

**Electronic Supplementary Information**

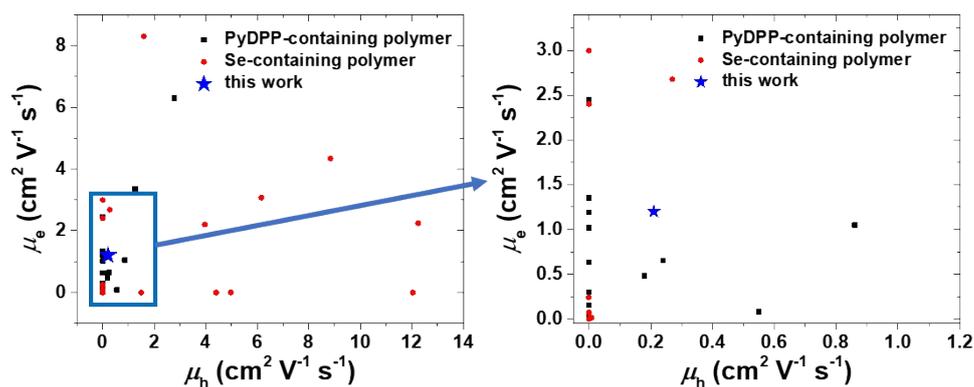
**Understanding of Simultaneous Pyridine and Selenophene-containing Copolymers and their Polarity Conversion in Transistors**

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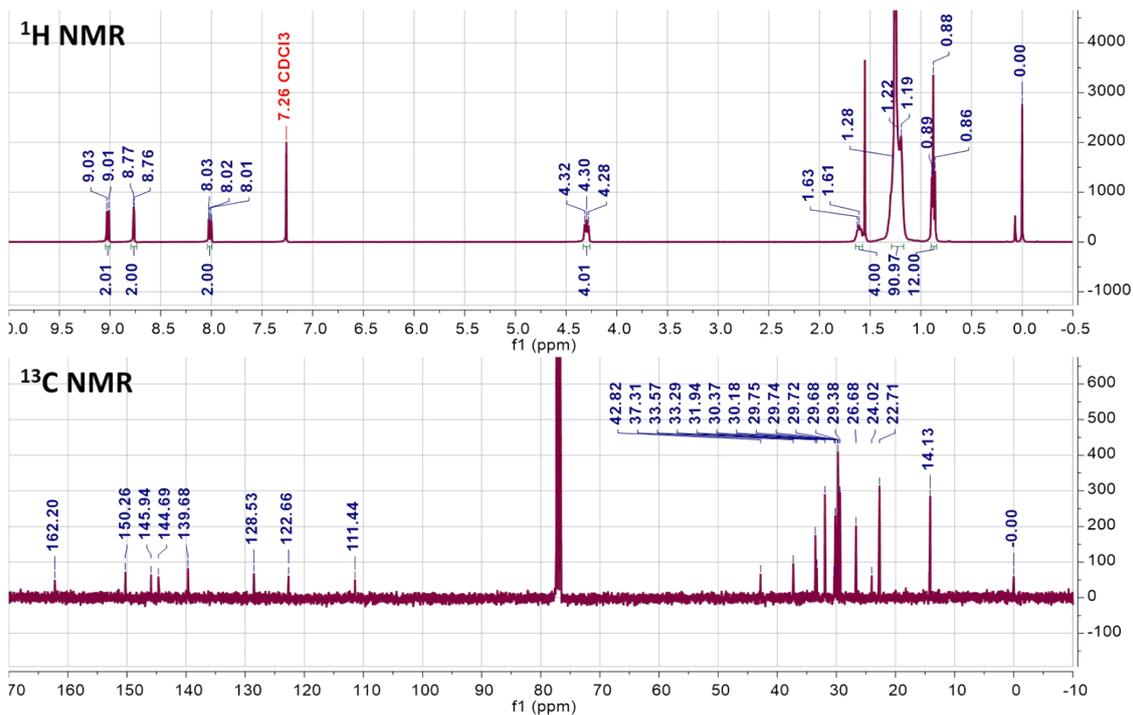


