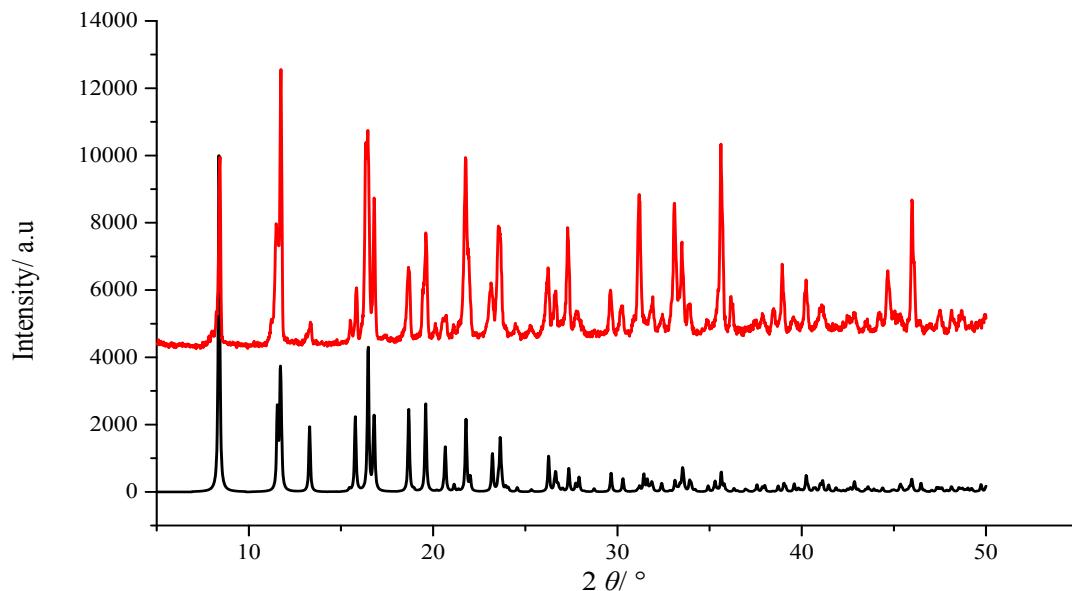


## Supplementary information

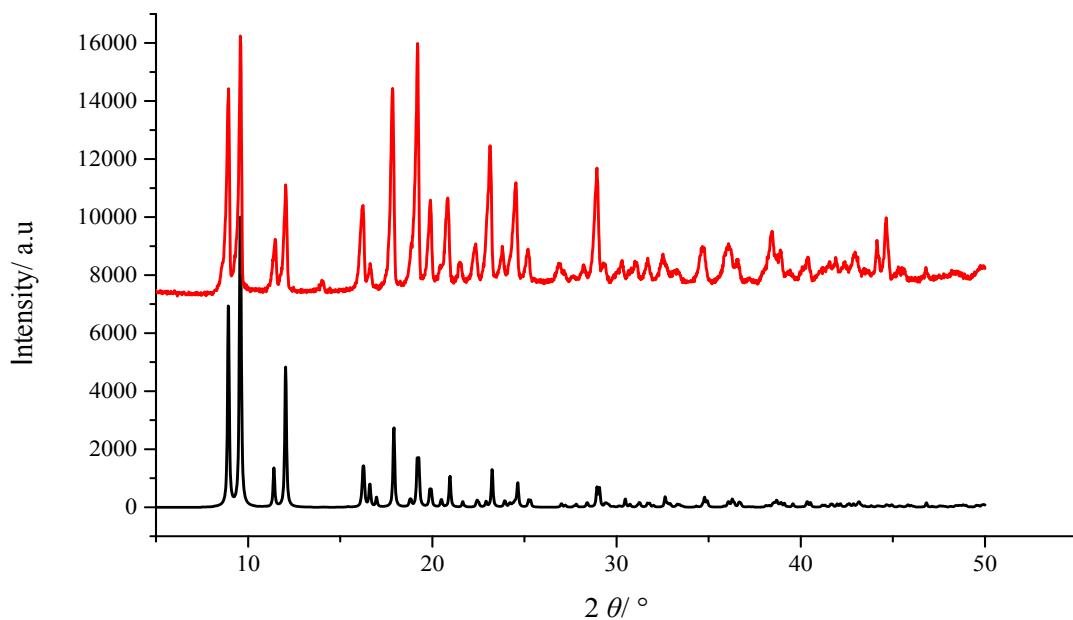
### Tuning of net architectures of Ni(II) and Zn(II) coordination polymers using isomeric organic linkers

