

Supporting Information

Isolated α -Turn and Incipient γ -Helix

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EXPERIMENTAL SECTION

General Methods: Unless otherwise specified, all non-aqueous reactions were performed under an inert argon atmosphere. Anhydrous DCM was obtained by passage through a solvent filtration system (Glass Contour, Irvine, CA) and transferred by syringe. Reaction mixture solutions (after aqueous workup) were dried over anhydrous Na₂SO₄, filtered, and rotary-evaporated under reduced pressure. Column chromatography was performed on 230-400 mesh silica gel, and thin-layer chromatography was performed on alumina plates coated with silica gel (Merck 60 F₂₅₄ plates). Visualization of the developed chromatogram was performed by UV absorbance or staining with iodine. Melting points were obtained on a Buchi melting point B-540 apparatus and are uncorrected. Accurate mass measurements were performed on an LC-MSD instrument in electrospray ionization (ESI-TOF) mode at the Université de Montréal Mass Spectrometry facility, and are listed as empirical formula confirmations [M + H]⁺. Nuclear magnetic resonance (NMR ¹H, ¹³C, COSY, HMBC) spectra were recorded on Bruker 300, 400, 500 and 700 MHz spectrometers. ¹H and ¹³C NMR spectra were respectively referenced to CDCl₃ (7.26 ppm and 77.16 ppm) or DMSO-d₆ (2.50 ppm, and 39.52 ppm). Coupling constant *J* values were measured in Hertz (Hz) and chemical shift values in parts per million (ppm). Infrared absorption spectra for characterization of compounds were recorded on a Bruker Alpha P FT-IR spectrometer equipped with a single reflection ATR sampling module which allows spectral acquisition from neat solid and liquid samples. Band positions are reported in reciprocal centimeters (cm⁻¹). The FT-IR absorption spectra in CDCl₃ (99.8% *d*; Merck) solution were recorded at 293 K using a Perkin-Elmer model 1720X FT-IR spectrophotometer, nitrogen flushed, equipped with a sample-shuttle device, at 2 cm⁻¹ nominal resolution, averaging 100 scans. Solvent (baseline) spectra were obtained under the same conditions. For spectral elaboration, the software SpectraCalc (Galactic) was employed. Cells were used with CaF₂ windows and path lengths of 0.1 mm, 1.0 mm, and 10.0 mm.

Reagents: Unless specified otherwise, commercially available reagents were purchased from Aldrich, A & C American Chemicals Ltd., Fluka and Advanced Chemtech™ and used without further purification, including adamantan-2-one, ammonium formate, formic acid, POCl_3 , and Et_3N .

SYNTHESIS

2-N-(Formyl)aminoadamantane-2-N'-(*iso*-propyl)carboxamide (4a**)**

A solution of *iso*-propyl isocyanide (**3a**, 1 g, 14.5 mmol, 1eq) in MeOH (2M) was treated with adamantan-2-one (2.2 g, 14.5 mmol, 1eq) and ammonium formate (1.1 g, 17.4 mmol, 1.2eq, dissolved in the minimum amount of H_2O). The mixture was stirred for 16h. The volatiles were evaporated. The residue was dissolved in CHCl_3 , washed with water and brine, and dried over Na_2SO_4 . The volatiles were evaporated and the residue was purified by column chromatography using 40% EtOAc in hexanes as eluent. Evaporation of the collected fractions gave Formyl-Adm-NH-*i*-Pr (**4a**, 2.4 g, 62%) as white powder: R_f 0.33 (3:97 MeOH: CH_2Cl_2); m.p. 181-185 °C; ^1H NMR (500 MHz, CDCl_3) δ 1.13 (d, J = 6.5, 6H), 1.65-1.74 (m, 6H), 1.80-1.82 (m, 2H), 1.92-1.96 (m, 4H), 2.64 (br, 2H), 4.01-4.07 (m, 1H), 6.07 (s, 1H), 6.96 (d, J = 7.5, 1H), 8.12 (d, J = 2.5, 1H); ^{13}C NMR (500 MHz, CDCl_3): δ 22.6, 26.5, 26.6, 32.1, 32.6, 34.0, 37.4, 41.4, 64.9, 161.7, 171.2; IR (neat) ν = 3320, 3286, 2903, 1652, 1518; HRMS (ESI) m/z calcd for $\text{C}_{15}\text{H}_{25}\text{N}_2\text{O}_2$ 265.1911; found [M + H] $^+$ 265.1916.

2-N-(Formyl)aminoadamantane-2-N'-(*tert*-butyl)carboxamide (4b**)**

Amide **4b** was synthesized from *tert*-butyl isocyanide (**3b**, 3 g, 41.6 mmol) according to the procedure described for the synthesis of formamide **4a** and purified by column chromatography using 30% EtOAc in hexanes as eluent. Evaporation of the collected fractions gave Formyl-Adm-NH-*t*-Bu (**4b**, 8.3g, 72%) as a white powder: R_f 0.2 (7:3 hexanes:EtOAc); m.p. 197-200 °C; ^1H NMR (500 MHz, CDCl_3): δ 1.32 (s, 9H), 1.3-1.72 (m, 6H), 1.79-1.84 (m, 2H), 1.92-2.00 (m, 4H), 2.60 (br, 2H), 6.16 (s, 1H), 6.95 (s, 1H), 8.11 (d, J = 2.0 Hz, 1H); ^{13}C NMR (500 MHz, CDCl_3) δ 26.4, 26.6, 28.6, 32.1, 32.6, 34.1, 37.3, 51.0, 65.4, 161.7, 171.0; IR (neat) ν : 3264, 2906, 1674,

1649, 1540, 1222, 741; HRMS (ESI) m/z calcd for $C_{16}H_{27}N_2O_2$ 279.2067; found $[M + H]^+$ 279.2066.

2-Isocyanoadamantane-2-*N'*-(*iso*-propyl)carboxamide (5a**)**

Formamide **4a** (2 g, 7.6 mmol) in CH_2Cl_2 (8 mL) was treated with Et_3N (6.36 g, 45.6 mmol), cooled to -5 °C, and treated dropwise with $POCl_3$ (1.07 g, 11.4 mmol). After stirring at this temperature for 1-2 h, the reaction mixture was vigorously stirred and treated with a solution of saturated $NaHCO_3$ (8mL). The organic phase was separated, washed with brine, and dried over Na_2SO_4 . Evaporation of the volatiles under reduced pressure afforded a yellow powder, which was purified by column chromatography using 30% EtOAc in hexanes as eluent. Evaporation of the collected fractions gave isonitrile **5a** (1.70 g, 90%) as white powder: R_f 0.53 (7:3 hexanes:EtOAc); m.p. 130-133 °C; 1H NMR (500 MHz, $CHCl_3$) δ 1.18 (d, $J = 7.0$, 6H), 1.72 (m, 2H), 1.76-1.78 (m, 4H), 1.81 (m, 1H), 1.89-1.91 (m, 1H), 1.96-1.99 (m, 2H), 2.23-2.26 (m, 2H), 2.31 (m, 2H), 4.07-4.14 (m, 1H), 5.74 (br, 1H); ^{13}C NMR (500 MHz, $CDCl_3$) δ 22.4, 26.2, 26.4, 33.3, 35.0 (2C), 37.3, 42.1, 68.6, 158.2, 166.7; IR (neat): ν = 3315, 2915, 2122, 1647, 1533, 653; HRMS (ESI) m/z calcd for $C_{15}H_{23}N_2O$ 247.1805; found $[M + H]^+$ 247.1810.

2-Isocyanoadamantane-2-*N'*-(*tert*-butyl)carboxamide (5b**)**

Isonitrile **5b** was synthesized from formamide **4b** (6.75 g, 24.24 mmol) according to the procedure described for the synthesis of isocyanide **5a** and purified by column chromatography using 10% EtOAc in hexanes as eluent. Evaporation of the collected fractions gave isocyanide **5b** (6 g, 94%) as white powder: R_f 0.36 (9:1 hexanes:EtOAc); m.p. 116-120 °C; 1H NMR (500 MHz, $CHCl_3$) δ 1.37 (s, 9H), 1.71 (m, 2H), 1.75-1.77 (m, 4H), 1.81 (m, 1H), 1.89 (m, 1H), 1.94-1.97 (m, 2H), 2.22-2.24 (m, 2H), 2.28 (m, 2H), 5.69 (s, 1H); ^{13}C NMR (500 MHz, $CDCl_3$) δ 26.2, 26.3, 28.5, 33.3, 35.1, 37.2, 39.4, 51.8, 68.9, 158.0, 166.6; IR (neat) ν : 3363, 2917, 2120, 1660, 1527, 1452, 1214, 583; HRMS (ESI) m/z calcd for $C_{16}H_{25}N_2O$ 261.19614; found $[M + H]^+$ 261.19712.

Formyl-Adm-Adm-NH-*i*-Pr (6a**)**

Dipeptide **6a** was synthesized from isonitrile **5a** (1.6 g, 6.47 mmol) according to the procedure described for the synthesis of formamide **4a** and purified by column chromatography using 40% hexanes in EtOAc as eluent. Evaporation of the collected fractions gave HCO-Adm-Adm-NH-*i*-Pr (**6a**, 2.5 g, 87%) as white powder: R_f 0.25 (4:6 hexanes:EtOAc); m.p. 192-198 °C; ^1H NMR (500 MHz, CHCl_3) δ 1.10 (d, J = 6.6 Hz, 6H), 1.62 (m, 1H), 1.65 (m, 1H), 1.69-1.70 (m, 9H), 1.76 (m, 1H), 1.79-1.83 (m, 4H), 1.93-1.95 (m, 6H), 2.00 (m, 1H), 2.03 (m, 1H), 2.64-2.66 (m, 4H), 3.99-4.07 (m, 1H), 5.74 (s, 1H), 6.81 (d, J = 7.9, 1H), 6.92 (s, 1H), 8.13 (d, J = 1.9, 1H); ^{13}C NMR (500 MHz, CDCl_3) δ 22.8, 26.4, 26.6 (2C), 26.9, 32.2, 32.6, 32.7, 32.8, 34.0, 34.2, 37.3, 37.7, 41.1, 64.7, 65.3, 161.3, 171.2, 172.1; IR (neat) v: 3360, 2904, 1677, 1511, 751. HRMS (ESI) m/z calcd for $\text{C}_{26}\text{H}_{40}\text{N}_3\text{O}_3$ 442.3060; found [M + H] $^+$ 442.3073.

Formyl-Adm-Adm-NH-*t*-Bu (6b**)**

Dipeptide **6b** was synthesized from isonitrile **5b** (2 g, 7.65 mmol) according to the procedure described for the synthesis of formamide **4a** and purified by column chromatography using 30% EtOAc in hexanes as eluent. Evaporation of the collected fractions gave HCO-Adm-Adm-NH-*t*-Bu (**6b**, 2.96 g, 85%) as white powder: R_f 0.21 (7:3 hexanes:EtOAc); m.p. 228-230 °C; ^1H NMR (500 MHz, CDCl_3) δ 1.30 (s, 9H), 1.62 (m, 1H), 1.64 (m, 1H), 1.68-1.72 (m, 8H), 1.75 (m, 1H), 1.78-1.82 (m, 5H), 1.94-1.96 (m, 6H), 2.00 (m, 1H), 2.03 (m, 1H), 2.63 (m, 2H), 2.68 (m, 2H), 5.55 (s, 1H), 6.83 (s, 1H), 6.93 (s, 1H), 8.13 (d, J = 1.9 Hz, 1H); ^{13}C NMR (500 MHz, CDCl_3): δ 26.3, 26.5, 26.6, 26.9, 28.8, 32.3, 32.6, 32.7, 32.8, 34.1, 34.2, 37.3, 37.6, 50.7, 65.3, 65.4, 161.0, 170.9, 171.9; IR (neat) v: 3256, 2917, 1684, 1655, 1538, 1499, 1217; HRMS (ESI) m/z calcd for $\text{C}_{27}\text{H}_{42}\text{N}_3\text{O}_3$ 456.3221; found [M + H] $^+$ 456.3242.

‘Isocyano’-Adm-Adm-NH-*i*-Pr (7a**)**

Isonitrile **7a** was synthesized from dipeptide **6a** (2.4 g, 5.4 mmol) according the procedure described for the synthesis of isocyanide **5a**, and purified by column chromatography using 20% EtOAc in hexanes as eluent. Evaporation of the collected fractions gave ‘CN’-Adm-Adm-NH-*i*-Pr

(7a, 2.23 g, 97%) as white powder: R_f 0.51 (7:3 hexanes:EtOAc); m.p 228-235 °C; ^1H NMR (500 MHz, CHCl_3) δ 1.12 (d, J = 6.5 Hz, 6H), 1.71-1.72 (m, 7H), 1.77 (m, 3H), 1.80 (m, 4H), 1.86-1.92 (m, 6H), 1.97-2.00 (m, 2H), 2.22-2.25 (m, 2H), 2.31 (m, 2H), 2.72 (m, 2H), 4.02-4.09 (m, 1H), 5.73 (s, 1H), 6.84 (d, J = 8.0 Hz, 1H); ^{13}C NMR (500 MHz, CDCl_3): δ 22.8, 26.1, 26.2, 26.4, 26.7, 32.4, 32.9, 33.2, 33.4, 34.2, 35.0, 37.1, 37.3, 41.3, 64.8, 69.0, 158.8, 167.2, 170.3; IR (neat) ν : 3357, 2856, 2135, 1677, 1632, 1527, 753. HRMS (ESI) m/z calcd for $\text{C}_{26}\text{H}_{38}\text{N}_3\text{O}_2$ 424.2959, found [M + H] $^+$ 424.2967.

‘Isocyano’-Adm-Adm-NH-*t*-Bu (7b)

Isonitrile **7a** was synthesized from dipeptide **6b** (1.16g, 2.54 mmol) according the procedure described for the synthesis of isocyanide **5a**, and purified by column chromatography using 30% EtOAc in hexanes as eluent. Evaporation of the collected fractions gave ‘CN’-Adm-Adm-NH-*i*-Pr (**7b**, 0.62g, 56%) as white powder: R_f 0.67 (7:3 Hex:EtOAc); m.p. 207-210 °C; ^1H NMR (500 MHz, CDCl_3) δ 1.32 (s, 9H), 1.69-1.72 (m, 7H), 1.77 (m, 3H), 1.79-1.82 (m, 4H), 1.87-1.90 (m, 6H), 1.98-2.00 (m, 2H), 2.22-2.25 (m, 2H), 2.33 (m, 2H), 2.68 (m, 2H), 5.30 (s, 1H), 6.91 (s, 1H); ^{13}C NMR (500 MHz, CDCl_3): δ 26.1, 26.2, 26.4, 26.7, 28.8, 32.6, 32.9, 33.2, 33.5, 34.2, 35.0, 37.1, 37.3, 51.0, 65.4, 68.9, 158.6, 167.1, 170.2; IR (neat) ν = 3321, 2919, 2123, 1654, 1537, 1509, 1452; HRMS (ESI) m/z calcd for $\text{C}_{27}\text{H}_{40}\text{N}_3\text{O}_2$ 438.3115; found [M + H] $^+$ 438.3131.

Formyl-Adm-Adm-Adm-NH-*i*-Pr (1)

Peptide **1** was synthesized from isocyanide **7a** (1.18 g, 2.8 mmol) according to the procedure described for the synthesis of formamide **4a**. After 24 h, solid ammonium formate (1 eq) was freshly added to the reaction mixture, which was worked up after 48 h, when complete disappearance of **7a** was observed by TLC. After evaporation of the volatiles, the residue was purified by column chromatography using 30% EtOAc in hexanes as eluent. Evaporation of the collected fractions gave HCO-Adm-Adm-Adm-NH-*i*-Pr (**1**, 1.06 g, 61%) as white powder: R_f 0.24 (7:3 Hex:EtOAc); m.p.> 250 °C; ^1H NMR (500 MHz, CDCl_3) δ 1.14 (d, J = 6.5 Hz, 6H), 1.63-1.72 (m, 16H), 1.76-1.78 (m, 5H), 1.83 (m, 3H), 1.90-1.93 (m, 4H), 1.95-1.97 (m, 6H), 2.08-2.11 (m,

2H), 2.63-2.66 (m, 6H), 3.91-3.98 (m, 1H), 5.57 (s, 1H), 7.00 (s, 1H), 7.01 (s, 1H), 7.17 (s, 1H), 8.10 (d, J = 2.0 Hz, 1H); ^{13}C NMR (quantitative analysis, 700 MHz, CDCl_3) δ 22.7 (2C), 26.4, 26.7 (3C), 26.9, 27.0, 32.2 (2C), 32.4 (2C), 32.6 (2C), 32.8 (2C), 33.1 (4C), 34.2 (6C), 37.3, 37.5, 37.7, 41.4, 64.3, 65.5, 65.8, 162.3, 170.9, 172.1, 173.3; IR (neat) v: 3309, 2909, 2860, 1681, 1647, 1525, 1497, 1470; HRMS (ESI) m/z calcd for $\text{C}_{37}\text{H}_{55}\text{N}_4\text{O}_4$ 619.4218; found $[\text{M} + \text{H}]^+$ 619.4226.

Formyl-Adm-Adm-Adm-NH-*t*-Bu (**2**)

Peptide **2** was synthesized from isonitrile **7b** (0.6 g, 1.37 mmol) according to a modified procedure as that described for the synthesis of formamide **4a**. The mixture was stirred for 48h. Each day freshly made sat. solution of $\text{NH}_4^+\text{HCO}_2^-$ (1eq) was added to the mixture. The volatiles were removed after completion of the reaction. The white solid was partitioned between H_2O and CH_2Cl_2 . The organic layer was washed with brine and dried over Na_2SO_4 . Evaporation of the volatiles gave a white solid which was purified by column chromatography using 30% EtOAc in hexanes as eluent. Evaporation of the collected fractions gave HCO-Adm-Adm-Adm-NH-*t*-Bu (**2**, 0.52, 61%) as white powder: R_f 0.32 (7:3 hexanes:EtOAc); m.p. > 250 °C; ^1H NMR (500 MHz, CDCl_3) δ 1.31 (s, 9H), 1.58 (m, 1H), 1.61-1.62 (m, 2H), 1.66-1.69 (m, 12H), 1.73 (m, 2H), 1.77-1.82 (m, 6H), 1.89-1.97 (m, 11H), 2.05-2.08 (m, 2H), 2.63-2.65 (m, 6H), 5.59 (s, 1H), 6.69 (s, 1H), 7.07 (s, 1H), 7.16 (s, 1H), 8.10 (d, J = 2.0 Hz, 1H); ^{13}C NMR (700 MHz, CDCl_3) δ 26.3, 26.6 (3C), 26.9 (2C), 28.7 (3C), 32.2 (2C), 32.5 (2C), 32.6 (2C), 32.7 (2C), 33.1 (4C), 34.1 (4C), 34.2 (2C), 37.3, 37.5, 37.7, 50.8, 65.0 (2C), 65.7, 162.1, 170.9, 171.4, 172.9; IR (neat) v: 2907, 2857, 1673, 1513, 749; HRMS (ESI) m/z calcd for $\text{C}_{38}\text{H}_{57}\text{N}_4\text{O}_4$ 633.4374; found $[\text{M} + \text{H}]^+$ 633.4394.

X-RAY DIFFRACTION

Tripeptide **1** was crystallized by dissolving 3-4 mg of peptide **1** in 1 mL of acetone and equilibrating hexane vapour into the mother liquor in a closed container to provide slow formation of crystals. Crystals of **1** were also obtained from other solvent systems: acetone with drops of CHCl_3 , large blocks; acetone, needles; EtOAc/Acetone/ CHCl_3 , needles. Crystals of peptide **2** were

similarly grown from a mixture of acetone/EtOAc and from DMSO. X-Ray diffraction data collection was performed at the Laboratoire de Diffraction des Rayons X de l'Université de Montréal with a Bruker Venture Metaljet diffractometer, equipped with a Gallium Liquid Metal Jet Source (Ga K α radiation, $\lambda = 1.34139 \text{ \AA}$), Helios MX Mirror Optics, a kappa goniometer, and a Photon 100 CMOS detector. Data collection, data reduction, and absorption correction were achieved by use of the APEX 2, SAINT, and SADABS software packages (Bruker AXS). Using Olex2 [S1], the structures were solved with the ShelXT [S2] structure solution program using Intrinsic Phasing, and refined by full-matrix least-squares procedures on F², using all data, with the XL [S3] refinement package. CCDC 1906513 – 1906517 contain the supplementary crystallographic data for this paper. The data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/structures.

Peptide 1 (LUB118)

from acetone

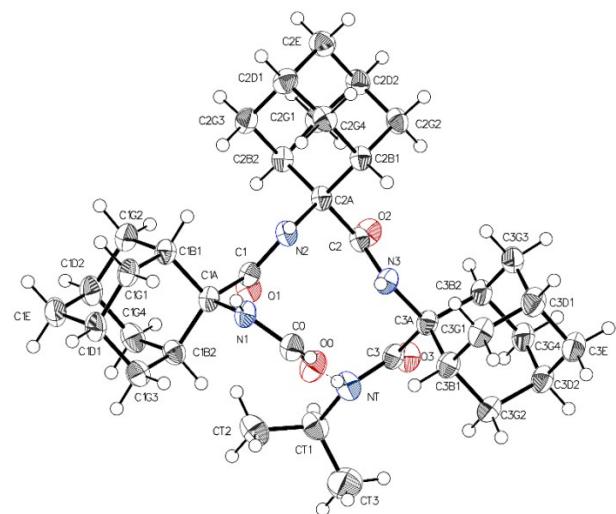


Table 1. Crystal data and structure refinement for lub118.

