

Electronic Supplementary Information

Nanoceria as a DNase I mimicking nanozyme

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Materials and Methods

Chemicals. All the DNA oligomers were purchased from Integrated DNA Technologies (IDT, Coralville, IA, USA), and their sequences and modifications are listed in **Table S1**. The DNase I and its reaction buffer (10×) was purchased from New England Biolabs. Nanoceria dispersion (catalog number: 289744, 20% dispersed in 2.5% acetic acid), and metal salts, and the other nanoparticles and their characterization were reported previously.^{1,2} Trisodium phosphate and ethylenediaminetetraacetic acid (EDTA) were from Mandel Scientific, Inc. (Guelph, Ontario, Canada). The nanoceria sample was washed three times using ultracentrifugation. Milli-Q water was used for all the experiments.

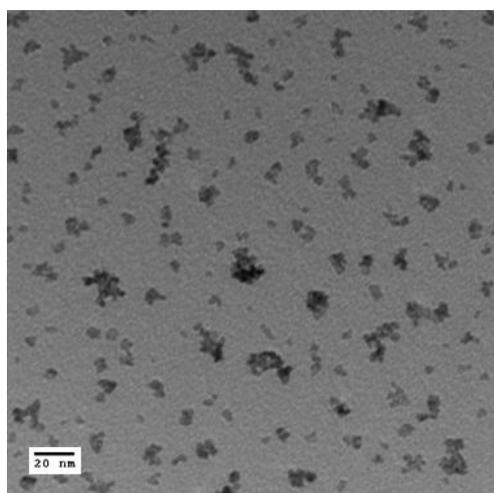
DNA cleavage by nanoceria. Detailed characterization of nanoceria used in this work was reported in our previous studies.^{3,4} Each DNA sample was dissolved in deionized water to a final concentration of 1.0 μM in the presence of nanoceria (100 $\mu\text{g}/\text{mL}$) under different temperatures (0-60°C). The cleavage products were separated using 15% denaturing polyacrylamide gel electrophoresis (dPAGE) and analyzed using a ChemDoc MP imaging system (Bio-Rad, USA). Cleavage by other metal oxide nanoparticles was performed using the same protocol.

DNA cleavage by DNase I. Each DNA sample was dissolved with 1× DNase I reaction buffer to a final concentration of 1.0 μM in the presence of DNase I (100 U/mL) at 37°C. The cleavage products were separated using 15% denaturing polyacrylamide gel electrophoresis (dPAGE) and analyzed using a ChemDoc MP imaging system (Bio-Rad, USA).

Mass spectrometry analysis of DNA cleavage products. For mass spectrometry studies, non-FAM-labeled A₅ and G₅ DNA were used. Each DNA was reacted at 1.0 μM with 100 $\mu\text{g}/\text{mL}$ of CeO₂ for 4 h at 60°C in a total volume of 5 mL. After the reaction, the cleavage products were further concentrated and desalted using a Sep-Pak C₁₈ cartridge (eluted with CH₃OH/CH₃CN mixture, 9:1, v:v) and dried in a vacuum centrifuge instrument. The dried samples were then dissolved by 50 μL Milli-Q water for mass spectrometry analysis. High resolution mass spectrometry analysis was performed using a Thermo Scientific Q-Exactive Orbitrap mass spectrometer (Thermo Scientific, USA) coupled with an electrospray ionization (ESI) source. To help the cleavage products to enter the gas phase, ammonium acetate (0.01M) was added.

Fluorescence analysis. DNA adsorption by nanoceria was analyzed by monitoring the change of fluorescence intensity upon the adding of nanoceria. The initial concentration of DNA (FAM-A₁₅) was 1.0 μ M. The fluorescence spectra of the FAM-A₁₅ samples were acquired after adding different concentrations of nanoceria. All fluorescence analysis was performed using a Cary Eclipse fluorometer (Varian, USA) by exciting at 485 nm.

(A)



(B)

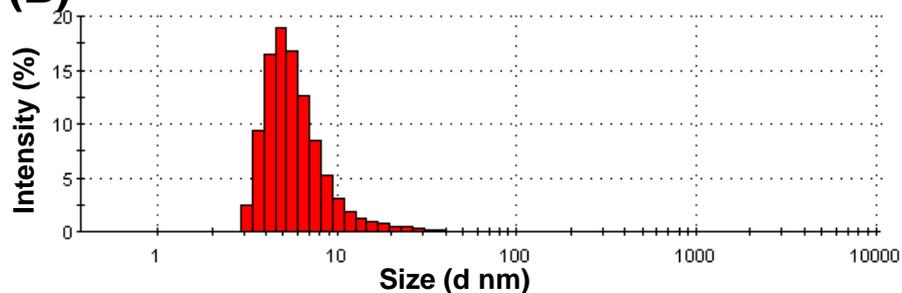


Fig. S1. (A) A TEM micrograph and (B) DLS size distribution of the CeO₂ nanoparticles used in this work. The average size was around 5 nm and slight aggregation was observed.

