

Supporting Information

Efficient Three-Components Coupling Reactions Catalyzed by Cu⁰-nanoparticles stabilized on Modified Montmorillonite Clay

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1. General information

Bentonite (procured from Gujarat, India) containing quartz, iron oxide etc. as impurities was purified by sedimentation¹ to collect the $< 2 \mu\text{m}$ Montmorillonite rich fraction. The basal spacing (d_{001}) of the air dried samples was about $12.5 \text{ \AA}$.² The specific surface area determined by N_2 adsorption was $101 \text{ m}^2/\text{g}$. The analytical oxide composition of the bentonite determined was SiO_2 : 49.42%; Al_2O_3 : 20.02%; Fe_2O_3 : 7.49%; MgO : 2.82%; CaO : 0.69%; Loss on ignition : 17.51%; others (Na_2O , K_2O and TiO_2): 2.05%.

The Montmorillonite (parent Mont.) was converted to the homoionic Na^+ -exchanged form by stirring in 2M NaCl solution for about 78 h, washed and dialysed against distilled water until the conductivity of the water approached that of distilled water. The cation exchange capacity (CEC) was 1.26 meq./g of clay (sample dried at 120°C).³

K_2PdCl_4 was purchased from Arora Matthey Ltd., India. Hydrazine hydrate and all the aryl halides, alkenes, alkynes, bases were purchased from M/S Sigma-Aldrich, USA. All the reagents were used as supplied.

IR spectra ($4000 - 400 \text{ cm}^{-1}$) were recorded on KBr discs in a Perkin-Elmer system 2000 FT-IR spectrophotometer. Powder XRD spectra were recorded on a Rigaku, Ultima IV X-ray diffractometer from $5 - 80^\circ 2\theta$ using $\text{CuK}\alpha$ source ($\lambda = 1.54 \text{ \AA}$). Specific surface area, pore volume, average pore diameter were measured with the Autosorb-1 (Quantachrome, USA). Specific surface area of the samples were measured by adsorption of nitrogen gas at 77 K and applying the Brunauer-Emmett-Teller (BET) calculation. Prior to adsorption, the samples were degassed at 250°C for 3 h. Pore size distributions were derived from desorption isotherms using Barrett-Joyner-Halenda (BJH) method. Scanning electron microscopy (SEM) images and energy dispersive X-ray spectroscopy (EDX) patterns were obtained with Leo 1430 vp operated at 3 and 10 KV. Prior to examination, the samples were coated with gold. Transmission electron microscopy (TEM) and high resolution transmission electron microscopy (HRTEM) images were recorded on a JEOL JEM-2011 electron microscope and the specimens were prepared by dispersing powdered samples in isopropyl alcohol, placing them on a carbon coated copper grid and allowing them to dry. The Pd contents were determined by Inductively Coupled Plasma Atomic Emission Spectroscopy (ICP-AES) using Perkin Elmer, OPTIMA 2000 instrument.

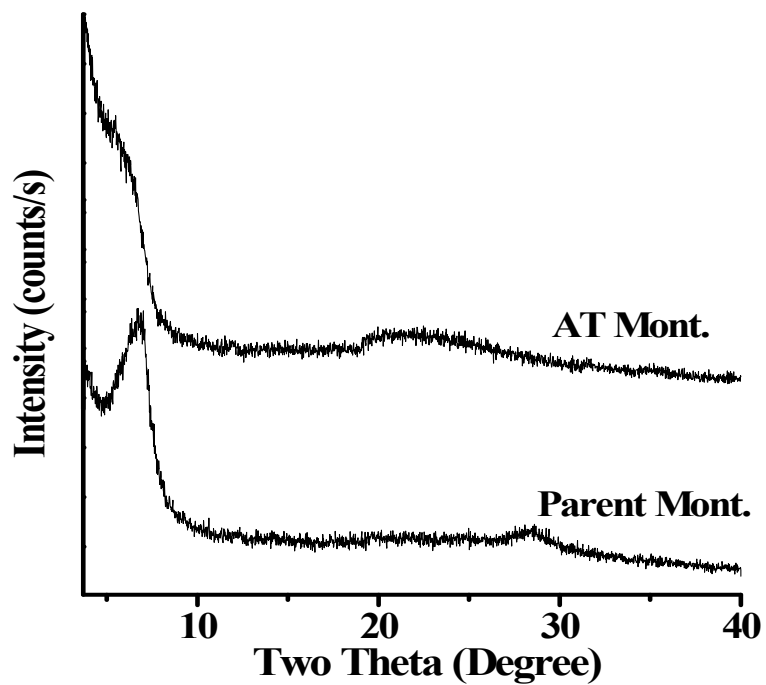


Fig. 1. Powder XRD pattern of different Montmorillonite clay.

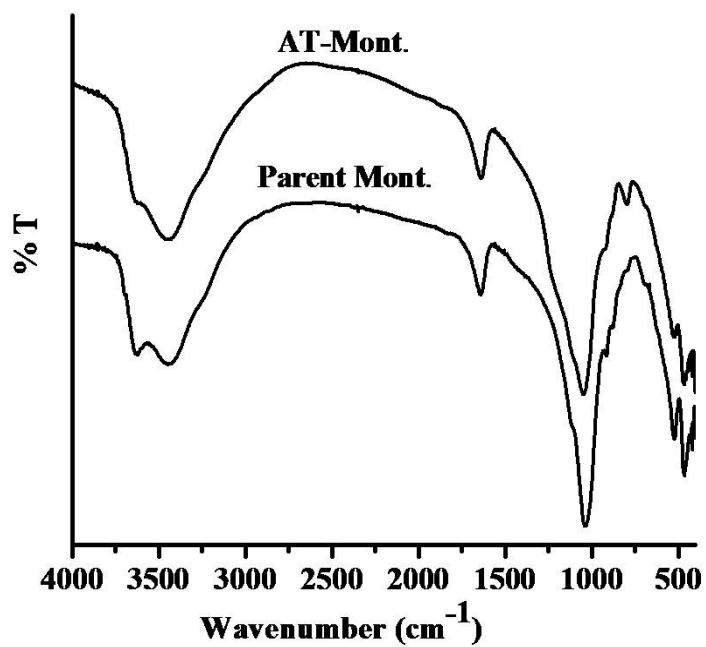


Fig. 2. FTIR spectra of different Montmorillonite clay.

