052535	0.169224	1	-2.136476	2.126086	0.171915	1	-1.785215	2.299367	0.028211
1	2.323297	-0.908731	1.535161	1	-2.350676	0.989191	1.531977	1	-2.102772	1.336712	1.498169
1	3.610259	-0.712107	0.354015	1	-3.567021	1.060285	0.264360	1	-3.354581	1.483622	0.272252
1	0.973637	1.898481	-0.564387	1	-1.522921	-2.039335	-0.607665	1	-1.874225	-1.962752	-0.388286
1	2.740108	1.694921	-0.661898	1	-3.211492	-1.481964	-0.465517	1	-3.438038	-1.128098	-0.186348
1	1.953753	1.619520	0.911667	1	-2.213992	-1.588114	0.980201	1	-2.356386	-1.280783	1.193139

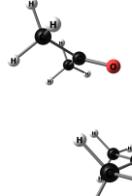
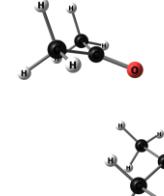
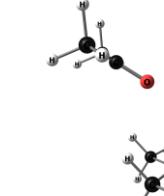
 <p>Acetone Dimer $O \cdots C = 3.0 \text{ \AA}$ $C=O \cdots C=O = 160^\circ$ MP2/6-311+G(2d,p)</p>			
6 1.923048 -0.085907 0.076941	6 1.936137 -0.015110 0.000062	6 0.464706 -1.467043 0.000000	
8 2.467320 -0.608857 -0.879391	8 2.552946 -0.074662 -1.048909	8 1.442583 -0.731326 0.000000	
6 1.957041 1.407763 0.299701	8 -0.848124 -0.043653 -1.116663	6 -0.170401 -1.941993 1.282590	
6 1.220570 -0.912476 1.134003	6 -1.762948 0.006625 -0.309584	6 -0.170401 -1.941993 -1.282590	
1 2.625101 1.878531 -0.419499	6 1.669901 1.299875 0.696361	1 0.386376 -1.565827 2.139472	
1 2.279822 1.633359 1.319968	6 1.446292 -1.263778 0.701423	1 -0.197548 -3.034599 1.307432	
1 0.947424 1.805085 0.172157	1 2.190085 2.107295 0.183776	1 -1.198984 -1.581214 1.321020	
1 0.876214 -1.850536 0.701608	1 1.991189 1.247204 1.740494	1 0.386376 -1.565827 -2.139472	
1 0.387636 -0.370041 1.584842	1 0.595451 1.496704 0.689866	1 -1.198984 -1.581214 -1.321020	
1 1.939870 -1.133401 1.929147	1 1.307546 -2.063536 -0.023938	1 -0.197548 -3.034599 -1.307432	
8 -0.839680 0.189471 -1.059505	1 0.519874 -1.082405 1.248546	8 -1.442583 0.731326 0.000000	
6 -1.727631 0.332117 -0.233827	1 2.206708 -1.569459 1.427334	6 -0.464706 1.467043 0.000000	
6 -2.124176 1.696773 0.282647	6 -2.327336 1.323407 0.173528	6 0.170401 1.941993 -1.282590	
6 -2.482533 -0.847820 0.332389	6 -2.390772 -1.244990 0.259563	6 0.170401 1.941993 1.282590	
1 -1.628517 2.476787 -0.293031	1 -1.848430 2.151027 -0.347107	1 -0.386376 1.565827 -2.139472	
1 -1.840613 1.782062 1.336326	1 -2.168850 1.421124 1.251587	1 1.198984 1.581214 -1.321020	
1 -3.208114 1.825509 0.231284	1 -3.407025 1.353707 0.005188	1 0.197548 3.034599 -1.307432	
1 -2.035842 -1.778455 -0.012977	1 -1.862969 -2.126049 -0.101107	1 -0.386376 1.565827 2.139472	
1 -3.525689 -0.800319 0.006755	1 -3.440626 -1.297099 -0.042347	1 0.197548 3.034599 1.307432	
1 -2.484635 -0.812082 1.425015	1 -2.372400 -1.216463 1.352443	1 1.198984 1.581214 1.321020	

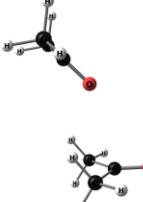
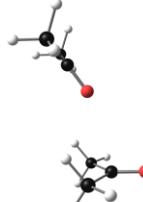
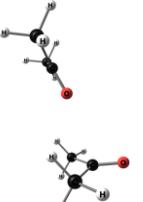
 <p>Acetone Dimer $O \cdots C = 2.9 \text{ \AA}$ $C=O \cdots C=O = 20^\circ$ MP2/6-311+G(2d,p)</p>			
6 1.548652 -0.050272 0.274452	6 -1.604916 0.002081 -0.275129	6 1.652136 0.041621 -0.240284	
8 0.996073 -0.260470 1.345274	8 -1.178011 -0.671136 -1.201634	8 1.471779 1.145970 -0.729068	
6 2.047230 -1.175728 -0.595831	6 -2.296031 -0.624149 0.909429	6 2.341150 -0.141338 1.088364	
6 1.772487 1.348578 -0.245864	6 -1.479005 1.507073 -0.256203	6 1.219800 -1.222895 -0.948421	
1 1.866753 -2.132927 -0.109550	1 -2.444250 -1.688556 0.735980	1 2.730503 0.812009 1.441125	
1 3.115591 -1.052875 -0.793178	1 -3.255520 -0.134620 1.095387	1 3.150856 -0.870699 1.000411	
1 1.522212 -1.143411 -1.553370	1 -1.668192 -0.481576 1.792406	1 1.615263 -0.528879 1.807127	
1 1.373972 2.081150 0.454276	1 -0.876283 1.848189 -1.096589	1 0.576519 -0.982560 -1.794374	
1 1.285995 1.453401 -1.218502	1 -1.034663 1.830511 0.687566	1 0.708579 -1.893773 -0.255244	
1 2.842612 1.521953 -0.389089	1 -2.476019 1.952453 -0.322073	1 2.109044 -1.743789 -1.316374	
8 -0.890308 -0.218067 -1.285458	8 0.843961 -0.241282 1.259074	8 -0.824676 0.005384 1.267726	
6 -1.528518 -0.009249 -0.263338	6 1.561435 -0.120002 0.276705	6 -1.572569 0.112278 0.306835	
6 -1.795044 1.388785 0.237415	6 2.090448 1.224071 -0.163941	6 -2.331956 -1.071540 -0.245135	
6 -2.051773 -1.138125 0.587003	6 1.938219 -1.303743 -0.576118	6 -1.767099 1.430389 -0.398271	
1 -1.383536 2.124023 -0.452100	1 1.666889 2.020244 0.445702	1 -2.000715 -1.992414 0.232086	
1 -1.338568 1.504735 1.224320	1 1.857567 1.388996 -1.219248	1 -2.202470 -1.136462 -1.328428	
1 -2.870528 1.548227 0.350516	1 3.180044 1.231384 -0.068626	1 -3.400149 -0.929500 -0.056549	
1 -1.907876 -2.089244 0.077155	1 1.720227 -2.230437 -0.048193	1 -1.384343 2.243302 0.215836	
1 -3.108766 -0.990096 0.820807	1 2.990906 -1.263850 -0.864639	1 -2.818931 1.591974 -0.644918	
1 -1.495992 -1.142886 1.529021	1 1.333518 -1.266489 -1.487351	1 -1.199421 1.408146 -1.333478	