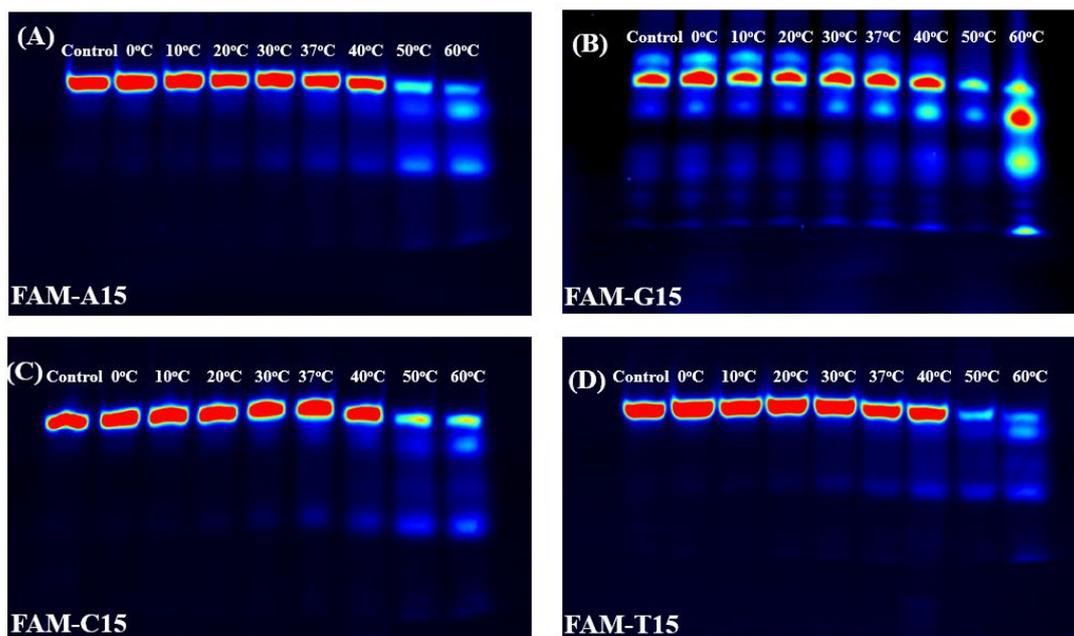


Fig. S2. Polyacrylamide gel electrophoresis analysis of DNA (1.0 μM) cleavage after incubated with nanoceria (100 $\mu\text{g}/\text{mL}$) at different temperatures (0-60 $^{\circ}\text{C}$) for 4 h. (A) FAM-A₁₅; (B) FAM-G₁₅; (C) FAM-C₁₅; (D) FAM-T₁₅. The cleavage yield increased with increasing temperature.

