

## Electronic Supplementary Information

### Nanoceria as a DNase I mimicking nanozyme

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## Materials and Methods

**Chemicals.** All the DNA oligomers were purchased from Integrated DNA Technologies (IDT, Coralville, IA, USA), and their sequences and modifications are listed in **Table S1**. The DNase I and its reaction buffer (10×) was purchased from New England Biolabs. Nanoceria dispersion (catalog number: 289744, 20% dispersed in 2.5% acetic acid), and metal salts, and the other nanoparticles and their characterization were reported previously.<sup>1,2</sup> Trisodium phosphate and ethylenediaminetetraacetic acid (EDTA) were from Mandel Scientific, Inc. (Guelph, Ontario, Canada). The nanoceria sample was washed three times using ultracentrifugation. Milli-Q water was used for all the experiments.

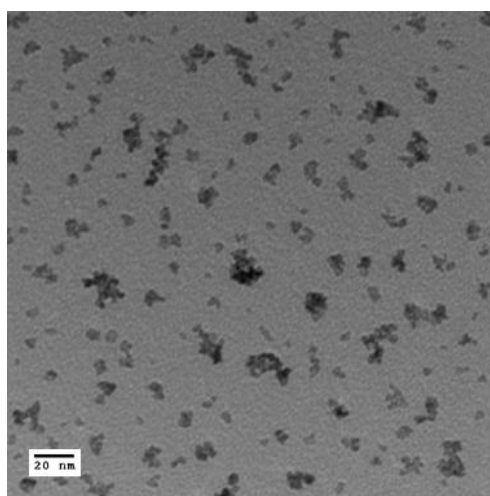
**DNA cleavage by nanoceria.** Detailed characterization of nanoceria used in this work was reported in our previous studies.<sup>3,4</sup> Each DNA sample was dissolved in deionized water to a final concentration of 1.0  $\mu\text{M}$  in the presence of nanoceria (100  $\mu\text{g}/\text{mL}$ ) under different temperatures (0-60°C). The cleavage products were separated using 15% denaturing polyacrylamide gel electrophoresis (dPAGE) and analyzed using a ChemDoc MP imaging system (Bio-Rad, USA). Cleavage by other metal oxide nanoparticles was performed using the same protocol.

**DNA cleavage by DNase I.** Each DNA sample was dissolved with 1× DNase I reaction buffer to a final concentration of 1.0  $\mu\text{M}$  in the presence of DNase I (100 U/mL) at 37°C. The cleavage products were separated using 15% denaturing polyacrylamide gel electrophoresis (dPAGE) and analyzed using a ChemDoc MP imaging system (Bio-Rad, USA).

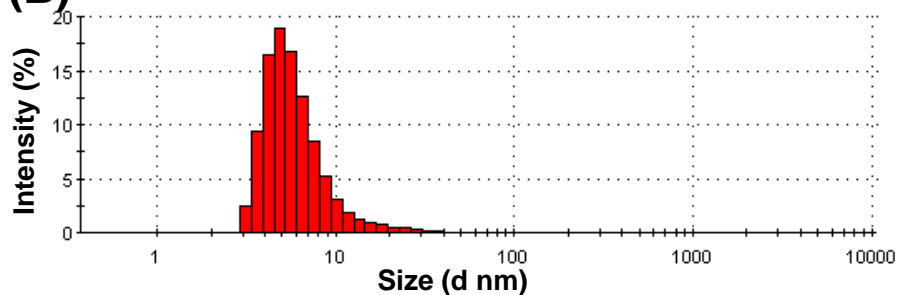
**Mass spectrometry analysis of DNA cleavage products.** For mass spectrometry studies, non-FAM-labeled A<sub>5</sub> and G<sub>5</sub> DNA were used. Each DNA was reacted at 1.0  $\mu\text{M}$  with 100  $\mu\text{g}/\text{mL}$  of CeO<sub>2</sub> for 4 h at 60°C in a total volume of 5 mL. After the reaction, the cleavage products were further concentrated and desalted using a Sep-Pak C<sub>18</sub> cartridge (eluted with CH<sub>3</sub>OH/CH<sub>3</sub>CN mixture, 9:1, v:v) and dried in a vacuum centrifuge instrument. The dried samples were then dissolved by 50  $\mu\text{L}$  Milli-Q water for mass spectrometry analysis. High resolution mass spectrometry analysis was performed using a Thermo Scientific Q-Exactive Orbitrap mass spectrometer (Thermo Scientific, USA) coupled with an electrospray ionization (ESI) source. To help the cleavage products to enter the gas phase, ammonium acetate (0.01M) was added.

