

## SUPPORTING INFORMATION

# Porous NiTiO<sub>3</sub>/TiO<sub>2</sub> nanostructures for photocatalytic hydrogen evolution

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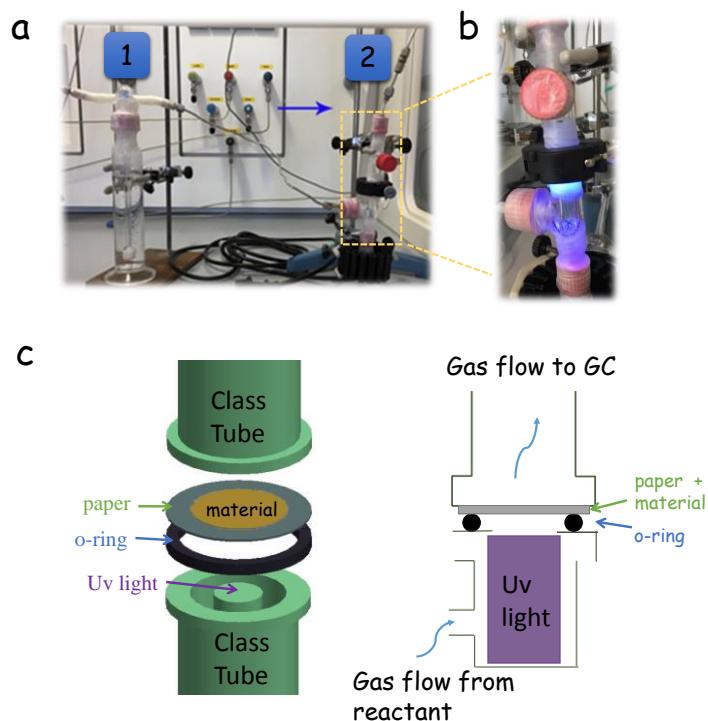
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# Content

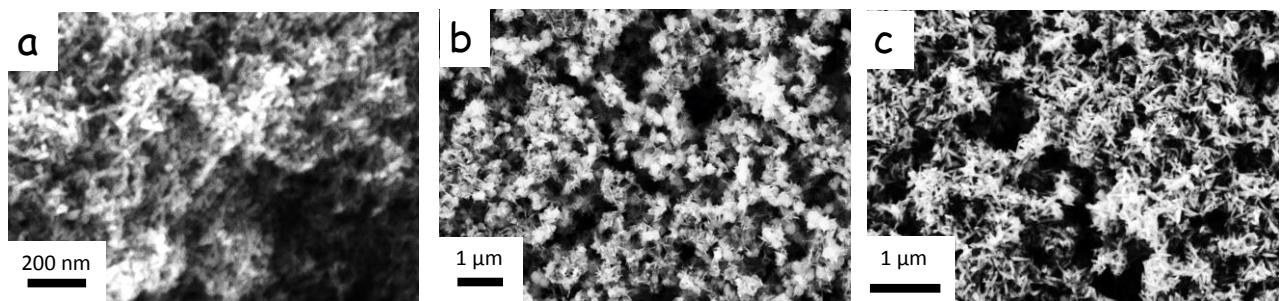
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## 1. Photoreactor



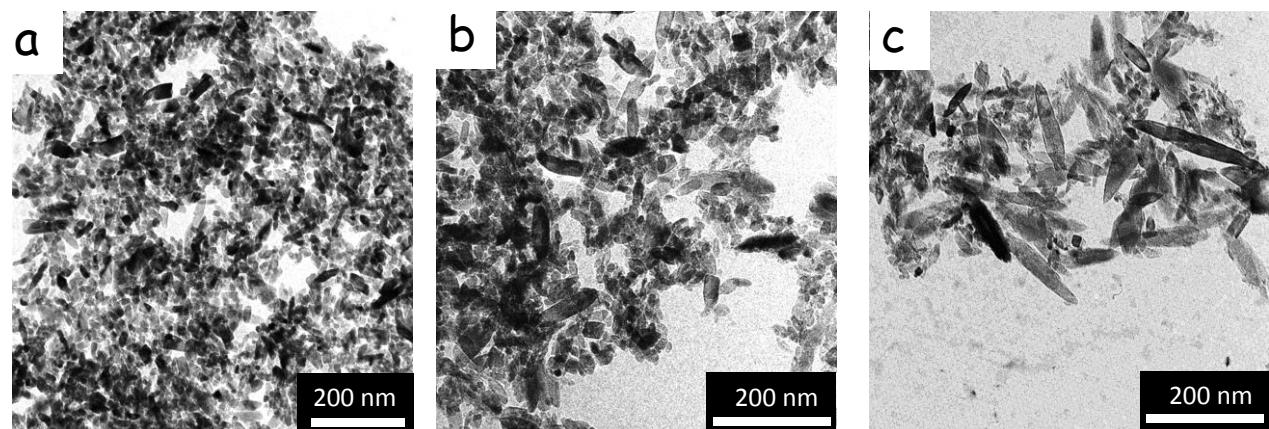
**Figure S1.** a) Photograph of the system used to test the photocatalytic hydrogen generation: 1 displays the Dreschel bottle containing the ethanol-water solution (1:9 molar). 2 displays the actual photoreactor. b) Scheme of the photoreactor.