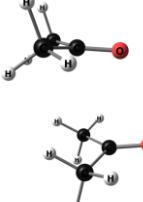
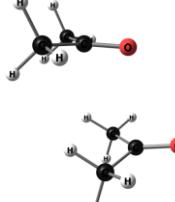
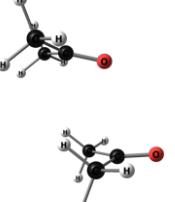
		
Acetone Dimer $O \cdots C = 2.9 \text{ \AA}$ $C=O \cdots C=O = 80^\circ$ MP2/6-311+G(2d,p)	Acetone Dimer $O \cdots C = 2.9 \text{ \AA}$ $C=O \cdots C=O = 100^\circ$ MP2/6-311+G(2d,p)	Acetone Dimer $O \cdots C = 2.9 \text{ \AA}$ $C=O \cdots C=O = 120^\circ$ MP2/6-311+G(2d,p)
6 -1.743596 0.122679 0.134990	6 -1.767182 0.153848 0.082786	6 1.891806 -0.116475 0.043350
8 -1.811935 1.263192 -0.293279	8 -1.968981 1.127403 -0.623031	8 2.281243 -0.932202 -0.774014
6 -2.288873 -1.054954 -0.633110	6 -2.169962 -1.238140 -0.338549	6 2.113648 1.366084 -0.134247
6 -1.141507 -0.183220 1.489829	6 -1.145754 0.281979 1.458015	6 1.205818 -0.545247 1.324342
1 -2.834819 -0.708553 -1.508835	1 -2.728014 -1.196744 -1.272257	1 2.764050 1.546389 -0.988467
1 -2.940774 -1.657874 0.004899	1 -2.774213 -1.710163 0.441221	1 2.550548 1.797006 0.770907
1 -1.451876 -1.682402 -0.948664	1 -1.269534 -1.840327 -0.479181	1 1.146852 1.845542 -0.304139
1 -0.559380 0.666520 1.845124	1 -0.679033 1.259200 1.572965	1 0.782706 -1.541878 1.205802
1 -0.527359 -1.085086 1.449519	1 -0.423660 -0.517229 1.639121	1 0.436567 0.169363 1.623114
1 -1.955054 -0.372040 2.197290	1 -1.937197 0.178721 2.207010	1 1.954926 -0.574329 2.122105
8 0.810825 -0.544390 -1.064988	8 0.802975 -0.516711 -1.081107	8 -0.779149 0.291928 -1.009838
6 1.617668 -0.027323 -0.306877	6 1.648775 -0.081911 -0.314776	6 -1.728913 0.034610 -0.286869
6 2.604095 -0.850342 0.487139	6 2.553243 -0.994678 0.479473	6 -2.478067 1.113533 0.460653
6 1.678879 1.470671 -0.129518	6 1.844091 1.404038 -0.127660	6 -2.207566 -1.383881 -0.088100
1 2.563817 -1.893372 0.177492	1 2.401844 -2.030626 0.180695	1 -2.091159 2.095502 0.193379
1 2.363100 -0.772114 1.551575	1 2.338204 -0.882205 1.546250	1 -2.372829 0.954686 1.537876
1 3.615937 -0.457346 0.358456	1 3.598709 -0.710892 0.332876	1 -3.545613 1.059576 0.230213
1 0.749794 1.925068 -0.470941	1 0.991899 1.943961 -0.538513	1 -1.541997 -2.076087 -0.601059
1 2.511850 1.855150 -0.726810	1 2.753400 1.707438 -0.655862	1 -3.222079 -1.487026 -0.483579
1 1.876119 1.737429 0.911153	1 1.984301 1.652133 0.926921	1 -2.251485 -1.622813 0.977832

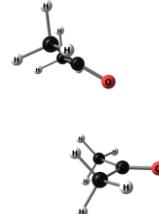
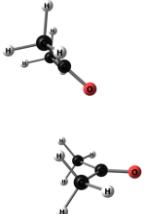
		
Acetone Dimer $O \cdots C = 2.9 \text{ \AA}$ $C=O \cdots C=O = 140^\circ$ MP2/6-311+G(2d,p)	Acetone Dimer $O \cdots C = 2.9 \text{ \AA}$ $C=O \cdots C=O = 160^\circ$ MP2/6-311+G(2d,p)	Acetone Dimer $O \cdots C = 2.9 \text{ \AA}$ $C=O \cdots C=O = 180^\circ$ MP2/6-311+G(2d,p)
6 1.885917 -0.078704 0.021014	6 1.883255 -0.121427 0.026481	6 1.863733 -0.028158 -0.042096
8 2.381811 -0.695265 -0.905371	8 2.454105 -0.484394 -0.986495	8 1.562721 0.027662 1.136865
8 -0.790581 0.143787 -1.072998	8 -0.788479 -0.036446 -1.098070	8 -0.859082 -0.050903 -1.039975
6 -1.724646 0.001102 -0.299745	6 -1.714114 -0.072755 -0.302709	6 -0.967643 -0.104990 -2.255085
6 1.950449 1.427400 0.108558	6 1.787726 1.335612 0.414112	6 2.202872 -1.336805 -0.718789
6 1.222898 -0.794812 1.180119	6 1.290137 -1.120672 0.998315	6 1.988934 1.222867 -0.885338
1 2.584631 1.820849 -0.684027	1 2.357351 1.946376 -0.284190	1 2.174683 -2.150140 0.004351
1 2.332803 1.734816 1.086036	1 2.161670 1.479367 1.432024	1 3.195574 -1.274710 -1.174476
1 0.940473 1.829064 0.000234	1 0.739876 1.642712 0.397283	1 1.485570 -1.532098 -1.518511
1 0.870025 -1.773743 0.858905	1 1.027142 -2.036479 0.471488	1 1.354080 2.006808 -0.476227
1 0.404075 -0.208282 1.601114	1 0.420568 -0.714526 1.518310	1 1.736412 1.035354 -1.930217
1 1.969004 -0.931998 1.969402	1 2.047491 -1.353493 1.753745	1 3.031375 1.555611 -0.850528
6 -2.332939 1.170252 0.441124	6 -2.271232 1.182378 0.330348	6 -1.019741 -1.423913 -2.992812
6 -2.325614 -1.357401 -0.026125	6 -2.357834 -1.376574 0.107578	6 -1.063811 1.143680 -3.101618
1 -1.867224 2.101906 0.123916	1 -1.839565 2.065126 -0.138635	1 -0.985927 -2.251378 -2.286027
1 -2.196515 1.035794 1.518245	1 -2.038792 1.183756 1.399874	1 -0.180319 -1.494856 -3.690906
1 -3.409713 1.213499 0.258028	1 -3.359980 1.205622 0.239418	1 -1.935291 -1.482322 -3.587526
1 -1.773836 -2.124395 -0.566773	1 -1.808289 -2.214956 -0.316685	1 -0.956360 2.028849 -2.477283
1 -3.373451 -1.367077 -0.338969	1 -3.391546 -1.398874 -0.249291	1 -2.032491 1.167574 -3.608753
1 -2.307651 -1.568056 1.046656	1 -2.391723 -1.459642 1.197149	1 -0.295021 1.136480 -3.879090

