

Supplementary Information for “A comparative MD study of the local structure of polymer semiconductors P3HT and PBT TT”

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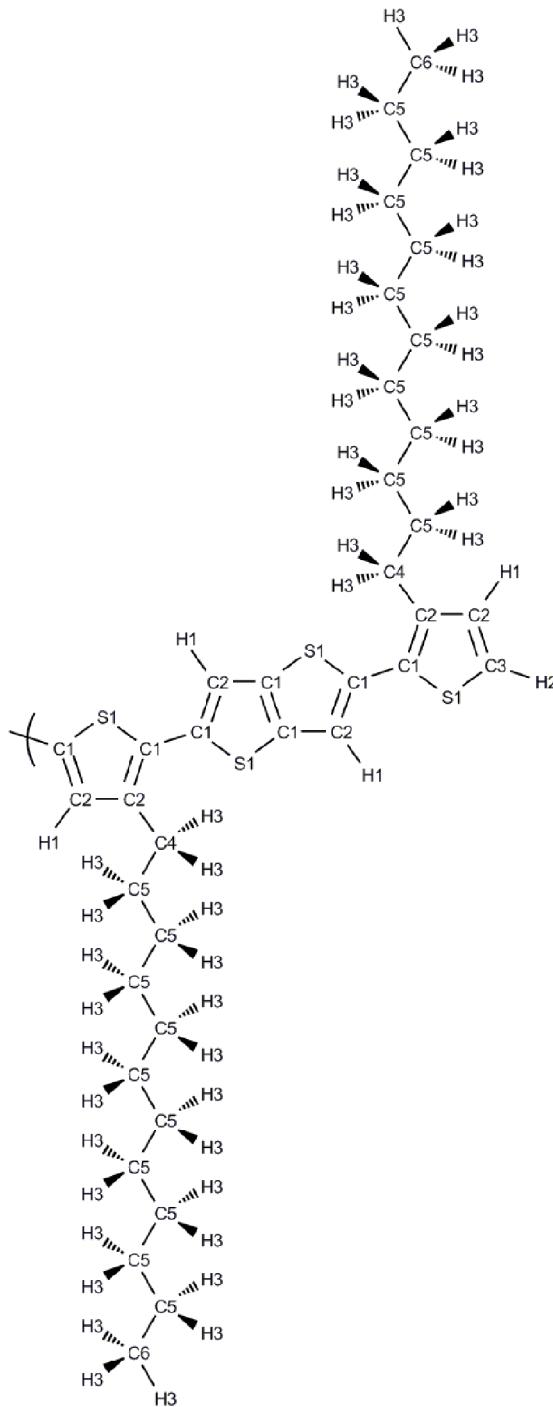


Fig. S1 Chemical structure of the end monomer in a poly(2,5-bis(3-alkylthiophen-2-yl)thieno[3,2-b]thiophene) (PBT TT) chain. Atoms of different type on our atomistic model are labelled with different numerical suffixes.

Table S1 Non-bonded parameters (LJ diameter σ_{ii} , LJ interaction strength ε_{ii} , and charge q_i) used in atomistic simulation models of PBTTT. PBTTT atom types are defined in Fig. S1.

Atom type i	$\sigma_{ii}/\text{\AA}$	$\varepsilon_{ii}/\text{kcal mol}^{-1}$	q_i/e
S1 ^a	3.550	0.250	-0.1496
C1 ^a	3.550	0.070	0.0748
C2 ^a	3.550	0.070	-0.1817
C3 ^a	3.550	0.070	-0.1412
C4 ^b	3.500	0.066	0.0617
C5 ^c	3.500	0.066	-0.1200
C6 ^c	3.500	0.066	-0.1800
H1 ^a	2.420	0.030	0.1817
H2 ^a	2.420	0.030	0.2160
H3 ^c	2.500	0.030	0.0600

^a Adapted from Ref. [1] (see our previous work for details [3]), with LJ parameters from OPLS-AA model [2].

^b From OPLS-AA model [2], with charge adjusted for monomer neutrality.

^c From OPLS-AA model [2].

Table S2 Parameters used in harmonic bond length potential $V_{\text{bond}}(l) = k_l(l - l_0)^2/2$ in atomistic simulation models of PBTTT. PBTTT atom types are defined in Fig. S1.

Bond type	$l_0/\text{\AA}$	$k_l/\text{kcal mol}^{-1} \text{\AA}^{-2}$
S1-C1 ^a	1.732	582.50
C1-C2 ^a	1.364	1028.54
C2-C2 ^a	1.428	906.20
C2-H1 ^a	1.080	741.26
C1-C1 ^a	1.430	784.58
C3-H2 ^a	1.080	741.26
C2-C4 ^b	1.510	634.00
C4/C5-C5/C6 ^b	1.529	536.00
C4/C5/C6-H3 ^b	1.090	680.00

^a Adapted from Ref. [1] (see our previous work for details [3]).

^b From Ref. [2].

15

Table S3 Parameters used in harmonic bond angle potential $V_{\text{angle}}(\theta) = k_\theta(\theta - \theta_0)^2/2$ in atomistic simulation models of PBTTT. PBTTT atom types are defined in Fig. S1.

Angle type	$\theta_0/\text{deg.}$	$k_\theta/\text{kcal mol}^{-1} \text{rad}^{-2}$
C1-S1-C1 ^a	92.774	172.720
S1-C1-C2 ^a	110.292	172.720
C1-C2-C2 ^a	113.322	79.164
C2-C2-H1 ^a	123.700	70.526
C2-C1-C1 ^a	130.140	109.388
C2-C3-H2 ^a	125.100	70.526
C1-C2-H1 ^a	122.979	70.526
S1-C1-C1 ^a	119.569	83.480
S1-C3-H2 ^a	124.609	57.574
C2-C2-C4 ^b	123.700	140.000
C1-C2-C4 ^b	122.979	140.000
C2-C4-C5 ^c	114.000	126.000
C2-C4-H3 ^c	109.500	70.000
C4/C5-C5/C6 ^c	112.700	116.700
C4/C5/C6-C4/C5/C6-H3 ^c	110.700	75.000
H3-C4/C5/C6-H3 ^c	107.800	66.000

^a Adapted from Ref. [1] (see our previous work for details [3]).

^b θ_0 adapted from Ref. [1]; k_θ from Ref. [2].

^c From Ref. [2].

25

References

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- 3 D. M. Huang, R. Faller, K. Do and A. J. Moule, *Journal of Chemical Theory and Computation*, 2010, **6**, 526-537.