Template effect of innocent and coordinating anions on the formation of interpenetrated 2D and 3D networks: methyl orange and iodine sorption studies

Fayaz Baig^a, Krishnan Rangan^b, Shibu M Eappen^c, Sanjay K. Mandal^{d*} and Madhushree Sarkar^{a*}

- ^{a.} Department of Chemistry, BITS Pilani, Pilani Campus, Rajasthan 333031, India. Email: <u>msarkar@pilani.bits-pilani.ac.in</u>
- ^{b.} Department of Chemistry, BITS Pilani, Hyderabad Campus, Jawahar Nagar Shameerpet Mandal Ranga Reddy District Hyderabad 500078, India.
- ^{c.} Sophiscated Test and Instrumentation Centre, Cochin University of Science and Technology Campus, Cochin 682022, India .
- ^{d.} Department of Chemical Sciences, Indian Institute of Science Education and Research Mohali, Sector 81, S.A.S. Nagar, Punjab 140 306, India. E-mail: <u>sanjaymandal@iisermohali.ac.in</u>

Supporting Information:

- IR, ¹H NMR, ¹³C NMR, Powder XRD and ORTEP of L1 (Fig. S1-S5)
- IR, Powder XRD and ORTEP of **CP1** (Fig. S6-S8), **[HL1]**[**ClO4**] (Fig. S9-S11), **CP2** (Fig. S12-S14) and **CP2_BN** (Fig. S15, S16 and S18)
- UV-Visible spectra for the dye degradation studies of CP2 (Fig. S17)
- Crystallographic data and refinement parameters of CP2_BN (Table S1)



Fig. S1 IR spectra of L1



Fig. S2 ¹H NMR of L1

L1







Fig. S4 Simulated and experimental powder XRD of L1



Fig. S5 ORTEP drawing of L1 showing thermal ellipsoids at the 50% probability level



Fig. S6 IR spectra of CP1



(b)



Fig. S7 (a) Simulated and (b) experimental powder XRD of CP1



Fig. S8 ORTEP drawing of **CP1** showing thermal ellipsoids at the 50% probability level; Symmetry codes: a: -x, 1-y, 1-z, b: 2-x, 1-y, -z; c: -1+x, +y, -1+z



Fig. S9 IR spectra of [HL1][ClO4]



Fig. S10 Simulated and experimental powder XRD of [HL1][ClO4]



Fig. S11 ORTEP drawing of [HL1][ClO4] showing thermal ellipsoids at the 50% probability level



Fig. S12 IR spectra of CP2



Fig. S13 (a) Simulated and (b) experimental PXRD of CP2



Fig. S14 ORTEP drawing of **CP2** showing thermal ellipsoids at the 50% probability level, Symmetry Codes: a: 1+x, y, z; b: x, y, 1+z; c: 1+x, 1+y, z; d: x-1, y, z; e: x-1, y-1, z; f: x, y, z-1







Fig. S16 Calculated and experimental PXRD of CP2_BN



Fig. S17 UV-Visible spectra of dye degradation by **CP2** from aqueous solutions of (a) 5×10^{-4} M MB, (b) 10^{-5} M MG, and (c) 2×10^{-5} M RB

Table S1 Crystallographic data and refinement parameters of CP2_BN

Compound CP2_BN	
-----------------	--

Empirical formula	C _{36.8} H ₃₄ Cd N _{8.4} O ₆ S ₂
Formula Wt.	866.45
Temperature/K	298.15
Crystal system	Triclinic
Space group	P-1
a/Å	9.8814(3)
b/Å	14.1900(4)
c/Å	15.7164(5)
α/°	94.950(3)
β/°	91.893(3)
γ/°	100.060(3)
$V/Å^3$	2159.13(11)
Ζ	2
$\rho_{calc}g/cm^3$	1.334
μ/mm^{-1}	0.649
F(000)	894
Crystal size/mm ³	0.3 imes 0.2 imes 0.2
Radiation	MoKα (λ = 0.71073)
2@ range for data collection/°	9.786 to 49.998
Index ranges	$-10 \le h \le 11$,
	$-16 \le k \le 16$,
	$-18 \le l \le 18$
Reflections collected	28332
Independent reflections	7608 [$R_{int} = 0.0227, R_{sigma} = 0.0194$]
Data / restraints / parameters	7539/0/464
Goodness-of-fit on F ²	1.065
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0551, wR_2 = 0.1377$
Final R indexes [all data]	$R_1 = 0.0591, wR_2 = 0.1399$



Fig S18 ORTEP of asymmetric unit of CP2_BN showing thermal ellipsoids at the 50% probability level