Electronic Supplementary Information (ESI):

Preparation of single crystal of pyrimidine-5-carbaldehyde 1:

Pyrimidine-5-carbaldehyde **1** *ca*. 10 mg was dissolved in 1 mL of acetone in sample tube. Vapor diffusion method with hexane under room temperature afforded single crystal of carbaldehyde **1** with enough quality for single crystal X-ray structural analysis and X-ray diffraction analysis.

Solid-vapor phase absolute asymmetric synthesis of 5-pyrimidyl alkanol 2:

Pyrimidine-5-carbaldehyde 1 was sublimed at 100 °C under vacuum. Sublimed powder crystals of the carbaldehyde 1 (1.5 mg) were placed in slender glass tubes (30 mm \times 4 mm ID). These slender tubes were placed into larger glass tubes ($40 \text{ mm} \times 11 \text{ mm}$ ID), which were then placed in a desiccator (93 mm \times 85 mm ID) filled with argon gas. One experimental set comprised 10 or 11 samples of carbaldehyde 1. A 1 M solution of *i*- Pr_2Zn in toluene (3 mL) was placed in a reaction vessel (15 mm \times 33 mm ID) in the desiccator. Pyrimidine-5-carbaldehyde 1 was exposed to *i*-Pr₂Zn vapor for 16 h at room temperature. The reaction was quenched with 1 M hydrochloric acid. The reaction mixture was dissolved in ethyl acetate and water, then extracted with ethyl acetate. The organic layer was passed through silica gel NH₂ using ethyl acetate as eluent. The molar ratio of carbaldehyde 1 and alkanol 2 was determined using gas chromatography (GC) (Phenomenex Zebron ZB-5HT, $30 \text{ m} \times 0.25 \text{ mm}$ ID $\times 0.25 \text{ µm}$; T = 150 °C until 300 °C, with an increment of 20 °C/min; N₂ flow rate 1.0 mL/min; retention time 3.7 min for 1 and 5.6 min for 2). Calibration revealed that the molar ratio of 1 and 2 was essentially the same as the ratio of peak areas of 1 and 2 determined by GC. The absolute configurations and ee of pyrimidyl alkanol 2 were determined using supercritical fluid chromatography (SFC) on a chiral stationary phase (Daicel Chiralpak IB column, 250×4.6 mm ID; eluent 9% methanol in CO₂ (v/v); flow rate 3.3 mL/min; 254 nm UV detector; retention time 1.8 min for (S)-2 and 2.1 min for (R)-2).

Run ^[a]	Molar ratio of $1 : 2^{[b]}$	Alkanol 2 ^[c]		
		ee [%]	Config.	
C23	47:53	5.0	S	
C24	55:45	7.8	R	
C25	56:44	2.0	R	
C26	55:45	9.0	S	
C27	54:46	0.8	R	
C28	62:38	4.5	R	
C29	70:30	6.3	S	
C30	55:45	0.2	<i>(S)</i>	
C31	61 : 39	4.4	S	
C32	49 : 51	7.1	S	
C33	71 : 29	24.1	R	
D34	39:61	2.2	S	
D35	62:38	13.4	S	
D36	50 : 50	5.0	R	
D37	39:61	3.3	R	
D38	38:62	2.6	R	
D39	36:64	3.3	R	
D40	33:67	5.7	R	
D41	17:83	14.4	S	
D42	16:84	4.9	R	
D43	8:92	1.7	S	
D44	19:81	3.2	S	
E45	78 : 22	5.6	S	
E46	13:87	9.5	R	
E47	66 : 34	1.2	R	
E48	56:44	0.5	S	
E49	70:30	5.8	S	
E50	43 : 57	0.1	<i>(S)</i>	
E51	55:45	0.7	R	
E52	43 : 57	1.9	S	
E53	48 : 52	1.6	S	

Table S1. Absolute asymmetric synthesis of pyrimidyl alkanol **2** from powder crystals of pyrimidine-5-carbaldehyde **1** and vapor of diisopropylzinc in conjunction with asymmetric autocatalysis.

