

Supplementary Information

A three-dimensional coordination polymer constructed from linear trinuclear copper(II) clusters with a ferromagnetic exchange coupling

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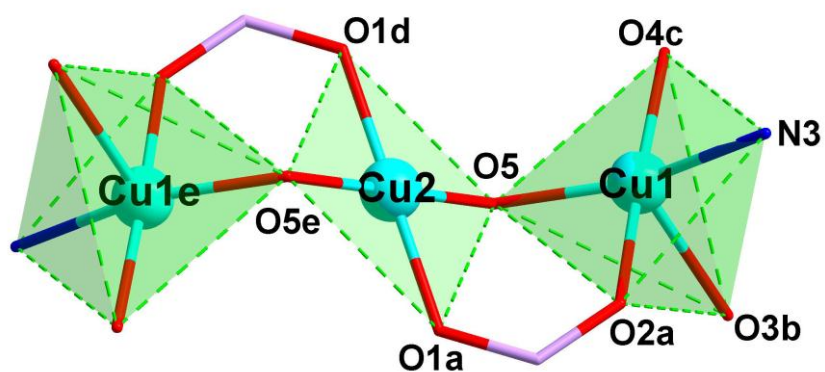


Fig. S1 The trinuclear copper(II) cluster in compound **1**.

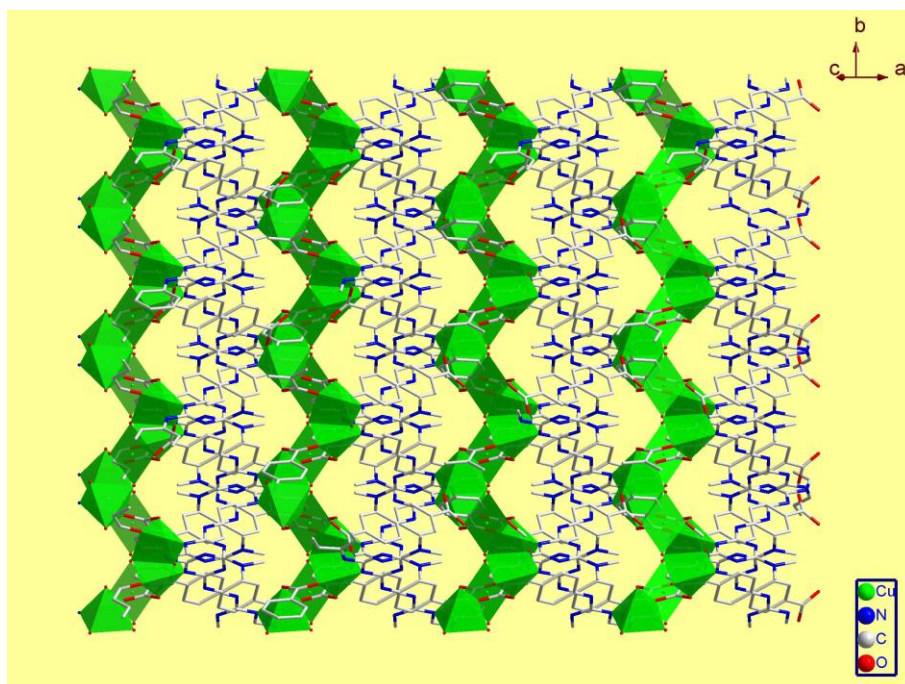


Fig. S2 View of the 3-D framework along *b* axis with the highlighted 1-D inorganic hybrid chain in compound **1**.

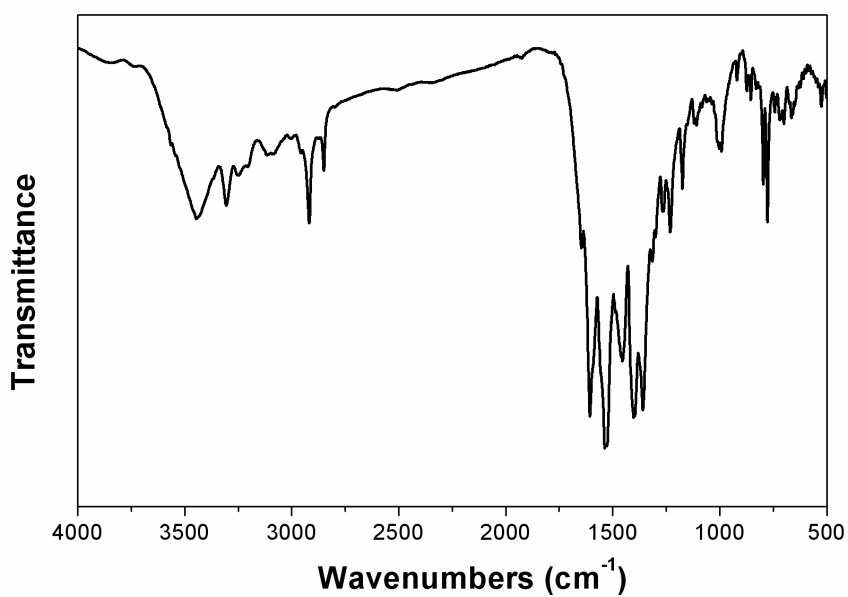


Fig. S3 IR spectrum of compound **1**.

