

## Electronic Supplementary Information for

# Electrochemical Detection of Insertion/Deletion Mutations Based on Enhanced Flexibility of Bulge-Containing Duplexes on Electrodes

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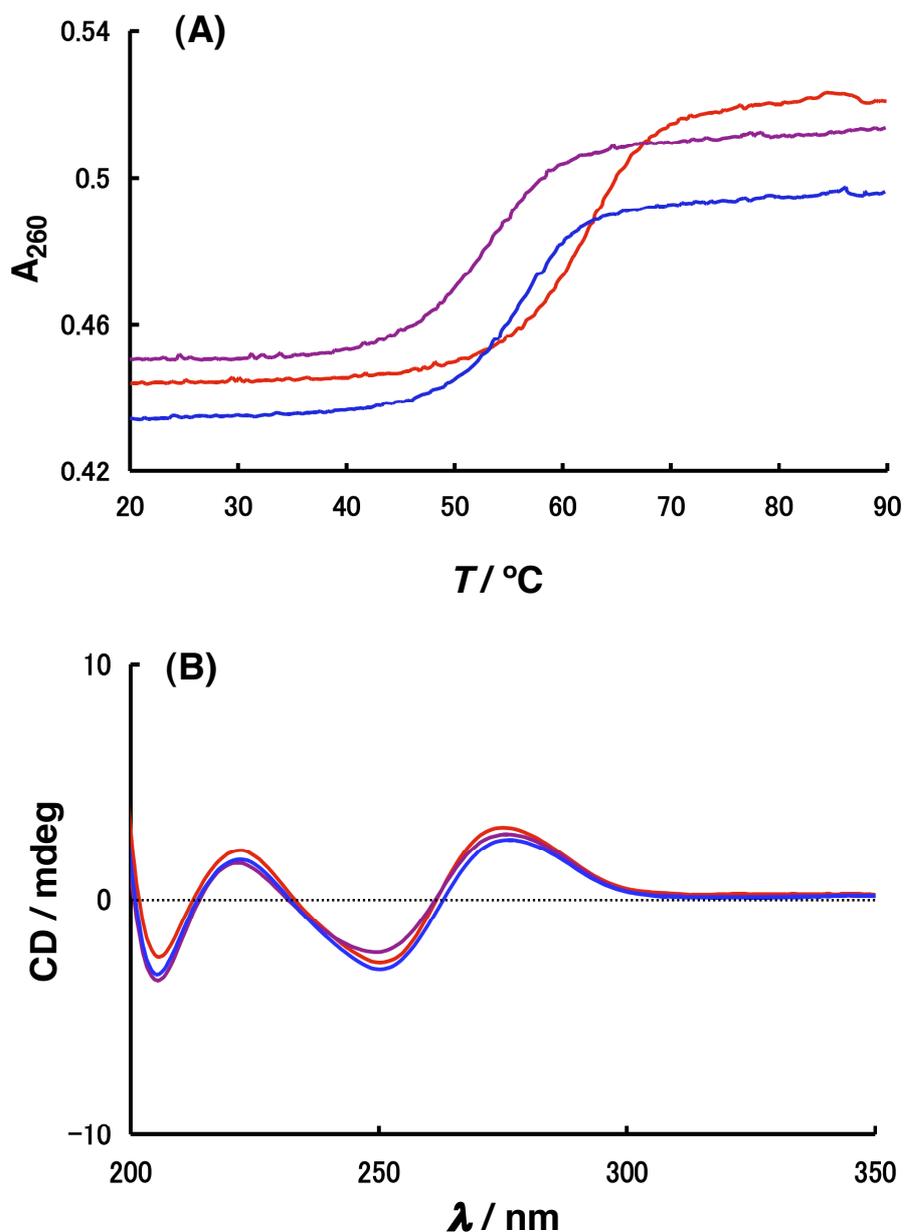
## Experimental section