**Fluorescence analysis.** DNA adsorption by nanoceria was analyzed by monitoring the change of fluorescence intensity upon the adding of nanoceria. The initial concentration of DNA (FAM-A<sub>15</sub>) was 1.0  $\mu$ M. The fluorescence spectra of the FAM-A<sub>15</sub> samples were acquired after adding different concentrations of nanoceria. All fluorescence analysis was performed using a Cary Eclipse fluorometer (Varian, USA) by exciting at 485 nm.

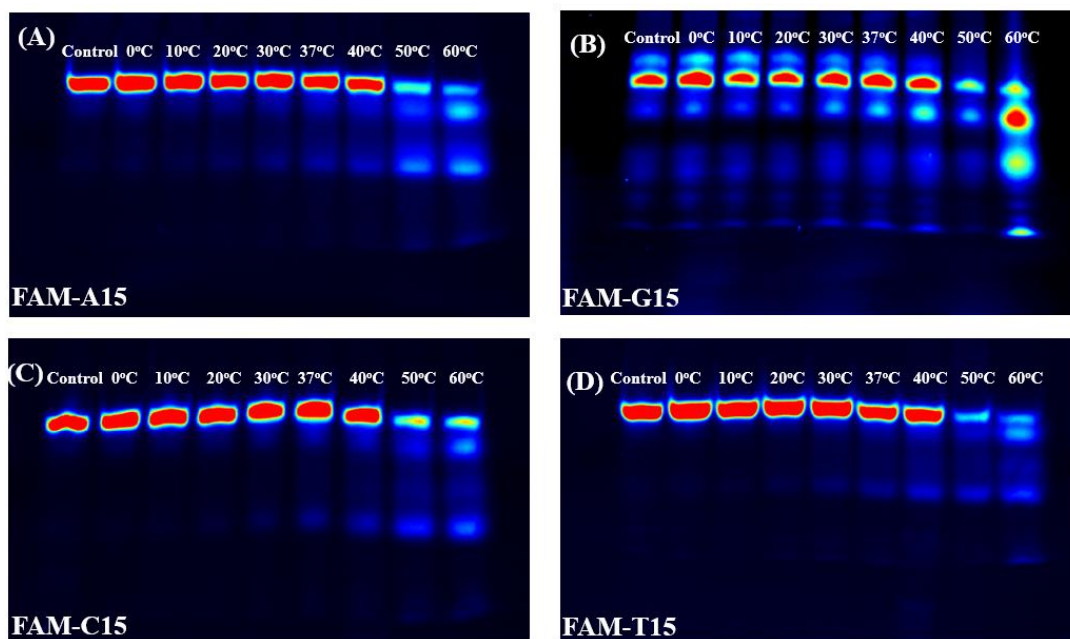
(A)