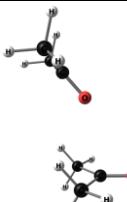
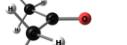
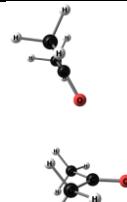
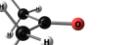
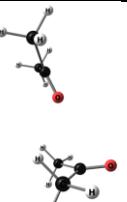
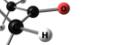
 Acetone Dimer $O \cdots C = 3.1 \text{ \AA}$ $C=O \cdots C = 90^\circ$ MP2/6-311+G(2d,p)			 Acetone Dimer $O \cdots C = 3.1 \text{ \AA}$ $C=O \cdots C = 100^\circ$ MP2/6-311+G(2d,p)			 Acetone Dimer $O \cdots C = 3.1 \text{ \AA}$ $C=O \cdots C = 110^\circ$ MP2/6-311+G(2d,p)					
6	2.306014	-1.658265	0.000000	6	2.254052	-1.699463	0.000000	6	2.183553	-1.745154	0.000000
8	1.508805	-0.733823	0.000000	8	1.532431	-0.714755	0.000000	8	1.548496	-0.702495	0.000000
6	2.781178	-2.299521	1.286247	6	2.699886	-2.361489	1.285232	6	2.590911	-2.429941	1.285238
6	2.781178	-2.299521	-1.286247	6	2.699886	-2.361489	-1.285232	6	2.590911	-2.429941	-1.285238
1	2.506413	-1.680278	2.138431	1	2.393866	-1.762796	2.141392	1	2.269320	-1.840602	2.142223
1	2.506413	-1.680278	-2.138431	1	2.393866	-1.762796	-2.141392	1	2.269320	-1.840602	-2.142223
1	3.858678	-2.476157	1.272793	1	3.782833	-2.508667	1.292876	1	3.674824	-2.572213	1.312621
1	2.297511	-3.276800	1.382731	1	2.242424	-3.353132	1.352469	1	2.136991	-3.423843	1.331044
1	2.297511	-3.276800	-1.382731	1	2.242424	-3.353132	-1.352469	1	2.136991	-3.423843	-1.331044
1	3.858678	-2.476157	-1.272793	1	3.782833	-2.508667	-1.292876	1	3.674824	-2.572213	-1.312621
6	3.864561	1.297691	0.000000	6	3.684149	1.531741	0.000000	6	3.491564	1.726729	0.000000
8	2.939448	0.209027	0.000000	8	2.807609	2.377867	0.000000	8	2.639039	2.597488	0.000000
6	4.499711	0.822500	-1.288526	6	4.282832	1.009501	-1.287277	6	4.070595	1.180734	-1.286244
6	4.499711	0.822500	1.288526	6	4.282832	1.009501	1.287277	6	4.070595	1.180734	1.286244
1	3.846488	1.049819	-2.129428	1	3.671050	1.319337	-2.132803	1	3.504219	1.556436	-2.136593
1	3.846488	1.049819	2.129428	1	3.671050	1.319337	2.132803	1	3.504219	1.556436	2.136593
1	4.724033	-0.244502	-1.257813	1	4.375190	-0.077486	-1.267417	1	4.058132	0.089531	-1.282869
1	5.450515	1.348832	-1.421318	1	5.291572	1.419268	-1.400428	1	5.114638	1.497074	-1.373743
1	5.450515	1.348832	1.421318	1	5.291572	1.419268	1.400428	1	5.114638	1.497074	1.373743
1	4.724033	-0.244502	1.257813	1	4.375190	-0.077486	1.267417	1	4.058132	0.089531	1.282869

 Acetone Dimer $O \cdots C = 3.1 \text{ \AA}$ $C=O \cdots C = 120^\circ$ MP2/6-311+G(2d,p)			 Acetone Dimer $O \cdots C = 3.1 \text{ \AA}$ $C=O \cdots C = 130^\circ$ MP2/6-311+G(2d,p)			 Acetone Dimer $O \cdots C = 3.1 \text{ \AA}$ $C=O \cdots C = 140^\circ$ MP2/6-311+G(2d,p)					
6	2.114764	-1.784826	0.000000	6	2.031167	-1.822461	0.000000	6	1.940158	-1.854072	0.000000
8	1.568404	-0.692878	0.000000	8	1.578519	-0.688222	0.000000	8	1.585051	-0.685474	0.000000
6	2.473423	-2.494321	1.285675	6	2.329138	-2.558569	1.285835	6	2.172115	-2.613421	1.285665
6	2.473423	-2.494321	-1.285675	6	2.329138	-2.558569	-1.285835	6	2.172115	-2.613421	-1.285665
1	2.166604	-1.896040	2.141859	1	2.070299	-1.937438	2.141589	1	1.971749	-1.971129	2.141443
1	2.166604	-1.896040	-2.141859	1	2.070299	-1.937438	-2.141589	1	1.971749	-1.971129	-2.141443
1	3.551205	-2.676534	1.324285	1	3.388862	-2.825890	1.325642	1	3.203031	-2.975968	1.324243
1	1.982229	-3.470396	1.321704	1	1.761326	-3.492101	1.321310	1	1.521389	-3.491141	1.322028
1	1.982229	-3.470396	-1.321704	1	1.761326	-3.492101	-1.321310	1	1.521389	-3.491141	-1.322028
1	3.551205	-2.676534	-1.324285	1	3.388862	-2.825890	-1.325642	1	3.203031	-2.975968	-1.324243
6	3.281630	1.903561	0.000000	6	3.050609	2.052141	0.000000	6	2.805371	2.175900	0.000000
8	2.413021	2.758737	0.000000	8	2.136432	2.858858	0.000000	8	1.833928	2.913162	0.000000
6	3.867160	1.363243	-1.285534	6	3.662317	1.541052	-1.285028	6	3.451079	1.708534	-1.284658
6	3.867160	1.363243	1.285534	6	3.662317	1.541052	1.285029	6	3.451079	1.708534	1.284658
1	3.323782	1.765764	-2.138616	1	3.115962	1.937035	-2.139293	1	2.888741	2.080439	-2.139406
1	3.323782	1.765764	2.138616	1	3.115962	1.937035	2.139293	1	2.888741	2.080439	2.139406
1	3.817456	0.273036	-1.295222	1	3.636160	0.450069	-1.302185	1	3.480318	0.617668	-1.306906
1	4.921409	1.648227	-1.353504	1	4.709941	1.851267	-1.341355	1	4.481372	2.072940	-1.333316
1	4.921409	1.648227	1.353504	1	4.709941	1.851267	1.341355	1	4.481372	2.072940	1.333316
1	3.817456	0.273036	1.295222	1	3.636160	0.450069	1.302185	1	3.480318	0.617668	1.306906

		
Acetone Dimer O···C = 3.1 Å C=O···C = 150° MP2/6-311+G(2d,p)	Acetone Dimer O···C = 3.1 Å C=O···C = 160° MP2/6-311+G(2d,p)	Acetone Dimer O···C = 3.1 Å C=O···C = 170° MP2/6-311+G(2d,p)
6 1.839509 -1.880024 0.000000 8 1.586617 -0.685115 0.000000 6 2.002595 -2.657193 1.285446 6 2.002595 -2.657193 -1.285446 1 1.862180 -1.999381 2.141319 1 1.862180 -1.999381 -2.141319 1 2.996092 -3.112187 1.323292 1 1.274302 -3.471691 1.322017 1 1.274302 -3.471691 -1.322017 1 2.996092 -3.112187 -1.323292 6 2.550476 2.272522 0.000000 8 1.517702 2.921489 0.000000 6 3.234191 1.862346 -1.284204 6 3.234191 1.862346 1.284204 1 2.642278 2.184177 -2.139317 1 2.642278 2.184177 2.139317 1 3.356955 0.777957 -1.304745 1 4.229329 2.314013 -1.332758 1 4.229329 2.314013 1.332758 1 3.356955 0.777957 1.304745	6 1.737048 -1.896674 0.000000 8 1.588006 -0.684423 0.000000 6 1.831064 -2.685239 1.285246 6 1.831064 -2.685239 -1.285246 1 1.749046 -2.017820 2.141217 1 1.749046 -2.017820 -2.141217 1 2.780623 -3.225857 1.323055 1 1.033740 -3.432301 1.321386 1 1.033740 -3.432301 -1.321386 1 2.780623 -3.225857 -1.323055 6 2.287289 2.346690 0.000000 8 1.200073 2.900047 0.000000 6 3.004026 1.996868 -1.283805 6 3.004026 1.996868 1.283805 1 2.392066 2.277273 -2.139428 1 2.392066 2.277273 2.139428 1 3.206054 0.924408 -1.307392 1 3.963127 2.520943 -1.327124 1 3.963127 2.520943 1.327124 1 3.206054 0.924408 1.307392	6 1.631175 -1.905581 0.000000 8 1.587957 -0.685011 0.000000 6 1.655429 -2.699351 1.285096 6 1.655429 -2.699351 -1.285096 1 1.632234 -2.027413 2.141098 1 1.632234 -2.027413 -2.141098 1 2.553878 -3.321151 1.322945 1 0.795500 -3.373403 1.320857 1 0.795500 -3.373403 -1.320857 1 2.553878 -3.321151 -1.322945 6 2.019383 2.395657 0.000000 8 0.884739 2.844359 0.000000 6 2.765087 2.112579 -1.283504 6 2.765087 2.112579 1.283504 1 2.132603 2.342013 -2.139419 1 2.132603 2.342013 2.139419 1 3.056950 1.060971 -1.308544 1 3.676447 2.715937 -1.324111 1 3.676447 2.715937 1.324111 1 3.056950 1.060971 1.308544

