

## On the Electronic Structure and Stability of Icosahedral $r\text{-X}_2\text{Z}_{10}\text{H}_{12}$ and $\text{Z}_{12}\text{H}_{12}^{2-}$ Clusters; $r = \{\text{ortho, meta, para}\}$ , $\text{X} = \{\text{C, Si}\}$ , $\text{Z} = \{\text{B, Al}\}$ <sup>\*</sup>

Josep M. Oliva<sup>(a)</sup>, Paul von Ragué Schleyer<sup>(b)</sup>, Gabriel Aullón<sup>(c)</sup>, José I. Burgos<sup>(d)</sup>, Antonio Fernández-Barbero<sup>(e)</sup>, Ibon Alkorta<sup>(f)</sup>

(a) Instituto de Química-Física Rocasolano (CSIC), Serrano 119, Madrid 28006, Spain

(b) Center for Computational Chemistry, University of Georgia, Athens, Georgia 30602, USA

(c) Departament de Química Inorgànica and Institut de Química Teòrica i Computacional (IQTCUB), Universitat de Barcelona, Diagonal 647, Barcelona 08028, Spain

(d) Instituto de Ciencias Matemáticas, CSIC-UAM-UCM-UC3M, Serrano 113, Madrid 28006, Spain

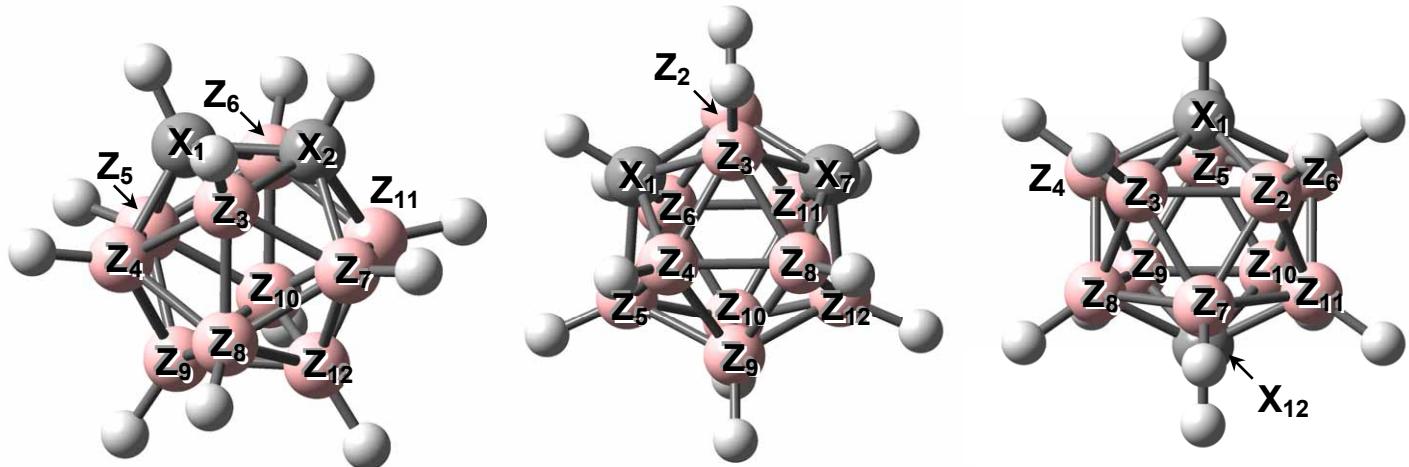
(e) Group of Complex Fluid Physics, Universidad de Almería, Almería 04120, Spain

(f) Instituto de Química Médica (CSIC), Juan de la Cierva 3, Madrid 28006, Spain

**SUPPLEMENTARY INFORMATION:** This section includes the experimental (where available) and computed symmetry-unique cage distances, Mulliken and natural bond orbital (NBO) charges [1], as well as the atom-in-molecules (AIM) topological analysis of the electron density for the  $\text{Z}_{12}\text{H}_{12}^{2-}$  and  $r\text{-X}_2\text{Z}_{10}\text{H}_{12}$  clusters;  $\text{X} = \{\text{C, Si}\}$ ,  $\text{Z} = \{\text{B, Al}\}$ ,  $r = \{\text{ortho, meta, para}\}$ . All computations carried out with the B3LYP/6-311+G(d,p) model. We also include the continuous shape measures  $S_{IC}$  [2] for the available crystal structures of the clusters included in this work (Table S3). In the colour PDF or HTML file, bond critical points, ring critical points and cage critical points - Figure S1, from (a) to (n) - are represented in small red, yellow and green spheres respectively.

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**Table S1.** Experimental (where available) and computed symmetry-unique cage distances for  $Z_{12}H_{12}^{2-}$  and  $r\text{-}X_2Z_{10}H_{12}$  clusters;  $X = \{\text{C, Si}\}$ ,  $Z = \{\text{B, Al}\}$ ,  $r = \{\text{ortho, meta, para}\}$ . All computations with the B3LYP/6-311+G(d,p) model. Distance = Computed/Experimental. Relative error:  $|\varepsilon_\Delta| = |(d_{\text{comp}} - d_{\text{exp}})/d_{\text{exp}}| \times 100$  (%).

$\downarrow Z \rightarrow X$	$X=Z$	Comp/Exp	$ \varepsilon_\Delta $							
Z=B	$Z_1Z_2$	1.786/1.785 [3]	0.1							
Z=Al	$Z_1Z_2$	2.702/2.702 [4]	0.0		Comp/Exp					
				Comp	R=Me		$\downarrow Z \rightarrow X$			
	Ortho	X=C [5]		X=Si	X=Si [6]	$ \varepsilon_\Delta $		Ortho	X=C	X=Si
Z=B	$X_1X_2$	1.625/1.626	0.4	2.314	2.325/2.308	0.7	Z=Al	$X_1X_2$	1.632	2.448
	$X_1Z_3$	1.720/1.717	0.2	2.139	2.138/2.117	1.0		$X_1Z_3$	2.158	2.620
	$X_1Z_4$	1.698/1.689	0.5	2.025	2.026/2.018	0.4		$X_1Z_4$	2.253	2.572
	$Z_3Z_4$	1.779/1.765	0.8	1.862	1.854/1.848	0.3		$Z_3Z_4$	2.566	2.714
	$Z_3Z_8$	1.764/1.760	0.2	1.777	1.779/1.770	0.5		$Z_3Z_8$	2.678	2.657
	$Z_4Z_5$	1.785/1.785	0.0	1.890	1.881/1.859	1.2		$Z_4Z_5$	2.579	2.752
	$Z_4Z_8$	1.781/1.773	0.5	1.775	1.773/1.774	0.1		$Z_4Z_8$	2.651	2.697
	$Z_4Z_9$	1.777/1.768	0.5	1.775	1.774/1.765	0.5		$Z_4Z_9$	2.650	2.680
	$Z_8Z_9$	1.793/1.789	0.2	1.789	1.788/1.783	0.3		$Z_8Z_9$	2.785	2.728
	$Z_9Z_{12}$	1.783/1.780	0.2	1.779	1.778/1.774	0.2		$Z_9Z_{12}$	2.711	2.699
Z=B	Meta	X=C [7]	$ \varepsilon_\Delta $	X=Si			Z=Al	Meta	X=C	X=Si
	$X_1Z_2$	1.693/1.693	0.0	2.033				$X_1Z_2$	2.176	2.562
	$X_1Z_4$	1.714/1.709	0.3	2.045				$X_1Z_4$	2.358	2.617
	$X_1Z_5$	1.711/1.708	0.2	2.032				$X_1Z_5$	2.207	2.586
	$Z_2Z_3$	1.789/1.778	0.6	2.036				$Z_2Z_3$	2.453	2.788
	$Z_3Z_4$	1.767/1.763	0.2	1.848				$Z_3Z_4$	2.515	2.688
	$Z_4Z_5$	1.783/1.777	0.3	1.872				$Z_4Z_5$	2.561	2.728
	$Z_4Z_8$	1.771/1.767	0.2	1.764				$Z_4Z_8$	2.651	2.664
	$Z_4Z_9$	1.780/1.777	0.2	1.784				$Z_4Z_9$	2.693	2.699
	$Z_5Z_9$	1.778/1.772	0.3	1.774				$Z_5Z_9$	2.678	2.682
	$Z_9Z_{10}$	1.793/1.782	0.6	1.798				$Z_9Z_{10}$	2.787	2.749
Z = B	Para	X=C [6]	$ \varepsilon_\Delta $ (%)	X=Si			Z=Al	Para	X=C	X=Si
	$X_1Z_2$	1.708/1.712	0.2	2.037				$X_1Z_2$	2.265	2.590
	$Z_2Z_3$	1.786/1.785	0.1	1.889				$Z_2Z_3$	2.575	2.755
	$Z_2-Z_7$	1.767/1.772	0.3	1.762				$Z_2-Z_7$	2.619	2.653

