Supporting Information

# Identification of $\beta$ scission products from free radical polymerizations of butyl acrylate at high temperature

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Figure S1: Molar mass distributions for polymer samples obtained from thermally initiated batch BA polymerizations at 120°C. #1 and #2 refer to the samples in Table 1.



- Figure S2: Molar mass distributions for polymer samples obtained from thermally initiated batch BA polymerizations at 140°C. #6 and #7 refer to the samples in Table 1.
- Table S1:Analytical SEC results for the fractions obtained after preparative SEC of a polymer<br/>obtained via semi-batch polymerization at 130°C.

fraction	content	<i>M</i> <sub>n</sub> / g⋅mol <sup>-1</sup>	<i>M</i> <sub>w</sub> / g·mol <sup>-1</sup>	D
1	0.161	9056	10280	1.14
2	0.372	5733	6679	1.17
3	0.200	3541	4348	1.23
4	0.128	2178	2940	1.35
5	0.066	1352	2118	1.57
6	0.034	876	1600	1.83
7	0.039	588	1192	2.02
6+7	0.073	757	1443	1.91
all		2148	4633	2.16



Figure S3: ESI-MS data of  $\beta$  scission products: semi-batch polymerization at 130°C after 1 h in xylene (S<sub>1</sub>: xylene end group).

MM <sub>n</sub> =CH2	+	Na <sup>+</sup>
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n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ $m/z$ / Da	intensity
	experimental	theoretical		
2	279.1567	279.1567	3.0E-05	591672
3	407.2404	407.2404	0.0E+00	558841
4	535.3240	535.3241	-1.3E-04	657030
5	663.4076	663.4079	-2.6E-04	390392
6	791.4911	791.4916	-4.9E-04	294774
7	919.5746	919.5753	-7.2E-04	242361
8	1047.6582	1047.6590	-8.5E-04	210042
9	1175.7415	1175.7428	-1.3E-03	127243
10	1303.8244	1303.8265	-2.1E-03	61132
11	1431.9065	1431.9102	-3.7E-03	24812
12	1559.9904	1559.9940	-3.6E-03	8650

MM<sub>n</sub><sup>H</sup> + Na⁺

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ $m/z$ / Da	intensity
	experimental	theoretical		
2	267.1568	267.1567	1.3E-04	709
3	395.2403	395.2404	-1.0E-04	16453
4	523.3239	523.3241	-2.3E-04	46569
5	651.4075	651.4079	-3.6E-04	34533
6	779.491	779.4916	-5.9E-04	40819
7	907.5744	907.5753	-9.2E-04	40246
8	1035.6579	1035.6590	-1.1E-03	42173
9	1163.7414	1163.7428	-1.4E-03	28933
10	1291.8244	1291.8265	-2.1E-03	18688
11	1419.9065	1419.9102	-3.7E-03	9589
12	1547.9903	1547.9940	-3.7E-03	4098
13	1676.0735	1676.0777	-4.2E-03	1496

 $S_1\text{-}MM_n^{=CH2}\text{ + }Na^+$ 

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ m/z / Da	intensity
	experimental	theoretical		
2	397.2350	397.2349	1.0E-04	27019
3	525.3190	525.3186	3.7E-04	40510
4	653.4027	653.4024	3.4E-04	43093
5	781.4867	781.4861	6.1E-04	40455
6	909.5707	909.5698	8.8E-04	41588
7	1037.6552	1037.6535	1.7E-03	43402
8	1165.7388	1165.7373	1.5E-03	34815
9	1293.8227	1293.8210	1.7E-03	22421
10	1421.9053	1421.9047	5.6E-04	12791
11	1549.9901	1549.9885	1.6E-03	6255
12	1678.0731	1678.0722	9.0E-04	2903

n	<i>m/z</i> / Da	<i>m/z</i> / Da	$\Delta m/z$ / Da	intensity
	experimental	theoretical		
2	385.2348	385.2349	-1.0E-04	46644
3	513.3185	513.3186	-1.3E-04	59076
4	641.4019	641.4024	-4.6E-04	49278
5	769.4855	769.4861	-5.9E-04	44869
6	897.5689	897.5698	-9.2E-04	44715
7	1025.6524	1025.6535	-1.1E-03	42558
8	1153.7357	1153.7373	-1.6E-03	30332
9	1281.8184	1281.8210	-2.6E-03	18299
10	1409.9005	1409.9047	-4.2E-03	9131
11	1537.9843	1537.9885	-4.2E-03	3997
12	1666.0663	1666.0722	-5.9E-03	1433



