

Electronic Supplementary Information (ESI) for

Unsymmetrical β -Functionalized 'Push-Pull' Porphyrins: Synthesis, Photophysical, Electrochemical and Nonlinear Optical Properties

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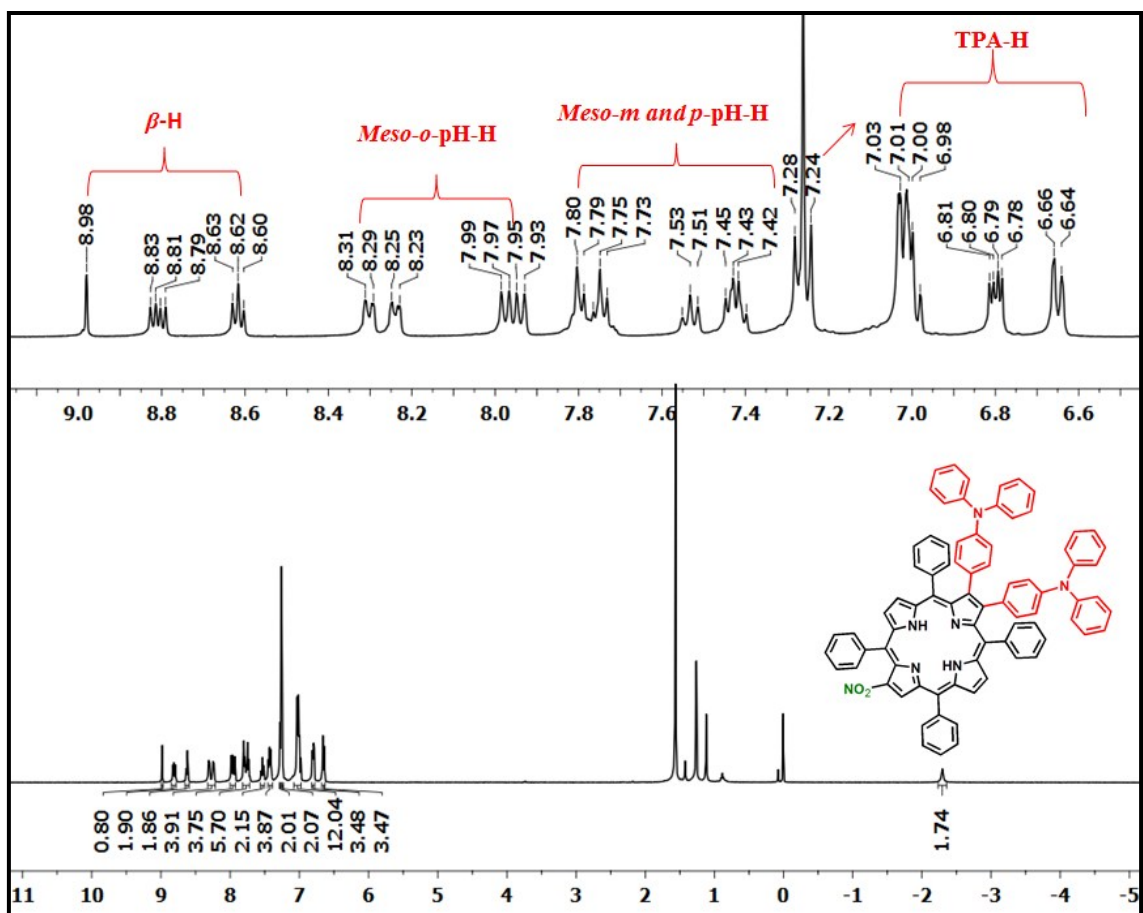


Figure S1. ^1H NMR spectrum of $\text{H}_2\text{TPP}(\text{TPA})_2\text{NO}_2$ in CDCl_3 at 298 K.

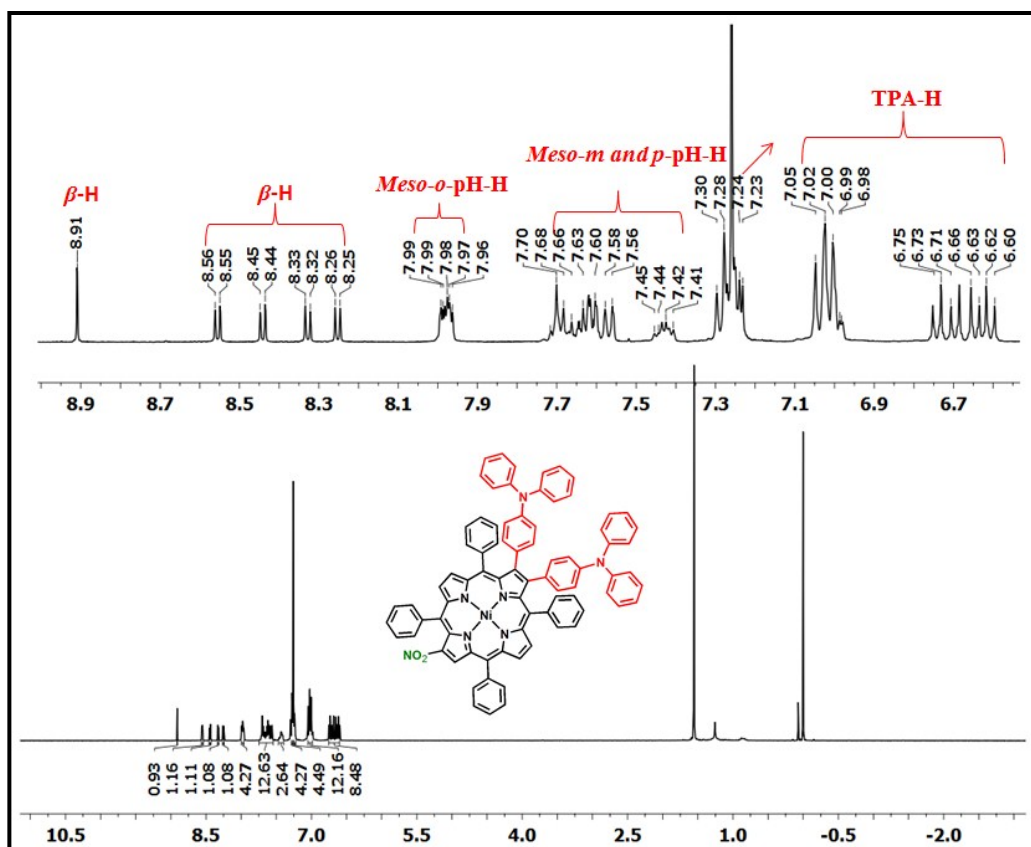


Figure S2. ¹H NMR spectrum of NiTPP(TPA)₂NO₂ in CDCl₃ at 298 K.

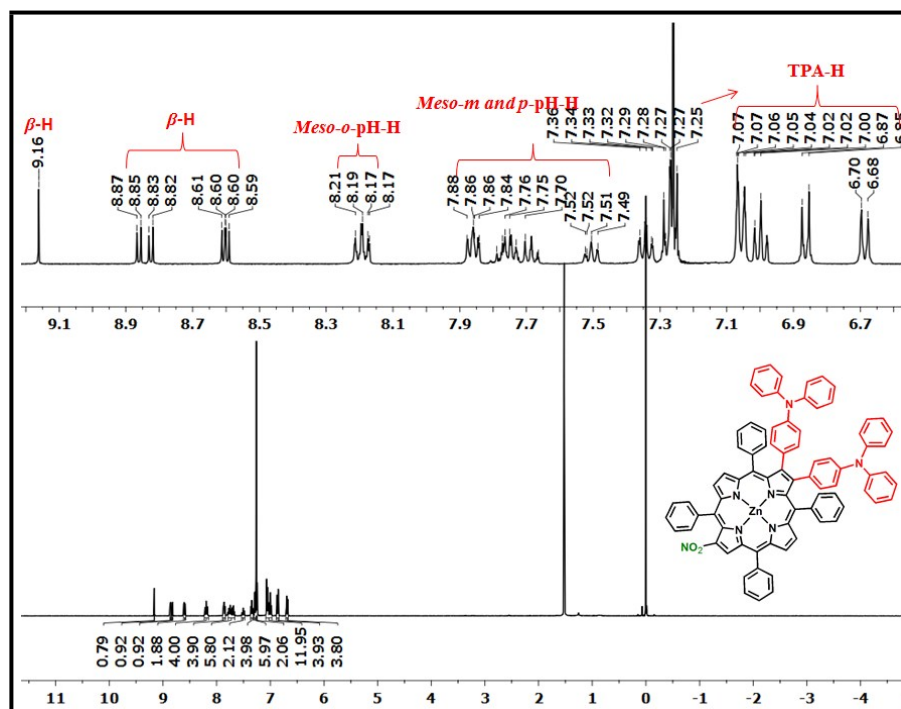


Figure S3. ¹H NMR spectrum of ZnTPP(TPA)₂NO₂ in CDCl₃ at 298 K.

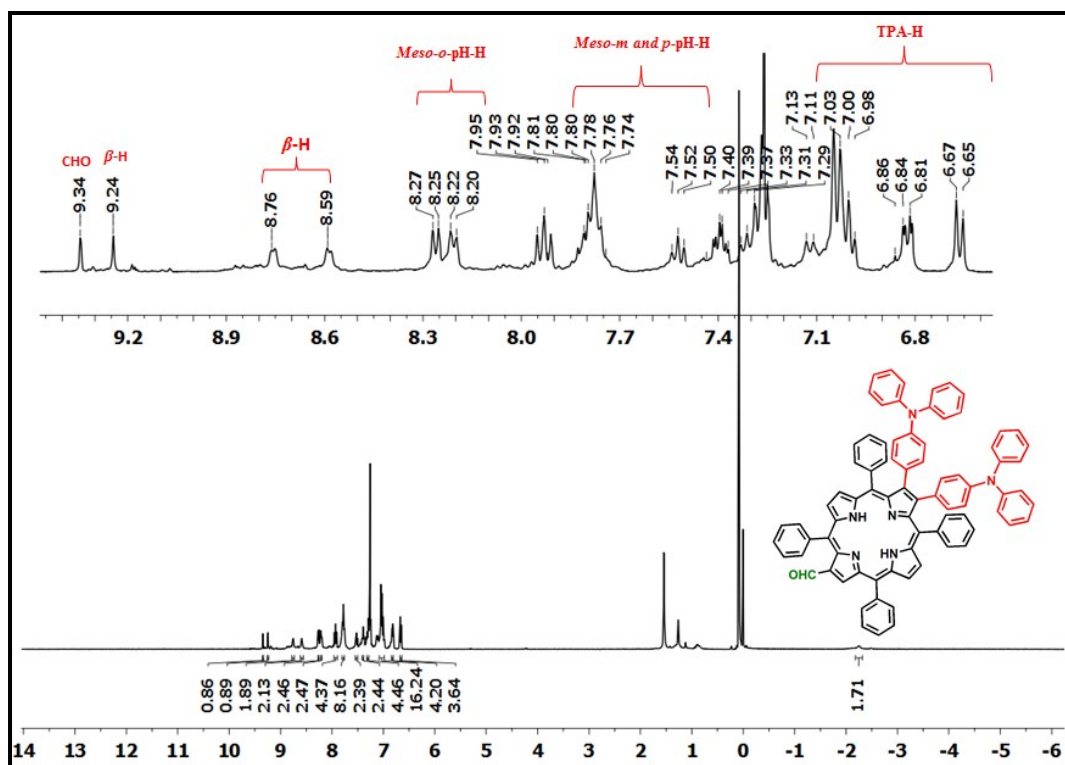


Figure S4. ^1H NMR spectrum of $\text{H}_2\text{TPP}(\text{TPA})_2\text{CHO}$ in CDCl_3 at 298 K.

