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*Supporting Information*

**Gas Storage Potential of ExBox<sup>4+</sup> and its Li-decorated derivative**

**Ranjita Das and Pratim Kumar Chattaraj\***

Department of Chemistry and Center for Theoretical Studies, Indian Institute of Technology

Kharagpur, Kharagpur – 721302, West Bengal, India

\*Author for correspondence: Email: [pkc@chem.iitkgp.ernet.in](mailto:pkc@chem.iitkgp.ernet.in)

**Table S1:** Geometrical parameters of hydrogen adsorbed complexes

<b>System</b>	<b>C-H bond length range(Å)</b>	<b>N-H bond length (Å)</b>	<b>Distance between hydrogen and centroid of ring (Å)</b>			<b>Geometrical parameter for ExBox</b>			
			<b>Pyridinium ring</b>	<b>Phenyl ring (T)</b>	<b>Phenyl ring (S)</b>	<b>C-C bond length (Å)</b>	<b>C-N bond length (Å)</b>	<b>&lt;C-C-C-C (deg)</b>	<b>&lt;N-C-C (deg)</b>
(2H <sub>2</sub> ) <sub>endo</sub> @ExBox <sup>4+</sup>	2.872-3.090	-	-	-	2.788	1.481	1.495	146.455	110.639
(3H <sub>2</sub> ) <sub>endo</sub> @ExBox <sup>4+</sup>	2.873-3.023	-	2.671	2.814	1.481	1.495	146.768	110.645	
(8H <sub>2</sub> ) <sub>exo</sub> @ExBox <sup>4+</sup>	2.978-3.148	2.967	2.690	2.691	2.669	1.481	1.494	146.007	110.585
(8H <sub>2</sub> ) <sub>exo</sub> +(2H <sub>2</sub> ) <sub>endo</sub> @ExBox <sup>4+</sup>	2.945(2.878) <sup>a</sup> -3.114(3.103) <sup>a</sup>	2.974	2.670	2.684	2.685 (2.790) <sup>a</sup>	1.480	1.494	145.586	110.470
(8H <sub>2</sub> ) <sub>exo</sub> +(3H <sub>2</sub> ) <sub>endo</sub> @ExBox <sup>4+</sup>	2.974(2.876) <sup>a</sup> -3.113(3.011) <sup>a</sup>	2.971	2.683	2.716 (3.457) <sup>a</sup>	2.686 (2.764) <sup>a</sup>	1.480	1.495	146.068	110.517
(12H <sub>2</sub> ) <sub>exo</sub> @ExBox <sup>4+</sup>	2.975-3.183	3.044	2.689	2.689	2.679	1.481	1.494	146.817	110.603
(12H <sub>2</sub> ) <sub>exo</sub> +(2H <sub>2</sub> ) <sub>endo</sub> @ExBox <sup>4+</sup>	2.970(2.876) <sup>a</sup> -3.290(3.117) <sup>a</sup>	3.045	2.686	2.690	2.654 (2.790) <sup>a</sup>	1.481	1.494	145.157	110.112
(12H <sub>2</sub> ) <sub>exo</sub> +(3H <sub>2</sub> ) <sub>endo</sub> @ExBox <sup>4+</sup>	2.980(2.880) <sup>a</sup> -3.301(3.004) <sup>a</sup>	3.046	2.699	2.695 (2.678) <sup>a</sup>	2.654 (2.775) <sup>a</sup>	1.480	1.495	145.442	110.465

a the values in parenthesis indicate the endohedral hydrogen

**Table S2:** Bond critical point data (a.u.) calculated at wB97x-D/6-311G (d, p) level for hydrogen bound ExBox<sup>4+</sup> complexes

<b>System</b>	<b>bond type</b>	$\rho$	$\nabla^2 \rho$	$G_{bcp}$	$V_{bcp}$	$H_{bcp}$
(2H <sub>2</sub> ) <sub>endo</sub> @ExBox <sup>4+</sup>	N-H	0.00375	0.01304	0.00243	-0.00161	0.00083
	C-H	0.00608	0.01661	0.00335	-0.00255	0.00080
(3H <sub>2</sub> ) <sub>endo</sub> @ExBox <sup>4+</sup>	N-H	0.00385	0.01348	0.00252	-0.00167	0.00085
	C-H	0.00586	0.01620	0.00325	-0.00244	0.00080
	H-H	0.00025	0.00095	0.00016	-0.00008	0.00008
3H <sub>2</sub> @model	N-H	0.00468	0.01601	0.00305	-0.00209	0.00096
	C-H	0.00468	0.01526	0.00292	-0.00202	0.00090
5H <sub>2</sub> @model	C-H	0.004055	0.01473	0.00275	-0.00182	0.00093
	H-H	0.003183	0.01031	0.00192	-0.00125	0.00066

**Table S3:** Energy decomposition analysis (kcal/mol) of (nM)<sub>endo/exo</sub>@ExBox<sup>4+</sup> (M= H<sub>2</sub>, CO; n=2, 8) complexes at B3LYP-D/DZP level