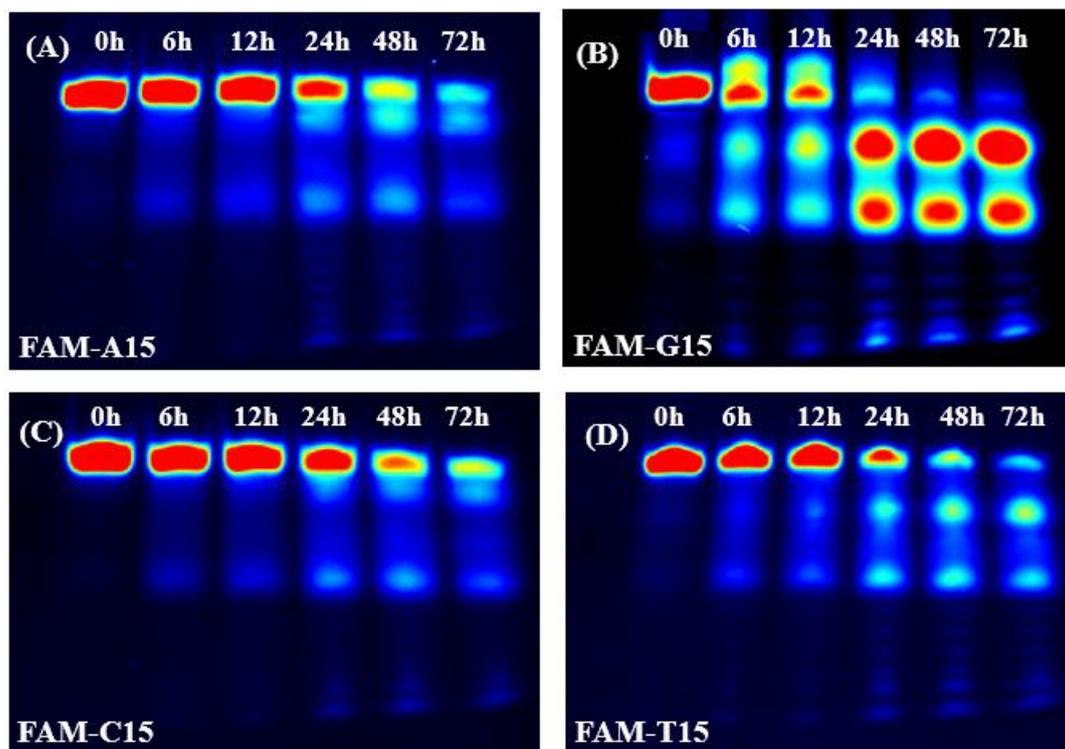


Fig. S3. Polyacrylamide gel electrophoresis analysis of DNA ($1.0 \mu\text{M}$) cleavage by nanoceria ($100 \mu\text{g/mL}$) at 37°C after 72 h incubation. (A) FAM-A₁₅; (B) FAM-G₁₅; (C) FAM-C₁₅; (D) FAM-T₁₅.

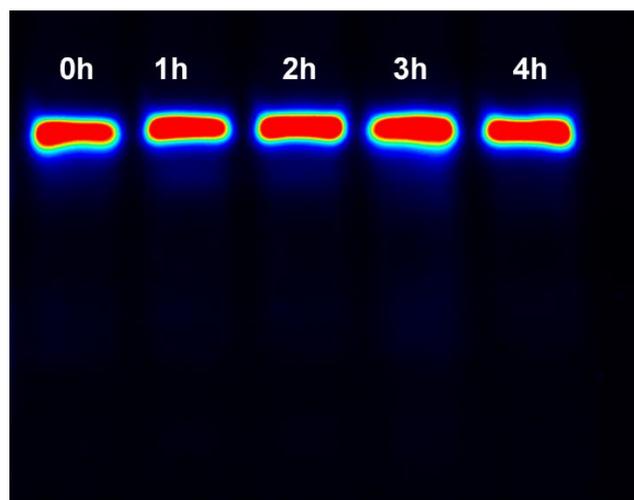


Fig. S4. Polyacrylamide gel electrophoresis analysis of DNA (1.0 μM) cleavage incubated by large CeO₂ particles (165 nm, 100 $\mu\text{g}/\text{mL}$) at 60°C for 4 h. The initial concentration of DNA (FAM-A₁₅) was 1.0 μM .