**Table S2.** Mulliken and NBO (Natural Bond Orbital) charges (in units of  $|e|$ ) for the  $Z_{12}H_{12}^{2-}$  and  $r\text{-}X_2Z_{10}H_{12}$  clusters included in this work,  $r = \{\text{ortho, meta, para}\}$ ,  $X = \{\text{C, Si}\}$ ,  $Z = \{\text{B, Al}\}$ . Computations on the optimized geometries at B3LYP/6-311+G(d,p) level of theory. Note: The atom labels follow from Figure 1 in main text and Table S1 above. The hydrogen labels are equivalent to cage atom labels, namely,  $H(j)$  is bound to  $Z(j)$  or  $X(j)$ .

Cluster	Atom	$q(\text{Mulliken})$	$q(\text{NBO})$	Cluster	Atom	$q(\text{Mulliken})$	$q(\text{NBO})$
$B_{12}H_{12}^{2-}$	$B_1$	-0.027	-0.175				
	$H_1$	-0.140	0.008				
$Al_{12}H_{12}^{2-}$	$Al_1$	0.033	0.117				
	$H_1$	-0.200	-0.283				
$o\text{-}C_2B_{10}H_{12}$	$C_1$	-0.146	-0.496	$o\text{-}Si_2B_{10}H_{12}$	$Si_1$	0.764	0.892
	$H_1$	0.247	0.300		$H_1$	0.024	-0.025
	$B_3$	-0.114	0.158		$B_3$	-0.655	-0.333
	$H_3$	0.019	0.054		$H_3$	0.061	0.058
	$B_4$	-0.158	0.000		$B_4$	-0.307	-0.286
	$H_4$	0.017	0.069		$H_4$	0.059	0.078
	$B_8$	0.127	-0.165		$B_8$	0.191	-0.185
	$H_8$	0.014	0.078		$H_8$	0.035	0.082
	$B_9$	0.122	-0.139		$B_9$	0.050	-0.153
	$H_9$	0.015	0.073		$H_9$	0.025	0.079
$m\text{-}C_2B_{10}H_{12}$	$C_1$	-0.052	-0.639	$m\text{-}Si_2B_{10}H_{12}$	$Si_1$	1.160	1.083
	$H_1$	0.233	0.299		$H_1$	0.020	-0.028
	$B_3$	-0.367	0.149		$B_3$	-0.720	-0.436
	$H_3$	0.037	0.066		$H_3$	0.070	0.074
	$B_4$	0.015	-0.021		$B_4$	-0.242	-0.280
	$H_4$	0.018	0.073		$H_4$	0.054	0.076
	$B_5$	-0.115	0.058		$B_5$	-0.414	-0.267
	$H_5$	0.016	0.056		$H_5$	0.045	0.074
	$B_9$	0.170	-0.176		$B_9$	0.182	-0.173
	$H_9$	0.014	0.080		$H_9$	0.035	0.079
$p\text{-}C_2B_{10}H_{12}$	$C_1$	-0.157	-0.664	$p\text{-}Si_2B_{10}H_{12}$	$Si_1$	0.995	1.094
	$H_1$	0.242	0.299		$H_1$	0.021	-0.028
	$B_2$	-0.038	0.005		$B_2$	-0.256	-0.289
	$H_2$	0.021	0.068		$H_2$	0.053	0.076

**Table S2 (cont)**

Cluster	Atom	<i>q</i> (Mulliken)	<i>q</i> (NBO)	Cluster	Atom	<i>q</i> (Mulliken)	<i>q</i> (NBO)
<i>o</i> -C <sub>2</sub> Al <sub>10</sub> H <sub>12</sub>	C <sub>1</sub>	-0.313	-1.239	<i>o</i> -Si <sub>2</sub> Al <sub>10</sub> H <sub>12</sub>	Si <sub>1</sub>	-0.011	-0.196
	H <sub>1</sub>	0.234	0.276		H <sub>1</sub>	0.012	-0.025
	Al <sub>3</sub>	0.292	0.805		Al <sub>3</sub>	0.108	0.392
	H <sub>3</sub>	-0.137	-0.255		H <sub>3</sub>	-0.103	-0.231
	Al <sub>4</sub>	0.037	0.442		Al <sub>4</sub>	0.031	0.294
	H <sub>4</sub>	-0.125	-0.230		H <sub>4</sub>	-0.102	-0.229
	Al <sub>8</sub>	0.239	0.240		Al <sub>8</sub>	0.136	0.186
	H <sub>8</sub>	-0.115	-0.246		H <sub>8</sub>	-0.101	-0.229
	Al <sub>9</sub>	0.092	0.239		Al <sub>9</sub>	0.204	0.201
	H <sub>9</sub>	-0.117	-0.244		H <sub>9</sub>	-0.104	-0.229
<i>m</i> -C <sub>2</sub> Al <sub>10</sub> H <sub>12</sub>	C <sub>1</sub>	-0.395	-1.623	<i>m</i> -Si <sub>2</sub> Al <sub>10</sub> H <sub>12</sub>	Si <sub>1</sub>	-0.028	-0.320
	H <sub>1</sub>	0.210	0.267		H <sub>1</sub>	0.004	-0.030
	Al <sub>3</sub>	0.254	0.819		Al <sub>3</sub>	0.118	0.408
	H <sub>3</sub>	-0.125	-0.231		H <sub>3</sub>	-0.104	-0.226
	Al <sub>4</sub>	0.126	0.503		Al <sub>4</sub>	0.084	0.296
	H <sub>4</sub>	-0.125	-0.238		H <sub>4</sub>	-0.103	-0.228
	Al <sub>5</sub>	0.242	0.505		Al <sub>5</sub>	0.169	0.313
	H <sub>5</sub>	-0.125	-0.231		H <sub>5</sub>	-0.106	-0.229
	Al <sub>9</sub>	0.058	0.208		Al <sub>9</sub>	0.086	0.176
	H <sub>9</sub>	-0.119	-0.243		H <sub>9</sub>	-0.101	-0.227
<i>p</i> -C <sub>2</sub> Al <sub>10</sub> H <sub>12</sub>	C <sub>1</sub>	-0.285	-1.604	<i>p</i> -Si <sub>2</sub> Al <sub>10</sub> H <sub>12</sub>	Si <sub>1</sub>	0.027	-0.324
	H <sub>1</sub>	0.210	0.271		H <sub>1</sub>	-0.001	-0.031
	Al <sub>2</sub>	0.143	0.500		Al <sub>2</sub>	0.098	0.299
	H <sub>2</sub>	-0.128	-0.233		H <sub>2</sub>	-0.103	-0.228

**Table S3.** Continuous shape measures ( $S_{IC}$ ) [2] for experimental available crystal structures of borane ( $B_{12}R_{12}^{2-}$ ), *o*-carborane, *m*-carborane, *p*-carborane, *o*-silaborane and alane  $Al_{12}R_{12}^{2-}$ .