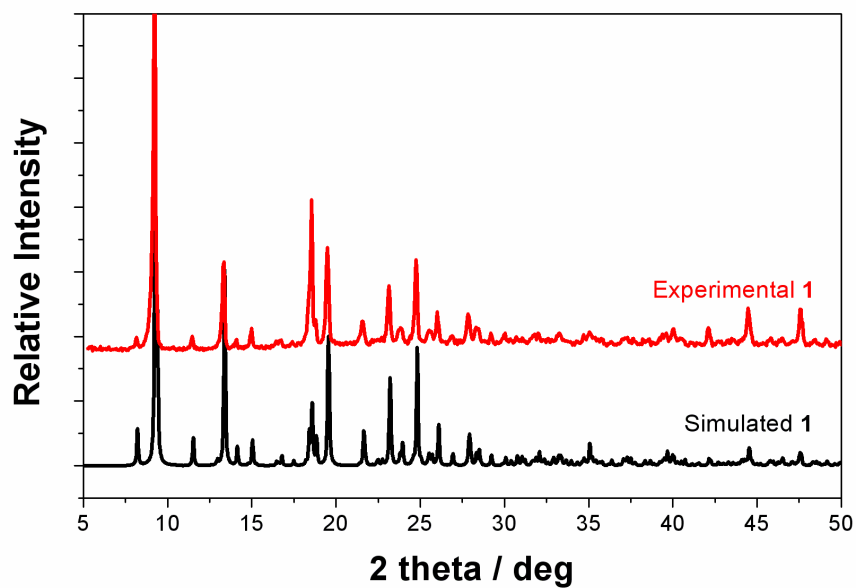


Fig. S4 Powder X-ray diffractions for simulated and experimental **1**.

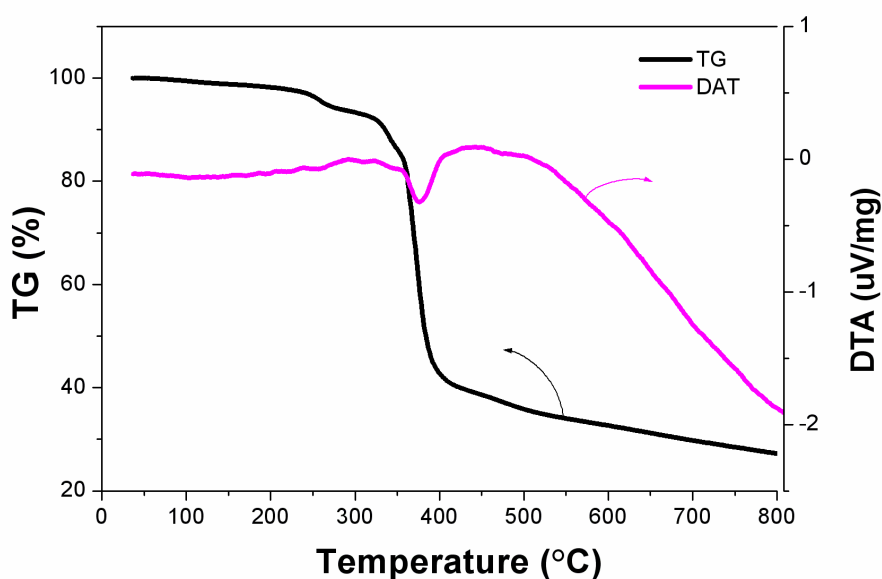


Fig. S5 TG and DTA curves of compound **1**.

Table S1. Selected Bond Lengths [Å] and Angles [°] for **1**^a

Cu1-O5	1.899(3)	Cu1-O3b	2.445(4)
Cu1-O2a	1.939(3)	Cu2-O5	1.895(3)
Cu1-O4c	1.951(3)	Cu2-O1a	1.948(3)
Cu1-N3	2.077(4)		
O5-Cu1-O2a	90.05(14)	N3-Cu1-O3b	102.70(12)
O5-Cu1-O4c	93.98(14)	O5e-Cu2-O5	180
O2a-Cu1-O4c	174.58(14)	O5e-Cu2-O1a	90.54(15)
O5-Cu1-N3	159.72(15)	O5-Cu2-O1a	89.46(15)
O2a-Cu1-N3	85.73(14)	O5e-Cu2-O1d	89.46(15)
O4c-Cu1-N3	89.24(13)	O5-Cu2-O1d	90.54(15)
O5-Cu1-O3b	96.76(13)	O1a-Cu2-O1d	180
O2a-Cu1-O3b	85.87(14)	Cu2-O5-Cu1	111.00(17)
O4c-Cu1-O3b	97.22(13)		

^aSymmetry transformations used to generate equivalent atoms: (a) $-x+3/2, y-1/2, -z+1/2$; (b) $x+1/2, -y+1/2, z+1/2$; (c) $-x+3/2, y+1/2, -z-1/2$; (d) $x+1/2, -y+1/2, z-1/2$; (e) $-x+2, -y, -z$.