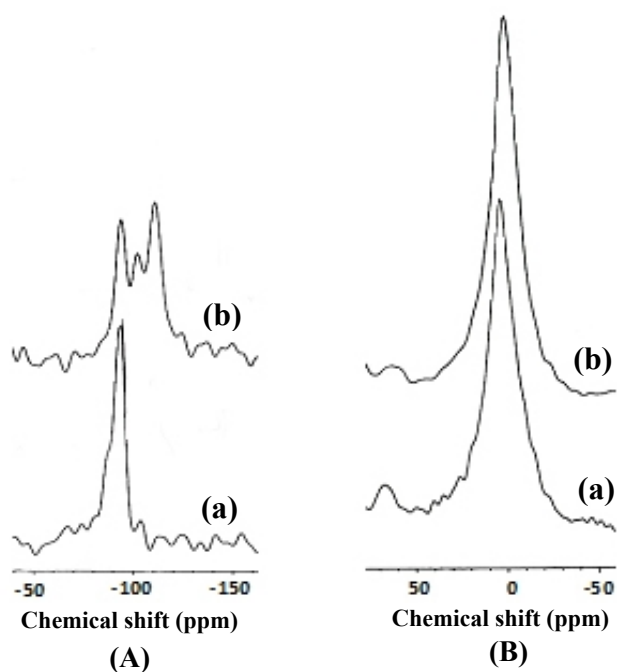


Fig. 3. (A) ^{29}Si and (B) ^{27}Al MAS-NMR spectra of (a) Parent Mont. (b) AT-Mont.

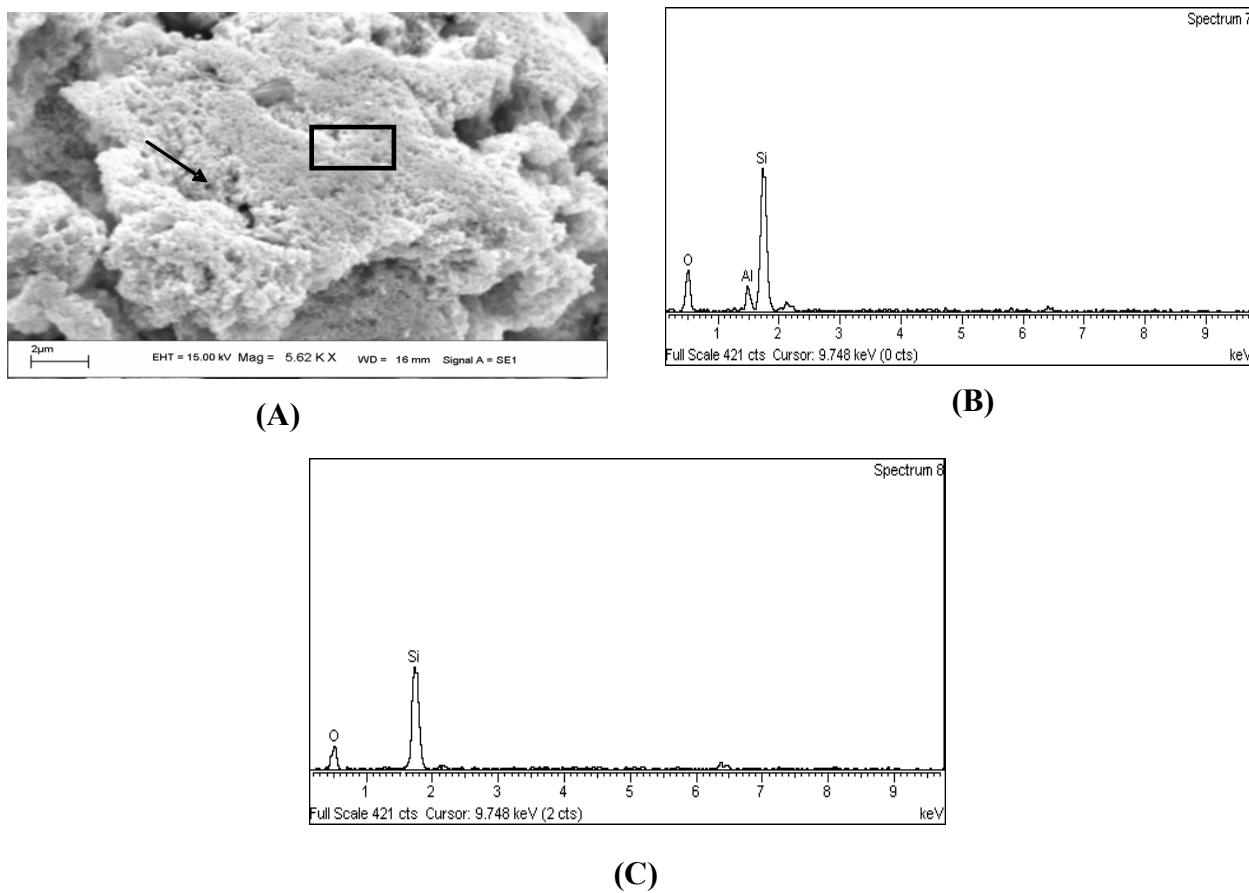


Fig. 4. (A) SEM image of the surface of AT-Mont.; (B) EDX analysis of the surface; (C) EDX spot analysis of pores (indicated by an arrow in (A)).

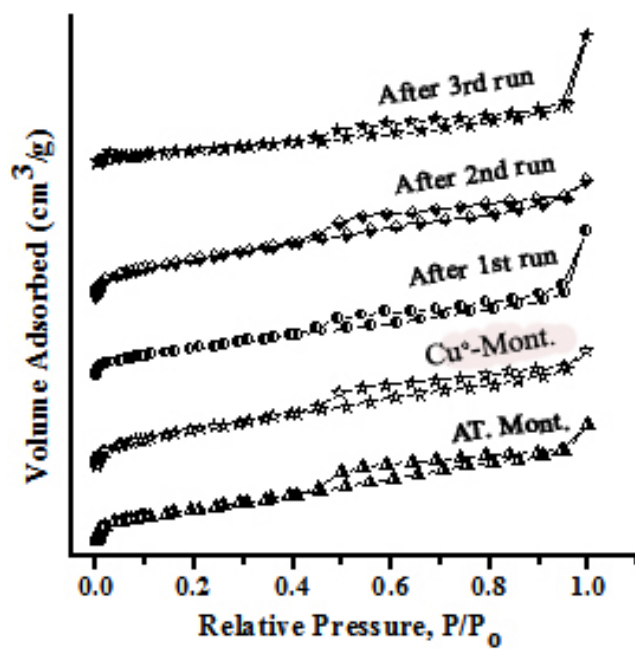


Fig. 5. N₂ adsorption / desorption isotherms of AT-Mont., Cu⁰-Mont. and recovered catalysts.