Energy terms	(2H <sub>2</sub> ) <sub>endo</sub> @ExBox <sup>4+</sup>	(8H <sub>2</sub> ) <sub>exo</sub> @ExBox <sup>4+</sup>	(2CO) <sub>endo</sub> @ExBox <sup>4+</sup>		
	Without counter ions	With counter ions	Without counter ions	With counter ions	Without counter ions
ΔV <sub>elstat</sub>	-4.61(29.6%)	-2.74(18.4%)	-8.87(20.4%)	-9.79(29.1%)	-18.26(44.8%)
ΔE <sub>pauli</sub>	7.48	8.04	17.67	15.83	24.36
ΔE <sub>oi</sub>	-5.19(33.2%)	-5.91(39.8%)	-20.5(47.1%)	-11.64(34.6%)	-9.08(22.2%)
ΔE <sub>disp</sub>	-5.81(37.2%)	-6.19(41.7%)	-14.12(32.5%)	-12.21(36.3%)	-13.41(33%)
ΔE <sub>int</sub>	-8.14	-6.8	-25.82	-17.8	-16.39
					-14.36

**Table S4:** CDA data for gas adsorbed complexes

	d	b	d-b	r
(2H <sub>2</sub> ) <sub>endo</sub> @ExBox <sup>4+</sup>	-0.005	0.005	-0.001	-0.070
(3H <sub>2</sub> ) <sub>endo</sub> @ExBox <sup>4+</sup>	-0.002	0.005	-0.004	-0.094
(8H <sub>2</sub> ) <sub>exo</sub> @ExBox <sup>4+</sup>	-0.070	0.008	-0.078	-0.188
(8H <sub>2</sub> ) <sub>exo</sub> +(2H <sub>2</sub> ) <sub>endo</sub> @ExBox <sup>4+</sup>	-0.062	0.012	-0.074	-0.254
(8H <sub>2</sub> ) <sub>exo</sub> +(3H <sub>2</sub> ) <sub>endo</sub> @ExBox <sup>4+</sup>	-0.066	0.011	-0.076	-0.279
(12H <sub>2</sub> ) <sub>exo</sub> @ExBox <sup>4+</sup>	-0.069	0.062	-0.130	-0.244
(12H <sub>2</sub> ) <sub>exo</sub> +(2H <sub>2</sub> ) <sub>endo</sub> @ExBox <sup>4+</sup>	-0.063	0.074	-0.137	-0.316
(12H <sub>2</sub> ) <sub>exo</sub> +(3H <sub>2</sub> ) <sub>endo</sub> @ExBox <sup>4+</sup>	-0.069	0.073	-0.136	-0.339
(CO) <sub>endo</sub> @ExBox <sup>4+</sup>	0.005	0.013	-0.008	-0.025
(2CO) <sub>endo</sub> @ExBox <sup>4+</sup>	0.015	0.034	-0.019	-0.100
(3CO) <sub>endo</sub> @ExBox <sup>4+</sup>	0.019	0.043	-0.024	-0.127

**Table S5:** Geometrical parameters of CO adsorbed complexes

System	Shortest C-C bond length (Å)	Shortest C-O bond length (Å)	Shortest N-C bond length (Å)	Shortest N-O bond length (Å)	Distance between CO and centroid of pyridinium ring (Å)	Geometrical parameter for ExBox			
						C-C bond length (Å)	C-N bond length (Å)	<C-C-C-C (deg)	<N-C-C (deg)
(CO) <sub>endo</sub> @ExBox <sup>4+</sup>	3.493	3.374	3.469	3.387	3.151(3.307) <sup>a</sup>	1.481	1.495	145.558	110.530
(2CO) <sub>endo</sub> @ExBox <sup>4+</sup>	3.393	3.471	3.463	3.383	3.142(3.293) <sup>a</sup>	1.481	1.495	145.895	110.661
(3CO) <sub>endo</sub> @ExBox <sup>4+</sup>	3.397	3.476	3.443	3.388	3.145(3.269) <sup>a</sup>	1.481	1.495	144.083	110.588

a the values in parenthesis indicate the bonding distance of C center of CO from centroid of ring

**Table S6:** Bond critical point data (a.u.) calculated at wB97x-D/6-311G (d, p) level for CO bound ExBox<sup>4+</sup> complexes