## 2. Additional SEM characterization of the precursor materials



**Figure S2.** SEM images of TiO<sub>2</sub>:Ni (5%) produced in the following conditions: a) without HDA, b) without H<sub>2</sub>O, c) in HDA-H<sub>2</sub>O-KCl.

### 3. Additional TEM characterization



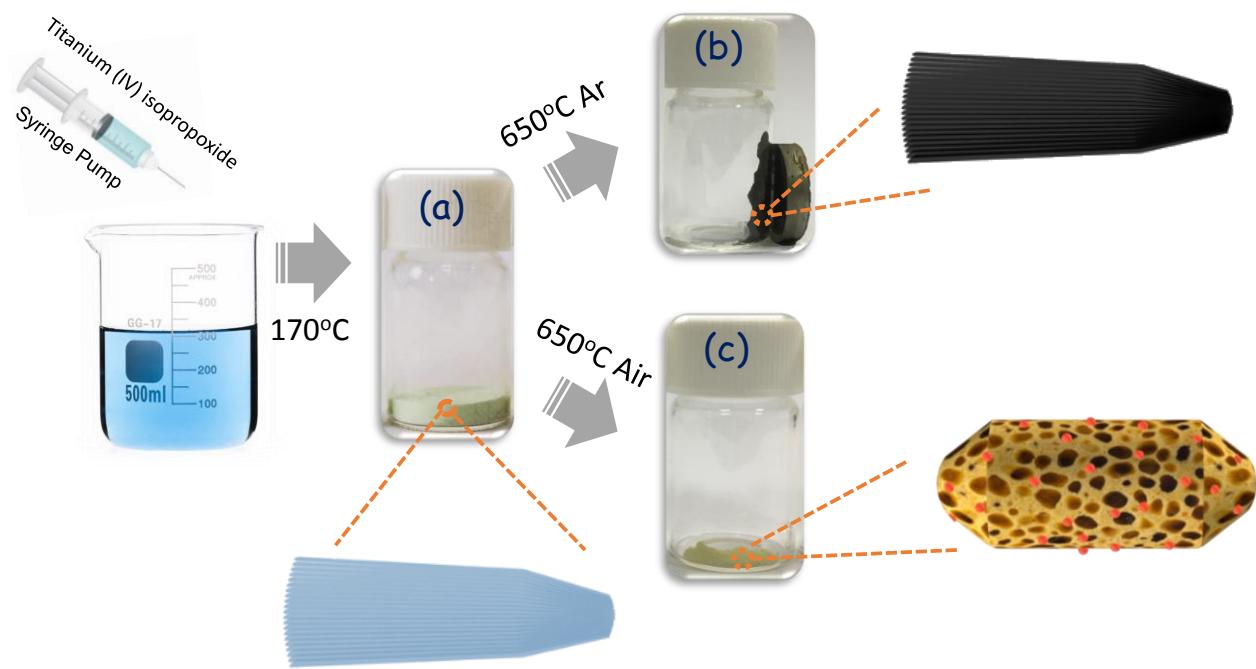
**Figure S3.** TEM images of (a)  $\text{TiO}_2$ , (b)  $\text{TiO}_2:\text{Ni}$  (1%), (c)  $\text{TiO}_2:\text{Ni}$  (2%).