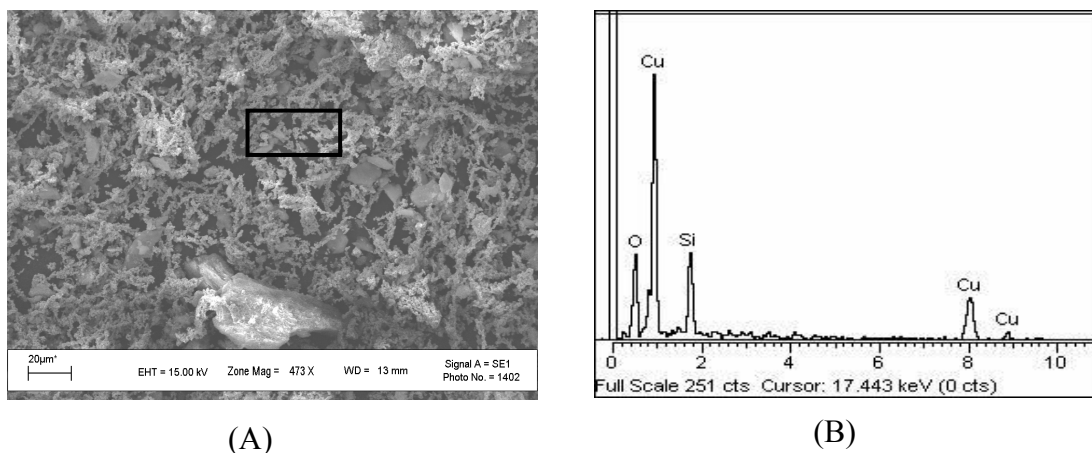


Fig. 6: (A) SEM image of the surface of AT-Mont. after supporting the Cu⁰-nanoparticles (B) EDX spot analysis of the pores.

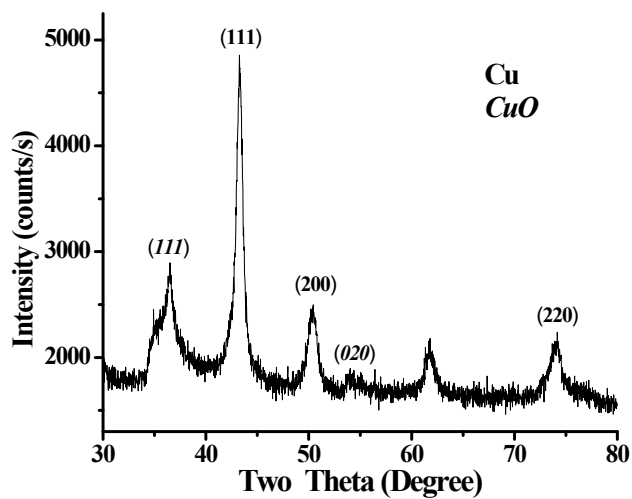
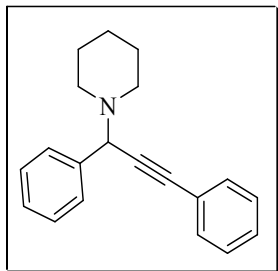


Fig. 7. Powder XRD pattern of recovered catalyst.

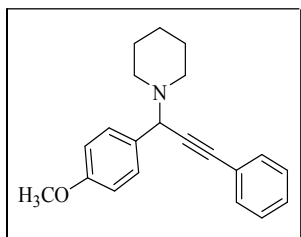
2. ^1H and ^{13}C NMR data of the synthesized Propargylamines:

(i) 1-(1,3-diphenylprop-2-ynyl)piperidine (**Entry 1**):



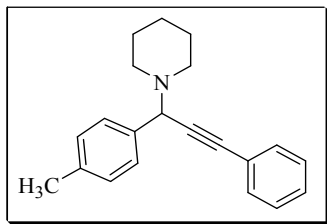
Pale yellow oily liquid; ^1H NMR (300 MHz, CDCl_3 , ppm): δ 1.38-1.65 (m, 6H), δ 2.54-2.57 (t, 4H), δ 4.79 (s, 1H), δ 7.24-7.35 (m, 6H), δ 7.50-7.53 (m, 2H), δ 7.61-7.69 (m, 2H); ^{13}C NMR (75 MHz, CDCl_3 , ppm): δ 24.66, 26.22, 50.74, 62.43, 86.10, 87.94, 123.38, 127.23, 128.13-128.34, 128.51-128.75, 131.87, 138.59; MS m/z 275 (M^+).

(ii) 1-(1-(4-Methoxyphenyl)-3-phenylprop-2-ynyl)-piperidine (**Entry 2**):



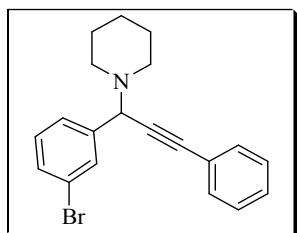
Dark yellowish oily liquid; ^1H NMR (300 MHz, CDCl_3 , ppm): δ 1.41-1.67 (m, 6H), δ 2.52-2.55 (t, 4H), δ 3.81 (s, 3H), δ 4.73 (s, 1H), δ 6.87-6.90 (m, 2H), δ 7.30-7.38 (m, 3H), δ 7.48-7.54 (m, 4H); ^{13}C NMR (75 MHz, CDCl_3 , ppm): δ 24.55, 26.21, 50.64, 55.27, 61.83, 86.44, 87.75, 113.46-113.68, 123.43, 128.10, 128.34, 129.75, 130.64, 131.85, 159.07; MS m/z 304 (M^+).

(iii) 1-(3-phenyl-1-p-tolylprop-2-ynyl)piperidine (**Entry 3**):



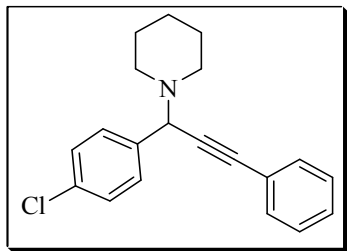
Yellow oily liquid; ^1H NMR (300 MHz, CDCl_3 , ppm): δ 1.34-1.60 (m, 6H), δ 2.34 (s, 3H), δ 2.41-2.62 (m, 4H), δ 4.75 (s, 1H), δ 7.05-7.14 (1H, m), δ 7.30-7.49 (m, 4H), δ 7.49-7.53 (m, 4H); ^{13}C NMR (75 MHz, CDCl_3 , ppm): δ 21.25, 24.61, 26.30, 50.78, 62.24, 81.75, 86.48, 121.89, 123.51, 128.14-129.93, 131.92, 132.60, 135.62, 137.19; MS m/z 289 (M^+).

(iv) 1-(1-(3-Bromophenyl)-3-phenylprop-2-ynyl)piperidine (**Entry 4**):



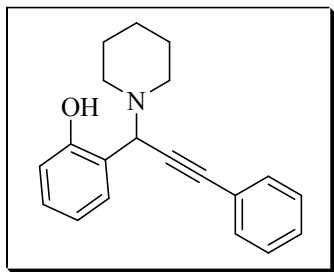
Yellow oily liquid; ^1H NMR (300 MHz, CDCl_3 , ppm): δ 1.33-1.66 (m, 6H), δ 2.51-2.52 (d, 4H), δ 2.41-2.62 (m, 4H), δ 4.74 (s, 1H), δ 7.16-7.40 (5H, m), δ 7.49-7.58 (m, 3H), δ 7.79 (s, 1H); ^{13}C NMR (75 MHz, CDCl_3 , ppm): δ 24.55, 26.31, 50.81, 61.94, 81.80, 85.18, 88.67, 121.91-123.17, 137.22-, 132.62, 141.37; MS m/z 355 (M^+).