**Chart S1.** The charge carrier transports of PyDPP-containing and Se-containing polymers in OFET devices.

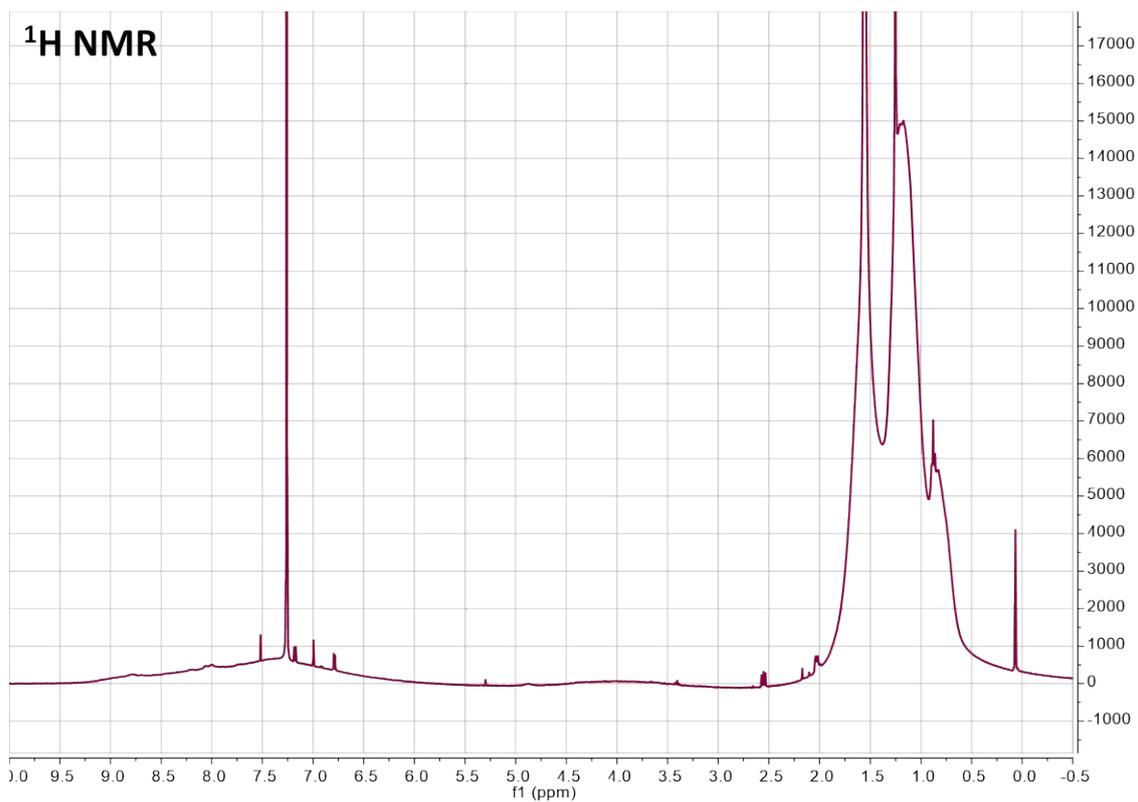
**Table S1.** The charge carrier transports of PyDPP-containing and Se-containing polymers in OFET devices.

