

## **Interface-Engineered Hematite Nanocones as Binder-Free Electrode for High-Performance Lithium-Ion Batteries**

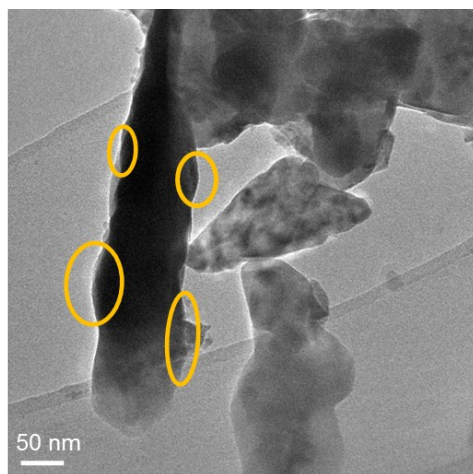
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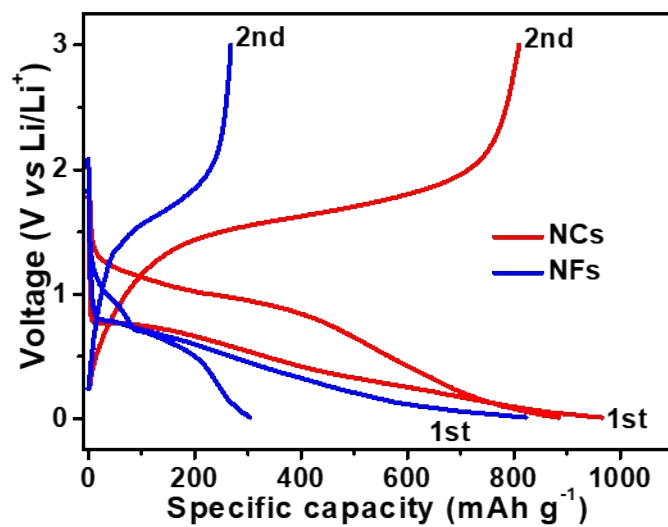
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**Figure S1.** TEM image of  $\text{Fe}_2\text{O}_3$  NCs with a knoblike structure on the surface.



**Figure S2.** The first two discharge/charge cycling profiles of NCs and NFs at a current density of  $0.125 \text{ A g}^{-1}$ .

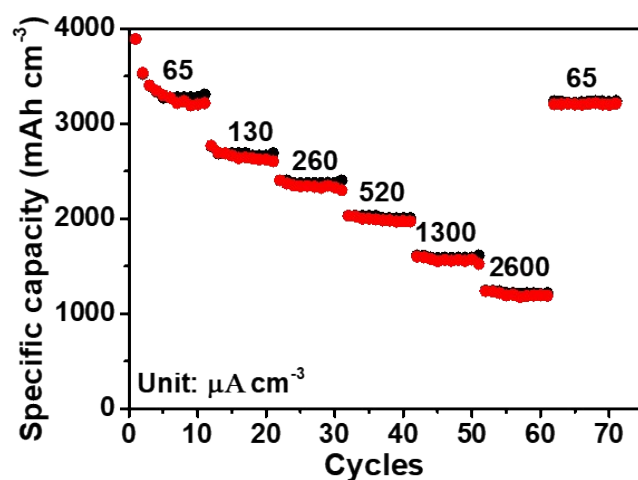


Figure S3. Rate performance.