### BORANES

Reference*	Mean B-B distance	Experimental values (Å)	$S_{IC}$
[8]		1.779	0.015
[8]		1.775	0.022
[9]		1.786	0.001
[10]		1.776	0.008
[11]		1.776	0.052
[12]		1.783	0.013
[12]		1.775	0.028
[13]		1.793	0.002
[14]		1.790	0.003
[14]		1.782	0.003
[15]		1.783	0.002
[16]		1.784	0.001
$(K^+)_2[B_{12}H_{12}]^{2-}$ [3]		1.779	0.001
$(Rb^+)_2[B_{12}H_{12}]^{2-}$ [3]		1.782	0.001
[17]		1.817	0.019
[18]		1.777	0.001
$(K^+)_3[B_{12}H_{12}]^{2-} I^-$ [3]		1.774	0.002
$(K^+)_3[B_{12}H_{12}]^{2-} I^-$ [3]		1.781	0.001
$(Cs^+)_3[B_{12}H_{12}]^{2-} I^-$ [3]		1.783	0.001
$(Rb^+)_3[B_{12}H_{12}]^{2-} Cl^-$ [18]		1.776	0.000
$(Rb^+)_3[B_{12}H_{12}]^{2-} Br^-$ [18]		1.778	0.000
$(Cs^+)_3[B_{12}H_{12}]^{2-} Br^-$ [18]		1.782	0.001
$(Cs^+)_3[B_{12}H_{12}]^{2-} Cl^-$ [18]		1.782	0.000
[19]		1.787	0.004
[19]		1.770	0.006
[20]		1.782	0.024
$(NH_4^+)_2[B_{12}H_{12}]^{2-}$ [3]		1.785	0.000
$(NH_4^+)_3[B_{12}H_{12}]^{2-} I^-$ [3]		1.781	0.000
[21]		1.776	0.001
$(NH_4^+)_3[B_{12}H_{12}]^{2-} Br^-$ [18]		1.782	0.000
[22]		1.775	0.002
[23]		1.763	0.008
[24]		1.785	0.007
[24]		1.785	0.031
[25]		1.782	0.001
[26]		1.776	0.004
[27]		1.773	0.005
[28]		1.792	0.002

\* Same reference for different row refers to a different crystal/counterion(s) (cations and anions other than  $B_{12}H_{12}^{2-}$ ). The ions are indicated in certain cases for clarity.

**Table S3 (cont)**

**ORTHO-CARBORANE**

Reference*	Experimental distances (Å)										$S_{IC}$
	C-C	C-B <sub>A</sub>	C-B <sub>B</sub>	B <sub>A</sub> -B <sub>B</sub>	B <sub>A</sub> -B <sub>C</sub>	B <sub>B</sub> -B <sub>B</sub>	B <sub>B</sub> -B <sub>C</sub>	B <sub>B</sub> -B <sub>D</sub>	B <sub>C</sub> -B <sub>D</sub>	B <sub>D</sub> -B <sub>D</sub>	
[29]	1.647	1.679	1.711	1.743	1.726	1.765	1.776	1.773	1.781	1.756	0.110
[5]	1.645	1.717	1.703	1.771	1.760	1.778	1.778	1.774	1.785	1.785	0.081
[5]	1.626	1.717	1.689	1.765	1.760	1.785	1.773	1.768	1.789	1.780	0.126
[30]	1.633	1.723	1.711	1.764	1.763	1.782	1.778	1.776	1.788	1.796	0.087
[30]	1.637	1.700	1.701	1.767	1.744	1.782	1.767	1.765	1.774	1.792	0.086
[30]	1.620	1.712	1.686	1.765	1.757	1.775	1.774	1.773	1.786	1.775	0.117
[31]	1.595	1.667	1.680	1.710	1.707	1.740	1.742	1.730	1.752	1.762	0.103
[32]	1.626	1.706	1.692	1.796	1.782	1.767	1.781	1.781	1.775	1.777	0.101
[33]	1.627	1.704	1.684	1.760	1.752	1.773	1.772	1.663	1.770	1.755	0.113
[33]	1.599	1.693	1.671	1.762	1.761	1.778	1.753	1.751	1.762	1.777	0.142
[33]	1.610	1.675	1.643	1.699	1.759	1.803	1.742	1.718	1.779	1.869	0.245
[34]	1.708	1.712	1.694	1.719	1.760	1.796	1.699	1.717	1.790	1.910	0.114
[34]	1.584	1.704	1.645	1.660	1.668	1.724	1.774	1.768	1.737	1.788	0.195
[34]	1.706	1.705	1.703	1.729	1.755	1.774	1.712	1.717	1.763	1.806	0.051
[35]	1.429	1.658	1.804	1.727	1.651	1.750	1.704	1.776	1.721	1.474	0.304
[36]	1.666	1.700	1.685	1.753	1.723	1.783	1.752	1.760	1.764	1.762	0.080
[37]	1.624	1.713	1.693	1.768	1.769	1.777	1.771	1.768	1.784	1.776	0.107
[37]	1.625	1.704	1.690	1.758	1.763	1.775	1.773	1.774	1.778	1.764	0.101
[38]	1.622	1.694	1.691	1.745	1.743	1.752	1.757	1.738	1.762	1.763	0.110
[39]	1.683	1.719	1.713	1.767	1.770	1.711	1.746	1.742	1.765	1.778	0.070

\* Same reference for a different row refers to a different crystal/counterion(s) (see reference).

**META-CARBORANE**

Ref*	Experimental distances (Å)										$S_{IC}$	
	C···C	B <sub>A</sub> -B <sub>A</sub>	B <sub>B</sub> -B <sub>B</sub>	B <sub>D</sub> -B <sub>D</sub>	C-B <sub>A</sub>	C-B <sub>B</sub>	C-B <sub>C</sub>	B <sub>A</sub> -B <sub>B</sub>	B <sub>B</sub> -B <sub>C</sub>	B <sub>B</sub> -B <sub>D</sub>	B <sub>C</sub> -B <sub>D</sub>	
[7]	2.614	1.778	1.767	1.782	1.693	1.709	1.708	1.763	1.777	1.777	1.772	0.101
[34]	2.715	1.674	1.706	1.737	1.683	1.671	1.706	1.684	1.720	1.727	1.724	0.062
[34]	2.748	1.693	1.707	1.747	1.696	1.704	1.747	1.702	1.743	1.731	1.740	0.055
[34]	2.601	1.779	1.768	1.785	1.687	1.707	1.710	1.761	1.777	1.772	1.770	0.109
[34]	2.695	1.656	1.685	1.735	1.672	1.676	1.723	1.689	1.637	1.728	1.720	0.068
[34]	2.666	1.769	1.731	1.767	1.695	1.715	1.735	1.735	1.755	1.757	1.750	0.050
[37]	2.609	1.774	1.762	1.768	1.693	1.705	1.708	1.771	1.767	1.771	1.756	0.103

\* Same reference for a different row refers to a different crystal/counterion(s) (see reference).

**PARA-CARBORANE**

Reference*	Experimental distances (Å)				$S_{IC}$
	C···C	B-B	C-B	B-B <sup>\$</sup>	
[7]	3.071	1.777	1.704	1.765	0.117
[7]	3.083	1.785	1.712	1.772	0.118
[37]	3.052	1.781	1.704	1.762	0.137
[37]	3.060	1.781	1.706	1.760	0.131

\* Same reference for a different row refers to a different crystal/counterion(s) (see reference).