Name	unit	$\mu_e$ ( $\text{cm}^2 \text{V}^{-1} \text{s}^{-1}$ )	$\mu_h$ ( $\text{cm}^2 \text{V}^{-1} \text{s}^{-1}$ )	ref
PDBPyTT	PyDPP	3.36	1.26	1 <sup>1</sup>
DPPPy-BT2CN	PyDPP	0.3	-	2 <sup>2</sup>
PDBPyDT2FBT	PyDPP	0.65	0.24	3 <sup>3</sup>
PDBPyBT	PyDPP	6.3	2.78	4 <sup>4</sup>
PPyDPP1-BT	PyDPP	1.05	0.86	5 <sup>5</sup>
PPyDPP1-4FBT	PyDPP	1.02	-	5 <sup>5</sup>
PPyDPP2-4FBT	PyDPP	2.45	-	5 <sup>5</sup>
PPyDPP1-4FTVT	PyDPP	1.19	-	5 <sup>5</sup>
PPyDPP2-4FTVT	PyDPP	1.35	-	5 <sup>5</sup>
PDPP[Py] <sub>2</sub> -T	PyDPP	0.63	-	6 <sup>6</sup>
PDPP[Py] <sub>2</sub> -TF2	PyDPP	0.15	-	6 <sup>6</sup>
PPyDPP-T	PyDPP	0.004	0.0008	7 <sup>7</sup>
PPyDPP-2FT	PyDPP	0.021	0.0003	7 <sup>7</sup>
PPyTDPP-TT	PyDPP	0.48	0.18	8 <sup>8</sup>
PPyTDPP-BT	PyDPP	0.08	0.55	8 <sup>8</sup>
PNDIBS	Se	0.07	-	9 <sup>9</sup>

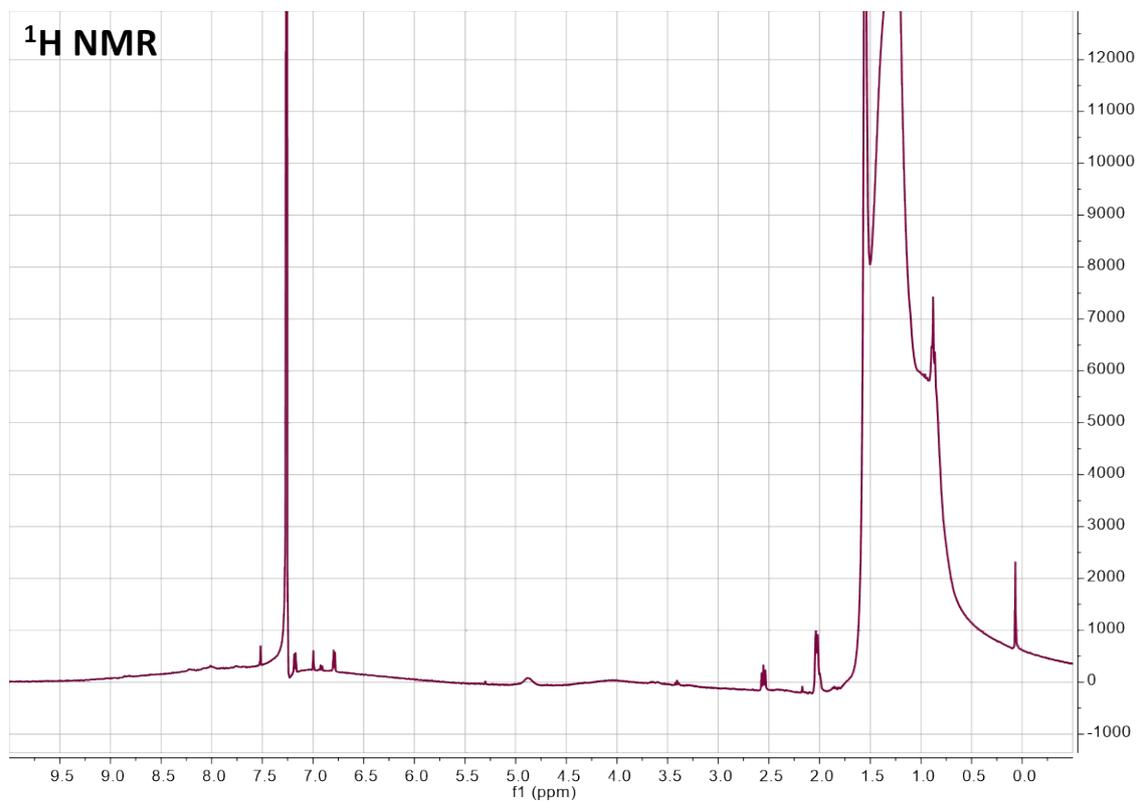
PNDI-SVS	Se	2.4	-	10 <sup>10</sup>
PTDPPSe-SiC4	Se	3.07	6.16	11 <sup>11</sup>
PTDPPSe-SiC5	Se	4.34	8.84	11 <sup>11</sup>
PTDPPSe-SiC6	Se	2.2	3.97	11 <sup>11</sup>
P-24-DPPDBSE	Se	-	4.4	12 <sup>12</sup>
P-29-DPPDBSE	Se	-	12.04	12 <sup>12</sup>
PNBS	Se	8.3	1.6	13 <sup>13</sup>
PNBSF	Se	3	-	13 <sup>13</sup>
PDPPDTSE	Se	-	4.97	14 <sup>14</sup>
PCDSeBT	Se	-	0.001	15 <sup>15</sup>
PDPP(SE)- $\epsilon$ -C8C15	Se	2.25	12.25	16 <sup>16</sup>
eNDIBS	Se	0.24	-	17 <sup>17</sup>
PNBDO-ST5	Se	2.68	0.27	18 <sup>18</sup>
PNBDO-SS5	Se	0.012	0.0084	18 <sup>18</sup>
P(DPP-alt-DTBS <sub>2</sub> e)	Se	-	1.5	19 <sup>19</sup>