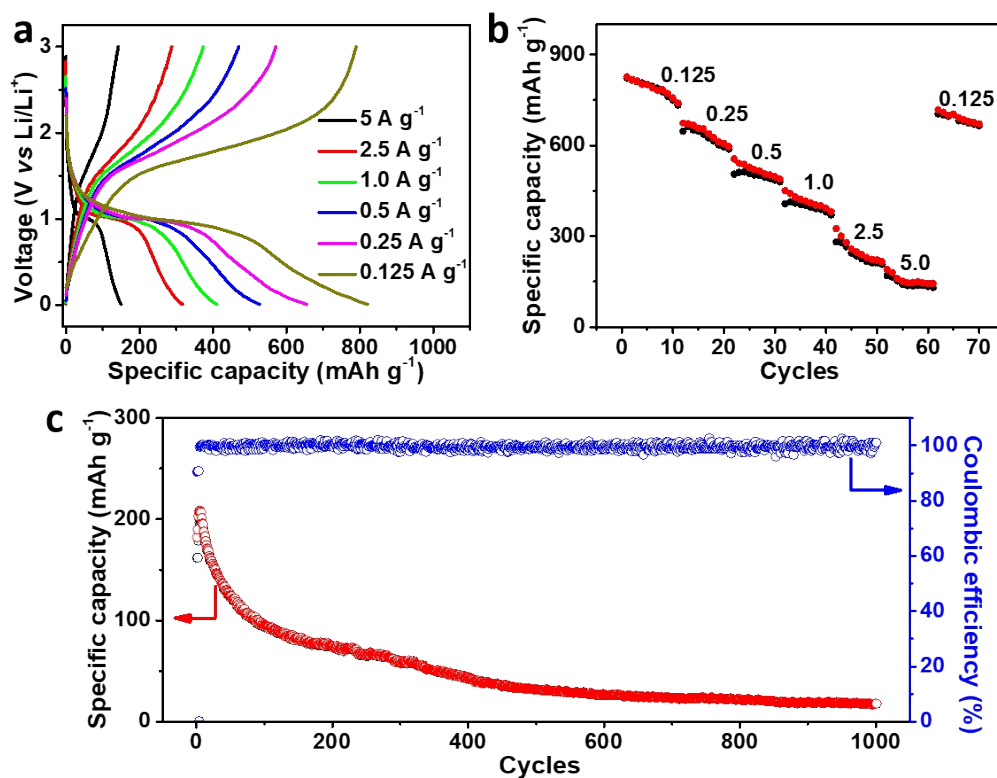
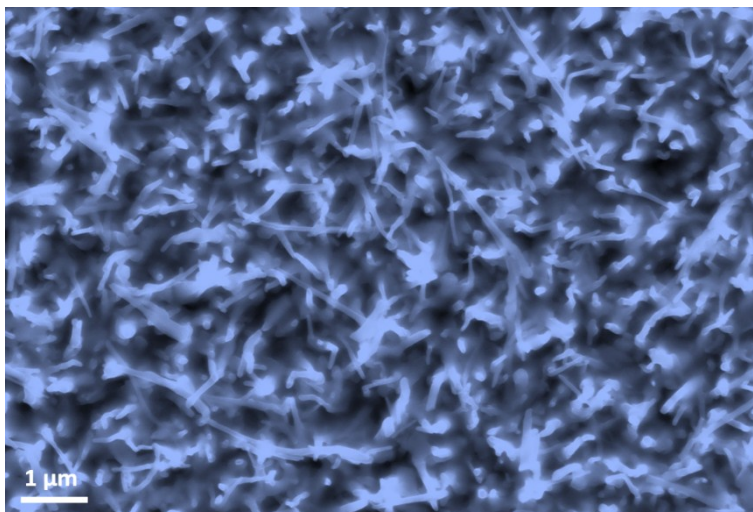
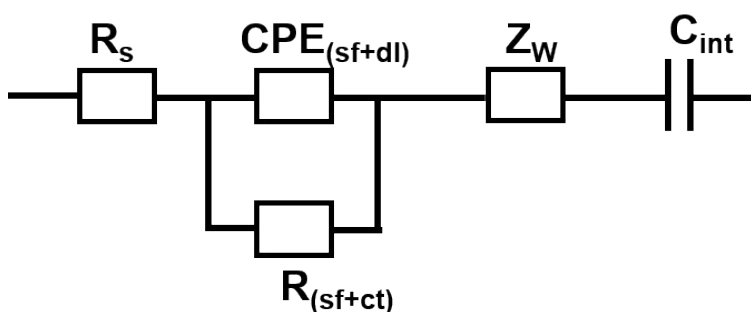


Figure S4. Electrochemical performance of NFs electrode. (a) Charge-discharge profiles at different current densities. (b) Rate performance. (c) Long-term cyclability at a current density of 5  $\text{A g}^{-1}$ .



**Figure S5.** SEM image of NCs electrode after long-term cycling test.



**Figure S6.** Equivalent electric circuit. The interception at the real axis in the high-frequency area means solution resistance ( $R_s$ ), which reflects the total resistance between electrolyte and electrode. The diameter of the semicircle or quasi-semicircle in medium frequency combines two resistances, involving Li-ion migration resistance ( $R_{sf}$ ) and charge-transfer resistance ( $R_{ct}$ ). The straight line in the low-frequency region represents the Warburg impedance ( $Z_w$ ), corresponding to Lithium-ion diffusion in the solid.

**Table S1. Comparison with the state-of-the-art Fe<sub>2</sub>O<sub>3</sub> anodes.**

Anode materials	Current density	Specific capacity (mAh g <sup>-1</sup> )	Areal capacity (mAh cm <sup>-2</sup> )	Volumetric capacity (mAh cm <sup>-3</sup> )	Reference
<b><math>\alpha</math>-Fe<sub>2</sub>O<sub>3</sub> NCs</b>	<b>0.125 A g<sup>-1</sup></b>	<b>968</b>	<b>0.774</b>	<b>3872</b>	<b>This work</b>
Fe <sub>2</sub> O <sub>3</sub> /Fe <sub>3</sub> C-graphene	50 $\mu$ A cm <sup>-2</sup>	-	0.427	3560	1
$\alpha$ -Fe <sub>2</sub> O <sub>3</sub> nanoflakes	68 mA g <sup>-1</sup>	680 $\pm$ 20	-	-	2
$\alpha$ -Fe <sub>2</sub> O <sub>3</sub> @CNF	50 mA g <sup>-1</sup>	604	-	-	3
Fe <sub>2</sub> O <sub>3</sub> -C	1.0 A g <sup>-1</sup>	812	-	-	4
Fe <sub>2</sub> O <sub>3</sub> @PANI	0.1 C	893	-	-	5
Graphene- Fe <sub>2</sub> O <sub>3</sub>	160 mA g <sup>-1</sup>	660	-	-	6
Fe <sub>2</sub> O <sub>3</sub> nanorod-C	0.2 C	758	-	-	7
$\alpha$ -Fe <sub>2</sub> O <sub>3</sub> -C	0.2 C	688	-	-	8
$\alpha$ -Fe <sub>2</sub> O <sub>3</sub>	1C	-	0.3831	-	9
C- $\alpha$ -Fe <sub>2</sub> O <sub>3</sub>	5.04 A g <sup>-1</sup>	420	0.557	-	10
Fe <sub>2</sub> O <sub>3</sub>	270 mA g <sup>-1</sup>	-	-	570	11
Fe <sub>2</sub> O <sub>3</sub> -graphite	200 mA g <sup>-1</sup>	-	-	1014	12
Fe <sub>2</sub> O <sub>3</sub> NPs	50 $\mu$ A cm <sup>-2</sup>	-	-	800	13

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