Sensitive photoluminescent chemosensor for cyanide in water based on a zinc coordination polymer bearing ditert-butyl-bipyridine.

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Table S1. Crystallographic data for the polymer, 1

Crystal data and structure refinement for Polym	ner (1)		
Identification code	mo_050GOI16_0m		
Empirical formula	C20 H28 N2 O7 Zn		
Formula weight	473.81		
Temperature	100(2) K		
Wavelength	0.71073 Å		
Crystal system	Monoclinic		
Space group	P2 ₁ /c		
Unit cell dimensions	a = 8.8493(6) Å	$\alpha = 90^{\circ}.$	
	b = 25.7425(15) Å	$\beta = 106.8621(12)^{\circ}.$	
	c = 9.7652(6) Å	$\gamma = 90^{\circ}.$	
Volume	2128.9(2) Å ³		
Z	4		
Density (calculated)	1.478 Mg/m ³		
Absorption coefficient	1.198 mm ⁻¹		
F(000)	992		
Crystal size	0.454 x 0.168 x 0.112 mm ³		
Theta range for data collection	2.318 to 27.439°.		
Index ranges	-11<=h<=11, -28<=k<=33, -10<=l<=12		
Reflections collected	23490		
Independent reflections	4876 [R(int) = 0.0358]		
Completeness to theta = 25.242°	99.9 %		
Absorption correction	None		
Refinement method	Full-matrix least-squares on	F ²	
Data / restraints / parameters	4876 / 13 / 297		
Goodness-of-fit on F ²	1.027		
Final R indices [I>2sigma(I)]	R1 = 0.0291, wR2 = 0.0716		
R indices (all data)	R1 = 0.0340, wR2 = 0.0741		
Extinction coefficient	n/a		
Largest diff. peak and hole	0.487 and -0.319 e.Å ⁻³		

Table S2. Crystallographic data for the polymer, 2

Crystal data and structure refinement for Polym	ner 2		
Identification code	mo_117MLRA15_0m		
Empirical formula	C20 H24 N2 O5 Zn		
Formula weight	437.78		
Temperature	293(2) K		
Wavelength	0.71073 Å		
Crystal system	Monoclinic		
Space group	P21/c		
Unit cell dimensions	a = 10.0221(5) Å	$\alpha = 90^{\circ}.$	
	b = 20.7033(9) Å	$\beta = 113.2768(10)^{\circ}.$	
	c = 10.1283(5) Å	$\gamma = 90^{\circ}.$	
Volume	1930.48(16) Å ³		
Z	4		
Density (calculated)	1.506 Mg/m ³		
Absorption coefficient	1.307 mm ⁻¹		
F(000)	912		
Crystal size	0.388 x 0.234 x 0.052 mm ³		
Theta range for data collection	1.967 to 26.371°.		
Index ranges	-12<=h<=12, -25<=k<=25, -12<=l<=12		
Reflections collected	38671		
Independent reflections	3944 [R(int) = 0.0664]		
Completeness to theta = 25.242°	99.9 %		
Absorption correction	None		
Refinement method	Full-matrix least-squares on	F ²	
Data / restraints / parameters	3944 / 3 / 261		
Goodness-of-fit on F ²	1.125		
Final R indices [I>2sigma(I)]	R1 = 0.0321, $wR2 = 0.0862$		
R indices (all data)	R1 = 0.0343, wR2 = 0.0873		
Extinction coefficient	n/a		
Largest diff. peak and hole	0.365 and -0.312 e.Å ⁻³		

Table S3. Crystallographic data for the polymer, 3

Crystal data and structure refinement for Polym	ner (3)		
Identification code	cu_096GOI16_0m		
Empirical formula	C52 H86 N4 O17 Zn2		
Formula weight	1169.98		
Temperature	100(2) K		
Wavelength	1.54178 Å		
Crystal system	Monoclinic		
Space group	Pn		
Unit cell dimensions	a = 9.2334(2) Å	$\alpha = 90^{\circ}$.	
	b = 11.2034(3) Å	$\beta = 95.9508(13)^{\circ}.$	
	c = 27.9409(7) Å	$\gamma = 90^{\circ}$.	
Volume	2874.78(12) Å ³		
Z	2		
Density (calculated)	1.352 Mg/m ³		
Absorption coefficient	1.612 mm ⁻¹		
F(000)	1244		
Crystal size	0.362 x 0.069 x 0.044 mm ³		
Theta range for data collection	3.180 to 68.242°.		
Index ranges	-10<=h<=11, -13<=k<=13, -33<=l<=33		
Reflections collected	24041		
Independent reflections	8618 [R(int) = 0.0319]		
Completeness to theta = 67.679°	99.9 %		
Absorption correction	None		
Refinement method	Full-matrix least-squares on	F ²	
Data / restraints / parameters	8618 / 97 / 774		
Goodness-of-fit on F ²	1.032		
Final R indices [I>2sigma(I)]	R1 = 0.0299, $wR2 = 0.0766$		
R indices (all data)	R1 = 0.0307, wR2 = 0.0773		
Absolute structure parameter	0.49(2)		
Extinction coefficient	n/a		
Largest diff. peak and hole	0.448 and -0.269 e.Å ⁻³		

Table S4 Hydrogen bonds for 1 [Å and °].