System	bond type	$\rho$	$\nabla^2 \rho$	$G_{bcp}$	$V_{bcp}$	$H_{bcp}$
$(CO)_{endo}@ExBox^{4+}$	C-O	0.00421	0.01337	0.00254	-0.00174	0.00080
	C-O(pyridinium ring)	0.00429	0.01595	0.00298	-0.00198	0.00100
	C-C	0.00569	0.01604	0.00319	-0.00238	0.00082
	C-C(pyridinium ring)	0.00493	0.01560	0.00301	-0.00212	0.00089
$(2CO)_{endo}@ExBox^{4+}$	C-O	0.00417	0.01322	0.00251	-0.00172	0.00079
	C-O(pyridinium ring)	0.00425	0.01535	0.00288	-0.00192	0.00096
	N-C	0.00485	0.01597	0.00306	-0.00213	0.00093
	C-C	0.00566	0.01591	0.00317	-0.00236	0.00081
	C-C(pyridinium ring)	0.00489	0.01520	0.00294	-0.00208	0.00086
$(3CO)_{endo}@ExBox^{4+}$	C-O	0.00409	0.01306	0.00248	-0.00169	0.00079
	C-C	0.01646	0.14705	0.02757	-0.01837	0.00920
	third CO molecule					
	C-O	0.00398	0.01292	0.00244	-0.00165	0.00079
	C-C	0.00539	0.01545	0.00305	-0.00224	0.00081
	H-C	0.00041	0.00164	0.00028	-0.00015	0.00013
	Interaction between 2CO-3CO					
	C-C	0.00377	0.01241	0.00233	-0.00156	0.00077
	C-O	0.00185	0.00799	0.00141	-0.00083	0.00059

**Table S7:** Adsorption energy ( $E_{ads}$ , kcal/mol), reaction enthalpy ( $\Delta H$ , kcal/mol), HOMO-LUMO gap (HLG, eV) for gas adsorbed ExBox.4Cl calculated at wB97x-D/6-311G (d, p) basis set

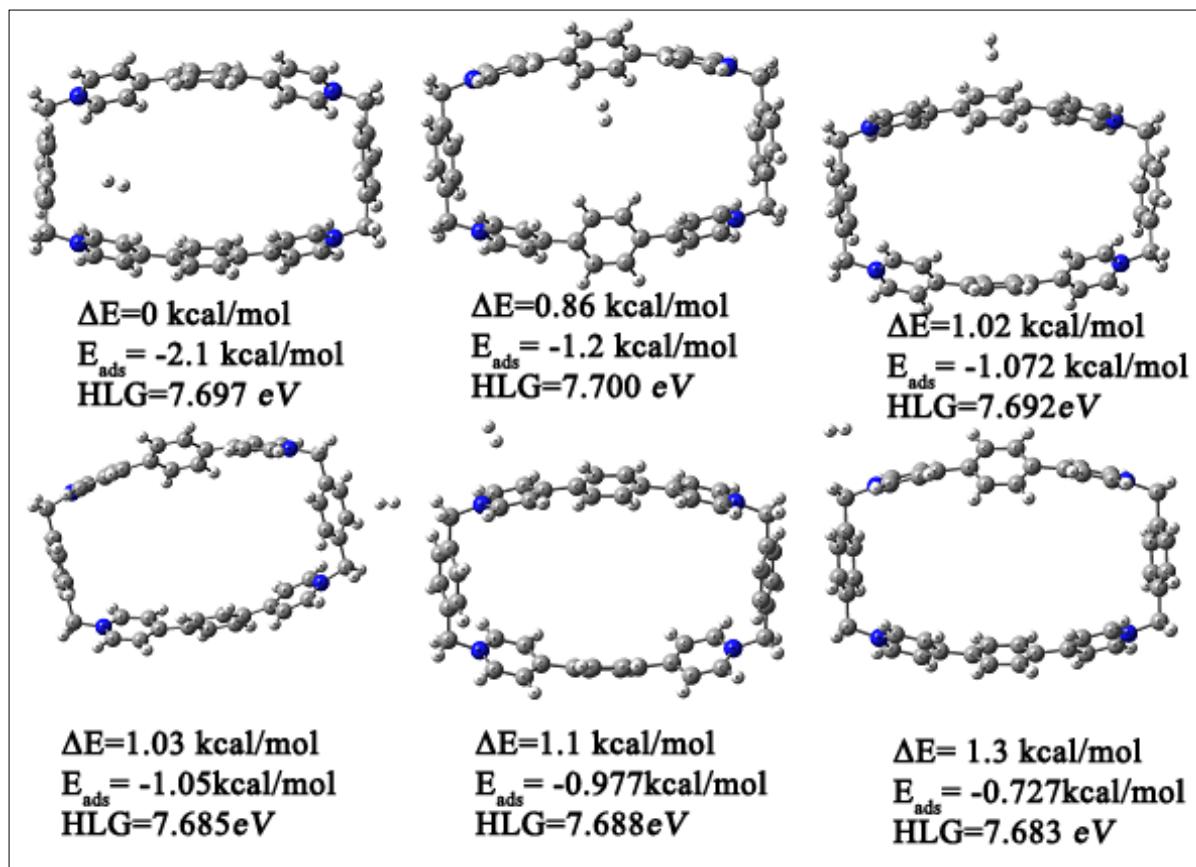
System	$E_{ads}$	$\Delta H$	HLG
$(2H_2)_{endo}@ExBox.4Cl$	-3.4	-2.2	5.477
$(3H_2)_{endo}@ExBox.4Cl$	-2.5	-1.3	5.419
$(8H_2)_{exo}@ExBox.4Cl$	-1.5	-0.3	5.492
$(12H_2)_{exo}@ExBox.4Cl$	-1.7	-0.4	5.551
$(2CO)_{endo}@ExBox.4Cl$	-5.5	-4.6	5.488