#### 4. Elemental composition

**Table S1.** Ti and Ni atomic concentrations of  $\text{TiO}_2$ ,  $\text{TiO}_2:\text{Ni}$  (1%, 2%, 5%) and  $\text{NiTiO}_3/\text{TiO}_2$  (1%, 2%, 5%). XRD data refers to the  $\text{TiO}_2$  and  $\text{NiTiO}_3$  crystallographic phases detected and it does not take into account any peak shift denoting the presence of Ni within the  $\text{TiO}_2$  structure.

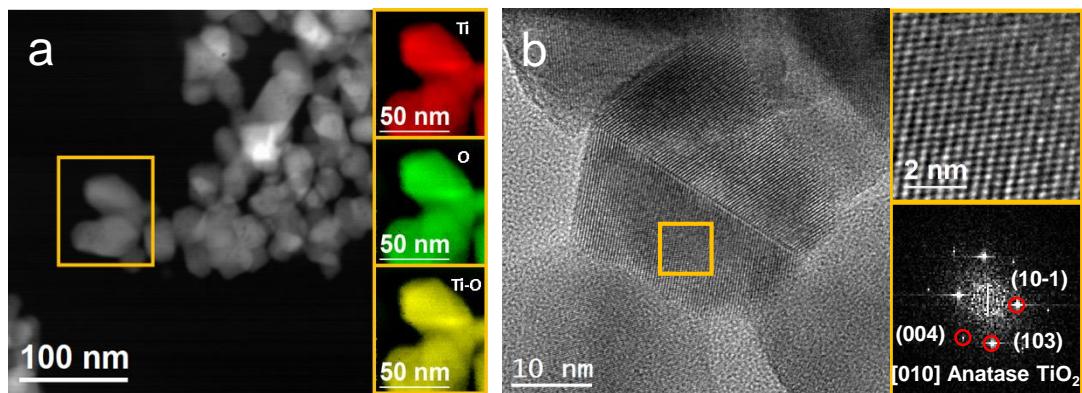
catalysts	EDX		XRD		XPS	
	Ti (atom%)	Ni (atom%)	Ti (atom%)	Ni (atom%)	Ti (atom%)	Ni (atom%)
$\text{TiO}_2$	100					
$\text{TiO}_2:\text{Ni}$ (1%)	96.4	3.6			94.3	5.7
$\text{TiO}_2:\text{Ni}$ (2%)	93.6	6.4				
$\text{TiO}_2:\text{Ni}$ (5%)	83.2	16.8				
$\text{NiTiO}_3/\text{TiO}_2$ (1%)	96.4	3.6	4.4	95.6	94.2	5.8
$\text{NiTiO}_3/\text{TiO}_2$ (2%)	94.3	5.7	6.5	93.5		
$\text{NiTiO}_3/\text{TiO}_2$ (5%)	86	14	13.3	86.7		
$\text{NiTiO}_3$	61.9	39.1				

## 5. Scheme of the preparation procedure



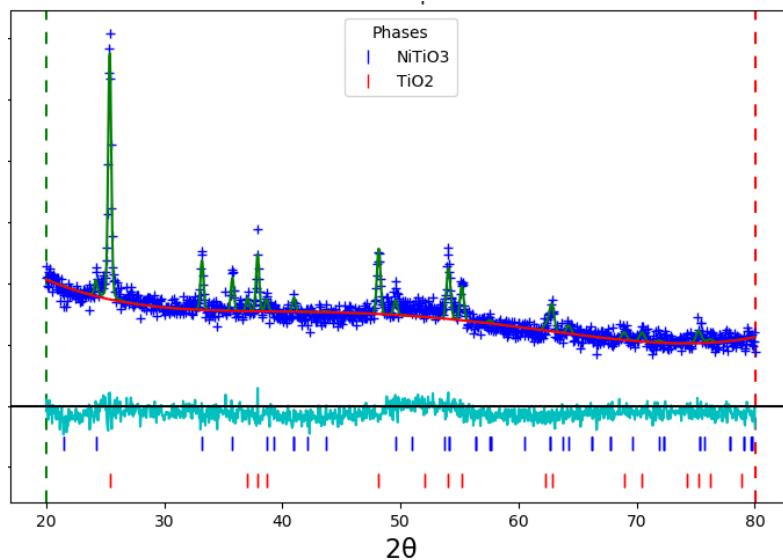
**Figure S4.** Scheme of the TiO<sub>2</sub>:Ni precursor preparation procedure: (a) TiO<sub>2</sub>:Ni, (b) NiO<sub>x</sub>/TiO<sub>2</sub>, (c) NiTiO<sub>3</sub>/TiO<sub>2</sub>.