*Shuang-De Liu, Bing-Chiuan Kuo, Yen-Wen Liu, Jing-Yi Lee, Kwun Yip Wong and Hon Man Lee\**

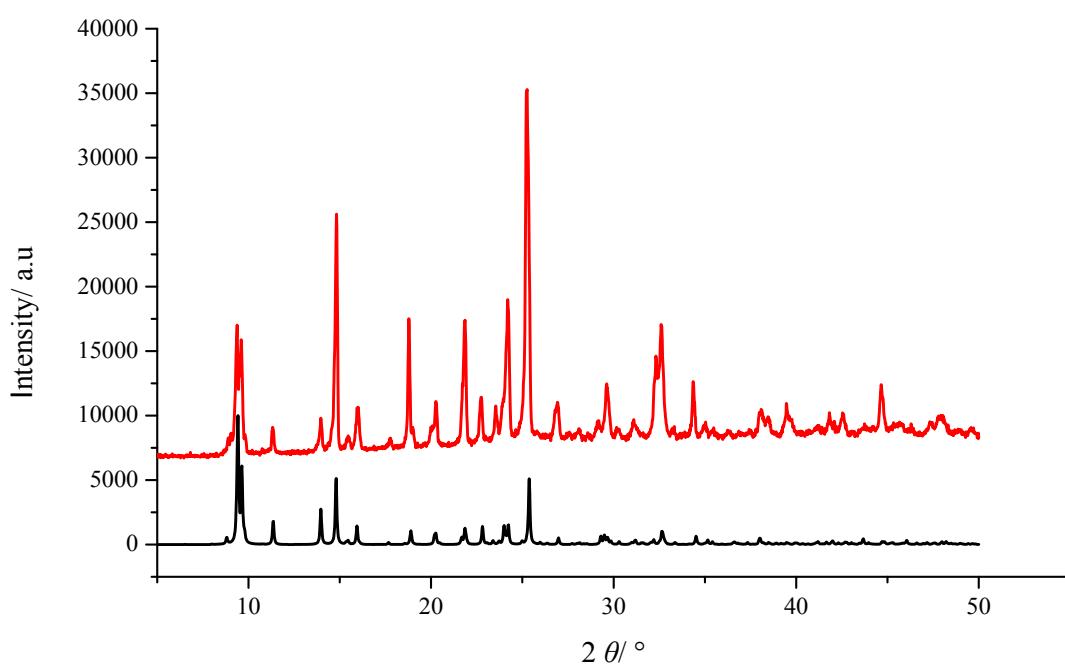
Department of Chemistry, National Changhua University of Education, Changhua 50058, Taiwan, R.O.C.



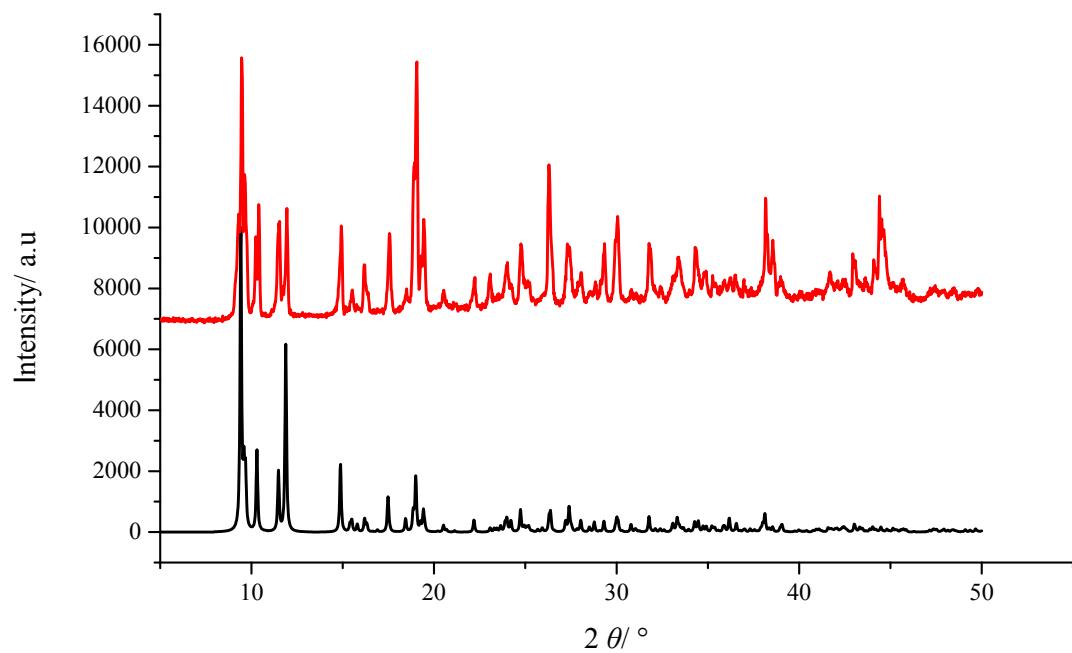
**Figure S1.** Comparison of the experimental PXRD patterns for as-synthesized **1** (upper trace) and the simulated pattern from single-crystal X-ray diffraction data (lower trace).