Figure S4: ESI-MS data of  $\beta$  scission products: batch polymerization at 130°C with  $c_{BA,0} = 2.06$ mol·L<sup>-1</sup> in dioxane (S<sub>2</sub>: dioxane end group).

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ $m/z$ / Da	intensity
	experimental	theoretical		
2	279.157	279.1567	3.3E-04	391271
3	407.2408	407.2404	4.0E-04	462495
4	535.3245	535.3241	3.7E-04	488422
5	663.4081	663.4079	2.4E-04	345787
6	791.4919	791.4916	3.1E-04	253310
7	919.5758	919.5753	4.8E-04	215149
8	1047.6595	1047.6590	4.5E-04	165288
9	1175.7431	1175.7428	3.2E-04	108448
10	1303.8263	1303.8265	-2.1E-04	55494
11	1431.9104	1431.9102	1.6E-04	23667
12	1559.9948	1559.9940	8.3E-04	9138
		-	-	-

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ m/z / Da	intensity
	experimental	theoretical		
2	267.1565	267.1567	-1.7E-04	558
3	395.2403	395.2404	-1.0E-04	8036
4	523.3239	523.3241	-2.3E-04	17225
5	651.4077	651.4079	-1.6E-04	17600
6	779.4912	779.4916	-3.9E-04	19433
7	907.5745	907.5753	-8.2E-04	18614
8	1035.6580	1035.6590	-1.0E-03	18023
9	1163.7412	1163.7428	-1.6E-03	13366
10	1291.8245	1291.8265	-2.0E-03	8657
11	1419.9076	1419.9102	-2.6E-03	4851
12	1547.9926	1547.9940	-1.4E-03	2111

1				
n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ m/z / Da	intensity
	experimental	theoretical		
2	379.2093	379.2091	2.0E-04	9398
3	507.2931	507.2928	2.7E-04	17464
4	635.3766	635.3766	4.1E-05	16127
5	763.4604	763.4603	1.1E-04	16967
6	891.5442	891.5440	1.8E-04	16107
7	1019.6283	1019.6277	5.5E-04	15392
8	1147.7116	1147.7115	1.2E-04	11410
9	1275.7945	1275.7952	-7.1E-04	7458
10	1403.8780	1403.8789	-9.4E-04	3915
11	1531.9630	1531.9627	3.3E-04	1749

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ m/z / Da	intensity
	experimental	theoretical		
2	367.2094	367.2091	3.0E-04	5683
3	495.2931	495.2928	2.7E-04	7904
4	623.3766	623.3766	4.1E-05	7497
5	751.4604	751.4603	1.1E-04	6864
6	879.5440	879.5440	-1.9E-05	6774
7	1007.6276	1007.6277	-1.5E-04	6294
8	1135.7112	1135.7115	-2.8E-04	4913
9	1263.7942	1263.7952	-1.0E-03	3273
10	1391.8779	1391.8789	-1.0E-03	1838



Figure S5: ESI-MS data of  $\beta$  scission products: batch polymerization at 120°C with  $c_{BA,0}$  = 2.92 mol·L<sup>-1</sup> in dioxane (S<sub>2</sub>: dioxane end group).

