Asymmetric PPCs: Strongly fluorescing NIR labels

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Experimental Section

General

Solvents were purified according to standard procedures. All commercially available reagents were used without further purification unless otherwise noted. Column chromatography was performed on Roth silica gel 60 (40-63 µm). All solvents used for UV/Vis/NIR and fluorescence measurements were spectroscopic grade and purchased from Fluka. NMR spectra were recorded on a 400 MHz Bruker Avance III spectrometer equipped with a BBFO probe with a Z gradient. The residual solvent peak was used as internal reference $^1$H: [CHCl$_3$ δ = 7.24 ppm; C$_2$DHCl$_4$ δ = 5.91 ppm; DMSO-d$_5$ δ = 2.50 ppm], $^{13}$C: [CDC$_3$ δ = 77.23 ppm]. For $^1$H-NMR, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, qui = quintet and m = multiplet) and coupling constant(s) were reported whenever possible. Spectra were analyzed with MestreNova. MALDI-TOF mass spectra were recorded on a Bruker Biflex III instrument in positive reflection mode. 2,5-Dihydroxybenzoic acid (DHB) or α-cyano-4-hydroxycinnamic acid (CHCA) were used as matrix. Elemental analyses were performed in the microanalytical lab of University of Konstanz with an CHN-analyzer Vario Micro Cube from Elementar. Absorption and emission were recorded at ambient temperature using quartz 1 cm cuvettes (3 ml). UV/Vis/NIR absorption were recorded using a Varian spectrometer, model Cary 50. The spectra were processed with Spekwin$^{[13]}$ to calculate the oscillator strenghts. UV/Vis-fluorescence was recorded at a self assembled spectrometer with a Xe flashlight lamp as light source and a nitrogen cooled CCD camera as detector.$^{[14]}$ NIR-fluorescence was recorded at a self assembled NIR fluorescence spectrometer with a nitrogen cooled Ge diode (Northcoast) as detector.$^{[15]}$ Either a diode laser (804 nm, 30 mW, model ACM30/1476 or 690 nm, 19 mW, model ACM19/1203) or a He-Ne laser (632.8 nm, 5 mW, Spectra Physics model 105-1) were used for excitation. The BF$_2$-PPCy 7e was used as reference to determine the quantum yields (QY = 0.59 in CHCl$_3$).$^{[6a]}$
2-(6-bromoquinoline-2-yl)acetonitrile (3f)

![Chemical structure of 3f](image)

6-bromochinaldine (3f’)[16]

100 g (581 mmol) 4-bromoaniline were refluxed in 300 ml 6 M HCl. 51 g (727 mmol) crotonaldehyde were added over a period of 2 h and the mixture was refluxed for 1 h. For purification 3f’ was precipitated as zinc complex. Therefore, 79.2 g (581 mmol) ZnCl$_2$ were added to the hot reaction mixture which then was cooled to 0°C. The precipitate was filtered and dried on air. The crude product was suspended in methylene chloride, filtered and washed with methylene chloride until the filtrate was colourless. The zinc complex was added to 300 ml water and 300 ml concentrated ammonia. The mixture was extracted (3x) with 200 ml methylene chloride. The combined organic layers were dried over MgSO$_4$. Removing the solvent yielded 70.21 g (320 mmol, 55 %) 3f’ as a yellow solid. $^1$H-NMR (400 MHz, CDCl$_3$): δ/ppm = 7.98 (d, $^3$J = 8.5 Hz, 1 H; H-4), 7.93 (m, 2 H; H-5, H-8), 7.75 (dd, $^3$J = 8.9 Hz, $^4$J = 2.2 Hz, 1 H; H-7), 7.31 (d, $^3$J = 8.5 Hz, 1 H; H-3), 2.75 (s, 3 H; CH$_3$).

2-chloromethyl-6-bromoquinoline (3f’)[17]

50 g (225 mmol) 3f’ were dissolved in 200 ml chloroform and 19.3 g (83 mmol) trichloroisocyanuric acid were added in portions. The mixture was heated to reflux for 1 h. Then the cyanuric acid was filtered over celite and washed with chloroform. The organic layer was washed with 0.01 M HCl and dried over MgSO$_4$. Removing the solvent yielded 49.3 g crude 3f’ as a yellow solid that was used without further purification. $^1$H-NMR (400 MHz, CDCl$_3$): δ/ppm = 8.12 (d, $^3$J = 8.5 Hz, 1 H; H-4), 7.99 (d, $^4$J = 2.2 Hz, 1 H; H-5), 7.94 (d, $^3$J = 8.8 Hz, 1 H; H-8), 7.80 (dd, $^3$J = 8.8 Hz, $^4$J = 2.2 Hz, 1 H; H-7), 7.63 (d, $^3$J = 8.5 Hz, 1 H; H-3), 4.82 (s, 2 H; CH$_2$).

2-(6-bromoquinoline-2-yl)acetonitrile (3f)

11.03 g (225 mmol) powdered NaCN, a small amount of NaI and 49.3 g 3f’ were stirred in 200 ml DMF at room temperature for 6 h. After removing the solvent the remaining residue was solved in methylene chloride and filtered over celite. The organic layer was washed with water, dried over MgSO$_4$ and the solvent was removed. Column chromatography (methylen chloride) yielded 17.18 g (70 mmol, 31 % relating to 3f’) 3f. $^1$H-NMR (400 MHz, CDCl$_3$): δ/ppm = 8.15 (d, $^3$J = 8.5 Hz, 1 H; H-4), 8.00 (d, $^4$J = 2.2 Hz, 1 H; H-5), 7.96 (d, $^3$J = 9.0 Hz, 1 H; H-8), 7.81 (dd, $^3$J = 9.0 Hz, $^4$J = 2.2 Hz, 1 H; H-7), 7.56 (d, $^3$J = 8.5 Hz, 1 H;
tert-butyl 5-(2-(cyanomethyl)quinolin-6-yl)pent-4-ynoate (3g)