(v) 1-(1-(4-Chlorophenyl)-3-phenylprop-2-ynyl)piperidine (**Entry 5**):



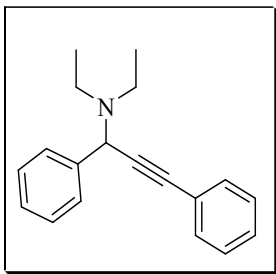
Yellow oily liquid; ^1H NMR (300 MHz, CDCl_3 , ppm): δ 1.31-1.51 (m, 6H), δ 2.51-2.54 (t, 4H), δ 4.76 (s, 1H), δ 7.30-7.34 (m, 6H), δ 7.49-7.58 (m, 4H); ^{13}C NMR (75 MHz, CDCl_3 , ppm): δ 24.53, 26.27, 50.74, 61.79, 74.29, 81.79, 85.45, 123.20, 128.33-128.56, 129.47, 129.97, 131.94, 132.60, 134.81; MS m/z 309 (M^+).

(vi) 2-(3-(4-phenyl-1(piperidin-1-yl)prop-2-ynyl)phenol (**Entry 6**):



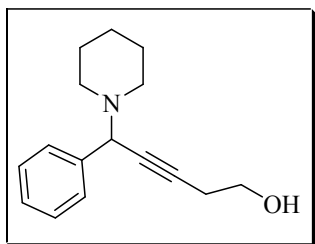
Yellow oily liquid; ^1H NMR (300 MHz, CDCl_3 , ppm): δ 1.21-1.67 (m, 6H), δ 2.68-2.75 (m, 4H), δ 5.09 (s, 1H), δ 6.83-6.85 (2H, m), δ 7.19-7.25 (m, 1H), δ 7.33-7.37 (m, 3H), δ 7.51-7.56 (m, 3H); ^{13}C NMR (75 MHz, CDCl_3 , ppm): δ 23.07, 24.06, 38.40, 61.13, 82.42, 89.90, 116.42, 117.64, 121.35, 122.66, 128.48-128.64, 129.27-129.43, 131.94, 132.54, 151.72; MS m/z 391 (M^+).

(vii) (1,3-Diphenyl-prop-2-ynyl)diethylamine (**Entry 7**):



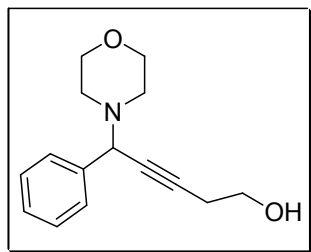
Pale yellow oily liquid; ^1H NMR (300 MHz, CDCl_3 , ppm): δ 0.97-1.04 (m, 6H), δ 2.40-2.62 (m, 4H), δ 4.97 (s, 1H), δ 7.24-7.27 (m, 5H), δ 7.41-7.44 (m, 3H), δ 7.59-7.61 (m, 2H); ^{13}C NMR (75 MHz, CDCl_3 , ppm): δ 13.64, 44.65, 57.09, 74.03, 81.63, 86.17, 121.86, 123.46, 127.33-129.26, 131.85, 132.56, 139.89; MS m/z 263 (M^+).

(viii) 5-phenyl-5-(piperidin-1-yl)pent-3-yn-1-ol (**Entry 8**):



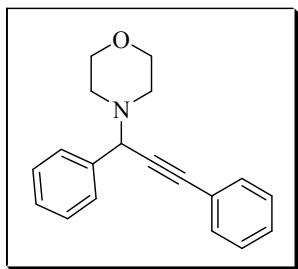
Yellow oily liquid; ^1H NMR (300 MHz, CDCl_3 , ppm): δ 1.38-1.42 (m, 4H), δ 2.08-2.17 (m, 4H), δ 2.44-2.61 (m, 5H), δ 3.72-3.79 (m, 2H), δ 4.53 (s, 1H), δ 7.26-7.61 (5H, m); ^{13}C NMR (75 MHz, CDCl_3 , ppm): δ 23.30, 24.57, 25.85, 50.68, 60.63, 61.22, 62.08, 66.70, 74.72, 78.57, 84.36, 126.76, 127.62, 128.08-128.67, 137.91; MS m/z 243 (M^+).

(ix) 5-Morpholino-5-phenylpent-3-yn-1-ol (**Entry 9**):



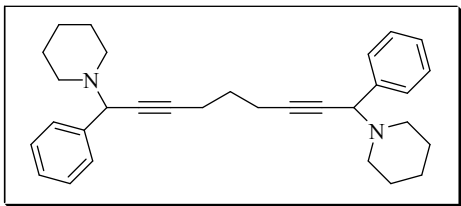
Pale yellow oily liquid; ^1H NMR (300 MHz, CDCl_3 , ppm): δ 2.53-2.62 (m, 4H), δ 3.40-3.80 (m, 8H), δ 4.53 (s, 1H), δ 5.30 (s, 1H), δ 7.26-7.61 (m, 4H), δ 8.06 (s, 1H); ^{13}C NMR (75 MHz, CDCl_3 , ppm): δ 23.22, 40.62, 45.81, 49.53, 67.08-67.20, 89.09, 127.07, 128.81, 129.74, 134.45, 136.40; MS m/z 245 (M^+).

(x) 4-(1,3-diphenylprop-2-ynyl)morpholine (**Entry 10**):