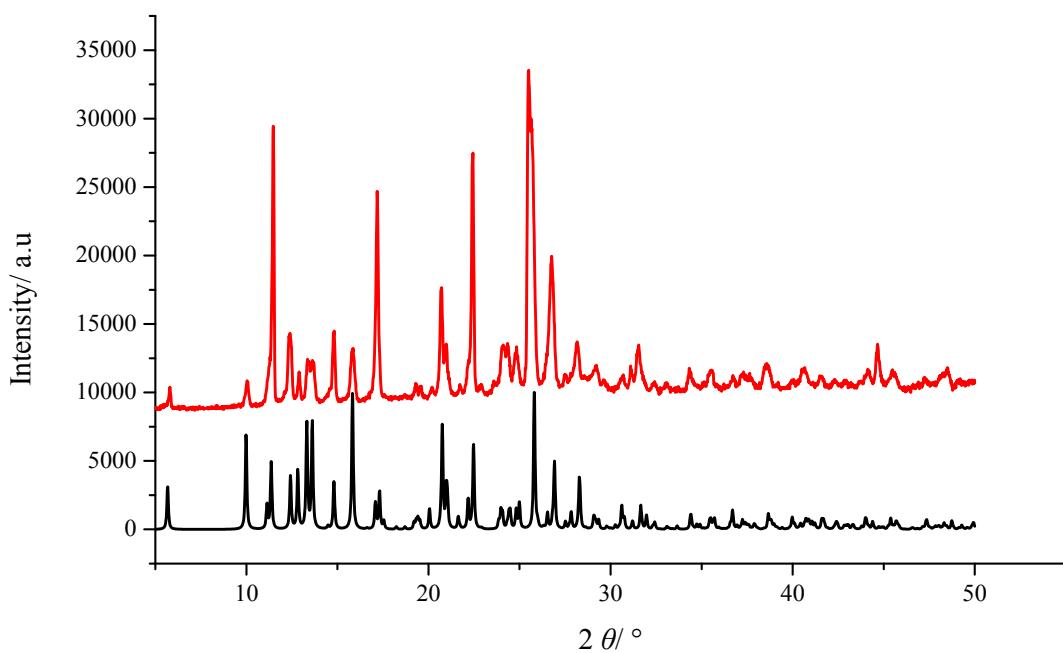
**Figure S2.** Comparison of the experimental PXRD patterns for as-synthesized **2** (upper trace) and the simulated pattern from single-crystal X-ray diffraction data (lower trace).



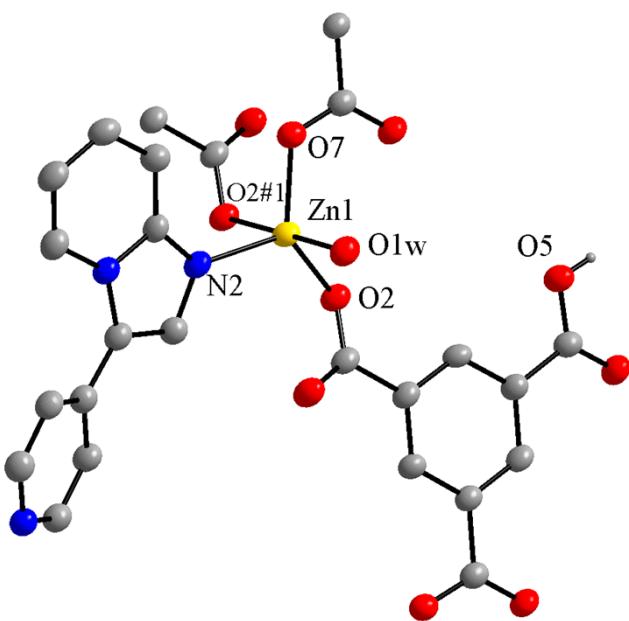
**Figure S3.** Comparison of the experimental PXRD patterns for as-synthesized **3** (upper trace) and the simulated pattern from single-crystal X-ray diffraction data (lower trace).