2.95 g (30 mmol) 4-pentynoic acid, 4.45 g (60 mmol) tert-butanole and 0.147 g (1.2 mmol) 4-(dimethyl-amino) pyridine were stirred in 10 ml methylene chloride at room temperature. A solution of 6.81 g (33 mmol) DCC in 10 ml methylene chloride was added at once. The mixture was stirred over night. The carbamide was filtered and washed with methylene chloride. The filtrate was washed with 0.5 M HCl (2x) and NaHCO₃ solution (2x), dried over MgSO₄ and the solvent was removed. Column chromatography (methylene chloride) yielded 3.98 g (25.8 mmol, 86%) 3h’. ¹H-NMR (400 MHz, CDCl₃): δ/ppm = 2.42 (m, 4 H; CH₂CH₂), 1.93 (m, 1 H; CH), 1.42 (s, 9 H; tert-butyl); ¹³C-NMR (100 MHz, CDCl₃): δ/ppm = 171.08, 82.77, 80.83, 68.77, 34.49, 28.08, 14.49.

tert-butyl 5-(2-(cyanomethyl)quinolin-6-yl)pent-4-ynoate (3h)

3.0 g (12.1 mmol) 3f and 3.0 g (19.5 mmol) 3h’ were dissolved in 25 ml abs. THF and 10 ml triethylamine. The mixture was degassed and 69 mg (0.03 mmol) CuI and 419 mg (0.03 mmol) tetrakis(triphenylphosphine)palladium(0) were added. The mixture was heated to 60°C over night. The mixture was added to 300 ml methylene chloride. The organic phase was washed (3x) with 300 ml water, dried over MgSO₄ and the solvent was removed. Column chromatography (petroleum ether/ethyl acetate 5/2) yielded 3.23 g (10.6 mmol, 83%) 3h. ¹H-NMR (400 MHz, CDCl₃): δ/ppm = 8.15 (d, ³J = 8.5 Hz, 1 H; H-4), 7.97 (d, ³J = 8.8 Hz, 1 H; H-8), 7.85 (d, ⁴J = 1.8 Hz, 1 H; H-5), 7.68 (dd, ³J = 8.8 Hz, ⁴J = 1.8 Hz, 1 H; H-7), 7.52 (d, ³J = 8.5 Hz, 1 H; H-3), 4.13 (s, 2 H; CH₂), 2.72 (t, ³J = 7.3 Hz, 2 H; COCH₂), 2.55 (t, ³J = 7.3 Hz, 2 H; COCH₂), 1.46 (s, 9 H; tert-butyl); ¹³C-NMR (100 MHz, CDCl₃): δ/ppm = 171.32, 150.80, 146.64, 137.99, 133.77, 130.80, 128.77, 127.18, 123.12, 120.48, 116.80, 90.94, 81.10, 80.58, 34.85, 28.32, 27.31, 15.82; elemental analysis calc’d (%) for C₂₀H₂₀N₂O₂ [M = 320.38 g/mol]: C 74.98 H 6.29 N 8.74; found: C 74.65 H 6.36 N 8.44.

H-3), 4.13 (s, 2 H; CH₂); ¹³C-NMR (100 MHz, CDCl₃): δ/ppm = 151.00, 146.06, 137.47, 134.37, 130.55, 129.94, 128.51, 121.57, 120.87, 116.65, 27.33; elemental analysis calc’d (%) for C₁₁H₇BrN₂ [M = 247.09 g/mol]: C 53.47 H 2.86 N 11.34; found: C 53.20 H 2.93 N 11.08.
5 g (9.17 mmol) 4 were refluxed in 6 ml POCl₃ for 2 h. The POCl₃ was distilled off and the residue was dried on vacuum. 2.11 g (9.18 mmol) 3d and 60 ml abs. THF were added and the mixture was heated to reflux for 3 h. The solvent was distilled off, 100 ml methanol and 10 ml water were added and the mixture was stirred over night. The solid was filtered off and washed with methanol, then suspended in methanol again, filtered off and washed. This procedure was repeated until the filtrate was colourless. 2.77 g (3.66 mmol, 40%) of 4d were obtained as a blue crystalline powder that was pure enough for further reactions. For analytic a small amount was further purified by column chromatography (gradient: methylene chloride to methylene chloride/ethyle acetate 5/1). ¹H-NMR (400 MHz, C₂D₂Cl₄): δ/ppm = 13.33 (s, 1 H; N-H), 8.59 (s, 1 H; N-H’), 8.20 (m, 2 H; o’), 7.75 (m, 1 H; H-4), 7.73 (m, 1 H; H-7), 7.50 (m, 2 H; o), 7.45 (m, 1 H; H-5), 6.98 (m, 4 H; m, m’), 3.98 (m, 4 H; OCH₂), 1.76 (m, 4 H; OCH₂CH₂), 1.5-1.15 (m, 20 H; alkyl), 1.31 (s, 9 H; tert-butyl), 0.83 (m, 6 H; CH₃); UV/Vis/NIR (chloroform): \( \lambda_{\text{max}}(\epsilon) = 618 \) (33000), 440 nm (18000 M⁻¹ cm⁻¹); MALDI-MS: m/z calcd: 756.4 [M+H]+; 778.4 [M+Na]+ found 756.5; 778.5; elementa analysis calcd (%) for C₄₇H₅₆N₄O₃S [M = 757.04 g/mol]: C 74.57 H 7.46 N 7.40; found: C 74.36 H 7.53 N 7.43.
4 g (7.34 mmol) 1 were refluxed in 5 ml POCl₃ for 1.5 h. The POCl₃ was distilled off und the residue was dried on vacuum. 1.65 g (7.34 mmol) 3e and 80 ml abs. THF were added and the mixture was heated to reflux for 2 h. The solvent was distilled off and 100 ml methanol and 10 ml water were added. The mixture was stirred over night. The solid was filtered off and washed with methanol, then suspended in methanol again, filtered off and washed. This procedure was repeated until the filtrate was colourless. 4.61 g (6.14 mmol, 84%) of 4e were obtained as a blue crystalline pouder that was pure enough for further reactions. For analytic a small amount was further purified by column chromatography (gradient: methylene chloride to methylene chloride/ethyle acetate 5/1). ¹H-NMR (400 MHz, C₂D₂Cl₄): δ/ppm = 14.91 (s, 1 H; N-H), 8.36 (m, 2 H; o’), 8.04 (d, ³J = 8.8 Hz, 1 H; H-4), 7.85 (d, ³J = 9.0 Hz, 1 H; H-8), 7.81 (dd, ³J = 9.0 Hz, ⁴J = 1.9 Hz, 1 H; H-7), 7.67 (d, ³J = 8.8 Hz, 1 H; H-3), 7.64 (d, ⁴J = 1.9 Hz, 1 H; H-5), 7.59 (s, 1 H; N-H’), 7.48 (m, 2 H; o), 7.10 (m, 2 H; m’), 7.01 (m, 2 H; m), 4.03 (t, ³J = 6.6 Hz, 2 H; OCH₂), 3.97 (t, ³J = 6.6 Hz, 2 H; OCH₂), 1.77 (m, 4 H; OCH₂CH₂), 1.45-1.15 (m, 20 H; alkyl), 1.37 (s, 9 H; tert-butyl), 0.83 (m, 6 H; CH₃); UV/Vis/NIR (chloroform): λ_max(ε) = 611 (37000), 413 nm (21000 M⁻¹cm⁻¹); MALDI-MS: m/z calcd: 751.5 [M+H]+, found 751.2; elemental analysis calcd (%) for C₄₉H₅₈N₄O₃ [M = 751.01 g/mol]: C 78.36 H 7.78 N 7.46; found: C 78.34 H 7.74 N 7.55.

