Supporting information:

#### C<sub>60</sub>/TiO<sub>x</sub> Bilayer for Conformal Growth of Perovskite Film for UV Stable

# **Perovskite Solar Cells**

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# **Experimental Section**

*1. Materials*: C<sub>60</sub> (99.5%) and TiCl<sub>4</sub> (98%) were respectively purchased form Alfa Aesar (U.S.) and Sinopharm Chemical Reagent Co.,Ltd (China). Lead (II) Iodide (99.99%, trace metals basis) and Lead (II) Bromine (purity, 99.99%) (for Perovskite precursor) was purchased from TCI (Japan), CH<sub>3</sub>NH<sub>3</sub>Br (MABr) and CH<sub>3</sub>CH<sub>2</sub>NH<sub>3</sub>I (FAI) from Dyesol (Australia). DMF, DMSO, 1,2-dichlorobenzene, 4-tertbutylpyridine (tBP) and lithium bis(trifluoromethylsulfonyl)amine (Li-TFSI) were obtained from Aldrich (U.S.). Tris(2-(1Hpyrazol-1-yl)-4-tert-butylpyridine)cobalt(III) tris(bis(trifluoromethylsul-fonyl)-imide) (FK209-cobalt(III)-TFSI) and spiro-MeOTAD was purchased from Xi'an Polymer Light Technology (China). All reagents were used as received.

2. Solar cells preparation: First, a non-compact fullerene ( $C_{60}$ ) film were deposited on the clean substrates (indium-doped tin oxide or polyethylene terephthalate/ ultrathin gold) by spin-coating a nearly saturated solution of 28 mM C<sub>60</sub> dissolved in 1,2-dichlorobenzene with a speed of 2000 rpm for 30 s and annealed at 60°C for 5 min. In order to establish an ultrathin layer of TiO<sub>x</sub>, an aqueous solution of TiCl<sub>4</sub> was diluted to the concentration varying from 1 mM to 9 mM. The ITO/ $C_{60}$ substrates were then immersed into these solutions and kept in an oven at 70°C for 30 min in petri dishes. The TiCl<sub>4</sub>-treated films were washed with deionized water and dried at 70°C in air for 1 hour. To compare the performance between the bilayer- and TiO<sub>2</sub>based devices, the reference TiO<sub>2</sub>-based devices were fabricated using TiO<sub>2</sub> ETLs by spray method at 400°C as previous reported.<sup>1</sup> Subsequently, FA<sub>0.85</sub>MA<sub>0.15</sub>PbI<sub>2.55</sub>Br<sub>0.45</sub> precursor solution was prepared of 1.1 M  $Pb^{2+}$  (PbI<sub>2</sub> and PbBr<sub>2</sub>) in a mixed solvent of DMSO and DMF (v/v=1:4). Both of the molar ratios for PbI<sub>2</sub>/PbBr<sub>2</sub> and FAI/MABr were fixed at 0.85:0.15. In a second step, the completely dissolved solution was spin coated onto ETLs in the nitrogen glovebox with the following procedure: first 1000 rpm for 10 s and second 5000 rpm for 30 s with ramps of 1000 and 2500 rpm·s<sup>-1</sup>,

respectively. 110  $\mu$ L of chlorobenzene was rapidly dripped on the rotating substrates during the second spin-coating step 15 s before the end of the procedure. The transparent perovskite film was then heated at 100 °C for 1.5 hours. A spiro-OMeTAD/ chlorobenzene solution (60 mM) with additives of 28  $\mu$ L 4-tert-butylpyridine (TBP) and 17.5  $\mu$ L Li-TFSI/acetonitrile (1.8 M) and 8 $\mu$ L FK209-cobalt(III)-TFSI/acetonitrile (0.2 M) was spin coated on top of the active layer at 4500 rpm for 20 s. Finally, 80-nmthick Au was deposited by thermal evaporation under high vacuum.