**Table S8:** Energy decomposition analysis (kcal/mol) of Li<sub>3</sub>model and its gas bound complexes at B3LYP-D/DZP level

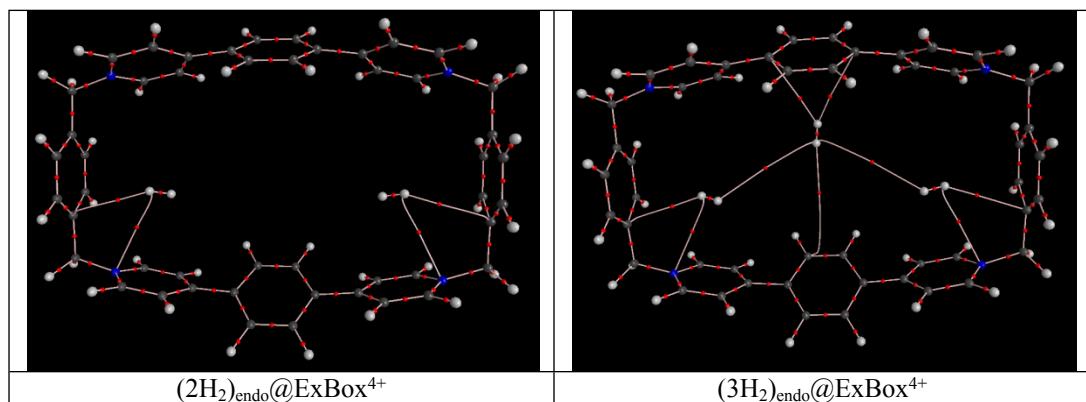
Energy terms	Li <sub>3</sub> model		$3H_2@Li_3$ model	$3CO@Li_3$ model
	Without counter ions	With counter ions		
$\Delta V_{elstat}$	-257.59(40.8%)	-293.99(47.4%)	-15.21(39.3%)	-22.05(37.7%)
$\Delta E_{pauli}$	455.06	491.85	8.07	15.09
$\Delta E_{oi}$	-343.03(54.4%)	-294.98(47.6%)	-14.11(36.5%)	-28.94(49.5%)
$\Delta E_{disp}$	-30.41(4.8%)	-30.25(~5%)	-9.35(24.1%)	-7.44(12.73%)
$\Delta E_{int}$	-175.97	-127.37	-15.21(39.3%)	-43.34

**Table S9:** CDA data for gas adsorbed  $\text{Li}_8\text{ExBox}^{4+}$  complexes

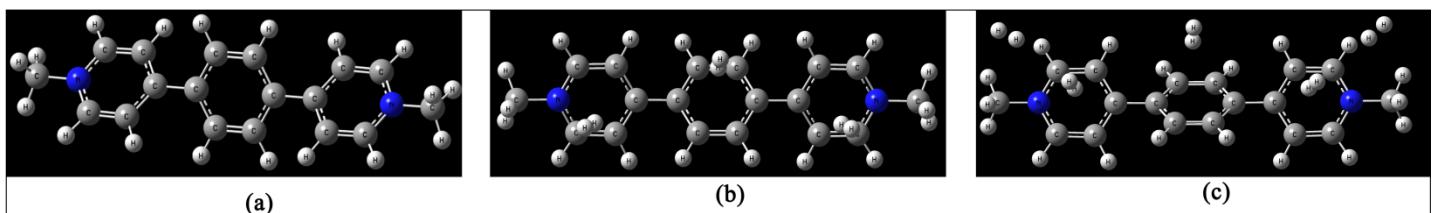
CDA analysis	8CO@ $\text{Li}_8\text{ExBox}^{4+}$			8H <sub>2</sub> @ $\text{Li}_8\text{ExBox}^{4+}$		
	ExBox to Li	ExBox to CO	Li-CO	ExBox to Li	ExBox to H <sub>2</sub>	Li-H <sub>2</sub>
d	1.29721	0.00350	-0.00927	1.44325	0.00267	-0.00719
b	0.00300	-0.00529	0.29693	0.01338	-0.00548	0.57090
d-b	1.29830	0.00880	-0.30620	1.42987	0.00815	-0.58628
r	-0.16241	-0.03195	-0.18178	-0.16292	-0.07131	-0.18322