**Figure S4.** Comparison of the experimental PXRD patterns for as-synthesized **4** (upper trace) and the simulated pattern from single-crystal X-ray diffraction data (lower trace).

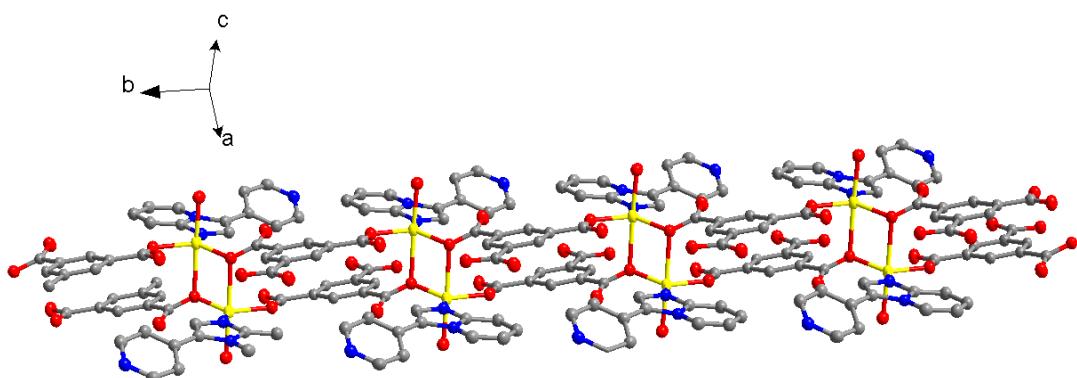


**Figure S5.** Comparison of the experimental PXRD patterns for as-synthesized **5** (upper trace) and the simulated pattern from single-crystal X-ray diffraction data (lower trace).

