A safe and fast-charging lithium-ion battery anode using MXene supported Li$_3$VO$_4$

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Figure S1. The binding between Ti atomic of MXene and O atomic of LVO.

Figure S2. SEM images of (a) MAX and (b) MXene.

Figure S3. (a) SEM image and (b) EDX mapping of MAX.
Figure S4. (a) SEM image and corresponding EDX mapping of (b) V, (c) O, (d) Ti and (e) C in the LVO/Ti$_3$C$_2$T$_x$ MXene composite.
Figure S5. The morphology of LVO composites from past works (six images on the two sides) and our work (center). The intrinsic electrical conductivity of LVO is low. Compositing LVO with conductive materials such as graphene\(^1,\) graphite\(^4,\) Ni\(^5\) and MoS\(_2\)\(^6\) have been tried before. However, most of the past works still contain a large amount of LVO clusters on the surface of MXene nanosheets (Figure S5). In our work, the uniform LVO nanoparticles were uniformly grown onto MXene nanosheets. We believe the reason is that MXene is composed of metal and carbon, and the metal ion termination gives it strong affinity with LVO, which allows the uniform growth of LVO on MXene, as illustrated in Figure S1.
Figure S6. SEM images of LVO/Ti$_3$C$_2$T$_x$ MXene composite after 10 minutes sonication.

Figure S7. Low-magnification TEM images of (a) Ti$_3$C$_2$T$_x$ MXene and (b) LVO/Ti$_3$C$_2$T$_x$ MXene.
Figure S8. The first cycle charge/discharge curves of LVO/Ti$_3$C$_2$Tx MXene composite anode in the voltage window of 0.1 V to 3.0 V at 0.1 C.

Figure S9. Rate performance of pure Ti$_3$C$_2$Tx MXene.
Figure S10. Discharge specific capacity and Coulombic efficiency of LVO/Ti₃C₂Tx MXene anode cycled at 5 C.

cycled at 5 C.
**Figure S11.** Capacity retention of LVO/Ti$_3$C$_2$Tx MXene cycled at 5 C and 10 C.

**Figure S12.** SEM images of LVO/Ti$_3$C$_2$Tx MXene composite anode (a) before and (b) after 1000 cycles at 5 C.
**Figure S13.** Nyquist plots of bare LVO and LVO/Ti$_3$C$_2$Tx MXene anodes. Inset: the equivalent circuit model used to simulate the impedance spectrum.

**Table S1.** Comparison of electrical conductivity between MXene, Graphene, Graphene Oxide (GO) and Reduced Graphene Oxide (RGO)

<table>
<thead>
<tr>
<th></th>
<th>Mxene$^7$</th>
<th>Graphene$^{8,9}$</th>
<th>GO$^{10}$</th>
<th>RGO$^{11}$</th>
</tr>
</thead>
<tbody>
<tr>
<td>Electrical Conductivity (S/m)</td>
<td>240000</td>
<td>500</td>
<td>Insulation</td>
<td>1.14</td>
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</tbody>
</table>

**Table S2.** Performance comparison of our LVO/MXene anode with other similar anodes.

<table>
<thead>
<tr>
<th>Anodes</th>
<th>High rate capacity</th>
<th>High rate cycle stability</th>
<th>Reference</th>
</tr>
</thead>
<tbody>
<tr>
<td>LVO@C$^{12}$</td>
<td>147 mAh g$^{-1}$ at 0.75C</td>
<td>Not reported</td>
<td><em>J Mater Chem A, 2015,</em> 3, 11253</td>
</tr>
<tr>
<td>O-Deficient LVO$^{13}$</td>
<td>90 mAh g$^{-1}$ at 5C</td>
<td>No reported</td>
<td><em>Adv. Sci. 2015, 2,</em> 1500090</td>
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<tr>
<td>Material</td>
<td>Capacity g⁻¹</td>
<td>Rate</td>
<td>Reference</td>
</tr>
<tr>
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<tr>
<td>LVO/CNT</td>
<td>87 at 5C, 62 at 10C</td>
<td>Not reported</td>
<td><em>ACS Nano</em> 2016, 10, 5398</td>
</tr>
<tr>
<td>LVO nanoparticle</td>
<td>107 at 5C</td>
<td>Not reported</td>
<td><em>ACS Appl Mater Inter.</em> 2016, 8, 23739</td>
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<tr>
<td>LVO-Rod</td>
<td>150 at 5C, 110 at 5C</td>
<td>Not provided</td>
<td><em>J. Mater. Chem. A</em> 2018, 6, 456</td>
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<tr>
<td>LVO@C nanofibers</td>
<td>141 at 5C, 98 at 10C</td>
<td>Not reported</td>
<td><em>Sci. Bull.</em> 2017, 62, 1081</td>
</tr>
<tr>
<td>hollow-structured LVO</td>
<td>140 at 5C, 110 at 10C</td>
<td>100 after 1000 cycles at 10C (20% carbon black)</td>
<td><em>Chem. Eur. J.</em> 2014, 20, 5608-5612</td>
</tr>
<tr>
<td>LVO nanoparticle</td>
<td>86 at 10C</td>
<td>71 after 1000 cycles at 10C (20% carbon black)</td>
<td><em>J. Solid State Electrochem.</em> 2017, 21, 2547</td>
</tr>
<tr>
<td>LVO@GNs</td>
<td>250 at 5C, 210 at 10C</td>
<td>163 after 5000 cycles at 5C</td>
<td><em>Chem. Commun.</em> 2015, 51, 229-231</td>
</tr>
<tr>
<td>LVO@N-doped C</td>
<td>141 at 5C, 60 at 10C</td>
<td>103 after 1000 cycles at 10C (20% acetylene black)</td>
<td><em>J. Alloy. Comp.</em> 2018, 767, 657</td>
</tr>
<tr>
<td>LVO/MXene</td>
<td>187 at 5C, 114 at 10C</td>
<td>146 after 1000 cycles at 5C, 81 after 1000 cycles at 10C (only 10% carbon black)</td>
<td>This work</td>
</tr>
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**Reference:**