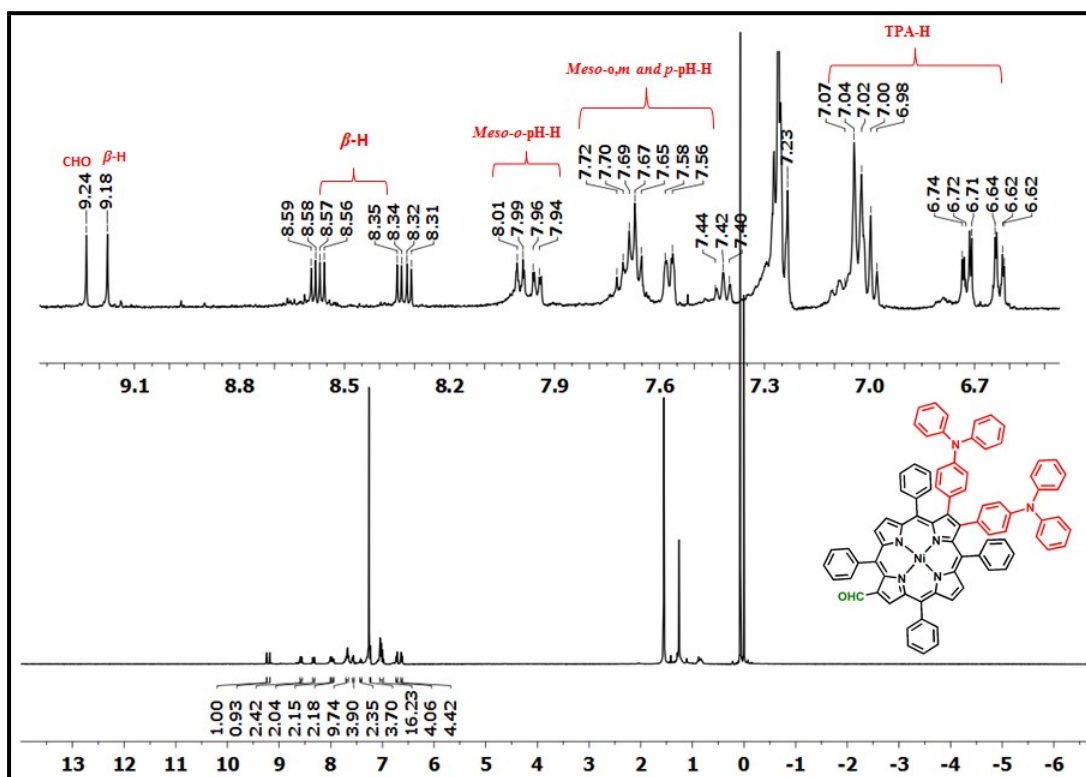
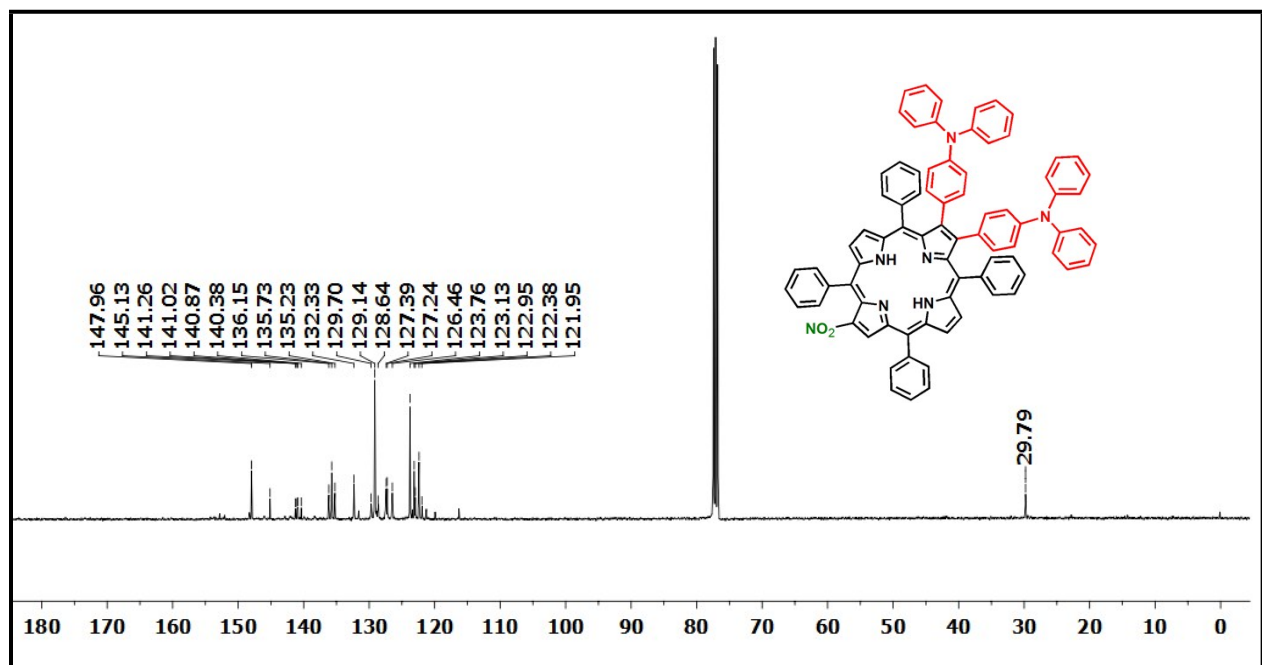
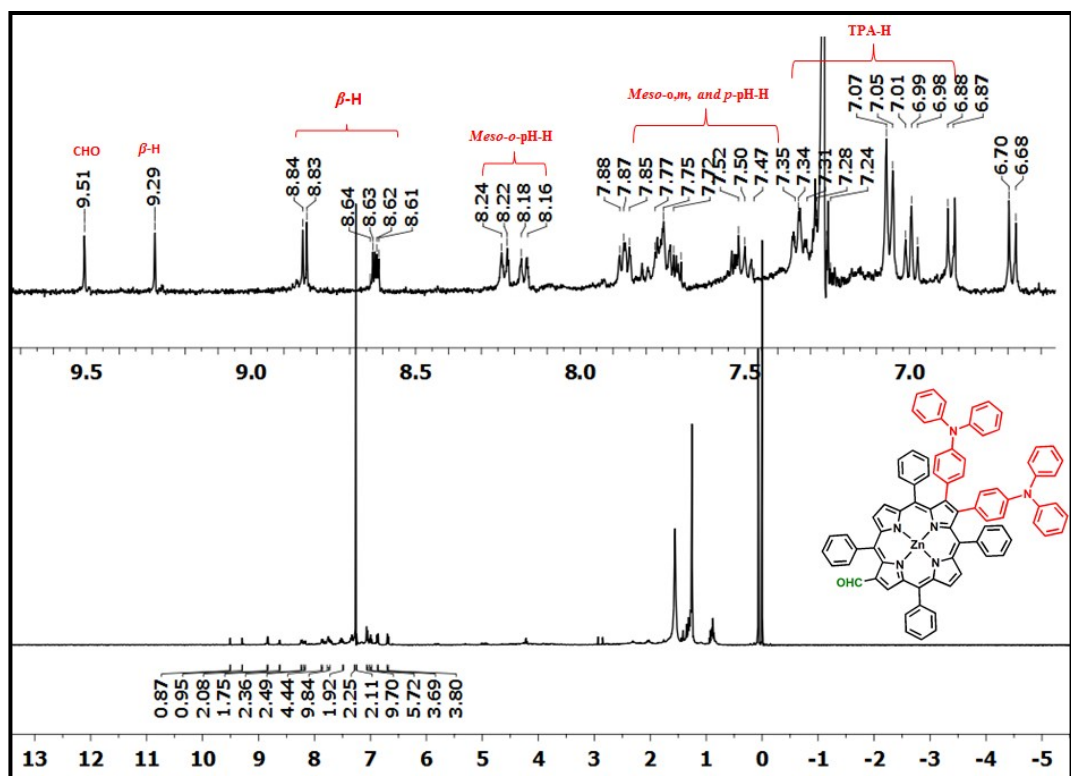


Figure S5. ^1H NMR spectrum of $\text{NiTPP}(\text{TPA})_2\text{CHO}$ in CDCl_3 at 298 K.



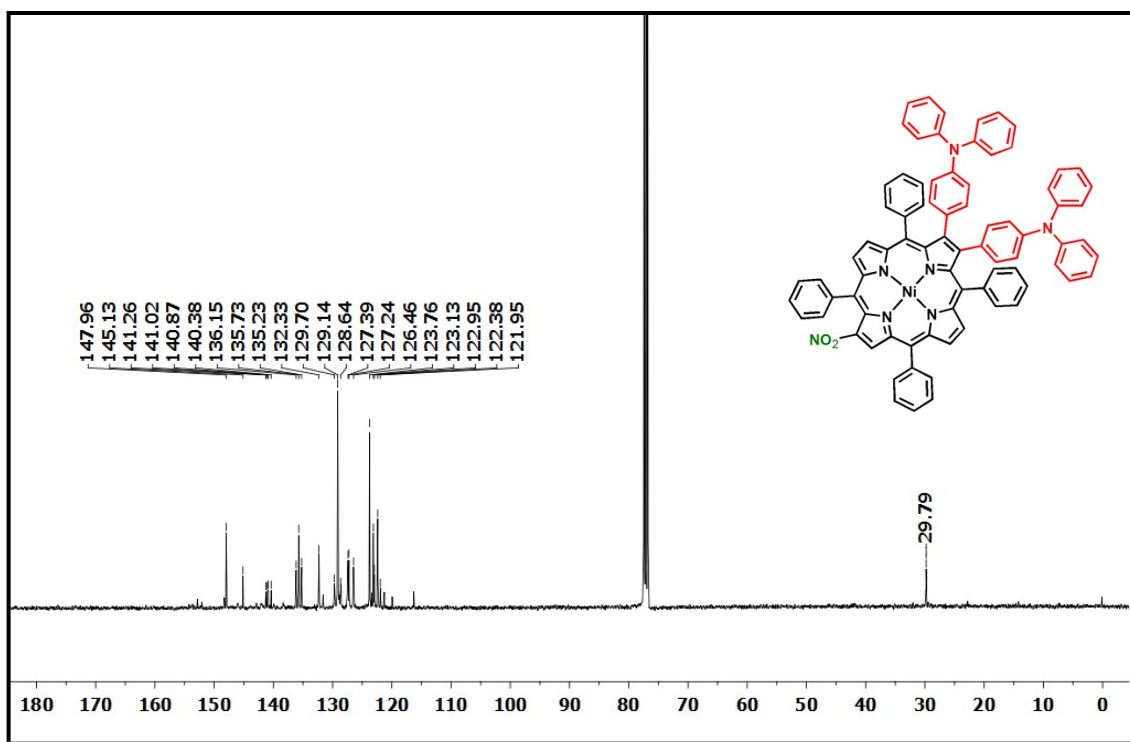


Figure S8. ^{13}C NMR spectrum of $\text{NiTPP}(\text{TPA})_2\text{NO}_2$ in CDCl_3 at 298 K