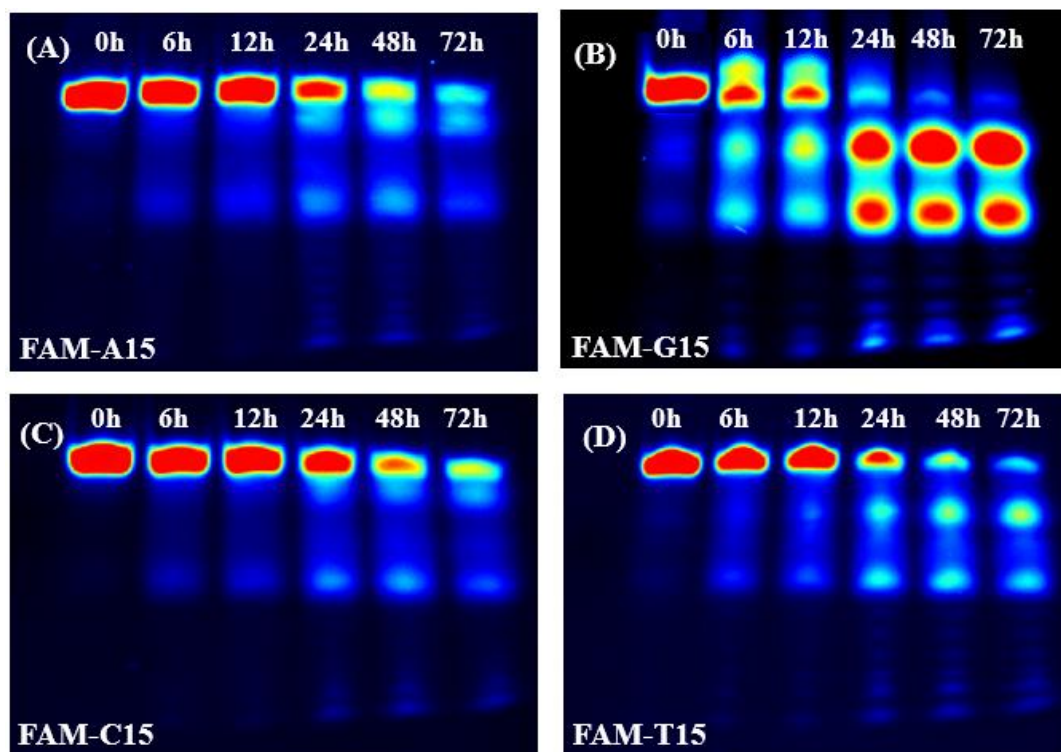
(B)



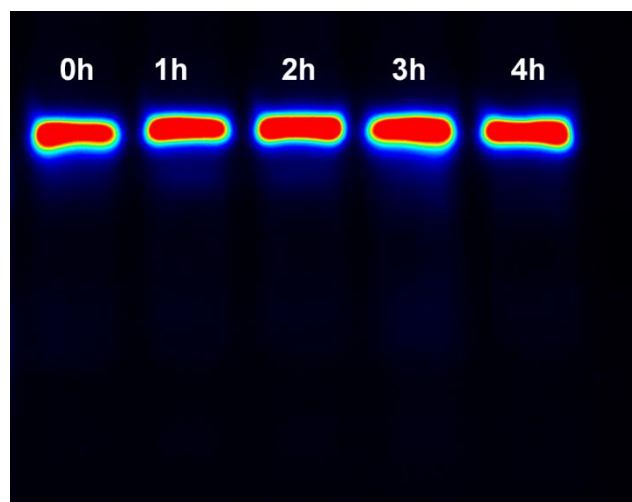
**Fig. S1.** (A) A TEM micrograph and (B) DLS size distribution of the CeO<sub>2</sub> nanoparticles used in this work. The average size was around 5 nm and slight aggregation was observed.



