

**Ultrathin IrRu Nanowires Network with high performance and durability for
Hydrogen Oxidation Reaction in alkaline anion exchange membrane fuel cell**

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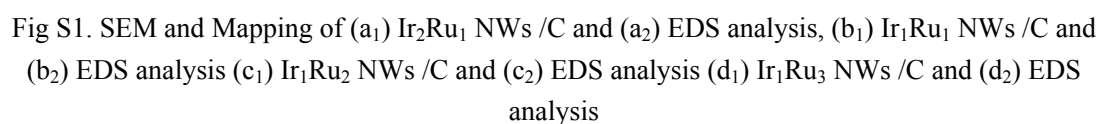


Fig S1. SEM and Mapping of (a₁) Ir₂Ru₁ NWs /C and (a₂) EDS analysis, (b₁) Ir₁Ru₁ NWs /C and (b₂) EDS analysis (c₁) Ir₁Ru₂ NWs /C and (c₂) EDS analysis (d₁) Ir₁Ru₃ NWs /C and (d₂) EDS analysis

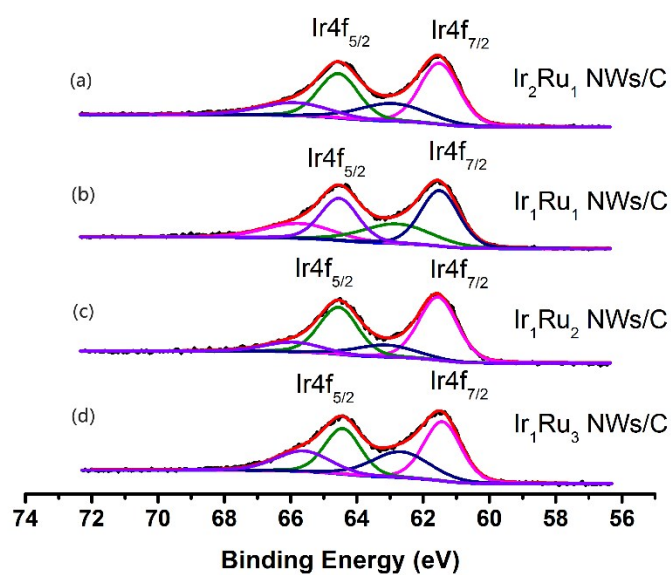


Fig S2. XPS spectra of Ir 4f for (a) Ir₂Ru₁ NWs/C, (b) Ir₁Ru₁ NWs/C, (c) Ir₁Ru₂ NWs/C, (d) Ir₁Ru₃ NWs/C.

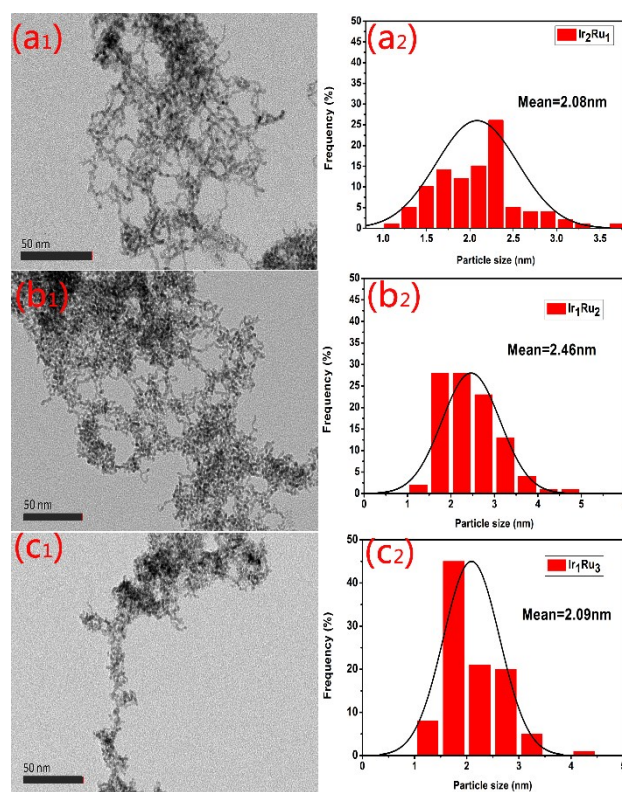


Fig S3. TEM images and the corresponding size distribution diagrams of Ir₂Ru₁ NWs (a₁ and a₂), Ir₁Ru₂ NWs (b₁ and b₂) and Ir₁Ru₃ NWs (c₁ and c₂).