## 6. EELS elemental maps and HRTEM of TiO<sub>2</sub>



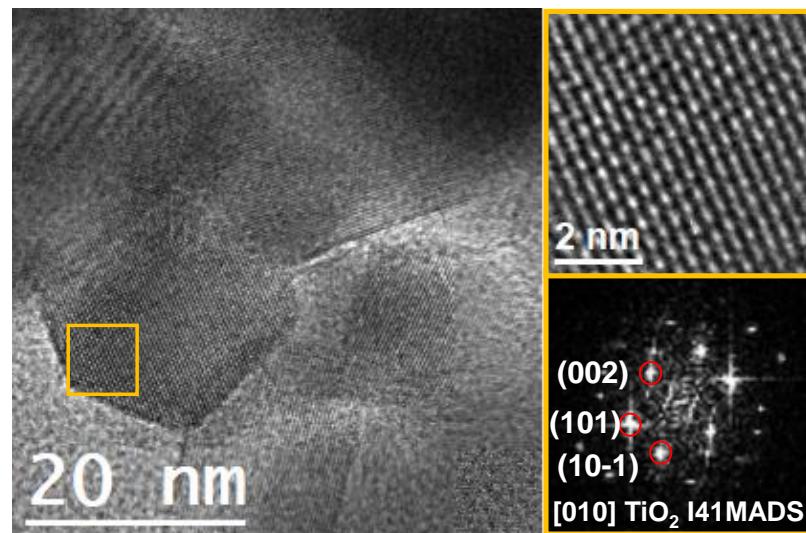
**Figure S5.** Undoped TiO<sub>2</sub> annealed at 650 °C in air: (a) STEM micrograph and EELS chemical composition maps obtained from the yellow squared area of the STEM micrograph. Individual Ti L<sub>2,3</sub>-edges at 456 eV (red) and O K-edge at 532 eV (green) as well as its composite. (b) HRTEM micrograph, detail of the yellow squared region and its corresponding power spectrum.

## 7. XRD Rietveld refinement



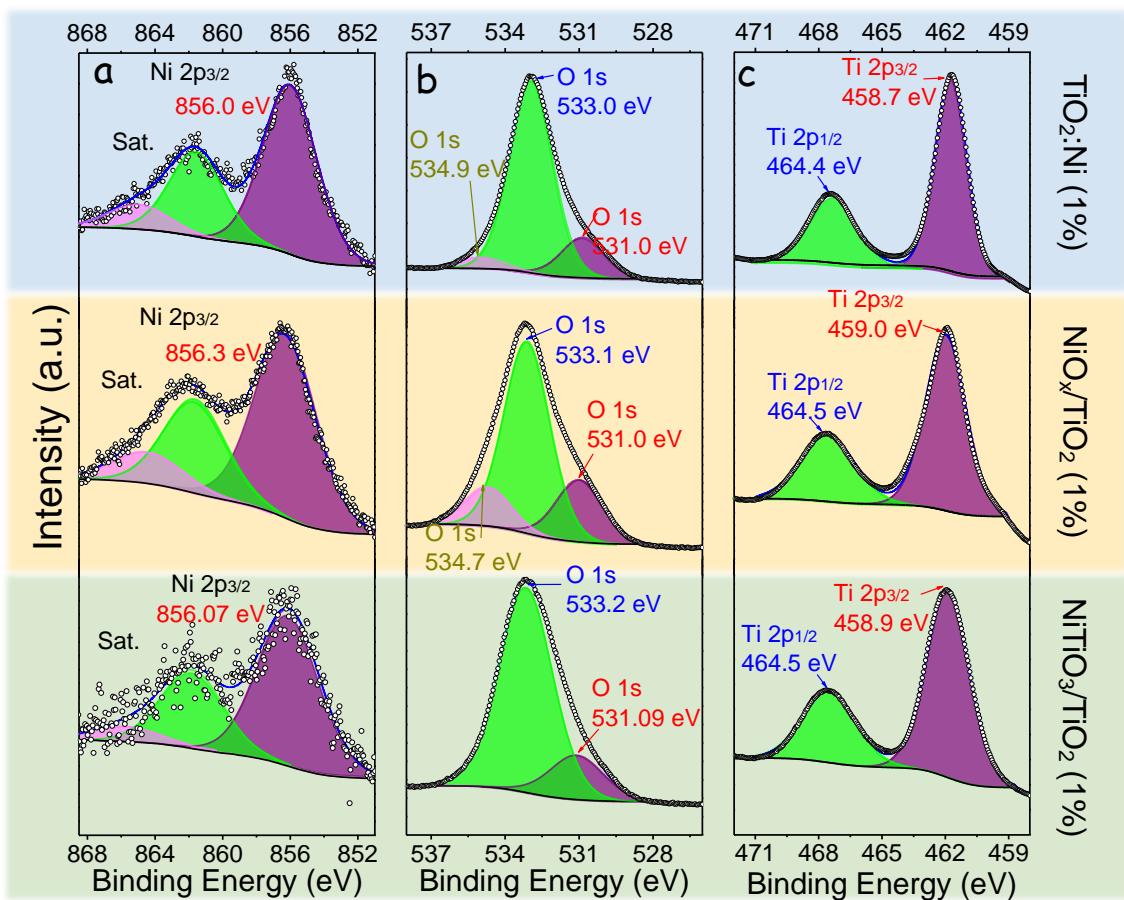
**Figure S6** Refined fitting of the NiTiO<sub>3</sub>/TiO<sub>2</sub>(5%) x-ray diffraction data. Blue symbols: experimental data; continuous red line: modified background; continuous green line: calculated modelled structure; continuous light blue line beneath pattern: difference between observed and calculated parameters. Blue tickmarks correspond to reflections of NiTiO<sub>3</sub> (R-3) unit cell, lower red ones to TiO<sub>2</sub> (I41/amd) unit cell. GOF = 1.21.  $R_w$  = 6.99%.