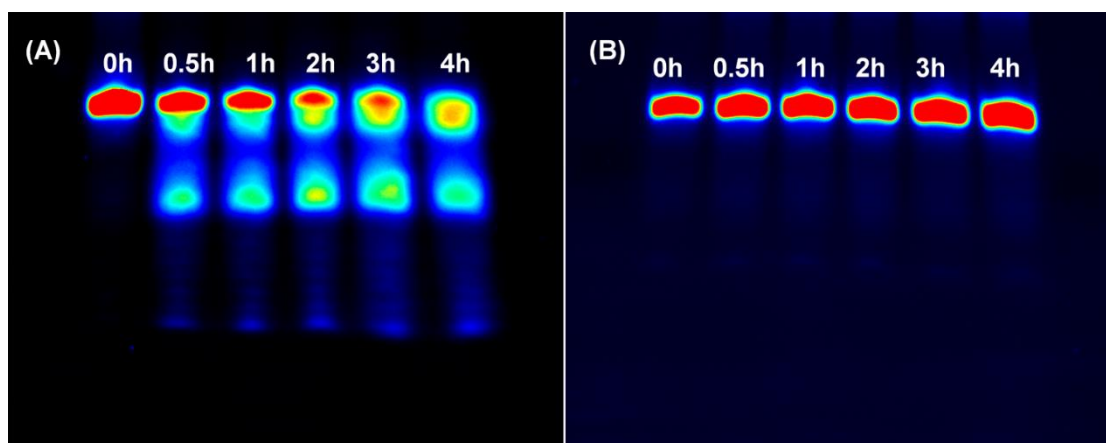
**Fig. S2.** Polyacrylamide gel electrophoresis analysis of DNA (1.0  $\mu\text{M}$ ) cleavage after incubated with nanoceria (100  $\mu\text{g}/\text{mL}$ ) at different temperatures (0-60°C) for 4 h. (A) FAM-A<sub>15</sub>; (B) FAM-G<sub>15</sub>; (C) FAM-C<sub>15</sub>; (D) FAM-T<sub>15</sub>. The cleavage yield increased with increasing temperature.



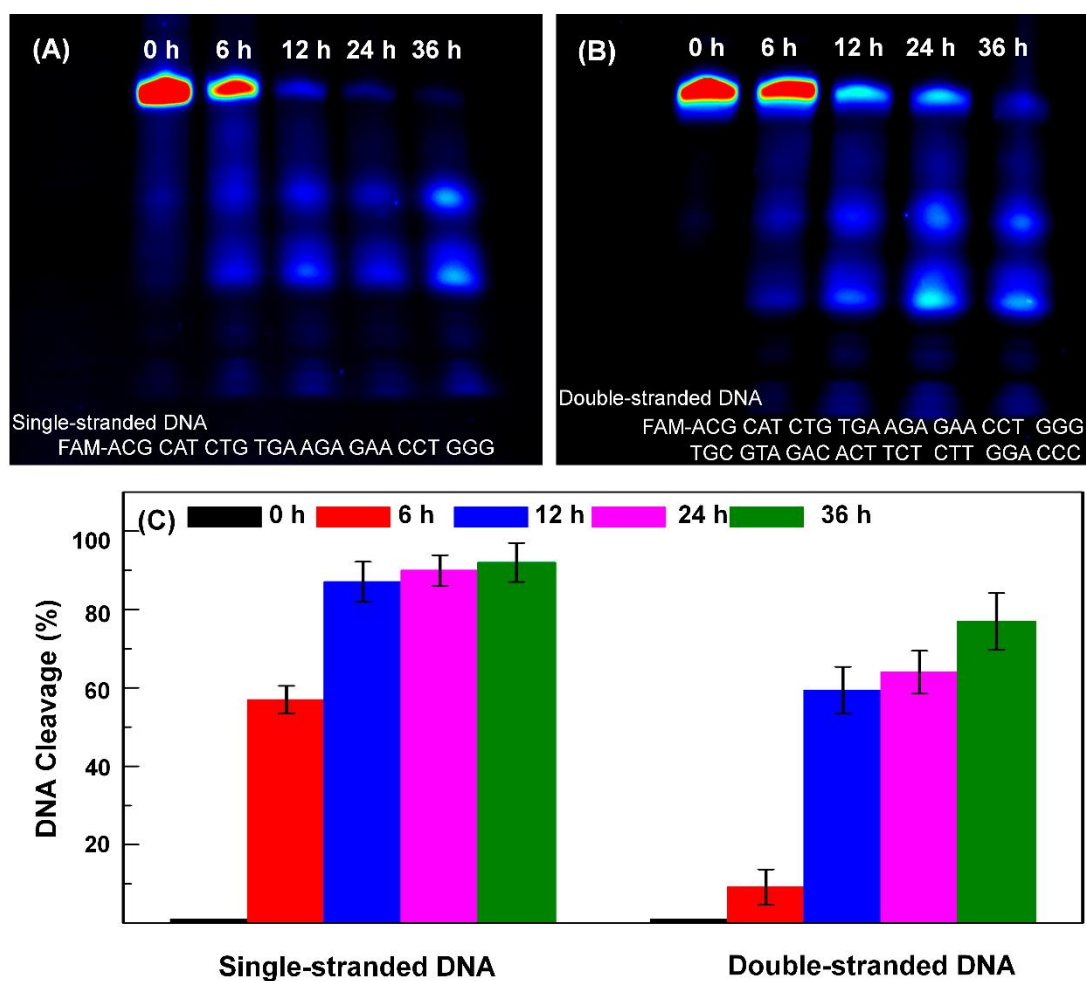
**Fig. S3.** Polyacrylamide gel electrophoresis analysis of DNA ( $1.0 \mu\text{M}$ ) cleavage by nanoceria ( $100 \mu\text{g/mL}$ ) at  $37^\circ\text{C}$  after 72 h incubation. (A) FAM-A<sub>15</sub>; (B) FAM-G<sub>15</sub>; (C) FAM-C<sub>15</sub>; (D) FAM-T<sub>15</sub>.