		
Acetone Dimer O···C = 3.0 Å C=O···C = 90° MP2/6-311+G(2d,p)	Acetone Dimer O···C = 3.0 Å C=O···C = 100° MP2/6-311+G(2d,p)	Acetone Dimer O···C = 3.0 Å C=O···C = 110° MP2/6-311+G(2d,p)
6 2.335277 -1.628421 0.000000 8 1.541268 -0.700771 0.000000 6 2.801626 -2.277009 1.285919 6 2.801626 -2.277009 -1.285919 1 2.536774 -1.654730 2.138938 1 2.536774 -1.654730 -2.138938 1 3.875664 -2.472909 1.272976 1 3.875664 -2.472909 -1.272976 1 2.300252 -3.245593 1.380795 1 2.300252 -3.245593 -1.380795 1 3.875664 -2.472909 -1.272976 6 3.820402 1.250011 0.000000 8 2.890837 2.036715 0.000000 6 4.464593 0.789029 -1.289461 6 4.464593 0.789029 1.289461 1 3.792548 0.974367 -2.125904 1 3.792548 0.974367 2.125904 1 4.747151 -0.263458 -1.251373 1 5.383563 1.365625 -1.437819 1 5.383563 1.365625 1.437819 1 4.747151 -0.263458 1.251373	6 2.275138 -1.669613 0.000000 8 1.559044 -0.680696 0.000000 6 2.713360 -2.337223 1.285094 6 2.713360 -2.337223 -1.285094 1 2.419518 -1.732657 2.141392 1 2.419518 -1.732657 -2.141392 1 3.793216 -2.504683 1.291024 1 3.793216 -2.504683 -1.291024 1 2.237195 -3.320017 1.353798 1 2.237195 -3.320017 -1.353798 1 3.793216 -2.504683 -1.291024 6 3.646428 1.474027 0.000000 8 2.752515 2.301408 0.000000 6 4.260821 0.972207 -1.288262 6 4.260821 0.972207 1.288262 1 3.618363 1.228989 -2.128874 1 3.618363 1.228989 2.128874 1 4.431939 -0.104710 -1.257439 1 5.235822 1.452251 -1.420882 1 5.235822 1.452251 1.420882 1 4.431939 -0.104710 1.257439	6 2.202258 -1.710331 0.000000 8 1.572153 -0.664536 0.000000 6 2.603948 -2.398435 1.285270 6 2.603948 -2.398435 -1.285270 1 2.293823 -1.802608 2.141970 1 2.293823 -1.802608 -2.141970 1 3.685431 -2.558130 1.310057 1 2.134135 -3.384866 1.334030 1 2.134135 -3.384866 -1.334030 1 3.685431 -2.558130 -1.310057 6 3.457259 1.669212 0.000000 8 2.573895 2.508192 0.000000 6 4.060589 1.151200 -1.286951 6 4.060589 1.151200 1.286951 1 3.452292 1.465023 -2.133468 1 3.452292 1.465023 2.133468 1 4.147016 0.063624 -1.269940 1 5.070939 1.558383 -1.395024 1 5.070939 1.558383 1.395024 1 4.147016 0.063624 1.269940

  Acetone Dimer $O \cdots C = 3.0 \text{ \AA}$ $C=O \cdots C = 120^\circ$ MP2/6-311+G(2d,p)	  Acetone Dimer $O \cdots C = 3.0 \text{ \AA}$ $C=O \cdots C = 130^\circ$ MP2/6-311+G(2d,p)	  Acetone Dimer $O \cdots C = 3.0 \text{ \AA}$ $C=O \cdots C = 140^\circ$ MP2/6-311+G(2d,p)
6 2.126467 -1.749510 0.000000	6 2.037119 -1.786654 0.000000	6 1.941118 -1.818193 0.000000
8 1.585407 -0.654953 0.000000	8 1.590275 -0.650219 0.000000	8 1.592262 -0.647830 0.000000
6 2.480387 -2.461385 1.285745	6 2.331092 -2.524388 1.285913	6 2.169243 -2.578685 1.285785
6 2.480387 -2.461385 -1.285745	6 2.331092 -2.524388 -1.285913	6 2.169243 -2.578685 -1.285785
1 2.183129 -1.857984 2.141709	1 2.078041 -1.900607 2.141473	1 1.972521 -1.934936 2.141326
1 2.183129 -1.857984 -2.141709	1 2.078040 -1.900606 -2.141473	1 1.972521 -1.934936 -2.141326
1 3.555746 -2.657693 1.322077	1 3.388650 -2.800336 1.324808	1 3.198182 -2.946903 1.324318
1 1.976611 -3.431003 1.324309	1 1.755822 -3.453339 1.322555	1 1.513796 -3.452883 1.322554
1 1.976611 -3.431003 -1.324309	1 1.755822 -3.453339 -1.322556	1 1.513796 -3.452883 -1.322554
1 3.555746 -2.657693 -1.322077	1 3.388650 -2.800336 -1.324808	1 3.198182 -2.946903 -1.324318
6 3.249769 1.841025 0.000000	6 3.023367 1.985354 0.000000	6 2.783792 2.105396 0.000000
8 2.344479 2.656758 0.000000	8 2.071894 2.747238 0.000000	8 1.776503 2.792299 0.000000
6 3.863592 1.333499 -1.286001	6 3.663551 1.510796 -1.285419	6 3.456825 1.678428 -1.284958
6 3.863592 1.333499 1.286001	6 3.663552 1.510796 1.285419	6 3.456825 1.678428 1.284958
1 3.272987 1.669814 -2.136365	1 3.070685 1.838234 -2.137672	1 2.850667 1.977725 -2.138226
1 3.272987 1.669814 2.136365	1 3.070685 1.838234 2.137672	1 2.850667 1.977725 2.138226
1 3.922747 0.243843 -1.280647	1 3.749325 0.422919 -1.287695	1 3.600785 0.596756 -1.292097
1 4.883917 1.719525 -1.374327	1 4.673875 1.924445 -1.361204	1 4.443192 2.147342 -1.352612
1 4.883917 1.719525 1.374327	1 4.673875 1.924444 1.361204	1 4.443192 2.147342 1.352612
1 3.922747 0.243843 1.280647	1 3.749325 0.422919 1.287695	1 3.600785 0.596756 1.292097