Identification code	lub118
Empirical formula	C ₃₇ H ₅₄ N ₄ O ₄
Formula weight	618.84

Temperature/K	150
Crystal system	monoclinic
Space group	P2 ₁ /c
a/Å	14.6491(6)
b/Å	14.7140(6)
c/Å	16.1495(7)
α/°	90
β/°	109.618(2)
γ/°	90
Volume/Å ³	3278.9(2)
Z	4
ρ _{calc} g/cm ³	1.254
μ/mm ⁻¹	0.413
F(000)	1344.0
Crystal size/mm ³	0.138 × 0.043 × 0.04
Radiation	GaKα (λ = 1.34139)
2Θ range for data collection/°	5.572 to 110.142
Index ranges	-17 ≤ h ≤ 17, -17 ≤ k ≤ 17, -19 ≤ l ≤ 19
Reflections collected	38793
Independent reflections	6204 [R _{int} = 0.0774, R _{sigma} = 0.0613]
Data/restraints/parameters	6204/0/409
Goodness-of-fit on F ²	1.020
Final R indexes [I>=2σ (I)]	R ₁ = 0.0577, wR ₂ = 0.1226
Final R indexes [all data]	R ₁ = 0.1106, wR ₂ = 0.1466
Largest diff. peak/hole / e Å ⁻³	0.26/-0.23
CCDC deposition No.	1906513

Table 2. Hydrogen Bonds for lub118.

D	H	A	d(D-H)/Å	d(H-A)/Å	d(D-A)/Å	D-H-A/°
N1 H1 O2 ¹			0.88	1.98	2.745 (2)	143.8
N1 H1 O3 ¹			0.88	2.61	3.181 (3)	123.1
N3 H3 O0			0.88	2.61	3.117 (3)	117.4
NTHTO0			0.88	2.22	3.095 (3)	177.6

¹ x, 1/2-y, 1/2+z

Table 3. Torsion Angles for lub118.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
C0	N1	C1A	C1B1	-170.2 (2)	C2B2	C2A	C2B1	C2G1	60.3 (2)
C0	N1	C1A	C1B2	71.7 (3)	C2B2	C2A	C2B1	C2G2	-59.4 (2)
C0	N1	C1A	C1	-52.4 (3)	C2B2	C2A	C2	O2	16.2 (3)
O0	C0	N1	C1A	5.2 (4)	C2B2	C2A	C2	N3	-169.11 (19)
N1	C1A	C1B1	C1G1	-58.6 (2)	C2B2	C2G3	C2D1	C2G1	-60.6 (2)
N1	C1A	C1B1	C1G2	-178.08 (19)	C2B2	C2G3	C2D1	C2E	58.8 (3)
N1	C1A	C1B2	C1G3	58.0 (2)	C2B2	C2G4	C2D2	C2G2	57.3 (3)
N1	C1A	C1B2	C1G4	177.16 (19)	C2B2	C2G4	C2D2	C2E	-62.8 (3)
N1	C1A	C1	O1	132.8 (2)	C2G1	C2B1	C2G2	C2D2	-58.5 (2)
N1	C1A	C1	N2	-49.0 (3)	C2G1	C2D1	C2E	C2D2	61.5 (3)
C1A	C1B1	C1G1	C1D1	-59.0 (2)	C2G2	C2B1	C2G1	C2D1	59.7 (2)
C1A	C1B1	C1G2	C1D2	62.3 (3)	C2G2	C2D2	C2E	C2D1	-60.3 (3)
C1A	C1B2	C1G3	C1D1	61.7 (2)	C2G3	C2B2	C2G4	C2D2	61.7 (2)
C1A	C1B2	C1G4	C1D2	-59.0 (3)	C2G3	C2D1	C2E	C2D2	-57.8 (3)
C1A	C1	N2	C2A	-169.76 (19)	C2G4	C2B2	C2G3	C2D1	-59.6 (2)
C1B1	C1A	C1B2	C1G3	-61.0 (2)	C2G4	C2D2	C2E	C2D1	59.8 (3)
C1B1	C1A	C1B2	C1G4	58.2 (2)	C2	C2A	C2B1	C2G1	-176.76 (19)
C1B1	C1A	C1	O1	-109.3 (3)	C2	C2A	C2B1	C2G2	63.6 (2)
C1B1	C1A	C1	N2	68.9 (2)	C2	C2A	C2B2	C2G3	179.30 (19)
C1B1	C1G1	C1D1	C1G3	57.7 (3)	C2	C2A	C2B2	C2G4	-62.4 (2)
C1B1	C1G1	C1D1	C1E	-62.4 (3)	C2	N3	C3A	C3B1	-178.1 (2)
C1B1	C1G2	C1D2	C1G4	-61.9 (3)	C2	N3	C3A	C3B2	64.2 (3)
C1B1	C1G2	C1D2	C1E	58.3 (3)	C2	N3	C3A	C3	-59.7 (3)
C1B2	C1A	C1B1	C1G1	60.7 (2)	O2	C2	N3	C3A	-2.3 (3)
C1B2	C1A	C1B1	C1G2	-58.8 (2)	N3	C3A	C3B1	C3G1	-61.7 (2)
C1B2	C1A	C1	O1	11.5 (3)	N3	C3A	C3B1	C3G2	178.84 (19)
C1B2	C1A	C1	N2	-170.23 (19)	N3	C3A	C3B2	C3G3	58.4 (3)
C1B2	C1G3	C1D1	C1G1	-60.1 (3)	N3	C3A	C3B2	C3G4	177.59 (18)
C1B2	C1G3	C1D1	C1E	59.7 (3)	N3	C3A	C3	O3	121.5 (2)
C1B2	C1G4	C1D2	C1G2	59.2 (3)	N3	C3A	C3	NT	-59.6 (3)
C1B2	C1G4	C1D2	C1E	-61.3 (3)	C3A	C3B1	C3G1	C3D1	-60.5 (3)
C1G1	C1B1	C1G2	C1D2	-58.5 (3)	C3A	C3B1	C3G2	C3D2	62.0 (3)
C1G1	C1D1	C1E	C1D2	61.0 (3)	C3A	C3B2	C3G3	C3D1	62.3 (3)
C1G2	C1B1	C1G1	C1D1	60.7 (3)	C3A	C3B2	C3G4	C3D2	-59.8 (3)
C1G2	C1D2	C1E	C1D1	-59.0 (3)	C3A	C3	NT	CT1	-172.4 (2)
C1G3	C1B2	C1G4	C1D2	60.1 (3)	C3B1	C3A	C3B2	C3G3	-59.7 (2)
C1G3	C1D1	C1E	C1D2	-59.3 (3)	C3B1	C3A	C3B2	C3G4	59.5 (2)
C1G4	C1B2	C1G3	C1D1	-59.6 (3)	C3B1	C3A	C3	O3	-121.4 (2)

C1G4	C1D2	C1E	C1D1	60.4 (3)	C3B1	C3A	C3	NT	57.5 (3)
C1	C1A	C1B1	C1G1	-174.75 (18)	C3B1	C3G1	C3D1	C3G3	59.7 (3)
C1	C1A	C1B1	C1G2	65.7 (2)	C3B1	C3G1	C3D1	C3E	-60.8 (3)
C1	C1A	C1B2	C1G3	177.53 (19)	C3B1	C3G2	C3D2	C3G4	-60.4 (3)
C1	C1A	C1B2	C1G4	-63.3 (3)	C3B1	C3G2	C3D2	C3E	59.7 (3)
C1	N2	C2A	C2B1	-177.24 (19)	C3B2	C3A	C3B1	C3G1	59.2 (2)
C1	N2	C2A	C2B2	64.8 (3)	C3B2	C3A	C3B1	C3G2	-60.3 (2)
C1	N2	C2A	C2	-59.3 (3)	C3B2	C3A	C3	O3	-2.4 (3)
O1	C1	N2	C2A	8.6 (3)	C3B2	C3A	C3	NT	176.5 (2)
N2	C2A	C2B1	C2G1	-59.6 (2)	C3B2	C3G3	C3D1	C3G1	-60.8 (3)
N2	C2A	C2B1	C2G2	-179.28 (18)	C3B2	C3G3	C3D1	C3E	58.6 (3)
N2	C2A	C2B2	C2G3	58.3 (2)	C3B2	C3G4	C3D2	C3G2	58.6 (3)
N2	C2A	C2B2	C2G4	176.62 (18)	C3B2	C3G4	C3D2	C3E	-61.5 (3)
N2	C2A	C2	O2	138.8 (2)	C3G1	C3B1	C3G2	C3D2	-58.6 (3)
N2	C2A	C2	N3	-46.5 (2)	C3G1	C3D1	C3E	C3D2	60.7 (3)
C2A	C2B1	C2G1	C2D1	-60.2 (3)	C3G2	C3B1	C3G1	C3D1	59.4 (3)
C2A	C2B1	C2G2	C2D2	62.4 (2)	C3G2	C3D2	C3E	C3D1	-60.2 (3)
C2A	C2B2	C2G3	C2D1	61.9 (2)	C3G3	C3B2	C3G4	C3D2	60.3 (3)
C2A	C2B2	C2G4	C2D2	-57.5 (3)	C3G3	C3D1	C3E	C3D2	-58.2 (3)
C2A	C2	N3	C3A	-177.04 (19)	C3G4	C3B2	C3G3	C3D1	-58.5 (3)
C2B1	C2A	C2B2	C2G3	-60.3 (2)	C3G4	C3D2	C3E	C3D1	59.4 (3)
C2B1	C2A	C2B2	C2G4	58.0 (2)	C3	C3A	C3B1	C3G1	-178.82 (19)
C2B1	C2A	C2	O2	-103.4 (2)	C3	C3A	C3B1	C3G2	61.7 (2)
C2B1	C2A	C2	N3	71.3 (2)	C3	C3A	C3B2	C3G3	179.8 (2)
C2B1	C2G1	C2D1	C2G3	58.7 (3)	C3	C3A	C3B2	C3G4	-61.0 (3)
C2B1	C2G1	C2D1	C2E	-61.4 (3)	C3	NT	CT1	CT2	-153.7 (2)
C2B1	C2G2	C2D2	C2G4	-60.8 (2)	C3	NT	CT1	CT3	82.0 (3)
C2B1	C2G2	C2D2	C2E	59.1 (3)	O3	C3	NT	CT1	6.6 (4)

Peptide 1 from acetone / CHCl₃

(LUB133)

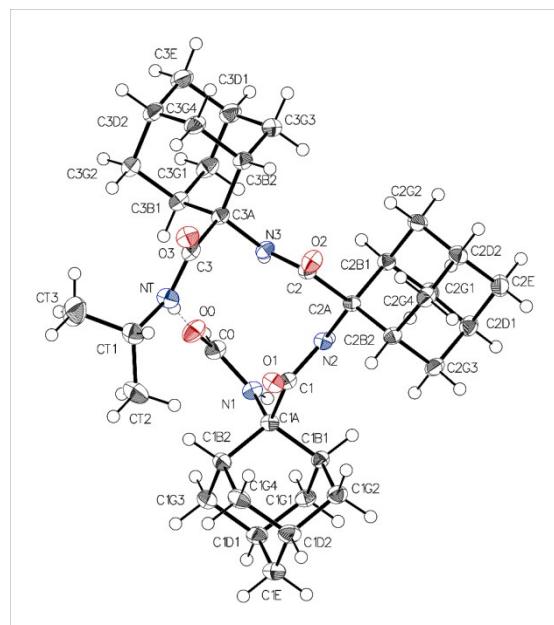


Table 4. Crystal data and structure refinement for lub133.

Identification code	lub133
Empirical formula	C ₃₇ H ₅₄ N ₄ O ₄
Formula weight	618.84
Temperature/K	150
Crystal system	monoclinic
Space group	P2 ₁ /c
a/Å	14.6489(4)
b/Å	14.7284(4)
c/Å	16.1631(4)
α/°	90
β/°	109.7380(10)
γ/°	90
Volume/Å ³	3282.38(15)
Z	4
ρ _{calc} g/cm ³	1.252
μ/mm ⁻¹	0.410
F(000)	1344.0
Crystal size/mm ³	0.19 × 0.19 × 0.11
Radiation	GaKα (λ = 1.34139)
2Θ range for data collection/°	5.576 to 121.43
Index ranges	-19 ≤ h ≤ 19, -19 ≤ k ≤ 19, -21 ≤ l ≤ 20

Reflections collected	49304
Independent reflections	7528 [$R_{\text{int}} = 0.0317$, $R_{\text{sigma}} = 0.0191$]
Data/restraints/parameters	7528/0/409
Goodness-of-fit on F^2	1.043
Final R indexes [$I >= 2\sigma(I)$]	$R_1 = 0.0454$, $wR_2 = 0.1159$
Final R indexes [all data]	$R_1 = 0.0499$, $wR_2 = 0.1205$
Largest diff. peak/hole / e Å ⁻³	0.38/-0.36
CCDC deposition No.	1906514

Table 5. Hydrogen Bonds for lub133.

D	H	A	d(D-H)/Å	d(H-A)/Å	d(D-A)/Å	D-H-A/°
N1 H1 O2 ¹			0.88	1.99	2.7452 (12)	143.1
N1 H1 O3 ¹			0.88	2.60	3.1790 (13)	123.8
N3 H3 O0			0.88	2.61	3.1150 (13)	117.6
NTHTO0			0.88	2.22	3.0958 (13)	177.0

¹ x, 1/2-y, 1/2+z

Table 6. Torsion Angles for lub133.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
C0	N1	C1A	C1B1	-170.36 (10)	C2B2	C2A	C2B1	C2G1	60.50 (11)
C0	N1	C1A	C1B2	71.46 (13)	C2B2	C2A	C2B1	C2G2	-59.03 (10)
C0	N1	C1A	C1	-52.75 (13)	C2B2	C2A	C2	O2	16.22 (14)
O0	C0	N1	C1A	5.58 (19)	C2B2	C2A	C2	N3	-168.87 (9)
N1	C1A	C1B1	C1G1	-58.62 (11)	C2B2	C2G3	C2D1	C2G1	-60.09 (12)
N1	C1A	C1B1	C1G2	-178.16 (9)	C2B2	C2G3	C2D1	C2E	59.09 (12)
N1	C1A	C1B2	C1G3	57.91 (11)	C2B2	C2G4	C2D2	C2G2	57.37 (11)
N1	C1A	C1B2	C1G4	176.93 (9)	C2B2	C2G4	C2D2	C2E	-62.94 (12)
N1	C1A	C1	O1	133.05 (11)	C2G1	C2B1	C2G2	C2D2	-58.30 (11)
N1	C1A	C1	N2	-48.55 (11)	C2G1	C2D1	C2E	C2D2	61.53 (12)
C1A	C1B1	C1G1	C1D1	-59.17 (11)	C2G2	C2B1	C2G1	C2D1	59.84 (11)
C1A	C1B1	C1G2	C1D2	62.31 (12)	C2G2	C2D2	C2E	C2D1	-60.33 (12)
C1A	C1B2	C1G3	C1D1	61.66 (12)	C2G3	C2B2	C2G4	C2D2	61.82 (11)
C1A	C1B2	C1G4	C1D2	-58.83 (12)	C2G3	C2D1	C2E	C2D2	-58.02 (12)
C1A	C1	N2	C2A	-169.77 (9)	C2G4	C2B2	C2G3	C2D1	-59.91 (11)
C1B1	C1A	C1B2	C1G3	-61.06 (11)	C2G4	C2D2	C2E	C2D1	59.92 (12)
C1B1	C1A	C1B2	C1G4	57.95 (11)	C2	C2A	C2B1	C2G1	-176.81 (9)
C1B1	C1A	C1	O1	-109.16 (12)	C2	C2A	C2B1	C2G2	63.66 (11)
C1B1	C1A	C1	N2	69.24 (11)	C2	C2A	C2B2	C2G3	179.47 (9)
C1B1	C1G1	C1D1	C1G3	57.83 (12)	C2	C2A	C2B2	C2G4	-62.26 (11)
C1B1	C1G1	C1D1	C1E	-62.06 (12)	C2	N3	C3A	C3B1	-178.27 (9)
C1B1	C1G2	C1D2	C1G4	-61.88 (12)	C2	N3	C3A	C3B2	63.83 (13)
C1B1	C1G2	C1D2	C1E	58.10 (12)	C2	N3	C3A	C3	-59.51 (13)

C1B2 C1A	C1B1 C1G1	60.78 (11)	O2	C2	N3	C3A	-2.41 (16)
C1B2 C1A	C1B1 C1G2	-58.75 (11)	N3	C3A	C3B1	C3G1	-61.43 (11)
C1B2 C1A	C1 O1	11.47 (15)	N3	C3A	C3B1	C3G2	178.74 (9)
C1B2 C1A	C1 N2	-170.13 (9)	N3	C3A	C3B2	C3G3	58.64 (12)
C1B2 C1G3 C1D1 C1G1		-59.85 (12)	N3	C3A	C3B2	C3G4	177.52 (9)
C1B2 C1G3 C1D1 C1E		59.81 (12)	N3	C3A	C3	O3	121.85 (11)
C1B2 C1G4 C1D2 C1G2		58.94 (12)	N3	C3A	C3	NT	-60.26 (12)
C1B2 C1G4 C1D2 C1E		-61.47 (12)	C3A	C3B1	C3G1	C3D1	-60.33 (12)
C1G1 C1B1 C1G2 C1D2		-58.29 (12)	C3A	C3B1	C3G2	C3D2	62.14 (12)
C1G1 C1D1 C1E	C1D2	60.70 (13)	C3A	C3B2	C3G3	C3D1	61.54 (12)
C1G2 C1B1 C1G1 C1D1		60.55 (12)	C3A	C3B2	C3G4	C3D2	-59.29 (12)
C1G2 C1D2 C1E	C1D1	-58.88 (13)	C3A	C3	NT	CT1	-172.63 (10)
C1G3 C1B2 C1G4 C1D2		60.32 (12)	C3B1	C3A	C3B2	C3G3	-59.54 (11)
C1G3 C1D1 C1E	C1D2	-59.43 (13)	C3B1	C3A	C3B2	C3G4	59.33 (12)
C1G4 C1B2 C1G3 C1D1		-59.76 (12)	C3B1	C3A	C3	O3	-120.78 (11)
C1G4 C1D2 C1E	C1D1	60.51 (13)	C3B1	C3A	C3	NT	57.10 (12)
C1 C1A C1B1 C1G1		-174.85 (9)	C3B1	C3G1	C3D1	C3G3	59.35 (12)
C1 C1A C1B1 C1G2		65.61 (11)	C3B1	C3G1	C3D1	C3E	-61.22 (13)
C1 C1A C1B2 C1G3		177.74 (9)	C3B1	C3G2	C3D2	C3G4	-60.19 (12)
C1 C1A C1B2 C1G4		-63.24 (12)	C3B1	C3G2	C3D2	C3E	59.72 (12)
C1 N2 C2A C2B1		-177.23 (9)	C3B2	C3A	C3B1	C3G1	59.29 (11)
C1 N2 C2A C2B2		64.61 (12)	C3B2	C3A	C3B1	C3G2	-60.54 (11)
C1 N2 C2A C2		-59.59 (12)	C3B2	C3A	C3	O3	-1.46 (15)
O1 C1 N2 C2A		8.68 (16)	C3B2	C3A	C3	NT	176.43 (9)
N2 C2A C2B1 C2G1		-59.67 (11)	C3B2	C3G3	C3D1	C3G1	-60.32 (12)
N2 C2A C2B1 C2G2		-179.20 (8)	C3B2	C3G3	C3D1	C3E	59.28 (13)
N2 C2A C2B2 C2G3		58.25 (11)	C3B2	C3G4	C3D2	C3G2	58.37 (12)
N2 C2A C2B2 C2G4		176.52 (9)	C3B2	C3G4	C3D2	C3E	-61.76 (12)
N2 C2A C2 O2		139.12 (10)	C3G1	C3B1	C3G2	C3D2	-58.86 (12)
N2 C2A C2 N3		-45.96 (12)	C3G1	C3D1	C3E	C3D2	60.84 (13)
C2A C2B1 C2G1 C2D1		-59.96 (12)	C3G2	C3B1	C3G1	C3D1	59.92 (12)
C2A C2B1 C2G2 C2D2		62.30 (11)	C3G2	C3D2	C3E	C3D1	-60.29 (13)
C2A C2B2 C2G3 C2D1		61.62 (11)	C3G3	C3B2	C3G4	C3D2	60.79 (12)
C2A C2B2 C2G4 C2D2		-57.39 (12)	C3G3	C3D1	C3E	C3D2	-58.52 (13)
C2A C2 N3 C3A		-177.35 (9)	C3G4	C3B2	C3G3	C3D1	-59.13 (12)
C2B1 C2A C2B2 C2G3		-60.56 (10)	C3G4	C3D2	C3E	C3D1	59.44 (13)
C2B1 C2A C2B2 C2G4		57.71 (11)	C3	C3A	C3B1	C3G1	-178.68 (9)
C2B1 C2A C2 O2		-103.17 (11)	C3	C3A	C3B1	C3G2	61.48 (11)
C2B1 C2A C2 N3		71.74 (11)	C3	C3A	C3B2	C3G3	179.59 (9)
C2B1 C2G1 C2D1 C2G3		58.46 (11)	C3	C3A	C3B2	C3G4	-61.53 (12)
C2B1 C2G1 C2D1 C2E		-61.66 (12)	C3	NT	CT1	CT2	-153.50 (12)
C2B1 C2G2 C2D2 C2G4		-61.01 (12)	C3	NT	CT1	CT3	82.71 (16)
C2B1 C2G2 C2D2 C2E		59.10 (12)	O3	C3	NT	CT1	5.27 (18)

Peptide 1 from acetone / EtOAc / CHCl₃

[LUB133(3)]

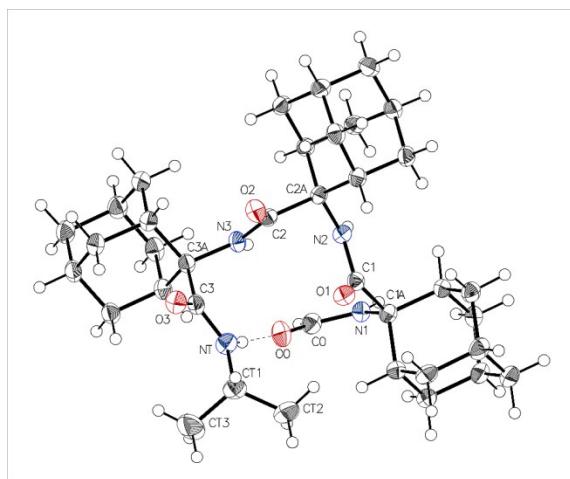


Table 7. Crystal data and structure refinement for lub133(3).