5d

561 mg (0.741 mmol) 4d and 1.2 ml (7.41 mmol) N,N-diisopropylethylamine were refluxed in 50 ml methylene chloride. 1.86 ml (14.8 mmol) boron trifluoride etherate were added and the mixture was refluxed for 30 min. The organic phase was washed (3x) with water, dried over MgSO₄ and the solvent was removed. Column chromatography (gradient: methylene chloride to methylene chloride/ethyle acetate 20/1) yielded 262 mg (0.326 mmol, 44%) 5d as a dark green crystalline pouder. ¹H-NMR (400 MHz, CDCl₃): δ/ppm = 8.14 (m, 2 H; o’), 7.97 (d, ³J = 8.8 Hz, 1 H; H-4), 7.69 (d, ⁴J = 1.9 Hz, 1 H; H-7), 7.63 (m, 2 H; o’), 7.51 (dd, ³J = 8.8 Hz, ⁴J = 1.9 Hz, 1 H; H-5), 7.24 (s, 1 H; N-H), 7.07 (m, 2 H; m), 7.02 (m, 2 H; m’), 4.05 (m, 4 H; OCH₂), 1.82 (m, 4 H; OCH₂CH₂), 1.5-1.2 (m, 20 H; alkyl), 1.35 (s, 9 H; tert-butyl), 0.88 (m, 6 H; CH₃); UV/Vis/NIR (chloroform): λ_max(ε) = 618 (43000),
430 nm (13000 M$^{-1}$cm$^{-1}$); MALDI-MS: m/z calcd: 805.4 [M+H]$^+$, found 805.0; elemental analysis calcd (%) for C$_{47}$H$_{55}$BF$_2$N$_4$O$_3$S [M = 804.84 g/mol]: C 70.14 H 6.89 N 6.96; found: C 69.97 H 6.68 N 6.84.

5e

500 mg (0.666 mmol) 4e and 1.1 ml (6.66 mmol) N,N-diisopropylethylamine were refluxed in 50 ml methylene chloride. 1.67 ml (13.3 mmol) boron trifluoride etherate were added and the mixture was refluxed for 30 min. The organic phase was washed (3x) with water, dried over MgSO$_4$ and the solvent was removed. Column chromatography (gradient: methylene chloride to methylene chloride/ethyle acetate 20/1) yielded 328 mg (0.478 mmol, 72%) 5e as a dark green crystalline powder.

$^1$H-NMR (400 MHz, C$_2$D$_2$Cl$_4$): $\delta$/ppm = 8.48 (d, $^3$J = 9.3 Hz, 1 H; H-8), 8.08 (m, 2 H; o’), 8.04 (d, $^3$J = 9.0 Hz, 1 H; H-4), 7.81 (s, 1 H; N-H), 7.73 (dd, $^3$J = 9.3 Hz, $^4$J = 2.0 Hz, 1 H; H-7), 7.63 (d, $^3$J = 9.0 Hz, 1 H; H-3), 7.59 (d, $^4$J = 2.0 Hz, 1 H; H-5), 7.53 (m, 2 H; o), 7.03 (m, 2 H; m), 7.00 (m, 2 H; m’), 4.01 (m, 4 H; OCH$_2$), 1.77 (m, 4 H; OCH$_2$CH$_2$), 1.45-1.15 (m, 20 H; alkyl), 1.32 (s, 9 H; tert-butyl), 0.83 (m, 6 H; CH$_3$); UV/Vis/NIR (chloroform): $\lambda_{\text{max}}$ (ε) = 630 (58000), 421 nm (15000 M$^{-1}$cm$^{-1}$); MALDI-MS: m/z calcd: 799.5 [M+H]$^+$, found 799.5; elemental analysis calcd (%) for C$_{49}$H$_{57}$BF$_2$N$_4$O$_3$ [M = 798.81 g/mol]: C 73.68 H 7.19 N 7.01; found: C 73.67 H 7.17 N 6.96.

