

Supporting information for

**A novel NIR fluorescent probe for double-site and ratiometric
detection of SO₂ derivatives and its application**

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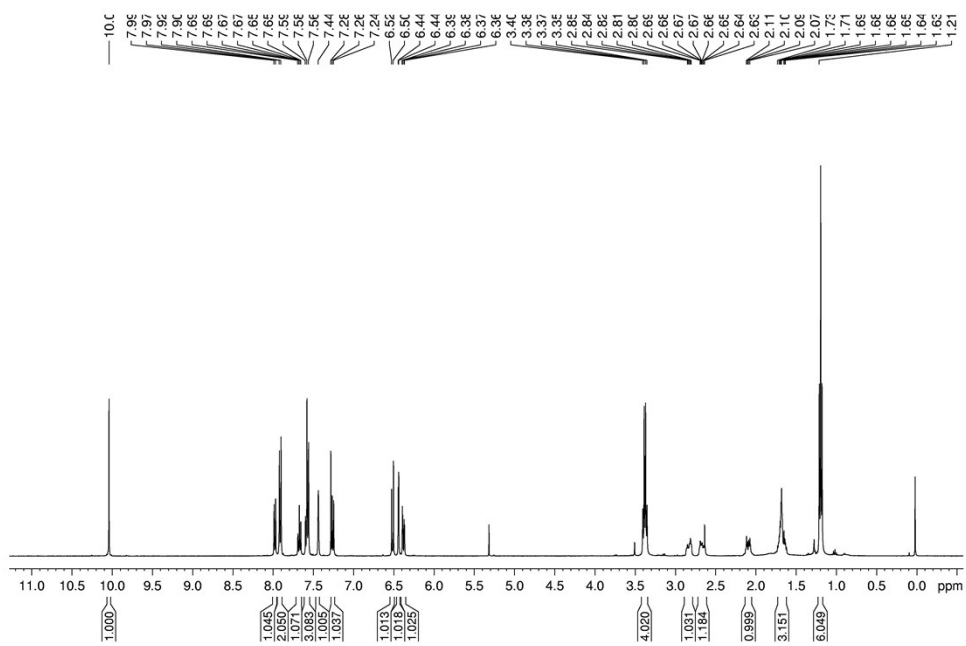


Fig. S1 ^1H NMR chart of probe **Q5** (CDCl_3 , 400 MHz).

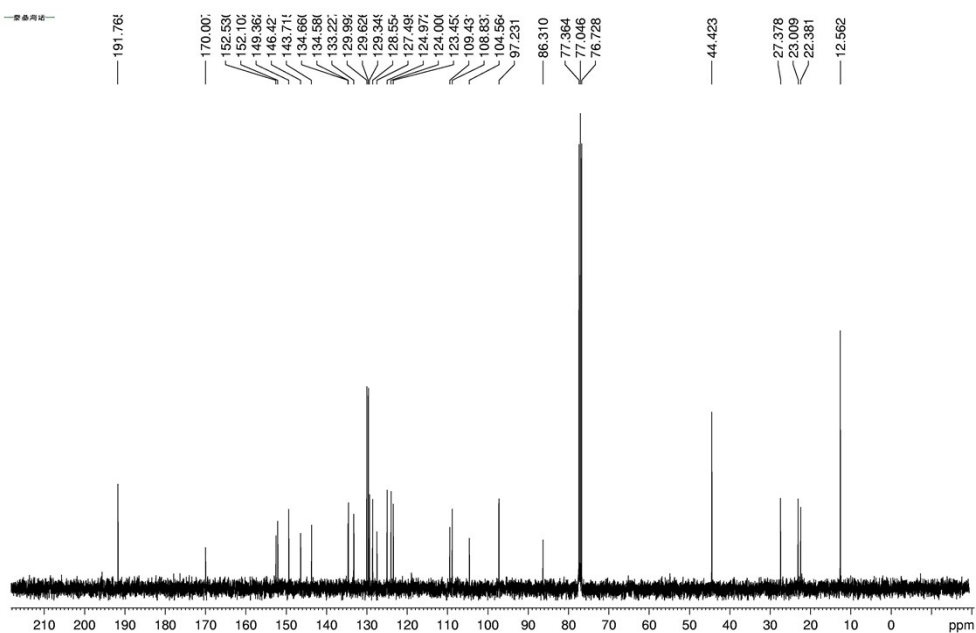


Fig. S2 ^{13}C NMR chart of probe **Q5** (CDCl_3 , 100 MHz)