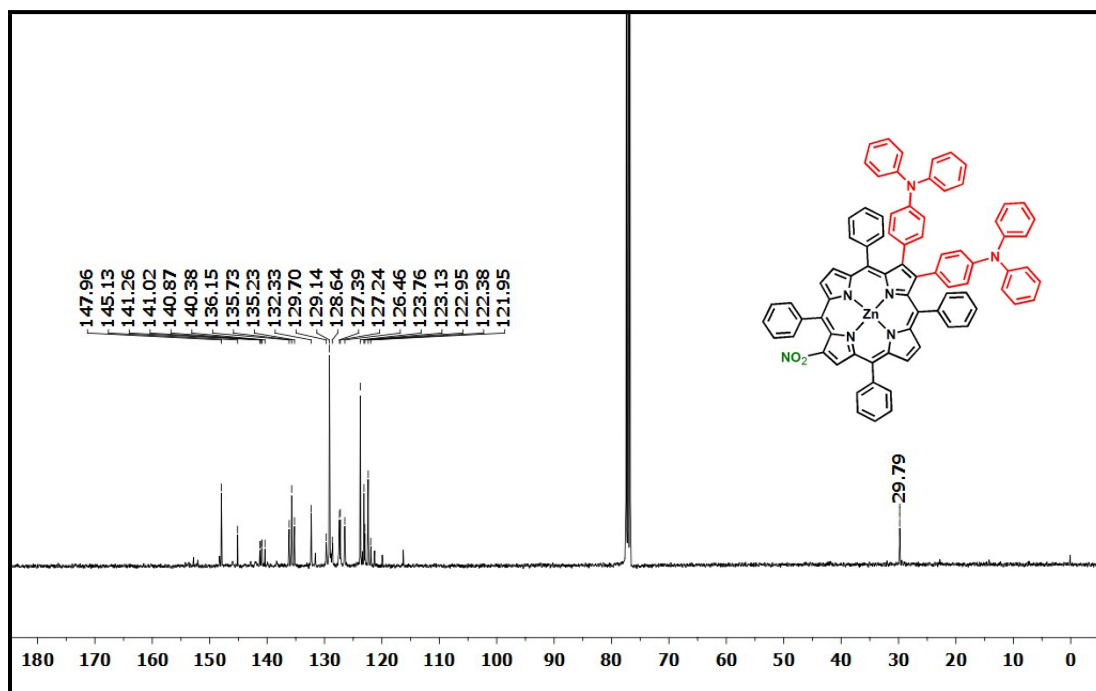


Figure S9. ^{13}C NMR spectrum of $\text{ZnTPP}(\text{TPA})_2\text{NO}_2$ in CDCl_3 at 298 K