General procedure for the synthesis of the asymmetric H-PPCy dyes 6:

Under nitrogen atmosphere, 1 mmol half converted PPCy (4) and 1.2 mmol heteroarylacetoniitril (3) are refluxed in absolute toluene and 4 mmol POCl$_3$ are added. The reaction is monitored by UV/Vis/NIR spectroscopy and thin-layer chromatography. As soon as 4 is used up the reaction is stopped. Solvent and excess POCl$_3$ are removed under vacuum. The crude product is treated with methanol in an ultrasonic bath and water is added until a solid precipitates. The solid is filtered off and washed with methanol until the filtrate is colourless. The remaining solid is purified by column chromatography.
H-PPCy 6a

Column chromatography (CH$_2$Cl$_2$/ethylacetate 50/1) yielded 70 mg (79.5 µmol, 66%) 6a as a greenish blue crystalline powder. $^1$H-NMR (400 MHz, CDCl$_3$): δ/ppm = 14.1 (br., 1 H; N-H), 13.56 (s, 1 H; N-H'), 7.93 (d, $^3$J = 8.8 Hz, 1 H; H-4), 7.82 (d, 1 H; H-8), 7.73 (m, 3 H; o, H-7), 7.64 (m, 2 H; o'), 7.61 (m, 1 H; H-5), 7.58 (d, $^3$J = 8.8 Hz, 1 H; H-3), 7.11 (m, 2 H; m), 6.96 (m, 2 H; m'), 6.67 (s, 1 H; H-5'), 4.07 (t, $^3$J = 6.6 Hz, 2 H; OCH$_2$), 3.98 (t, $^3$J = 6.6 Hz, 2 H; OCH$_2$'), 2.46 (s, 6 H; CH$_3$), 1.83 (m, 4 H; OCH$_2$CH$_2$), 1.55-1.2 (m, 20 H; alkyl), 1.41 (s, 9 H; tert-butyl), 0.90 (m, 6 H; CH$_3$); UV/Vis/NIR (chloroform): $\lambda_{\text{max}}$ (ε) = 708 (100000), 645 nm (39000 M$^{-1}$cm$^{-1}$); MALDI-MS: m/z calcd: 880.5 [M+H]$^+$, found 880.2; elemental analysis calcd (%) for C$_{57}$H$_{65}$N$_7$O$_2$ [M = 880.17 g/mol]: C 77.78 H 7.44 N 11.14; found: C 77.37 H 7.27 N 11.02.

H-PPCy 6b

Column chromatography (CH$_2$Cl$_2$) yielded 330 mg (0.35 mmol, 87%) 6b as a dark-green crystalline powder. $^1$H-NMR (400 MHz, C$_2$D$_2$Cl$_4$): δ/ppm = 14.90 (s, 1 H; N-H), 12.24 (s, 1 H; N-H'), 8.02 (d, $^3$J = 8.8 Hz, 1 H; H-4), 7.75 (m, 4 H; o, H-7, H-8), 7.66 (m, 3 H; o', H-3), 7.61 (m, 1 H; H-5), 7.54 (m, 1 H; H-4'), 7.35 (d, $^3$J = 8.8 Hz, 1 H; H-7'), 7.26 (m, 1 H; H-6'), 7.15 (m, 2 H; m), 7.10 (m, 2 H; m'), 4.04 (m, 4 H; OCH$_2$), 1.79 (m, 4 H; OCH$_2$CH$_2$), 1.5-1.2
(m, 20 H; alkyl), 1.36 (s, 9 H; tert-butyl), 1.33 (s, 9 H; tert-butyl), 0.84 (m, 6 H; CH₃);
**UV/Vis/NIR** (chloroform): $\lambda_{\text{max}}(\varepsilon) = 720$ (108000), 655 nm (44000 M⁻¹cm⁻¹);
**MALDI-MS:** m/z calcd: 947.6 [M+H]+, found 947.3; **elemental analysis** calcd (%) for C₆₂H₇₀N₆O₃
[M = 947.26 g/mol]: C 78.61 H 7.45 N 8.87; found: C 78.41 H 7.57 N 8.90.

**H-PPCy 6c**

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Column chromatography (CH₂Cl₂) yielded 150 mg (0.17 mmol, 87%) 6c as a dark-green crystalline powder. **¹H-NMR** (400 MHz, C₂D₂Cl₄): δ/ppm = 14.76 (s, 1 H; N-H), 14.12 (s, 1 H; N-H'), 7.98 (d, 3J = 8.8 Hz, 1 H; H-4), 7.7 (m, 8 H; o, o', H-3, H-5, H-7, H-8), 7.54 (m, 1 H; H-4'), 7.37 (m, 1 H; H-3'), 7.11 (m, 2 H; m), 7.07 (m, 2 H; m'), 6.86 (m, 1 H; H-5'), 4.01 (m, 4 H; OCH₂), 2.52 (s, 3 H; CH₃), 1.78 (m, 4 H; OCH₂CH₂), 1.50-1.15 (m, 20 H; alkyl), 1.35 (s, 9 H; tert-butyl), 0.84 (m, 6 H; CH₃); **UV/Vis/NIR** (chloroform): $\lambda_{\text{max}}(\varepsilon) = 718$ (97000), 653 nm (39000 M⁻¹cm⁻¹); **MALDI-MS:** m/z calcd: 865.5 [M+H]+, found 865.2; **elemental analysis** calcd (%) for C₅₇H₆₄N₆O₂ [M = 865.16 g/mol]: C 79.13 H 7.46 N 9.71; found: C 79.16 H 7.53 N 9.69.

**H-PPCy 6d**

![Image](image_url)

**Batch:**

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Column chromatography (CH$_2$Cl$_2$) yielded 202 mg (0.21 mmol, 79%) 6d as a green crystalline powder. $^1$H-NMR (400 MHz, C$_2$D$_2$Cl$_4$): $\delta$/ppm = 14.85 (s, 1 H; N-H), 13.04 (s, 1 H; N-H'), 8.00 (d, $^3$J = 8.8 Hz, 1 H; H-4), 7.75 (m, 4 H; o, H-7, H-7'), 7.69 (d, 1 H; H-8), 7.68 (d, 1 H; H-4'), 7.64 (m, 3 H; o', H-3), 7.61 (d, $^4$J = 2.0 Hz, 1 H; H-5), 7.44 (dd, $^3$J = 8.8 Hz, $^4$J = 2.0 Hz, 1 H; H-5'), 7.14 (m, 2 H; m), 7.09 (m, 2 H; m'), 4.04 (m, 4 H; OCH$_2$), 1.79 (m, 4 H; OCH$_2$CH$_2$), 1.36 (s, 9 H; tert-butyl), 1.32 (s, 9 H; tert-butyl), 0.84 (m, 6 H; CH$_3$); UV/Vis/NIR (chloroform): $\lambda_{\text{max}}$(c) = 734 (116000), 667 nm (47000 M$^{-1}$cm$^{-1}$); MALDI-MS: m/z calcld: 963.5 [M+H]$^+$, found 963.2; elemental analysis calcld (%) for C$_{62}$H$_{70}$N$_6$O$_2$S [M = 963.32 g/mol]: C 77.30 H 7.32 N 8.72; found: C 77.42 H 7.31 N 8.65.