E54	31:69	7.0	S
E55	65 : 35	3.2	R
F56	84:16	2.6	R
F57	77:23	0.1	<i>(S)</i>
F58	86:14	2.6	R
F59	80:20	1.6	R
F60	83:17	3.9	R
F61	79:21	3.3	S
F62	78:22	2.3	R
F63	88:12	2.8	S
F64	80:20	7.6	S
F65	79:21	1.1	R
F66	67:33	0.5	R
G67	48:52	4.4	R
G68	61 : 39	4.9	R
G69	44 : 56	9.5	R
G70	50:50	2.7	R
G71	41 : 59	2.1	S
G72	24:76	1.4	S
G73	48:52	1.9	S
G74	54:46	9.0	R
G75	15:85	1.5	S
G76	48:52	3.0	S
G77	63 : 37	10.1	S
H78	71:29	2.6	S
H79	74:26	5.1	R
H80	72:28	6.9	S
H81	79:21	3.4	R
H82	89:11	$12.6^{[d]}$	R
H83	67:33	5.2	R
H84	88:12	4.6	S
H85	67:33	6.0	S
H86	69:31	0.7	S
H87	57:43	0.2	<i>(S)</i>
H88	48 : 52	5.7	S
I89	93:7	2.2	S

190	88:12	3.0	S
I91	86 : 14	0.5	S
192	84:16	1.1	R
193	94 : 6	4.0	R
I94	88:12	0.3	<i>(S)</i>
195	81:19	0.7	S
I96	88:12	1.4	S
I97	92:8	1.2	R
I98	85 : 15	1.4	R
J99	87:13	0.3	(R)
J100	88:12	0.5	R
J101	91:9	0.1	(R)
J102	89:11	3.1	R
J103	86 : 14	1.2	R
J104	83:17	2.3	R
J105	92:8	0.2	<i>(S)</i>
J106	91:9	5.0	S
J107	89:11	2.6	R
J108	87 : 13	4.4	S
K109	86:14	3.4	S
K110	75 : 25	1.2	S
K111	83:17	3.7	R
K112	89:11	1.2	R
K113	78:22	1.5	R
K114	83:17	0.2	(R)
K115	80:20	0.1	(R)
K116	78:22	0.9	S
K117	72:28	2.3	R
K118	71 : 29	4.9	S
L119	81 : 19	6.1	S
L120	90:10	4.7	S
L121	88:12	2.8	S
L122	73 : 27	1.8	R
L123	93:7	6.0	S
L124	88:12	4.1	R
L125	84:16	3.4	R

L126	85:15	5.2	R
L127	83:17	3.6	R
L128	88:12	2.8	R
L129	91:9	27.2	S

[a] In the series A–L, each set of reactions was run simultaneously in the same desiccator. Each glass tube was used in only one reaction.

[b] Determined using GC.

[c] Determined using supercritical fluid chromatography (SFC) analysis on a chiral stationary phase.

[d] Chiral SFC chromatogram was shown in Fig. 3S.

Fig. S1. Single crystal of pyrimidine-5-carbaldehyde 1.



Powder X-ray diffractometry (XRD) of crystals of pyrimidine-5-carbaldehyde 1: It can be seen that the diffraction peaks at 5.5° , 10.9° , 22.0° , 27.6° , 33.3° , and 39.1° correspond to (0 0 2), (0 0 4), (0 0 8), (0 0 10), (0 0 12), and (0 0 14) crystal planes respectively.

Fig. S2. X-ray powder diffraction patterns. (a) Single crystal of pyrimidine-5carbaldehyde 1 ($P2_1/n$: observed pattern). (b) Sublimated powder crystal of carbaldehyde 1 (observed pattern).



Chi-squared test: Chi-squared test for goodness of fit (χ_{GF}^2) and Chi-squared test for independence (χ_1^2) were shown in Tables S2 and S3, respectively. The results affording alkanol 2 with <0.5% ee were excluded.

	Pyrimidyl alkanol 2 ^[a]		Total
	R	S	
Observed frequency O_i	61	58	119
Theoretical probability	0.5	0.5	1
Theoretical frequency E _i	59.5	59.5	119
$\chi_{ m GF}^2$	$P = \sum_{i=1}^{k} \frac{(O_i - E_i)}{E_i}$	$\frac{)^2}{2} = 0.07563$	

Table S2 Chi d tost fo 4... cc+(2)

(Significant probability: p = 0.78)

Table S3.	Chi-squared	l test for	independence	(γ_I^2) .

		Neighboring sample		Total
		R	S	
Sample	R	32	29	61
	S	29	29	58
	Total	61	58	119



(Significant probability: p = 0.79)

Fig. S3. Chiral SFC chromatogram of the reaction mixture (Table S1, run H82). The reaction mixture was analyzed after working up. SFC conditions: column, Daicel Chiralpak IB ($250 \times 4.6 \text{ mm ID}$); eluent, 9% methanol in CO₂ (v/v); flow rate, 3.3 mL/min; detection, 254 nm UV detector; retention time (t_R) 1.96 min for (*S*)-alkanol **2** and 2.21 min for (*R*)-alkanol **2**). The areas of 43.687% for (*S*)-**2** and 56.313% for (*R*)-**2**, thus, analysed product is *R*-configured alkanol **2** with 12.6% ee.