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ m/z / Da	intensity
	experimental	theoretical		
2	279.1562	279.1567	-4.7E-04	192560
3	407.2398	407.2404	-6.0E-04	99913
4	535.3233	535.3241	-8.3E-04	151569
5	663.4066	663.4079	-1.3E-03	53809
6	791.4902	791.4916	-1.4E-03	54280
7	919.5739	919.5753	-1.4E-03	32481
8	1047.6577	1047.6590	-1.3E-03	36991
9	1175.7406	1175.7428	-2.2E-03	16319
10	1303.8237	1303.8265	-2.8E-03	9034
11	1431.9066	1431.9102	-3.6E-03	3587
12	1559.9918	1559.9940	-2.2E-03	1479

 $MM_n^H + Na^+$ 

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta m/z$ / Da	intensity
	experimental	theoretical		
3	395.2391	395.2404	-1.3E-03	614
4	523.3226	523.3241	-1.5E-03	1664
5	651.4059	651.4079	-2.0E-03	1756
6	779.4896	779.4916	-2.0E-03	2129
7	907.5729	907.5753	-2.4E-03	2029
8	1035.6567	1035.6590	-2.3E-03	2019
9	1163.7390	1163.7428	-3.8E-03	1552
10	1291.8227	1291.8265	-3.8E-03	1103
11	1419.9047	1419.9102	-5.5E-03	621
12	1547.9893	1547.9940	-4.7E-03	318

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ m/z / Da	intensity
	experimental	theoretical		
1	251.1247	251.1254	-6.7E-04	502
2	379.2086	379.2091	-5.0E-04	2443
3	507.2918	507.2928	-1.0E-03	2165
4	635.3757	635.3766	-8.6E-04	2150
5	763.4585	763.4603	-1.8E-03	1686
6	891.5417	891.5440	-2.3E-03	1701
7	1019.6251	1019.6277	-2.6E-03	1643
8	1147.7085	1147.7115	-3.0E-03	1159
9	1275.7909	1275.7952	-4.3E-03	768
10	1403.8751	1403.8789	-3.8E-03	464
11	1531.9609	1531.9627	-1.8E-03	241

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta m/z$ / Da	intensity
	experimental	theoretical		
1	239.1248	239.1254	-5.7E-04	3234
2	367.2087	367.2091	-4.0E-04	1724
3	495.2917	495.2928	-1.1E-03	1848
4	623.3752	623.3766	-1.4E-03	1561
5	751.4585	751.4603	-1.8E-03	1264
6	879.5422	879.5440	-1.8E-03	1215
7	1007.6264	1007.6277	-1.3E-03	1120
8	1135.7094	1135.7115	-2.1E-03	853
9	1263.7908	1263.7952	-4.4E-03	549
10	1391.8752	1391.8789	-3.7E-03	310



Figure S6: ESI-MS data of  $\beta$  scission products: batch polymerization at 120°C with  $c_{BA,0}$  = 3.69 mol·L<sup>-1</sup> in dioxane (S<sub>2</sub>: dioxane end group).

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta m/z$ / Da	intensity
	experimental	theoretical		
2	279.1567	279.1567	3.0E-05	154718
3	407.2402	407.2404	-2.0E-04	76861
4	535.3238	535.3241	-3.3E-04	180612
5	663.4073	663.4079	-5.6E-04	80100
6	791.4908	791.4916	-7.9E-04	90155
7	919.5742	919.5753	-1.1E-03	39812
8	1047.658	1047.6590	-1.0E-03	41404
9	1175.7415	1175.7428	-1.3E-03	17176
10	1303.8253	1303.8265	-1.2E-03	10424
11	1431.9083	1431.9102	-1.9E-03	3881
12	1559.9919	1559.9940	-2.1E-03	1694

 $MM_n^H + Na^+$ 

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta m/z$ / Da	intensity
	experimental	theoretical		
3	395.2397	395.2404	-7.0E-04	625
4	523.3236	523.3241	-5.3E-04	2098
5	651.4073	651.4079	-5.6E-04	2863
6	779.4906	779.4916	-9.9E-04	3756
7	907.5732	907.5753	-2.1E-03	3392
8	1035.6571	1035.6590	-1.9E-03	3073
9	1163.7406	1163.7428	-2.2E-03	2284
10	1291.8244	1291.8265	-2.1E-03	1520
11	1419.9068	1419.9102	-3.4E-03	817
12	1547.9911	1547.9940	-2.9E-03	454