Sample Name	L5	Position	P1-A1	Instrument Name	Instrument 1	User Name	Agilent FSE
Inj Vol	0.5	InjPosition		SampleType	Sample	IRM Calibration Status	Some Ions Missed
Data Filename	L5-0.2ul.d	ACQ Method	test.m	Comment		Acquired Time	7/26/17 Wed 10:34:08

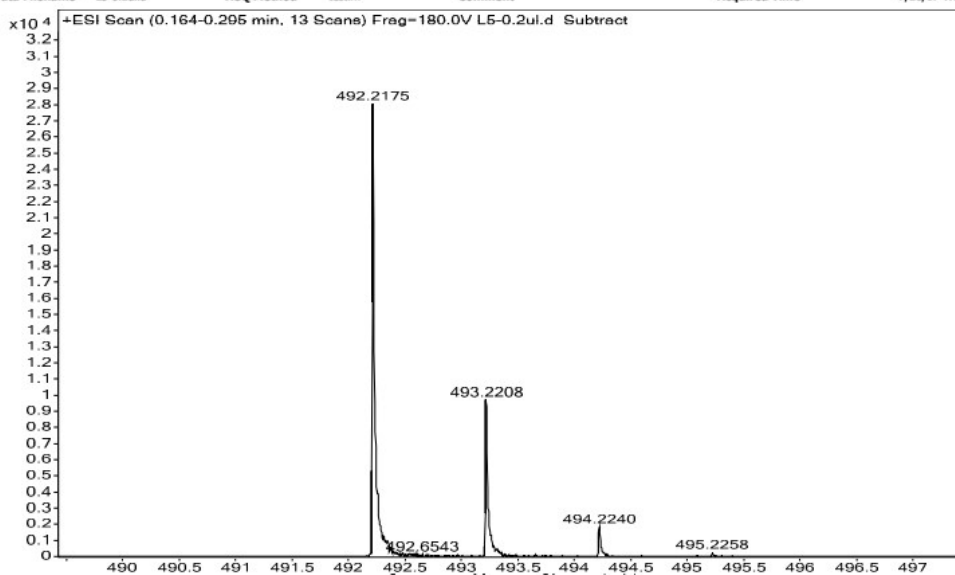


Fig. S3 ESI-HRMS spectrum of probe **Q5**.

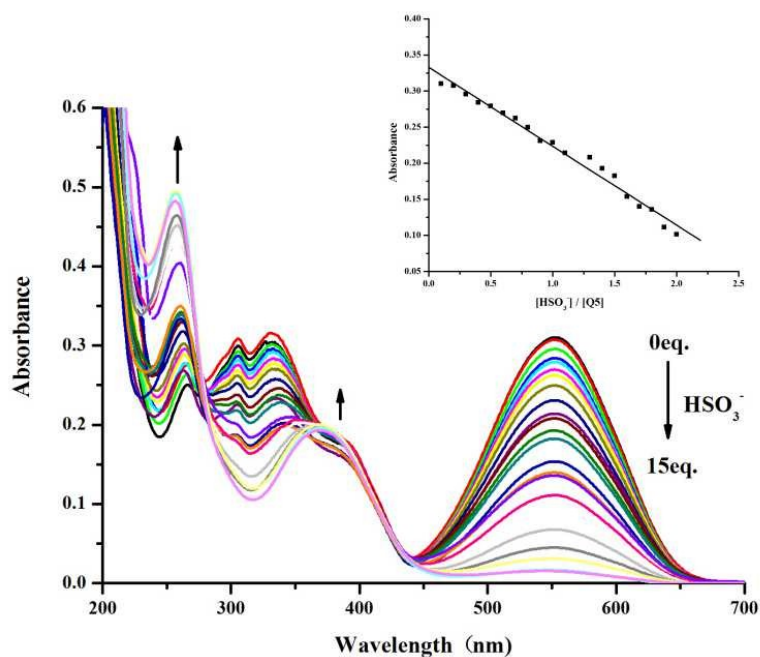


Fig. S4. Absorption spectra of **Q5** (10 μ M) with gradual addition of various amounts of HSO_3^- (0-15 eq.) in MeOH/PBS buffer (3/7, v/v, 10 mM, pH = 7.40) solution. Inset shows linear relationship between the UV peak of the probe at 550 nm and HSO_3^- concentration.