B-B: same pentagon

B-B<sup>\$</sup>: between pentagons

### ***ORTHO-SILABORANE***

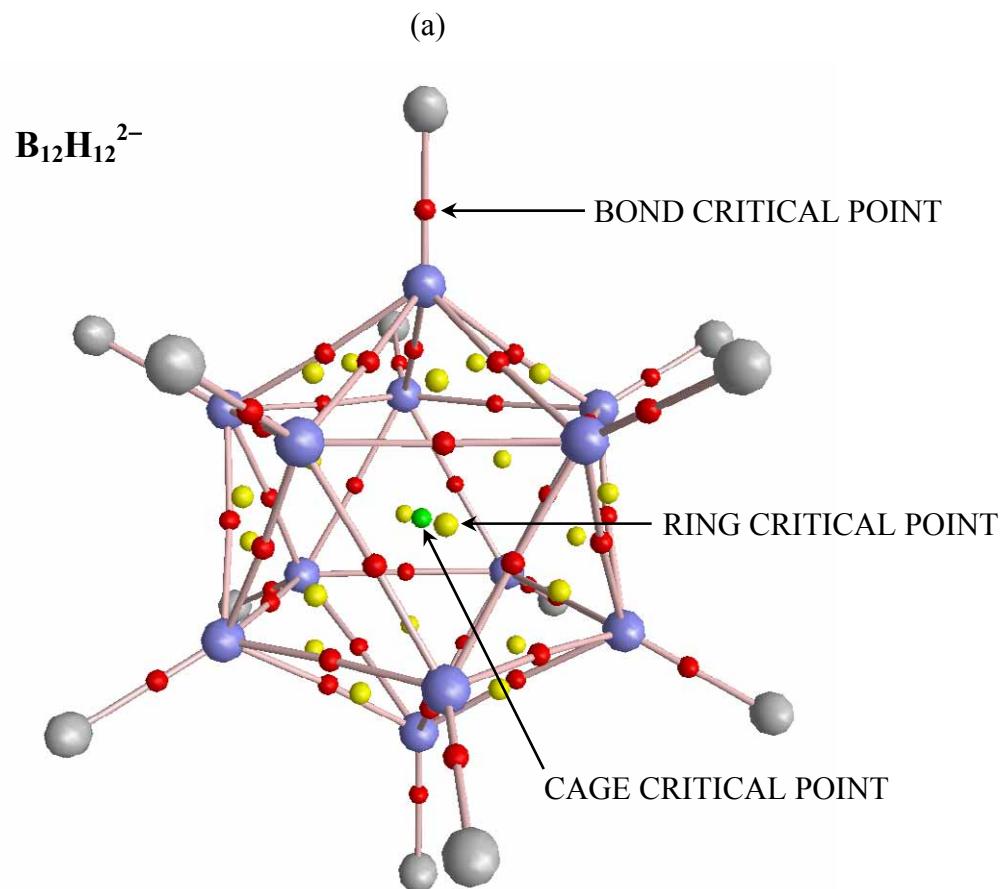
Reference	Experimental distances ( $\text{\AA}$ )										$S_{\text{IC}}$
	Si-Si	Si-B <sub>A</sub>	Si-B <sub>B</sub>	B <sub>A</sub> -B <sub>B</sub>	B <sub>A</sub> -B <sub>C</sub>	B <sub>B</sub> -B <sub>B</sub>	B <sub>B</sub> -B <sub>C</sub>	B <sub>B</sub> -B <sub>D</sub>	B <sub>C</sub> -B <sub>D</sub>	B <sub>D</sub> -B <sub>D</sub>	
[40]	2.314	2.111	2.014	1.854	1.780	1.863	1.771	1.766	1.782	1.780	0.575
[41]	2.304	2.110	2.021	1.844	1.780	1.864	1.759	1.764	1.775	1.750	0.605
[6]	2.308	2.115	2.017	1.849	1.770	1.859	1.770	1.767	1.782	1.774	0.585
[42]	2.308	2.115	2.017	1.849	1.770	1.859	1.770	1.767	1.782	1.774	0.585

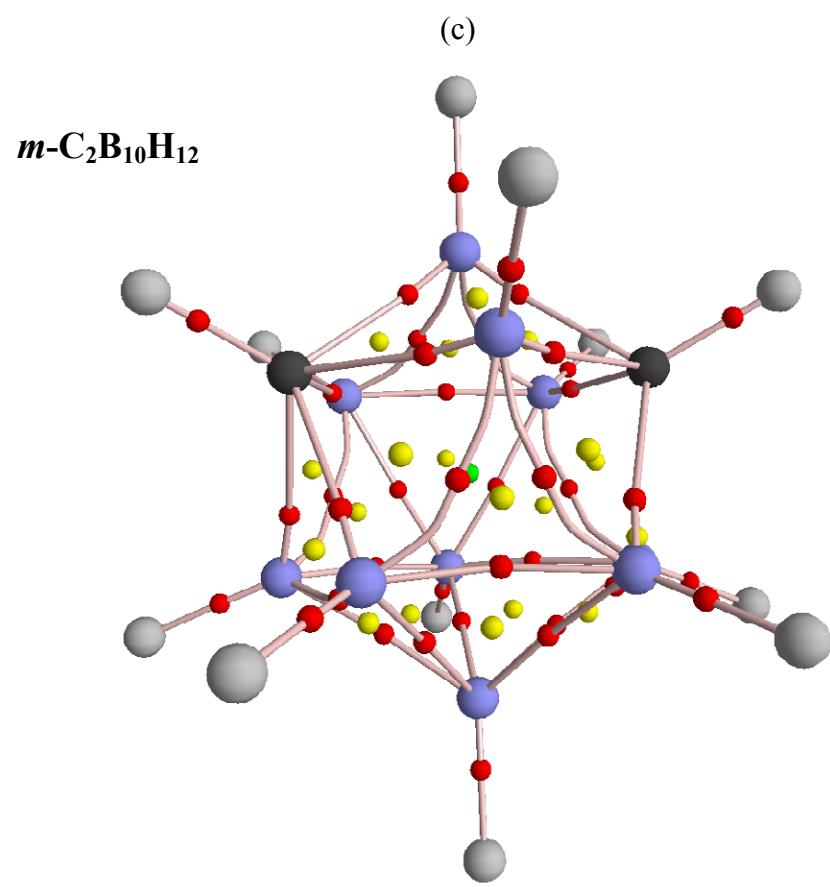
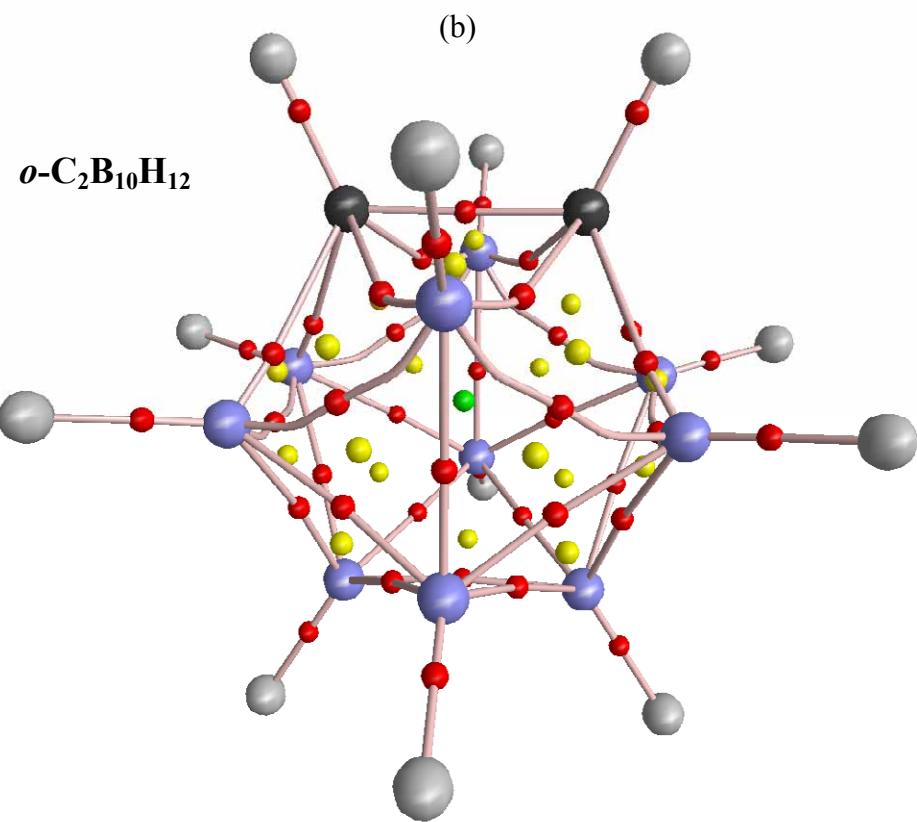
### **ALANES**