Identification code	lub133(3)
Empirical formula	C ₃₇ H ₅₄ N ₄ O ₄
Formula weight	618.84
Temperature/K	150
Crystal system	monoclinic
Space group	P2 ₁ /c
a/Å	14.6463(6)
b/Å	14.7404(6)
c/Å	16.1538(7)
α/°	90
β/°	109.735(2)
γ/°	90
Volume/Å ³	3282.6(2)
Z	4
ρ _{calc} g/cm ³	1.252
μ/mm ⁻¹	0.410
F(000)	1344.0
Crystal size/mm ³	0.35 × 0.06 × 0.03
Radiation	GaKα (λ = 1.34139)
2Θ range for data collection/°	5.578 to 114.496
Index ranges	-17 ≤ h ≤ 18, -18 ≤ k ≤ 18, -18 ≤ l ≤ 18
Reflections collected	31865
Independent reflections	6452 [R _{int} = 0.0454, R _{sigma} = 0.0341]
Data/restraints/parameters	6452/0/409
Goodness-of-fit on F ²	1.057

Final R indexes [$I \geq 2\sigma(I)$] $R_1 = 0.0469$, $wR_2 = 0.1017$
 Final R indexes [all data] $R_1 = 0.0655$, $wR_2 = 0.1134$
 Largest diff. peak/hole / e Å⁻³ 0.27/-0.33
 CCDC deposition No. 1906515

Table 8. Hydrogen Bonds for lub133(3).

D	H	A	d(D-H)/Å	d(H-A)/Å	d(D-A)/Å	D-H-A/°
N1	H1	O2 ¹	0.88	1.99	2.7448 (18)	143.2
N1	H1	O3 ¹	0.88	2.61	3.1839 (19)	123.8
N3	H3	O0	0.88	2.61	3.1147 (19)	117.7
NTHTO0			0.88	2.21	3.094 (2)	177.2

¹ x, 1/2-y, 1/2+z

Table 9. Torsion Angles for lub133(3).

A	B	C	D	Angle/°	A	B	C	D	Angle/°
C0	N1	C1A	C1B1	-170.35 (15)	C2B2	C2A	C2B1	C2G1	60.56 (17)
C0	N1	C1A	C1B2	71.64 (19)	C2B2	C2A	C2B1	C2G2	-59.24 (16)
C0	N1	C1A	C1	-52.77 (19)	C2B2	C2A	C2	O2	16.2 (2)
O0	C0	N1	C1A	5.4 (3)	C2B2	C2A	C2	N3	-168.90 (13)
N1	C1A	C1B1	C1G1	-58.55 (17)	C2B2	C2G3	C2D1	C2G1	-60.18 (18)
N1	C1A	C1B1	C1G2	-178.10 (13)	C2B2	C2G3	C2D1	C2E	59.32 (18)
N1	C1A	C1B2	C1G3	57.78 (17)	C2B2	C2G4	C2D2	C2G2	57.25 (18)
N1	C1A	C1B2	C1G4	177.14 (13)	C2B2	C2G4	C2D2	C2E	-63.01 (17)
N1	C1A	C1	O1	133.00 (16)	C2G1	C2B1	C2G2	C2D2	-58.42 (17)
N1	C1A	C1	N2	-48.77 (17)	C2G1	C2D1	C2E	C2D2	61.49 (18)
C1A	C1B1	C1G1	C1D1	-59.25 (17)	C2G2	C2B1	C2G1	C2D1	59.79 (17)
C1A	C1B1	C1G2	C1D2	62.23 (18)	C2G2	C2D2	C2E	C2D1	-60.32 (18)
C1A	C1B2	C1G3	C1D1	61.87 (17)	C2G3	C2B2	C2G4	C2D2	61.98 (17)
C1A	C1B2	C1G4	C1D2	-58.84 (18)	C2G3	C2D1	C2E	C2D2	-58.17 (18)
C1A	C1	N2	C2A	-169.85 (13)	C2G4	C2B2	C2G3	C2D1	-59.95 (17)
C1B1	C1A	C1B2	C1G3	-61.28 (17)	C2G4	C2D2	C2E	C2D1	59.94 (18)
C1B1	C1A	C1B2	C1G4	58.08 (18)	C2	C2A	C2B1	C2G1	-176.72 (13)
C1B1	C1A	C1	O1	-109.08 (17)	C2	C2A	C2B1	C2G2	63.48 (16)
C1B1	C1A	C1	N2	69.14 (17)	C2	C2A	C2B2	C2G3	179.43 (13)
C1B1	C1G1	C1D1	C1G3	57.94 (18)	C2	C2A	C2B2	C2G4	-62.28 (17)
C1B1	C1G1	C1D1	C1E	-62.17 (17)	C2	N3	C3A	C3B1	-178.39 (14)
C1B1	C1G2	C1D2	C1G4	-61.66 (19)	C2	N3	C3A	C3B2	63.84 (19)
C1B1	C1G2	C1D2	C1E	58.28 (18)	C2	N3	C3A	C3	-59.58 (19)
C1B2	C1A	C1B1	C1G1	60.79 (17)	O2	C2	N3	C3A	-2.4 (2)
C1B2	C1A	C1B1	C1G2	-58.76 (17)	N3	C3A	C3B1	C3G1	-61.40 (17)

C1B2 C1A C1 O1	11.4 (2)	N3	C3A	C3B1	C3G2	178.76 (13)
C1B2 C1A C1 N2	-170.33 (13)	N3	C3A	C3B2	C3G3	58.53 (17)
C1B2 C1G3 C1D1 C1G1	-59.91 (19)	N3	C3A	C3B2	C3G4	177.54 (14)
C1B2 C1G3 C1D1 C1E	59.81 (18)	N3	C3A	C3	O3	121.89 (16)
C1B2 C1G4 C1D2 C1G2	58.78 (19)	N3	C3A	C3	NT	-60.22 (18)
C1B2 C1G4 C1D2 C1E	-61.62 (18)	C3A	C3B1	C3G1	C3D1	-60.35 (18)
C1G1 C1B1 C1G2 C1D2	-58.41 (18)	C3A	C3B1	C3G2	C3D2	62.25 (18)
C1G1 C1D1 C1E C1D2	60.89 (19)	C3A	C3B2	C3G3	C3D1	61.88 (18)
C1G2 C1B1 C1G1 C1D1	60.56 (17)	C3A	C3B2	C3G4	C3D2	-59.36 (18)
C1G2 C1D2 C1E C1D1	-59.15 (19)	C3A	C3	NT	CT1	-172.47 (15)
C1G3 C1B2 C1G4 C1D2	60.76 (18)	C3B1	C3A	C3B2	C3G3	-59.67 (17)
C1G3 C1D1 C1E C1D2	-59.37 (19)	C3B1	C3A	C3B2	C3G4	59.34 (18)
C1G4 C1B2 C1G3 C1D1	-59.91 (18)	C3B1	C3A	C3	O3	-120.65 (17)
C1G4 C1D2 C1E C1D1	60.39 (19)	C3B1	C3A	C3	NT	57.24 (18)
C1 C1A C1B1 C1G1	-174.64 (13)	C3B1	C3G1	C3D1	C3G3	59.42 (19)
C1 C1A C1B1 C1G2	65.81 (16)	C3B1	C3G1	C3D1	C3E	-61.25 (19)
C1 C1A C1B2 C1G3	177.51 (13)	C3B1	C3G2	C3D2	C3G4	-60.22 (18)
C1 C1A C1B2 C1G4	-63.13 (18)	C3B1	C3G2	C3D2	C3E	59.69 (18)
C1 N2 C2A C2B1	-177.32 (13)	C3B2	C3A	C3B1	C3G1	59.26 (17)
C1 N2 C2A C2B2	64.82 (18)	C3B2	C3A	C3B1	C3G2	-60.58 (17)
C1 N2 C2A C2	-59.46 (18)	C3B2	C3A	C3	O3	-1.4 (2)
O1 C1 N2 C2A	8.4 (2)	C3B2	C3A	C3	NT	176.48 (14)
N2 C2A C2B1 C2G1	-59.47 (17)	C3B2	C3G3	C3D1	C3G1	-60.51 (19)
N2 C2A C2B1 C2G2	-179.26 (12)	C3B2	C3G3	C3D1	C3E	58.96 (19)
N2 C2A C2B2 C2G3	58.18 (17)	C3B2	C3G4	C3D2	C3G2	58.42 (18)
N2 C2A C2B2 C2G4	176.47 (13)	C3B2	C3G4	C3D2	C3E	-61.78 (18)
N2 C2A C2 O2	139.09 (15)	C3G1	C3B1	C3G2	C3D2	-58.80 (18)
N2 C2A C2 N3	-45.97 (17)	C3G1	C3D1	C3E	C3D2	60.9 (2)
C2A C2B1 C2G1 C2D1	-60.23 (18)	C3G2	C3B1	C3G1	C3D1	59.95 (18)
C2A C2B1 C2G2 C2D2	62.47 (17)	C3G2	C3D2	C3E	C3D1	-60.37 (19)
C2A C2B2 C2G3 C2D1	61.71 (17)	C3G3	C3B2	C3G4	C3D2	60.79 (18)
C2A C2B2 C2G4 C2D2	-57.36 (18)	C3G3	C3D1	C3E	C3D2	-58.4 (2)
C2A C2 N3 C3A	-177.35 (13)	C3G4	C3B2	C3G3	C3D1	-58.87 (18)
C2B1 C2A C2B2 C2G3	-60.53 (16)	C3G4	C3D2	C3E	C3D1	59.35 (19)
C2B1 C2A C2B2 C2G4	57.76 (17)	C3	C3A	C3B1	C3G1	-178.65 (13)
C2B1 C2A C2 O2	-103.06 (17)	C3	C3A	C3B1	C3G2	61.51 (17)
C2B1 C2A C2 N3	71.87 (17)	C3	C3A	C3B2	C3G3	179.45 (13)
C2B1 C2G1 C2D1 C2G3	58.64 (18)	C3	C3A	C3B2	C3G4	-61.55 (18)
C2B1 C2G1 C2D1 C2E	-61.67 (17)	C3	NT	CT1	CT2	-153.65 (17)
C2B1 C2G2 C2D2 C2G4	-60.90 (18)	C3	NT	CT1	CT3	82.7 (2)
C2B1 C2G2 C2D2 C2E	59.11 (18)	O3	C3	NT	CT1	5.4 (3)

Peptide 2 (lub132)

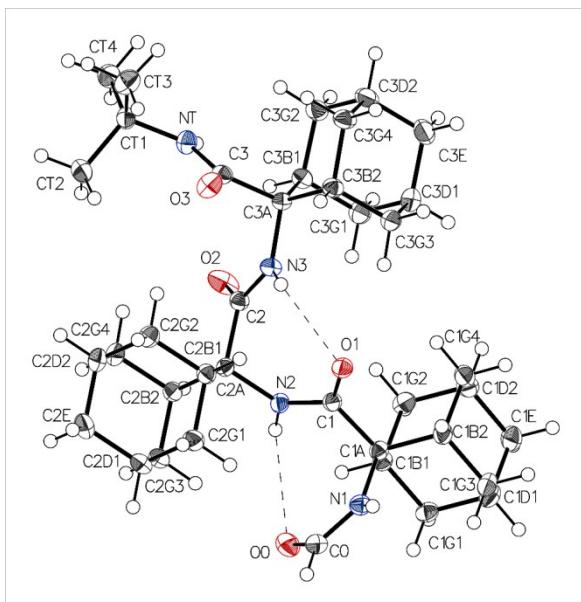


Table 10. Crystal data and structure refinement for lub132.

Identification code	lub132
Empirical formula	C ₃₈ H ₅₆ N ₄ O ₄
Formula weight	632.86
Temperature/K	150
Crystal system	Triclinic
Space group	P-1
a/Å	7.7427(3)
b/Å	10.1000(3)
c/Å	20.7549(7)
α/°	90.0300(10)
β/°	95.7130(10)
γ/°	90.8690(10)
Volume/Å ³	1614.81(10)
Z	2
ρ _{calc} g/cm ³	1.302
μ/mm ⁻¹	0.427
F(000)	688.0
Crystal size/mm ³	0.15 × 0.12 × 0.08
Radiation	GaKα (λ = 1.34139)
2θ range for data collection/°	3.722 to 121.502
Index ranges	-10 ≤ h ≤ 10, -13 ≤ k ≤ 13, -27 ≤ l ≤ 26
Reflections collected	42140
Independent reflections	7412 [R _{int} = 0.0354, R _{sigma} = 0.0242]
Data/restraints/parameters	7412/0/435
Goodness-of-fit on F ²	1.071
Final R indexes [I>=2σ (I)]	R ₁ = 0.0495, wR ₂ = 0.1279
Final R indexes [all data]	R ₁ = 0.0543, wR ₂ = 0.1319

Largest diff. peak/hole / e Å⁻³ 0.42/-0.41
 CCDC deposition No. 1906516

Table 11. Hydrogen Bonds for lub132.

D	H	A	d(D-H)/Å	d(H-A)/Å	d(D-A)/Å	D-H-A/°
N1	H1	O2 ¹	0.95 (2)	1.99 (2)	2.9161 (15)	164.6 (17)
C1B1H1B1O0			1.00	2.37	2.9486 (16)	116.2
C1B2H1B2O2 ¹			1.00	2.66	3.4311 (18)	134.4
N2	H2	O0	0.833 (19)	2.572 (18)	3.1665 (15)	129.4 (15)
C2B1H2B1O1			1.00	2.41	3.0250 (15)	119.2
N3	H3	O1	0.86 (2)	2.31 (2)	2.8804 (14)	124.5 (16)

¹ -1+x, y, z

Table 12. Torsion Angles for lub132.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
C0	N1	C1A	C1B1	-42.33 (17)	C2B2	C2A	C2B1	C2G2	-60.77 (13)
C0	N1	C1A	C1B2	-161.01 (12)	C2B2	C2A	C2	O2	-20.05 (16)
C0	N1	C1A	C1	82.12 (15)	C2B2	C2A	C2	N3	161.76 (11)
O0	C0	N1	C1A	6.7 (2)	C2B2	C2G3	C2D1	C2G1	-59.47 (14)
N1	C1A	C1B1	C1G1	-58.16 (14)	C2B2	C2G3	C2D1	C2E	60.49 (14)
N1	C1A	C1B1	C1G2	-178.06 (10)	C2B2	C2G4	C2D2	C2G2	59.44 (14)
N1	C1A	C1B2	C1G3	62.11 (13)	C2B2	C2G4	C2D2	C2E	-60.74 (14)
N1	C1A	C1B2	C1G4	-177.93 (10)	C2G1	C2B1	C2G2	C2D2	-59.14 (14)
N1	C1A	C1	O1	92.17 (13)	C2G1	C2D1	C2E	C2D2	59.71 (16)
N1	C1A	C1	N2	-84.06 (12)	C2G2	C2B1	C2G1	C2D1	58.78 (14)
C1A	C1B1	C1G1	C1D1	-60.55 (14)	C2G2	C2D2	C2E	C2D1	-60.15 (15)
C1A	C1B1	C1G2	C1D2	61.63 (14)	C2G3	C2B2	C2G4	C2D2	60.30 (14)
C1A	C1B2	C1G3	C1D1	60.84 (14)	C2G3	C2D1	C2E	C2D2	-59.62 (15)
C1A	C1B2	C1G4	C1D2	-60.64 (14)	C2G4	C2B2	C2G3	C2D1	-60.20 (14)
C1A	C1	N2	C2A	171.96 (10)	C2G4	C2D2	C2E	C2D1	59.67 (16)
C1B1	C1A	C1B2	C1G3	-60.20 (13)	C2	C2A	C2B1	C2G1	179.85 (10)
C1B1	C1A	C1B2	C1G4	59.76 (13)	C2	C2A	C2B1	C2G2	60.07 (13)
C1B1	C1A	C1	O1	-143.84 (12)	C2	C2A	C2B2	C2G3	176.77 (10)
C1B1	C1A	C1	N2	39.93 (15)	C2	C2A	C2B2	C2G4	-63.67 (13)
C1B1	C1G1	C1D1	C1G3	59.56 (14)	C2	N3	C3A	C3B1	-47.42 (17)
C1B1	C1G1	C1D1	C1E	-60.61 (14)	C2	N3	C3A	C3B2	-166.18 (12)
C1B1	C1G2	C1D2	C1G4	-60.39 (14)	C2	N3	C3A	C3	80.34 (15)
C1B1	C1G2	C1D2	C1E	59.26 (14)	O2	C2	N3	C3A	13.3 (2)
C1B2	C1A	C1B1	C1G1	59.89 (13)	N3	C3A	C3B1	C3G1	-57.68 (13)
C1B2	C1A	C1B1	C1G2	-60.01 (13)	N3	C3A	C3B1	C3G2	-176.40 (10)

C1B2 C1A C1 O1	-21.92 (16)	N3	C3A	C3B2	C3G3	61.30 (13)
C1B2 C1A C1 N2	161.85 (11)	N3	C3A	C3B2	C3G4	-178.56 (10)
C1B2 C1G3 C1D1 C1G1	-59.56 (15)	N3	C3A	C3	O3	65.25 (14)
C1B2 C1G3 C1D1 C1E	60.04 (14)	N3	C3A	C3	NT	-113.17 (12)
C1B2 C1G4 C1D2 C1G2	59.82 (14)	C3A	C3B1	C3G1	C3D1	-61.27 (14)
C1B2 C1G4 C1D2 C1E	-59.67 (15)	C3A	C3B1	C3G2	C3D2	58.85 (14)
C1G1 C1B1 C1G2 C1D2	-58.74 (14)	C3A	C3B2	C3G3	C3D1	59.64 (14)
C1G1 C1D1 C1E C1D2	60.46 (15)	C3A	C3B2	C3G4	C3D2	-63.13 (14)
C1G2 C1B1 C1G1 C1D1	59.58 (14)	C3A	C3	NT	CT1	174.19 (11)
C1G2 C1D2 C1E C1D1	-60.03 (15)	C3B1	C3A	C3B2	C3G3	-60.70 (13)
C1G3 C1B2 C1G4 C1D2	59.93 (14)	C3B1	C3A	C3B2	C3G4	59.44 (13)
C1G3 C1D1 C1E C1D2	-59.14 (15)	C3B1	C3A	C3	O3	-169.83 (12)
C1G4 C1B2 C1G3 C1D1	-59.93 (14)	C3B1	C3A	C3	NT	11.74 (16)
C1G4 C1D2 C1E C1D1	58.72 (15)	C3B1	C3G1	C3D1	C3G3	59.53 (14)
C1 C1A C1B1 C1G1	-176.66 (10)	C3B1	C3G1	C3D1	C3E	-59.59 (15)
C1 C1A C1B1 C1G2	63.45 (13)	C3B1	C3G2	C3D2	C3G4	-59.11 (14)
C1 C1A C1B2 C1G3	174.20 (10)	C3B1	C3G2	C3D2	C3E	61.21 (14)
C1 C1A C1B2 C1G4	-65.84 (13)	C3B2	C3A	C3B1	C3G1	60.68 (13)
C1 N2 C2A C2B1	-51.10 (16)	C3B2	C3A	C3B1	C3G2	-58.05 (13)
C1 N2 C2A C2B2	-169.21 (11)	C3B2	C3A	C3	O3	-47.88 (15)
C1 N2 C2A C2	74.34 (14)	C3B2	C3A	C3	NT	133.69 (12)
O1 C1 N2 C2A	-4.1 (2)	C3B2	C3G3	C3D1	C3G1	-57.95 (14)
N2 C2A C2B1 C2G1	-59.39 (13)	C3B2	C3G3	C3D1	C3E	61.96 (14)
N2 C2A C2B1 C2G2	-179.17 (10)	C3B2	C3G4	C3D2	C3G2	61.99 (14)
N2 C2A C2B2 C2G3	61.96 (13)	C3B2	C3G4	C3D2	C3E	-57.73 (15)
N2 C2A C2B2 C2G4	-178.48 (10)	C3G1	C3B1	C3G2	C3D2	-60.33 (14)
N2 C2A C2 O2	95.94 (14)	C3G1	C3D1	C3E	C3D2	58.92 (15)
N2 C2A C2 N3	-82.25 (13)	C3G2	C3B1	C3G1	C3D1	59.45 (14)
C2A C2B1 C2G1 C2D1	-61.33 (14)	C3G2	C3D2	C3E	C3D1	-59.72 (15)
C2A C2B1 C2G2 C2D2	61.22 (14)	C3G3	C3B2	C3G4	C3D2	57.92 (14)
C2A C2B2 C2G3 C2D1	60.20 (14)	C3G3	C3D1	C3E	C3D2	-60.73 (15)
C2A C2B2 C2G4 C2D2	-60.92 (14)	C3G4	C3B2	C3G3	C3D1	-60.41 (14)
C2A C2 N3 C3A	-168.55 (11)	C3G4	C3D2	C3E	C3D1	59.09 (15)
C2B1 C2A C2B2 C2G3	-58.84 (13)	C3	C3A	C3B1	C3G1	-177.49 (10)
C2B1 C2A C2B2 C2G4	60.73 (13)	C3	C3A	C3B1	C3G2	63.78 (14)
C2B1 C2A C2 O2	-140.25 (13)	C3	C3A	C3B2	C3G3	171.63 (10)
C2B1 C2A C2 N3	41.56 (15)	C3	C3A	C3B2	C3G4	-68.23 (13)
C2B1 C2G1 C2D1 C2G3	60.56 (14)	C3	NT	CT1	CT2	-70.49 (17)
C2B1 C2G1 C2D1 C2E	-59.54 (14)	C3	NT	CT1	CT3	52.15 (17)
C2B1 C2G2 C2D2 C2G4	-59.80 (15)	C3	NT	CT1	CT4	170.11 (13)
C2B1 C2G2 C2D2 C2E	60.40 (14)	O3	C3	NT	CT1	-4.1 (2)
C2B2 C2A C2B1 C2G1	59.01 (13)					

