## MoS<sub>2</sub> nanoflowers encapsulated in carbon nanofibers containing

## amorphous SnO<sub>2</sub> as an anode for lithium-ion batteries

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Fig. S1. TG curve of CNF precursor films under  $N_2$  atmosphere.



Fig.S2. (a-b) The SEM images of MoS<sub>2</sub> nanoflowers at different magnification.



Fig.S3. (a-b) The SEM images of MoS<sub>2</sub>@CNF composites at different magnifications; mapping of Mo (c), S (d), C (e) and N(f) elements.



Fig.S4. (a-b) The SEM images of SnO<sub>2</sub>@CNF composites at different magnifications; mapping of Sn (c), O (d) , C (e) and N(f) elements.



Fig.S5. XRD patterns of the MoS<sub>2</sub> nanoflowers.



Fig. S6. (a) XPS survey scan spectrum and core level spectra of (b) Mo 3d, (c) S 2p, (d) C 1s and (e) N 1s for MoS<sub>2</sub>@CNF.



Fig. S7. (a) XPS survey scan spectrum and core level spectra of (b) Sn 3d, (c) O 1s, (d) C 1s and (e) N 1s for MoS<sub>2</sub>-SnO<sub>2</sub>@CNF.



Fig.S8. The first three consecutive CV curves of  $MoS_2$  @CNF (a) and  $SnO_2$ @CNF (b).



Fig. S9. Discharge–charge curves of the MoS<sub>2</sub>@CNF (a) and SnO<sub>2</sub>@CNF (b) electrode at current density of 200 mA g<sup>-1</sup>.



Fig. S10. SEM image of SnO\_2@CNF electrode after the 430th cycle at 2000 mA g^-1.



Fig. S11. The equivalent circuit of the sample.



Fig.S12. (a) The MoS<sub>2</sub>-SnO<sub>2</sub>@CNF under bending states; (b) photograph of a flexible battery and a LED lightened by the flexible battery

under (c) flatting and (d) bending (180°) states.

	Reversible capacity (mAhg <sup>-1</sup> )/Cycle number			_	
Materials	Current	Current	Current	Current	Ref.
	density	density	density 500mA	density	
	100mA g <sup>-1</sup>	200mA g <sup>-1</sup>	g-1	2000mA g <sup>-1</sup>	
SnO <sub>2</sub> @C@VO <sub>2</sub>	765/100th		424/600th		1
SnO <sub>2-x</sub> :RGO		950/100th		700/100th	2
MoC-N-C				675/500th	3
SnS/C		648/500th	548/500th		4
SnS/MoS <sub>2</sub> –C		989/60th		718/700th	5
NDG/MoS <sub>2</sub> /NDG	750/100th				6
MoS <sub>2</sub> /NC		904/100th		534/400th	7
MoS <sub>2</sub> @EPF			854/200th	531/200th	8
SnO₂/GA	700/80th	512/10th			9
Sn@SnO <sub>x</sub> @MoS <sub>2</sub> @C			791/100th	530/800th	10
MoS <sub>2</sub> -SnO <sub>2</sub> @CNF		983/100th		710/800th	This work

 $\label{eq:stable_stab$ 

Table. S2.  $R_{\rm ct}$ ,  $\sigma$  and  $D_{\rm Li^+}$  values determined from the EIS for all the electrodes

	$R_{\rm ct}$ (Ω)	$\sigma (\Omega  \mathrm{cm}^2  \mathrm{s}^{-0.5})$	$D_{\text{Li+}}$ (cm <sup>2</sup> s <sup>-1</sup> )
MoS <sub>2</sub> -SnO <sub>2</sub> @CNF-1	182.6	56.8	4.9×10 <sup>-15</sup>
MoS <sub>2</sub> -SnO <sub>2</sub> @CNF-2	147.6	43.8	8.2×10 <sup>-15</sup>
MoS <sub>2</sub> -SnO <sub>2</sub> @CNF-3	197.2	69.6	3.3×10 <sup>-15</sup>
MoS₂@CNF	266.1	82.3	2.3×10 <sup>-15</sup>
SnO₂@CNF	254.8	79.4	2.5×10 <sup>-15</sup>

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