

Electronic Supporting Information

Measurements of Rhenium Isotopic Composition In Low-Abundance Samples

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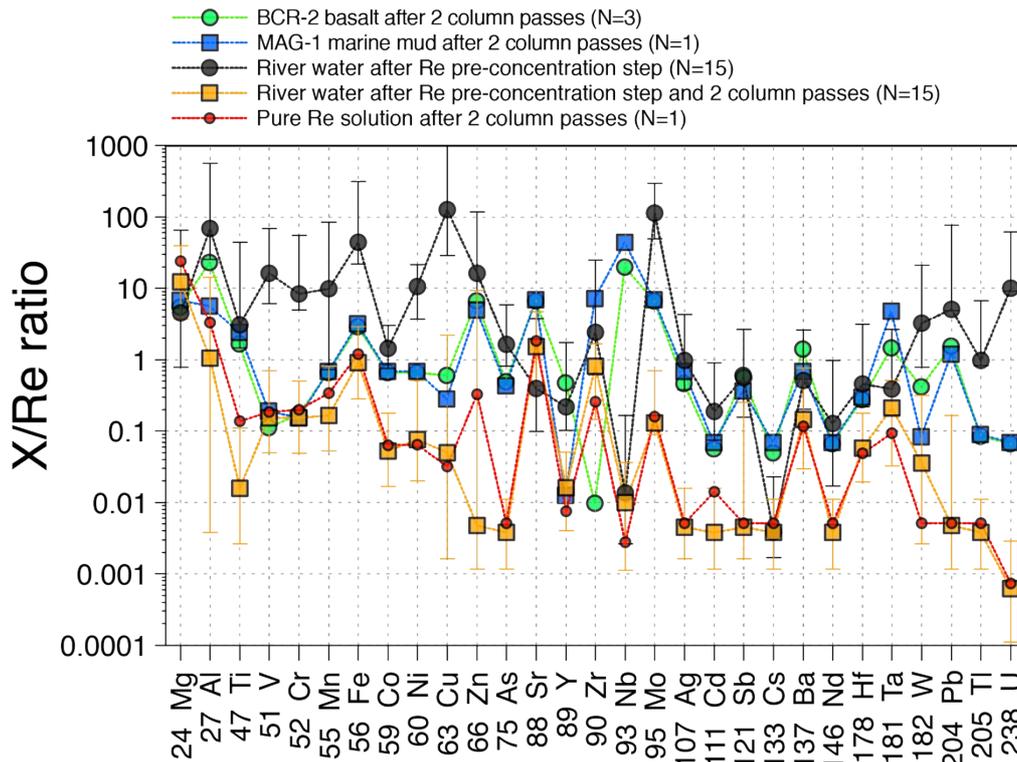
Cup configuration:

Cup	L4	L3	L2	L1	C	H1	H2	H3	H4
Main isotope	-	¹⁸³ W	¹⁸⁴ W	¹⁸⁵ Re	¹⁸⁶ W	¹⁸⁷ Re	¹⁸⁸ Os	¹⁸⁹ Os	-
Collector	-	10 ¹¹	10 ¹¹	10 ¹³	10 ¹¹	10 ¹³	10 ¹¹	10 ¹¹	-

Figure S1: Cup configuration used for simultaneous measurement of Re and W isotopes

Tests to determine the influence of residual inorganic matrix on Re isotopes:

To determine the efficiency of the separation method for removing inorganic matrix, we measured the ratio between chemical elements and Re for different types of materials (chondrite, basalt, shale, river water) having distinct Re concentration (Figure S2). For basalt SRM BCR-2 and shale SRM MAG-1, we can see that after 2 column passes, the ratio X/Re (in g/g) for most chemical elements is < 50. The highest residual concentration elements being Al, Nb, Zr and Mo. For BCR-2, the second column chemistry lead to significant further removal of Al, Mg, Fe and Tl while the X/Re ratio for most of other elements does not significantly changed between the first and second column. For river waters, the X/Re ratio after two column separation is lower than 1 for most of the elements showing that inorganic matrix residue is even significant than in the rock samples.