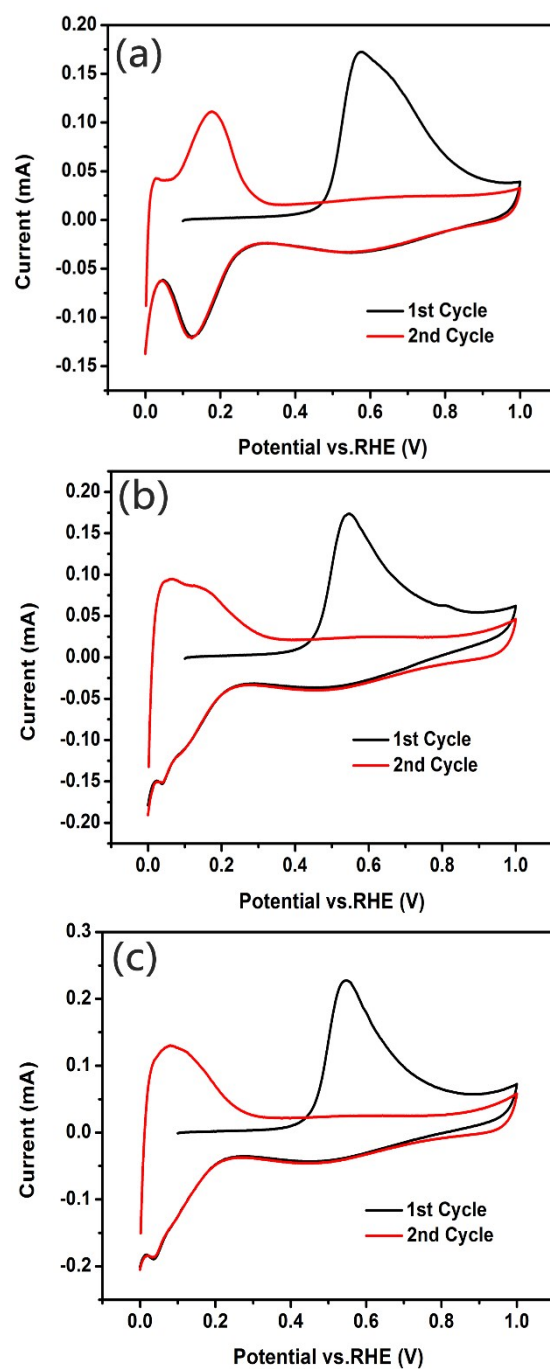


Fig S4. CO-stripping voltammetry for of Ir_2Ru_1 NWs (a), Ir_1Ru_2 NWs (b) and Ir_1Ru_3 NWs (c).

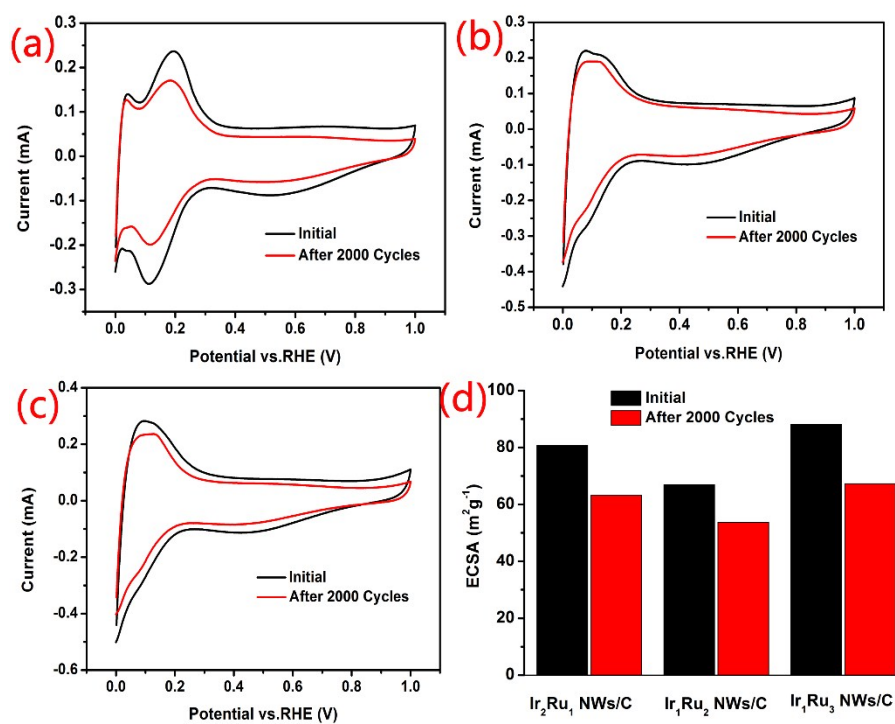


Fig S5. Cyclic voltammograms of Ir₂Ru₁ NWs (a), Ir₁Ru₂ NWs (b) and Ir₁Ru₃ NWs (c) and the corresponding ECSA (d) before and after half-cell-ADTs .