Peptide 2 bis-DMSO solvate (LUB125)

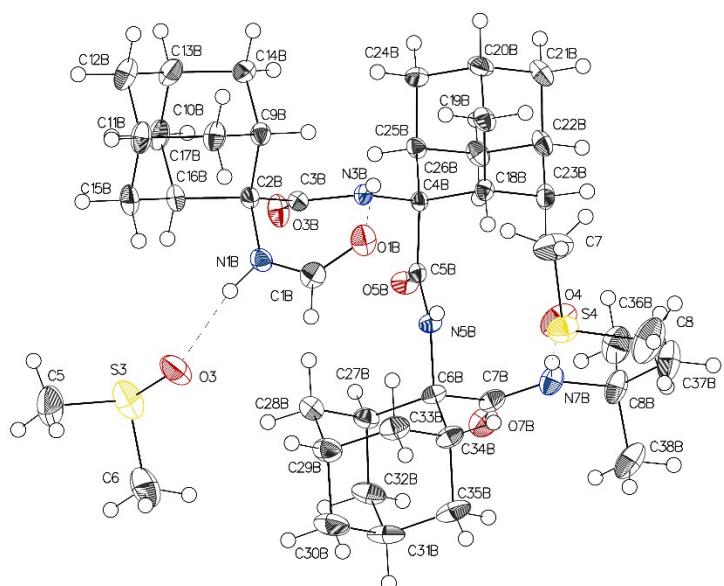
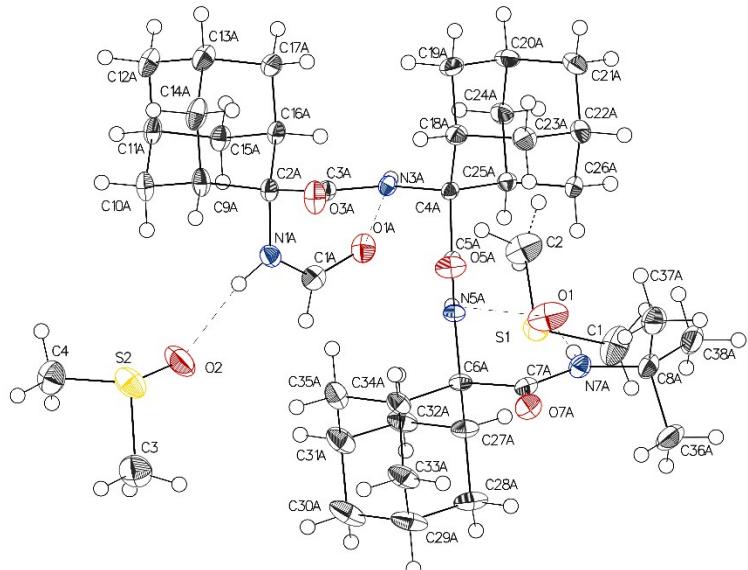


Table 13. Crystal data and structure refinement for LUB125.

Identification code	LUB125
Empirical formula	C ₄₂ H ₆₈ N ₄ O ₆ S ₂
Formula weight	789.12
Temperature/K	150
Crystal system	Orthorhombic
Space group	Pbcn
a/Å	9.8603(3)
b/Å	43.0350(14)
c/Å	38.9591(12)
α/°	90
β/°	90
γ/°	90

Volume/ \AA^3	16531.8(9)
Z	16
ρ_{calc} g/cm 3	1.268
μ/mm^{-1}	1.022
F(000)	6848.0
Crystal size/mm 3	0.38 \times 0.13 \times 0.06
Radiation	GaK α ($\lambda = 1.34139$)
2 Θ range for data collection/ $^\circ$	3.572 to 121.596
Index ranges	-12 \leq h \leq 12, -55 \leq k \leq 55, -50 \leq l \leq 50
Reflections collected	305415
Independent reflections	19053 [$R_{\text{int}} = 0.0575$, $R_{\text{sigma}} = 0.0229$]
Data/restraints/parameters	19053/0/993
Goodness-of-fit on F^2	1.110
Final R indexes [$I \geq 2\sigma(I)$]	$R_1 = 0.0625$, $wR_2 = 0.1585$
Final R indexes [all data]	$R_1 = 0.0715$, $wR_2 = 0.1645$
Largest diff. peak/hole / e \AA^{-3}	0.81/-0.79
CCDC deposition No.	1906517

Table 14. Hydrogen Bonds for LUB125.

D	H	A	d(D-H)/ \AA	d(H-A)/ \AA	d(D-A)/ \AA	D-H-A/ $^\circ$
N1B	H1B	O3	0.88	1.93	2.787 (2)	165.8
N3B	H3B	O1B	0.88	2.12	2.797 (2)	133.4
N5B	H5B	O4	0.88	2.64	3.171 (2)	119.5
N7B	H7B	O4	0.88	2.19	3.072 (2)	176.6
N1A	H1A	O2	0.88	1.96	2.812 (2)	162.4
N3A	H3A	O1A	0.88	2.22	2.846 (2)	127.5
N5A	H5A	O1	0.88	2.45	3.016 (2)	122.5
N7A	H7A	O1	0.88	2.30	3.183 (2)	175.5

Table 15. Torsion Angles for LUB125.

A	B	C	D	Angle/ $^\circ$	A	B	C	D	Angle/ $^\circ$
N1B	C2B	C3B	O3B	108.3 (2)	N1A	C2A	C3A	O3A	-103.1 (2)
N1B	C2B	C3B	N3B	-69.87 (19)	N1A	C2A	C3A	N3A	74.81 (18)
N1B	C2B	C9B	C10B	-57.2 (2)	N1A	C2A	C9A	C10A	-60.5 (2)
N1B	C2B	C9B	C14B	-176.91 (15)	N1A	C2A	C9A	C14A	-179.54 (15)
N1B	C2B	C16B	C15B	60.6 (2)	N1A	C2A	C16A	C15A	57.22 (19)
N1B	C2B	C16B	C17B	179.88 (16)	N1A	C2A	C16A	C17A	177.12 (14)

N3B	C4B	C5B	O5B	112.48(19)	N3A	C4A	C5A	O5A	-114.77(19)
N3B	C4B	C5B	N5B	-66.89(19)	N3A	C4A	C5A	N5A	64.06(19)
N3B	C4B	C18B	C19B	-60.43(18)	N3A	C4A	C18A	C19A	-57.2(2)
N3B	C4B	C18B	C23B	179.72(14)	N3A	C4A	C18A	C23A	-175.57(15)
N3B	C4B	C25B	C24B	57.72(19)	N3A	C4A	C25A	C24A	60.45(18)
N3B	C4B	C25B	C26B	176.25(14)	N3A	C4A	C25A	C26A	-179.66(14)
N5B	C6B	C7B	O7B	-122.2(2)	N5A	C6A	C7A	O7A	125.23(19)
N5B	C6B	C7B	N7B	58.5(2)	N5A	C6A	C7A	N7A	-55.5(2)
N5B	C6B	C27B	C28B	-57.8(2)	N5A	C6A	C27A	C28A	-179.72(17)
N5B	C6B	C27B	C32B	-176.74(16)	N5A	C6A	C27A	C32A	-59.8(2)
N5B	C6B	C34B	C33B	60.1(2)	N5A	C6A	C34A	C33A	176.57(15)
N5B	C6B	C34B	C35B	179.77(17)	N5A	C6A	C34A	C35A	58.0(2)
C1B	N1B	C2B	C3B	74.6(2)	C1A	N1A	C2A	C3A	-75.4(2)
C1B	N1B	C2B	C9B	-48.8(3)	C1A	N1A	C2A	C9A	166.47(18)
C1B	N1B	C2B	C16B	-166.51(19)	C1A	N1A	C2A	C16A	48.6(2)
C2B	N1B	C1B	O1B	-3.1(3)	C2A	N1A	C1A	O1A	4.0(3)
C2B	C9B	C10B	C11B	-61.2(2)	C2A	C9A	C10A	C11A	-60.7(2)
C2B	C9B	C14B	C13B	61.8(2)	C2A	C9A	C14A	C13A	60.7(2)
C2B	C16B	C17B	C13B	-60.0(2)	C2A	C16A	C17A	C13A	-61.43(19)
C3B	N3B	C4B	C5B	-60.0(2)	C3A	N3A	C4A	C5A	59.8(2)
C3B	N3B	C4B	C18B	-179.87(16)	C3A	N3A	C4A	C18A	-62.5(2)
C3B	N3B	C4B	C25B	62.1(2)	C3A	N3A	C4A	C25A	179.93(16)
C3B	C2B	C9B	C10B	-176.44(15)	C3A	C2A	C9A	C10A	-176.02(16)
C3B	C2B	C9B	C14B	63.80(19)	C3A	C2A	C9A	C14A	65.0(2)
C3B	C2B	C16B	C15B	175.74(16)	C3A	C2A	C16A	C15A	177.27(15)
C3B	C2B	C16B	C17B	-65.0(2)	C3A	C2A	C16A	C17A	-62.83(19)
C4B	N3B	C3B	O3B	-3.2(3)	C4A	N3A	C3A	O3A	3.3(3)
C4B	N3B	C3B	C2B	175.00(15)	C4A	N3A	C3A	C2A	-174.54(15)
C4B	C18B	C19B	C20B	-60.8(2)	C4A	C18A	C19A	C20A	-60.5(2)
C4B	C18B	C23B	C22B	61.76(19)	C4A	C18A	C23A	C22A	59.3(2)
C4B	C25B	C26B	C22B	-59.8(2)	C4A	C25A	C26A	C22A	-62.1(2)
C5B	N5B	C6B	C7B	61.1(2)	C5A	N5A	C6A	C7A	-59.0(2)
C5B	N5B	C6B	C27B	-62.2(2)	C5A	N5A	C6A	C27A	-177.35(16)
C5B	N5B	C6B	C34B	179.80(16)	C5A	N5A	C6A	C34A	64.5(2)
C5B	C4B	C18B	C19B	-178.61(14)	C5A	C4A	C18A	C19A	-177.89(15)
C5B	C4B	C18B	C23B	61.54(19)	C5A	C4A	C18A	C23A	63.7(2)
C5B	C4B	C25B	C24B	178.67(14)	C5A	C4A	C25A	C24A	178.07(15)
C5B	C4B	C25B	C26B	-62.80(18)	C5A	C4A	C25A	C26A	-62.04(18)
C6B	N5B	C5B	O5B	3.8(3)	C6A	N5A	C5A	O5A	-5.0(3)
C6B	N5B	C5B	C4B	-176.88(15)	C6A	N5A	C5A	C4A	176.22(15)
C6B	C27B	C28B	C29B	-61.3(2)	C6A	C27A	C28A	C29A	62.2(2)
C6B	C27B	C32B	C31B	58.7(2)	C6A	C27A	C32A	C31A	-60.0(2)
C6B	C34B	C35B	C31B	-62.2(2)	C6A	C34A	C35A	C31A	61.3(2)
C7B	N7B	C8B	C36B	63.6(3)	C7A	N7A	C8A	C36A	61.0(3)
C7B	N7B	C8B	C37B	-177.1(2)	C7A	N7A	C8A	C37A	-61.2(2)

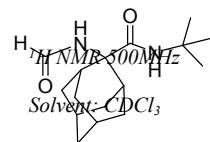
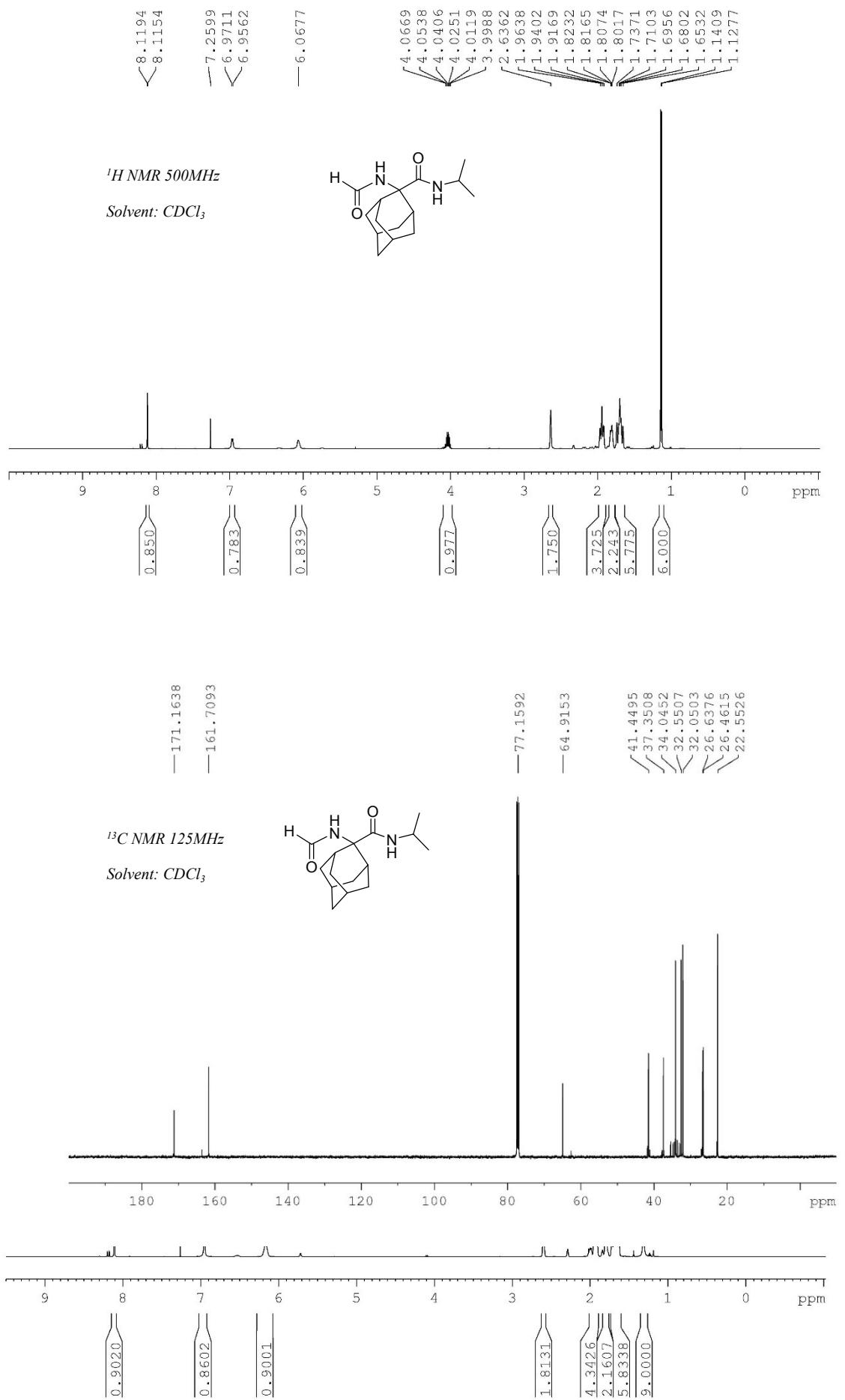
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C7B	C6B	C27B	C28B	-178.05(16)	C7A	C6A	C27A	C28A	63.7(2)
C7B	C6B	C27B	C32B	63.0(2)	C7A	C6A	C27A	C32A	-176.44(15)
C7B	C6B	C34B	C33B	176.99(17)	C7A	C6A	C34A	C33A	-63.2(2)
C7B	C6B	C34B	C35B	-63.4(2)	C7A	C6A	C34A	C35A	178.22(16)
C8B	N7B	C7B	O7B	-5.4(3)	C8A	N7A	C7A	O7A	0.7(3)
C8B	N7B	C7B	C6B	173.91(19)	C8A	N7A	C7A	C6A	-178.57(17)
C9B	C2B	C3B	O3B	-128.8(2)	C9A	C2A	C3A	O3A	13.2(3)
C9B	C2B	C3B	N3B	53.0(2)	C9A	C2A	C3A	N3A	-168.94(16)
C9B	C2B	C16B	C15B	-60.0(2)	C9A	C2A	C16A	C15A	-60.59(19)
C9B	C2B	C16B	C17B	59.3(2)	C9A	C2A	C16A	C17A	59.32(18)
C9B	C10B	C11B	C12B	-60.2(2)	C9A	C10A	C11A	C12A	-60.3(2)
C9B	C10B	C11B	C15B	59.8(2)	C9A	C10A	C11A	C15A	59.7(2)
C10B	C9B	C14B	C13B	-58.4(2)	C10A	C9A	C14A	C13A	-59.4(2)
C10B	C11B	C12B	C13B	60.3(2)	C10A	C11A	C12A	C13A	58.8(2)
C10B	C11B	C15B	C16B	-59.3(2)	C10A	C11A	C15A	C16A	-60.7(2)
C11B	C12B	C13B	C14B	-59.8(2)	C11A	C12A	C13A	C14A	-58.3(2)
C11B	C12B	C13B	C17B	59.0(2)	C11A	C12A	C13A	C17A	60.3(2)
C11B	C15B	C16B	C2B	60.7(2)	C11A	C15A	C16A	C2A	62.1(2)
C11B	C15B	C16B	C17B	-60.4(2)	C11A	C15A	C16A	C17A	-58.1(2)
C12B	C11B	C15B	C16B	60.4(2)	C12A	C11A	C15A	C16A	59.4(2)
C12B	C13B	C14B	C9B	59.3(2)	C12A	C13A	C14A	C9A	59.7(2)
C12B	C13B	C17B	C16B	-60.6(2)	C12A	C13A	C17A	C16A	-59.8(2)
C14B	C9B	C10B	C11B	58.7(2)	C14A	C9A	C10A	C11A	59.6(2)
C14B	C13B	C17B	C16B	58.9(2)	C14A	C13A	C17A	C16A	60.2(2)
C15B	C11B	C12B	C13B	-58.9(2)	C15A	C11A	C12A	C13A	-60.3(2)
C15B	C16B	C17B	C13B	60.5(2)	C15A	C16A	C17A	C13A	58.4(2)
C16B	C2B	C3B	O3B	-7.6(3)	C16A	C2A	C3A	O3A	133.76(19)
C16B	C2B	C3B	N3B	174.19(16)	C16A	C2A	C3A	N3A	-48.4(2)
C16B	C2B	C9B	C10B	60.2(2)	C16A	C2A	C9A	C10A	60.1(2)
C16B	C2B	C9B	C14B	-59.52(19)	C16A	C2A	C9A	C14A	-58.94(19)
C17B	C13B	C14B	C9B	-60.4(2)	C17A	C13A	C14A	C9A	-59.8(2)
C18B	C4B	C5B	O5B	-130.31(18)	C18A	C4A	C5A	O5A	8.3(2)
C18B	C4B	C5B	N5B	50.3(2)	C18A	C4A	C5A	N5A	-172.85(15)
C18B	C4B	C25B	C24B	-59.68(18)	C18A	C4A	C25A	C24A	-59.90(18)
C18B	C4B	C25B	C26B	58.86(19)	C18A	C4A	C25A	C26A	59.99(18)
C18B	C19B	C20B	C21B	-60.2(2)	C18A	C19A	C20A	C21A	-59.9(2)
C18B	C19B	C20B	C24B	59.3(2)	C18A	C19A	C20A	C24A	59.5(2)
C19B	C18B	C23B	C22B	-58.5(2)	C19A	C18A	C23A	C22A	-60.2(2)
C19B	C20B	C21B	C22B	59.7(2)	C19A	C20A	C21A	C22A	58.9(2)
C19B	C20B	C24B	C25B	-59.2(2)	C19A	C20A	C24A	C25A	-59.1(2)
C20B	C21B	C22B	C23B	-59.7(2)	C20A	C21A	C22A	C23A	-59.3(2)
C20B	C21B	C22B	C26B	59.7(2)	C20A	C21A	C22A	C26A	59.9(2)
C20B	C24B	C25B	C4B	60.4(2)	C20A	C24A	C25A	C4A	60.4(2)
C20B	C24B	C25B	C26B	-59.6(2)	C20A	C24A	C25A	C26A	-59.8(2)