D-HA	d(D-H)	d(HA)	d(DA)	<(DHA)
O(5)-H(5A)O(6)	0.827(15)	1.934(16)	2.7535(19)	171(2)
O(5)-H(5B)O(1)	0.834(15)	2.55(2)	2.9423(18)	110.3(17)
O(5)-H(5B)O(2)	0.834(15)	1.799(15)	2.6193(18)	167(2)
O(6)-H(6A)O(2)#3	0.839(15)	1.922(15)	2.7567(18)	173(2)
O(6)-H(6B)O(7)#4	0.842(18)	1.997(17)	2.8328(19)	172(4)
O(6)-H(6C)O(7)	0.830(18)	1.917(18)	2.741(2)	172(4)
O(7)-H(7A)O(4)#1	0.850(15)	1.946(16)	2.7837(18)	169(2)
O(7)-H(7B)O(6)	0.848(18)	1.91(2)	2.741(2)	168(4)
O(7)-H(7C)O(6)#4	0.838(18)	2.007(19)	2.8328(19)	169(4)

Symmetry transformations used to generate equivalent atoms:

#1 x+1,y,z #2 x-1,y,z #3 -x+1,-y+1,-z #4 -x+2,-y+1,-z

D-HA	d(D-H)	d(HA)	d(DA)	<(DHA)	
C(5)-H(5)O(4)#3	0.93	2.34	3.197(3)	153.5	
O(5)-H(5A)O(3)#4	0.828(17)	1.830(18)	2.657(3)	176(4)	
O(5)-H(5B)O(2)	0.843(17)	1.811(18)	2.618(3)	160(3)	

Table S5 Hydrogen bonds for 2 [Å and °].

Symmetry transformations used to generate equivalent atoms:

#1 x-1,-y+1/2,z-1/2 #2 x+1,-y+1/2,z+1/2 #3 -x+1,y+1/2,-z+3/2 #4 x-1,y,z

Table S6	. Hvdrogen	bonds t	for 3	ſÅ	and	°].
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D-HA	d(D-H)	d(HA)	d(DA)	<(DHA)
O(5)-H(5A)O(12)	0.84(2)	1.94(3)	2.732(4)	156(4)
O(5)-H(5B)O(4)#1	0.83(2)	1.84(3)	2.645(4)	162(5)
O(10)-H(10A)O(11)	0.83(2)	1.85(2)	2.676(4)	171(5)
O(10)-H(10B)O(8)#2	0.83(2)	1.85(2)	2.673(4)	168(5)
O(11)-H(11A)O(4)	0.88(2)	1.92(2)	2.765(4)	162(4)
O(11)-H(11B)O(13)	0.85(2)	1.97(3)	2.751(5)	151(5)
O(12)-H(12D)O(13)	0.84(2)	1.93(3)	2.709(5)	153(5)
O(12)-H(12E)O(8)	0.85(2)	1.88(3)	2.706(4)	165(6)
O(13)-H(13D)O(16)	0.87(2)	1.90(2)	2.737(5)	162(5)
O(13)-H(13E)O(14)	0.87(2)	1.85(2)	2.713(5)	171(5)
O(14)-H(14D)O(15)	0.92(2)	1.90(3)	2.818(5)	171(5)
O(14)-H(14E)O(6)	0.88(2)	1.90(3)	2.763(4)	168(4)
O(15)-H(15A)O(8)#2	0.90(2)	2.05(3)	2.939(5)	167(5)
O(15)-H(15B)O(17)	0.89(2)	2.00(3)	2.844(4)	156(5)
O(16)-H(16D)O(2)	0.88(2)	1.87(3)	2.748(4)	175(5)
O(16)-H(16E)O(17)#1	0.88(2)	2.08(3)	2.905(5)	156(4)
O(17)-H(17D)O(12)#2	0.89(2)	2.04(3)	2.901(5)	161(5)
O(17)-H(17E)O(4)	0.86(2)	2.06(3)	2.909(5)	170(6)

Symmetry transformations used to generate equivalent atoms:

#1 x-1,y,z #2 x+1,y,z

Bond	Distance (Å)	Bonds	Angles (°)
Zn(1)-O(1)	2.041(1)	O(1)-Zn(1)-N(2)	171.22(5)
Zn(1)-O(5)	2.079(1)	O(5)-Zn(1)-O(3)	163.52(5)
Zn(1)-N(1)	2.109(1)	N(1)- $Zn(1)$ - $O(4)$	145.48(5)
Zn(1)-O(4)	2.142(1)	O(5)-Zn(1)-N(1)	108.24(6)
Zn(1)-N(2)	2.163(1)	O(5)-Zn(1)-O(4)	104.16(5)
Zn(1)-O(3)	2.256(1)	O(1)-Zn(1)-O(4)	96.80(5)
		O(4)-Zn(1)-N(2)	91.22(5)
		N(2)-Zn(1)-O(3)	87.12(5)
		N(1)- $Zn(1)$ - $O(3)$	87.07(5)
		N(1)- $Zn(1)$ - $N(2)$	77.15(6)
		O(4)-Zn(1)-O(3)	59.65(5)