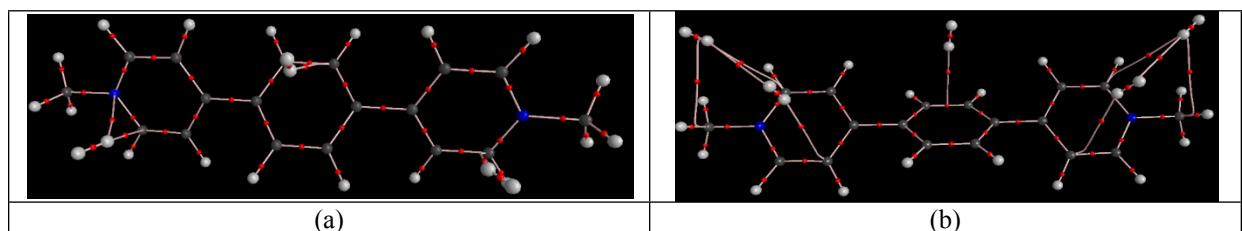
**Fig. S1:** Optimized geometries of different isomers of  $\text{H}_2@\text{ExBox}^{4+}$



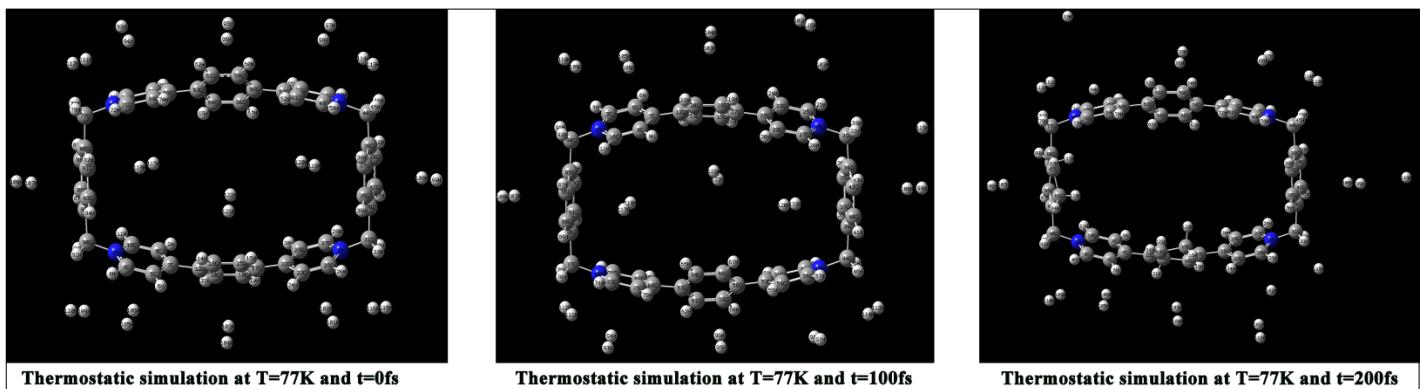
**Fig. S2:** Molecular graphs showing bond paths, bond critical points (BCP, red dots) of  $n\text{H}_2@\text{ExBox}^{4+}$



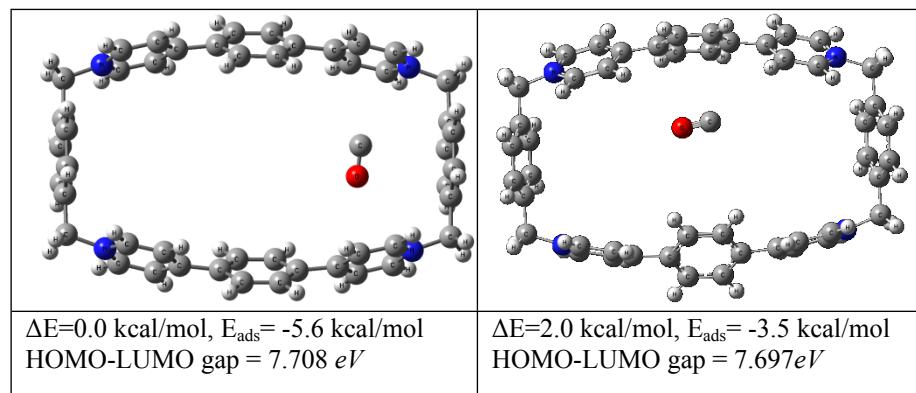
**Fig. S3:** Structures of model and  $n\text{H}_2@\text{model}$  ( $n=3,5$ )