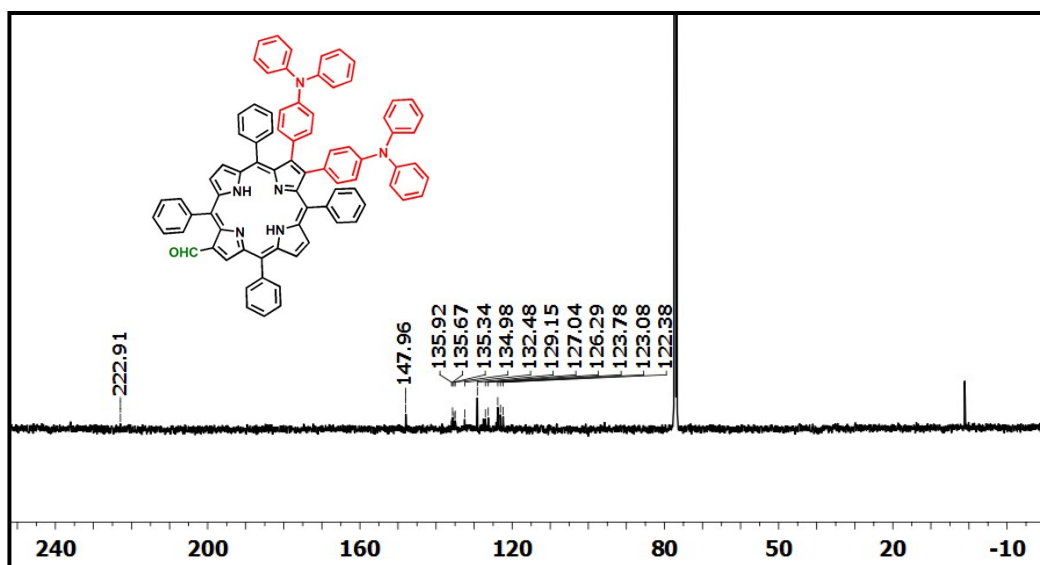


Figure S10. ^{13}C NMR spectrum of $H_2TPP(TPA)_2CHO$ in $CDCl_3$ at 298 K

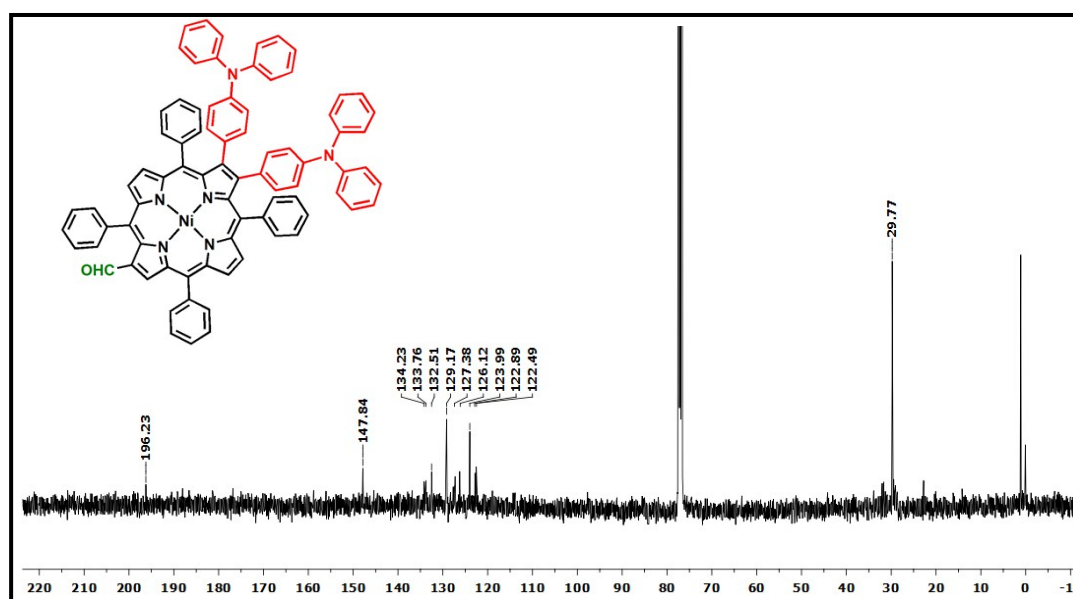


Figure S11. ^{13}C NMR spectrum of $NiTPP(TPA)_2CHO$ in $CDCl_3$ at 298 K

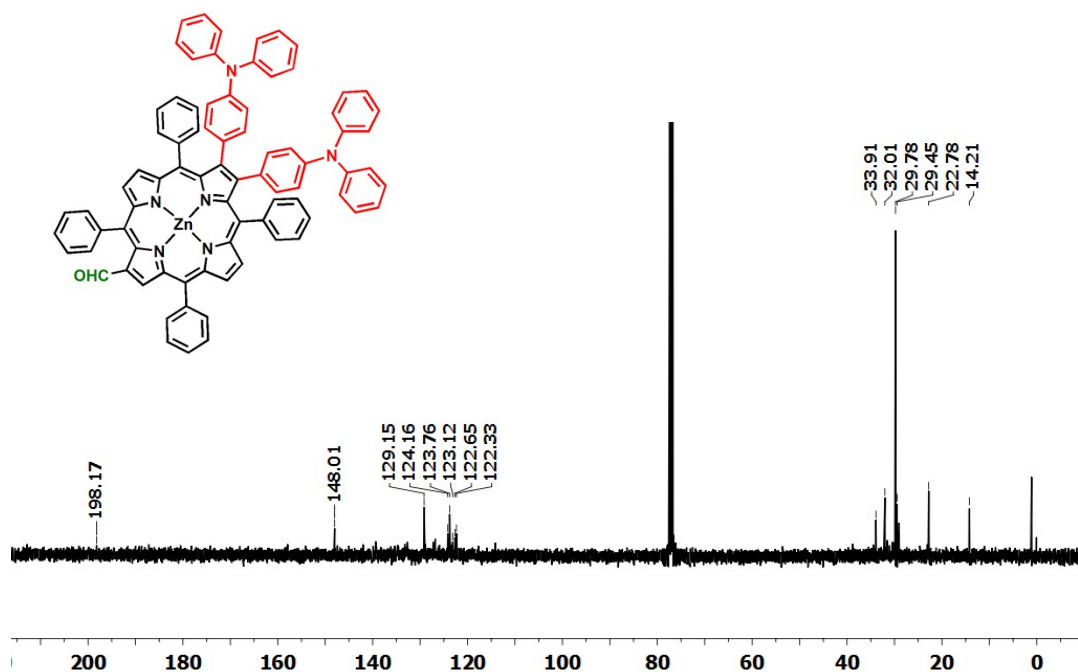


Figure S12. ¹³C NMR spectrum of ZnTPP(TPA)₂CHO in CDCl₃ at 298 K

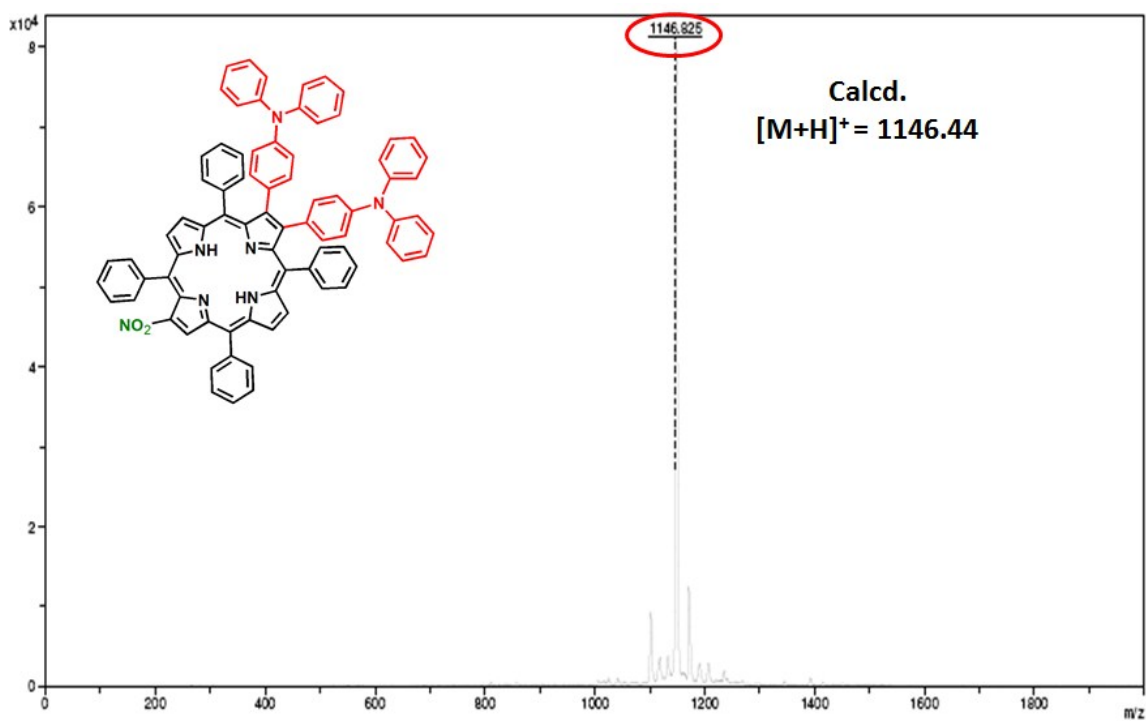


Figure S13. MALDI-TOF mass spectrum of H₂TPP(TPA)₂NO₂.

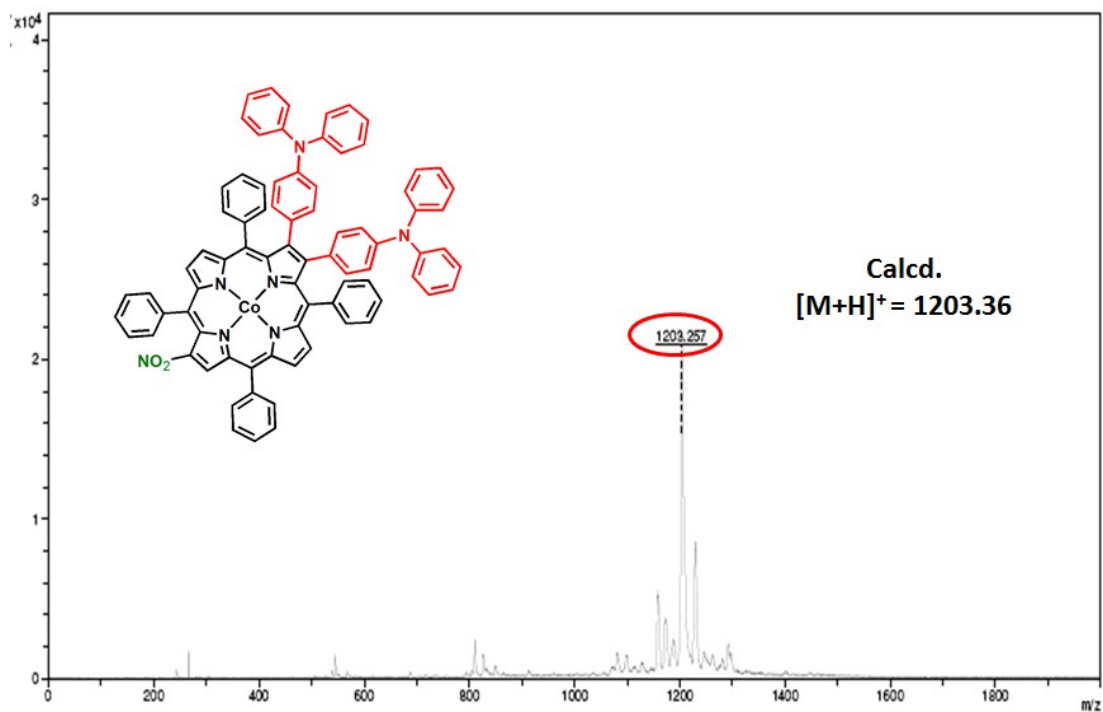


Figure S14. MALDI-TOF mass spectrum of CoTPP(TPA)₂NO₂.

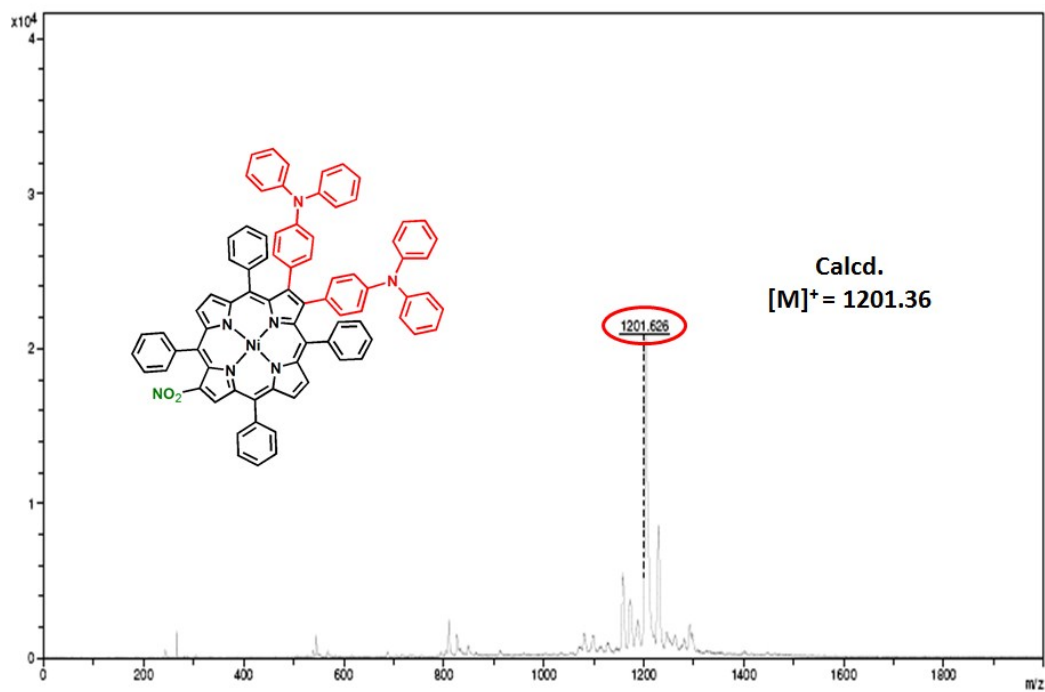


Figure S15. MALDI-TOF mass spectrum of NiTPP(TPA)₂NO₂.

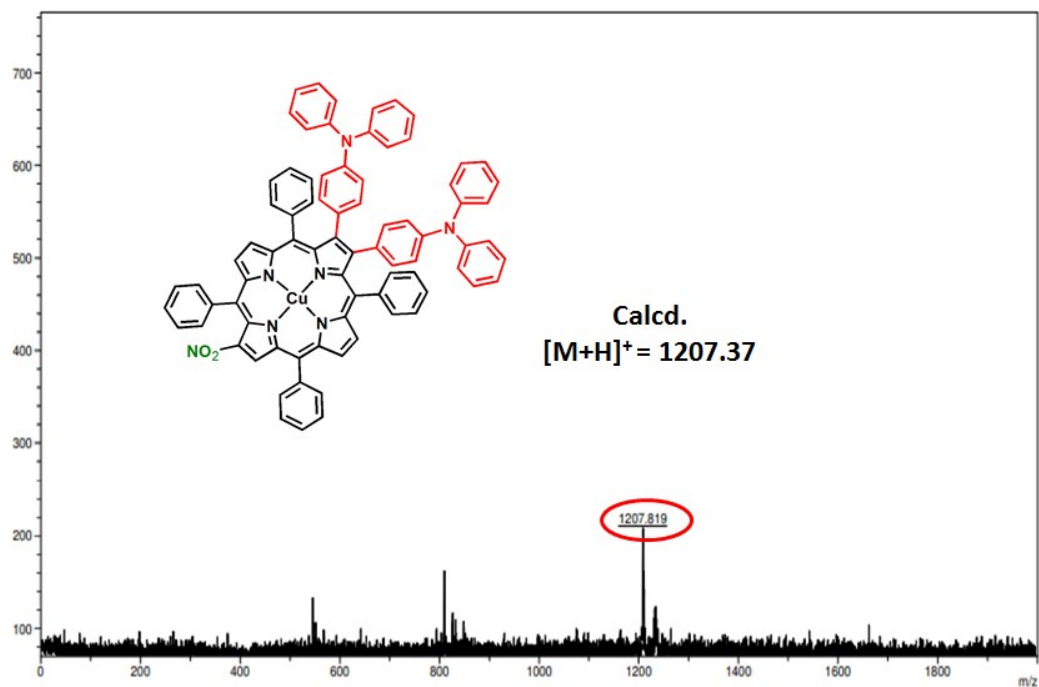


Figure S16. MALDI-TOF mass spectrum of $\text{CuTPP(TPA)}_2\text{NO}_2$.

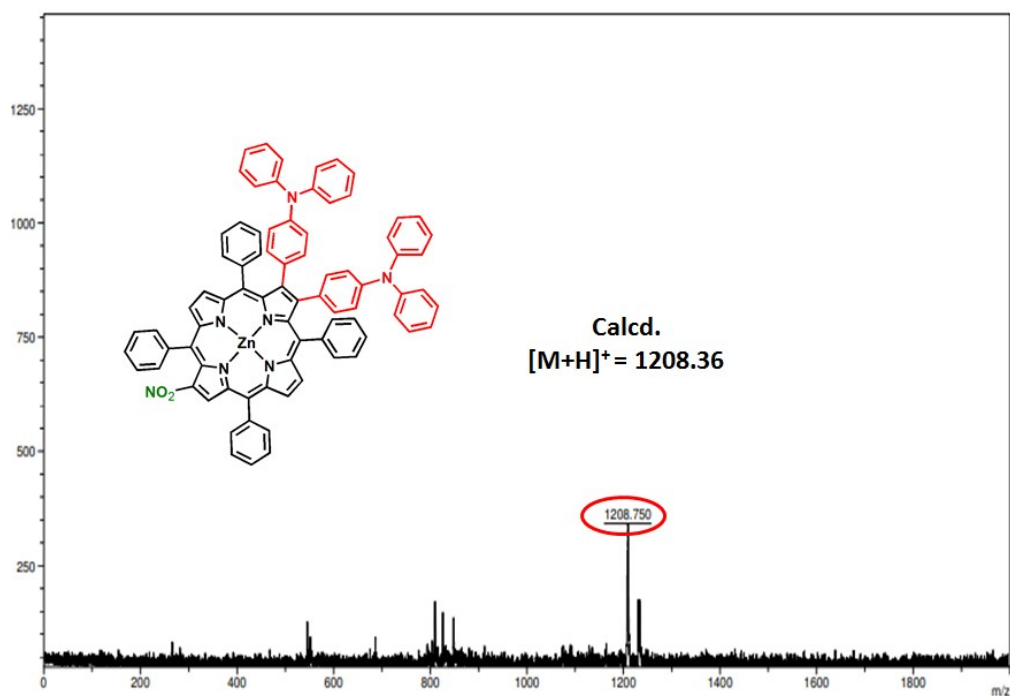


Figure S17. MALDI-TOF mass spectrum of $\text{ZnTPP(TPA)}_2\text{NO}_2$.