**Materials.** DNA probes were synthesized as a procedure previously reported.<sup>1-3</sup>

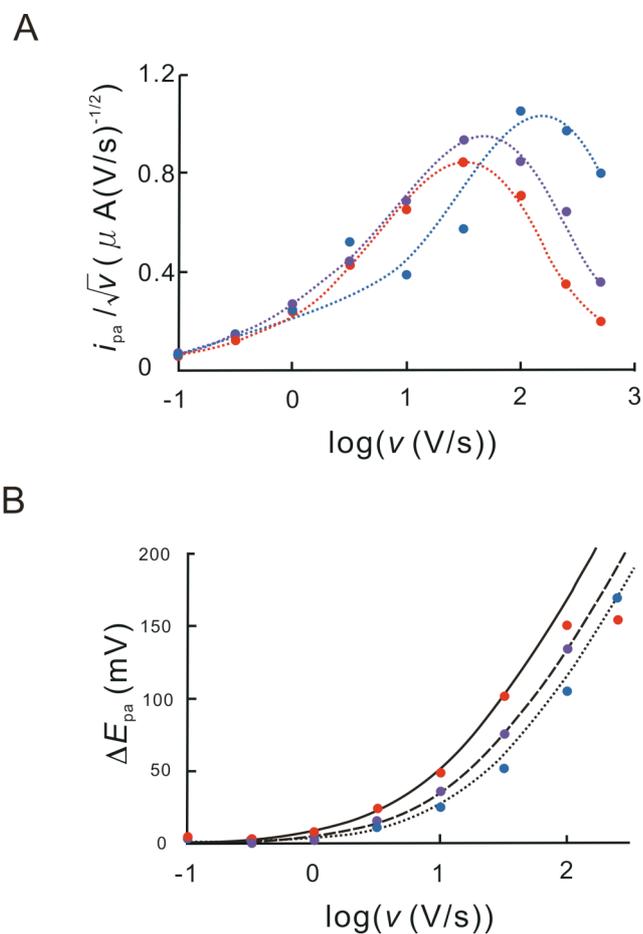
**DNA probe characterization.** DNA probes were characterized by using MALDI-TOF mass spectrometry measurements recorded on a Bruker-Daltonics-Autoflex mass spectrometer with 3-hydroxypicolinic acid as the matrix (Figure S4): m/z for **CYP3A4**: (calcd: 5170.83 [M-H]<sup>-</sup>; found: 5171.62), **CYP2D6** (calcd: 5043.80 [M-H]<sup>-</sup>; found: 5046.76), **C6-CYP3A4** (calcd: 5212.89 [M-H]<sup>-</sup>; found: 5216.51), **C6-CYP2D6** (calcd: 5085.85 [M-H]<sup>-</sup>; found: 5085.65), **p53** (calcd: 5091.82 [M-H]<sup>-</sup>; found: 5092.14), **ABCA** (calcd: 5096.81 [M-H]<sup>-</sup>; found: 5099.59), **PCM1** (calcd: 5078.79 [M-H]<sup>-</sup>; found: 5082.36), **PDK4** (calcd: 5202.85 [M-H]<sup>-</sup>; found: 5203.15).

**Immobilization of DNA probes onto gold electrodes.** Commercially available gold electrodes (Tanaka Kikinzoku, Tokyo, Japan) were cleaned as previously reported<sup>1-5</sup> and dried under argon stream before use. For immobilization of DNA probes, a probe DNA (100 μM) in a buffer solution (1 μL, 10 mM phosphate buffer that contained 1 M NaClO<sub>4</sub>, pH 7) was placed on the gold electrode and kept in a closed container under high humidity for 90 min at room temperature. After having been rinsed with the buffer solution (300 μL), the probe DNA-modified gold electrode was soaked in a solution of 1 mM 6-mercaptohexan-1-ol in the buffer solution contained 1% Tween 20 (300 μL) for 90 min at room temperature. Then, it was thoroughly washed with Milli-Q water and the buffer solution successively. For hybridization of target DNAs, the modified gold electrode was soaked into a solution of 100 μM (down to 1 μM) target DNAs in the buffer solution (50 μL) for 90 min at room temperature, then it was rinsed with the buffer solution (300 μL).