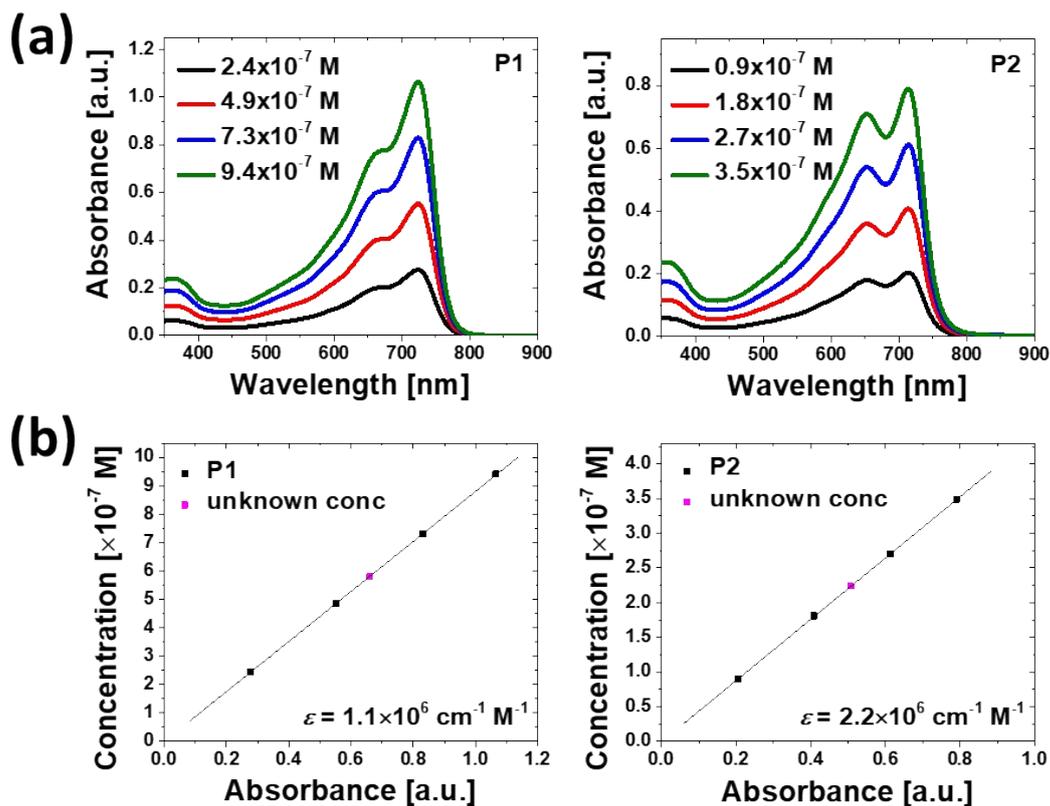
**Fig. S1** The <sup>1</sup>H and <sup>13</sup>C NMR spectroscopy of 5PyDPP.