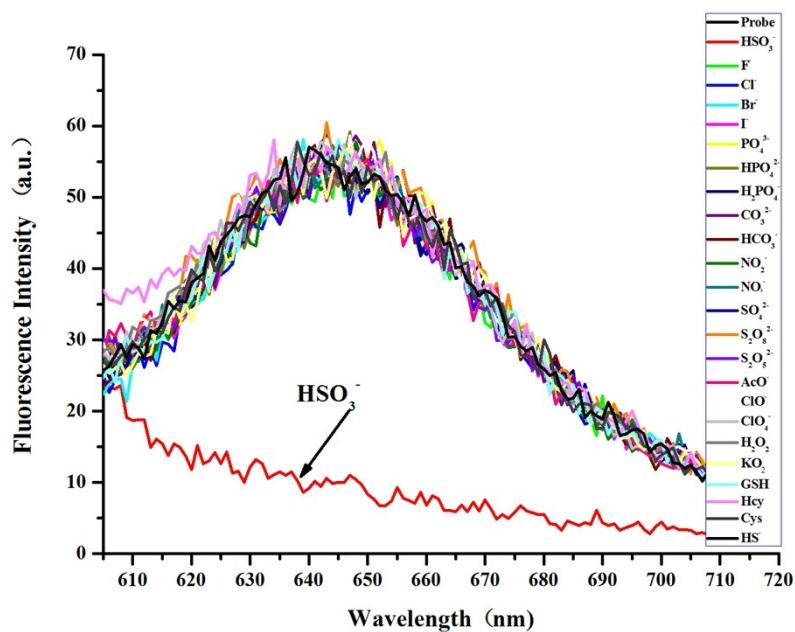
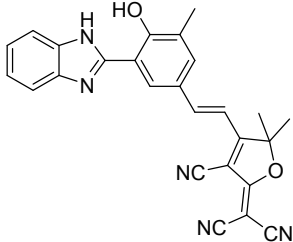
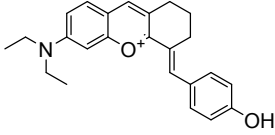
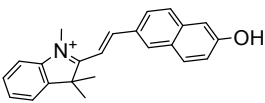
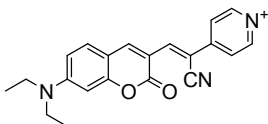
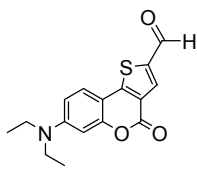
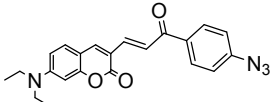
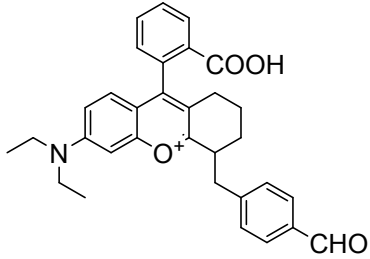


Fig. S5 Fluorescence intensity of probe **Q5** (10 μ M) in the presence of 10 equiv different analytes in MeOH/PBS buffer (3/7, v/v, 10 mM, pH = 7.40) solution. ($\lambda_{\text{ex}} = 580$ nm).

Table S1 Reported fluorescent probes for the detection of HSO_3^- .

Ref	Probe structures	$\lambda_{\text{ex}} / \lambda_{\text{em}}$	LOD (M)	Double-site
1		520 nm/ 580 nm	7×10^{-8}	NO
2		377 nm/ 563 nm	3.3×10^{-7}	NO
3		450nm/ 483 nm/633nm	1.7×10^{-8}	NO
4		350nm/ 428/ 508nm	7×10^{-7}	NO

5		400nm/ 483 nm	8.2×10^{-7}	NO
6		570nm/ 650nm	1.2×10^{-7}	NO
7		405nm/ 550nm	8.5×10^{-7}	NO
8		550nm/ 630 nm	2.8×10^{-6}	NO
9		385 nm/ 475 nm	2.3×10^{-7}	NO
10		410 nm/ 460/ 590 nm	1.0×10^{-7}	NO
Our Work		410 nm/ 485/ 650 nm	8.9×10^{-8}	YES