H-PPCy 6e$^6$

\[ \text{H-PPCy 6e} \]

H-PPCy 6f

\[ \text{H-PPCy 6f} \]

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Column chromatography (CH$_2$Cl$_2$) yielded 1.13 g (1.15 mmol, 87%) 6f as a green crystalline powder. $^1$H-NMR (400 MHz, C$_2$D$_2$Cl$_4$): $\delta$/ppm = 14.79 (s, 1 H; N-H), 14.44 (s, 1 H; N-H), 7.99 (d, $^3$J = 8.8 Hz, 1 H; H-4), 7.87 (d, $^3$J = 8.8 Hz, 1 H; H-4'), 7.82 (d, $^4$J = 1.9 Hz, 1 H; H-5'), 7.70 (m, 7 H; o, o', H-7, H-8, H-7'), 7.62 (m, 4 H; H-3, H-3',H-5, H-8'), 7.12 (m, 4 H; m, m'), 4.05 (m, 4 H; OCH$_2$), 1.79 (m, 4 H; OCH$_2$CH$_2$), 1.5-1.15 (m, 20 H; alkyl),
1.36 (s, 9 H; tert-butyl), 0.84 (m, 6 H; CH₃); UV/Vis/NIR (chloroform): λₓₘₐₓ(ε) = 734 (122000), 666 nm (45000 M⁻¹cm⁻¹); MALDI-MS: m/z calcld: 979.4 [M+H]⁺, found 979.5; elemental analysis calcld (%) for C₆₀H₆₃BrN₆O₂ [M = 980.09 g/mol]: C 73.53 H 6.48 N 8.57; found: C 73.49 H 6.48 N 8.62.

H-PPCy 6g

**Batch:**

- 4e 500 mg 0.66 mmol
- 3g 135 mg 0.80 mmol
- POCl₃ 405 mg 2.64 mmol 250 µl
- Toluol 20 ml

Column chromatography (CH₂Cl₂/ethylacetate 50/1) yielded 343 mg (0.38 mmol, 58%) 6g as a green crystalline powder. ¹H-NMR (400 MHz, C₂D₂Cl₄): δ/ppm = 14.80 (s, 1 H; N-H), 13.79 (s, 1 H; N-H'), 8.79 (s, 1 H; H-3'), 7.90 (d, 3J = 8.3 Hz, 1 H; H-8'), 7.85 (d, 3J = 8.9 Hz, 1 H; H-4), 7.76 (d, 3J = 8.3 Hz, 1 H; H-5'), 7.77 (m, 7 H; H-o,o', H-7, H-8, H-6'), 7.57 (m, 1 H; H-5), 7.55 (m, 1 H; H-7'), 7.42 (d, 3J = 8.9 Hz, 1 H; H-3), 7.09 (m, 4 H; m, m'), 4.03 (m, 4 H; OCH₂), 1.80 (m, 4 H; OCH₂CH₂), 1.5-1.15 (m, 20 H; alkyl), 1.36 (s, 9 H; tert-butyl), 0.83 (m, 6 H; CH₃); UV/Vis/NIR (chloroform): λₓₘₐₓ(ε) = 739 (132000), 671 nm (47000 M⁻¹cm⁻¹); MALDI-MS: m/z calcld: 902.5 [M+H]⁺; 924.5 [M+Na]⁺; 940.5 [M+K]⁺ found 903.0; 925.0; 940.9; elemental analysis calcld (%) for C₅₉H₆₃N₇O₂ [M = 902.18 g/mol]: C 78.55 H 7.04 N 10.87; found: C 78.19 H 7.05 N 10.92.

**General procedure for the synthesis of the Unsymmetric BF₂-PPCy dyes 7:**

1 mmol H-PPCy dye (6) and 20 mmol N,N-diisopropylethylamine are refluxed in methylene chloride. 40 mmol boron trifluoride etherate are added and the mixture is refluxed for 10 min. The mixture is washed with water and dried over MgSO₄. After removing the solvent the crude product is purified by column chromatography.
**BF₂-PPCy 7a**

Column chromatography (CH₂Cl₂) yielded 185 mg (0.19 mmol, 86%) 7a as a green crystalline powder. ¹H-NMR (400 MHz, C₂D₂Cl₄): δ/ppm = 8.33 (d, ³J = 9.2 Hz, 1 H; H-8), 7.93 (d, ³J = 9.1 Hz, 1 H; H-4), 7.70 (m, 1 H; H-7), 7.64 (m, 6 H; o,o’, H-3, H-5), 6.99 (m, 2 H; m), 6.96 (m, 2 H; m’), 6.75 (s, 1 H; H-5’), 4.00 (m, 4 H; OCH₂), 2.58 (s, 3 H; CH₃), 2.48 (s, 3 H; CH₃), 1.78 (m, 4 H; OCH₂CH₂), 1.55-1.2 (m, 20 H; alkyl), 1.29 (s, 9 H; tert-butyl), 0.82 (m, 6 H; CH₃); UV/Vis/NIR (chloroform): λmax(ε) = 719 (178000), 654 nm (43000 M⁻¹cm⁻¹); MALDI-MS: m/z calcd: 976.5 [M+H]+; 998.5 [M+Na]+; 1014.5 [M+K]+ found 975.9; 997.9; 1013.9; elemental analysis calcd (%) for C₅₇H₆₃B₂F₄N₇O₂ [M = 975.77 g/mol]: C 70.16 H 6.51 N 10.05; found: C 70.34 H 6.59 N 10.02.