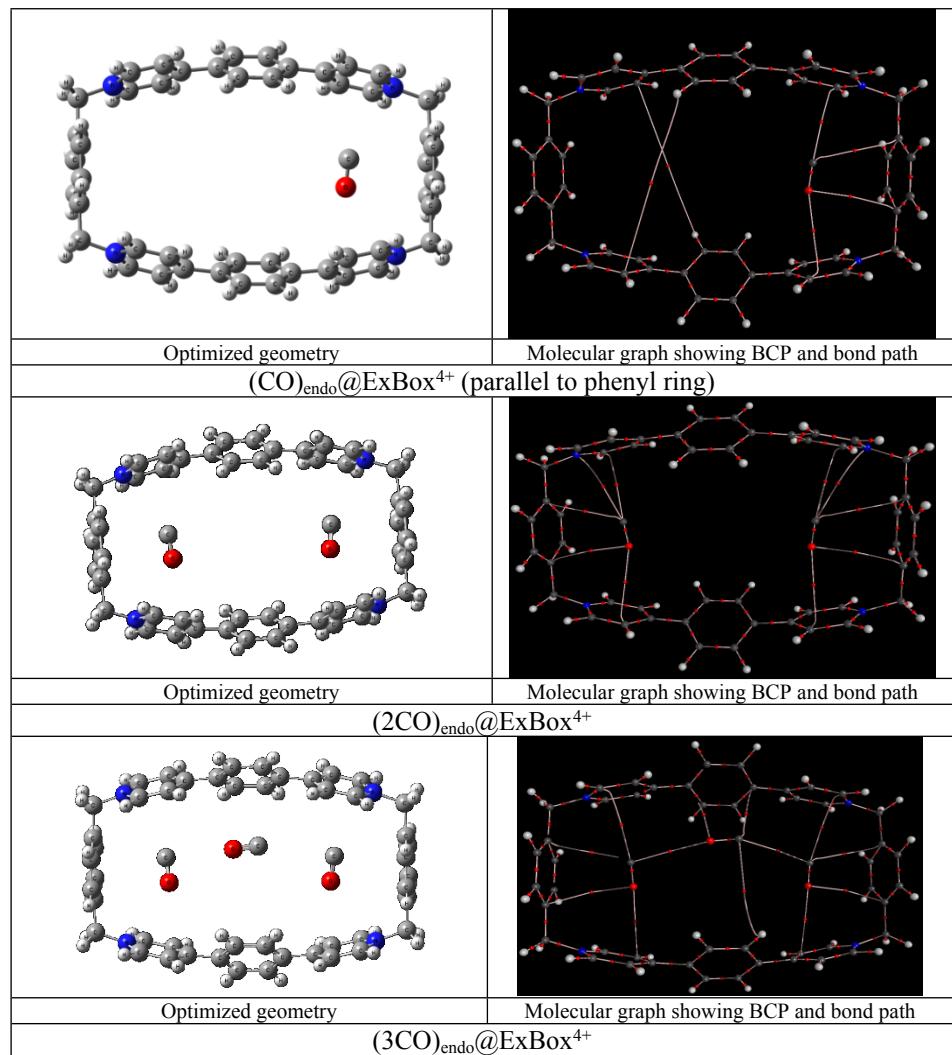
**Fig. S4:** Molecular graphs showing bond paths, bond critical points (BCP, red dots) of  $n\text{H}_2@\text{model}$  ( $n=3,5$ )



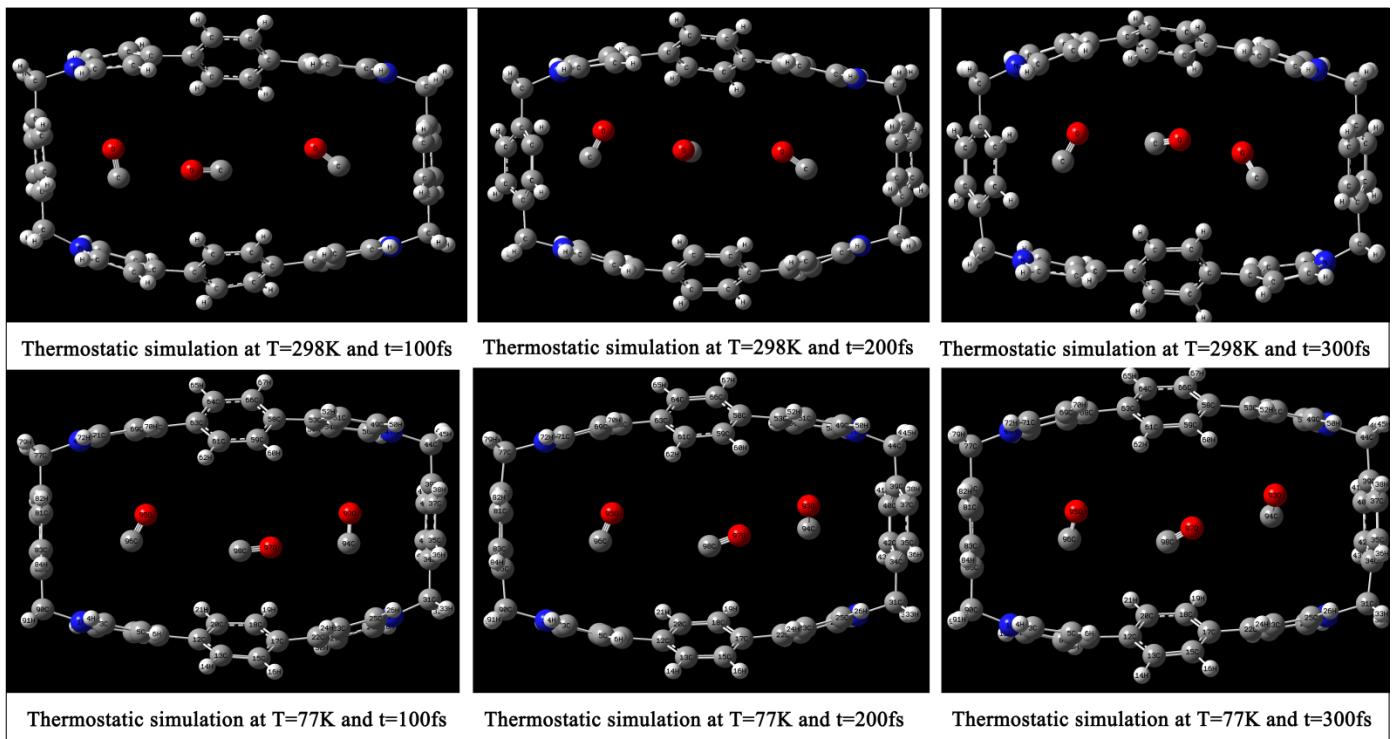
**Fig. S5:** Snap shots of  $(12\text{H}_2)_{\text{exo}} + (3\text{H}_2)_{\text{endo}}@\text{ExBox}^{4+}$  thermostatic simulation at 77K temperature at different time steps