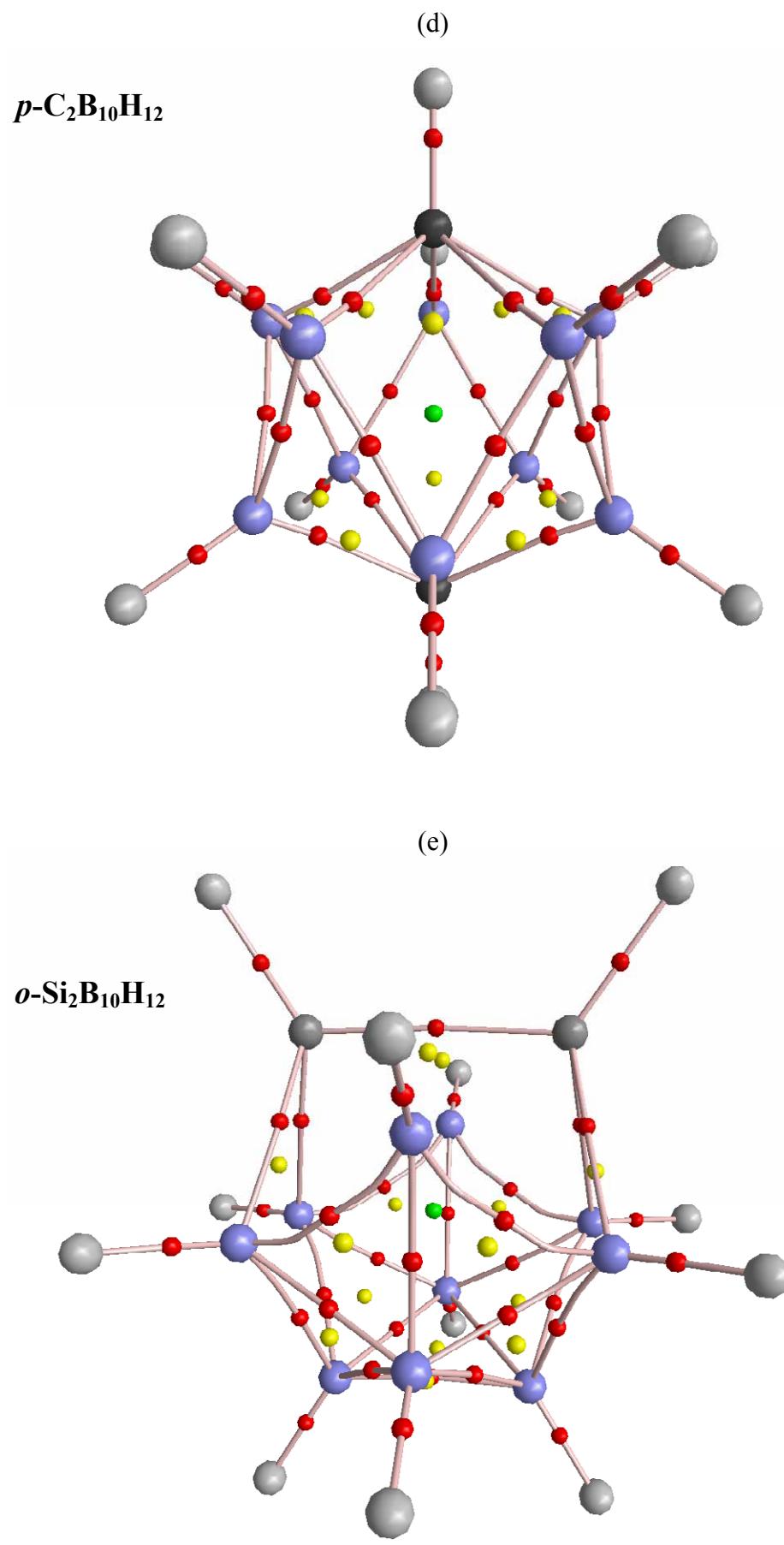
Reference*	Experimental values ( $\text{\AA}$ )		$S_{\text{IC}}$
	Mean Al-Al distance		
[43]	2.686		0.001
[4]	2.701		0.157
[4]	2.702		0.146
[44]	2.684		0.019
[44]	2.685		0.024