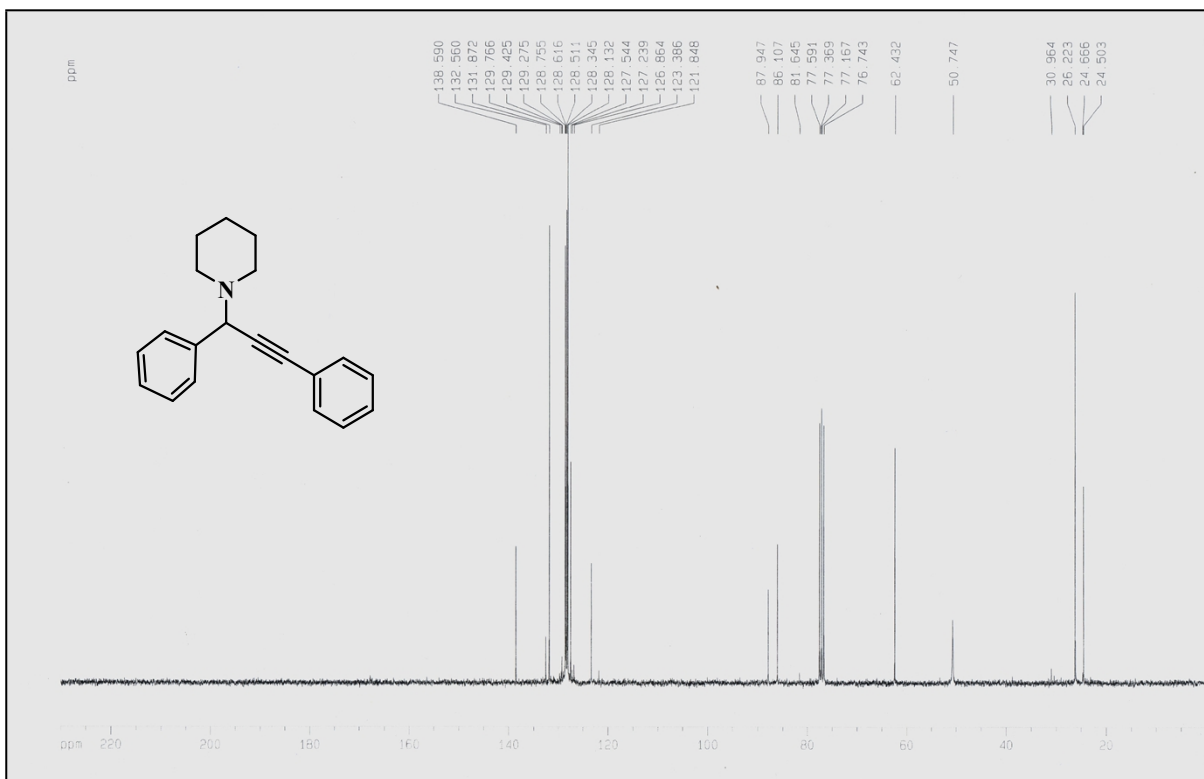
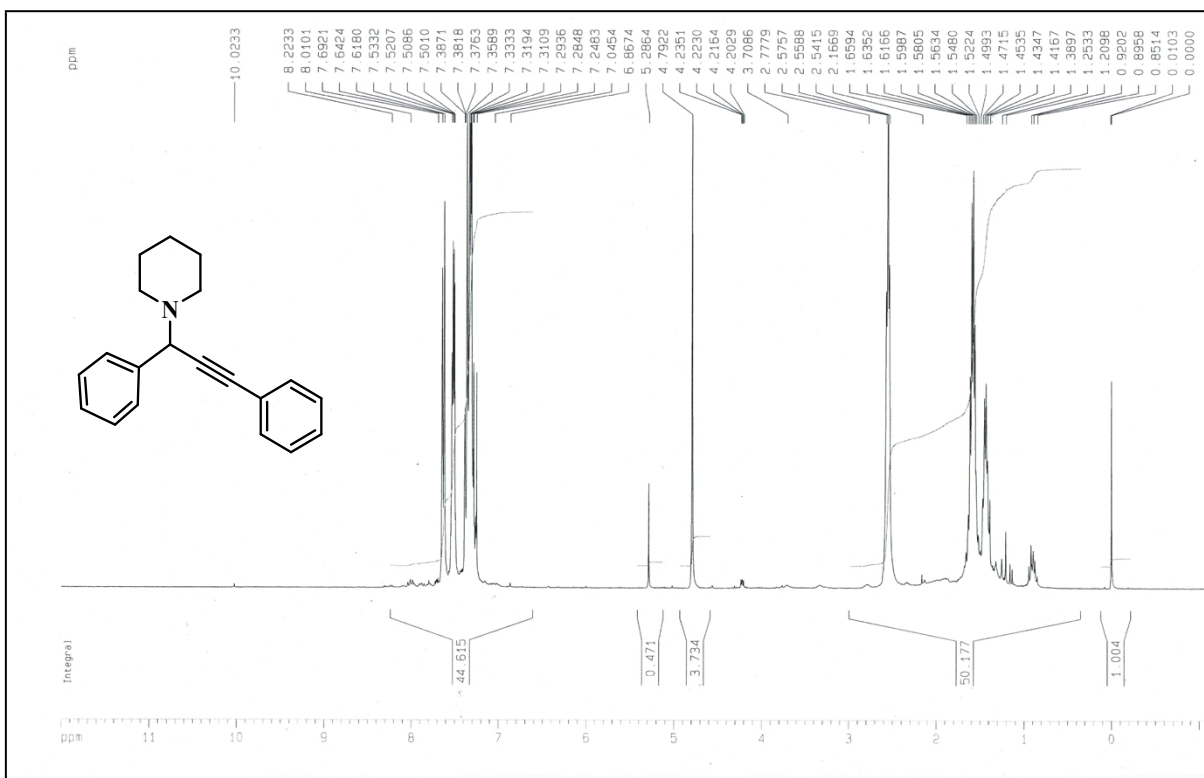
Pale yellow oily liquid; ^1H NMR (300 MHz, CDCl_3 , ppm): δ 2.62-2.65 (t, 4H), δ 3.72-3.75 (t, 4H), δ 4.79 (s, 1H), δ 7.24-7.39 (m, 6H), δ 7.49-7.53 (m, 2H), δ 7.62-7.64 (m, 2H); ^{13}C NMR (75 MHz, CDCl_3 , ppm): δ 49.89, 62.06, 67.16, 85.02, 88.54, 122.97, 127.82, 128.34, 128.62, 131.83, 137.77; MS m/z 277 (M^+).