**Fig. S4.** Polyacrylamide gel electrophoresis analysis of DNA (1.0  $\mu\text{M}$ ) cleavage incubated by large CeO<sub>2</sub> particles (165 nm, 100  $\mu\text{g}/\text{mL}$ ) at 60°C for 4 h. The initial concentration of DNA (FAM-A<sub>15</sub>) was 1.0  $\mu\text{M}$ .

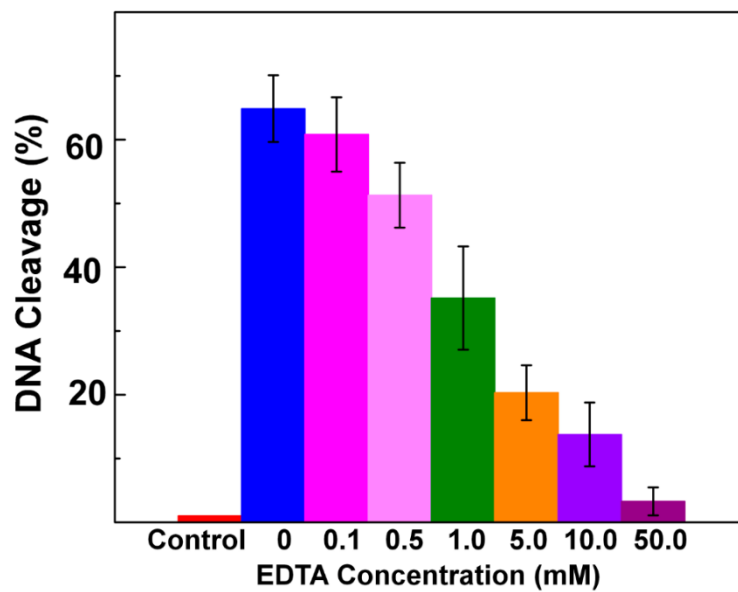


**Fig. S5.** Gel micrographs showing cleavage of DNA (FAM-A<sub>15</sub>) by heat treated (A) 100  $\mu\text{g/mL}$  CeO<sub>2</sub> at 60°C, and (B) DNase I (100 U/mL) at 37°C for 4 h. Before the test, both the CeO<sub>2</sub> and DNase I solutions were heated at 90°C for 10 min. The initial concentration of DNA (FAM-A<sub>15</sub>) was 1.0  $\mu\text{M}$ . The DNase I sample showed no cleavage indicating that it was deactivated by the heat treatment.



**Fig. S6.** Gel micrographs showing cleavage of a random sequenced 24-mer single-stranded DNA (A) and its duplex (B) by CeO<sub>2</sub> (100 µg/mL) at 37°C. The initial concentrations of single-stranded DNA and double stranded DNA were 1.0 µM. (C) Quantification of DNA cleavage yields of the single-stranded DNA (left) and double-stranded DNA (right) at different incubation times.