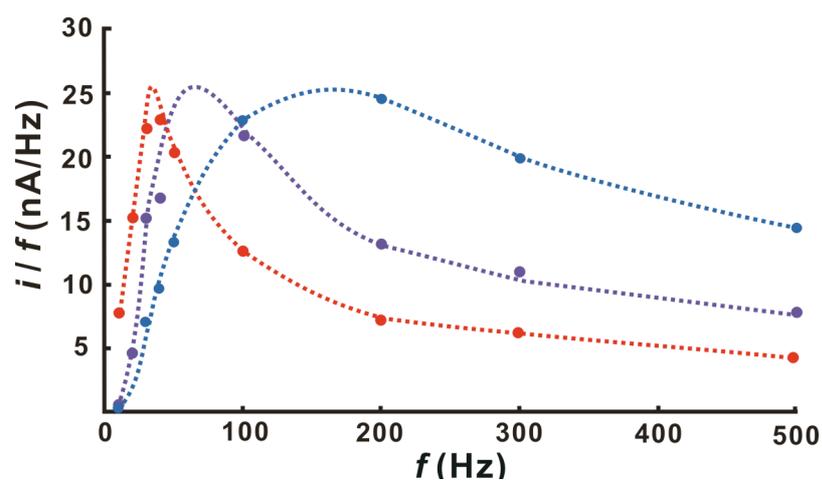
**Electrochemical Measurements.** CV and SWV measurements were carried out as previously reported.<sup>4,5</sup> The buffer solution (10 mM phosphate buffer that contained 1 M NaClO<sub>4</sub>, pH 7) was used as the electrolyte solution for all electrochemical studies. CV and SWV measurements were carried out at 25 °C on the probe-modified electrodes by means of a normal three-electrode configuration consisting of the gold working electrode, a saturated Ag/AgCl reference electrode, and a platinum wire auxiliary electrode. The working compartment of the electrochemical cell was separated from the reference compartment by a glass frit.



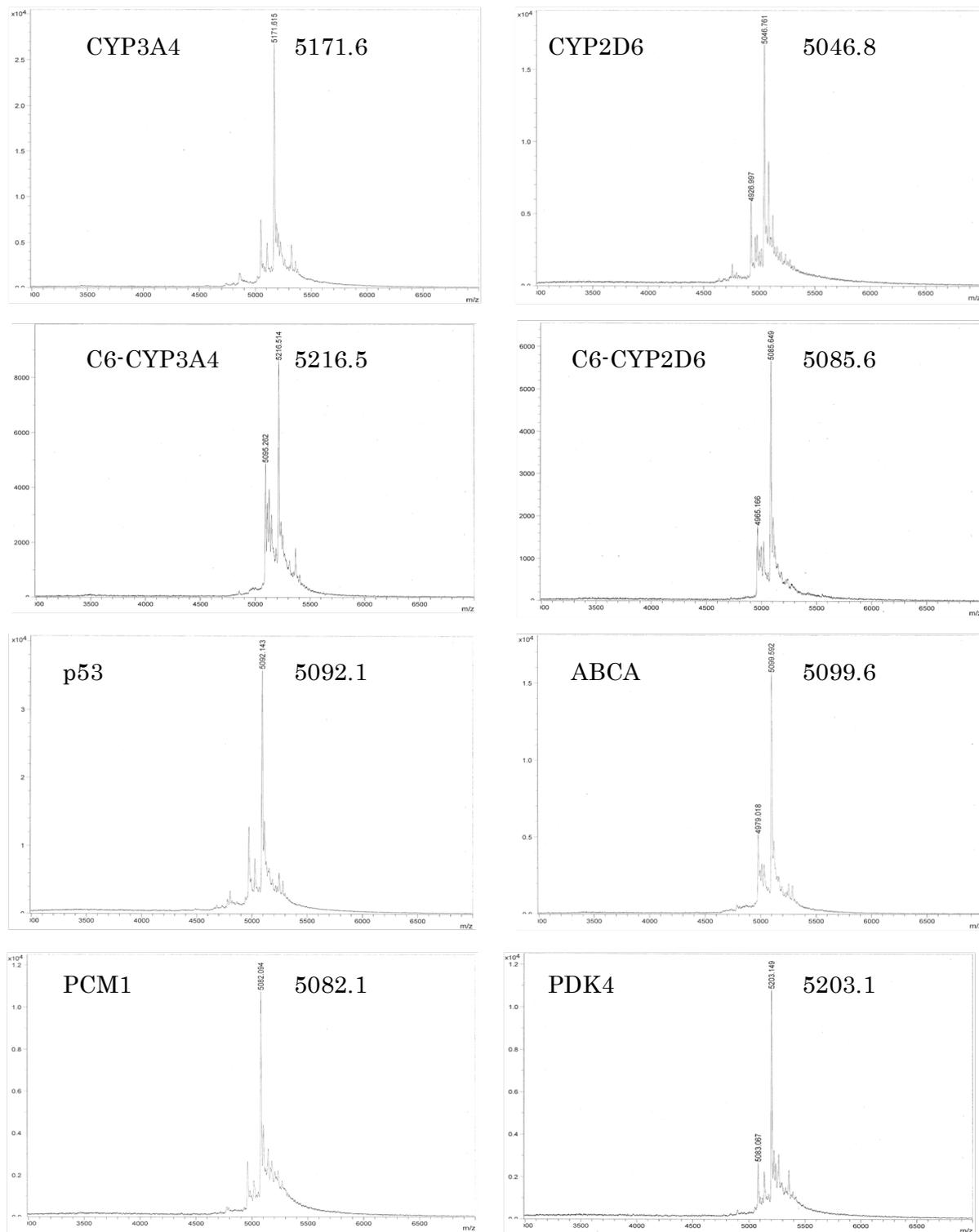
**Figure S1.** (A) UV-melting curves at 260 nm and (B) CD spectra at 25 °C of fully-matched **Iq1'-wild** (red line), insertion-type **Iq1'-insC** (purple line), and deletion-type **Iq1'-delT** (blue line). The solution for the measurements by 1 mm path-length contained **Iq1'** (1  $\mu\text{M}$ ) and 1 equiv of **wild**, **insC**, or **delT** (1  $\mu\text{M}$ ) in 10 mM phosphate buffer (pH 7) with 1 M  $\text{NaClO}_4$ . Absorbance vs. temperature profiles of the duplexes were measured by using a JASCO-V-560 spectrophotometer with a peltier and a temperature controller in a range from 20 to 90 °C. CD spectra were recorded on a JASCO-J-720WI spectropolarimeter.