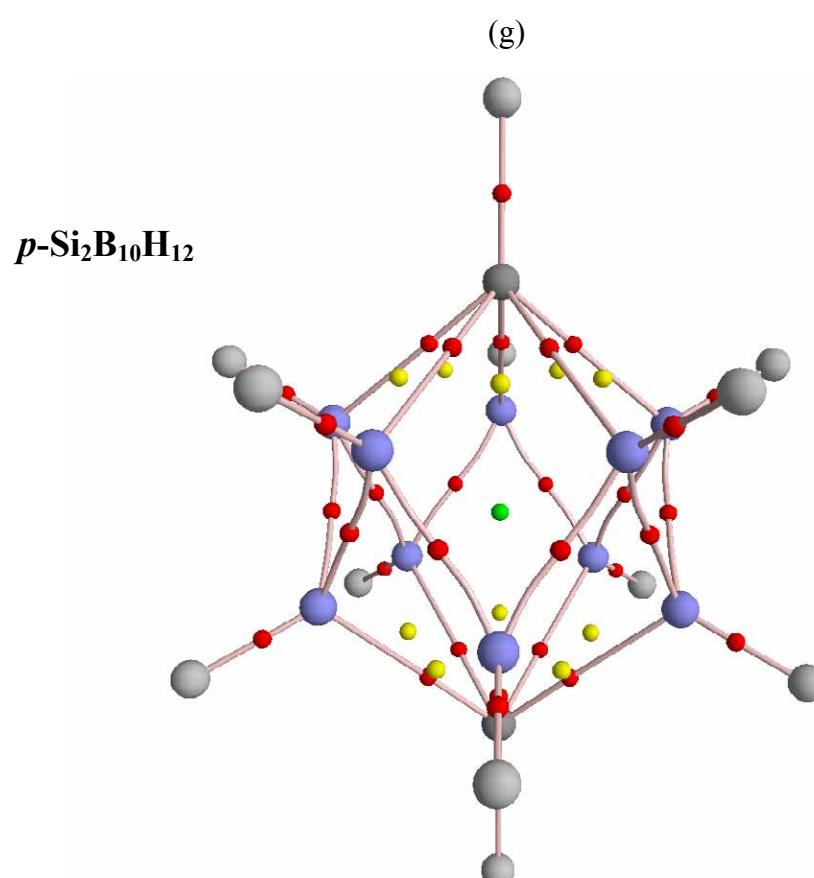
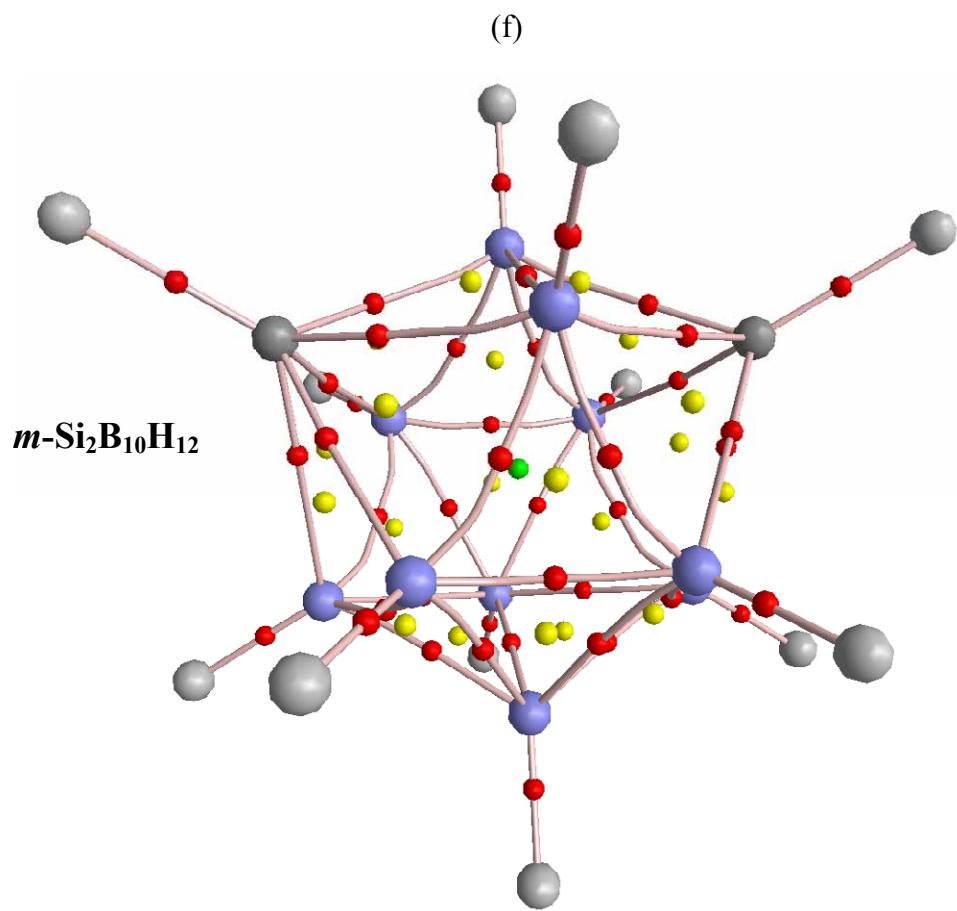
\* Same reference for a different row refers to a different crystal/counterion(s) (see reference).

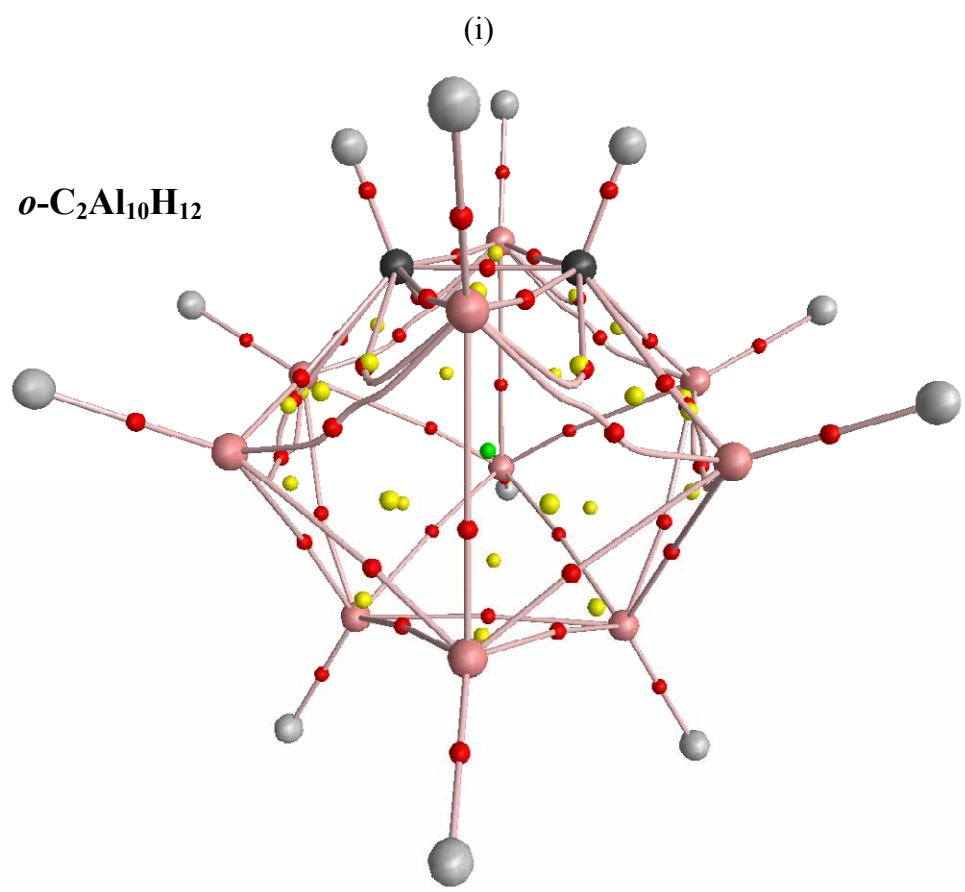
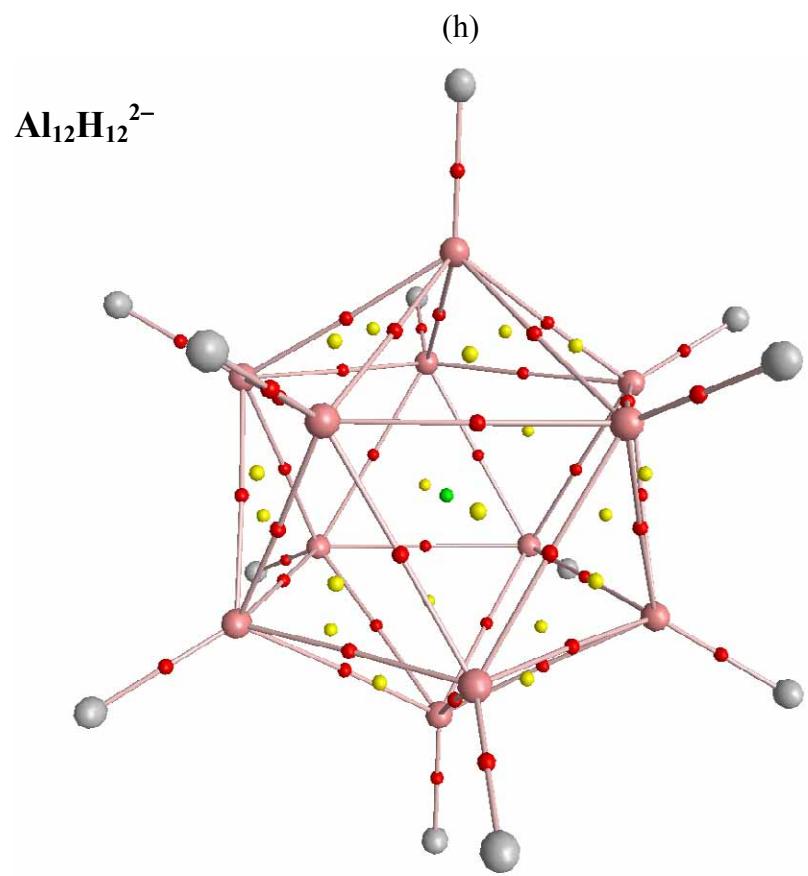
**Figure S1.** Topological analysis of the electron density in  $Z_{12}H_{12}^{2-}$  and  $r\text{-}X_2Z_{10}H_{12}$  clusters with  $X = \{\text{C}, \text{Si}\}$  and  $Z=\{\text{B}, \text{Al}\}$ . (a)  $\text{B}_{12}\text{H}_{12}^{2-}$ , (b) *ortho*- $\text{C}_2\text{B}_{10}\text{H}_{12}$ , (c) *meta*- $\text{C}_2\text{B}_{10}\text{H}_{12}$ , (d) *para*- $\text{C}_2\text{B}_{10}\text{H}_{12}$ , (e) *ortho*- $\text{Si}_2\text{B}_{10}\text{H}_{12}$ , (f) *meta*- $\text{Si}_2\text{B}_{10}\text{H}_{12}$ , (g) *para*- $\text{Si}_2\text{B}_{10}\text{H}_{12}$ , (h)  $\text{Al}_{12}\text{H}_{12}^{2-}$ , (i) *ortho*- $\text{C}_2\text{Al}_{10}\text{H}_{12}$ , (j) *meta*- $\text{C}_2\text{Al}_{10}\text{H}_{12}$ , (k) *para*- $\text{C}_2\text{Al}_{10}\text{H}_{12}$ , (l) *ortho*- $\text{Si}_2\text{Al}_{10}\text{H}_{12}$  , (m) *meta*- $\text{Si}_2\text{Al}_{10}\text{H}_{12}$  , (n) *para*- $\text{Si}_2\text{Al}_{10}\text{H}_{12}$ . Computations with the B3LYP/6-311+G(d,p) model. All geometries correspond to energy minima.