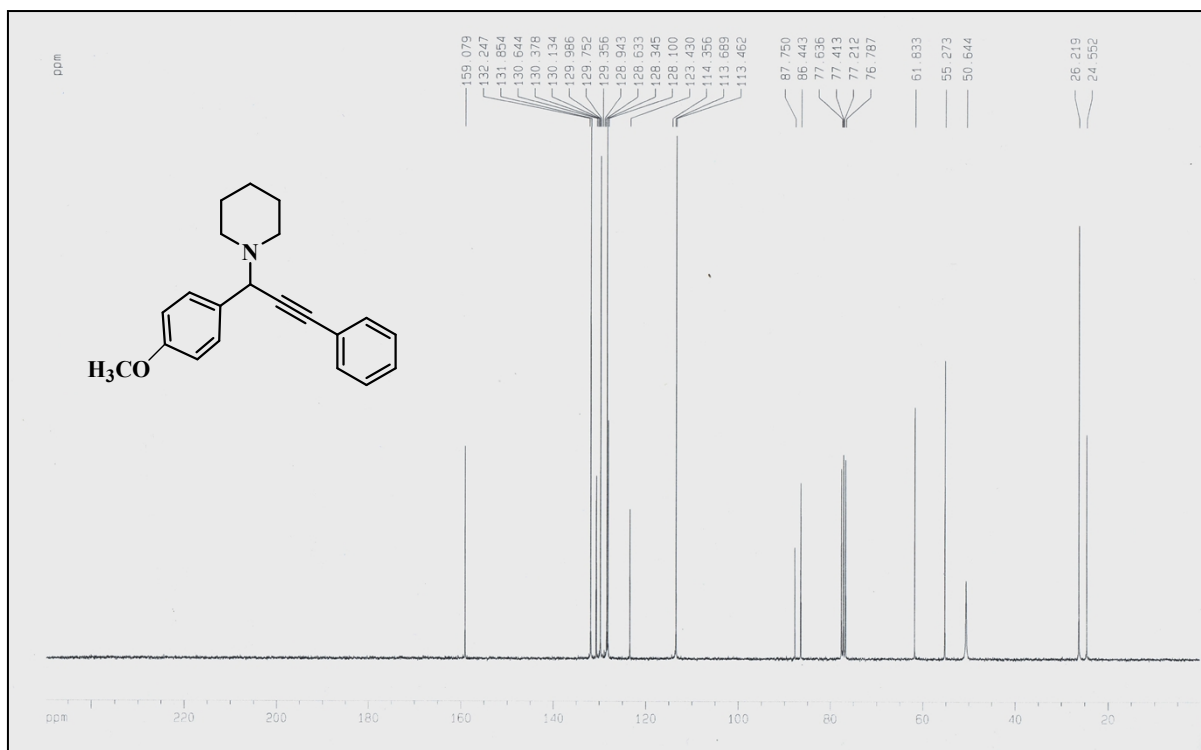
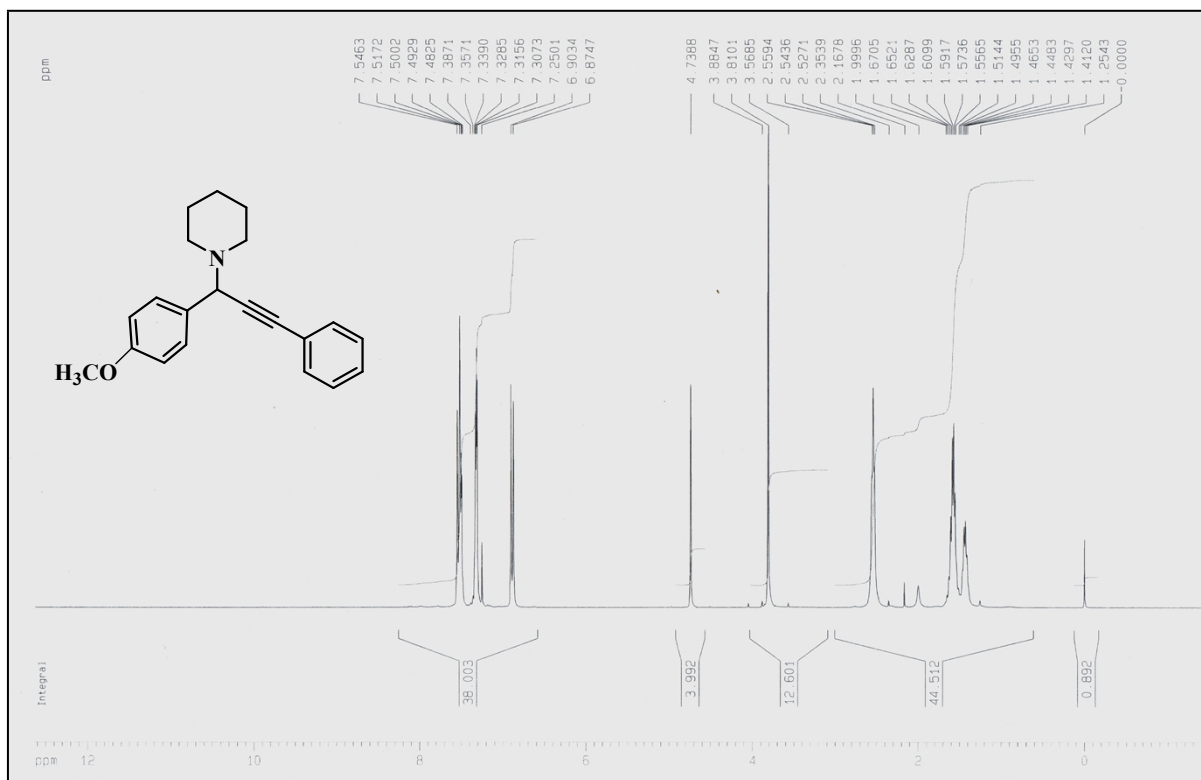
(xi) 1-(1,9-diphenyl-9-(piperidin-1-yl)nona-2,7-diynyl)piperidine (**Entry 11**):

