Porous cobalt oxides with tunable hierarchical morphologies for

supercapacitor electrodes

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Supporting Information:

Figure S1. SEM image of cobalt hydroxide prepared using cobalt chloride under a low magnification, showing microspherical particle are formed.

Figure S2. SEM images of cobalt hydroxides prepared using (a) chloride and (b) acetate, where 0.02ml NaOH substituted ammonia solution, showing that small plate-like products are generated instead of hierarchical structure.

Figure S3. SEM images of cobalt hydroxides prepared using cobalt nitrate for 40 min reaction, showing nanocolumns consist of also (0001) hexagonal $Co(OH)_2$ nanosheets.

Figure S4. SEM image of the cobalt hydroxide prepared with reduced cobalt nitrate concentration (0.05 Mol/L) and ammonia solution at 70 °C for 1 hour.

Figure S5. XPS spectrum of cobalt oxide prepared by annealing cobalt hydroxide using acetate salt. The Co 2p XPS spectua of the cobalt oxide exhibit two peaks at about 795 and 780 eV, corresponding to the Co $2p_{1/2}$ and Co $2p_{3/2}$ spin-orbit peaks of Co₃O₄.

Figure S6. Discharge curves of for the Co_3O_4 samples fabricated with different cobalt salts, (a) nitrate, (b) chloride, and (c) acetate at different rates. The calculated specific capacitance values of the three samples prepared using nitrate, chloride and acetate are 218, 187 and 171 F g⁻¹ at 1.25 A g⁻¹, respectively, consisting with these values calculated from CV curves.

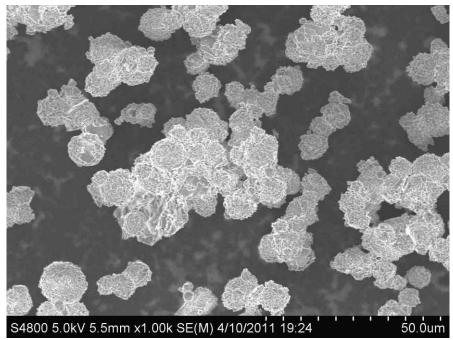


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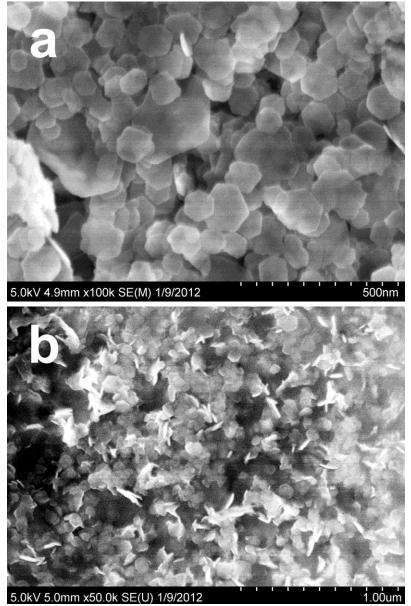


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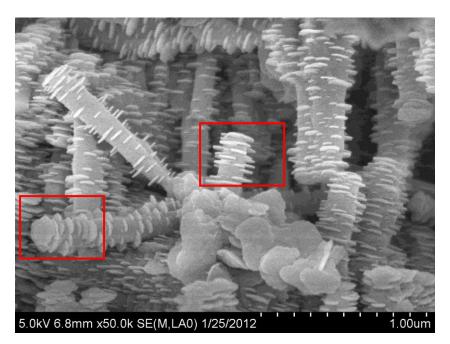


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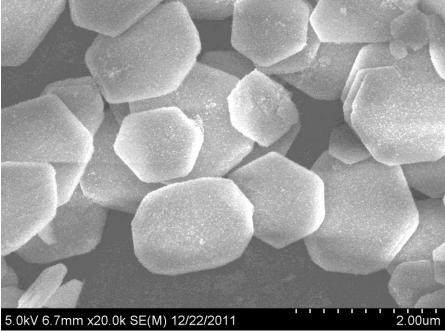


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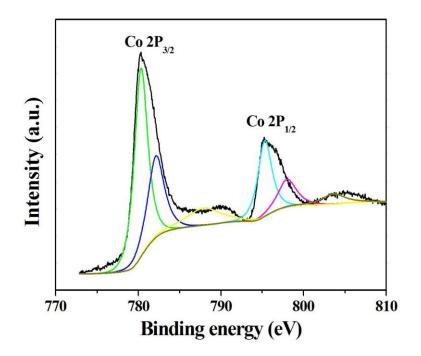


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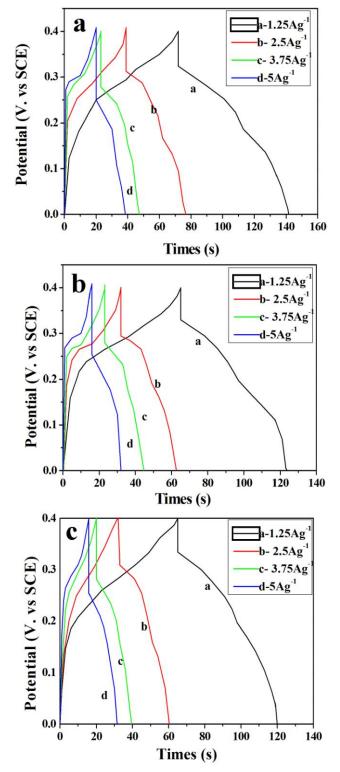


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