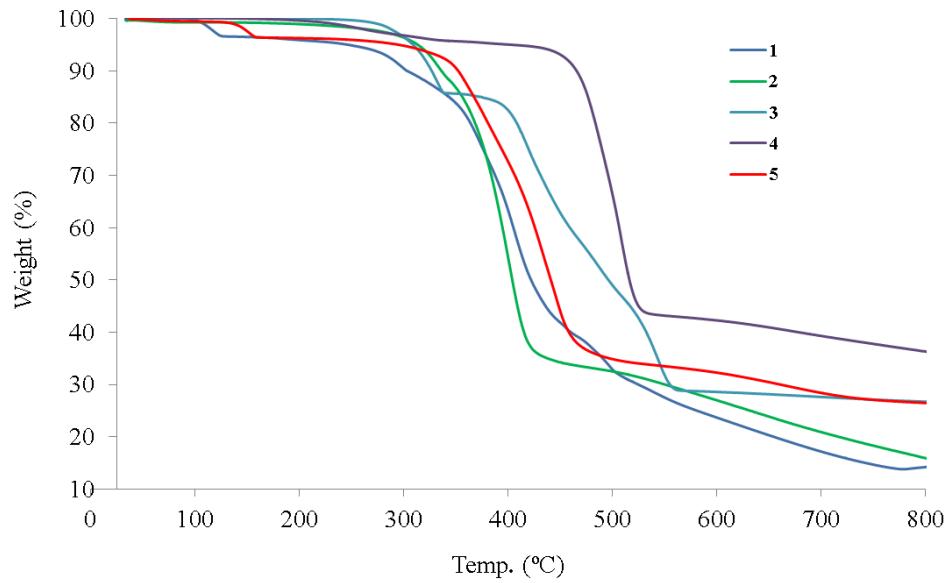


**Figure S6.** Coordination environment of the Zn atoms in **6** with thermal ellipsoids shown at 50 % probability.

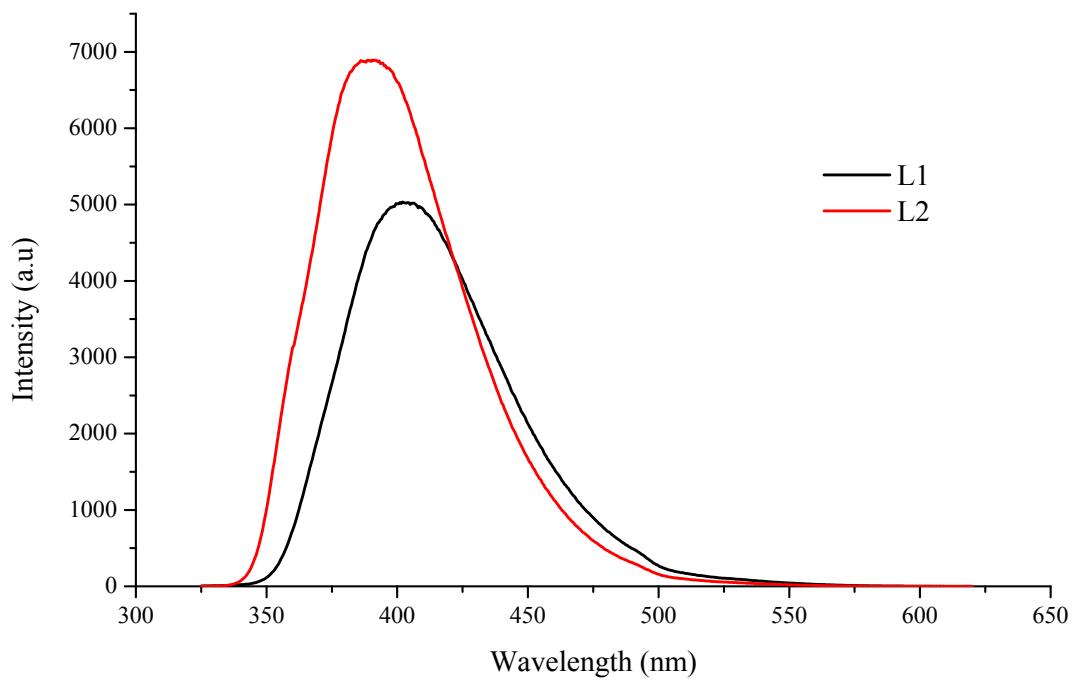
The hydrogen atoms, except that on O5, are omitted for clarity. Symmetry code: #1 =  $-x + 1, -y, -z + 2$



