

Electronic Supplementary Information

Excited-state hydrogen relay along a blended-alcohol chain as a model system of a proton wire: deuterium effect on the reaction dynamics

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Table S1 List of abbreviations

| Symbol | Word | Symbol | Word |
|----------------------|---|------------------------|-------------------------------------|
| 7HQ | 7-Hydroxyquinoline | ESPT | Excited-state proton transfer |
| 7DQ | Deuterated 7HQ | ESDT | Excited-state deuteron transfer |
| ROH | Normal alcohol | H bond | Hydrogen bond |
| ROD ^a | Deuterated alcohol | <i>K</i> | Association constant |
| N | Normal species of 7DQ | α | Proton-donating ability (acidity) |
| T | Tautomeric species of 7DQ | β | Proton-accepting ability (basicity) |
| Nc | Cyclic 7DQ·(ROD) ₂ complex | <i>k</i> _{PT} | Rate constant of ESPT |
| S^b | Relatively strong-acidic alcohol in Nc | <i>k</i> _{DT} | Rate constant of ESDT |
| W^b | Relatively weak-acidic alcohol in Nc | KIE ^c | Kinetic isotope effect |

^a MeOD, EtOD, and BuOD are symbols for deuterated methanol, ethanol, and *tert*-butanol, respectively.

^b In a blended alcohol chain. ^c Calculated as *k*_{PT}/*k*_{DT}.

Table S2 Alcohol-dependent K and k_{DT} in n -heptane

| Alcohol | α^a | β^a | K/M^{-2} | $k_{\text{DT}}^{-1}/\text{ps}$ | KIE^b |
|----------------------|------------|-----------|---------------------------------------|--------------------------------|----------------|
| Methanol | 0.93 | 0.62 | 8400 ^c (9100) ^d | 430 | 6.8 |
| Ethanol | 0.83 | 0.77 | 5200 (6800) | 560 | 7.8 |
| <i>tert</i> -Butanol | 0.68 | 1.01 | 3400 (4200) | 920 | 5.6 |

^a Proton-donating (α) and -accepting abilities (β), which signify the acidity and the basicity, respectively, of ROH because the values of ROD are not available in literatures.^{46–48} ^b Kinetic isotope effect calculated as $k_{\text{PT}}/k_{\text{DT}}$.³⁴ ^c For 7DQ·(ROD)₂.³³ ^d For 7HQ·(ROH)₂.³³

Table S3 Fluorescence kinetic constants of 7DQ in ROD-added *n*-heptane^a

| Alcohol ^b | N* fluorescence decay | | T* fluorescence | |
|----------------------|-------------------------------------|--------------------------|--------------------------|---------------------------|
| | $\tau_{\text{fast}}^{\text{c}}$ /ps | τ_{slow} /ps | τ_{rise} /ps | τ_{decay} /ps |
| MeOD | 430 | 1800 (10%) ^d | 430 | 4900 |
| EtOD | 560 | 1700 (13%) | 560 | 5300 |
| BuOD | 920 | 1600 (13%) | 920 | 5000 |

^a N* and T* fluorescence were monitored at 420 nm and 550 nm, respectively, after excitation of Nc at 355 nm. ^b The concentration of each alcohol was kept at 30 mM in *n*-heptane. ^c ESDT time (k_{DT}^{-1}).
^d Initial amplitude percentage of the slow decay component.