Supporting Information

A novel visible-light-response plasmonic photocatalyst CNT/Ag/AgBr and its photocatalytic properties

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Fig. S1 (a) SEM image and (b) back scattering electron (BSE) image of the Ag/AgBr composite, in which both the (a) and (b) micrographs are taken in the same field. (c) The overall SEM image of the Ag/AgBr.
As shown in Fig. S2 a, the CNT/Ag/AgBr is mainly in the shape of particle. The surface of the particle is enlarged in Fig. S2 b. It is clear that CNTs cover the surface of the Ag/AgBr particles, and the diameter size of the CNTs are mainly at about 20-30 nm.

**Fig. S2** The SEM images of CNT/Ag/AgBr with a high CNT content of about 41.4 at% (CNT : AgNO₃=5 wt%)
As shown in Fig. S3, it is clear that the addition of the graphite does not enhance the photocatalytic ability of Ag/AgBr obviously. And the CNT/Ag/AgBr showed much higher photocatalytic ability than Ag/AgBr and graphite/Ag/AgBr. The emphasis of this work is investigating the effect of CNTs on the photoactivity of the CNT/Ag/AgBr composite. This is just a comparison of common carbon material and CNTs in this system.

**Fig. S3** The comparison of the photocatalytic ability of pure Ag/AgBr, 1.4 at % graphite/Ag/AgBr and 1.4 at % CNT/Ag/AgBr.
The degradation of 4-chlorophenol under visible-light irradiation with 1 g L\(^{-1}\) catalyst.

Fig. S4 The degradation of 4-chlorophenol (4-CP) under visible-light irradiation.
In order to investigate the degradation products clearly, the LC/MS were used to investigate the dye solution (50 mg L\(^{-1}\)) which had been degraded for 40 min and 3 h by 1.4 at % CNT/Ag/AgBr. The MS results of the degradation products with Ag/AgBr can be seen as follows (Fig. S5). The degradation products with Ag/AgBr showed the similar results.

![Graph (a)](image1)

![Graph (b)](image2)
Fig. S5 The degradation products analyzed by MS analysis (a, b, c and d).
Fig. S6 The TOC removal of the dye solution in the presence of the catalyst for 1 h.