C21B	C20B	C24B	C25B	60.1 (2)	C21A	C20A	C24A	C25A	61.1 (2)
C21B	C22B	C23B	C18B	59.2 (2)	C21A	C22A	C23A	C18A	61.0 (2)
C21B	C22B	C26B	C25B	-60.6 (2)	C21A	C22A	C26A	C25A	-59.1 (2)
C23B	C18B	C19B	C20B	59.4 (2)	C23A	C18A	C19A	C20A	59.5 (2)
C23B	C22B	C26B	C25B	59.7 (2)	C23A	C22A	C26A	C25A	60.6 (2)
C24B	C20B	C21B	C22B	-59.4 (2)	C24A	C20A	C21A	C22A	-60.7 (2)
C24B	C25B	C26B	C22B	59.7 (2)	C24A	C25A	C26A	C22A	58.6 (2)
C25B	C4B	C5B	O5B	-10.5 (2)	C25A	C4A	C5A	O5A	128.59 (18)
C25B	C4B	C5B	N5B	170.13 (15)	C25A	C4A	C5A	N5A	-52.6 (2)
C25B	C4B	C18B	C19B	60.16 (18)	C25A	C4A	C18A	C19A	59.50 (18)
C25B	C4B	C18B	C23B	-59.69 (19)	C25A	C4A	C18A	C23A	-58.88 (19)
C26B	C22B	C23B	C18B	-60.7 (2)	C26A	C22A	C23A	C18A	-58.8 (2)
C27B	C6B	C7B	O7B	0.1 (3)	C27A	C6A	C7A	O7A	-117.5 (2)
C27B	C6B	C7B	N7B	-179.27 (17)	C27A	C6A	C7A	N7A	61.7 (2)
C27B	C6B	C34B	C33B	-59.8 (2)	C27A	C6A	C34A	C33A	58.2 (2)
C27B	C6B	C34B	C35B	59.8 (2)	C27A	C6A	C34A	C35A	-60.4 (2)
C27B	C28B	C29B	C30B	-59.4 (2)	C27A	C28A	C29A	C30A	58.9 (2)
C27B	C28B	C29B	C33B	59.7 (2)	C27A	C28A	C29A	C33A	-61.2 (2)
C28B	C27B	C32B	C31B	-61.1 (2)	C28A	C27A	C32A	C31A	60.0 (2)
C28B	C29B	C30B	C31B	58.6 (3)	C28A	C29A	C30A	C31A	-60.3 (3)
C28B	C29B	C33B	C34B	-58.6 (2)	C28A	C29A	C33A	C34A	59.0 (2)
C29B	C30B	C31B	C32B	-59.6 (3)	C29A	C30A	C31A	C32A	61.0 (3)
C29B	C30B	C31B	C35B	60.1 (3)	C29A	C30A	C31A	C35A	-58.4 (3)
C29B	C33B	C34B	C6B	59.6 (2)	C29A	C33A	C34A	C6A	-59.0 (2)
C29B	C33B	C34B	C35B	-59.9 (2)	C29A	C33A	C34A	C35A	60.2 (2)
C30B	C29B	C33B	C34B	61.3 (2)	C30A	C29A	C33A	C34A	-61.2 (2)
C30B	C31B	C32B	C27B	61.7 (3)	C30A	C31A	C32A	C27A	-61.5 (2)
C30B	C31B	C35B	C34B	-58.9 (2)	C30A	C31A	C35A	C34A	59.2 (2)
C32B	C27B	C28B	C29B	59.6 (2)	C32A	C27A	C28A	C29A	-58.4 (2)
C32B	C31B	C35B	C34B	60.9 (2)	C32A	C31A	C35A	C34A	-59.7 (2)
C33B	C29B	C30B	C31B	-60.9 (3)	C33A	C29A	C30A	C31A	59.3 (3)
C33B	C34B	C35B	C31B	58.3 (2)	C33A	C34A	C35A	C31A	-59.1 (2)
C34B	C6B	C7B	O7B	120.5 (2)	C34A	C6A	C7A	O7A	2.6 (3)
C34B	C6B	C7B	N7B	-58.8 (2)	C34A	C6A	C7A	N7A	-178.08 (16)
C34B	C6B	C27B	C28B	60.3 (2)	C34A	C6A	C27A	C28A	-59.4 (2)
C34B	C6B	C27B	C32B	-58.7 (2)	C34A	C6A	C27A	C32A	60.45 (19)
C35B	C31B	C32B	C27B	-58.4 (3)	C35A	C31A	C32A	C27A	58.7 (2)

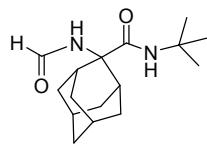
Table 16. Atomic Occupancy for LUB125.

Atom	Occupancy	Atom	Occupancy	Atom	Occupancy
S1	0.894 (2)	S1B	0.106 (2)	H2AA	0.894 (2)
H2AB	0.894 (2)	H2AC	0.894 (2)	H2BD	0.106 (2)
H2BE	0.106 (2)	H2BF	0.106 (2)		

NMR

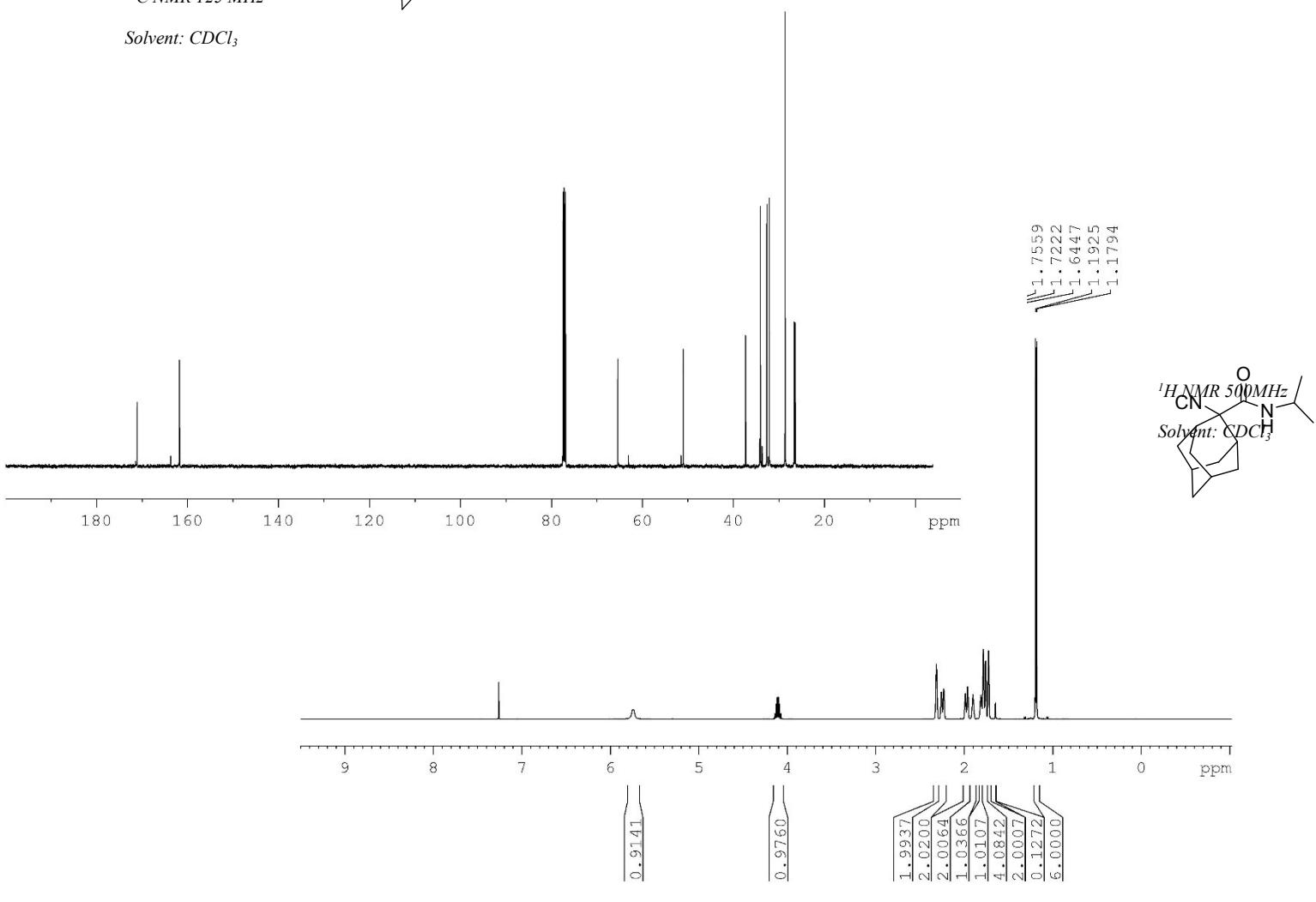


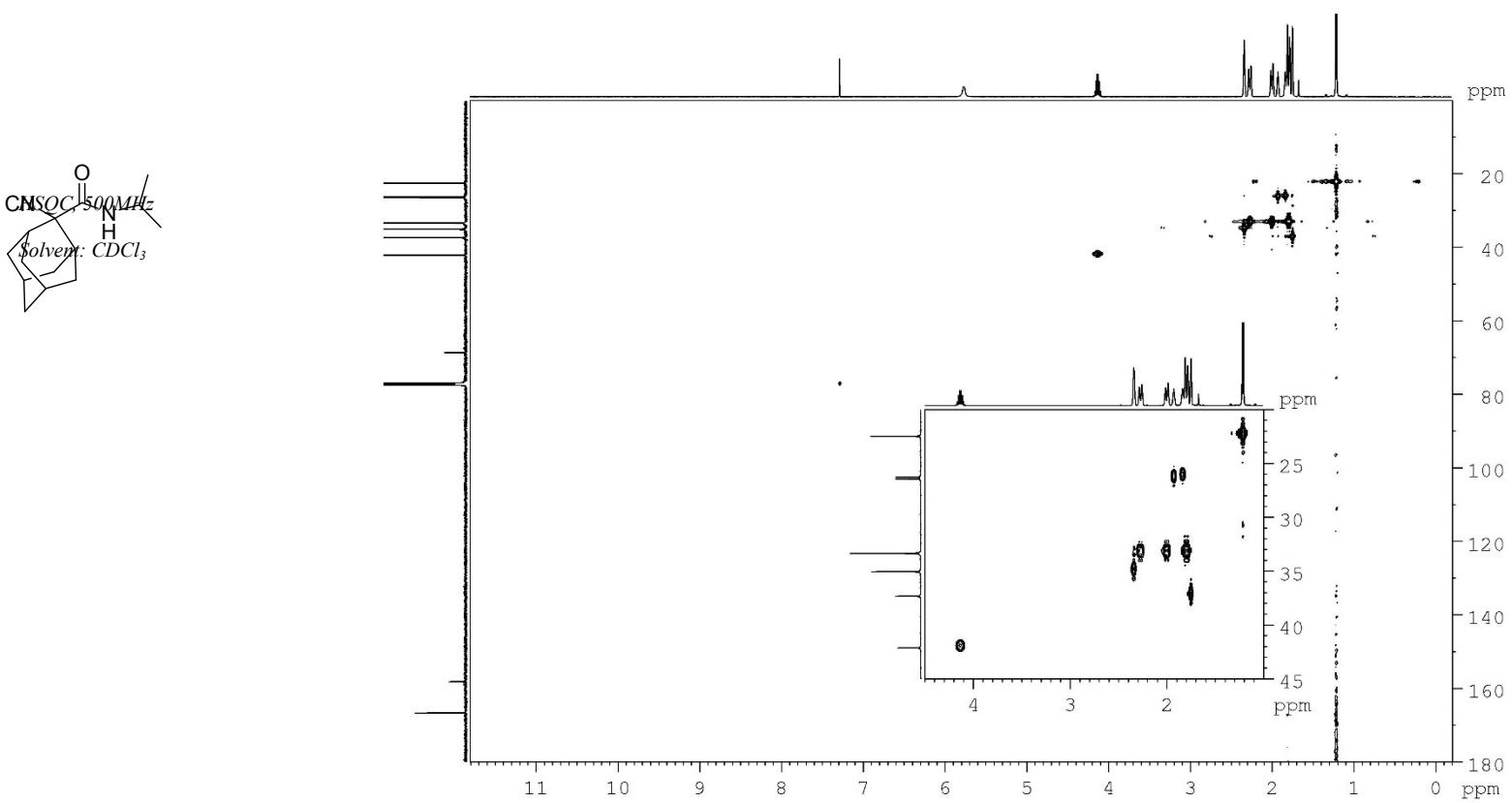
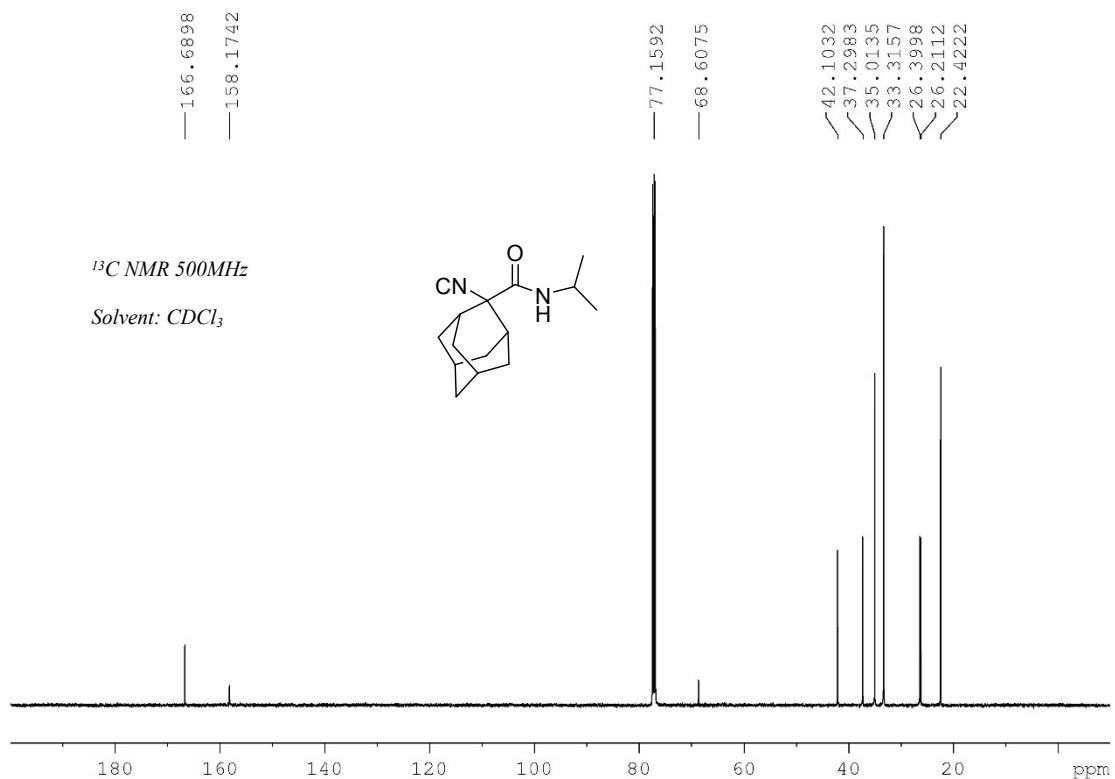
^{13}C NMR 125 MHz

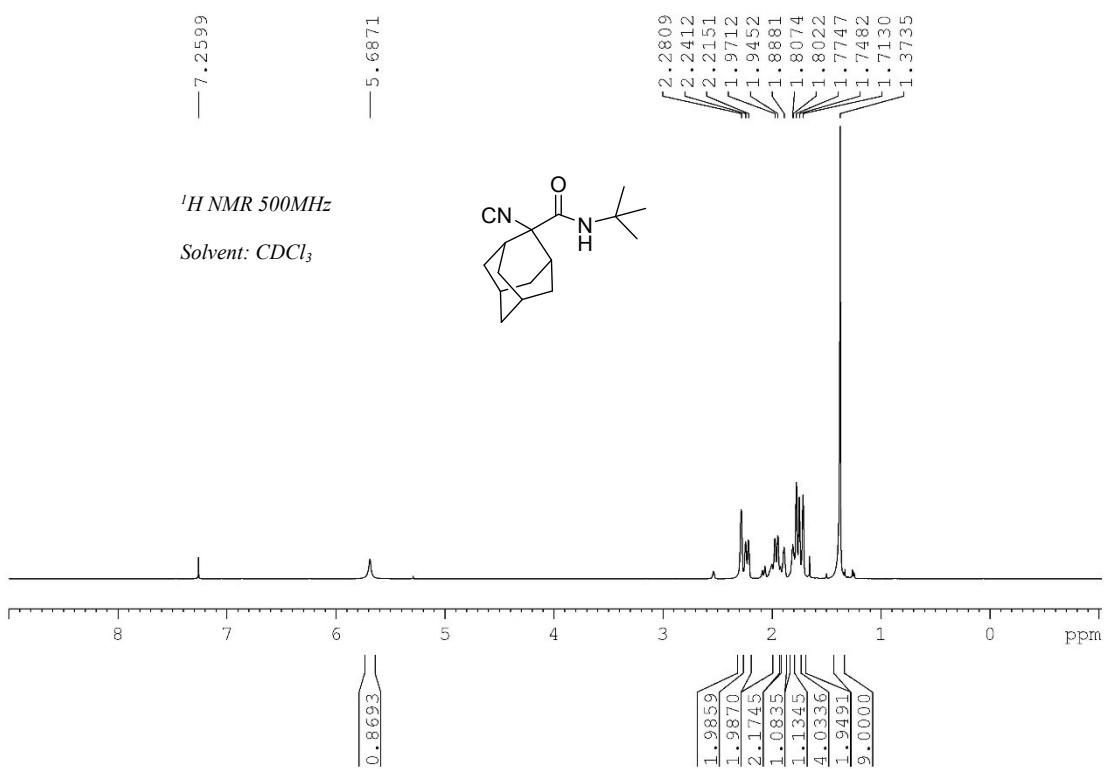


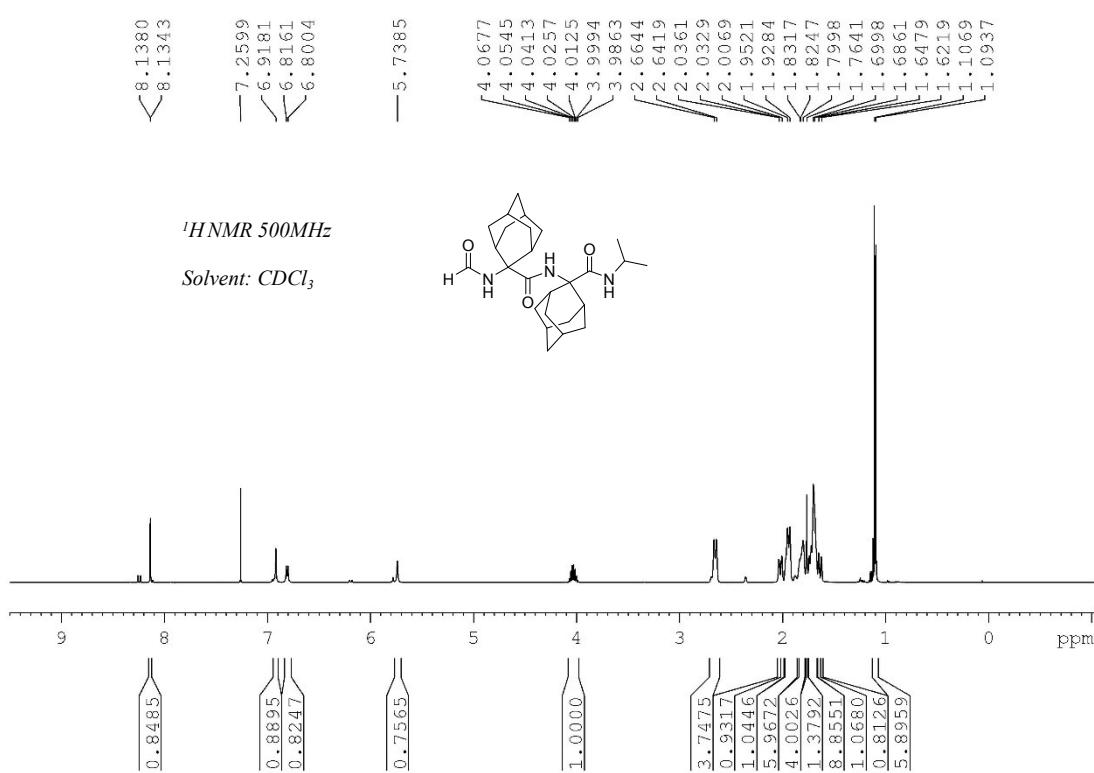
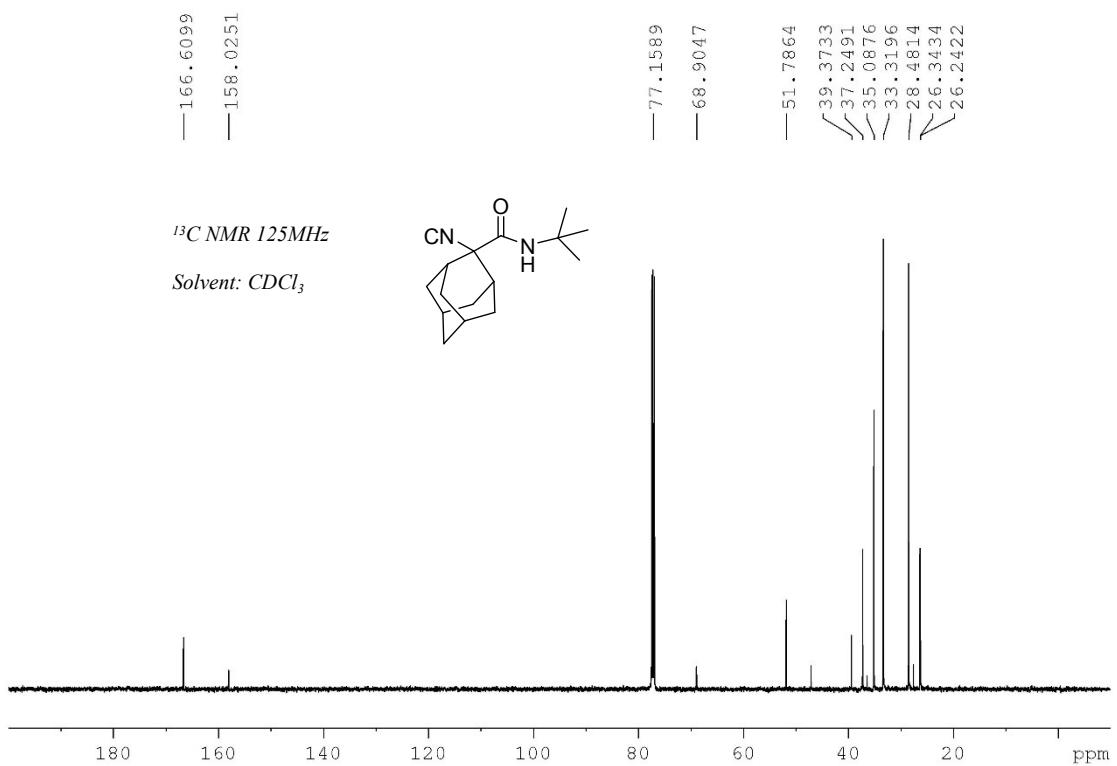
— 171.0449
— 161.7412
— 77.1590
— 65.3856
— 51.0146
/ 37.2977
/ 34.0558
/ 32.179
/ 32.1011
/ 28.6272
/ 26.6452
/ 26.3930