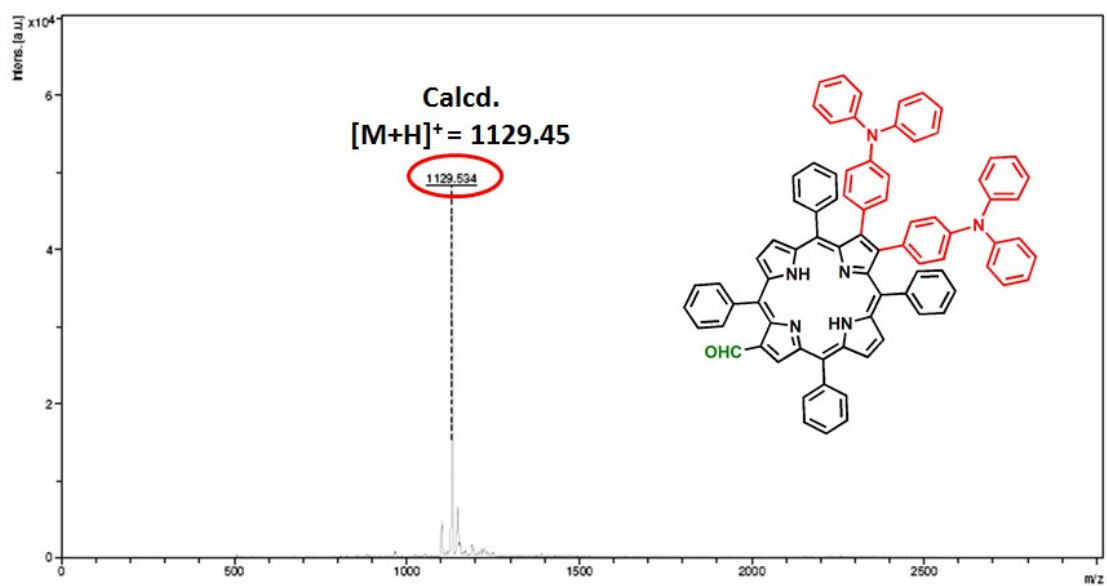


Figure S18. MALDI-TOF mass spectrum of $\text{H}_2\text{TPP}(\text{TPA})_2\text{CHO}$.

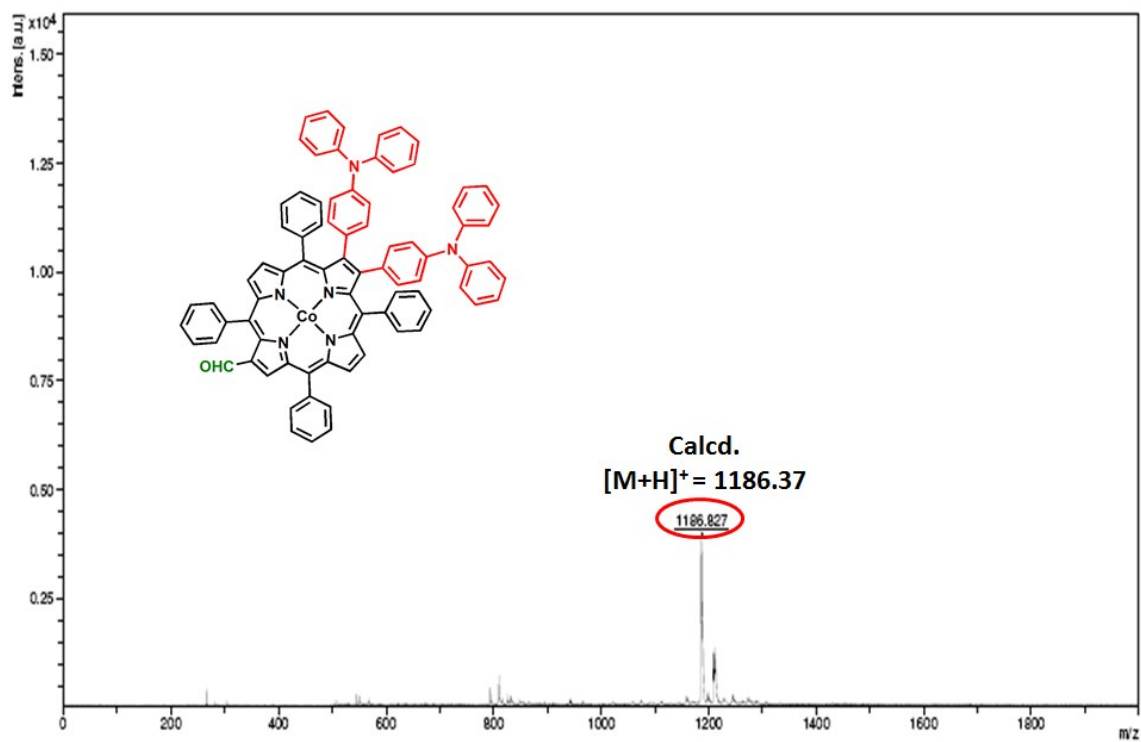


Figure S19. MALDI-TOF mass spectrum of $\text{CoTPP}(\text{TPA})_2\text{CHO}$.

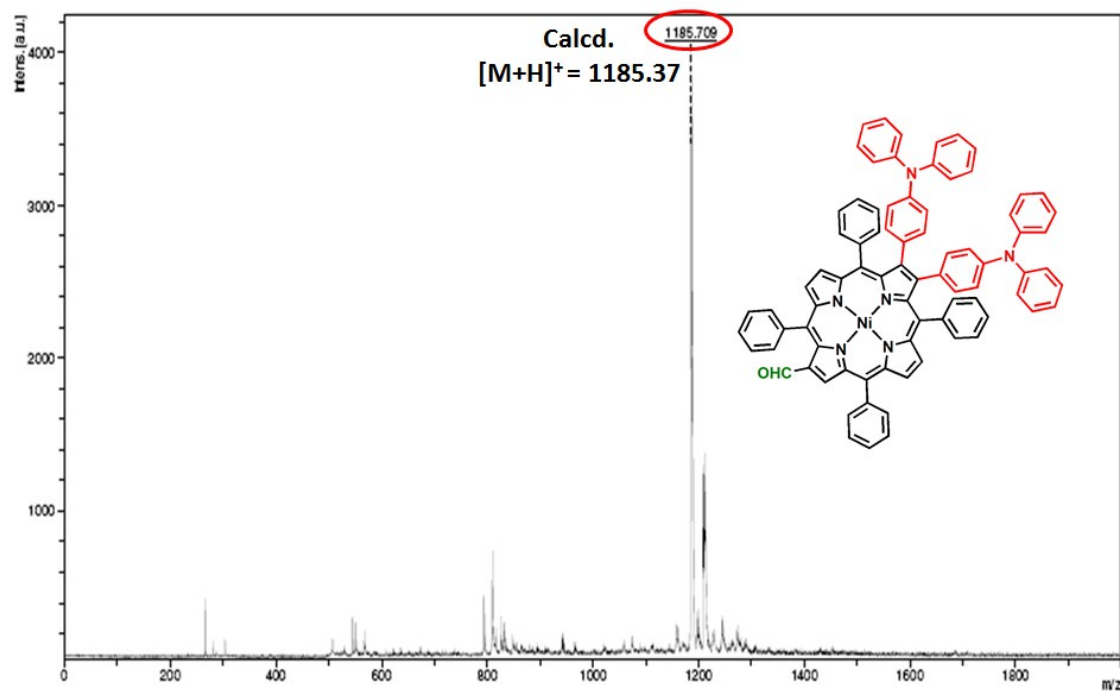


Figure S20. MALDI-TOF mass spectrum of NiTPP(TPA)₂CHO.

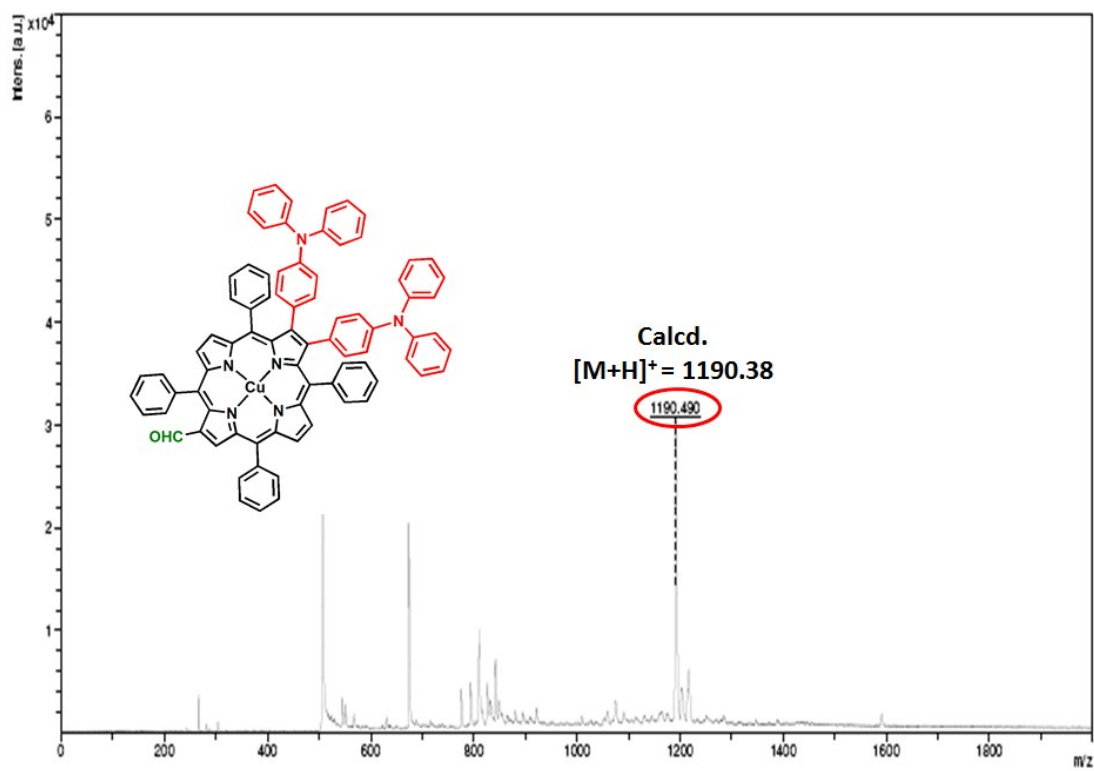


Figure S21. MALDI-TOF mass spectrum of CuTPP(TPA)₂CHO.