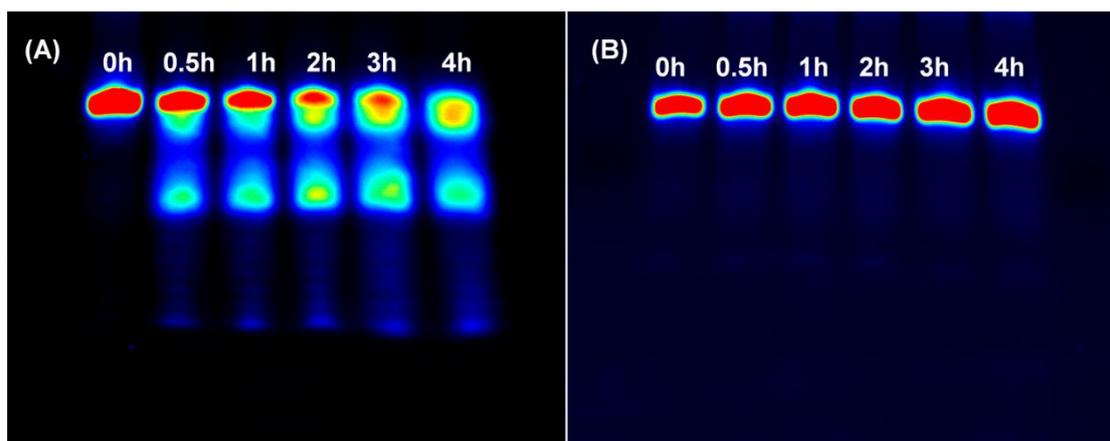


Fig. S5. Gel micrographs showing cleavage of DNA (FAM-A₁₅) by heat treated (A) 100 $\mu\text{g/mL}$ CeO_2 at 60°C, and (B) DNase I (100 U/mL) at 37°C for 4 h. Before the test, both the CeO_2 and DNase I solutions were heated at 90°C for 10 min. The initial concentration of DNA (FAM-A₁₅) was 1.0 μM . The DNase I sample showed no cleavage indicating that it was deactivated by the heat treatment.

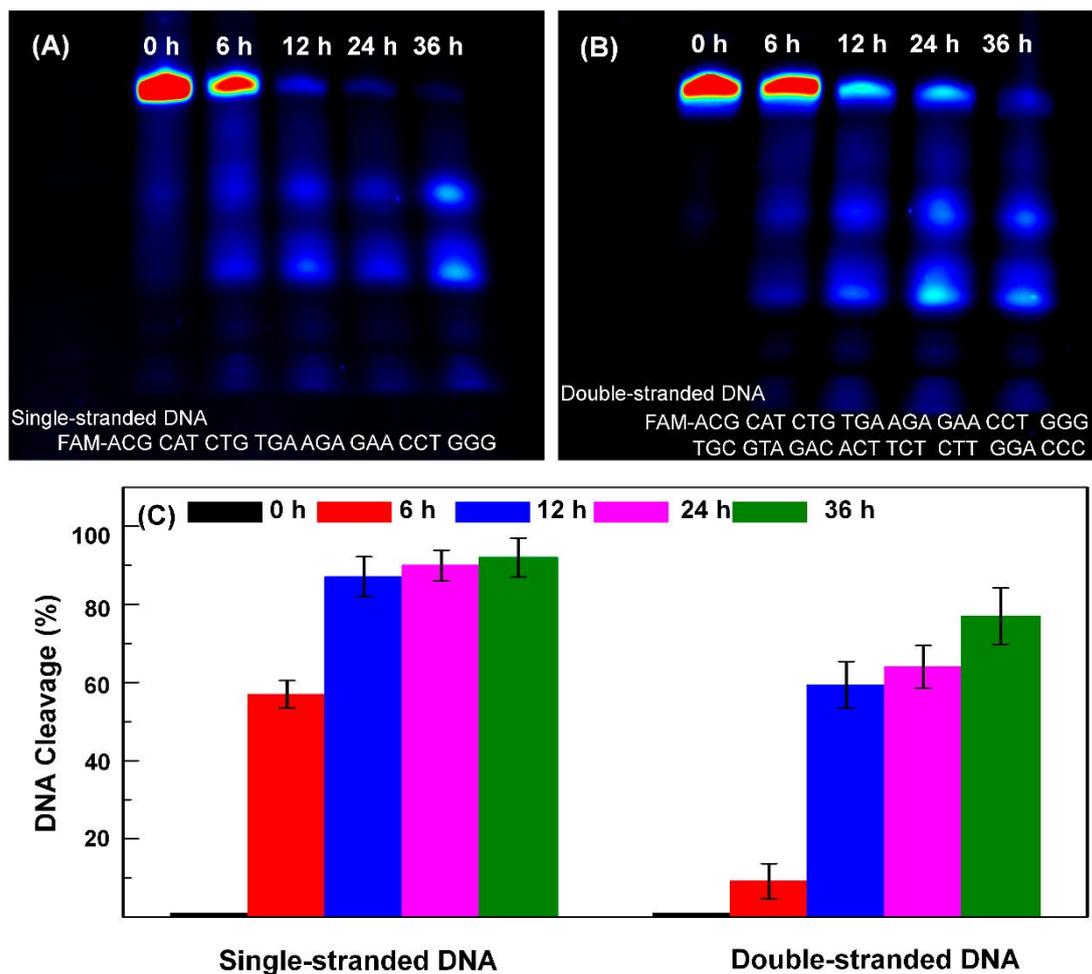


Fig. S6. Gel micrographs showing cleavage of a random sequenced 24-mer single-stranded DNA (A) and its duplex (B) by CeO₂ (100 μg/mL) at 37°C. The initial concentrations of single-stranded DNA and double stranded DNA were 1.0 μM. (C) Quantification of DNA cleavage yields of the single-stranded DNA (left) and double-stranded DNA (right) at different incubation times.