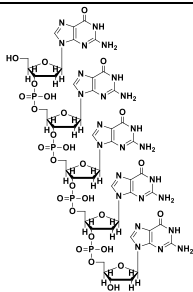
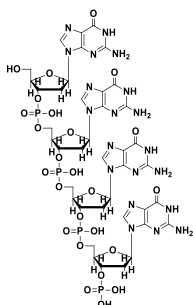
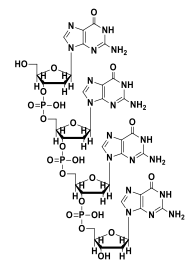
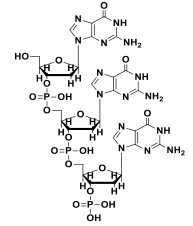
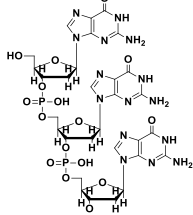
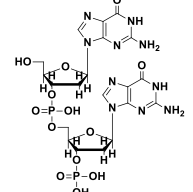


**Fig. S7.** Effect of EDTA (0-50 mM) on the FAM-A<sub>15</sub> DNA cleavage yield by nanoceria at 60°C for 2 h. The initial concentration of the DNA was 1.0  $\mu$ M. All experiments were performed in triplicate.

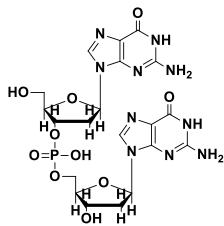
**Table S1.** The sequences of DNA used in this work

<b>DNA</b>	<b>Sequences (from 5'to 3') and modifications</b>
FAM-A <sub>5</sub>	FAM-AAAAA
FAM-G <sub>5</sub>	FAM-GGGGG
FAM-C <sub>5</sub>	FAM-CCCCC
FAM-T <sub>5</sub>	FAM-TTTTT
FAM-A <sub>15</sub>	FAM-AAAAAAAAAAAAAAA
FAM-G <sub>15</sub>	FAM-GGGGGGGGGGGGGGG
FAM-C <sub>15</sub>	FAM-CCCCCCCCCCCCCCC
FAM-T <sub>15</sub>	FAM-TTTTTTTTTTTTTTT
FAM-A <sub>20</sub>	FAM-AAAAAAAAAAAAAAAAAAA
FAM-A <sub>30</sub>	FAM-AAAAAAAAAAAAAAAAAAAAAAAAAA
FAM-A <sub>45</sub>	FAM-AAAAAAAAAAAAAAAAAAAAAAAAAAAAAA
FAM-A <sub>72</sub>	FAM-AAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAA AAAAAAAAAAAAAAAAAAAAAAAAA
FAM-A <sub>90</sub>	FAM-AAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAA AAAAAAAAAAAAA AAAAAAAAAAAAAAAAAAAAAAAAAA

**Table S2.** Main DNA cleavage products of G<sub>5</sub>

No.	Possible cleavage products of DNA-G <sub>5</sub>		MS analysis
	Chemical molecular formula	Theoretical mass	Experimental results
1		$[M-H^+] = 1582.29956$ $[M-2H^+]/2 = 790.64614$ $[M-3H^+]/3 = 526.76166$	$[M-2H^+]/2 = 790.64655$ $[M-3H^+]/3 = 526.76094$
2		$[M-H^+] = 1333.21337$ $[M-2H^+]/2 = 666.10304$ $[M-3H^+]/3 = 443.732937$	<b>Not detected</b>
3		$[M-H^+] = 1253.24704$ $[M-2H^+]/2 = 626.11988$ $[M-3H^+]/3 = 417.07749$	$[M-2H^+]/2 = 626.11970$
4		$[M-H^+] = 1004.16085$ $[M-2H^+]/2 = 501.57678$	<b>Not detected</b>
5		$[M-H^+] = 924.19452$ $[M-2H^+]/2 = 461.59362$	$[M-H^+] = 924.19459$ $[M-2H^+]/2 = 461.59359$
6		$[M-H^+] = 675.10883$ $[M-2H^+]/2 = 337.05052$	<b>Not detected</b>

