

Supporting Information

Design and Synthesis of Room Temperature Stable Li-Argyrodite Superionic Conductors *via* Cation Doping

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Table S1. Energy difference (ΔE) of various ionic substitutions (Al, B, Ge, Si) argyrodites and pure Li₇PS₆ between HT and LT phase

<i>substitution</i>	E_{HT} (eV/Å)	E_{LT} (eV/Å)	$\Delta E(E_{HT}-E_{LT})$ (eV/Å)	<i>Stable phase</i>
None	-234.79627	-234.82979	-0.03352	LT phase
Si	-234.95776	-234.78552	0.17224	HT phase
Ge	-230.67633	-230.52328	0.15305	HT phase
Al	-230.05914	-231.27840	-1.21926	LT phase
B	-235.19432	-235.28478	-0.09046	LT phase

Table S2. Composition dependence of the Ionic Conductivity of Cold-Pressed Powder at Room Temperature

$\text{Li}_{7.2}\text{B}_{0.1}\text{P}_{0.9}\text{S}_6$	$\text{Li}_{7.2}\text{Al}_{0.1}\text{P}_{0.9}\text{S}_6$
Ionic conductivity σ / S cm^{-1}	Ionic conductivity σ / S cm^{-1}
9.88×10^{-6}	8.52×10^{-5}

Table S3. Rietveld refinements of X-ray diffraction data for $\text{Li}_{7.25}\text{Ge}_{0.25}\text{P}_{0.75}\text{S}_6$ at 298 K.

Li _{7.25} Ge _{0.25} P _{0.75} S ₆ structure from X-ray diffraction data (space group $F\bar{4}3m$) ; $a = 9.9543(1)$ Å Fit residuals (R_{wp} , R_{exp} , χ^2): 6.46%, 4.72%, 2.86						
Atom	Site	x	y	z	Occ.	$U_{iso}/ \text{Å}^2$
Li1	48h	0.823(2)	0.498(1)	0.676(2)	0.60(1)	0.1043(9)
S1	16e	0.623(1)	x(S1)	x(S1)	1	0.0244(4)
S2	4c	0.250	x(S2)	x(S2)	1	0.0164(9)
S3	4a	0	x(S3)	x(S3)	1	0.0302(1)
Ge1	4b	0.500	x(Ge1)	x(Ge1)	0.23(2)	0.0531(3)
P1	4b	x(Ge1)	x(Ge1)	x(Ge1)	1- Occ.(Ge1)	U_{iso} (Ge1)

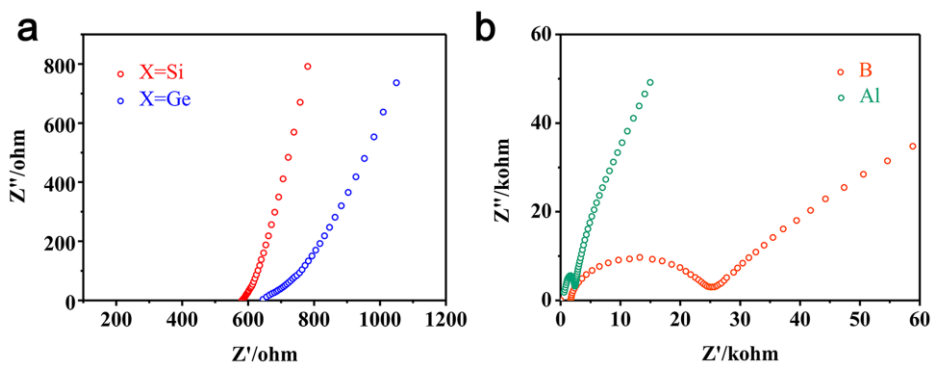


Figure S1. a. Nyquist plots of $Li_{7.1}M_{0.1}P_{0.9}S_6$ ($M = Si, Ge$), b. Nyquist plots of $Li_{7.2}N_{0.1}P_{0.9}S_6$ ($N = B, Al$) at room temperature.