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ m/z / Da	intensity
	experimental	theoretical		
1	251.1255	251.1254	1.3E-04	251
2	379.2088	379.2091	-3.0E-04	1099
3	507.2925	507.2928	-3.3E-04	1735
4	635.3765	635.3766	-5.9E-05	2402
5	763.4589	763.4603	-1.4E-03	2039
6	891.5423	891.5440	-1.7E-03	1939
7	1019.6247	1019.6277	-3.0E-03	1738
8	1147.7096	1147.7115	-1.9E-03	1289
9	1275.7926	1275.7952	-2.6E-03	900
10	1403.8763	1403.8789	-2.6E-03	544
11	1531.9603	1531.9627	-2.4E-03	283

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta m/z$ / Da	intensity
	experimental	theoretical		
1	239.1251	239.1254	-2.7E-04	896
2	367.209	367.2091	-1.0E-04	871
3	495.292	495.2928	-8.3E-04	1357
4	623.3761	623.3766	-4.6E-04	1437
5	751.4596	751.4603	-6.9E-04	1348
6	879.5428	879.5440	-1.2E-03	1320
7	1007.6271	1007.6277	-6.5E-04	1136
8	1135.7099	1135.7115	-1.6E-03	854
9	1263.7771	1263.7952	-1.8E-02	767
10	1391.8779	1391.8789	-1.0E-03	328



Figure S7: ESI-MS data of  $\beta$  scission products: batch polymerization at 130°C with  $c_{BA,0}$  = 2.92 mol·L<sup>-1</sup> in dioxane (S<sub>2</sub>: dioxane end group).

 $MM_n^{=CH2} + Na^+$ 

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta m/z$ / Da	intensity
	experimental	theoretical		
2	279.1567	279.1567	3.0E-05	150724
3	407.2404	407.2404	0.0E+00	108442
4	535.3239	535.3241	-2.3E-04	129539
5	663.4073	663.4079	-5.6E-04	75672
6	791.4910	791.4916	-5.9E-04	56691
7	919.5747	919.5753	-6.2E-04	40972
8	1047.6583	1047.6590	-7.5E-04	32089
9	1175.7414	1175.7428	-1.4E-03	18723
10	1303.8246	1303.8265	-1.9E-03	9742
11	1431.9078	1431.9102	-2.4E-03	4454
12	1559.9929	1559.9940	-1.1E-03	1761

 $MM_n^H + Na^+$ 

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta m/z$ / Da	intensity
	experimental	theoretical		
2	267.1573	267.1567	6.3E-04	262
3	395.2399	395.2404	-5.0E-04	970
4	523.3237	523.3241	-4.3E-04	2618
5	651.4071	651.4079	-7.6E-04	2912
6	779.4904	779.4916	-1.2E-03	3237
7	907.5735	907.5753	-1.8E-03	2924
8	1035.6572	1035.6590	-1.8E-03	2719
9	1163.7402	1163.7428	-2.6E-03	1977
10	1291.8225	1291.8265	-4.0E-03	1447
11	1419.9063	1419.9102	-3.9E-03	748

12 1547.9889 1547.9940 -5.1E-03 411					
	12	1547.9889	1547.9940	-5.1E-03	411

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ m/z / Da	intensity
	experimental	theoretical		
1	251.1251	251.1254	-2.7E-04	834
2	379.2092	379.2091	1.0E-04	2188
3	507.2926	507.2928	-2.3E-04	3288
4	635.3759	635.3766	-6.6E-04	2945
5	763.4596	763.4603	-6.9E-04	2848
6	891.5431	891.5440	-9.2E-04	2626
7	1019.6266	1019.6277	-1.1E-03	2379
8	1147.7103	1147.7115	-1.2E-03	1696
9	1275.7924	1275.7952	-2.8E-03	1191
10	1403.8751	1403.8789	-3.8E-03	668
11	1531.9591	1531.9627	-3.6E-03	311

n	m/z / Da	m/z / Da	$\Lambda m/z$ / Da	intensity
	experimental	theoretical		
1	239.1252	239.1254	-1.7E-04	1353
2	367.2092	367.2091	1.0E-04	1337
3	495.2927	495.2928	-1.3E-04	1610
4	623.3760	623.3766	-5.6E-04	1430
5	751.4597	751.4603	-5.9E-04	1354
6	879.5433	879.5440	-7.2E-04	1168
7	1007.6273	1007.6277	-4.5E-04	1086
8	1135.7104	1135.7115	-1.1E-03	848
9	1263.7920	1263.7952	-3.2E-03	524
10	1391.8730	1391.8789	-5.9E-03	292