**BF₂-PPCy 7b**

Column chromatography (CH₂Cl₂) yielded 128 mg (0.123 mmol, 78%) 7b as a green crystalline powder. ¹H-NMR (400 MHz, C₂D₂Cl₄): δ/ppm = 8.40 (d, ³J = 8.9 Hz, 1 H; H-8), 8.17 (d, ³J = 9.0 Hz, 1 H; H-4), 7.76 (m, 1 H; H-7), 7.68 (m, 6 H; o,o’, H-3, H-5), 7.53 (m, 1 H; H-4’), 7.39 (m, 2 H; H-6’, H-7’), 7.02 (m, 4 H; m; m’), 4.02 (m, 4 H; OCH₂), 1.79 (m, 4 H; OCH₂CH₂), 1.5-1.15 (m, 20 H; alkyl), 1.31 (s, 9 H; tert-butyl), 1.27 (s, 9 H; tert-butyl), 0.85 (m, 6 H; CH₃); UV/Vis/NIR (chloroform): λmax(ε) = 718 (162000), 654 nm
(45000 M$^{-1}$cm$^{-1}$); MALDI-MS: m/z calcd: 1043.6 [M+H]$^+$; 1065.6 [M+Na]$^+$ found 1043.0; 1064.9; elemental analysis calcd (%) for C$_{62}$H$_{68}$B$_2$F$_4$N$_6$O$_3$ [M = 1042.86 g/mol]: C 71.41 H 6.57 N 8.06; found: C 71.43 H 6.74 N 8.10.

**BF$_2$-PPCy 7c**

Column chromatography (CH$_2$Cl$_2$) yielded 39 mg (41 µmol, 68%) 7c as a green crystalline powder. $^1$H-NMR (400 MHz, CDC$_3$): $\delta$/ppm = 8.40 (d, $^3$J = 9.3 Hz, 1 H; H-8), 8.02 (d, $^3$J = 9.2 Hz, 1 H; H-4), 7.67 (m, 9 H; o, o', H-3, H-5, H-7, H-4', H-3'), 7.03 (m, 5 H; m, m', H-5'), 4.05 (m, 4 H; OCH$_2$), 2.73 (s, 3 H; CH$_3$), 1.81 (m, 4 H; OCH$_2$CH$_2$), 1.55-1.2 (m, 20 H; alkyl), 1.34 (s, 9 H; tert-butyl), 0.89 (m, 6 H; CH$_3$); UV/Vis/NIR (chloroform): $\lambda_{\text{max}}$(€) = 730 (179000), 663 nm (46000 M$^{-1}$cm$^{-1}$); MALDI-MS: m/z calcd: 961.5 [M+H]$^+$; 983.5 [M+Na]$^+$; 999.5 [M+K]$^+$ found 961.1; 983.1; 999.1; elemental analysis calcd (%) for C$_{57}$H$_{62}$B$_2$F$_4$N$_6$O$_2$ [M = 960.76 g/mol]: C 71.26 H 6.50 N 8.75; found: C 71.06 H 6.54 N 8.46.

**BF$_2$-PPCy 7d**

Column chromatography (CH$_2$Cl$_2$) yielded 37 mg (35 µmol, 60%) 7d as an olive-green crystalline powder. $^1$H-NMR (400 MHz, C$_2$D$_2$Cl$_4$): $\delta$/ppm = 8.37 (d, $^3$J = 9.3 Hz, 1 H; H-8), 8.14 (d, $^3$J = 9.1 Hz, 1 H; H-4), 7.81 (d, $^3$J = 8.8 Hz, 1 H; H-4'), 7.74 (dd, $^3$J = 9.3 Hz, $^4$J
= 2.0 Hz, 1 H; H-7), 7.65 (m, 7 H; o, o’, H-3, H-5, H-7’), 7.45 (dd, 3J = 8.8 Hz, 4J = 1.8 Hz, 1 H; H-5’), 7.01 (m, 4 H; m, m’), 4.02 (m, 4 H; OCH₂), 1.79 (m, 4 H; OCH₂CH₂), 1.5-1.15 (m, 20 H; alkyl), 1.30 (s, 9 H; tert-buty), 0.85 (m, 6 H; CH₃); UV/Vis/NIR (chloroform): λmax(ε) = 742 (191000), 672 nm (48000 M⁻¹cm⁻¹); MALDI-MS: m/z calcd: 1059.5 [M+H]+; 1081.5 [M+Na]+ found 1059.0; 1081.1; elemental analysis calcd (%) for C₆₂H₆₈B₂F₄N₆O₂S [M = 1058.92 g/mol]: C 70.32 H 6.47 N 7.94; found: C 70.27 H 6.45 N 7.89.

BF₂-PPCy 7e[6]

[Diagram]

BF₂-PPCy 7f

[Diagram]

Batch:

6f 36 mg 0.037 mmol
BF₃ · Et₂O 209 mg 1.47 mmol 0.18 ml
DIPEA 95 mg 0.74 mmol 0.12 ml
CH₂Cl₂ 20 ml

Column chromatography (CH₂Cl₂) yielded 32 mg (30 µmol, 80%) 7f as a green crystalline powder. ¹H-NMR (400 MHz, C₂D₂Cl₄): δ/ppm = 8.37 (d, 3J = 9.3 Hz, 1 H; H-8), 8.27 (d, 3J = 9.3 Hz, 1 H; H-8’), 8.13 (d, 3J = 9.0 Hz, 1 H; H-4), 7.91 (d, 3J = 9.2 Hz, 1 H; H-4’), 7.80 (d, 4J = 2.0 Hz, 1 H; H-5’), 7.73 (dd, 3J = 9.3 Hz, 4J = 1.9 Hz, 1 H; H-7), 7.64 (m, 8 H; o, o’, H-3, H-5, H-3’, H-7’), 7.01 (m, 4 H; m, m’), 4.02 (m, 4 H; OCH₂), 1.79 (m, 4 H; OCH₂CH₂), 1.5-1.15 (m, 20 H; alkyl), 1.30 (s, 9 H; tert-buty), 0.85 (m, 6 H; CH₃); UV/Vis/NIR (chloroform): λmax(ε) = 758 (234000), 686 nm (50000 M⁻¹cm⁻¹); MALDI-MS: m/z calcd: 1075.4 [M+H]+; 1097.4 [M+Na]+ found 1076.0; 1098.0; elemental analysis
calcd (%) for C₆₀H₆₁B₂BrF₄N₆O₂ [M = 1075.68 g/mol]: C 66.99 H 5.72 N 7.81; found: C 66.61 H 5.76 N 7.76.