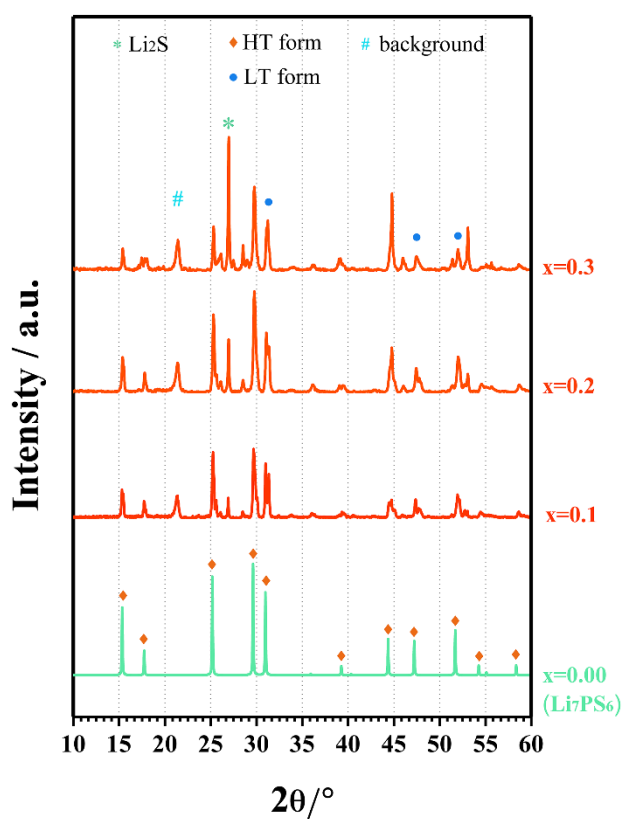


Figure S2. X-ray diffraction (XRD) patterns of $Li_{7+2x}Al_xP_{1-x}S_6$ ($x = 0.1, 0.2, 0.3$).

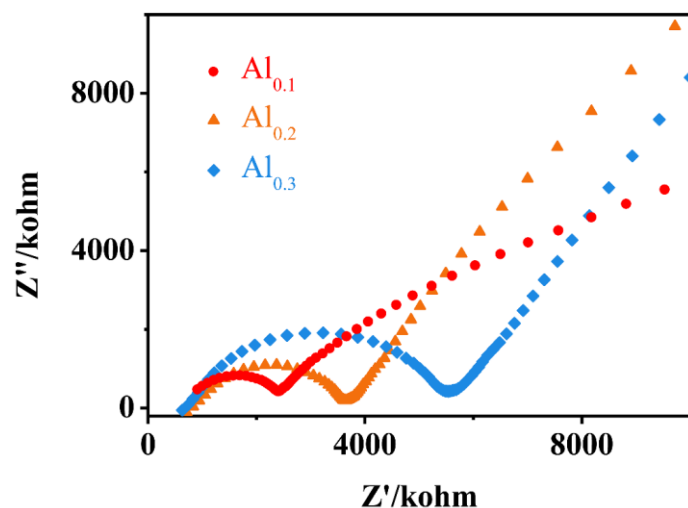


Figure S3. Nyquist plots of $\text{Li}_{7+2x}\text{Al}_x\text{P}_{1-x}\text{S}_6$ ($x = 0.1, 0.2, 0.3$).

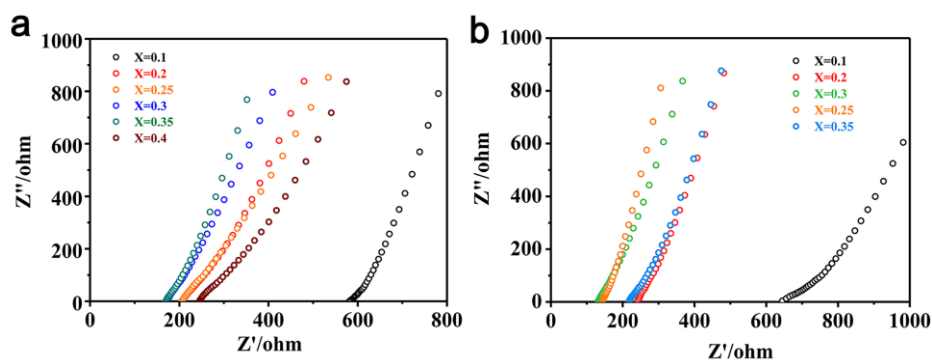


Figure S4. Nyquist plots of $\text{Li}_{7+x}\text{M}_x\text{P}_{1-x}\text{S}_6$ ($M = \text{Si}, \text{Ge}$, $x = 0.1-0.4$) at room temperature. a. $M = \text{Si}$. b. $M = \text{Ge}$.