Figure S8: ESI-MS data of  $\beta$  scission products: batch polymerization at 130°C with  $c_{BA,0}$  = 3.69 mol·L<sup>-1</sup> in dioxane (S<sub>2</sub>: dioxane end group).

	1.15	1 15		
n	m/z / Da	<i>m/z /</i> Da	$\Delta m/z$ / Da	intensity
	experimental	theoretical		
2	279.1561	279.1567	-5.7E-04	126665
3	407.2395	407.2404	-9.0E-04	83707
4	535.3228	535.3241	-1.3E-03	94514
5	663.4063	663.4079	-1.6E-03	55015
6	791.4897	791.4916	-1.9E-03	39079
7	919.5732	919.5753	-2.1E-03	26225
8	1047.657	1047.6590	-2.0E-03	18745
9	1175.7406	1175.7428	-2.2E-03	10663
10	1303.824	1303.8265	-2.5E-03	5336
11	1431.9076	1431.9102	-2.6E-03	2387
12	1559.9906	1559.9940	-3.4E-03	938

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ $m/z$ / Da	intensity
	experimental	theoretical		
2	267.1559	267.1567	-7.7E-04	203
3	395.2397	395.2404	-7.0E-04	896
4	523.3225	523.3241	-1.6E-03	2067
5	651.4063	651.4079	-1.6E-03	2218
6	779.4891	779.4916	-2.5E-03	2432
7	907.5723	907.5753	-3.0E-03	2069
8	1035.6555	1035.6590	-3.5E-03	1793
9	1163.7396	1163.7428	-3.2E-03	1335
10	1291.8237	1291.8265	-2.8E-03	873
11	1419.9054	1419.9102	-4.8E-03	469
12	1547.9881	1547.9940	-5.9E-03	242

 $S_2$ - $MM_n^{=CH2}$  +  $Na^+$ 

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ $m/z$ / Da	intensity
	experimental	theoretical		
1	251.1247	251.1254	-6.7E-04	577
2	379.2083	379.2091	-8.0E-04	1657
3	507.2916	507.2928	-1.2E-03	2288
4	635.3747	635.3766	-1.9E-03	2058
5	763.4584	763.4603	-1.9E-03	1902
6	891.5415	891.5440	-2.5E-03	1639
7	1019.6248	1019.6277	-2.9E-03	1416
8	1147.7084	1147.7115	-3.1E-03	990
9	1275.7929	1275.7952	-2.3E-03	634
10	1403.8758	1403.8789	-3.1E-03	370
11	1531.9598	1531.9627	-2.9E-03	181

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ m/z / Da	intensity
	experimental	theoretical		
1	239.1246	239.1254	-7.7E-04	856
2	367.2082	367.2091	-9.0E-04	1144
3	495.2917	495.2928	-1.1E-03	1271
4	623.3745	623.3766	-2.1E-03	1097
5	751.459	751.4603	-1.3E-03	908
6	879.5419	879.5440	-2.1E-03	831
7	1007.6259	1007.6277	-1.8E-03	688
8	1135.709	1135.7115	-2.5E-03	465
9	1263.7929	1263.7952	-2.3E-03	296
10	1391.8754	1391.8789	-3.5E-03	172



Figure S9: ESI-MS data of  $\beta$  scission products: batch polymerization at 140°C with  $c_{BA,0}$  = 2.92 mol·L<sup>-1</sup> in dioxane (S<sub>2</sub>: dioxane end group).