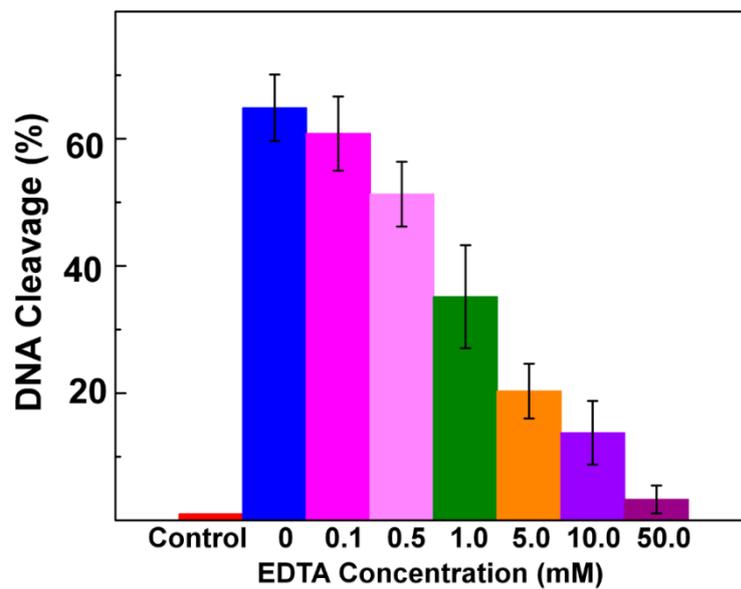
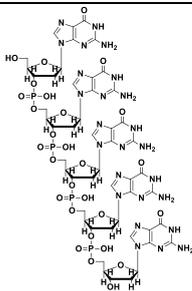
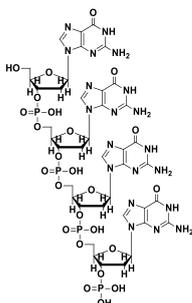
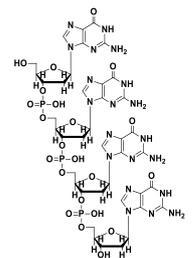
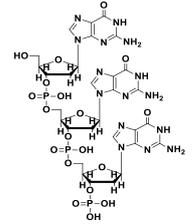
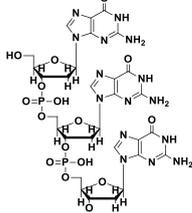
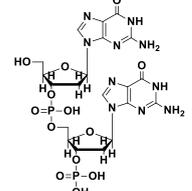


Fig. S7. Effect of EDTA (0-50 mM) on the FAM-A₁₅ DNA cleavage yield by nanoceria at 60°C for 2 h. The initial concentration of the DNA was 1.0 μ M. All experiments were performed in triplicate.

Table S1. The sequences of DNA used in this work

DNA	Sequences (from 5'to 3') and modifications
FAM-A ₅	FAM-AAAAA
FAM-G ₅	FAM-GGGGG
FAM-C ₅	FAM-CCCCC
FAM-T ₅	FAM-TTTTT
FAM-A ₁₅	FAM-AAAAAAAAAAAAAAA
FAM-G ₁₅	FAM-GGGGGGGGGGGGGGG
FAM-C ₁₅	FAM-CCCCCCCCCCCCCCC
FAM-T ₁₅	FAM-TTTTTTTTTTTTTTT
FAM-A ₂₀	FAM-AAAAAAAAAAAAAAAAAAA
FAM-A ₃₀	FAM-AAAAAAAAAAAAAAAAAAAAAAAAAA
FAM-A ₄₅	FAM-AAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAA
FAM-A ₇₂	FAM-AAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAA AAAAAAAAAAAAAAAAAAAAAAAAA
FAM-A ₉₀	FAM-AAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAAA AAAAAAAAAAAAA AAAAAAAAAAAAAAAAAAAAAAAAAA