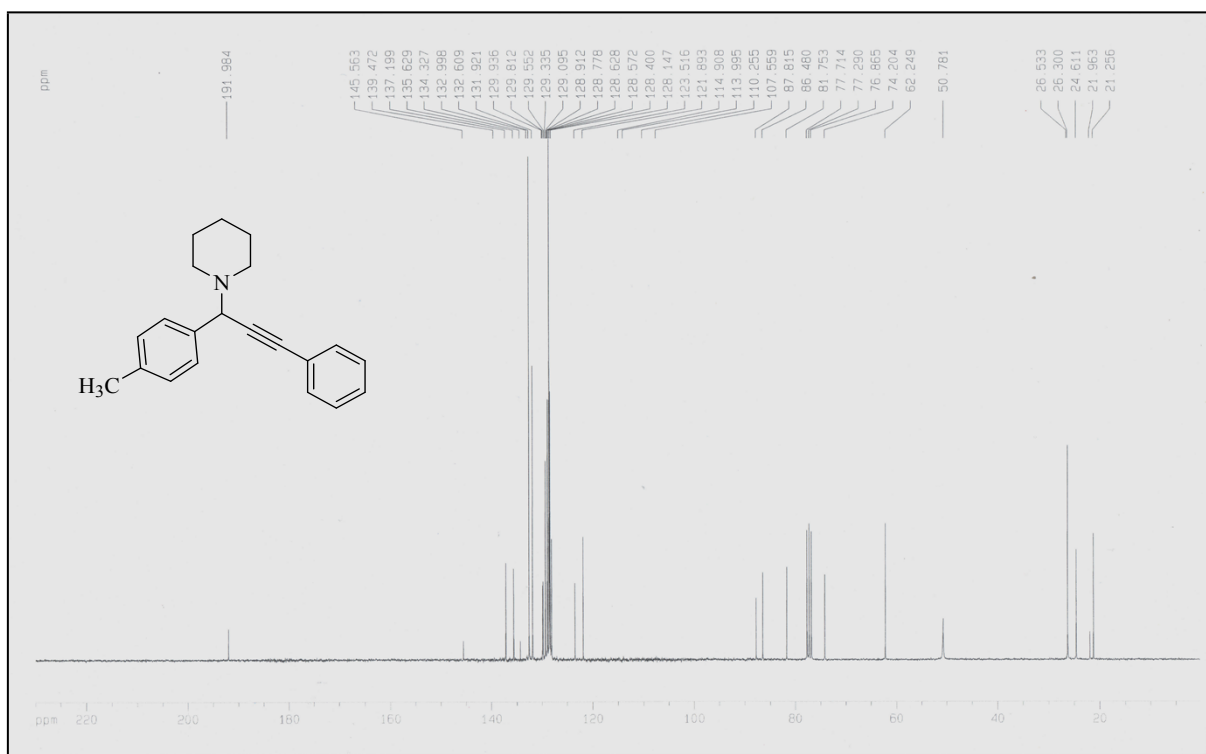
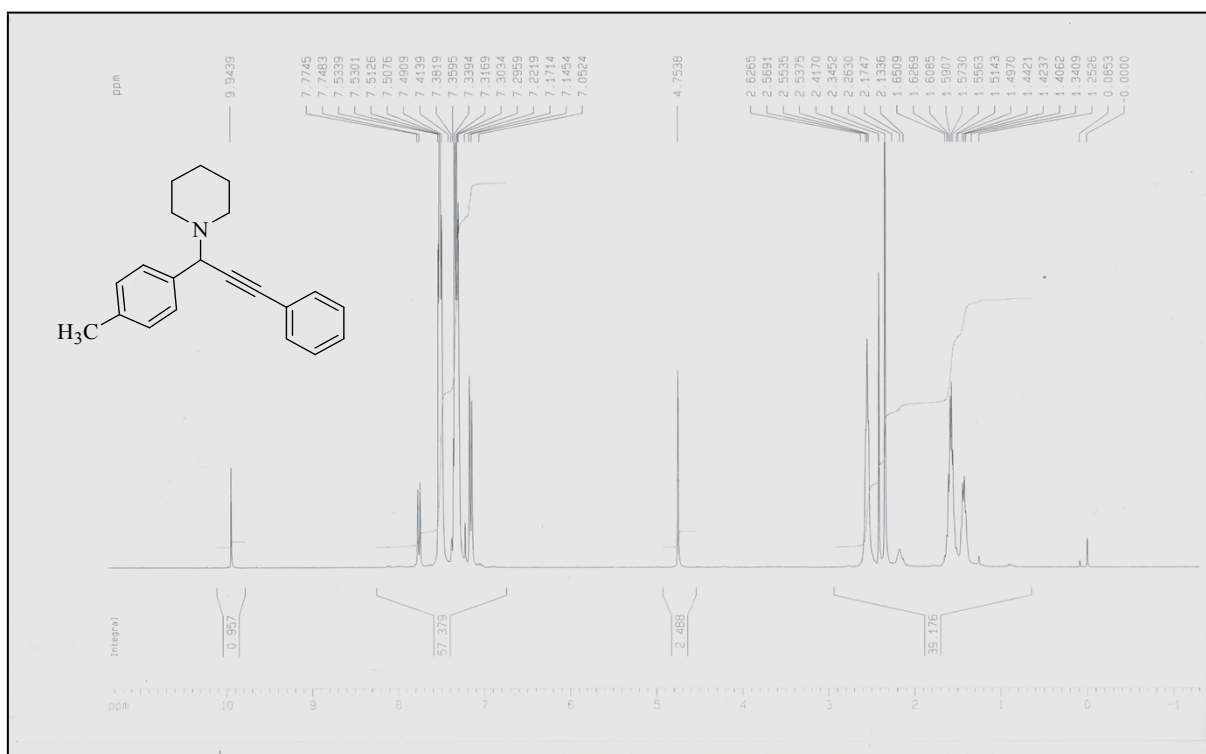