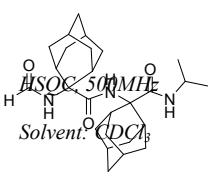
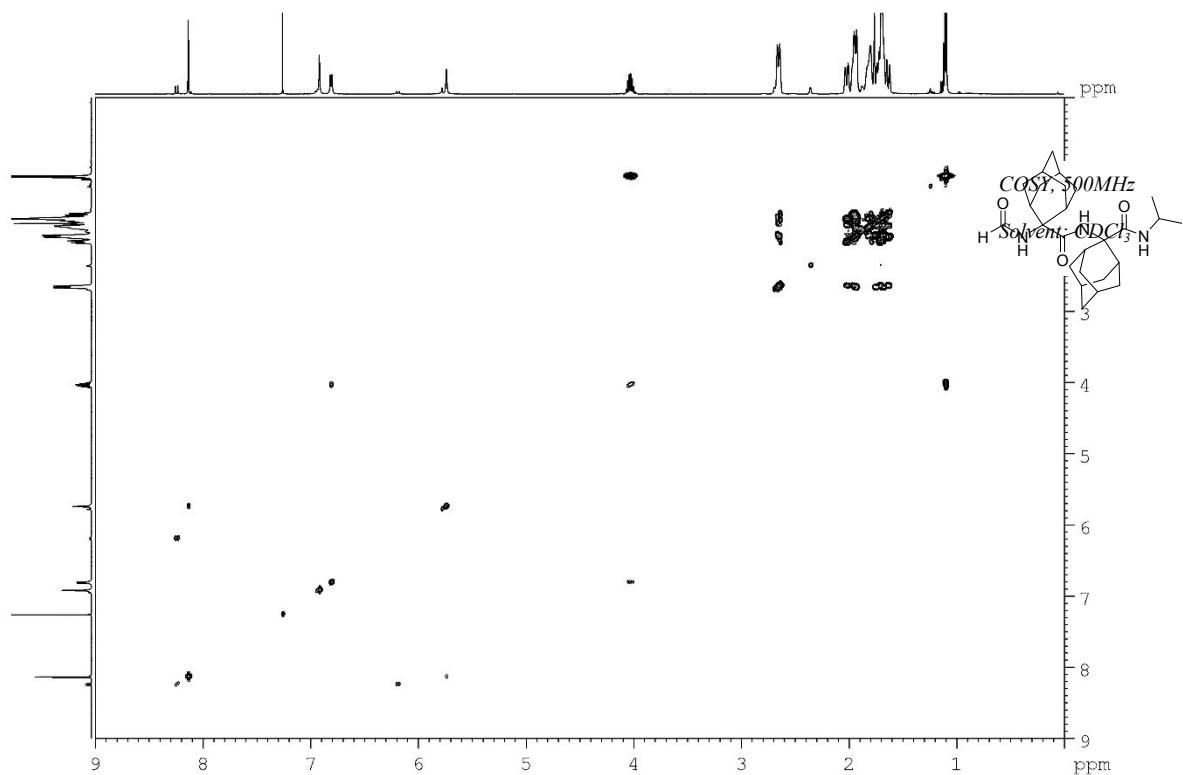
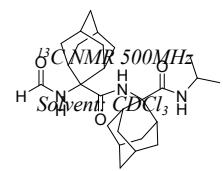
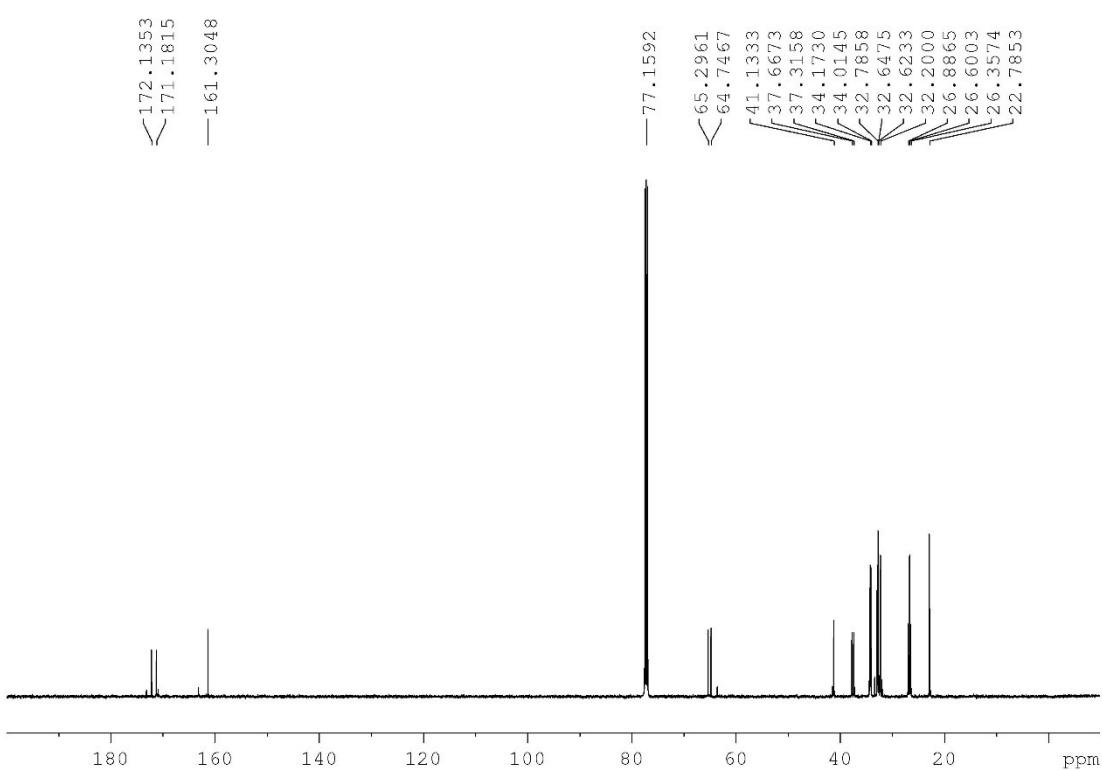
Solvent: CDCl_3