  Acetone Dimer $O \cdots C = 3.0 \text{ \AA}$ $C=O \cdots C = 150^\circ$ MP2/6-311+G(2d,p)	  Acetone Dimer $O \cdots C = 3.0 \text{ \AA}$ $C=O \cdots C = 160^\circ$ MP2/6-311+G(2d,p)	  Acetone Dimer $O \cdots C = 3.0 \text{ \AA}$ $C=O \cdots C = 170^\circ$ MP2/6-311+G(2d,p)
6 1.839376 -1.842782 0.000000	6 1.732886 -1.859877 0.000000	6 1.623644 -1.868985 0.000000
8 1.591866 -0.646814 0.000000	8 1.589680 -0.646992 0.000000	8 1.586586 -0.648273 0.000000
6 1.999057 -2.620592 1.285558	6 1.822922 -2.648872 1.285342	6 1.643575 -2.662848 1.285179
6 1.999057 -2.620592 -1.285558	6 1.822922 -2.648872 -1.285342	6 1.643575 -2.662848 -1.285179
1 1.862101 -1.961744 2.141211	1 1.745356 -1.980639 2.141108	1 1.625169 -1.990504 2.140995
1 1.862101 -1.961744 -2.141211	1 1.745356 -1.980639 -2.141108	1 1.625169 -1.990504 -2.140995
1 2.990259 -3.080625 1.323288	1 2.769109 -3.195414 1.322842	1 2.537853 -3.290661 1.322709
1 1.266700 -3.431434 1.322620	1 1.020967 -3.390951 1.322127	1 0.779139 -3.331102 1.321570
1 1.266700 -3.431434 -1.322620	1 1.020967 -3.390951 -1.322127	1 0.779139 -3.331102 -1.321570
1 2.990259 -3.080625 -1.323288	1 2.769109 -3.195414 -1.322842	1 2.537853 -3.290661 -1.322709
6 2.534214 2.201340 0.000000	6 2.278119 2.272948 0.000000	6 2.017653 2.320596 0.000000
8 1.471195 2.798789 0.000000	8 1.164709 2.770777 0.000000	8 0.862092 2.711534 0.000000
6 3.240540 1.831153 -1.284378	6 3.014409 1.966326 -1.283982	6 2.778602 2.082256 -1.283808
6 3.240540 1.831153 1.284378	6 3.014409 1.966326 1.283982	6 2.778602 2.082256 1.283808
1 2.617856 2.091388 -2.138576	1 2.375241 2.180771 -2.138758	1 2.125077 2.246825 -2.138826
1 2.617856 2.091388 2.138576	1 2.375241 2.180771 2.138758	1 2.125077 2.246825 2.138826
1 3.457893 0.761784 -1.294566	1 3.313266 0.916795 -1.296955	1 3.161928 1.060559 -1.299804
1 4.192344 2.367282 -1.345868	1 3.921987 2.574651 -1.340704	1 3.633425 2.762933 -1.336596
1 4.192344 2.367282 1.345868	1 3.921987 2.574651 1.340704	1 3.633425 2.762933 1.336596
1 3.457893 0.761784 1.294566	1 3.313266 0.916795 1.296955	1 3.161928 1.060559 1.299804

<p>Acetone Dimer $O \cdots C = 2.9 \text{ \AA}$ $C=O \cdots C = 90^\circ$ MP2/6-311+G(2d,p)</p>	<p>Acetone Dimer $O \cdots C = 2.9 \text{ \AA}$ $C=O \cdots C = 100^\circ$ MP2/6-311+G(2d,p)</p>	<p>Acetone Dimer $O \cdots C = 2.9 \text{ \AA}$ $C=O \cdots C = 110^\circ$ MP2/6-311+G(2d,p)</p>
6 2.375549 -1.590095 0.000000 8 1.582235 -0.661337 0.000000 6 2.834316 -2.245479 1.285362 6 2.834316 -2.245479 -1.285362 1 2.575457 -1.622874 2.139848 1 2.575457 -1.622874 -2.139848 1 3.905570 -2.455530 1.273946 1 2.319822 -3.207630 1.376414 1 2.319822 -3.207630 -1.376414 1 3.905570 -2.455530 -1.273946 6 3.787330 1.222159 0.000000 8 2.857683 2.008413 0.000000 6 4.438614 0.773245 -1.290426 6 4.438614 0.773245 1.290426 1 3.751914 0.921186 -2.122454 1 3.751914 0.921186 2.122454 1 4.771708 -0.263785 -1.246643 1 5.327059 1.392601 -1.452519 1 5.327059 1.392601 1.452519 1 4.771708 -0.263785 1.246643	6 2.309888 -1.632120 0.000000 8 1.595784 -0.641305 0.000000 6 2.742175 -2.303960 1.284878 6 2.742175 -2.303960 -1.284878 1 2.456947 -1.695821 2.141498 1 2.456947 -1.695821 -2.141498 1 3.819470 -2.486730 1.290114 1 2.252107 -3.279976 1.353850 1 2.252107 -3.279976 -1.353850 1 3.819470 -2.486730 -1.290114 6 3.618248 1.437069 0.000000 8 2.714266 2.253116 0.000000 6 4.243982 0.951671 -1.289291 6 4.243982 0.951671 1.289291 1 3.578444 1.162603 -2.124785 1 3.578444 1.162603 2.124785 1 4.480054 -0.112351 -1.250291 1 5.186502 1.488414 -1.439383 1 5.186502 1.488414 1.439383 1 4.480054 -0.112351 1.250291	6 2.223045 -1.676792 0.000000 8 1.597184 -0.628253 0.000000 6 2.619286 -2.368277 1.285137 6 2.619286 -2.368277 -1.285137 1 2.317902 -1.768304 2.142035 1 2.317902 -1.768304 -2.142035 1 3.698488 -2.542495 1.308568 1 2.136363 -3.348391 1.334980 1 2.136363 -3.348391 -1.334980 1 3.698488 -2.542495 -1.308568 6 3.428802 1.620118 0.000000 8 2.529344 2.441455 0.000000 6 4.046876 1.121430 -1.287849 6 4.046876 1.121430 1.287849 1 3.408432 1.383730 -2.129745 1 3.408432 1.383730 2.129745 1 4.209391 0.043017 -1.259660 1 5.025032 1.596486 -1.415217 1 5.025032 1.596486 1.415217 1 4.209391 0.043017 1.259660