1	I			
n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ $m/z$ / Da	intensity
	experimental	theoretical		
2	279.1558	279.1567	-8.7E-04	45667
3	407.2393	407.2404	-1.1E-03	44493
4	535.3226	535.3241	-1.5E-03	39521
5	663.4058	663.4079	-2.1E-03	27413
6	791.4894	791.4916	-2.2E-03	20387
7	919.5730	919.5753	-2.3E-03	16363
8	1047.6565	1047.6590	-2.5E-03	12174
9	1175.7395	1175.7428	-3.3E-03	7970
10	1303.8221	1303.8265	-4.4E-03	4199
11	1431.9053	1431.9102	-4.9E-03	2091
12	1559.9893	1559.9940	-4.7E-03	836
-		-		

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ m/z / Da	intensity
	experimental	theoretical		
3	395.2391	395.2404	-1.3E-03	927
4	523.3226	523.3241	-1.5E-03	1536
5	651.4057	651.4079	-2.2E-03	1601
6	779.4889	779.4916	-2.7E-03	1705
7	907.5723	907.5753	-3.0E-03	1537
8	1035.6561	1035.6590	-2.9E-03	1433
9	1163.7387	1163.7428	-4.1E-03	1043
10	1291.8214	1291.8265	-5.1E-03	736
11	1419.9040	1419.9102	-6.2E-03	384
12	1547.9882	1547.9940	-5.8E-03	200

-				1
n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ m/z / Da	intensity
	experimental	theoretical		
2	379.2083	379.2091	-8.0E-04	975
3	507.2914	507.2928	-1.4E-03	1475
4	635.3747	635.3766	-1.9E-03	1421
5	763.4580	763.4603	-2.3E-03	1419
6	891.5419	891.5440	-2.1E-03	1322
7	1019.6251	1019.6277	-2.6E-03	1178
8	1147.7081	1147.7115	-3.4E-03	872
9	1275.7915	1275.7952	-3.7E-03	560
10	1403.8741	1403.8789	-4.8E-03	341
11	1531.9586	1531.9627	-4.1E-03	152

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ m/z / Da	intensity
	experimental	theoretical		
2	367.2080	367.2091	-1.1E-03	492
3	495.2912	495.2928	-1.6E-03	575
4	623.3745	623.3766	-2.1E-03	545
5	751.4577	751.4603	-2.6E-03	510
6	879.5420	879.5440	-2.0E-03	486
7	1007.6251	1007.6277	-2.6E-03	454
8	1135.7090	1135.7115	-2.5E-03	337
9	1263.7915	1263.7952	-3.7E-03	229
10	1391.8735	1391.8789	-5.4E-03	126



Figure S10: ESI-MS data of  $\beta$  scission products: batch polymerization at 140°C with  $c_{BA,0}$  = 3.69 mol·L<sup>-1</sup> in dioxane (S<sub>2</sub>: dioxane end group).

<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ $m/z$ / Da	intensity
experimental	theoretical		
279.1558	279.1567	-8.7E-04	150376
407.2391	407.2404	-1.3E-03	115060
535.3224	535.3241	-1.7E-03	100496
663.4058	663.4079	-2.1E-03	72539
791.4891	791.4916	-2.5E-03	49879
919.5724	919.5753	-2.9E-03	39468
1047.656	1047.6590	-3.0E-03	26957
1175.7396	1175.7428	-3.2E-03	17989
1303.8232	1303.8265	-3.3E-03	9417
1431.9059	1431.9102	-4.3E-03	4720
1559.9895	1559.9940	-4.5E-03	1976
	<i>m/z /</i> Da experimental 279.1558 407.2391 535.3224 663.4058 791.4891 919.5724 1047.656 1175.7396 1303.8232 1431.9059 1559.9895	m/z / Dam/z / Daexperimentaltheoretical279.1558279.1567407.2391407.2404535.3224535.3241663.4058663.4079791.4891791.4916919.5724919.57531047.6561047.65901175.73961175.74281303.82321303.82651431.90591431.91021559.98951559.9940	$m/z$ / Da $m/z$ / Da $\Delta m/z$ / Daexperimentaltheoretical $\Delta m/z$ / Da279.1558279.1567-8.7E-04407.2391407.2404-1.3E-03535.3224535.3241-1.7E-03663.4058663.4079-2.1E-03791.4891791.4916-2.5E-03919.5724919.5753-2.9E-031047.6561047.6590-3.0E-031175.73961175.7428-3.2E-031303.82321303.8265-3.3E-031431.90591431.9102-4.3E-031559.98951559.9940-4.5E-03

