

Supporting Information for

Ambipolar Charge Transport in an Organic/Inorganic van der Waals *p-n* Heterojunction

Jie Yan,^{ab} Yang Hao,^{ab} Yutao Cui,^{ab} Jiajia Zhang,^{ab} Ye Zou,^{ab} Weifeng Zhang,^{ab} Gui Yu,^{ab}

Jian Zheng,^{ab} Wei Xu*^{ab} and Daoben Zhu*^{ab}

a: Beijing National Laboratory for Molecular Sciences, CAS Key Laboratory of Organic Solids, Institute of Chemistry, Chinese Academy of Sciences, Beijing 100190, People's Republic of China.

b: University of Chinese Academy of Sciences, Beijing 100049, People's Republic of China.

*E-mail: wxu@iccas.ac.cn, zhudb@iccas.ac.cn

1. Langmuir curve of PDVT-10

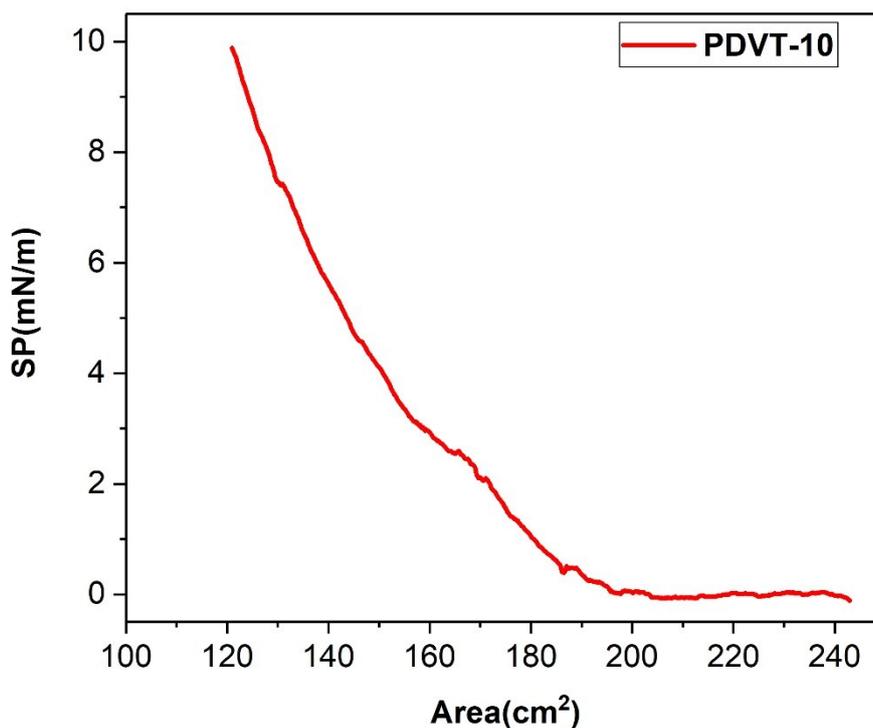


Figure S1. Surface pressure versus area isotherm of PDVT-10 (spreading solution of 4 mg of PDVT-10 in 1 mL of *o*-dichlorobenzene) on a water subphase.

2. Output characteristics of the PDVT-10/MoS₂ ambipolar *p-n* heterojunctions.

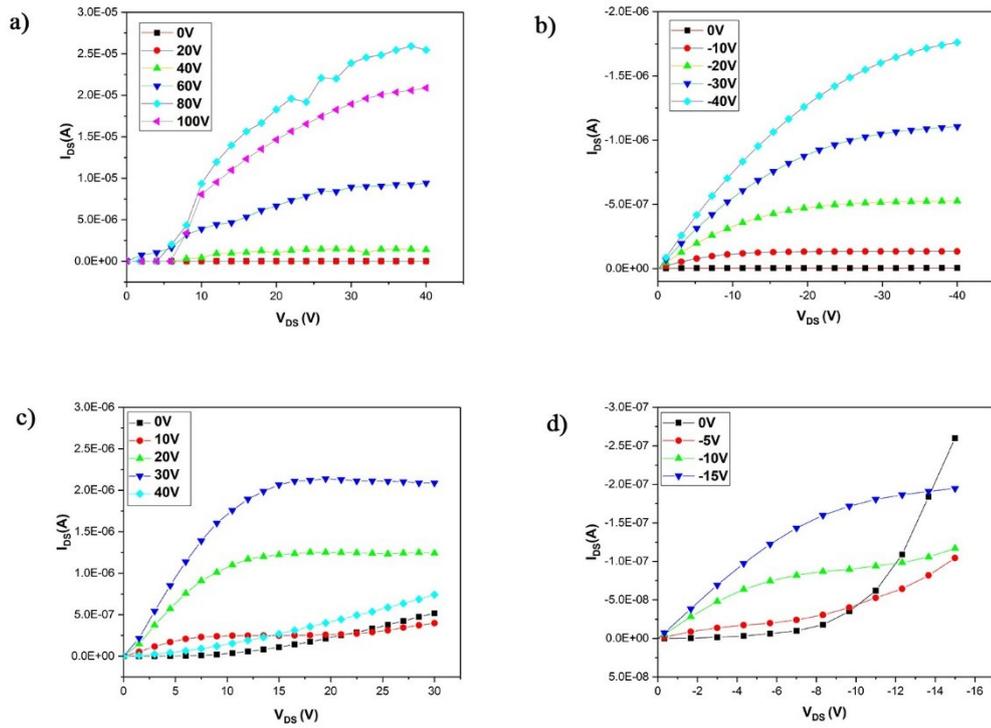


Figure S2. Output characteristics of the PDVT-10/MoS₂ ambipolar *p-n* heterojunctions. (a) Output characteristics of MoS₂. (b) Output characteristics of PDVT-10. (c, d) Output characteristics of the *p-n* junction.