**Fig. S2** The <sup>1</sup>H NMR spectroscopy of P1.



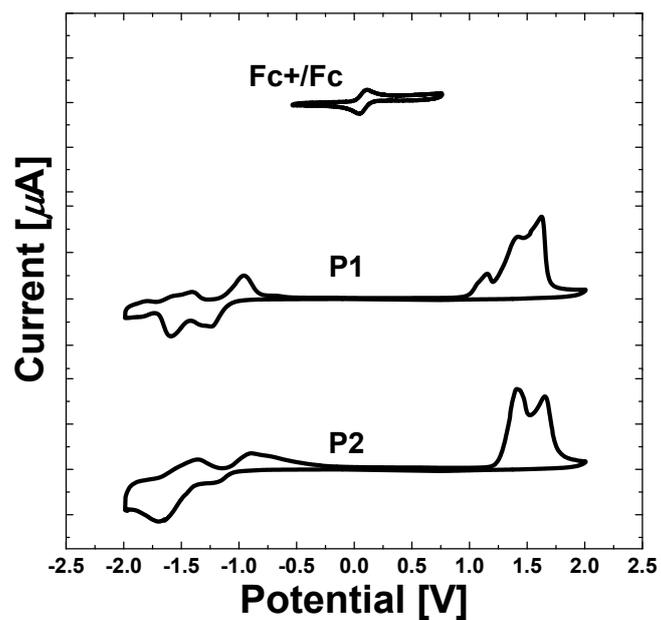
**Fig. S3** The <sup>1</sup>H NMR spectroscopy of P2.



**Fig. S4** (a) UV-vis absorption spectra at different concentrations of the polymers in chlorobenzene and (b) calibration plots of the concentration of chlorobenzene-polymer solutions according to the measured absorbances at each  $\lambda_{\text{max}}$ .