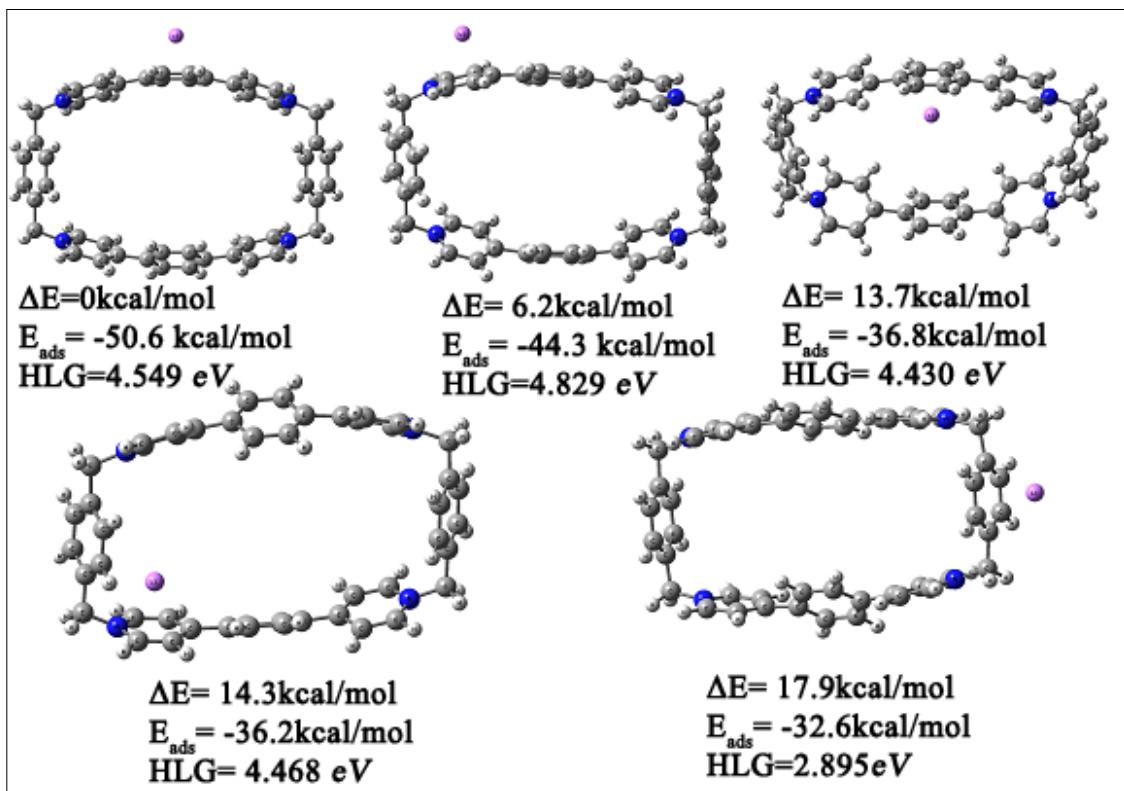
**Fig. S6:** Optimized geometries of different isomers of  $(\text{CO})_{\text{endo}}@\text{ExBox}^{4+}$