Table S2. Main DNA cleavage products of G₅

No.	Possible cleavage products of DNA-G ₅		MS analysis
	Chemical molecular formula	Theoretical mass	Experimental results
1		$[M-H^+]=1582.29956$ $[M-2H^+]/2=790.64614$ $[M-3H^+]/3=526.76166$	$[M-2H^+]/2=790.64655$ $[M-3H^+]/3=526.76094$
2		$[M-H^+]=1333.21337$ $[M-2H^+]/2=666.10304$ $[M-3H^+]/3=443.732937$	Not detected
3		$[M-H^+]=1253.24704$ $[M-2H^+]/2=626.11988$ $[M-3H^+]/3=417.07749$	$[M-2H^+]/2=626.11970$
4		$[M-H^+]=1004.16085$ $[M-2H^+]/2=501.57678$	Not detected
5		$[M-H^+]=924.19452$ $[M-2H^+]/2=461.59362$	$[M-H^+]=924.19459$ $[M-2H^+]/2=461.59359$
6		$[M-H^+]=675.10883$ $[M-2H^+]/2=337.05052$	Not detected

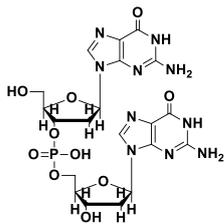
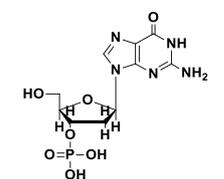
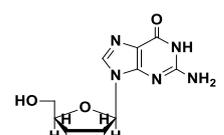
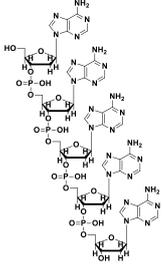
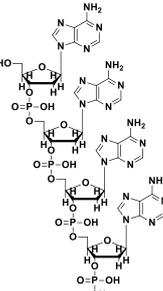
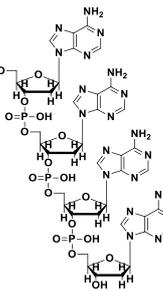
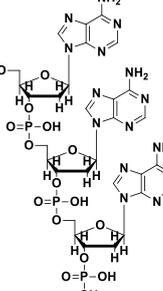
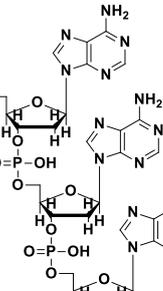
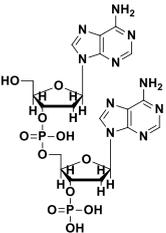
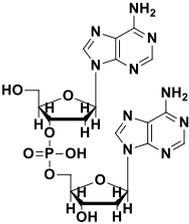
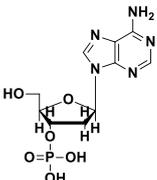
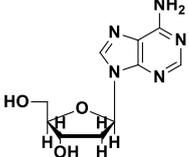
7		$[M-H^+] = 595.14200$	$[M-H^+] = 595.14189$
8		$[M-H^+] = 346.05581$	Not detected
9		$[M-H^+] = 266.09675$	$[M-H^+] = 266.21305$

Table S3. Main DNA cleavage products of A₅

No.	Possible cleavage products of DNA-A ₅		MS analysis
	Chemical molecular formula	Theoretical exact mass	Experimental results
1		$[M-H^+]=1502.32498$ $[M-2H^+]/2=750.65885$ $[M-3H^+]/3=500.10347$	$[M-2H^+]/2=750.65886$
2		$[M-H^+]=1269.23371$ $[M-2H^+]/2=634.11321$	Not detected
3		$[M-H^+]=1189.26738$ $[M-2H^+]/2=594.13005$	$[M-2H^+]/2=594.12949$
4		$[M-H^+]=956.17610$ $[M-2H^+]/2=477.58441$	Not detected
5		$[M-H^+]=876.20977$ $[M-2H^+]/2=437.60125$	$[M-2H^+]/2=437.13792$

6		$[M-H^+]=643.11850$ $[M-2H^+]/2=321.05506$	Not detected
7		$[M-H^+]=563.15217$	$[M-H^+]=563.15177$
8		$[M-H^+]=330.06089$	Not detected
9		$[M-H^+]=250.09456$	$[M-H^+]=250.09482$

References

1. B. Liu and J. Liu, *ACS Appl. Mater. Interfaces*, 2015, **7**, 24833-24838.
2. B. Liu, L. Ma, Z. Huang, H. Hu, P. Wu and J. Liu, *Mater. Horizons*, 2018, **5**, 65-69.
3. B. Liu, Z. Sun, P.-J. J. Huang and J. Liu, *J. Am. Chem. Soc.*, 2015, **137**, 1290-1295.
4. Y. Zhao, Y. Wang, A. Mathur, Y. Wang, V. Maheshwari, H. Su and J. Liu, *Nanoscale*, 2019.