## **Supplementary Information**

## Imprinted MoS<sub>2</sub> achieving high-efficient self-separative molecule extraction

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**Supplementary Information S1:** 



Figure S1. Imprinting in animals: a) ducklings on Corgi,<sup>w1</sup> b) new-hatched ducks on hen,<sup>w2</sup> and c) wolf cub on Rottweiler.<sup>w3</sup>

Log in the websites below for more detailed information about imprinting in animals:

w1. Ducklings on Corgi: http://www.thatcutesite.com/two-ducklings-imprint-on-a-

<u>corgi/</u>

w2. Wolf cub on Rottweiler: <u>http://www.dailymail.co.uk/news/article-</u>

1192276/Barking-true-The-touching-bond-Rottweiler-wolf-cub.html

## w3. New-hatched ducks on hen:

https://www.deathandtaxesmag.com/181417/hilda-the-hen-sits-on-the-wrong-nest-

hatches-crazy-cute-baby-ducks/

## **Supplementary Information S2:**

Sips model which shares both attributes of Langmuir and Freundlich models is defined as follow:

$$q_{e} = \frac{q_{m}K_{s}C_{e}^{n_{s}}}{1 + K_{s}C_{e}^{n_{s}}}$$
(1)

Here,  $q_{\rm m}$  is the maximum adsorption capacity,  $q_{\rm e}$  is the equilibrium adsorption capacity,  $C_{\rm e}$  is the equilibrium concentration,  $K_{\rm s}$  is the Sips isotherm constant related to the energy of adsorption, and  $n_{\rm s}$  is the sorbent surface heterogeneity parameter. If the value of  $n_{\rm s}$  is unity, the Sips model describes the typical Langmuir adsorption behavior, otherwise the Sips model reflects a complex multilayer adsorption ( $n_{\rm s}$  not equal to unity). Optionally, as the value of  $C_{\rm e}$  or  $K_{\rm s}$  approaches zero, the Sips isotherm describes Freundlich isotherm behavior.



Figure S2. Xps spectra of pristine  $MoS_2$  (red line),  $MoS_2$  with adsorbed RhB (black line), and  $MoS_2$  with adsorbed LA (blue line): a) Element scanning, b) Mo 3d, and c) S 2p.



Figure S3. SEM image of as-prepared flower-like MoS<sub>2</sub>.



**Figure S4.** HRTEM of the (a) edge and (b) plane of  $MoS_2$  nanosheet, and (c) Mo and S element mapping on  $MoS_2$  edge.



Figure S5. Water contact angle (WCA) test of n-hexane- (top graph) and water-

imprinted MoS<sub>2</sub> (bottom graph) versus time.



Figure S6. <sup>1</sup>H solid-state MAS NMR spectra of pristine, water- and n-hexane-

imprinted MoS<sub>2</sub>.



Figure S7. Recycling ability of imprinted  $MoS_2$  for heterophasic adsorption.



Figure S8. Imprinted MoS<sub>2</sub> d) wrapped by water droplet submerged in n-hexane and

e) wrapped by n-hexane droplet submerged in water.



Figure S9. Fourier transformed infrared spectra (FTIR) of lauric acid (LA), LA adsorbed by  $MoS_2$  in monophase (both  $MoS_2$  and LA in n-hexane) and heterophase (MoS<sub>2</sub> in water but LA in n-hexane).

Figure S10 indicates that  $MoS_2$  soaked by water and n-hexane have no change in its flower-like morphology, compared with as-prepared  $MoS_2$ .



Figure S10. SEM images of a) as-prepared  $MoS_2$ , b) water-soaked  $MoS_2$ , and c) n-hexane-soaked  $MoS_2$ .