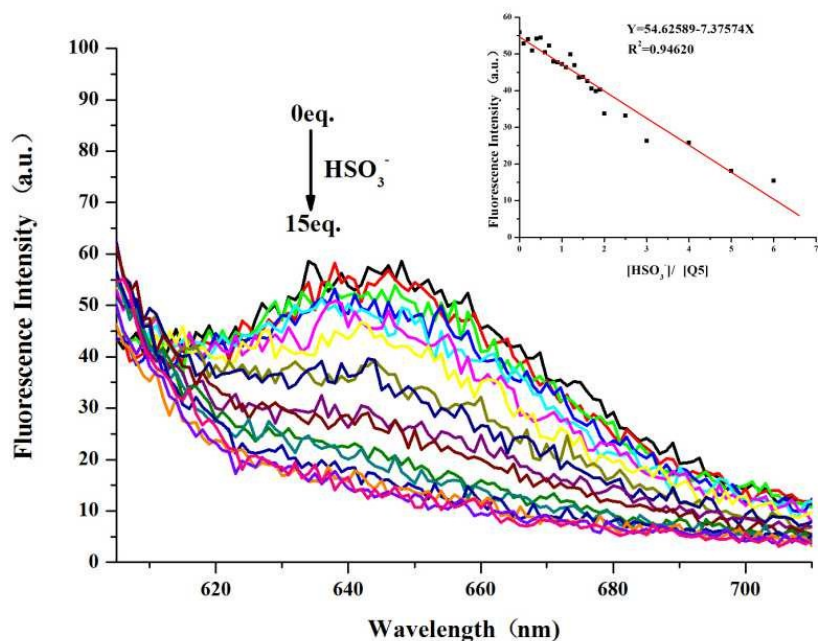


Fig. S6 Fluorescence emission spectra of compound **Q5** (10 μM) in the presence of different concentrations of HSO_3^- (0-15 equiv) in MeOH/PBS buffer (3/7, v/v, 10 mM, pH = 7.40) solution. Inset shows the linear responses with HSO_3^- concentrations ($\lambda_{\text{ex}} = 580 \text{ nm}$).

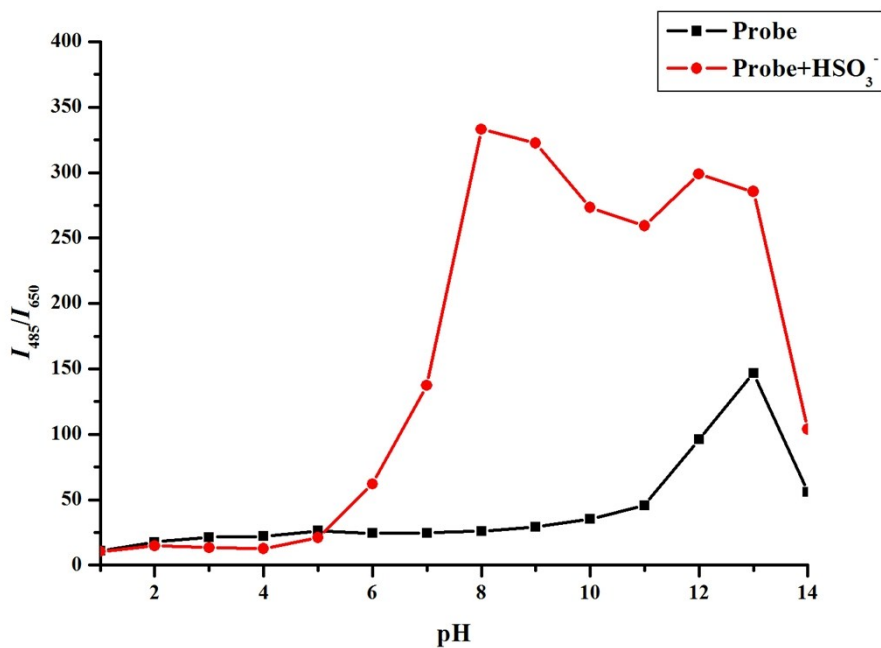


Fig. S7 The ratiometric fluorescence responses (F_{485}/F_{650}) of free **Q5** (10 μM) and in the presence of 10 eq. HSO_3^- in MeOH/PBS buffer (3/7, v/v, 10 mM) solution with different pH conditions ($\lambda_{\text{ex}} = 410 \text{ nm}$, slit = 10 nm).