**BF₂-PPCy 7g**

Column chromatography (CH₂Cl₂) yielded 182 mg (0.18 mmol, 83%) 7g as an olive-green crystalline powder. ¹H-NMR (400 MHz, C₂D₂Cl₄): δ/ppm = 8.98 (s, 1 H; H-3’), 8.37 (d, 3J = 9.1 Hz, 1 H; H-8), 8.18 (m, 2 H; H-4, H-8’), 7.89 (d, 3J = 7.8 Hz, 1 H; H-5’), 7.75 (m, 1 H; H-7), 7.64 (m, 7 H; o,o’, H-3, H-5, H-7’), 7.51 (m, 1 H; H-6’), 6.99 (m, 4 H; m, m’), 4.00 (m, 4 H; OCH₂), 1.76 (m, 4 H; OCH₂CH₂), 1.45-1.1 (m, 20 H; alkyl), 1.28 (s, 9 H; tert-butyl), 0.81 (m, 6 H; CH₃); UV/Vis/NIR (chloroform): λmax (ε) = 770 (213000), 700 nm (52000 M⁻¹cm⁻¹);

**MALDI-MS:** m/z calcd: 998.5 [M+H]⁺; 1020.5 [M+Na]⁺; 1036.5 [M+K]⁺ found 998.2; 1020.2; 1036.2; elemental analysis calcd (%) for C₅₉H₆₁B₂BrF₄N₇O₂ [M = 997.78 g/mol]: C 71.02 H 6.16 N 9.83; found: C 71.20 H 6.27 N 9.81.

**H-PPCy 6h**

Under nitrogen atmosphere 500 mg (0.666 mmol) 4e were refluxed in 0.6 ml POCl₃ for 1 h. The POCl₃ was distilled off and the remaining solid was dried in vacuo. A solution of 235 mg (0.732 mmol) 3h in 10 ml abs. THF was added and the mixture was heated to reflux for 30 min. The mixture was quenched in 150 ml CH₂Cl₂ and 150 ml water. The organic layer
was washed with 150 ml water (3x), dried over MgSO₄ and the solvent was removed. Column chromatography (CH₂Cl₂) yielded 471 mg (0.447 mmol, 67%) 6h as a green crystalline powder.

**¹H-NMR** (400 MHz, C₂D₂Cl₄): δ/ppm = 14.75 (s, 1 H; N-H), 14.51 (s, 1 H; N-H), 7.88 (d, ³J = 8.8 Hz, 1 H; H-4), 7.76 (d, ³J = 8.8 Hz, 1 H; H-4'), 7.73 (d, ⁴J = 2.0 Hz, 1 H; H-5), 7.68 (m, 6 H; o, o', H-8, H-8'), 7.58 (m, 3 H; H-7, H-7', H-5'), 7.49 (d, 1 H; H-3), 7.47 (d, 1 H; H-3'), 7.09 (m, 4 H; m, m'), 4.04 (m, 4 H; OCH₂), 2.65 (t, ³J = 7.2 Hz, 2 H; CH₂), 2.51 (t, ³J = 7.2 Hz, 2 H; CH₂), 1.79 (m, 4 H; OCH₂CH₂), 1.44 (m, 4 H; O(CH₂)₂CH₂), 1.41 (s, 9 H; tert-butyl), 1.38-1.15 (m, 16 H; alkyl), 1.36 (s, 9 H; tert-butyl), 0.84 (m, 6 H; CH₃); **UV/Vis/NIR** (chloroform): λmax(ε) = 739 (134000), 670 nm (49000 M⁻¹cm⁻¹);

**MALDI-MS**: m/z calcd: 1053.6 [M+H]⁺, found 1053.5; **elemental analysis** calcd (%) for C₆₉H₇₆N₆O₄ [M = 1053.38 g/mol]: C 78.67 H 7.27 N 7.98; found: C 78.63 H 7.36 N 8.01.

**H-PPCy 6i**

440 mg (0.418 mmol) 6h were refluxed in 10 ml CH₂Cl₂ and 1 ml trifluoroacetic acid over night. The mixture was diluted in 300 ml CH₂Cl₂ and the organic layer was washed with 300 ml water (3x). The solvent was removed and the residue was suspended in methanol, filtered off and washed with methanol. Drying on air yielded 390 mg (0.391 mmol, 94%) 9 as a green crystalline powder. **UV/Vis/NIR** (chloroform): λmax(ε) = 739 (132000), 670 nm (48000 M⁻¹cm⁻¹);

**MALDI-MS**: m/z calcd: 997.6 [M+H]⁺, found 997.4; **elemental analysis** calcd (%) for C₆₅H₆₈N₆O₄ [M = 997.27 g/mol]: C 78.28 H 6.87 N 8.43; found: C 78.31 H 6.98 N 8.43.
**BF$_2$-PPCy 7i**

100 mg (0.100 mmol) 6i and 0.33 ml (2 mmol) $N,N$-di-*iso*-propylethylamine were refluxed in 15 ml CH$_2$Cl$_2$. 0.5 ml (4 mmol) boron trifluoride etherate were added and the mixture was refluxed for 10 min. The mixture was washed with water, dried over MgSO$_4$ and the solvent was removed. Column chromatography (gradient: CH$_2$Cl$_2$ to CH$_2$Cl$_2$/ethylacetate 10/1) yielded 88 mg (0.081 mmol, 81%) 7i as a green crystalline powder. $^1$H-NMR (400 MHz, C$_2$D$_2$Cl$_4$): $\delta$/ppm = 8.36 (m, 1 H; H-8), 8.30 (m, 1 H; H-8'), 8.12 (d, $J=9.2$ Hz, 1 H; H-4), 7.94 (d, $J=9.3$ Hz, 1 H; H-4'), 7.72 (dd, $J=9.5$ Hz, $J=9.5$ Hz, 1 H; H-7), 7.66 (m, 5 H; o, o', H-3), 7.58 (m, 4 H; H-3', H-5, H-5', H-7'), 7.00 (m, 4 H; m, m); 4.01 (m, 4 H; OCH$_2$), 2.68 (m, 4 H; (CH$_2$)$_2$), 1.78 (m, 4 H; OCH$_2$CH$_2$), 1.44 (m, 4 H; O(CH$_2$)$_2$CH$_2$), 1.29 (s, 9 H; tert-butyl), 1.37-1.12 (m, 16 H; alkyl), 0.83 (m, 6 H; CH$_3$); UV/Vis/NIR (chloroform): $\lambda_{max}(\epsilon) = 762$ (243000), 690 nm (54000 M$^{-1}$cm$^{-1}$); MALDI-MS: m/z calcd: 1093.6 [M+H]$^+$, found 1092.8; elemental analysis calcd (%) for C$_{65}$H$_{66}$B$_2$F$_4$N$_6$O$_4$ [M = 1092.87 g/mol]: C 71.44 H 6.09 N 7.69; found: C 71.39 H 6.17 N 7.64.