## 8. HRTEM characterization of NiTiO<sub>3</sub>/TiO<sub>2</sub> (1%)

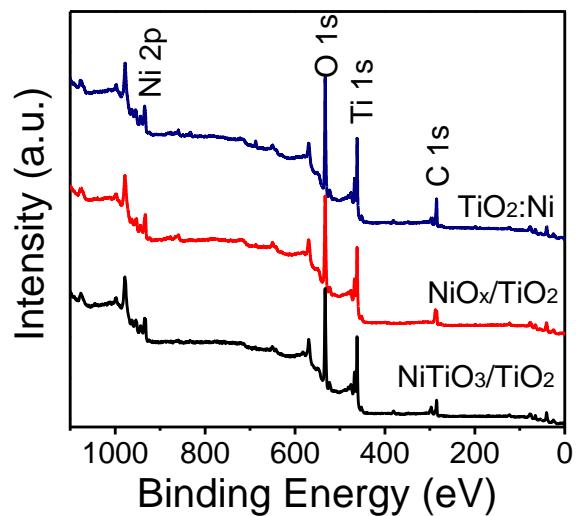


**Figure S7** HRTEM micrographs of NiTiO<sub>3</sub>/TiO<sub>2</sub> (1%), detail of the yellow squared region and its corresponding power spectrum.

## 9. XPS analyses

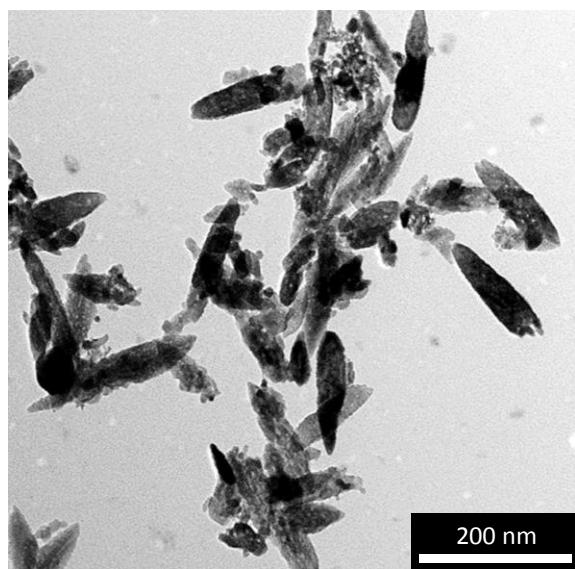


**Figure S8.** Ni 2p<sub>3/2</sub> region (a), O 1s region (b) and Ti 2p region (c) of the XPS spectra of (1) TiO<sub>2</sub>:Ni (1%) (2) NiO<sub>x</sub>/TiO<sub>2</sub> (1%) and (3) NiTiO<sub>3</sub>/TiO<sub>2</sub> (1%) samples.