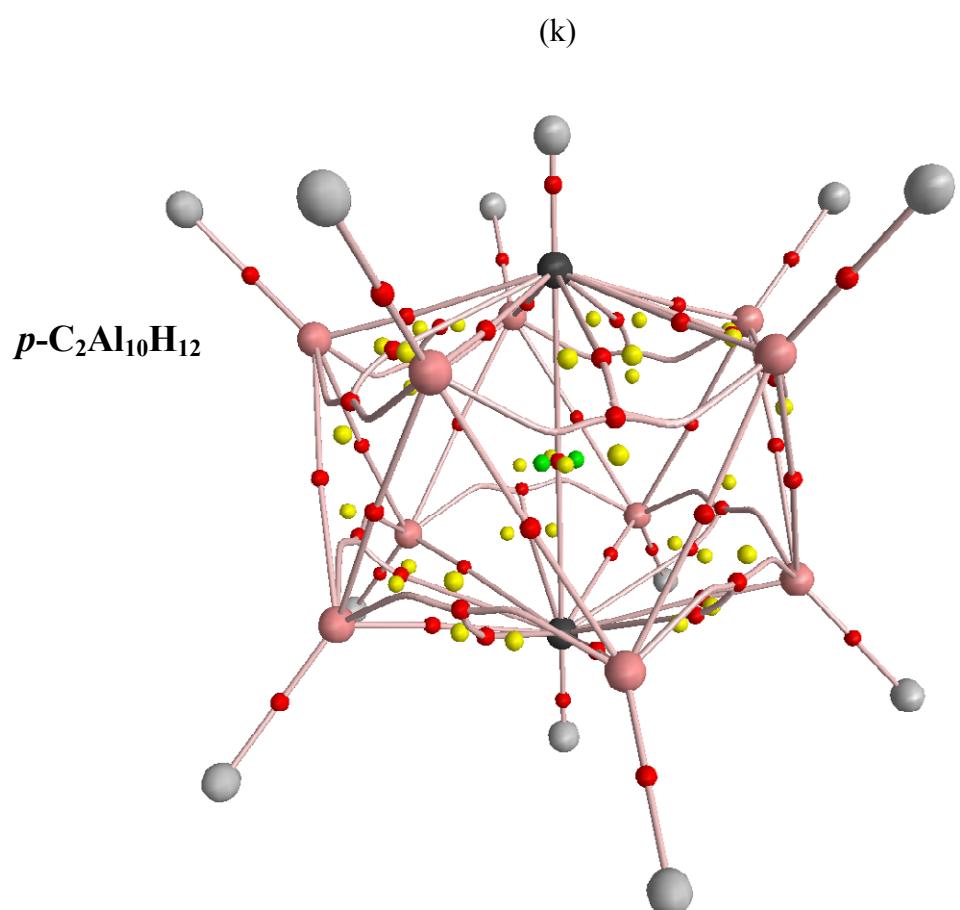
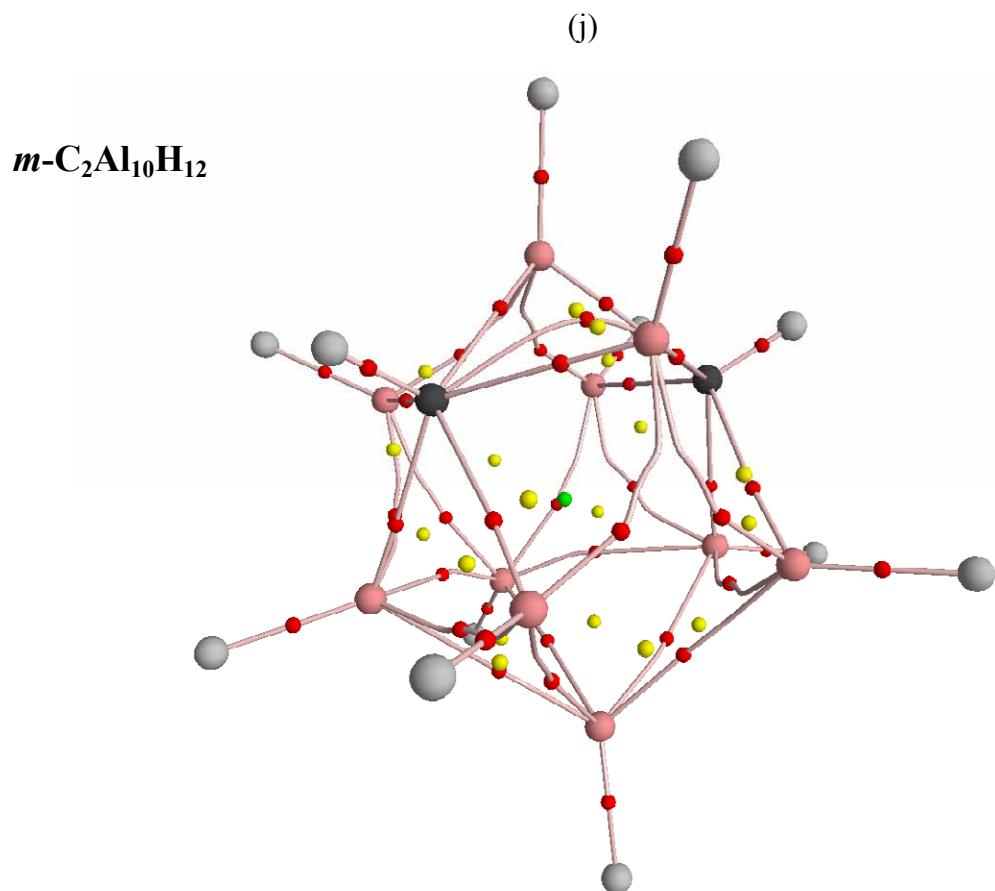


