Supporting Information

Solvatochromic Probes for Detecting Hydrogen-Bond-Donating Solvents

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MADAT – $^{13}$C NMR – 400MHz – DMSO

![Chemical structure and NMR spectrum](image-url)
S1. AMADAT emission in various solvents. Concentration = 0.15 mM in DCM, 0.41 mM in ethyl acetate and acetonitrile, 4.1 mM in methanol. A large concentration of AMADAT was used for methanol due to the large quenching of the fluorescence.

S2. Emission of MADAT in phenol and trifluoroethanol. Concentration = 4.1 mM.
S3. Linear correlation between concentration of thymine 1 and fluorescence intensity of MADAT (0.12 mM) in dichloromethane, showing efficient binding at sub-stoichiometric levels of thymine 1.

S4. Addition of N-Me thymine 2 to MADAT in dichloromethane. Concentration = 0.12mM. Excitation = 330

S5. Linear relation between water concentration and fluorescence intensity.

S6. Linear relation between concentration and intensity of MADAT in ethyl acetate indicating no dimerization.

S7. Linear relation between concentration and intensity of AMADAT in ethyl acetate indicating no dimerization.

S8. Quantum Yield of MADAT: Dichloromethane = 0.88, Methanol = 0.0068 using anthracene (QY 0.27) as reference.