**Table S2.** Solubility of the Polymers in Chlorobenzene (mg/mL).

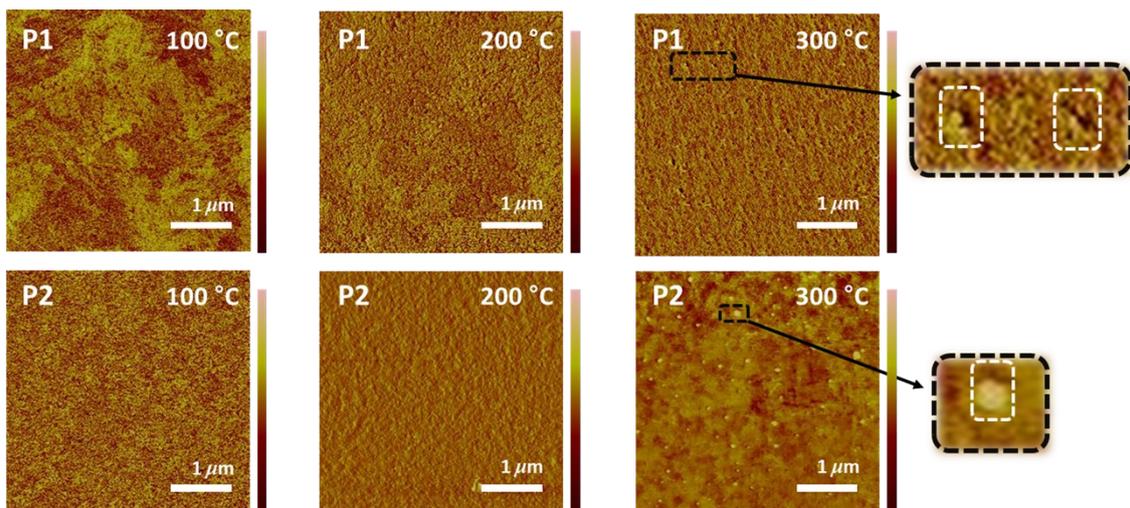
	P1	P2
Solubility [mg/ mL]	5.80	2.24



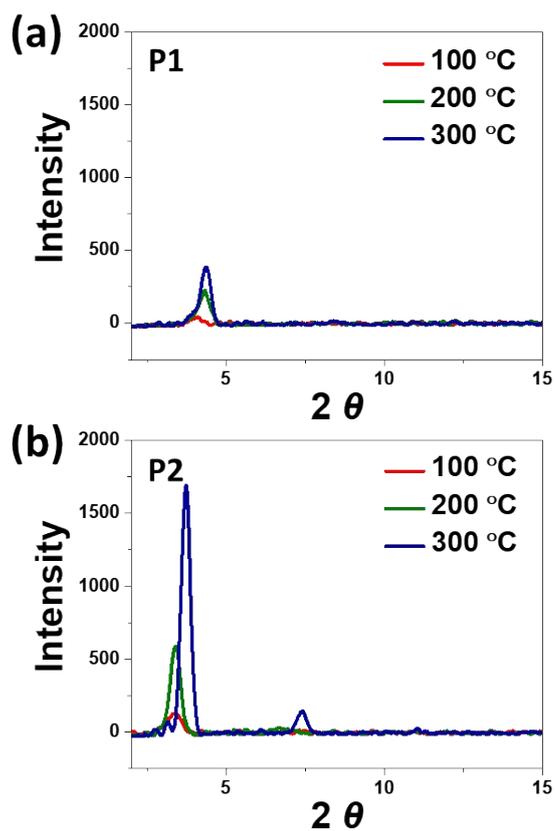
**Fig. S5** Cyclic voltammograms of the polymer films in  $n\text{-Bu}_4\text{NPF}_6/\text{CH}_3\text{CN}$  solution (scan rate,  $100\text{ mVs}^{-1}$ ).

**Table S3.** Calculated dipole moments and energy levels of the BT units by DFT.

	Vector			Dipole Moment [D]	$E_{\text{HOMO}}^{\text{DFT}}$ [eV]	$E_{\text{LUMO}}^{\text{DFT}}$ [eV]
	x	y	z			
<b>T1</b>	1.4123	3.4552	2.3629	4.4178	-5.11	-3.51
<b>T2</b>	1.4103	4.1109	1.1614	4.4986	-5.08	-3.49



**Fig. S6** AFM phase images of annealed polymer films. The polymer films were formed by spin casting on SiO<sub>2</sub>/Si substrates.



**Fig. S7** 1D out-of-plane XRD patterns of spin-casted (a) P1 and (b) P2 films on SiO<sub>2</sub> substrates at different annealing condition.

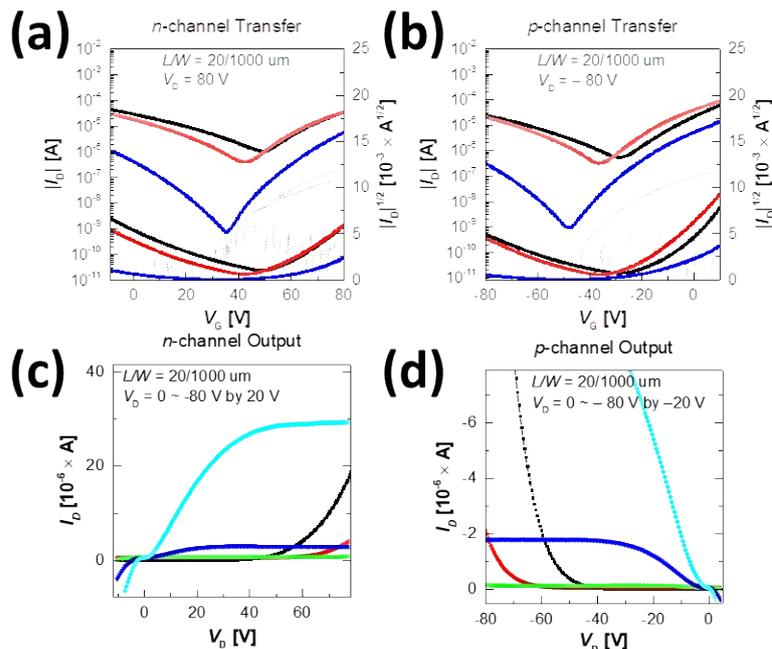
**Table S4.** Crystallographic parameters of spin-cast polymer films obtained from XRD at different annealing condition.