Table S7. Selected bond distances and bond angles for 1

Table S8. Selected bond distances and bond angles for 2

Bond	Distance (Å)	Bonds	Angles (°)
Zn(1)-O(1)	1.988(2)	O(5)-Zn(1)-N(2)	169.49(9)
Zn(1)-O(3)	2.042(2)	O(4)-Zn(1)-O(1)	155.26(7)
Zn(1)-O(5)	2.050(2)	O(3)-Zn(1)-N(1)	136.61(8)
Zn(1)-N(1)	2.102(2)	O(1)-Zn(1)-N(1)	121.20(8)
Zn(1)-N(2)	2.172(2)	O(1)-Zn(1)-O(3)	99.86(8)
Zn(1)-O(4)	2.557(2)	O(5)-Zn(1)-N(1)	97.69(9)
		O(3)-Zn(1)-O(5)	95.36(10)
		O(3)-Zn(1)-N(2)	94.83(8)
		O(1)-Zn(1)-N(2)	85.48(8)
		N(1)- $Zn(1)$ - $N(2)$	76.49(8)
		O(4)-Zn(1)-O(3)	55.40(7)

Table S9. Selected bond distances and bond angles for 3

Bond	Distance (Å)	Bonds	Angles (°)
Zn(1)-O(1)	2.031(3)	O(3)-Zn(1)-N(2)	165.29(11)
Zn(1)-O(5)	2.057(3)	O(1)- $Zn(1)$ - $O(5)$	151.33(12)
Zn(1)-O(3)	2.070(3)	N(1)-Zn(1)-O(2)	148.82(10)
Zn(1)-N(1)	2.111(3)	O(5)- $Zn(1)$ - $N(1)$	113.32(11)
Zn(1)-N(2)	2.151(3)	O(3)- $Zn(1)$ - $O(2)$	103.59(10)
Zn(1)-O(2)	2.488(3)	O(1)-Zn(1)-O(3)	95.32(11)
		O(1)- $Zn(1)$ - $N(2)$	93.80(12)
		N(2)-Zn(1)-O(2)	91.05(11)
		O(5)- $Zn(1)$ - $N(2)$	88.42(12)
		N(1)- $Zn(1)$ - $N(2)$	76.28(12)
		O(1)-Zn(1)-O(2)	57.06(11)



Fig. S1. 1D hydrogen-bonding supramolecular array of 1; hydrogens are omitted for clarity.



Fig. S2. 2D hydrogen-bonding supramolecular packing of 2; 5dmb ligand is omitted for clarity.



Fig. S3. IR-ATR of polymer, 1



Fig. S4. IR-ATR of polymer, 2



Fig. S5. IR-ATR of polymer, 3



Fig. S6. TGA plots for polymer, 1



Fig. S7. TGA plots for polymer, 2



Fig. S8. TGA plots for polymer, 3



Fig. S9. PXRD pattern (bottom) and simulated patterns (top) for 3.



Fig. S10. Superimposed molecules 1,4-chdc and 4,4'-dtbb using the program Mercury. (a) According with the numbering of SCXRD colored in blue is the six-membered ring (C21, C22, C23, C24, C25, C26) and colored in red is the ring (C45, C46, C47, C48, C49, C50). (b) According with the numbering of SCXRD colored in blue are the atoms of 4,4'-dtbb molecule (C27, C28, C29, C30, C31, C32, C33, C34, C35, C36) and colored in red are the atoms (C1, C2, C3, C4, C5, C6, C7, C8, C9, C10).



Fig. S11. ¹³C CPPI and CPMAS NMR spectra (spinning rate at 7kHz) of **3**. In the case of CPPI experiment 30 μ s of a polarization inversion period was used.



Fig. S12. ¹³C CPPI and CPMAS NMR (spinning rate at 7kHz) spectra of 4,4'-dtbb. In the case of CPPI experiment 30 μ s of a polarization inversion period was used.



Figure S13. ¹H NMR spectrum of sodium salt of 1,4-cyclohexanedicarboxylato in D₂O



Fig. S14. Connectivity RCOO-Zn for the polymers 1-2.



Fig. S15. Connectivity RCOO-Zn for the polymer 3.



Fig. S16. Fluorescence response of **3** (10 μ M) upon addition of CN⁻ at pH= 7.0 in the presence of sodium salts of F⁻, Cl⁻, H₂PO₄⁻, HCO₃⁻, NO₃⁻, H₂AsO₃⁻ and SO₄⁻² (1.0 mM each).