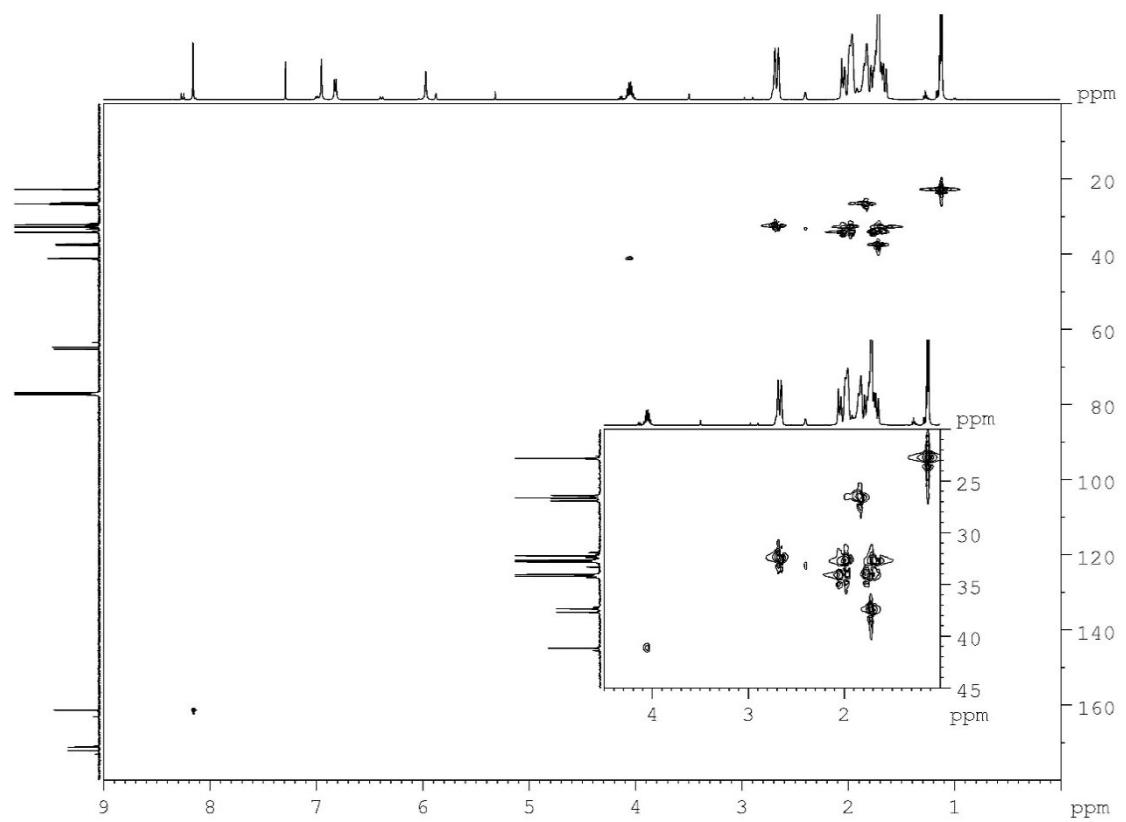


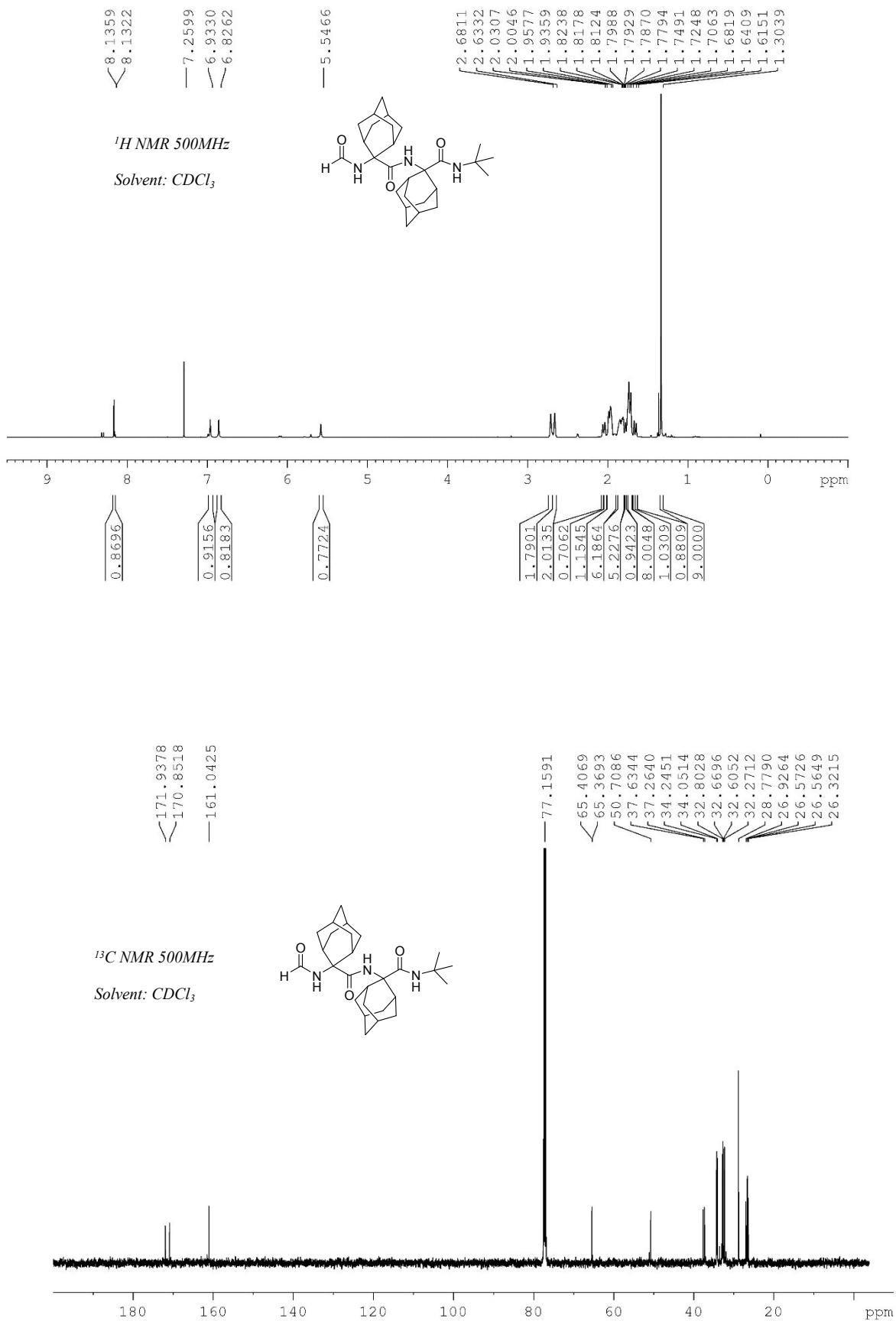


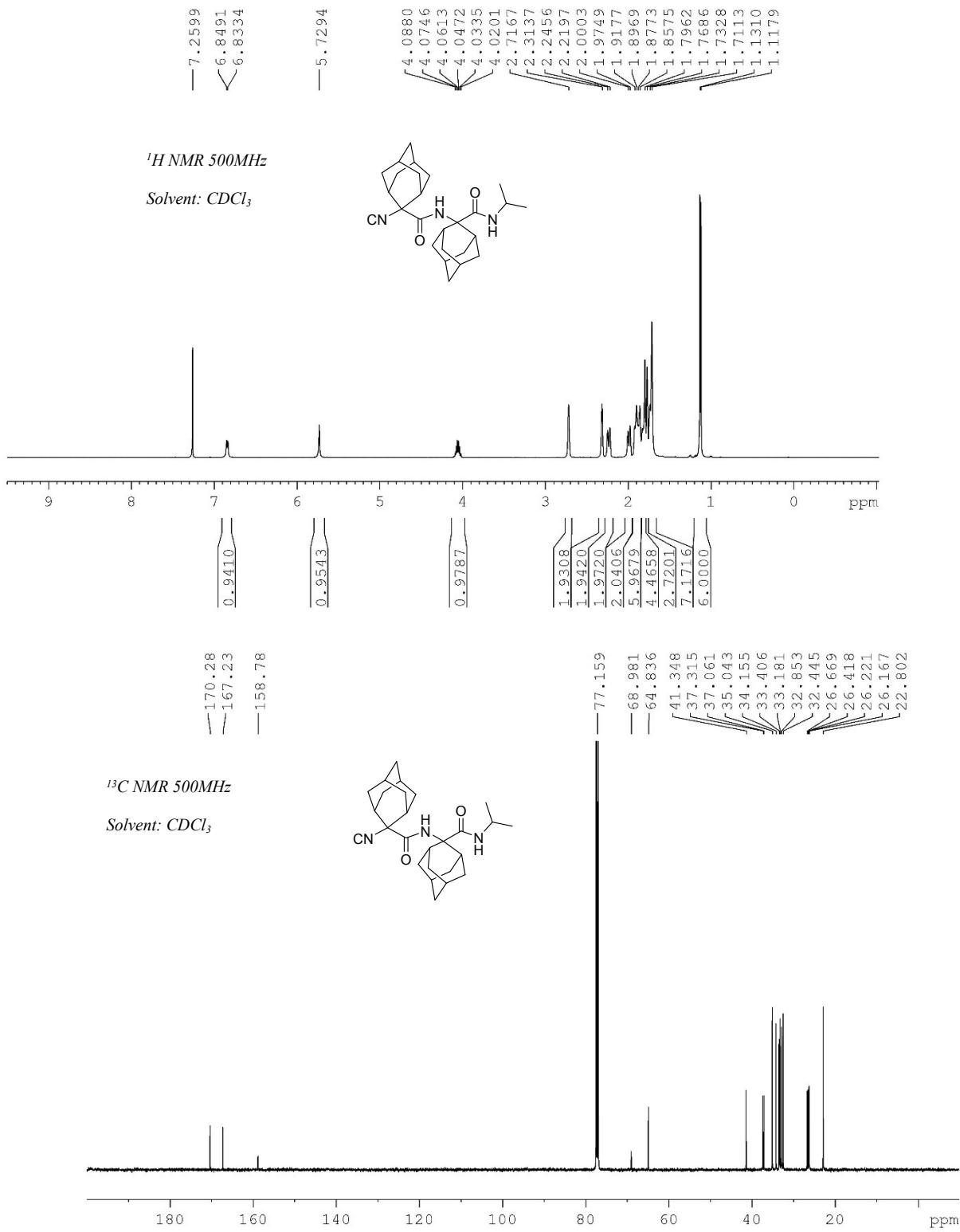


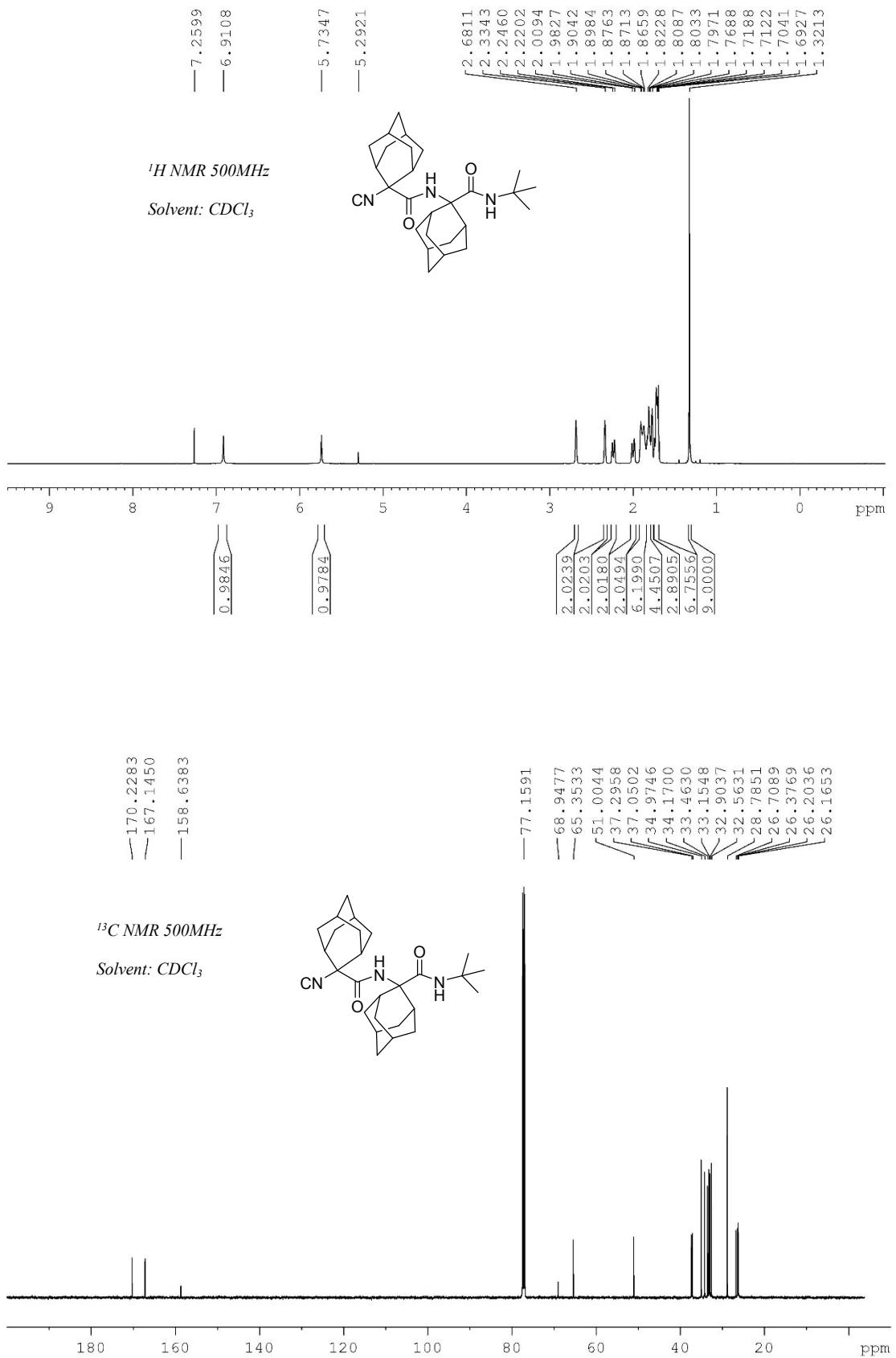


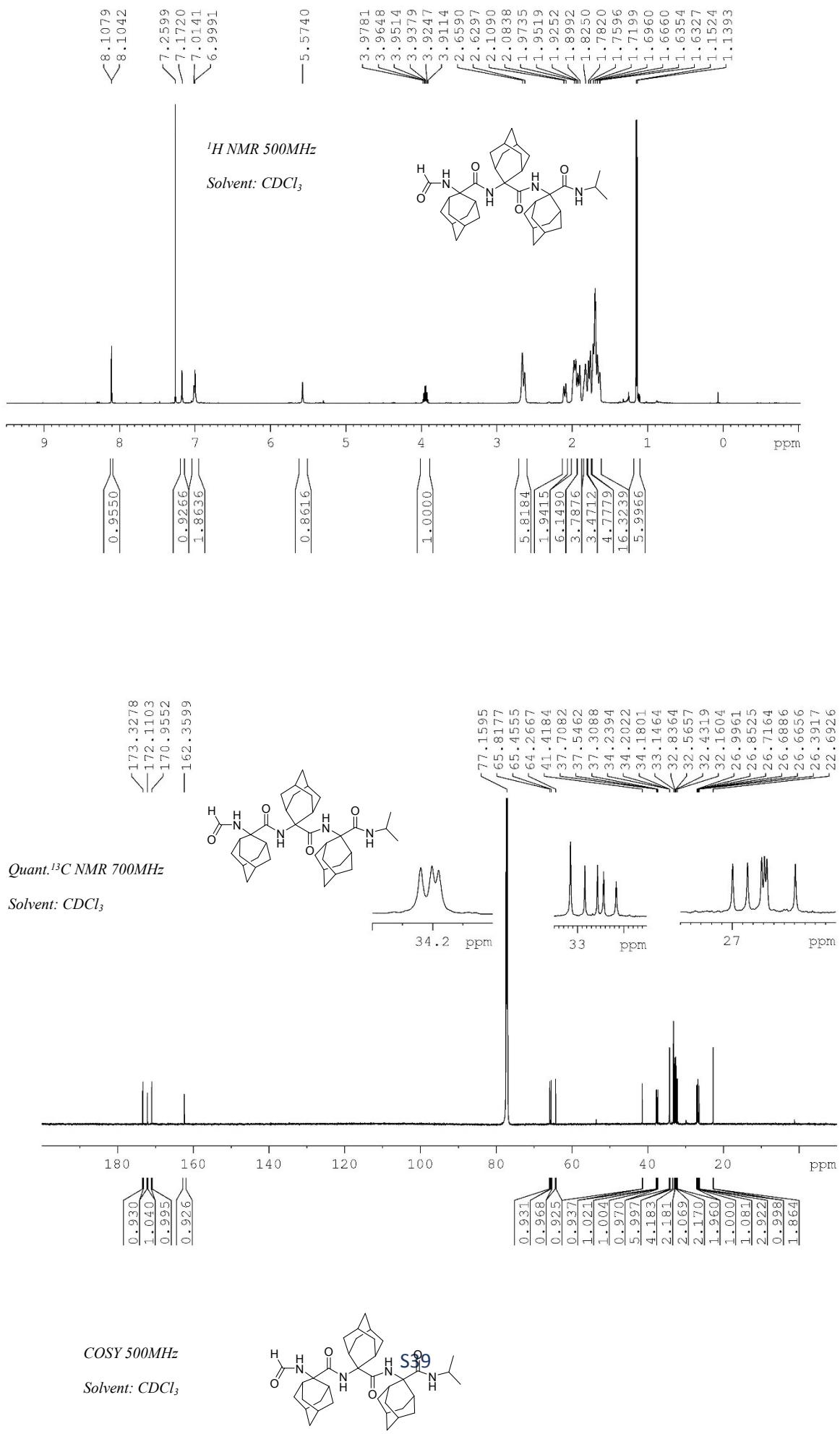


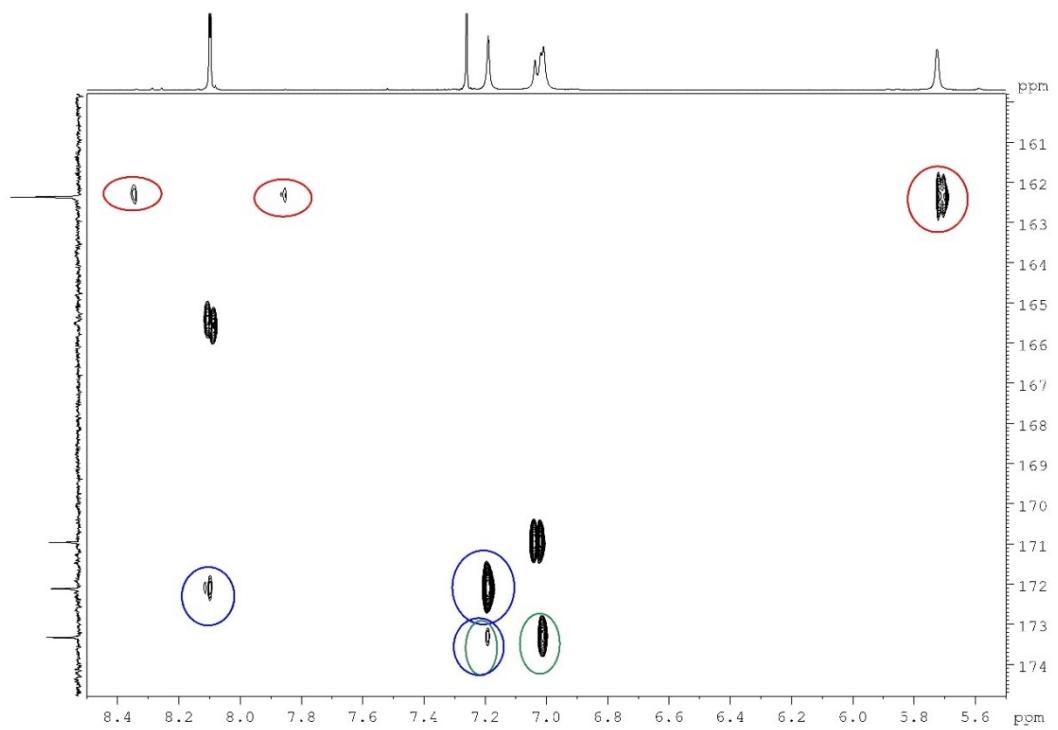
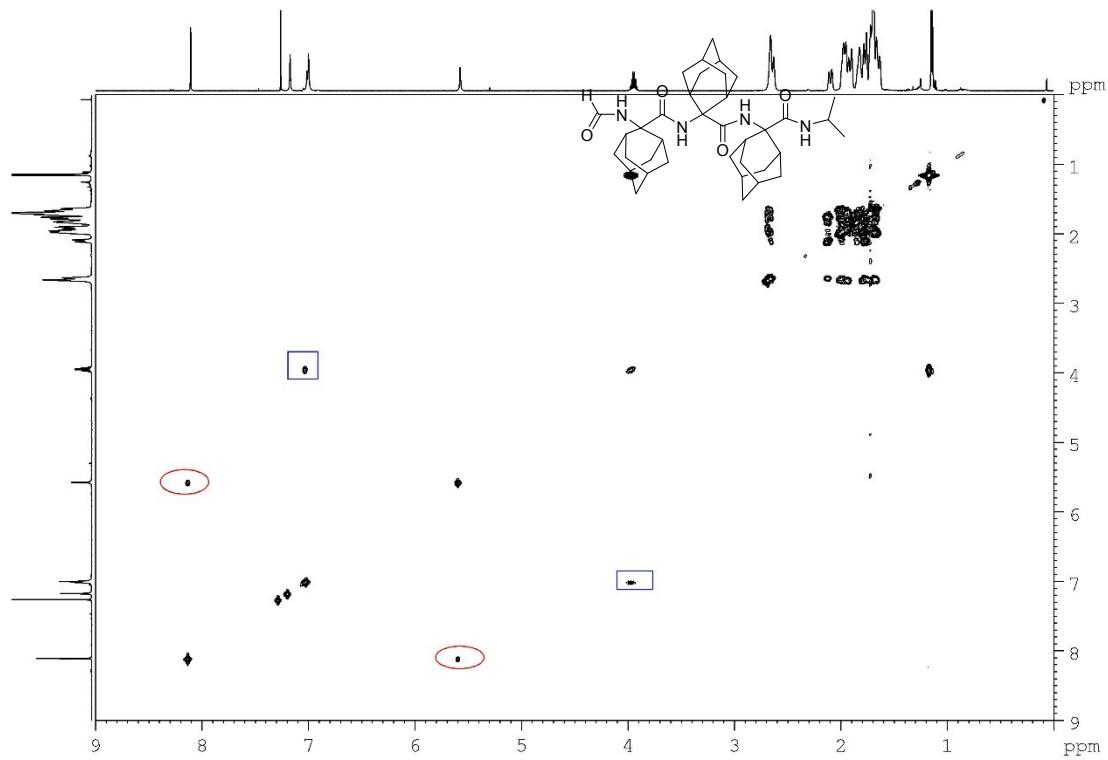


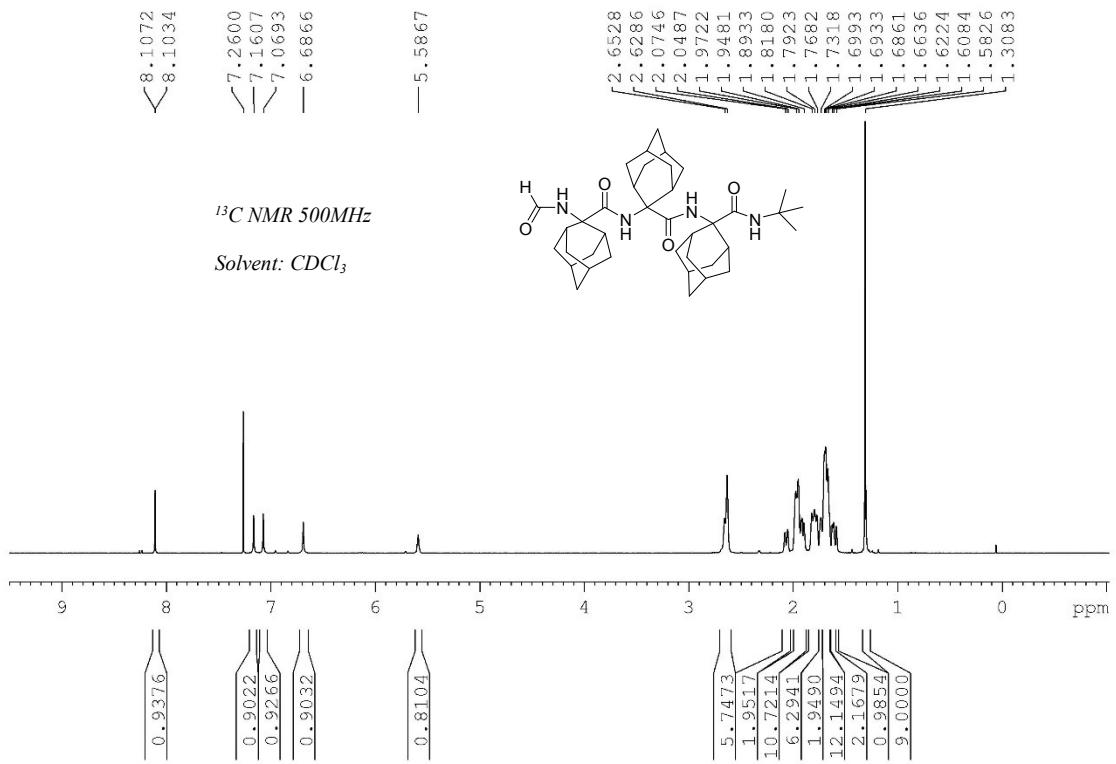


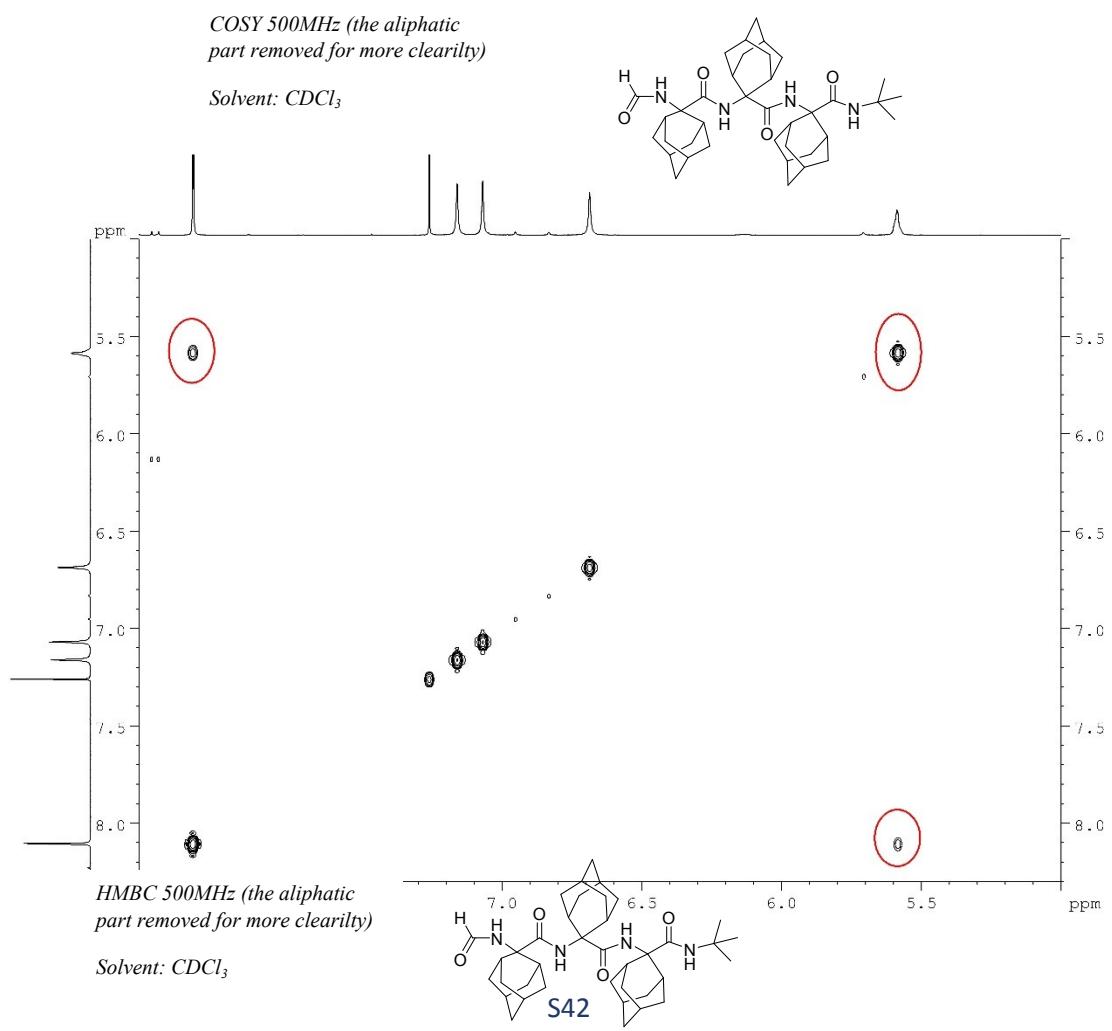
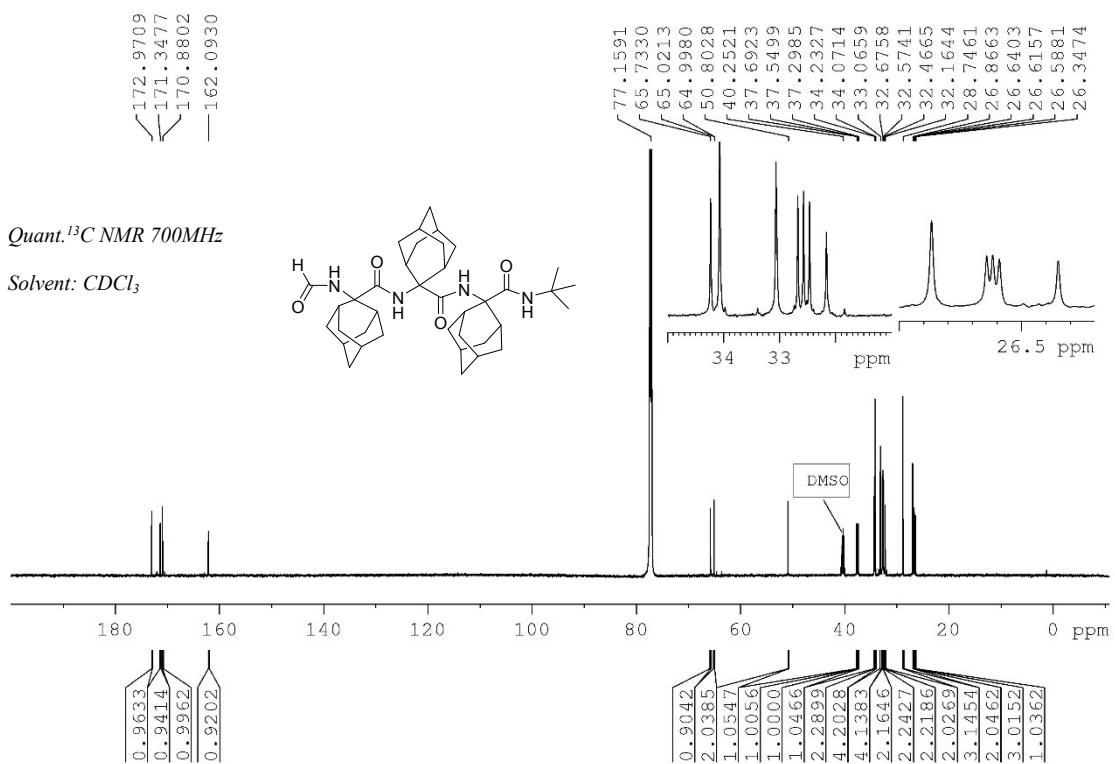