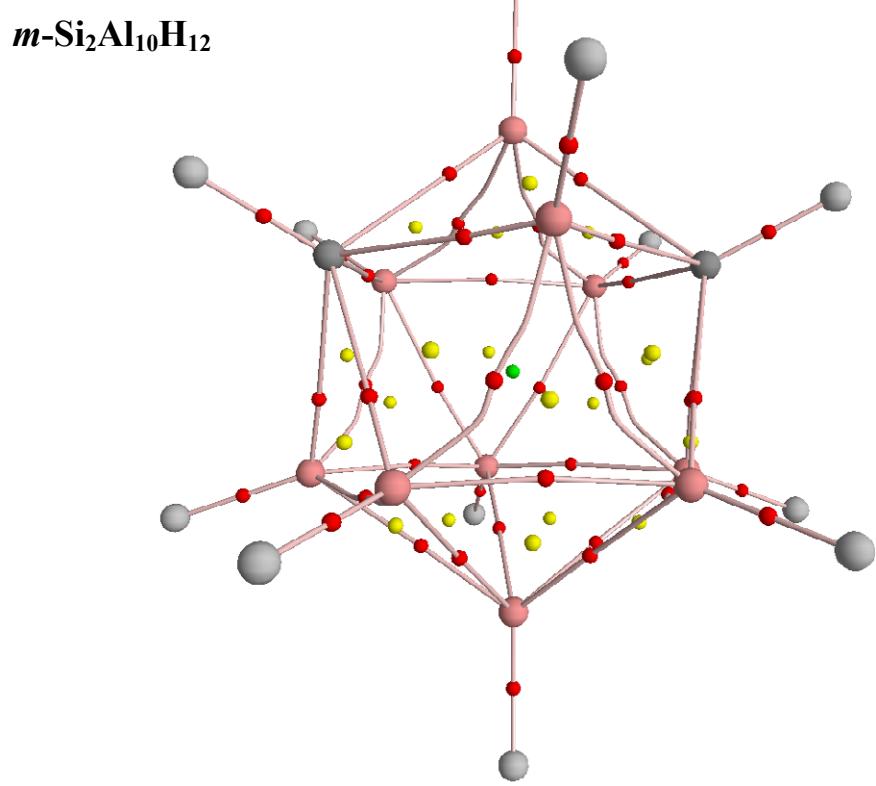
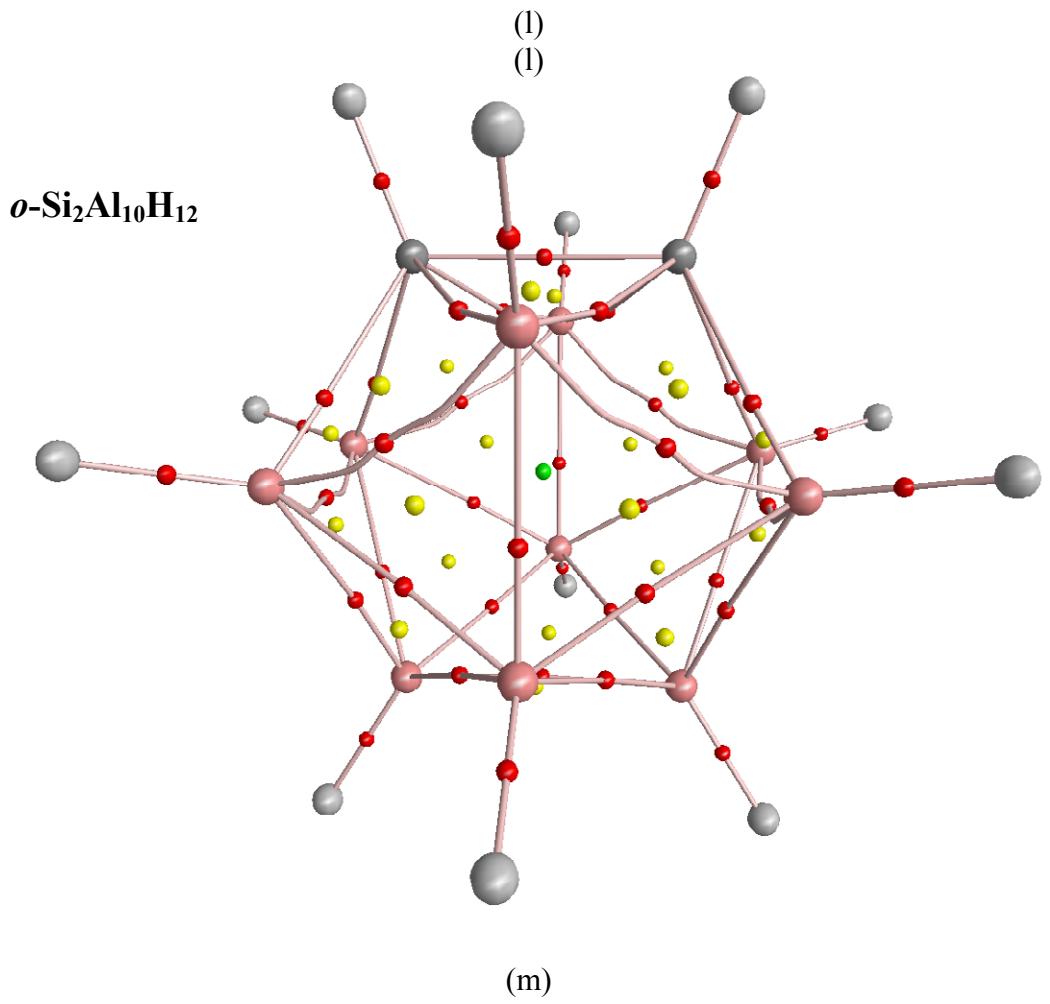


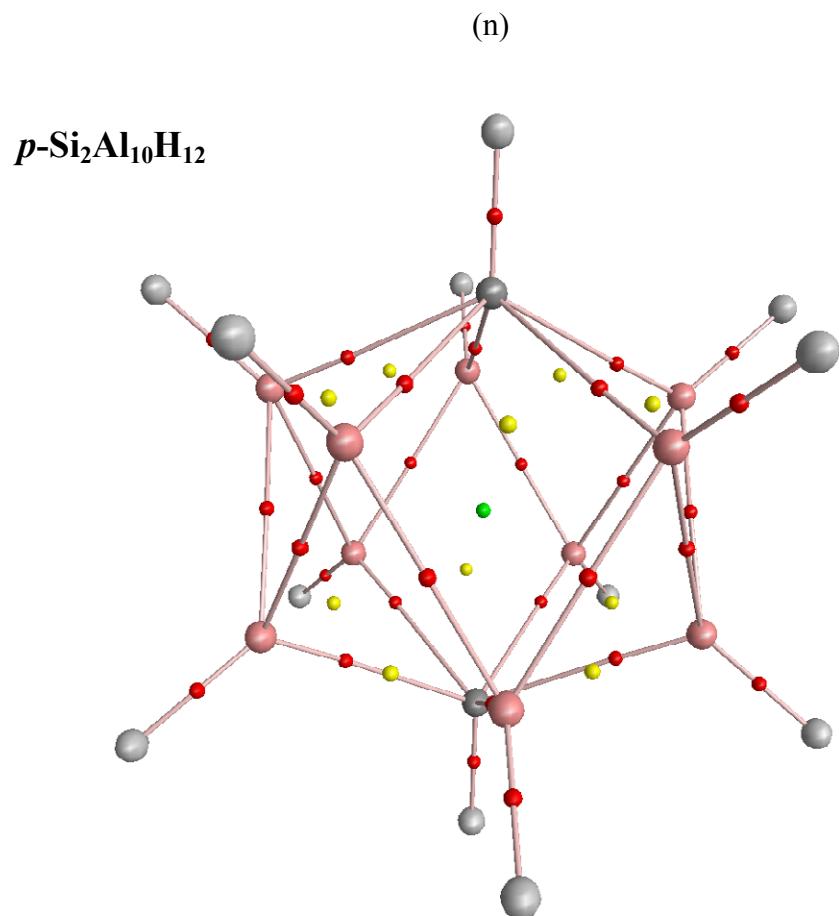












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