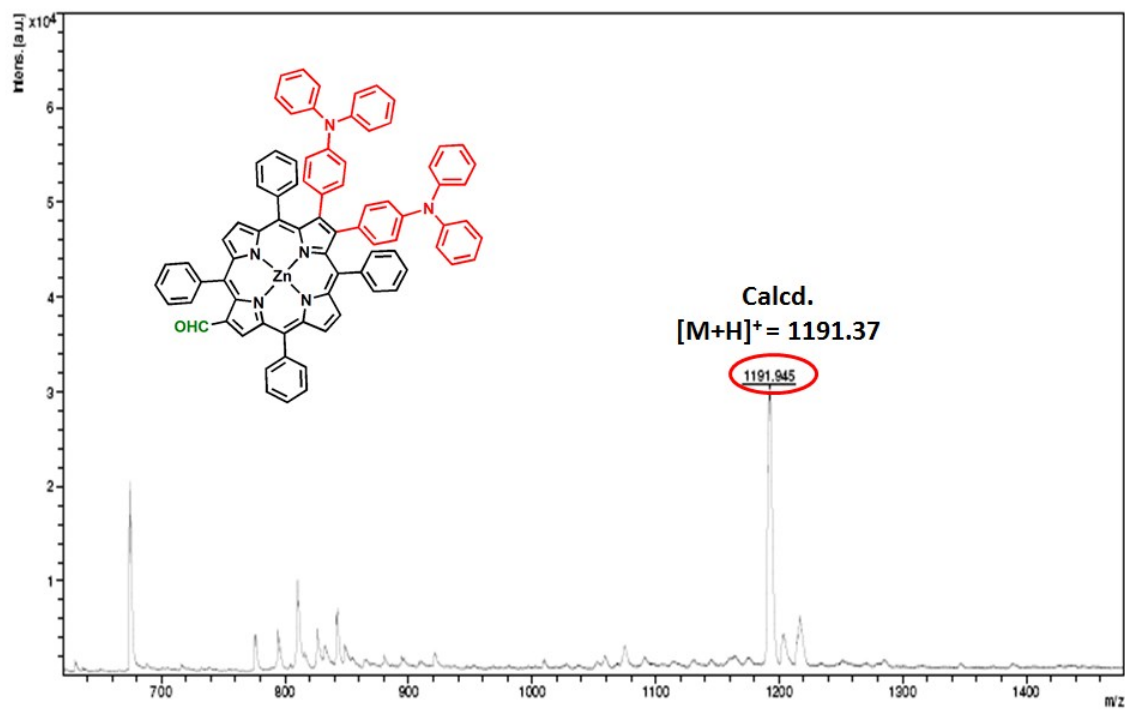


Figure S22. MALDI-TOF mass spectrum of ZnTPP(TPA)₂CHO.

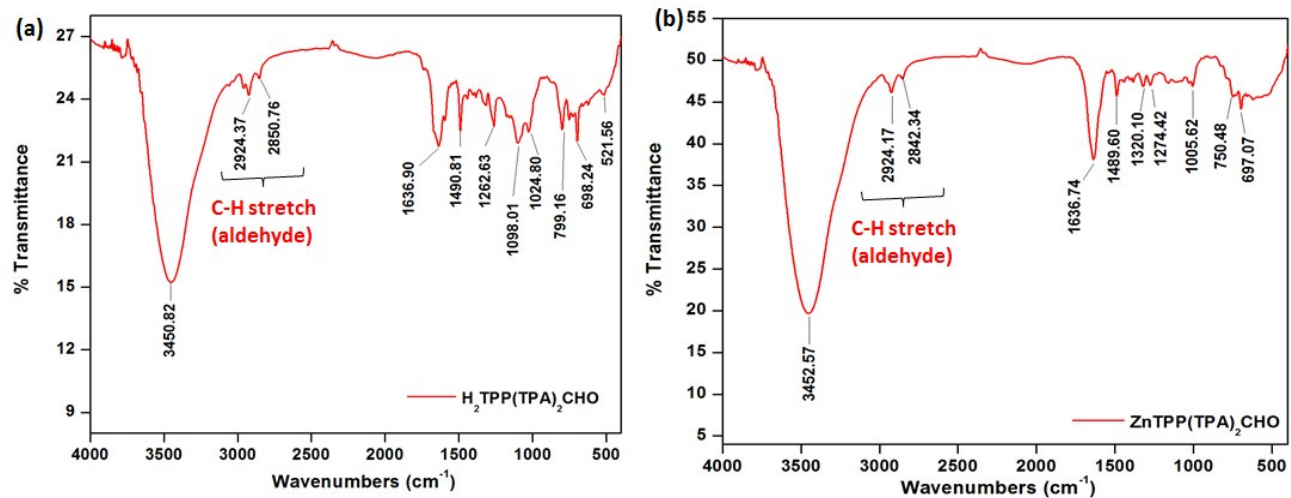


Figure S23. IR spectra of (a) $\text{H}_2\text{TPP}(\text{TPA})_2\text{CHO}$ and (b) $\text{ZnTPP}(\text{TPA})_2\text{CHO}$.

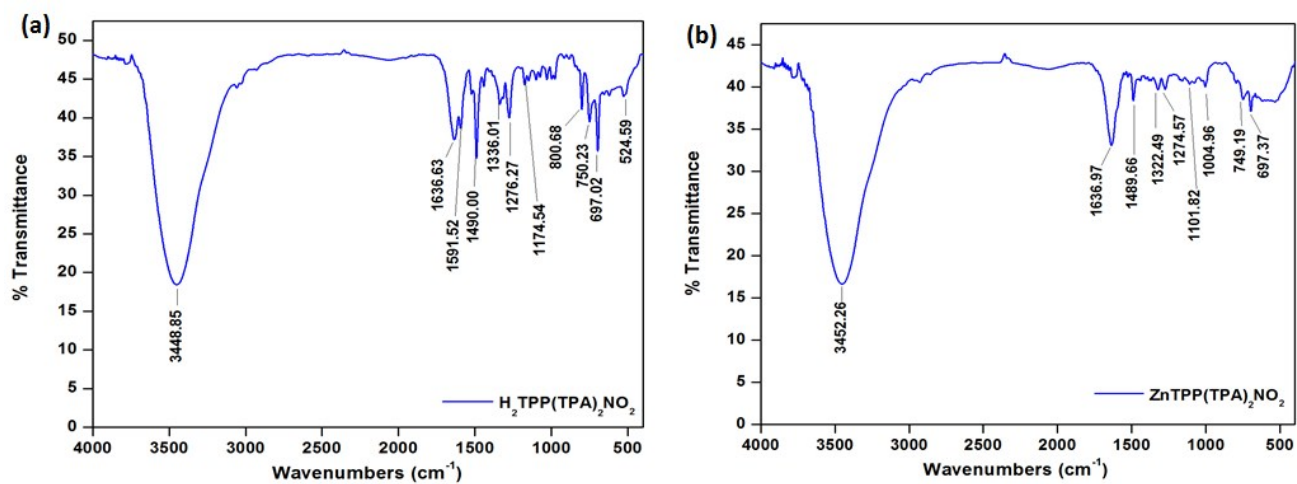


Figure S24. IR spectra of (a) $\text{H}_2\text{TPP}(\text{TPA})_2\text{NO}_2$ and (b) $\text{ZnTPP}(\text{TPA})_2\text{NO}_2$.

Table S1. Photophysical data of MTPP(TPA)₂NO₂ and MTPP(TPA)₂CHO (2H, Co^{II}, Ni^{II}, Cu^{II}, Zn^{II}) in CH₂Cl₂ at 298 K

Porphyrin	$\lambda_{\text{excitation, nm}}$	$\lambda_{\text{emission, nm}}$	Φ_f	FWHM	τ [ns]
H ₂ TPP(TPA) ₂ NO ₂	308(4.65), 438(5.13), 539(4.12), 690(3.83)	754	0.0019	40	0.52
H ₂ TPP(TPA) ₂ CHO	308(4.69), 438(5.24), 535(4.06), 576(3.88), 612(3.72), 678(3.61)	701	0.019	29	4.88
CoTPP(TPA) ₂ NO ₂	308(4.76), 432(5.01), 551(4.11), 597(4.10)			60	
CoTPP(TPA) ₂ CHO	308(4.70), 434(5.11), 555(4.16), 593(4.09)			45	
NiTPP(TPA) ₂ NO ₂	309(4.77), 438(5.11), 553(4.17), 602(4.14)			44	
NiTPP(TPA) ₂ CHO	309(4.76), 437(5.34), 554(4.11), 598(4.08)			32	
CuTPP(TPA) ₂ NO ₂	309(4.73), 432(5.17), 559(4.16), 606(4.10)			41	
CuTPP(TPA) ₂ CHO	309(4.81), 434(5.46), 559(4.25), 600(4.20)			25	
ZnTPP(TPA) ₂ NO ₂	309(4.85), 434(5.50), 560(4.47), 609(4.36)	698	0.003	37	0.58
ZnTPP(TPA) ₂ CHO	311(4.42), 434(5.16), 562(3.91), 603(3.83)	647	0.009	23	0.89

Table S2. Redox Potential Data of all Synthesized Porphyrins with Comparative Porphyrins containing 0.1 M TBAPF₆ with scan rate 0.1 Vs⁻¹ at 298K.

Porphyrin	Oxidation				ΔE (V)	Reduction			$M^{II/III}$	$M^{II/I}$
	I	II	III	IV		I	II	III		
H ₂ TPP	1.00	1.34			2.23	-1.23	-1.54			
H ₂ TPPBr ₂ NO ₂	1.11	1.21			1.87	-0.75	-0.82			
H ₂ TPP(TPA) ₂ NO ₂	1.02	1.62			1.85	-0.83	-0.98			
H ₂ TPPBr ₂ CHO	1.11	1.22			2.00	-0.89				
H ₂ TPP(TPA) ₂ CHO	1.10				2.05	-0.94				
CoTPP	1.06	1.31			2.44	-1.38			0.85	-0.86
CoTPPBr ₂ NO ₂	1.22	1.44			2.43	-1.20			0.92	-0.51
CoTPP(TPA) ₂ NO ₂	1.22	1.51			2.47	-1.25			0.96	-0.60
CoTPPBr ₂ CHO	1.26	1.41			2.28	-1.02	-1.49		0.95	-0.57
CoTPP(TPA) ₂ CHO	1.23	1.45			2.36	-1.13			0.98	-0.67
NiTPP	1.02	1.32			2.30	-1.28	-1.72			
NiTPPBr ₂ NO ₂	1.24				2.07	-0.83	-1.06			
NiTPP(TPA) ₂ NO ₂	0.95	1.23			1.82	-0.87	-1.14	-1.4		
NiTPPBr ₂ CHO	1.14				2.09	-0.95	-1.21			
NiTPP(TPA) ₂ CHO	1.03	1.26	1.5	2.0	2.09	-1.05	-1.05			
CuTPP	0.97	1.35			2.30	-1.33	-1.70			
CuTPPBr ₂ NO ₂	1.07	1.49			1.92	-0.85	-1.07			
CuTPP(TPA) ₂ NO ₂	1.00	1.50			1.89	-0.89	-1.14	-1.3		
CuTPPBr ₂ CHO	1.08	1.47			2.02	-0.94	-1.20			
CuTPP(TPA) ₂ CHO	1.06	1.47			2.03	-0.97	-1.34			
ZnTPP	0.84	1.15			2.20	-1.36	-1.77			
ZnTPPBr ₂ NO ₂	0.94	1.18			1.88	-0.93	-1.08			
ZnTPP(TPA) ₂ NO ₂	0.98	1.23			2.05	-1.07	-1.30			
ZnTPPBr ₂ CHO	0.97	1.22			2.02	-1.05	-1.20			
ZnTPP(TPA) ₂ CHO	1.04	1.55			2.16	-1.11	-1.36			