**Figure S7.** Schematic view of the 1D chain in **6**.

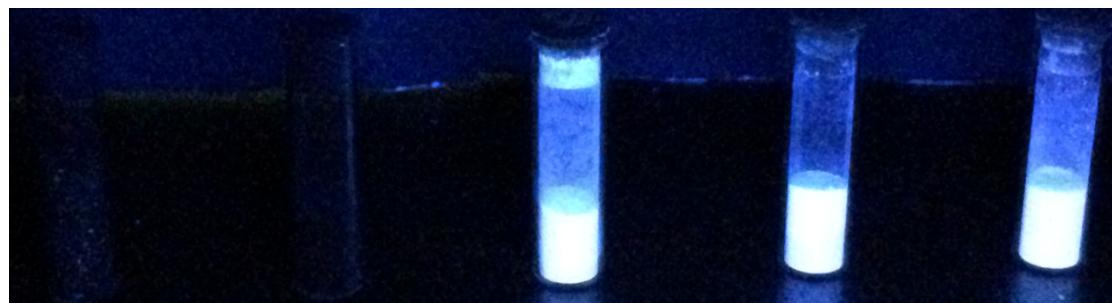


**Figure S8.** TG curves of **1–5**

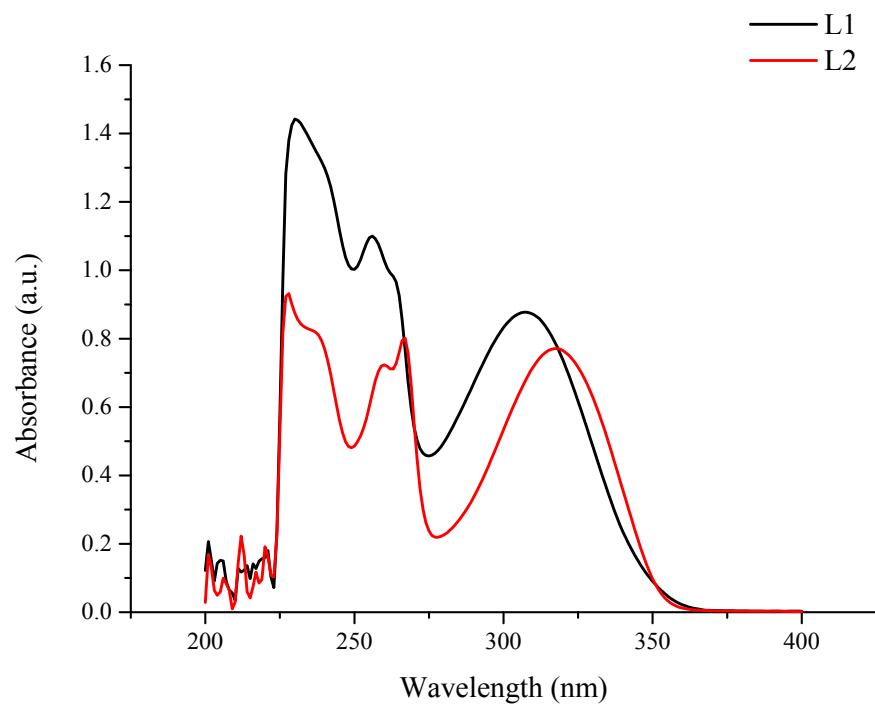


**Figure S9.** Fluorescence spectra of L1 and L2 ( $6 \times 10^{-5}$  M in DCM; L1:  $\lambda_{\text{ex}} = 308$  nm,  $\lambda_{\text{max}} = 401$  nm. L2:

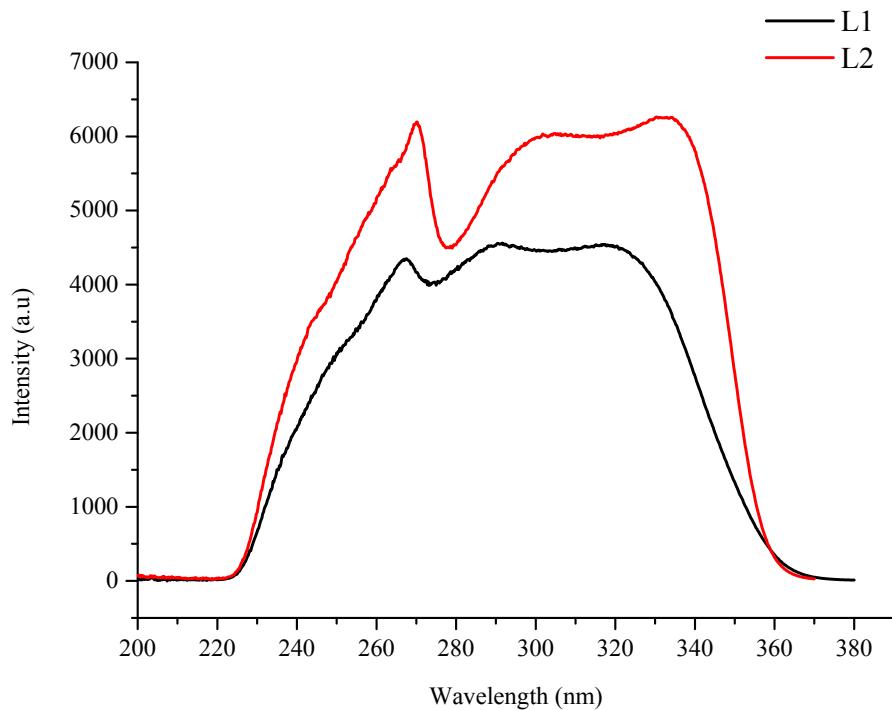
$\lambda_{\text{ex}} = 318$  nm,  $\lambda_{\text{max}} = 391$  nm)