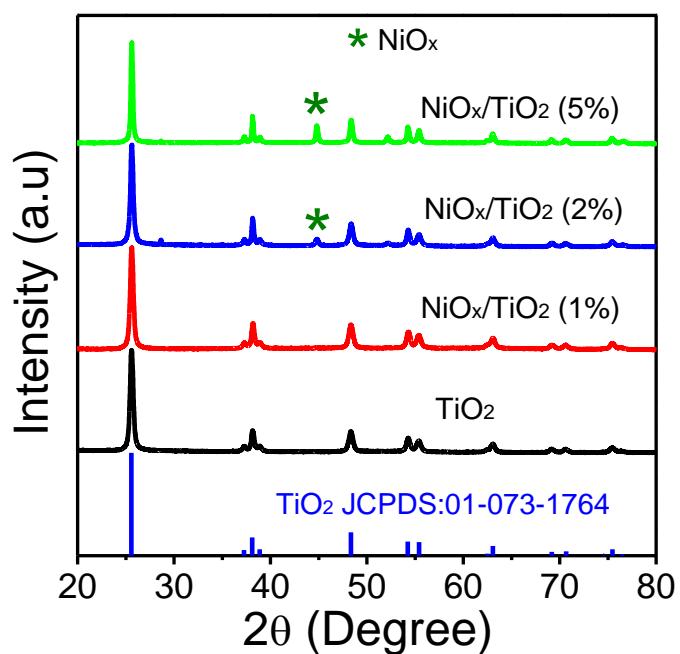


**Figure S9.** Survey XPS spectra of TiO<sub>2</sub>:Ni (1%), NiO<sub>x</sub>/TiO<sub>2</sub> (1%) and NiTiO<sub>3</sub>/TiO<sub>2</sub> (1%).