**BF$_2$-PPCy 8**

Under nitrogen atmosphere 50 mg (0.046 mmol) 7i were dissolved in 10 ml dry CH$_2$Cl$_2$ and 28.7 mg (0.114 mmol) BBr$_3$ were added. The mixture was stirred at room temperature over night. The mixture was quenched with water and the CH$_2$Cl$_2$ was removed on vacuum.
green solid was filtered off, washed several times with water and dried in vacuum. 31 mg (0.036 mmol) of a green solid were obtained. Due to its low solubility no further purification could be carried out and the crude product was used for the coupling reaction. MALDI-MS: m/z calcd: 869.3 [M+H]+, found 870.6.

**Peptide synthesis**

H-R<sub>9</sub>-NH<sub>2</sub> was synthesized on an Applied Biosystems ABI 433A peptide synthesizer using Fmoc-Arg(Pbf)-OH, PS-AM-RAM resin and standard Fmoc-SPPS coupling protocols<sup>[19]</sup>. Cleavage from the resin was achieved using a cleavage solution of 95% TFA, 2.5% TIS and 2.5% H<sub>2</sub>O. The peptide was purified by C18 RP-HPLC (Macherey-Nagel VP 250/10 Nucleosil 100-5 C18) using an Aekta purifier system with a gradient 0 - 10% B in 10 CV (A: 95% H<sub>2</sub>O, 5% MeCN, 0.1% TFA; B: 95% MeCN, 5% H<sub>2</sub>O, 0.1% TFA). The collected fractions were analyzed by MALDI-TOF mass spectrometry, pooled and freeze-dried MALDI-MS: m/z calcd: 1423.9 [M+H]+, found 1423.2.

**Coupling of 8 to H-R<sub>9</sub>-NH<sub>2</sub><sup>[10b]</sup>**

8 (1 eq.), COMU (1-[(1-Cyano-2-ethoxy-2-oxoethylideneaminooxy-dimethylamino-morpholinomethylene)]-methanaminium hexafluorophosphate) (0.95 eq.) and DIPEA (1.9 eq.) were dissolved in NMP and stirred at RT for 15 minutes. H-R<sub>9</sub>-NH<sub>2</sub> was dissolved in NMP and added to the solution. The mixture was stirred at RT over night. The green solution was concentrated to approximately 1 ml and desalted with a Sephadex PD10 column. The obtained fractions were analyzed by MALDI-TOF MS, concentrated and purified by C18 RP-HPLC as described above. The gradient used was 30 - 50% B in 15 CV. MALDI-MS: m/z calcd: 2274.2 [M+H]+, found 2270.4.

**Live cell imaging<sup>[12]</sup>**

Two days before the experiment 100,000 HeLa cells were seeded in a 35 mm μ-dish (Ibidi) and the cells were incubated at 37°C and 5% CO<sub>2</sub>. On the day of the experiment a 2 μM solution of the labeled polyarginine in HKR buffer (5 mM HEPES, 137 mM NaCl, 2.68 mM KCl, 2.05 mM MgCl<sub>2</sub>·6 H<sub>2</sub>O, 1.8 mM CaCl<sub>2</sub>·2 H<sub>2</sub>O and 1 g/l glucose) was prepared. The cell culture medium was aspirated from the cells and 2 ml of the prepared peptide solution were added. After 30 minutes of incubation the peptide-containing solution was removed and the cells were washed twice with 1 ml HKR buffer. Before live cell imaging a colorless medium including Hoechst 33342 for labeling the nucleus was added and the cells were observed for several hours with a confocal microscope (Zeiss LSM 510 Meta) at 37°C and 5% CO<sub>2</sub>. Hoechst 33342 was excited at 405 nm and fluorescence was detected by using a band pass filter 420-480 nm. Arg<sub>9</sub>-8 was excited at 633 nm and fluorescence was detected by using a long pass
filter 650 nm. At this setup a longer wavelength excitation was not available; due to the low absorption of 8 at 633 nm the fluorescence intensity is relatively low.

To obtain the emission spectrum of Arg₉-8 in the used HKR buffer as well as inside of living cells, 100.000 HeLa cells were seeded in a 35 mm µ-dish (Ibidi). Two days later the cells were incubated with Arg₉-8. 1 h after incubation live cell imaging was performed (Figure 1) using a confocal microscope (Zeiss Axiovert 200) equipped with a pulsed laser (LDH-P-C-690, Picoquant) and a TCSPC module (HydraHarp 400, Picoquant). Spectra were simultaneously recorded by using a PS 857 dispersion prism (Thorlabs) and a CCD-camera (Andor Newton DU970N-BV).

Figure 1: Normalized fluorescence spectra of Arg₉-8 in HKR buffer (black line) and inside the endocytotic vesicle (red line) marked in the inset.
Optical Spectra

Absorption (and fluorescence for 5d and 5e) spectra of the H-Chelates 4 and 6 (black) and the BF2-complexes 5 and 7 (red) in chloroform at room temperature.
$6c \ X = H$

$7c \ X = BF_2$

$6d \ X = H$

$7d \ X = BF_2$
**Supplementary Material (ESI) for Chemical Communications**

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![Graph](image-url)

\[6f \ X = H\]

\[7f \ X = BF_2\]

![Graph](image-url)

\[6g \ X = H\]

\[7g \ X = BF_2\]
Supplementary Material (ESI) for Chemical Communications
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\[ \text{6i} \quad X = H \]
\[ \text{7i} \quad X = BF_2 \]