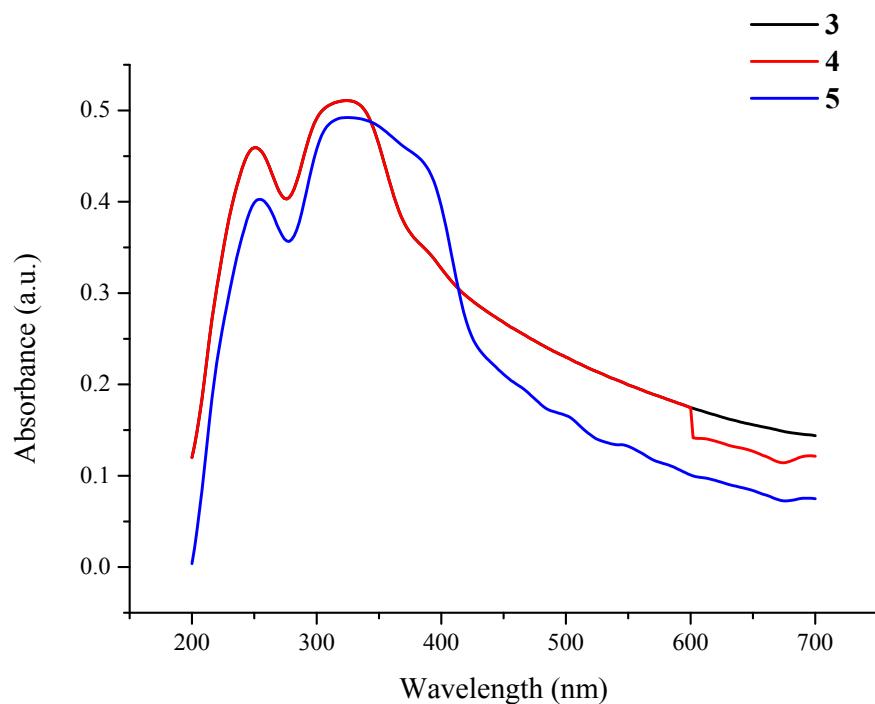
**Figure S10.** Irradiation of **1–5** (from left to right) under UV light (325 nm)



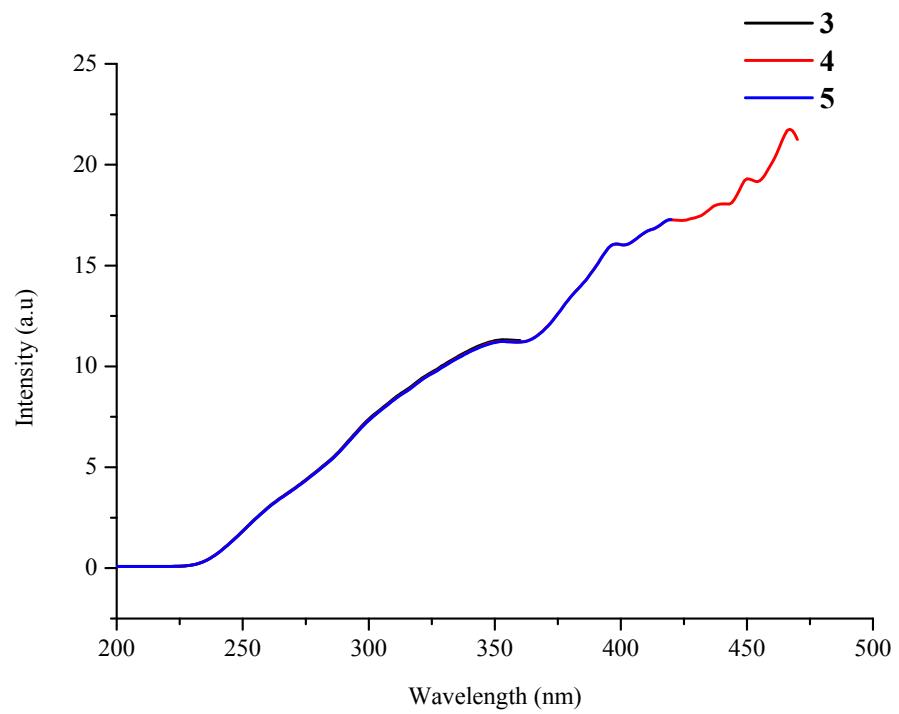
**Figure S11.** Absorption spectrum of L1 and L2 ( $6 \times 10^{-5}$  M in DCM)