**Figure S2.** Scan rate  $\nu$  dependence of the anodic peak current  $i_{pa}$  and of the anodic peak potential  $E_{pa}$  of the cyclic voltammograms recorded at the **Iq1**-modified gold electrodes with **wild** (red symbols), **insC** (purple symbols), and **delA** (blue symbols). (A) Bell shaped variation of the anodic peak-current function  $i_{pa}/\sqrt{\nu}$  vs  $\log(\nu)$ . (B) Variation of  $\Delta E_{pa}$  vs  $\log(\nu)$ . Simulated curves are shown in the case for  $k = 90 \text{ s}^{-1}$  (continuous line),  $160 \text{ s}^{-1}$  (dashed line), and  $180 \text{ s}^{-1}$  (dotted line).



**Figure S3.** Frequency  $f$  dependence of SWV peak current at the electrodes modified with the duplexes of wild-type **Iq1•wild** (red symbols), the insertion-type **Iq1•insC** (purple symbols), and the deletion-type **Iq1•delA** (blue symbols). SWV peak current is normalized versus  $f$ .



**Figure S4.** MALDI-TOF mass spectra of the newly synthesized DNA probes.

## Supporting References

1. M. Inouye, R. Ikeda, M. Takase, T. Tsuru and J. Chiba, *Proc. Natl. Acad. Soc. USA*, 2005, **102**, 11606.
2. R. Ikeda, J. Chiba and M. Inouye, *e-J. Surf. Sci. Nanotechnol.*, 2005, **3**, 393.
3. R. Ikeda, A. Akaishi, J. Chiba and M. Inouye, *ChemBioChem*, 2007, **8**, 2219.
4. R. Ikeda, S. Kobayashi, J. Chiba and M. Inouye, *Chem. Eur. J.*, 2009, **15**, 4822.
5. R. Ikeda, S. Kitagawa, J. Chiba and M. Inouye, *Chem. Eur. J.*, 2009, **15**, 7048.