<p>Acetone Dimer $O \cdots C = 2.9 \text{ \AA}$ $C=O \cdots C = 120^\circ$ MP2/6-311+G(2d,p)</p>	<p>Acetone Dimer $O \cdots C = 2.9 \text{ \AA}$ $C=O \cdots C = 130^\circ$ MP2/6-311+G(2d,p)</p>	<p>Acetone Dimer $O \cdots C = 2.9 \text{ \AA}$ $C=O \cdots C = 140^\circ$ MP2/6-311+G(2d,p)</p>
6 2.140994 -1.715325 0.000000 8 1.604611 -0.618391 0.000000 6 2.490688 -2.429270 1.285715 6 2.490688 -2.429270 -1.285715 1 2.200296 -1.822489 2.141633 1 2.200296 -1.822489 -2.141633 1 3.564014 -2.636505 1.320835 1 1.977226 -3.393822 1.325583 1 1.977226 -3.393822 -1.325583 1 3.564014 -2.636505 -1.320835 6 3.223819 1.787469 0.000000 8 2.298266 2.579670 0.000000 6 3.855090 1.302623 -1.286615 6 3.855090 1.302623 1.286615 1 3.232375 1.585349 -2.133603 1 3.232375 1.585349 2.133603 1 3.995591 0.220726 -1.269396 1 4.843052 1.761953 -1.393045 1 4.843052 1.761953 1.393045 1 3.995591 0.220726 1.269396	6 2.046549 -1.751529 0.000000 8 1.604770 -0.613111 0.000000 6 2.337122 -2.490530 1.285958 6 2.337121 -2.490530 -1.285958 1 2.088164 -1.864914 2.141378 1 2.088163 -1.864913 -2.141378 1 3.393059 -2.772744 1.324429 1 1.756440 -3.416109 1.323218 1 1.756440 -3.416109 -1.323218 1 3.393059 -2.772744 -1.324430 6 3.001429 1.928413 0.000000 8 2.029246 2.663208 0.000000 6 3.659032 1.478703 -1.285854 6 3.659032 1.478702 1.285854 1 3.035736 1.751446 -2.135637 1 3.035736 1.751445 2.135637 1 3.826834 0.400645 -1.276009 1 4.634985 1.965596 -1.378541 1 4.634986 1.965595 1.378541 1 3.826835 0.400645 1.276009	6 1.946699 -1.782294 0.000000 8 1.603023 -0.610410 0.000000 6 2.171757 -2.543563 1.285862 6 2.171757 -2.543563 -1.285862 1 1.978112 -1.898615 2.141207 1 1.978112 -1.898615 -2.141207 1 3.199054 -2.916432 1.324308 1 1.512432 -3.414836 1.323030 1 1.512432 -3.414836 -1.323030 1 3.199054 -2.916432 -1.324308 6 2.766592 2.045924 0.000000 8 1.740265 2.703442 0.000000 6 3.455675 1.645458 -1.285260 6 3.455675 1.645458 1.285260 1 2.822623 1.889729 -2.136512 1 2.822623 1.889729 2.136512 1 3.679795 0.577651 -1.280775 1 4.404309 2.184958 -1.368678 1 4.404309 2.184958 1.368678 1 3.679795 0.577651 1.280775

Acetone Dimer O···C = 2.9 Å C=O···C = 150° MP2/6-311+G(2d,p)	Acetone Dimer O···C = 2.9 Å C=O···C = 160° MP2/6-311+G(2d,p)	Acetone Dimer O···C = 2.9 Å C=O···C = 170° MP2/6-311+G(2d,p)
6 1.841730 -1.806528 0.000000 8 1.599419 -0.609512 0.000000 6 1.998197 -2.584879 1.285637 6 1.998197 -2.584879 -1.285637 1 1.864513 -1.925120 2.141116 1 1.864513 -1.925120 -2.141116 1 2.987247 -3.049576 1.323233 1 1.262067 -3.392291 1.323116 1 1.262067 -3.392291 -1.323116 1 2.987247 -3.049576 -1.323233 6 2.522295 2.139724 0.000000 8 1.442328 2.705378 0.000000 6 3.242580 1.797990 -1.284745 6 3.242580 1.797990 1.284745 1 2.597442 2.003510 -2.137182 1 2.597442 2.003510 2.137182 1 3.537324 0.747486 -1.284519 1 4.152818 2.400825 -1.360711 1 4.152818 2.400825 1.360711 1 3.537324 0.747486 1.284519	6 1.732088 -1.823353 0.000000 8 1.594134 -0.609870 0.000000 6 1.818671 -2.612638 1.285415 6 1.818671 -2.612638 -1.285415 1 1.744922 -1.943753 2.141024 1 1.744922 -1.943753 -2.141024 1 2.761956 -3.164210 1.322646 1 1.012794 -3.350447 1.322733 1 1.012794 -3.350447 -1.322733 1 2.761956 -3.164210 -1.322646 6 2.271823 2.209835 0.000000 8 1.144174 2.673834 0.000000 6 3.019629 1.933038 -1.284316 6 3.019629 1.933038 1.284316 1 2.362949 2.093506 -2.137640 1 2.362949 2.093506 2.137640 1 3.391834 0.907407 -1.287796 1 3.882089 2.602968 -1.354160 1 3.882089 2.602968 1.354160 1 3.391834 0.907407 1.287796	6 1.619617 -1.832340 0.000000 8 1.587864 -0.611483 0.000000 6 1.635852 -2.626194 1.285246 6 1.635852 -2.626194 -1.285246 1 1.621538 -1.953531 2.140904 1 1.621538 -1.953531 -2.140904 1 2.526563 -3.259085 1.322497 1 0.767628 -3.289510 1.322184 1 0.767628 -3.289510 -1.322184 1 2.526563 -3.259085 -1.322497 6 2.017022 2.256587 0.000000 8 0.850264 2.611750 0.000000 6 2.787028 2.049012 -1.283950 6 2.787028 2.049012 1.283950 1 2.121208 2.160820 -2.137989 1 2.121208 2.160820 2.137989 1 3.238690 1.055778 -1.290934 1 3.593865 2.785348 -1.348363 1 3.593865 2.785348 1.348363 1 3.238690 1.055778 1.290934

Formaldehyde Dimer O···C = 2.72 Å C=O···C = 100° MP2/6-311+G(2d,p)	Formaldehyde Dimer O···C = 2.72 Å C=O···C = 110° MP2/6-311+G(2d,p)	Formaldehyde Dimer O···C = 2.72 Å C=O···C = 120° MP2/6-311+G(2d,p)
6 -1.181638 1.184331 0.000000 8 0.000000 1.470196 0.000000 1 -1.956514 1.967845 0.000000 1 -1.517021 0.134721 0.000000 6 1.108516 -1.017994 0.000000 8 0.089337 -1.680452 0.000000 1 1.598785 -0.709271 0.936595 1 1.598785 -0.709271 -0.936595	6 -1.198149 1.287641 0.000000 8 0.002456 1.470827 0.000000 1 -1.907525 2.131007 0.000000 1 -1.618205 0.267951 0.000000 6 1.309512 -0.919048 0.000000 8 0.363470 -1.681348 0.000000 1 1.765555 -0.560886 0.936641 1 1.765555 -0.560886 -0.936641	6 -1.202532 1.402322 0.000000 8 0.009626 1.472124 0.000000 1 -1.830934 2.307883 0.000000 1 -1.719077 0.428162 0.000000 6 1.504962 -0.804687 0.000000 8 0.705013 -1.718279 0.000000 1 1.893194 -0.372963 0.936346 1 1.893194 -0.372963 -0.936346
Formaldehyde Dimer O···C = 2.72 Å C=O···C = 130° MP2/6-311+G(2d,p)	Formaldehyde Dimer O···C = 2.72 Å C=O···C = 140° MP2/6-311+G(2d,p)	Formaldehyde Dimer O···C = 2.72 Å C=O···C = 150° MP2/6-311+G(2d,p)
6 -1.198808 1.517742 0.000000 8 0.014544 1.474251 0.000000 1 -1.740223 2.477980 0.000000 1 -1.806090 0.597748 0.000000 6 1.689601 -0.673790 0.000000 8 1.064696 -1.714552 0.000000 1 1.995482 -0.179873 0.936133 1 1.995482 -0.179873 -0.936133	6 -1.187993 1.630992 0.000000 8 0.016287 1.476736 0.000000 1 -1.639451 2.636719 0.000000 1 -1.877801 0.771336 0.000000 6 1.863598 -0.525097 0.000000 8 1.412132 -1.651726 0.000000 1 2.086908 0.011477 0.935974 1 2.086908 0.011477 -0.935974	6 -1.169634 1.741399 0.000000 8 0.015825 1.479372 0.000000 1 -1.529057 2.783602 0.000000 1 -1.934392 0.947766 0.000000 6 2.025310 -0.359613 0.000000 8 1.736321 -1.538273 0.000000 1 2.170556 0.203129 0.935955 1 2.170556 0.203129 -0.935955