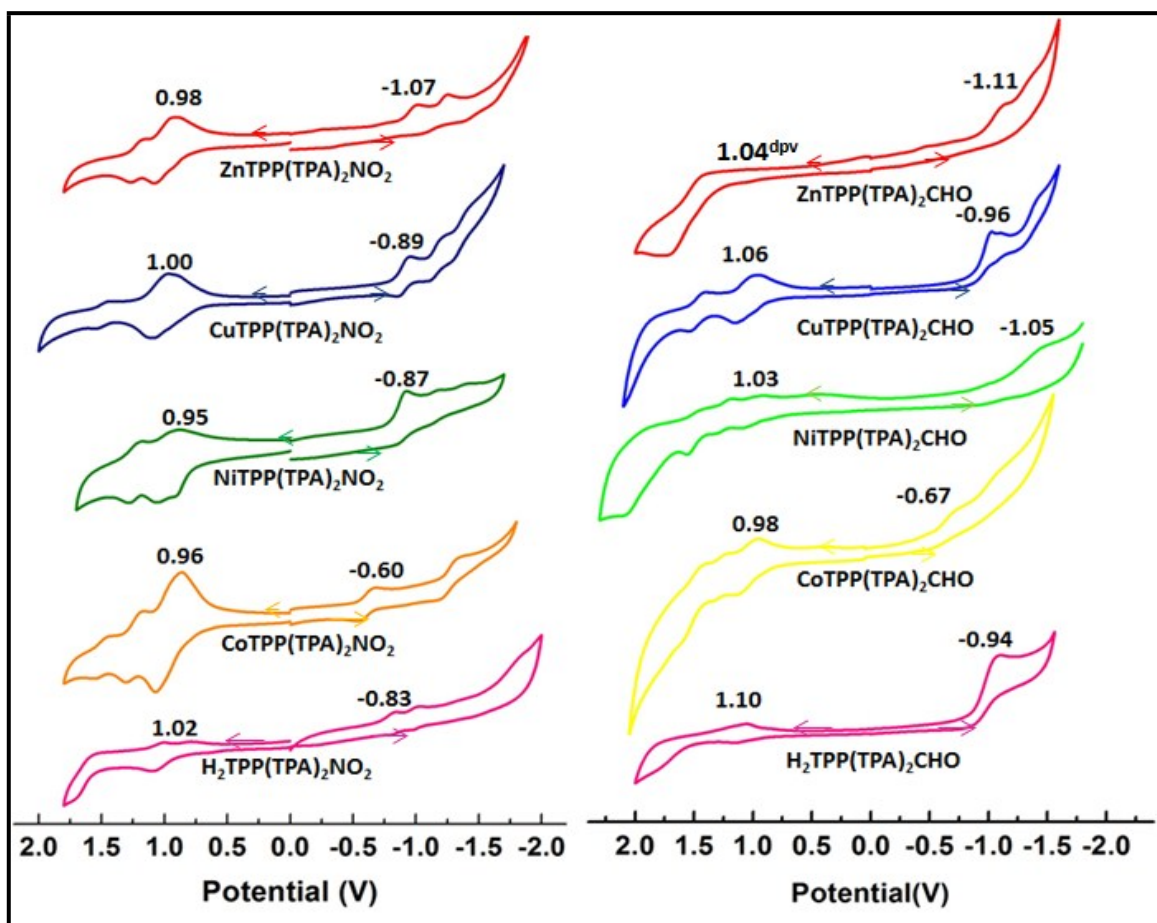


Figure S25. Cyclic Voltammograms of Porphyrins (a) MTPP(TPA)₂NO₂ and MTPP(TPA)₂CHO (M = 2H, Co(II), Cu(II), Ni(II), Zn(II)) and in CH₂Cl₂ with a Scan Rate of 0.1 V/s at 298 K.

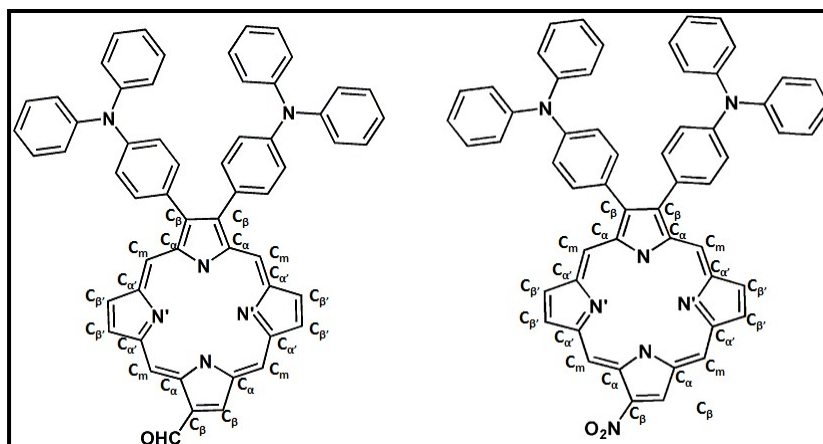


Table S3. Selected Average Bond Lengths (Å) and Bond Angles (°) for the B3LYP/LANL2DZ Optimized Geometry of $\text{H}_2\text{TPP}(\text{TPA})_2\text{NO}_2/\text{CHO}$ and $\text{ZnTPP}(\text{TPA})_2\text{NO}_2/\text{CHO}$.

	$\text{H}_2\text{TPPBr}_2\text{CHO}$	$\text{H}_2\text{TPP}(\text{TPA})_2\text{CHO}$	$\text{H}_2\text{TPPBr}_2\text{NO}_2$	$\text{H}_2\text{TPP}(\text{TPA})_2\text{NO}_2$	$\text{ZnTPP}(\text{TPA})_2\text{CHO}$	$\text{ZnTPP}(\text{TPA})_2\text{NO}_2$
Bond Length (Å)						
M-N					2.080	2.090
M-N'					2.040	2.050
N-C $_{\alpha}$	1.363	1.390	1.388	1.391	1.395	1.394
N'-C $_{\alpha'}$	1.375	1.390	1.392	1.390	1.396	1.395
C $_{\alpha}$ -C $_{\beta}$	1.462	1.450	1.460	1.440	1.462	1.459
C $_{\alpha'}$ -C $_{\beta'}$	1.433	1.471	1.394	1.472	1.458	1.457
C $_{\beta}$ -C $_{\beta'}$	1.363	1.399	1.375	1.394	1.390	1.386
C $_{\beta'}$ -C $_{\beta''}$	1.365	1.367	1.380	1.366	1.375	1.372
C $_{\alpha}$ -C $_{\text{m}}$	1.414	1.417	1.420	1.417	1.422	1.422
C $_{\alpha'}$ -C $_{\text{m}}$	1.403	1.419	1.413	1.418	1.417	1.416
ΔC_{β} (Å)	0.575	0.636	0.667	0.642	0.593	0.580
$\Delta 24$ (Å)	0.266	0.312	0.315	0.317	0.282	0.275
ΔMetal (Å)					0.011	0.033
Bond Angle (deg)						
N-C $_{\alpha}$ -C $_{\text{m}}$	124.52	124.97	124.70	125.12	124.57	124.71
N'-C $_{\alpha'}$ -C $_{\text{m}}$	127.19	126.37	127.00	126.33	126.40	126.26
N-C $_{\alpha}$ -C $_{\beta}$	109.91	106.33	109.20	110.28	108.91	108.81
N'-C $_{\alpha'}$ -C $_{\beta'}$	106.29	110.26	106.40	106.11	108.86	108.95
C $_{\beta}$ -C $_{\alpha}$ -C $_{\text{m}}$	125.47	123.33	125.90	123.34	126.41	126.45
C $_{\beta'}$ -C $_{\alpha'}$ -C $_{\text{m}}$	126.46	128.56	126.60	128.58	124.90	124.36
C $_{\alpha}$ -C $_{\text{m}}$ -C $_{\alpha'}$	124.73	124.13	123.80	123.95	124.13	124.70
C $_{\alpha}$ -C $_{\beta}$ -C $_{\beta'}$	106.46	106.80	107.00	106.78	107.09	106.96
C $_{\alpha'}$ -C $_{\beta'}$ -C $_{\beta''}$	108.26	107.96	108.20	108.12	107.42	107.45
C $_{\alpha}$ -N-C $_{\alpha}$	107.00	105.76	107.20	110.95	107.58	107.74
C $_{\alpha}$ -N'-C $_{\alpha'}$	110.83	111.11	110.60	110.73	107.30	107.22
M-N-C $_{\alpha}$					125.18	125.27
M-N'-C $_{\alpha'}$					125.49	125.36
N-M-N					177.54	179.75
N'-M-N'					178.70	177.23

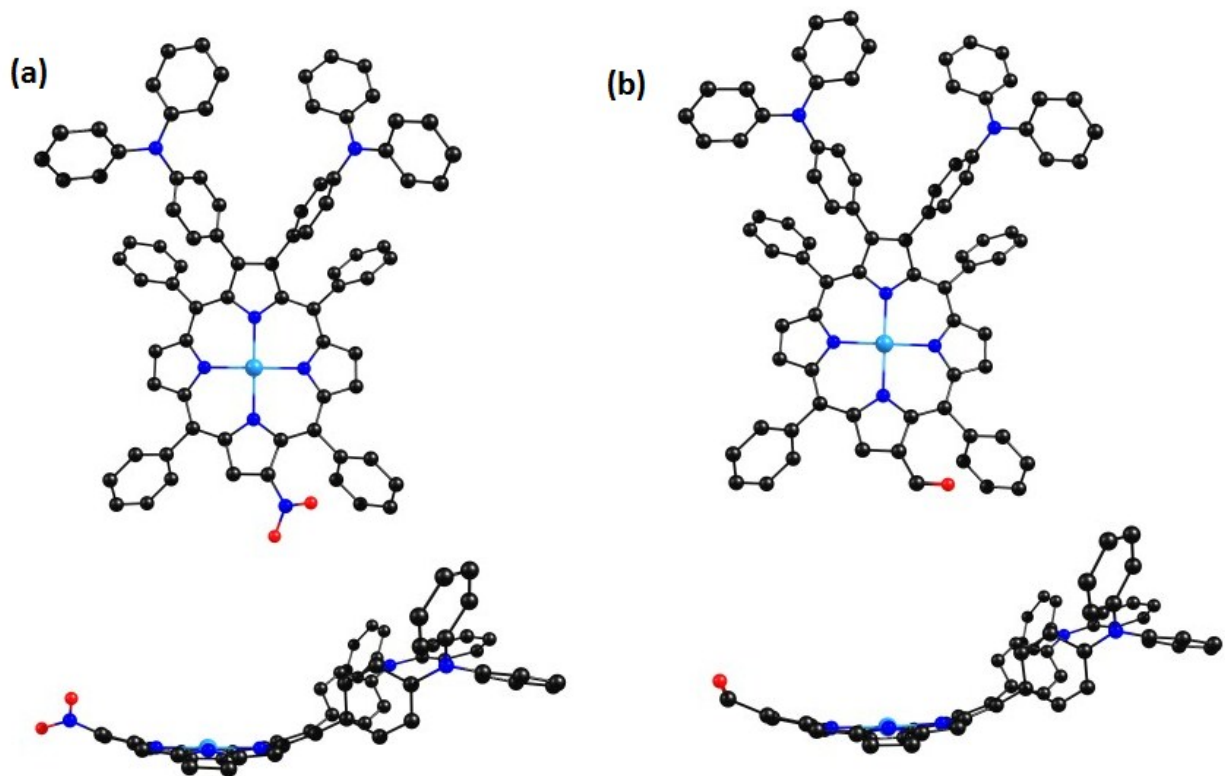


Figure S26. Optimized gas phase geometries of (a) ZnTPP(TPA)₂NO₂ and (b) ZnTPP(TPA)₂CHO.