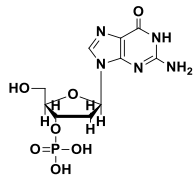
7



$[M-H^+] = 595.14200$

$[M-H^+] = 595.14189$

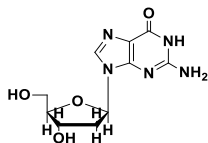
8



$[M-H^+] = 346.05581$

**Not detected**

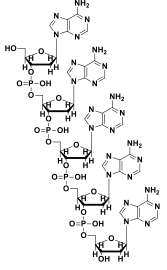
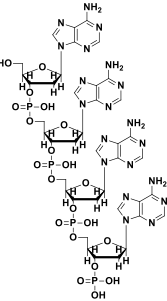
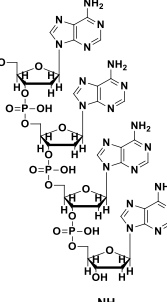
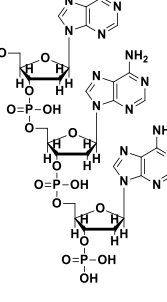
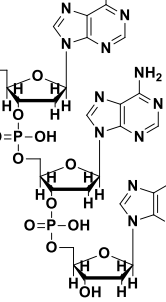
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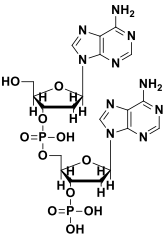
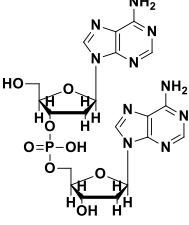
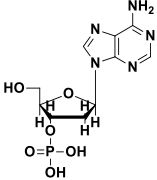
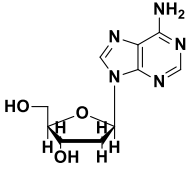


$[M-H^+] = 266.09675$

$[M-H^+] = 266.21305$

**Table S3.** Main DNA cleavage products of A<sub>5</sub>

No.	Possible cleavage products of DNA-A <sub>5</sub>		MS analysis
	Chemical molecular formula	Theoretical exact mass	Experimental results
1		$[M-H^+]=1502.32498$ $[M-2H^+]/2=750.65885$ $[M-3H^+]/3=500.10347$	$[M-2H^+]/2=750.65886$
2		$[M-H^+]=1269.23371$ $[M-2H^+]/2=634.11321$	<b>Not detected</b>
3		$[M-H^+]=1189.26738$ $[M-2H^+]/2=594.13005$	$[M-2H^+]/2=594.12949$
4		$[M-H^+]=956.17610$ $[M-2H^+]/2=477.58441$	<b>Not detected</b>
5		$[M-H^+]=876.20977$ $[M-2H^+]/2=437.60125$	$[M-2H^+]/2=437.13792$

6		$[M-H^+]=643.11850$ $[M-2H^+]/2=321.05506$	<b>Not detected</b>
7		$[M-H^+]=563.15217$	$[M-H^+]=563.15177$
8		$[M-H^+]=330.06089$	<b>Not detected</b>
9		$[M-H^+]=250.09456$	$[M-H^+]=250.09482$

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## References

1. B. Liu and J. Liu, *ACS Appl. Mater. Interfaces*, 2015, **7**, 24833-24838.
2. B. Liu, L. Ma, Z. Huang, H. Hu, P. Wu and J. Liu, *Mater. Horizons*, 2018, **5**, 65-69.
3. B. Liu, Z. Sun, P.-J. J. Huang and J. Liu, *J. Am. Chem. Soc.*, 2015, **137**, 1290-1295.
4. Y. Zhao, Y. Wang, A. Mathur, Y. Wang, V. Maheshwari, H. Su and J. Liu, *Nanoscale*, 2019.