<p>Formaldehyde Dimer $O \cdots C = 2.72 \text{ \AA}$ $C=O \cdots C = 160^\circ$ MP2/6-311+G(2d,p)</p>	<p>Formaldehyde Dimer $O \cdots C = 2.72 \text{ \AA}$ $C=O \cdots C = 170^\circ$ MP2/6-311+G(2d,p)</p>	<p>Formaldehyde Dimer $O \cdots C = 2.8 \text{ \AA}$ $C=O \cdots C = 100^\circ$ MP2/6-311+G(2d,p)</p>																																																																																																
<table> <tbody> <tr><td>6</td><td>-1.143452</td><td>1.848489</td><td>0.000000</td></tr> <tr><td>8</td><td>0.013935</td><td>1.481971</td><td>0.000000</td></tr> <tr><td>1</td><td>-1.409091</td><td>2.918475</td><td>0.000000</td></tr> <tr><td>1</td><td>-1.975978</td><td>1.126284</td><td>0.000000</td></tr> <tr><td>6</td><td>2.172937</td><td>-0.178932</td><td>0.000000</td></tr> <tr><td>8</td><td>2.032644</td><td>-1.384263</td><td>0.000000</td></tr> <tr><td>1</td><td>2.246429</td><td>0.397519</td><td>0.935956</td></tr> <tr><td>1</td><td>2.246429</td><td>0.397519</td><td>-0.935956</td></tr> </tbody> </table>	6	-1.143452	1.848489	0.000000	8	0.013935	1.481971	0.000000	1	-1.409091	2.918475	0.000000	1	-1.975978	1.126284	0.000000	6	2.172937	-0.178932	0.000000	8	2.032644	-1.384263	0.000000	1	2.246429	0.397519	0.935956	1	2.246429	0.397519	-0.935956	<table> <tbody> <tr><td>6</td><td>-1.109784</td><td>1.950938</td><td>0.000000</td></tr> <tr><td>8</td><td>0.010933</td><td>1.484145</td><td>0.000000</td></tr> <tr><td>1</td><td>-1.280820</td><td>3.040078</td><td>0.000000</td></tr> <tr><td>1</td><td>-2.002550</td><td>1.304672</td><td>0.000000</td></tr> <tr><td>6</td><td>2.305429</td><td>0.016091</td><td>0.000000</td></tr> <tr><td>8</td><td>2.290559</td><td>-1.197211</td><td>0.000000</td></tr> <tr><td>1</td><td>2.318306</td><td>0.597065</td><td>0.935938</td></tr> <tr><td>1</td><td>2.318306</td><td>0.597065</td><td>-0.935938</td></tr> </tbody> </table>	6	-1.109784	1.950938	0.000000	8	0.010933	1.484145	0.000000	1	-1.280820	3.040078	0.000000	1	-2.002550	1.304672	0.000000	6	2.305429	0.016091	0.000000	8	2.290559	-1.197211	0.000000	1	2.318306	0.597065	0.935938	1	2.318306	0.597065	-0.935938	<table> <tbody> <tr><td>6</td><td>-1.197822</td><td>1.216745</td><td>0.000000</td></tr> <tr><td>8</td><td>-0.016929</td><td>1.504252</td><td>0.000000</td></tr> <tr><td>1</td><td>-1.974878</td><td>1.998490</td><td>0.000000</td></tr> <tr><td>1</td><td>-1.532088</td><td>0.166582</td><td>0.000000</td></tr> <tr><td>6</td><td>1.126256</td><td>-1.051746</td><td>0.000000</td></tr> <tr><td>8</td><td>0.094009</td><td>-1.693136</td><td>0.000000</td></tr> <tr><td>1</td><td>1.620853</td><td>-0.750540</td><td>0.936968</td></tr> <tr><td>1</td><td>1.620853</td><td>-0.750540</td><td>-0.936968</td></tr> </tbody> </table>	6	-1.197822	1.216745	0.000000	8	-0.016929	1.504252	0.000000	1	-1.974878	1.998490	0.000000	1	-1.532088	0.166582	0.000000	6	1.126256	-1.051746	0.000000	8	0.094009	-1.693136	0.000000	1	1.620853	-0.750540	0.936968	1	1.620853	-0.750540	-0.936968
6	-1.143452	1.848489	0.000000																																																																																															
8	0.013935	1.481971	0.000000																																																																																															
1	-1.409091	2.918475	0.000000																																																																																															
1	-1.975978	1.126284	0.000000																																																																																															
6	2.172937	-0.178932	0.000000																																																																																															
8	2.032644	-1.384263	0.000000																																																																																															
1	2.246429	0.397519	0.935956																																																																																															
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8	0.010933	1.484145	0.000000																																																																																															
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1	-2.002550	1.304672	0.000000																																																																																															
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8	-0.016929	1.504252	0.000000																																																																																															
1	-1.974878	1.998490	0.000000																																																																																															
1	-1.532088	0.166582	0.000000																																																																																															
6	1.126256	-1.051746	0.000000																																																																																															
8	0.094009	-1.693136	0.000000																																																																																															
1	1.620853	-0.750540	0.936968																																																																																															
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<p>Formaldehyde Dimer $O \cdots C = 2.8 \text{ \AA}$ $C=O \cdots C = 110^\circ$ MP2/6-311+G(2d,p)</p>	<p>Formaldehyde Dimer $O \cdots C = 2.8 \text{ \AA}$ $C=O \cdots C = 120^\circ$ MP2/6-311+G(2d,p)</p>	<p>Formaldehyde Dimer $O \cdots C = 2.8 \text{ \AA}$ $C=O \cdots C = 130^\circ$ MP2/6-311+G(2d,p)</p>																																																																																																
<table> <tbody> <tr><td>6</td><td>-1.215430</td><td>1.320405</td><td>0.000000</td></tr> <tr><td>8</td><td>-0.014819</td><td>1.503381</td><td>0.000000</td></tr> <tr><td>1</td><td>-1.924998</td><td>2.163770</td><td>0.000000</td></tr> <tr><td>1</td><td>-1.636214</td><td>0.301175</td><td>0.000000</td></tr> <tr><td>6</td><td>1.328309</td><td>-0.953448</td><td>0.000000</td></tr> <tr><td>8</td><td>0.381531</td><td>-1.714588</td><td>0.000000</td></tr> <tr><td>1</td><td>1.782145</td><td>-0.592719</td><td>0.936847</td></tr> <tr><td>1</td><td>1.782145</td><td>-0.592719</td><td>-0.936847</td></tr> </tbody> </table>	6	-1.215430	1.320405	0.000000	8	-0.014819	1.503381	0.000000	1	-1.924998	2.163770	0.000000	1	-1.636214	0.301175	0.000000	6	1.328309	-0.953448	0.000000	8	0.381531	-1.714588	0.000000	1	1.782145	-0.592719	0.936847	1	1.782145	-0.592719	-0.936847	<table> <tbody> <tr><td>6</td><td>-1.222699</td><td>1.433617</td><td>0.000000</td></tr> <tr><td>8</td><td>-0.010552</td><td>1.503388</td><td>0.000000</td></tr> <tr><td>1</td><td>-1.851011</td><td>2.339354</td><td>0.000000</td></tr> <tr><td>1</td><td>-1.740287</td><td>0.460060</td><td>0.000000</td></tr> <tr><td>6</td><td>1.526471</td><td>-0.837031</td><td>0.000000</td></tr> <tr><td>8</td><td>0.726267</td><td>-1.750255</td><td>0.000000</td></tr> <tr><td>1</td><td>1.912629</td><td>-0.403766</td><td>0.936611</td></tr> <tr><td>1</td><td>1.912629</td><td>-0.403766</td><td>-0.936611</td></tr> </tbody> </table>	6	-1.222699	1.433617	0.000000	8	-0.010552	1.503388	0.000000	1	-1.851011	2.339354	0.000000	1	-1.740287	0.460060	0.000000	6	1.526471	-0.837031	0.000000	8	0.726267	-1.750255	0.000000	1	1.912629	-0.403766	0.936611	1	1.912629	-0.403766	-0.936611	<table> <tbody> <tr><td>6</td><td>-1.221384</td><td>1.547389</td><td>0.000000</td></tr> <tr><td>8</td><td>-0.008090</td><td>1.503850</td><td>0.000000</td></tr> <tr><td>1</td><td>-1.762770</td><td>2.507723</td><td>0.000000</td></tr> <tr><td>1</td><td>-1.829316</td><td>0.627809</td><td>0.000000</td></tr> <tr><td>6</td><td>1.713638</td><td>-0.704238</td><td>0.000000</td></tr> <tr><td>8</td><td>1.089096</td><td>-1.745146</td><td>0.000000</td></tr> <tr><td>1</td><td>2.016754</td><td>-0.208876</td><td>0.936404</td></tr> <tr><td>1</td><td>2.016754</td><td>-0.208876</td><td>-0.936404</td></tr> </tbody> </table>	6	-1.221384	1.547389	0.000000	8	-0.008090	1.503850	0.000000	1	-1.762770	2.507723	0.000000	1	-1.829316	0.627809	0.000000	6	1.713638	-0.704238	0.000000	8	1.089096	-1.745146	0.000000	1	2.016754	-0.208876	0.936404	1	2.016754	-0.208876	-0.936404