**Figure S12.** Excitation spectrum of L1 and L2 ( $6 \times 10^{-5}$  M in DCM)



**Figure S13.** Solid-state Absorption spectrum of Zn 3–5



**Figure S14.** Solid-state excitation spectrum of Zn 3–5

**Table S1. Crystallographic data**

	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>	<b>5</b>	<b>6</b>
empirical formula	C <sub>20</sub> H <sub>17</sub> N <sub>3</sub> NiO <sub>6</sub>	C <sub>40</sub> H <sub>28</sub> N <sub>6</sub> Ni <sub>2</sub> O <sub>9</sub>	C <sub>24</sub> H <sub>16</sub> N <sub>3</sub> O <sub>7</sub> Zn <sub>2</sub>	C <sub>21</sub> H <sub>13</sub> N <sub>3</sub> O <sub>7</sub> Zn <sub>2</sub> ·0.5H <sub>2</sub> O	C <sub>22</sub> H <sub>15</sub> N <sub>3</sub> O <sub>9</sub> Zn <sub>2</sub>	C <sub>21</sub> H <sub>15</sub> N <sub>3</sub> O <sub>7</sub> Zn·H <sub>2</sub> O
formula weight	454.08	854.10	589.14	558.09	596.11	504.75
crystal system	orthorhombic	orthorhombic	triclinic	monoclinic	monoclinic	triclinic
space group	Fdd2	Fdd2	P $\bar{1}$	C2/c	P2 <sub>1</sub> /c	P $\bar{1}$
<i>a</i> , Å	17.297(2)	11.723(2)	10.4229(5)	20.233(4)	7.5106(2)	7.2889(4)
<i>b</i> , Å	30.446(4)	22.957(6)	10.5611(6)	10.901(2)	31.1959(7)	9.7098(5)
<i>c</i> , Å	14.7303(19)	31.011(11)	12.2119(6)	20.355(4)	10.5537(3)	13.6547(7)
$\alpha$ , deg	90	90	114.782(3)	90	90	95.558(1)
$\beta$ , deg	90	90	109.889(4)	115.467(10)	118.658	90.979(1)
$\gamma$ , deg	90	90	96.087(3)	90	90	93.688(1)
<i>V</i> , Å <sup>3</sup>	7757.3(17)	8346(4)	1097.91(10)	4053.3(13)	2169.82(10)	959.59(9)
<i>T</i> , K	150(2)	150(2)	150(2)	150(2)	150(2)	150(2)
<i>Z</i>	16	8	2	8	4	2
no. of unique data	5007	4545	4795	4202	4716	3957
no. of params refined	281	263	326	315	325	298
<i>R</i> <sub>1</sub> <sup>a</sup> [ <i>I</i> > 2σ <i>I</i> ]	0.0684	0.0432	0.0537	0.0842	0.0774	0.0788
<i>wR</i> <sub>2</sub> <sup>b</sup> (all data)	0.1824	0.1076	0.1533	0.2432	0.2229	0.2498

<sup>a</sup>*R*<sub>1</sub> = Σ(||*F*<sub>o</sub>| - |*F*<sub>c</sub>|)/Σ|*F*<sub>o</sub>|. <sup>b</sup> *wR*<sub>2</sub> = [Σ(|*F*<sub>o</sub>|<sup>2</sup> - |*F*<sub>c</sub>|<sup>2</sup>)<sup>2</sup>/Σ(*F*<sub>o</sub><sup>2</sup>)]<sup>1/2</sup>

**Table S2.** Classical hydrogen bonds linking the 2D layers into a 3D net in **1** [Å and °]

D–H···A	d(D–H)	d(H···A)	d(D···A)	<(DHA)
O3–H3B···O5#1	0.84	1.96	2.789(6)	167
O4–H4A···O2#2	0.82	2.00	2.805(4)	167

Symmetry transformations used to generate equivalent atoms: #1: 1/4 + x, 1/4 – y, –3/4 + z; #2: 1/4 + x, 1/4 – y, 1/4 + z

**Table S3.** Important Bond Distances (Å) and Angles (°) of **6**.

Zn1–N2	2.019(5)
Zn1–O1w	2.125(4)
Zn1–O2	1.950(4)
Zn1–O7	1.950(4)
Zn1–O2#1	2.433(4)
O1w–Zn1–O2	97.11(16)
O1w–Zn1–O7	95.57(16)
O2–Zn1–O7	128.96(17)
O1w–Zn1–N2	93.94(17)
O2–Zn1–N2	119.22(18)
O7–Zn1–N2	108.89(18)
O7–Zn1–O2#1	85.57(15)
O2–Zn1–O2#1	79.71(15)
N2–Zn1–O2#1	88.72(16)
O1w–Zn1–O2#1	176.59(14)

Symmetry code: #1 = –x + 1, –y, –z + 2