3. Absorption spectra of PDVT-10, MoS₂ and the junction

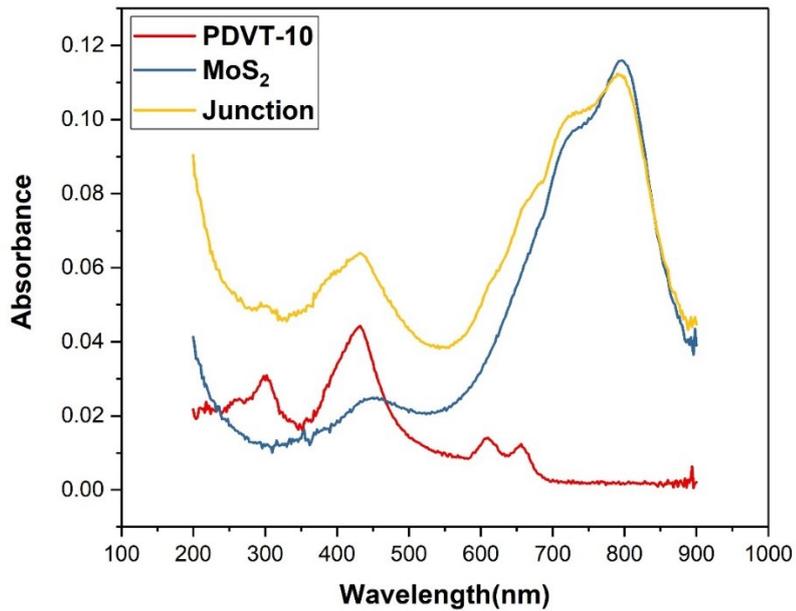


Figure S3. UV-Vis-NIR spectra of PDVT-10, MoS₂ and the junction.

4. UPS spectra of PDVT-10.

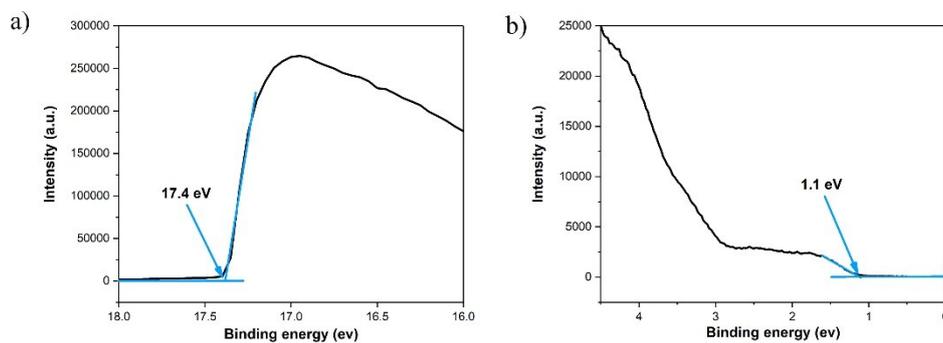


Figure S4. UPS spectra of PDVT-10.

5. Hole mobility as a function of gate voltage.

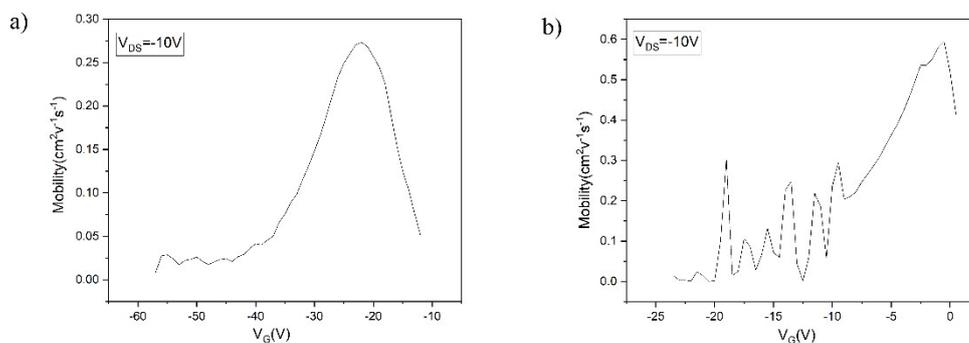


Figure S5. A plot of hole mobility as a function of gate voltage. (a) PDVT-10. Within $-60 \text{ V} < V_{GS} < -40 \text{ V}$, the mobility was calculated with the saturated equation. Within $-40 \text{ V} < V_{GS} < 10 \text{ V}$, the mobility was calculated with the liner equation. (b) The junction. Within $-25 \text{ V} < V_{GS} < -12 \text{ V}$, the mobility was calculated with the saturated equation. Within $-12 \text{ V} < V_{GS} < 0 \text{ V}$, the mobility was calculated with the liner equation.