6	-1.215430	1.320405	0.000000																																																																																															
8	-0.014819	1.503381	0.000000																																																																																															
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1	-1.636214	0.301175	0.000000																																																																																															
6	1.328309	-0.953448	0.000000																																																																																															
8	0.381531	-1.714588	0.000000																																																																																															
1	1.782145	-0.592719	0.936847																																																																																															
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6	-1.222699	1.433617	0.000000																																																																																															
8	-0.010552	1.503388	0.000000																																																																																															
1	-1.851011	2.339354	0.000000																																																																																															
1	-1.740287	0.460060	0.000000																																																																																															
6	1.526471	-0.837031	0.000000																																																																																															
8	0.726267	-1.750255	0.000000																																																																																															
1	1.912629	-0.403766	0.936611																																																																																															
1	1.912629	-0.403766	-0.936611																																																																																															
6	-1.221384	1.547389	0.000000																																																																																															
8	-0.008090	1.503850	0.000000																																																																																															
1	-1.762770	2.507723	0.000000																																																																																															
1	-1.829316	0.627809	0.000000																																																																																															
6	1.713638	-0.704238	0.000000																																																																																															
8	1.089096	-1.745146	0.000000																																																																																															
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<p>Formaldehyde Dimer $O \cdots C = 2.8 \text{ \AA}$ $C=O \cdots C = 140^\circ$ MP2/6-311+G(2d,p)</p>	<p>Formaldehyde Dimer $O \cdots C = 2.8 \text{ \AA}$ $C=O \cdots C = 150^\circ$ MP2/6-311+G(2d,p)</p>	<p>Formaldehyde Dimer $O \cdots C = 2.8 \text{ \AA}$ $C=O \cdots C = 160^\circ$ MP2/6-311+G(2d,p)</p>																																																																																																
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6	-1.211474	1.661038	0.000000																																																																																															
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<p>Formaldehyde Dimer $O \cdots C = 2.9 \text{ \AA}$ $C=O \cdots C = 150^\circ$ MP2/6-311+G(2d,p)</p> <table border="1"> <tbody> <tr><td>6</td><td>-1.227854</td><td>1.809185</td><td>0.000000</td></tr> <tr><td>8</td><td>-0.043939</td><td>1.541043</td><td>0.000000</td></tr> <tr><td>1</td><td>-1.582088</td><td>2.853265</td><td>0.000000</td></tr> <tr><td>1</td><td>-1.997558</td><td>1.020136</td><td>0.000000</td></tr> <tr><td>6</td><td>2.085246</td><td>-0.427857</td><td>0.000000</td></tr> <tr><td>8</td><td>1.898098</td><td>-1.627089</td><td>0.000000</td></tr> <tr><td>1</td><td>2.176790</td><td>0.145914</td><td>0.936234</td></tr> <tr><td>1</td><td>2.176790</td><td>0.145914</td><td>-0.936234</td></tr> </tbody> </table>	6	-1.227854	1.809185	0.000000	8	-0.043939	1.541043	0.000000	1	-1.582088	2.853265	0.000000	1	-1.997558	1.020136	0.000000	6	2.085246	-0.427857	0.000000	8	1.898098	-1.627089	0.000000	1	2.176790	0.145914	0.936234	1	2.176790	0.145914	-0.936234	<p>Formaldehyde Dimer $O \cdots C = 2.9 \text{ \AA}$ $C=O \cdots C = 160^\circ$ MP2/6-311+G(2d,p)</p> <table border="1"> <tbody> <tr><td>6</td><td>-1.205204</td><td>1.913885</td><td>0.000000</td></tr> <tr><td>8</td><td>-0.050504</td><td>1.539447</td><td>0.000000</td></tr> <tr><td>1</td><td>-1.463584</td><td>2.985710</td><td>0.000000</td></tr> <tr><td>1</td><td>-2.043293</td><td>1.197920</td><td>0.000000</td></tr> <tr><td>6</td><td>2.235784</td><td>-0.244619</td><td>0.000000</td></tr> <tr><td>8</td><td>2.236656</td><td>-1.458351</td><td>0.000000</td></tr> <tr><td>1</td><td>2.236999</td><td>0.336535</td><td>0.936092</td></tr> <tr><td>1</td><td>2.236999</td><td>0.336535</td><td>-0.936092</td></tr> </tbody> </table>	6	-1.205204	1.913885	0.000000	8	-0.050504	1.539447	0.000000	1	-1.463584	2.985710	0.000000	1	-2.043293	1.197920	0.000000	6	2.235784	-0.244619	0.000000	8	2.236656	-1.458351	0.000000	1	2.236999	0.336535	0.936092	1	2.236999	0.336535	-0.936092	<p>Formaldehyde Dimer $O \cdots C = 2.9 \text{ \AA}$ $C=O \cdots C = 170^\circ$ MP2/6-311+G(2d,p)</p> <table border="1"> <tbody> <tr><td>6</td><td>-1.174628</td><td>2.013717</td><td>0.000000</td></tr> <tr><td>8</td><td>-0.058164</td><td>1.537201</td><td>0.000000</td></tr> <tr><td>1</td><td>-1.336221</td><td>3.104339</td><td>0.000000</td></tr> <tr><td>1</td><td>-2.073486</td><td>1.375724</td><td>0.000000</td></tr> <tr><td>6</td><td>2.370892</td><td>-0.046996</td><td>0.000000</td></tr> <tr><td>8</td><td>2.536904</td><td>-1.249331</td><td>0.000000</td></tr> <tr><td>1</td><td>2.292542</td><td>0.529095</td><td>0.935933</td></tr> <tr><td>1</td><td>2.292542</td><td>0.529095</td><td>-0.935933</td></tr> </tbody> </table>	6	-1.174628	2.013717	0.000000	8	-0.058164	1.537201	0.000000	1	-1.336221	3.104339	0.000000	1	-2.073486	1.375724	0.000000	6	2.370892	-0.046996	0.000000	8	2.536904	-1.249331	0.000000	1	2.292542	0.529095	0.935933	1	2.292542	0.529095	-0.935933
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