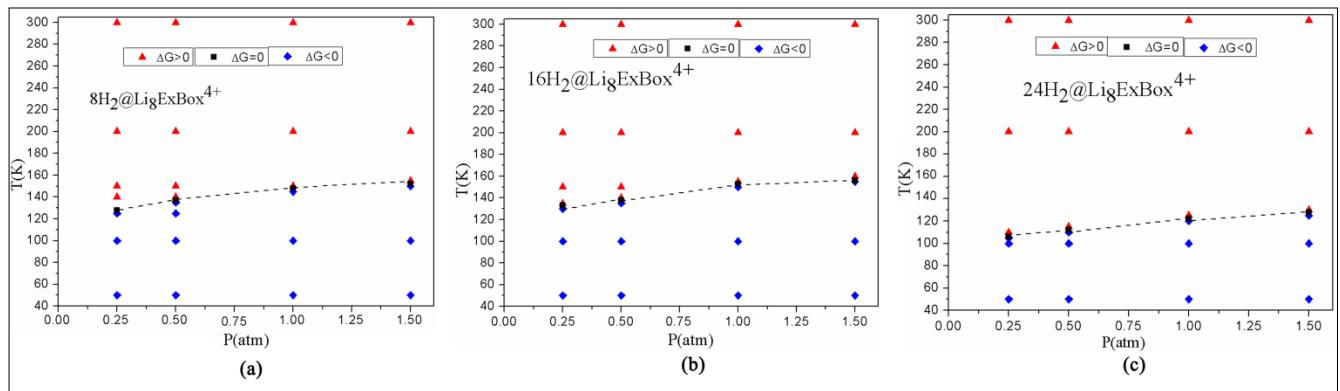
**Fig. S7:** Optimized geometries and molecular graphs showing bond paths, bond critical points (BCP, red dots) of  $(nCO)_{\text{endo}}@\text{ExBox}^{4+}$ ;  $n=1,2,3$



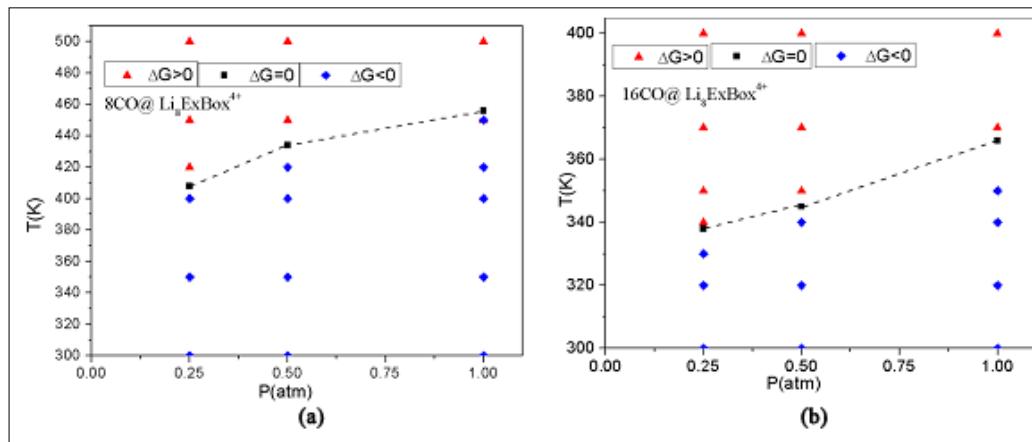
**Fig. S8:** Snap shots of  $(3\text{CO})_{\text{endo}}@\text{ExBox}^{4+}$  under different simulation conditions at different time steps



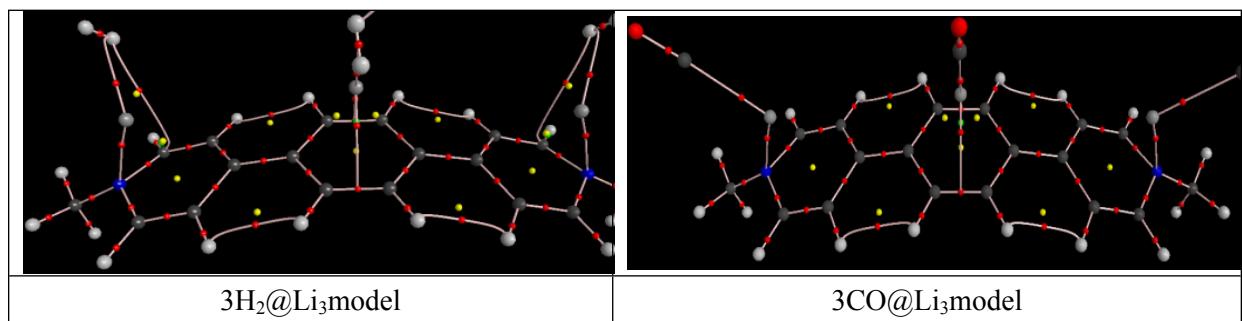
**Fig. S9:** Optimized geometries of different isomers of  $\text{Li}@\text{ExBox}^{4+}$



**Fig. S10:** T-P phase diagrams showing variation of  $\Delta G$  for (a)  $8\text{H}_2@\text{Li}_8\text{ExBox}^{4+}$  (b)  $16\text{H}_2@\text{Li}_8\text{ExBox}^{4+}$  and (c)  $24\text{H}_2@\text{Li}_8\text{ExBox}^{4+}$



**Fig. S11:** T-P phase diagrams showing variation of  $\Delta G$  for (a)  $8\text{CO}@ \text{Li}_8\text{ExBox}^{4+}$  and (b)  $16\text{CO}@ \text{Li}_8\text{ExBox}^{4+}$



**Fig. S12:** Molecular graphs showing bond paths, bond critical points (BCP, red dots)