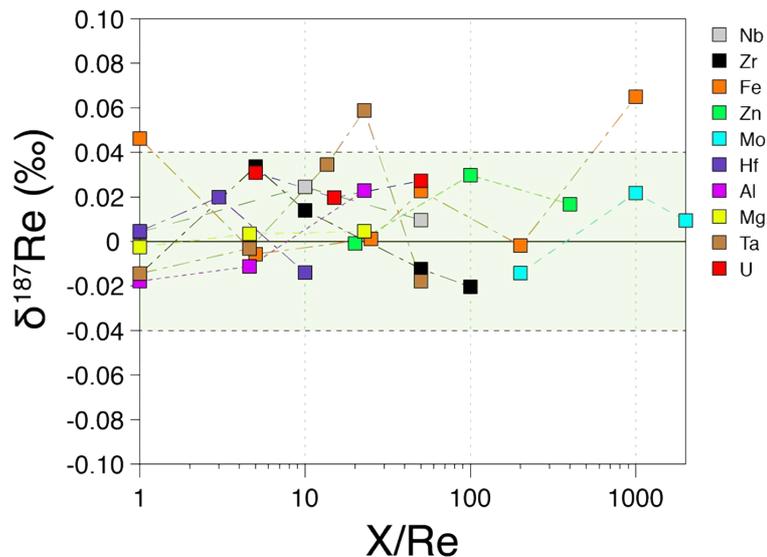


33

34 **Figure S2.** X/Re ratio values (with X any chemical element, in g/g) measured in Re fraction after column chemistry
 35 for different type of materials (Basalt, Marine mud, river water, pure Re solution). Are represented the average
 36 value for several independent samples or digestions (except for MAG-1, only 1 measurement) with the error bar
 37 corresponding to the SD of all the measurements.

38 Using these results, we performed a series of tests where a Re standard solution was doped with
 39 various elements (Al, Fe, Mg, Nb, Zr, Ta, Mo, Zn, U, Hf) at different concentrations, covering a range
 40 of value of X/Re (where X is the doped element) corresponding to the X/Re ratios measured in solutions
 41 after 2 column passes. Results are presented in figure SXX and show no significant influence of
 42 inorganic elements on the measured Re isotopic composition.

43



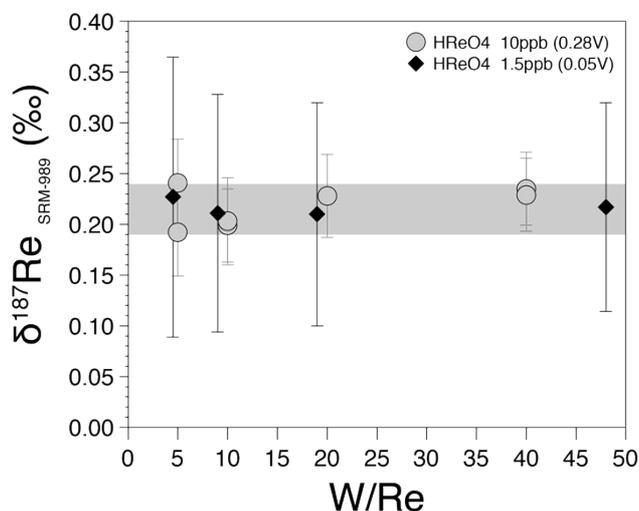
44

45 **Figure S3.** $\delta^{187}\text{Re}$ (relative to NIST-SRM989) as a function of the ratio between chemical elements (Nb, Zr, Fe,
 46 Zn, Mo, Hf, Al, Mg, Ta and U) and Re. All measurements were performed for beam intensity of 0.2V on ^{187}Re
 47 (corresponding to 5 ng of Re).

48 **Influence of W/Re ratios:**

49 The influence of the amount of tungsten external standard added to the sample for isotope measurement
50 (W/Re ratio) was tested by measuring HReO4 standard doped with various amount of W (the bracketing
51 standard have the same W/Re ratio) for two different Re concentrations, 10 and 1.5 ppb Re (Figure
52 SXX). We observe no influence of the W/Re ratio for values ranging from 5 to 40.

53



54

55 **Figure S4.** $\delta^{187}\text{Re}$ (relative to NIST-SRM989) as a function of the W/Re used for measurement (W standard added
56 after column chemistry, same W/Re between bracketing standard SRM 989 and HReO4 measured standard. The
57 Grey shaded area corresponds to the average $\delta^{187}\text{Re}$ of HReO4.

58

59 **Assessment of data quality:**

60 One way to assess data quality is to compare the $\delta^{187}\text{Re}$ value calculated with the Standard-sample
61 bracketing method ($\delta^{187}\text{Re}^{\text{SSB}}$) with that calculated by both W-corrected and SSB methods ($\delta^{187}\text{Re}$).
62 External normalization to W is used to correct for both the changes in the plasma conditions and matrix
63 effects. In general, we find that the external error for the SSB method of mass bias correction is greater
64 than for the combined EN-SSB method and relatively independent of the Re concentration but quite
65 sensitive to instrumental stability (plasma, nebuliser uptake stability, spray chamber) during any given
66 analytical session. Thus, correcting for changes of the mass bias using combination of W-normalization
67 and SSB gives more accurate results.

68 In the case of relatively stable or constant mass bias (and in the absence of matrix effect) the $\delta^{187}\text{Re}$
69 determined by $\delta^{187}\text{Re}^{\text{SSB}}$ and $\delta^{187}\text{Re}$ should be equal ($\delta^{187}\text{Re}^{\text{SSB}} - \delta^{187}\text{Re} = 0$). In the case of a non-stable
70 mass-bias, which happened during some of our sequences, we measure a maximum ($\delta^{187}\text{Re}^{\text{SSB}} - \delta^{187}\text{Re}$)
71 of +0.27‰ and a minimum of -0.14‰, showing that plasma instabilities can lead to inaccurate $\delta^{187}\text{Re}^{\text{SSB}}$
72 but correct $\delta^{187}\text{Re}$ values. For geological samples, two BCR-2 measurements and one soil doped with
73 Re of known composition yielded $\delta^{187}\text{Re}^{\text{SSB}} - \delta^{187}\text{Re} = +0.30\text{‰}$ (See supplementary tables XX), but
74 correct $\delta^{187}\text{Re}$ values. These results actually illustrate the importance of applying W correction as this
75 corrects for small variations in mass bias, presumably resulting from matrix effects, that SSB alone
76 would not correct for. Nevertheless, as a precautionary measure we reject any measurements for which
77 ($\delta^{187}\text{Re}^{\text{SSB}} - \delta^{187}\text{Re}$) exceeds 0.30‰.