3. Characterizations: The J-V characteristics of the devices were measured with a Keithley 2400 sourcemeter equipped with a sunlight simulator (XES-300T1, SAN-EI Electric, AM 1.5), which was calibrated using a standard silicon reference cell. The J-V curves of all devices were measured by masking the active area with a metal mask of 1 cm<sup>2</sup>. Incident photon to current efficiency (IPCE) was measured as a function of wavelength from 300 to 900 nm (Enli Technology) with dual Xenon/quartz halogen light source. The absorption spectra were recorded using UV/Vis spectrometer (Shimadzu, UV-3600) in the 300 nm - 900 nm range. Confocal PL mapping was carried out with a laser confocal Raman spectrometer (Princeton Instruments, Acton Standard Series SP-2558) and a 485 nm laser (PicoQuant LDH-P-C-485, 0.4 mW with a 1% optical density filter), using a home-built confocal microscope on a 5 × 5  $\mu$ m<sup>2</sup> sample area.



**Figure S1.** (a) Optical transmission spectra of  $FTO/C_{60}$  and  $FTO/C_{60}/u$ -TiO<sub>x</sub> bilayer samples; (b) UPS measurements of  $C_{60}/u$ -TiO<sub>x</sub> bilayer and schematics of band alignment for devices based on different ETLs.



Figure S2. The surface SEM images of partial perovskite films on (a)  $C_{60}$  layer and (b)  $C_{60}/u$ -TiO<sub>x</sub> bilayer.

Table S1. The fitting date from the TRPL spectra of  $C_{60}$ /perovskite and  $C_{60}$ /u-TiO<sub>x</sub>/perovskite

samples.
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Туре	$ au_1$	$ au_2$	$A_1$	A <sub>2</sub>	$ au_{ m ave}$
C <sub>60</sub>	15.09	78.29	49.34%	50.66%	68.30
C <sub>60</sub> /u-TiO <sub>x</sub>	9.42	72.18	55.68%	44.32%	63.34



**Figure S3.** (a) Transmittance and reflectance, (b) light harvesting efficiency (LHE) and (c) absorbance of the ETLs/perovskite/*spiro*-OMeTAD film; (d) absorbed photon-to-current conversion efficiency (APCE) spectra of  $C_{60}$  and  $C_{60}/u$ -TiO<sub>x</sub> bilayer-based PSCs.



**Figure S4.** (a) EQE spectra of PSCs with  $C_{60}$  and  $C_{60}/u$ -TiO<sub>x</sub> bilayer ETLs. (b) Steady-state measurement of photocurrent ( $J_{sc}$ ) and PCE of PSCs with  $C_{60}$  and  $C_{60}/u$ -TiO<sub>x</sub> bilayer.

	$J_{\rm sc} ({\rm mA}{\cdot}{\rm cm}^{-2})$	$V_{ m oc}$ (V)	FF (%)	PCE (%)
0 mM	22.51	1.01	62.69	14.27
1 mM	22.07	1.02	70.37	15.78
3 mM	22.38	1.06	75.82	18.01
5 mM	22.65	1.09	77.55	19.21
7 mM	22.72	1.09	77.16	19.05
9 mM	22.39	1.08	76.89	18.63

**Table S2.** Photovoltaic parameters of PSCs fabricated with  $C_{60}/u$ -TiO<sub>x</sub> bilayer ETLs with a TiCl<sub>4</sub> treatment process varying in solution concentration from 0 to 50 mM measured under AM1.5 illumination.



**Figure S5.** XRD patterns of perovskite films on different ETLs under constant 10 mW·cm<sup>-2</sup> UV irradiation ( $\lambda = 340$  nm) in air ( $\approx 45\%$  humidity) for (a-c) 312 h and (d-f) 5 h.



**Figure S6.** The transmittance of the polyethylene terephthalate (PET) and PET/ ultrathin (8 nm) gold (PETUG) substrates.

# **Reference:**

[1] Y.-K. Ren, X.-H. Ding, Y.-H. Wu, J. Zhu, T. Hayat, A. Alsaedi, Y.-F. Xu, Z.-Q. Li,

S.-F. Yang, S.-Y. Dai, Journal of Materials Chemistry A, 2017, 5, 20327-20333.