## 10. $\text{TiO}_2:\text{Ni}$ annealed in argon

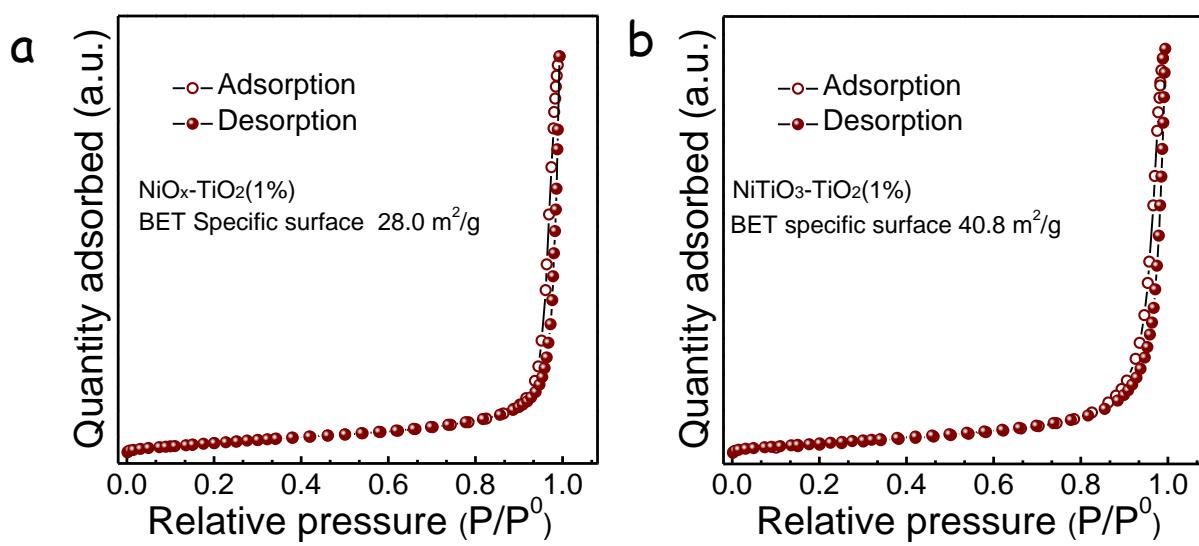


**Figure S10.** TEM image of  $\text{NiO}_x/\text{TiO}_2$  (5%) annealed in argon.



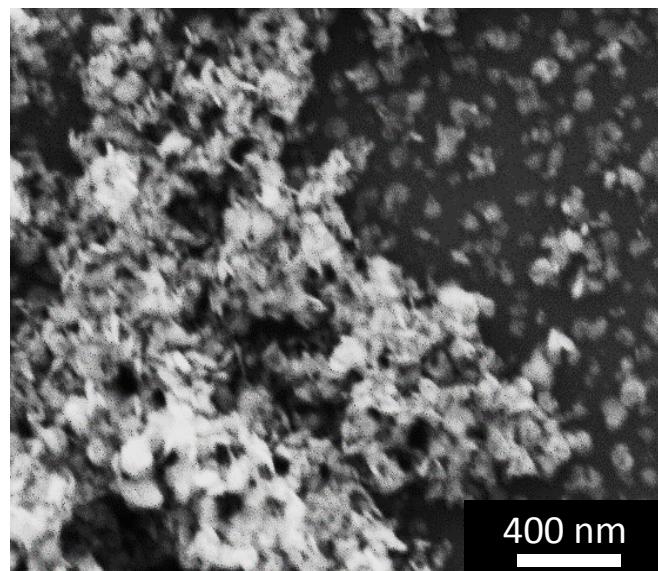
**Figure S11.** XRD pattern of  $\text{TiO}_2$  and  $\text{NiO}_x/\text{TiO}_2$  (1%, 2%, 5%) annealed in argon.

## 11.Nitrogen adsorption-desorption isotherms

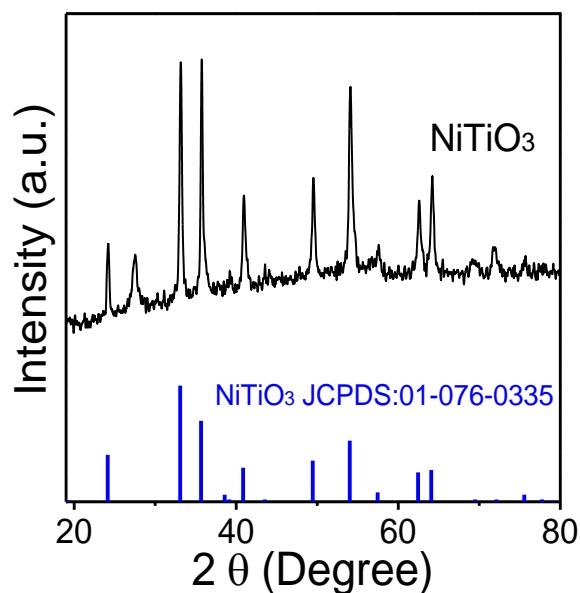


**Figure S12.** Nitrogen adsorption (open symbols) and desorption (filled symbols) isotherms measured of (a)  $\text{NiO}_x\text{-TiO}_2(1\%)$  and (b)  $\text{NiTiO}_3\text{-TiO}_2(1\%)$  at 77.3 K.