78

79 **Preparation of Re-free matrix standards for Re isotope accuracy tests:**

80 To assess the accuracy of our Re isotope measurements, we prepared two Re-free typical geological
81 matrix by collecting volume cuts before and after Re elution peaks for different materials. These cuts
82 were then re-combined, evaporated and then treated like a “normal” sample, i.e. passed three times
83 through column chemistry. After three column passes, a small aliquot (2% of sample mass) was taken
84 to check the absence of Re. Then, a known amount of ^{187}Re rich (i.e. very different $\delta^{187}\text{Re}$ relative to

85 current range of measured $\delta^{187}\text{Re}$ for geological materials) in-house standard DURH-Re-1 was added to
86 the “matrix” sample and the $\delta^{187}\text{Re}$ was measured.

87

88 Matrix 1: a mass of 4.5 g of different river sediment samples (original Re concentration of ~0.2 ppb)
89 containing < 1% TOC, doped with 12 ng of DURH-Re-1 standard.

90

91 Matrix 2: a mass of 0.4 g of Posidonia Shale sample from NW Germany (original Re concentration of
92 ~45 ppb) containing > 1% TOC, doped with 2.5 ng of DURH-Re-1 standard.

93

94 One limitation of the previous test is that it does not account for the fact that some elements or organic
95 molecules may have identical behaviour to Re during our column chemistry method (i.e. elution from
96 the column only with 12 mL of 4M HNO_3) and create matrix effect which would bias $\delta^{187}\text{Re}$
97 measurements. To test this, we prepared an additional set of matrix materials corresponding to samples
98 “naturally” containing low Re concentration and doped with DURH-Re-1:

99

100 Matrix 3: a mass of 1.5 g of Granite SRM “G2” (Re concentration of 0.011 ppb) doped with 3 ng of of
101 DURH-Re-1 standard.

102

103 Matrix 4: a mass of 1.5 g of New Zealand soil sample NZ14-54 (Re concentration of 0.036 ppb) and
104 containing Total Organic Carbon (TOC) of 1.3%, with 3 ng of of DURH-Re-1 standard.

105

106 Matrix 5: a mass of 1.5 g of New Zealand soil sample NZ14-62 (Re concentration of 0.021 ppb) and
107 containing Total Organic Carbon (TOC) of 7.5%, with 3 ng of of DURH-Re-1 standard.

108

109 Matrix 6: a mass of 2.5 g of New Zealand soil sample NZ14-54 (Re concentration of 0.036 ppb) and
110 containing Total Organic Carbon (TOC) of 1.3%, doped with 2.5 ng of of DURH-Re-1 standard.

111

112 **Samples and materials used for column chemistry tests:**

113 The samples used for testing the effect of the number of column passes on the measured Re isotopic
114 composition (Fig. 5) are:

115 “River UK” = River Wear sampled in Durham

116 “River 1, 2, 3 Canada” = Rivers from the Mackenzie River basin (Canada)

117 “River France” = Small river in the French Alps from the Rhone river basin (France)

118 “Black shale” = Posidonia Shale sample from NW Germany

119 “Chondrite” = Allende Chondrite

120 “Basalt” = BCR-2 basalt reference material

121

122 **Supplementary tables:**

123 We provide two spreadsheets (one with simplified tables, the other one with raw data) with tables and
124 supplementary tables:

125 - **Table S1**: Tests with different W/Re of the bracketing standard relative to the sample (here the
126 bracketing standard and the “sample” are the same solution).

127 - **Table S2**: Tests of Re isotope fractionation during incomplete elution through column chemistry for
128 the column containing 1 mL AG1-X8 resin and for large column containing 3 mL AG1-X8 resin used
129 for pre-concentration of Re from water sample on the field.

130 - **Table S3**: Doping tests with different elements and various X/Re ratio (the doped standard and the
131 bracketing standard are the same solution), data corresponding to the figure SXX.

132 - **Table S4**: Results of standard reference materials and matrix test material measured for Re isotopes.
133 Each line corresponds to a single complete measurement (from digestion to Re isotope analysis).

134 - **Table S5**: Influence of the W/Re of both bracketing standard and measured solution on the measured
135 Re isotopic composition (for two concentration, 10 ppb and 1.5 ppb)

136 - **Table S6**: Results for Re concentration measurements for a wide range of SRMs