polymer film	Temperature [°C]	$2\theta$ [°]	d-spacing [Å]
P1	100	4.115	21.46
	200	4.3429	20.33
	300	4.3718	20.20
P2	100	3.3305	26.51
	200	3.4173	25.83
	300	3.7355	23.63

**Table S5.** 2D-GIXD crystallographic parameters of spin-cast polymer films at different annealing conditions.<sup>a</sup>

Polymer film	$T_a$ [°C]	Lamellar spacing <sup>b</sup>			$\pi$ - $\pi$ spacing <sup>c</sup>		
		$q_z$ [Å <sup>-1</sup> ]	$d_z$ [Å]	$L_c$ [Å]	$q_{xy}$ [Å <sup>-1</sup> ]	$d_{xy}$ [Å]	$L_c$ [Å]
P1	As-cast	0.274	22.90	122.2	1.741	3.61	90.7
	100	0.284	22.15	162.6	1.747	3.60	108.3
	200	0.303	20.77	309.0	1.743	3.61	121.9
	300	0.311	20.22	387.1	1.738	3.62	136.0
P2	As-cast	0.230	27.36	167.1	1.758	3.57	150.3
	100	0.231	27.18	217.5	1.758	3.57	170.0
	200	0.232	27.06	242.5	1.754	3.58	182.5
	300	0.261	24.09	425.8	1.754	3.58	201.8

<sup>a</sup>The parameters were calculated from GIXD profiles; The parameters for <sup>b</sup>the lamellar spacing and <sup>c</sup>the  $\pi$ - $\pi$  spacing were derived from the peaks along  $q_z$  and  $q_{xy}$  axis, respectively.

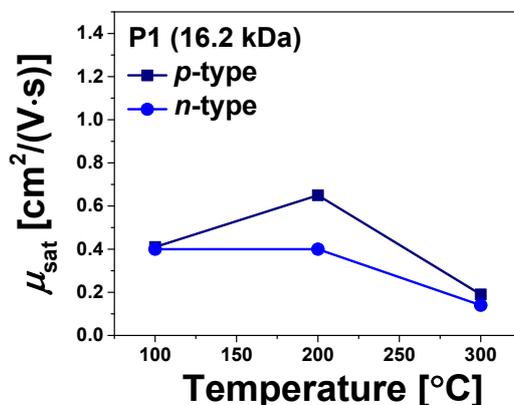


**Fig. S8** (a,b) Transfer and (c,d) output characteristics of the *n*-type and *p*-type devices of a low molecular weight P1 at different annealing conditions: 100 °C (black), 200 °C (red) and 300 °C (blue).

**Table S6.** Electrical characteristics of a low molecular weight P1 OFET Devices at different annealing conditions.

	$T_a$ [°C]	<i>n</i> -channel			<i>p</i> -channel		
		$\mu_e$ [cm <sup>2</sup> V <sup>-1</sup> s <sup>-1</sup> ]	$V_{th}$ [V]	$I_{on}$ / $I_{off}$	$\mu_e$ [cm <sup>2</sup> V <sup>-1</sup> s <sup>-1</sup> ]	$V_{th}$ [V]	$I_{on}$ / $I_{off}$
<b>P1</b> (16.2 kDa)	100	0.4 (± 0.1)	47 (± 8)	10	0.41 (± 0.2)	-26 (± 4)	10 <sup>2</sup>
	200	0.4 (± 0.02)	45 (± 2)	10 <sup>2</sup>	0.65 (± 0.03)	-35 (± 0.1)	10 <sup>2</sup>
	300	0.14 (± 0.03)	36 (± 1)	10 <sup>4</sup>	0.19 (± 0.01)	-35 (± 0.2)	10 <sup>4</sup>

4 devices for each condition were fabricated and used for mobility calculation.



**Fig. S9** Hole and electron mobilities of a low molecular weight P1 at different annealing conditions.

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