## 12. NiTiO<sub>3</sub> reference material



**Figure S13.** SEM images of NiTiO<sub>3</sub>



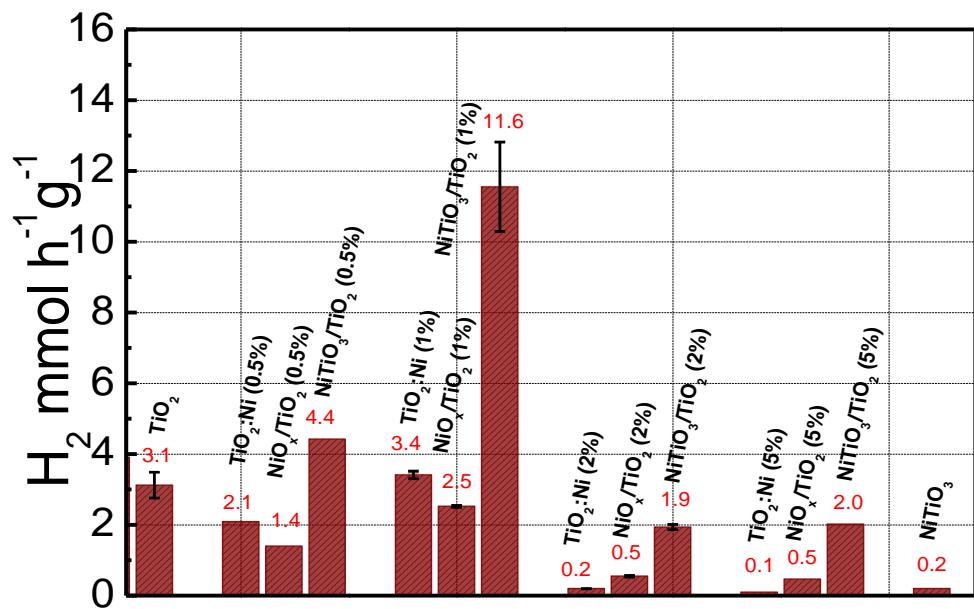
**Figure S14.** XRD pattern of NiTiO<sub>3</sub>

### 13. Literature comparison

**Table S2.** Comparison of the hydrogen evolution rate and the apparent quantum yield with reported Ni-Ti-O systems.

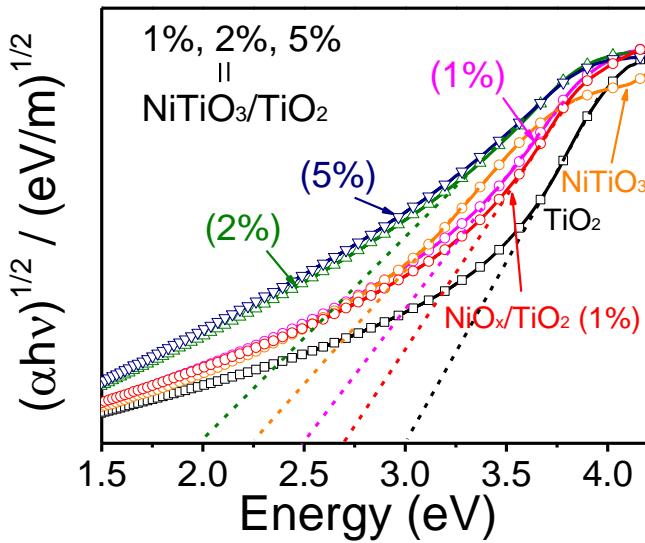
Photocatalyst	H <sub>2</sub> Evolution Rate μmol h <sup>-1</sup> g <sup>-1</sup>	Reaction conditions	AQY %	Ref.
Ni/NiO/N-TiO <sub>2-x</sub>	185	110W λ>420nm	7.5	S1
0.23%Ni(OH) <sub>2</sub> on TiO <sub>2</sub>	900	3W 365nm	12.4	S2
0.25wt% NiO-TiO <sub>2</sub>	261	3W 365nm	1.7	S3
TiO <sub>2</sub> -Ni(HCO <sub>3</sub> ) <sub>2</sub> -2.5%	377	300W 380nm (±5nm)	6.24	S4
0.32% Ni(NO <sub>3</sub> ) <sub>2</sub> -TiO <sub>2</sub>	163	3W 365nm	8.1	S5
Mesoporous NiO/TiO <sub>2</sub>	240	3W 365nm	1.7	S6
Pt NiO/TiO <sub>2</sub> 1:1 molar ratio	1,250	400W UV	7.8	S7
Hollow NiTiO <sub>3</sub> /TiO <sub>2</sub> (1%)	11,500	365nm	11.6	This work

## 14. Photocatalytic H<sub>2</sub> production rates



**Figure S15.** Photocatalytic H<sub>2</sub> production rates obtained from TiO<sub>2</sub>, NiTiO<sub>3</sub>, TiO<sub>2</sub>:Ni (1%), NiO<sub>x</sub>/TiO<sub>2</sub> (1%), NiTiO<sub>3</sub>/TiO<sub>2</sub> (1%, 2%, 5%).

## 15.UV-vis analysis



**Figure S16.** Kubelka-Munk function for  $\text{TiO}_2$ ,  $\text{NiTiO}_3/\text{TiO}_2$  (1, 2, 5%) and  $\text{NiO}_x/\text{TiO}_2$  including a linear fit (dashed lines) to determine the band gap energy.

## References

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