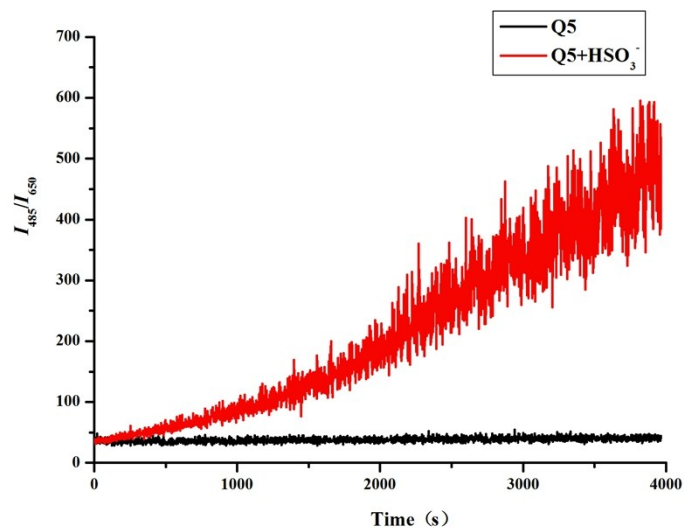


Fig. S8 Kinetics of fluorescence responses (F_{485}/F_{650}) of free **Q5** (10 μM) and after the addition of HSO_3^- (100 μM) in PBS (pH = 7.40, 10 mM, containing 30% MeOH). ($\lambda_{\text{ex}} = 410 \text{ nm}$, slit = 10 nm).

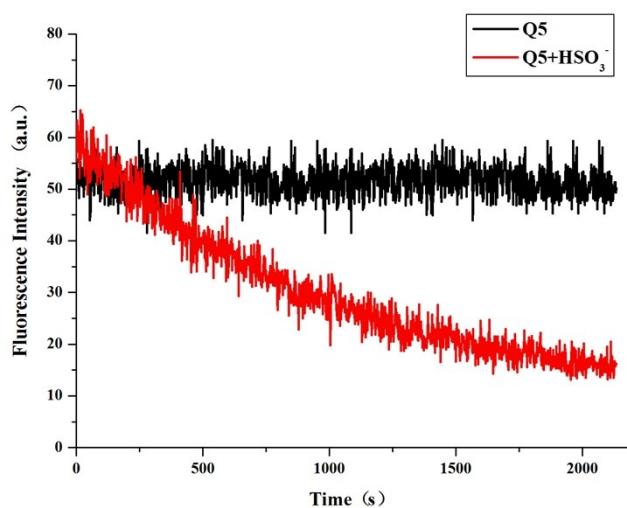


Fig. S9 Kinetics of the fluorescence responses (F_{650}) of **Q5** (10 μM) after the addition of HSO_3^- (100 μM) in PBS (pH = 7.40, 10 mM, containing 30% MeOH). ($\lambda_{\text{ex}} = 580 \text{ nm}$, slit = 10 nm).