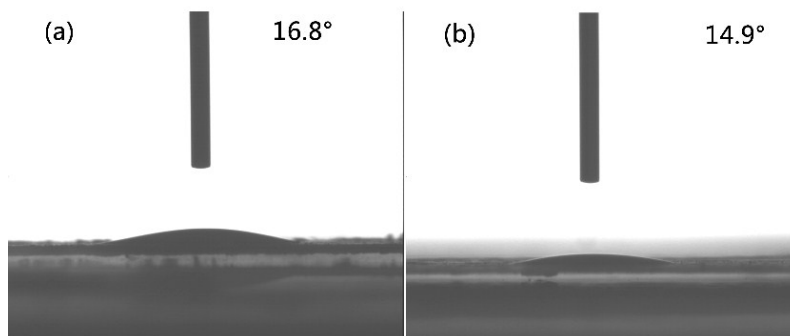


Fig S6. Contact angle of GDE with (a) Pt/C and (b) Ir₁Ru₁ NWs /C as anode catalysts

Table S1. The compositions of four different IrRu NWs/C catalysts.

catalyst	Nominal Ir:Ru (atomic)	Composition from EDS analysis			Composition from ICP analysis		
		Ir (wt %)	Ru (wt %)	Ir:Ru (atomic)	Ir (wt %)	Ru (wt %)	Ir:Ru (atomic)
	2:1	15.96	4.93	60.21:39.79	13.30	4.54	60.63:39.37
IrRu	1:1	14.99	6.68	51.05:48.95	14.31	6.79	52.55:47.45
NWs/C	1:2	9.88	9.55	32.94:67.06	8.96	9.41	33.37:66.63
	1:3	8.47	11.67	25.89:74.11	7.33	10.87	26.16:73.84

Table S2 Exchange current density, mass activity@50mV, Specific activity @50mV, for the HOR in base electrolytes on different electrocatalytic materials.

Material	Experiment Conditions	Method	i_0 (mA cm _{meta} ⁻²)	Mass activity@ 50mV (A g ⁻¹)	Specific activity@50 mV (mA cm _{metal} ⁻²)	Ref.
Pt (110)	0.1 M KOH, 25 °C,	micro	0.7	---	1.0	1
Pt (110)	0.1 M KOH, 25 °C,	micro	0.05	---	0.068	1
Pt(111)	0.1 M KOH, 25 °C,	micro	0.04	---	0.068	1
46% Pt/C (Tanaka)	0.1 M KOH, 25 °C	micro	0.05	---	0.34	2
50% Ru/C (2.2 nm)	0.1 M KOH, 25 °C	micro	0.030	---	---	3
50% Ru/C (2.4 nm)	0.1 M KOH, 25 °C	micro	0.043	---	---	3
50% Ru/C (3.1 nm)	0.1 M KOH, 25 °C	micro	0.063	162	0.64	3
50% Ru/C (3.4 nm)	0.1 M KOH, 25 °C	micro	0.056	---	---	3
50% Ru/C (3.9 nm)	0.1 M KOH, 25 °C	micro	0.044	---	---	3

50% Ru/C (4.1 nm)	0.1 M KOH, 25 °C	micro	0.045	---	---	3
50% Ru/C (5.1 nm)	0.1 M KOH, 25 °C	micro	0.046	---	---	3
50% Ru/C (6.6 nm)	0.1 M KOH, 25 °C	micro	0.038	---	---	3
20% Ir/C (Premetek)	0.1 M NaOH, 40 °C	micro	0.38	448.4	0.76	4
20% Ir/C (Premetek)	0.1 M KOH, 20 °C	---	---	313	0.49	5
Pt/C	0.1 M KOH, 25 °C	micro	0.107	338	0.712	This work
Ir ₁ Ru ₁ NWs/C	0.1 M KOH, 25 °C	micro	0.126	1416	2.682	This work

References:

1. N. M. Markovic, S. T. Sarraf, H. A. Gasteiger and P. N. Ross, *Journal of the Chemical Society-Faraday Transactions*, 1996, **92**, 3719-3725.
2. G. Couturier, D. W. Kirk, P. J. Hyde and S. Srinivasan, *Electrochimica Acta*, 1987, **32**, 995-1005.
3. J. Ohyama, T. Sato, Y. Yamamoto, S. Arai and A. Satsuma, *Journal of the American Chemical Society*, 2013, **135**, 8016-8021.
4. J. Durst, A. Siebel, C. Simon, F. Hasché, J. Herranz and H. A. Gasteiger, *Energy & Environmental Science*, 2014, **7**, 2255.
5. J. Zheng, Z. Zhuang, B. Xu and Y. Yan, *Acs Catalysis*, 2015, **5**, 4449-4455.