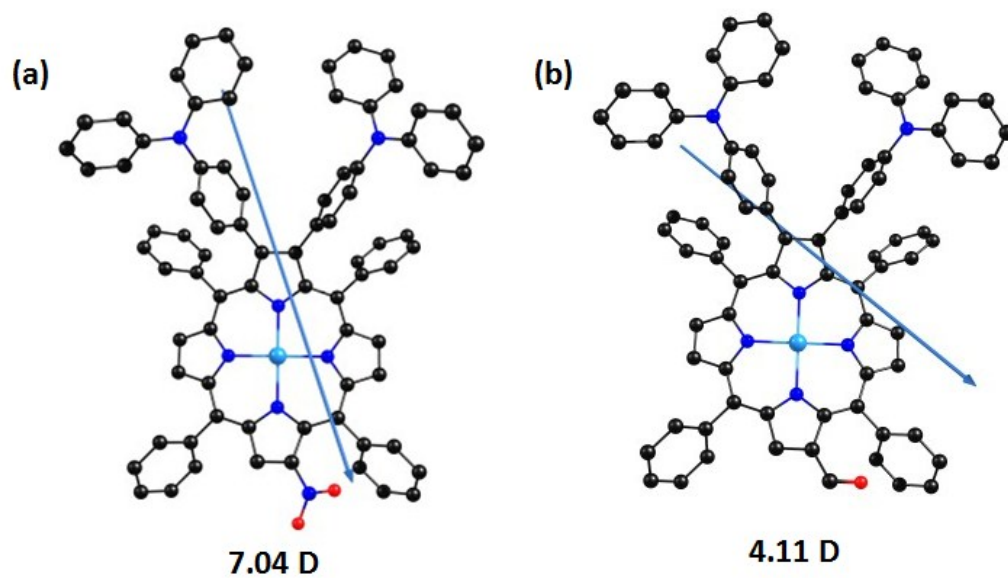


Figure S27. Theoretically calculated dipole moment direction of (a) ZnTPP(TPA)₂NO₂ and (b) ZnTPP(TPA)₂CHO.

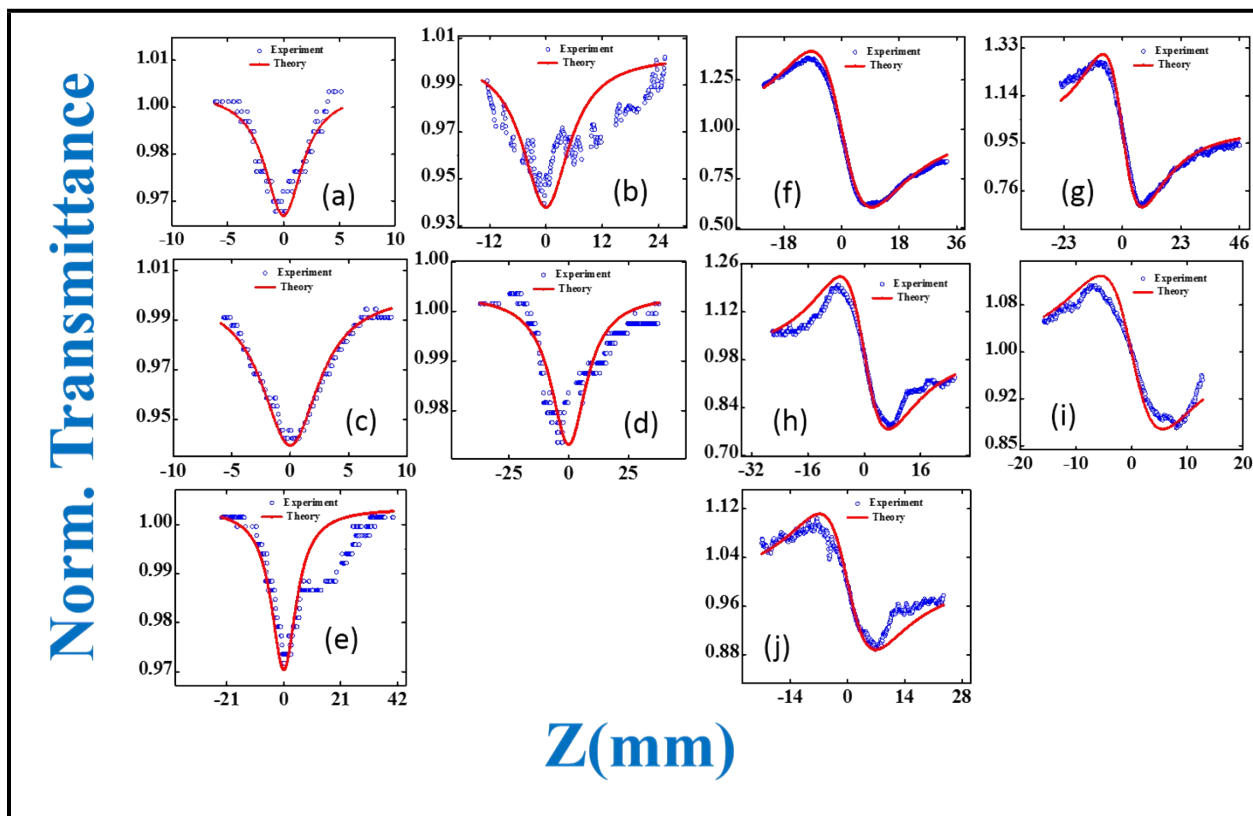


Figure S28: Experimental and theoretically fitted Z-scan data for sample $\text{Zn(TPA)}_2\text{NO}_2$ in OA mode at (a) 680 nm (b) 700 nm (c) 750 nm (d) 800 nm (e) 850 nm and CA mode at (f) 680 nm (g) 700 nm (h) 750 nm (i) 800 nm (j) 850 nm. Open symbols are the experimental data points while the solid lines are the theoretical fits.

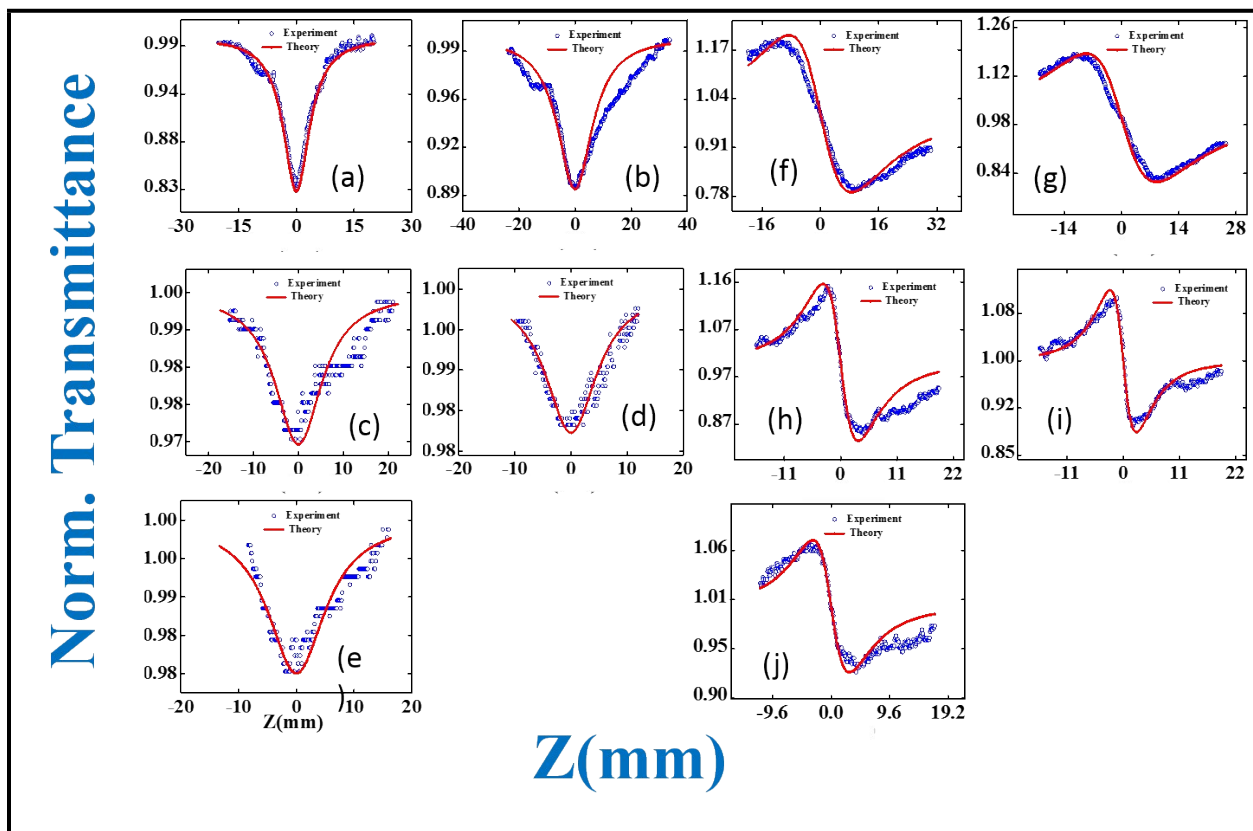


Figure S29: Experimental and theoretically fitted Z-scan data for sample $\text{Zn}(\text{TPA})_2\text{CHO}$ in OA mode at (a) 680 nm (b) 700 nm (c) 750 nm (d) 800 nm (e) 850 nm and CA mode at (f) 680 nm (g) 700 nm (h) 750 nm (i) 800 nm (j) 850 nm. Open symbols are the experimental data points while the solid lines are the theoretical fits.