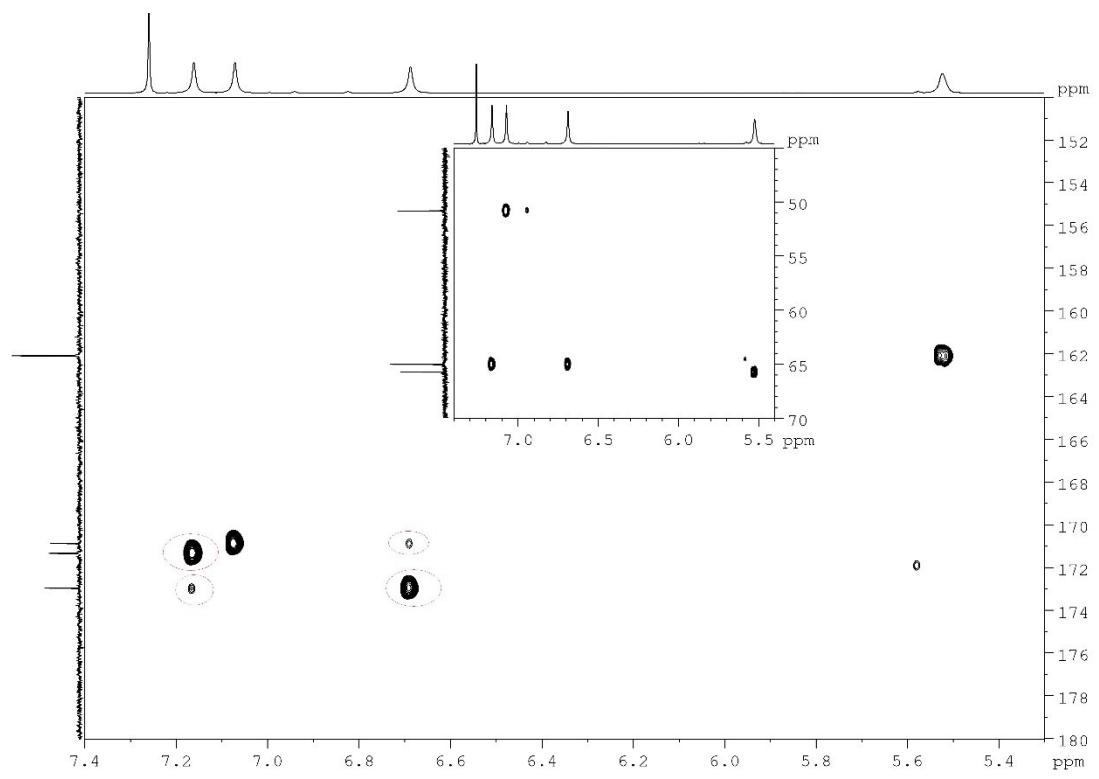


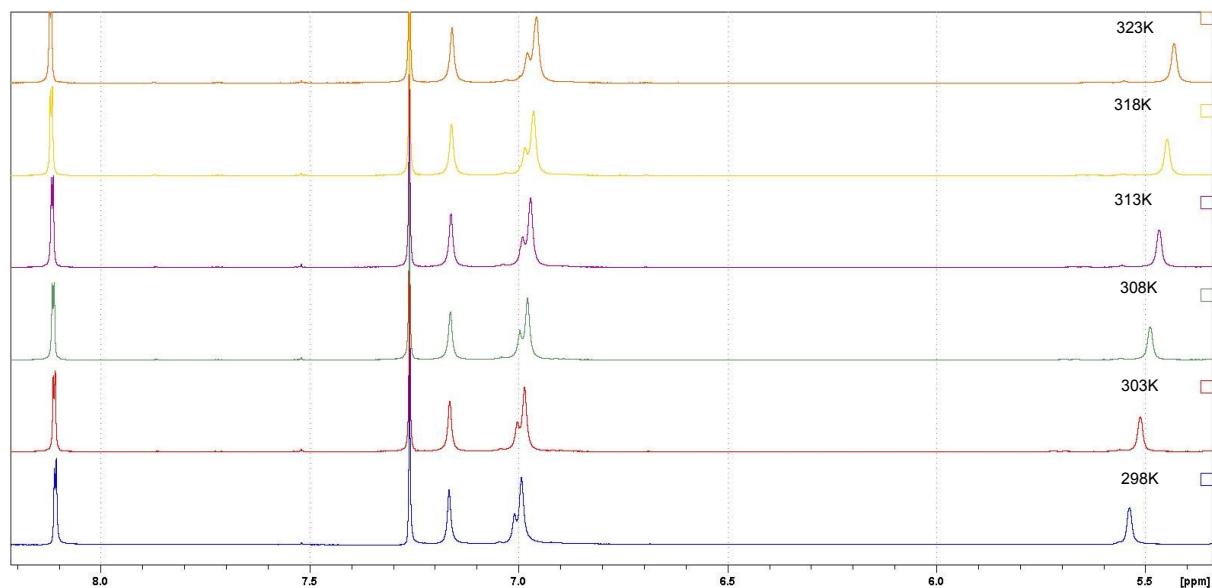




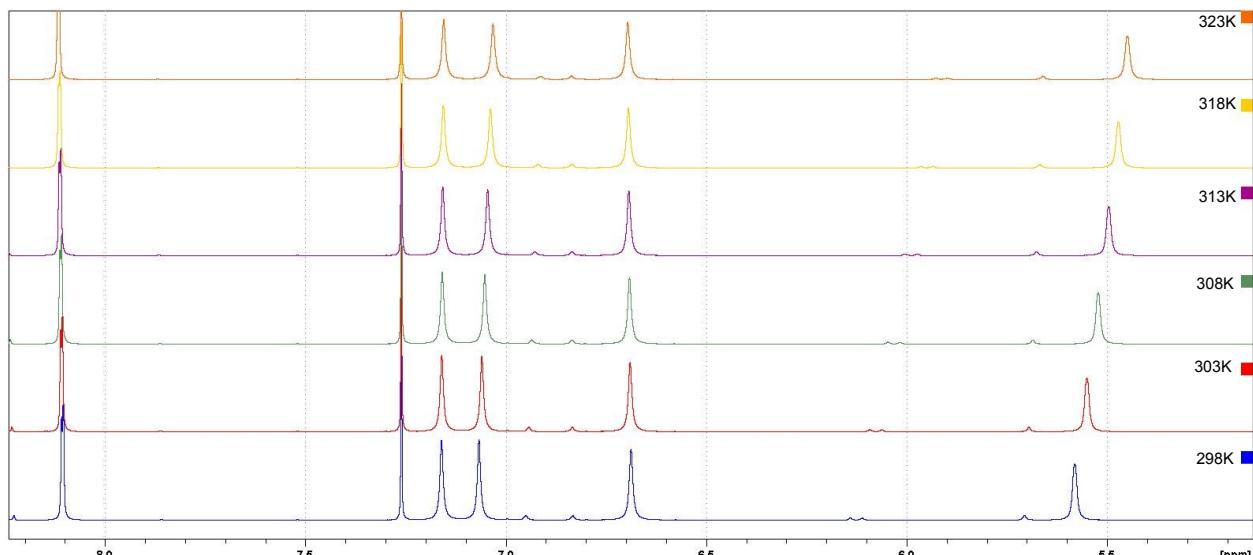




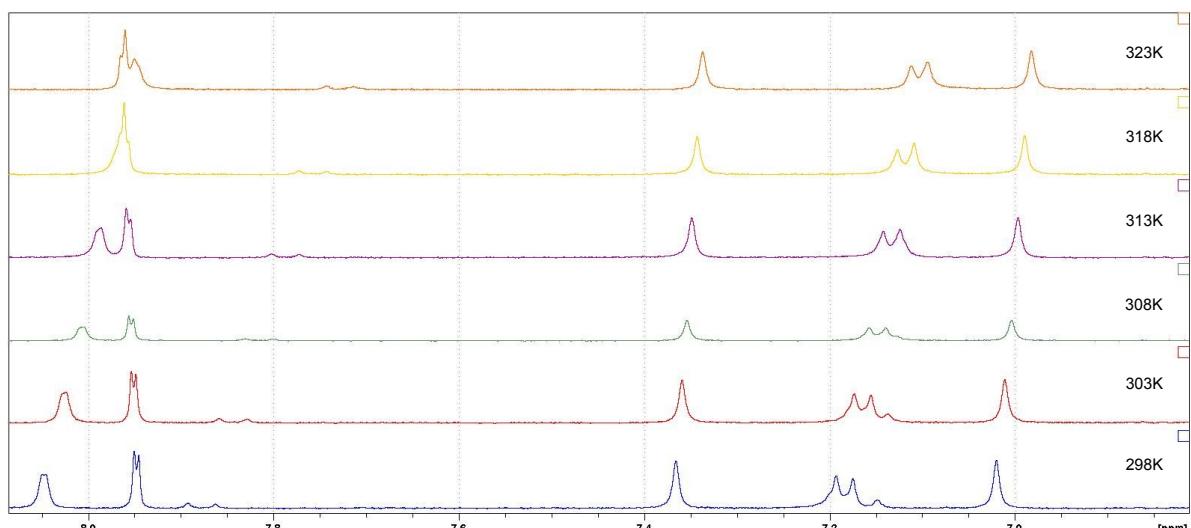




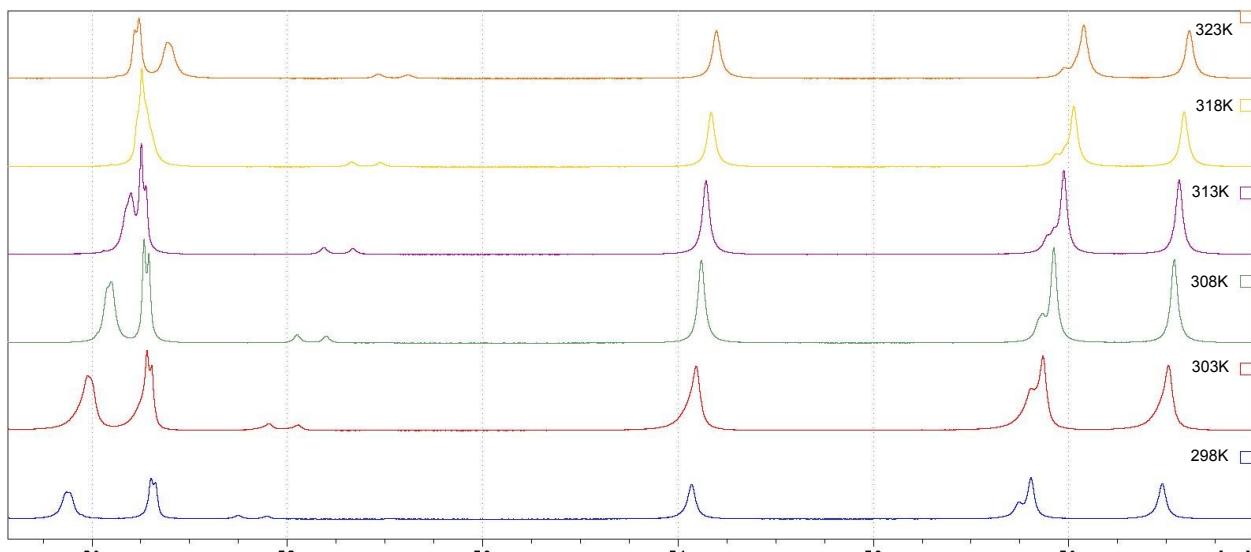
Effect of temperature changes on N-H signal chemical shifts (400 MHz) of **1** in CDCl_3 .



Effect of temperature changes on N-H signal chemical shifts (400 MHz) of **2** in CDCl_3 .



Effect of temperature changes on N-H signal chemical shifts (400 MHz) of **1** in DMSO-*d*6.



Effect of temperature changes on N-H signal chemical shifts (400 MHz) of **2** in DMSO-*d*6.

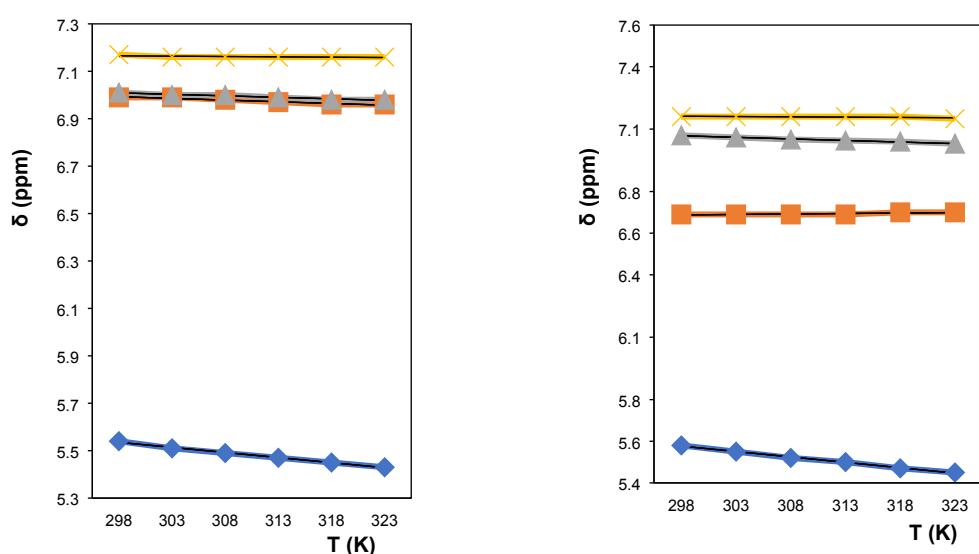


Figure S(1). Plot of changes of NH signal chemical shifts in the NMR spectra of peptides **1** (left) and **2** (right) as a function of temperature in CDCl₃.

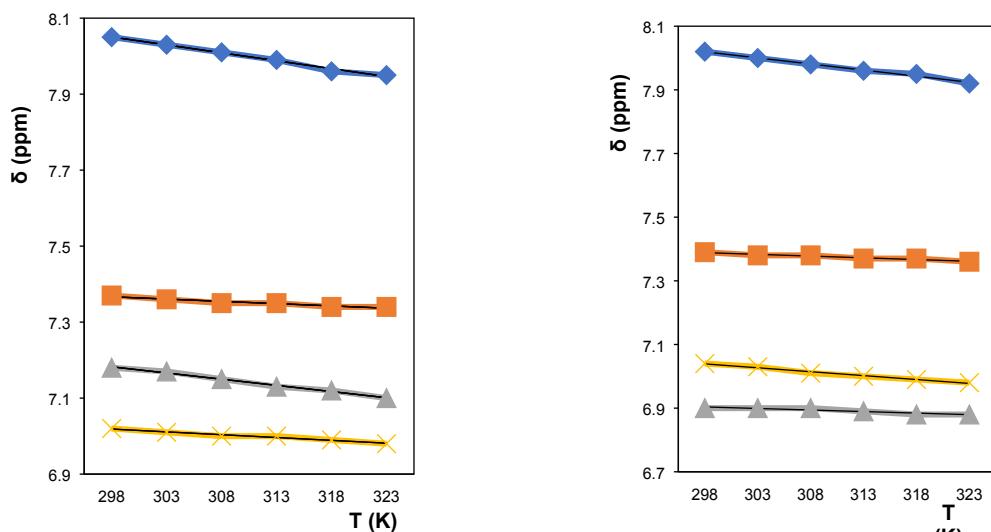


Figure S(2). Plot of changes of NH signal chemical shifts in the NMR spectra of peptides **1** (left) and **2** (left) as a function of temperature in DMSO-*d*6

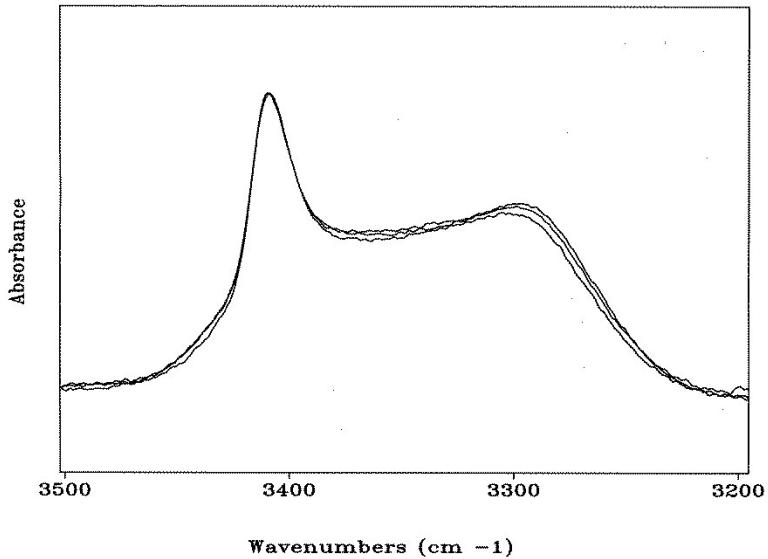


Figure S(3). Overlay of the FT-IR absorption spectra (N-H stretching region) of peptide **1** in CDCl₃ solution at the concentrations 10.0 mM, 1.0 mM, and 0.1 mM.

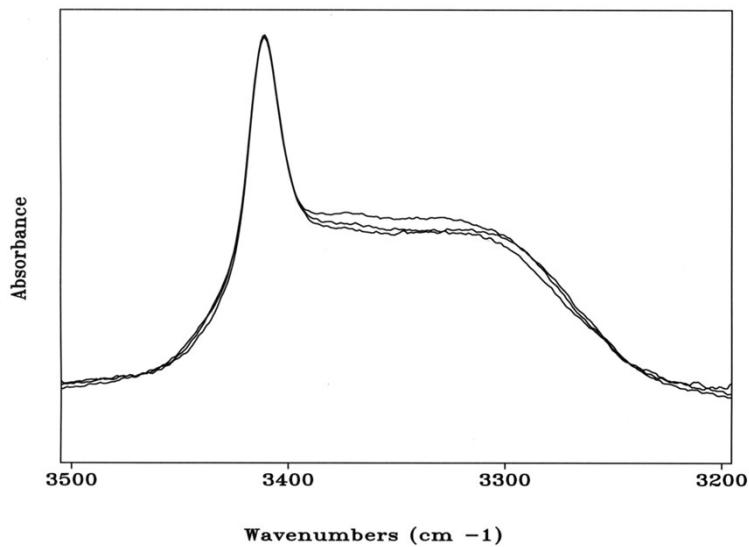


Figure S(4). Overlay of the FT-IR absorption spectra (N-H stretching region) of peptide 2 in CDCl_3 solution at the concentrations 10.0 mM, 1.0 mM, and 0.1 mM.

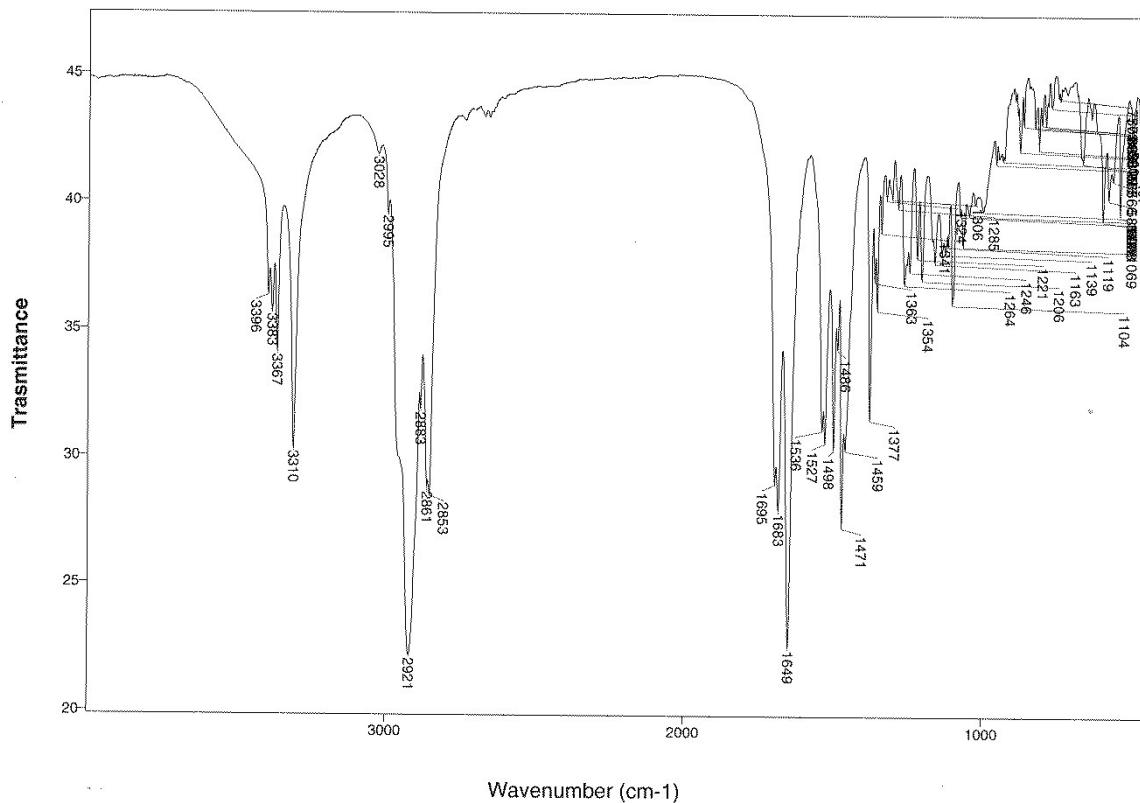


Figure S(5). Solid-state FT-IR absorption spectrum (transmittance mode; KBr disk technique) of crystals of peptide **1** grown from acetone.

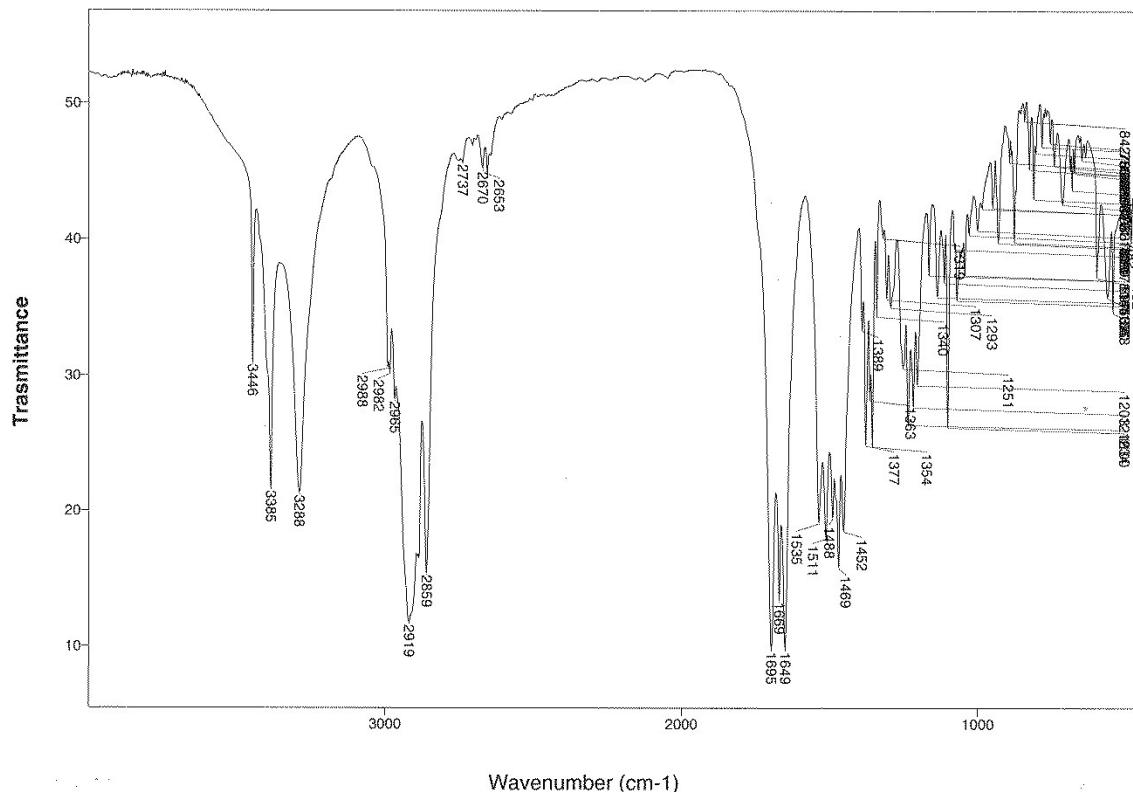


Figure S(6). Solid-state FT-IR absorption spectrum (transmittance mode; KBr disk technique) of crystals of peptide **2** grown from acetone / EtOAc.

SUPPORTING REFERENCES

- S1. Dolomanov, O.V., Bourhis, L.J., Gildea, R.J., Howard, J.A.K. & Puschmann, H. (2009), *J. Appl. Crystallogr.* 42, 339-341.
- S2. Sheldrick, G.M. (2015). *Acta Crystallogr. A*71, 3-8.
- S3. Sheldrick, G.M. (2015). *Acta Crystallogr. C*71, 3-8.