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ m/z / Da	intensity
	experimental	theoretical		
2	267.1561	267.1567	-5.7E-04	319
3	395.2391	395.2404	-1.3E-03	2090
4	523.3223	523.3241	-1.8E-03	3657
5	651.4055	651.4079	-2.4E-03	3925
6	779.4886	779.4916	-3.0E-03	4094
7	907.5718	907.5753	-3.5E-03	3574
8	1035.6547	1035.6590	-4.3E-03	3276
9	1163.7384	1163.7428	-4.4E-03	2421
10	1291.8221	1291.8265	-4.4E-03	1659
11	1419.9046	1419.9102	-5.6E-03	939
12	1547.9877	1547.9940	-6.3E-03	492

 $S_2$ - $MM_n^{=CH2}$  +  $Na^+$ 

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ $m/z$ / Da	intensity
	experimental	theoretical		
1	251.1245	251.1254	-8.7E-04	803
2	379.2079	379.2091	-1.2E-03	2642
3	507.2911	507.2928	-1.7E-03	3349
4	635.3744	635.3766	-2.2E-03	3065
5	763.4577	763.4603	-2.6E-03	3099
6	891.5407	891.5440	-3.3E-03	2779
7	1019.6242	1019.6277	-3.5E-03	2414
8	1147.7078	1147.7115	-3.7E-03	1782
9	1275.7918	1275.7952	-3.4E-03	1202
10	1403.8749	1403.8789	-4.0E-03	718
11	1531.958	1531.9627	-4.7E-03	357

n	<i>m/z /</i> Da	<i>m/z /</i> Da	$\Delta$ m/z / Da	intensity
	experimental	theoretical		
1	239.1246	239.1254	-7.7E-04	1194
2	367.2079	367.2091	-1.2E-03	1305
3	495.2909	495.2928	-1.9E-03	1320
4	623.3744	623.3766	-2.2E-03	1286
5	751.4578	751.4603	-2.5E-03	1100
6	879.5412	879.5440	-2.8E-03	1013
7	1007.6244	1007.6277	-3.3E-03	886
8	1135.7082	1135.7115	-3.3E-03	695
9	1263.7919	1263.7952	-3.3E-03	466
10	1391.8741	1391.8789	-4.8E-03	282

# Figures S11 – S16 give the following NMR spectra of MM<sub>3</sub><sup>=CH2</sup>

<sup>1</sup>H spectrum (600 MHz)
<sup>13</sup>C {<sup>1</sup>H} spectrum (150 MHz)
<sup>13</sup>C-DEPT135 spectrum (150 MHz)
<sup>1</sup>H, <sup>1</sup>H-COSY (DQF-COSY, 600 MHz)
<sup>1</sup>H, <sup>13</sup>C-HSQC (600 MHz)
<sup>1</sup>H, <sup>13</sup>C-HMBC (600 MHz)



Figure S11: <sup>1</sup>H spectrum of the unsaturated trimer ( $MM_3^{=CH2}$ ) (600 MHz).



Figure S12:  ${}^{13}C{}^{1}H$  spectrum of MM<sub>3</sub><sup>=CH2</sup> (150 MHz).



Figure S13: <sup>13</sup>C-DEPT135 spectrum of  $MM_3^{=CH2}$  (150 MHz).



Figure S14: <sup>1</sup>H,<sup>1</sup>H-COSY (DQF-COSY) of MM<sub>3</sub><sup>=CH2</sup> (600 MHz).



Figure S15:  ${}^{1}$ H, ${}^{13}$ C-HSQC of MM $_{3}^{=CH2}$  (600 MHz).



Figure S16:  ${}^{1}$ H, ${}^{13}$ C-HMBC of MM3<sup>=CH2</sup> (600 MHz).