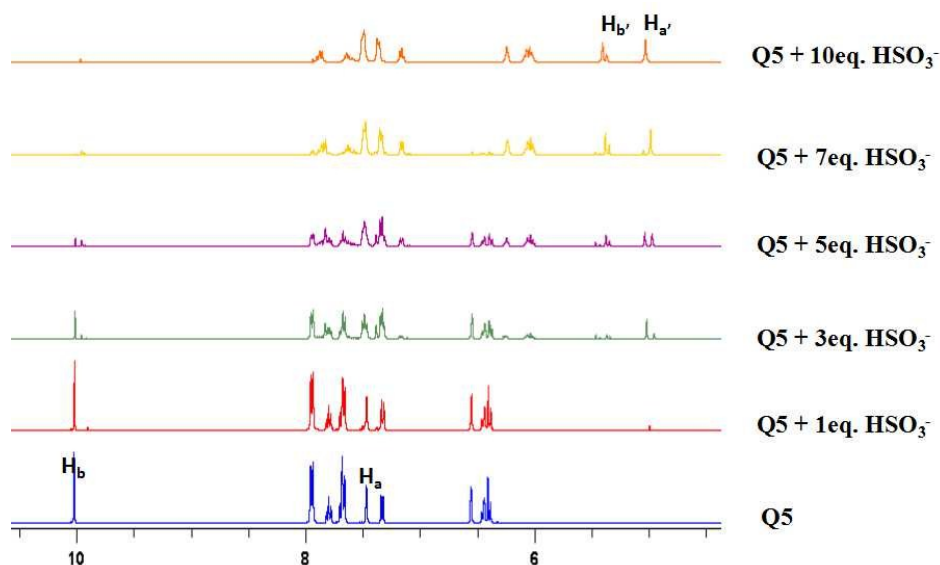


Fig. S10 The stack ^1H NMR spectrum of the mixture of probe **Q5** with different concentrations of HSO_3^- (0-10 equiv) in DMSO-d_6 .

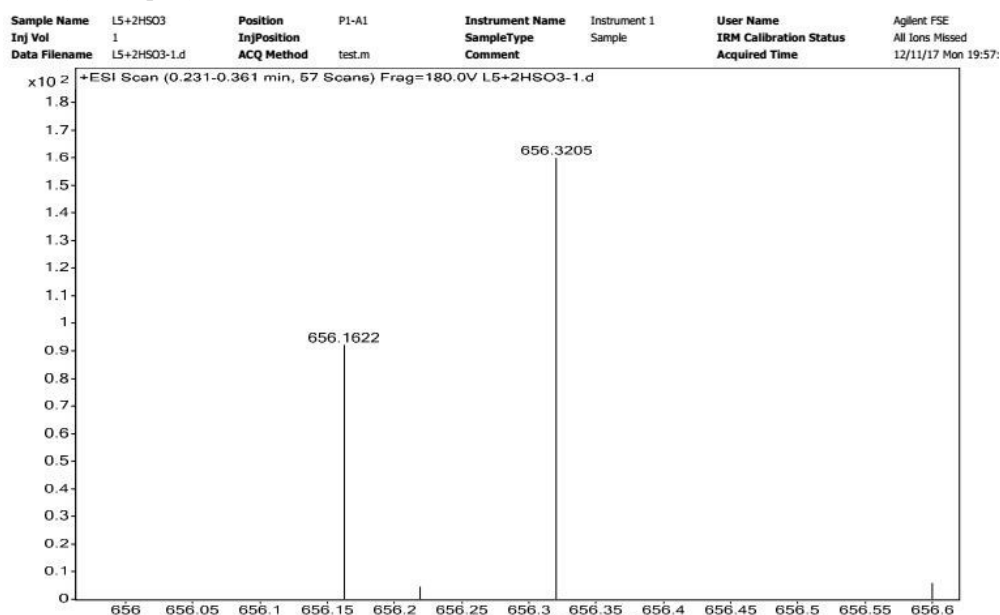


Fig. S11 ESI-HRMS spectrum of **Q5**.

Reference

1. D. Zhang, W. Liu and K. Chen, *RSC Adv.*, 2016, **6**, 103905–103909.
2. X. Liu, Q. Yang, W. Chen, L. Mo and J. Kang, *Org. Biomol. Chem.*, 2015, **13**, 8663–8668.
3. W. Liu, D. Zhang, B. Ni, J. Li, H. Weng and Y. Ye, *Sens. Actuators, B*, 2019, **284**, 330-336.
4. L. Zhang, Z. Wang, and X. Cao, *Sensors. Actuators, B*, 2016, **236**, 741–748.
5. T. T. Niu, T. Yu, G. X. Yin and H.T. Li, *Analyst*, 2019, **144**, 1546–1554.
6. W. J. Zhang, F. J. Huo and Y. B. Zhang, *Sensors. Actuators, B*, 2019, **297**, 126747-126752.
7. C. C. Gao, Y. Tian, R. B. Zhang, J. Jing and X. L. Zhang, *New J. Chem.*, 2019, **43**, 5255–15259.
8. P. Jana, N. Patel, V. Soppin, and S. Kanvah, *New J. Chem.*, 2019, **43**, 584–592.
9. L. L. Yang, M. F. Liu, K. J. Sheng and X. L. Li, *New J. Chem.*, 2019, **43**, 4188–4195.
10. H. Tian, J. Qian, Q. Sun, C. Jiang, R. Zhang and W. Zhang, *Analyst.*, 2014, **139**, 3373–3377.