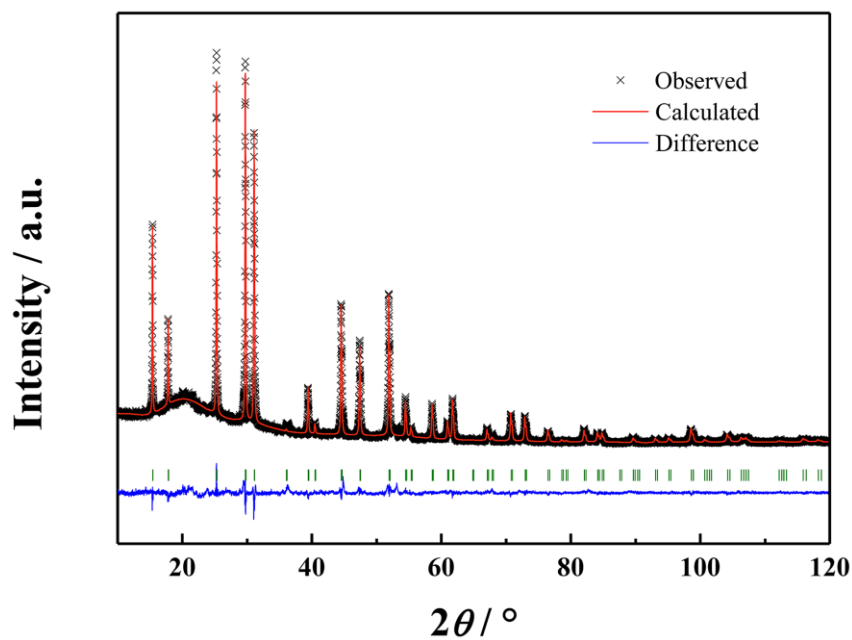


Figure S5. Rietveld refinement pattern of $\text{Li}_{7.25}\text{Ge}_{0.25}\text{P}_{0.75}\text{S}_6$ at 298 K. Experimental data are shown as points; Red: calculated intensities; black: observed intensities; blue: difference profile. The green markets indicate the positions of the diffraction lines.

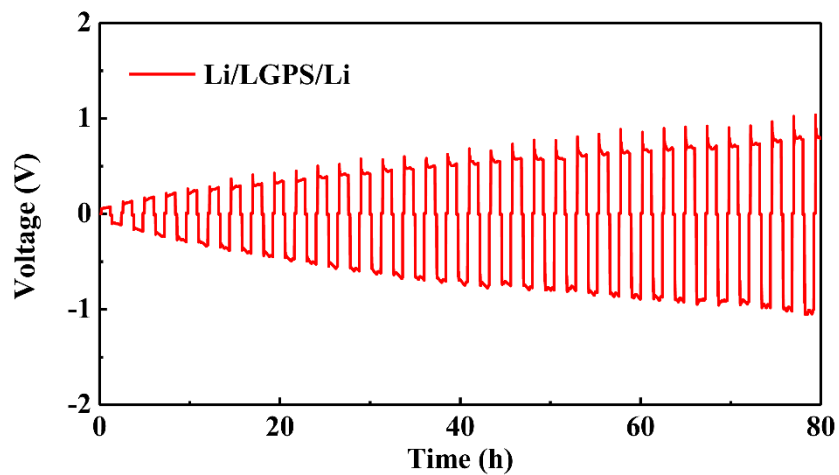


Figure S6. Potential profiles during galvanostatic cycling for a Li/LGPS/Li cell.

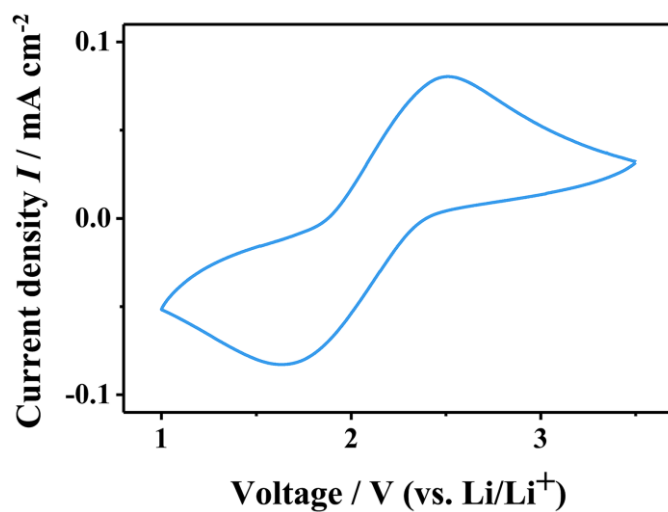


Figure S7. Cyclic voltammogram for a Li/Li_{7.25}Si_{0.25}P_{0.75}S₆/TiS₂ cell; from 1 to 3.5 V at a scan rate of 1 mV s⁻¹.

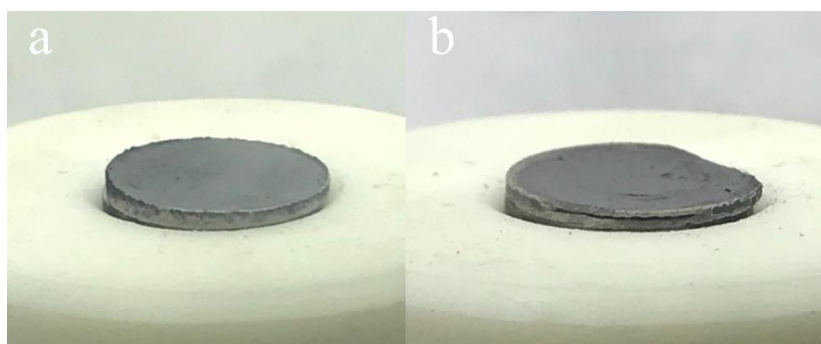


Figure S8. Photographs of TiS₂/SE/Li all-solid-state cell before (a) and after (b) charge-discharge cycle.