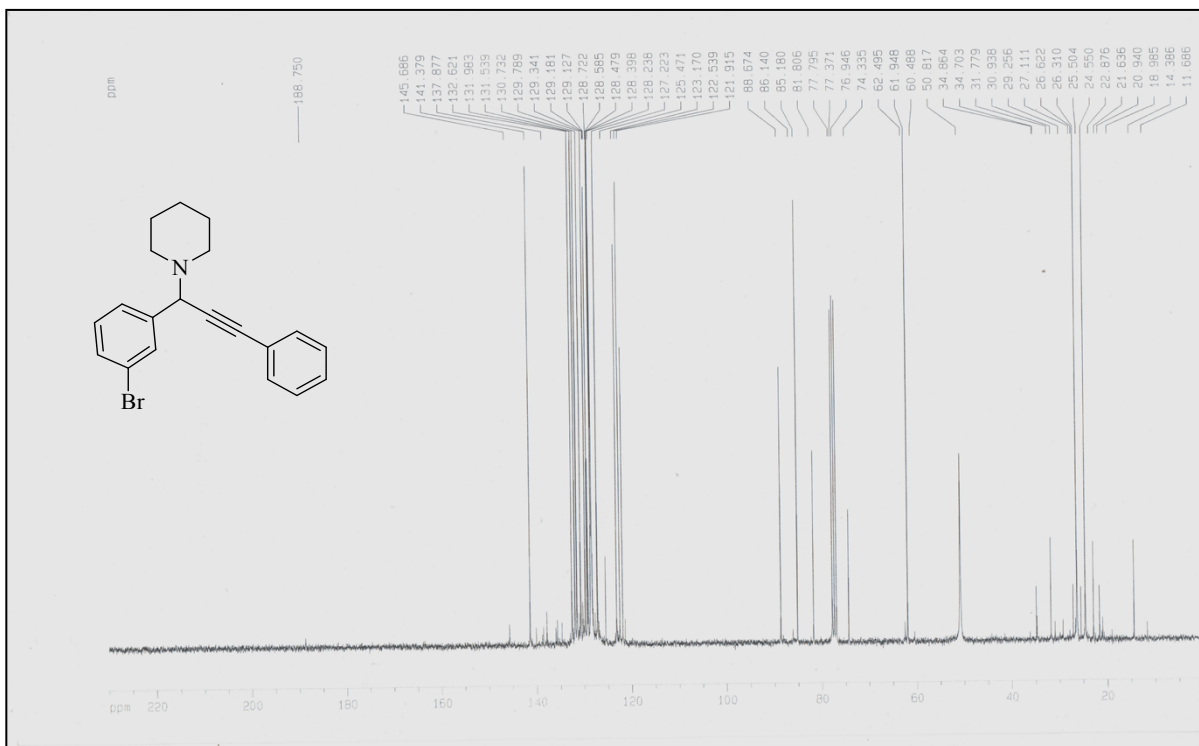
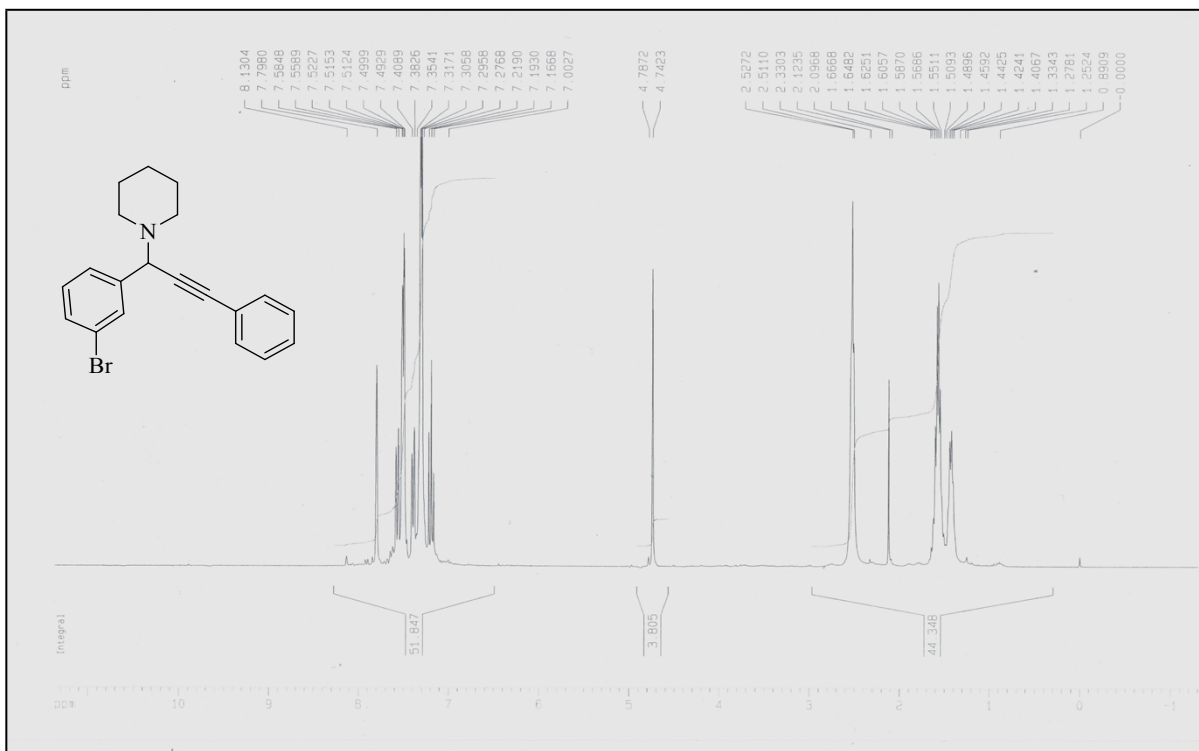
Yellow oily liquid; ^1H NMR (300 MHz, CDCl_3 , ppm): δ 1.25-1.59 (m, 10H), δ 1.77-1.82 (m, 4H), δ 2.36-2.47 (m, 12H), δ 4.54 (s, 2H), δ 7.23-7.42 (6H, m), δ 7.53-7.55 (m, 4H). ^{13}C NMR (75 MHz, CDCl_3 , ppm): δ 17.56, 17.68, 24.42, 26.06, 27.92, 50.54, 61.95, 68.84, 83.62, 86.60, 127.34, 127.96, 128.54, 138.78; MS m/z 438 (M^+).

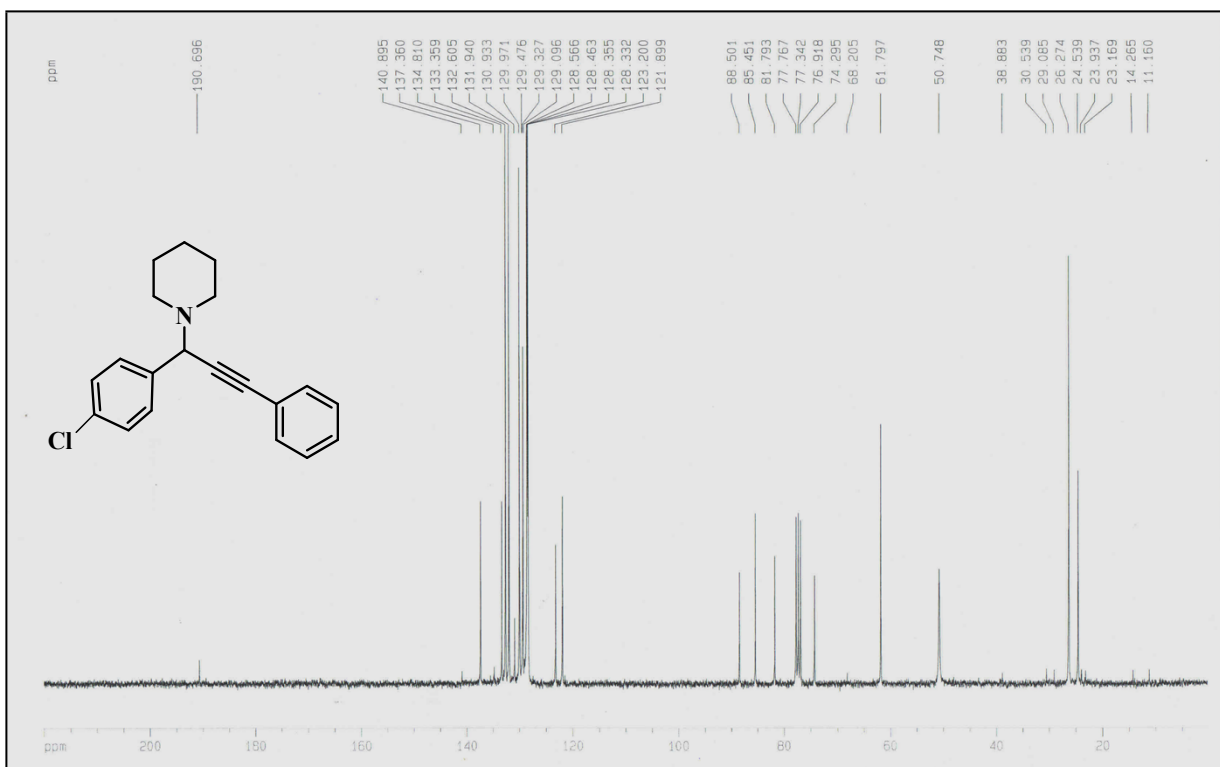
3. ^1H NMR and ^{13}C NMR spectra of compounds

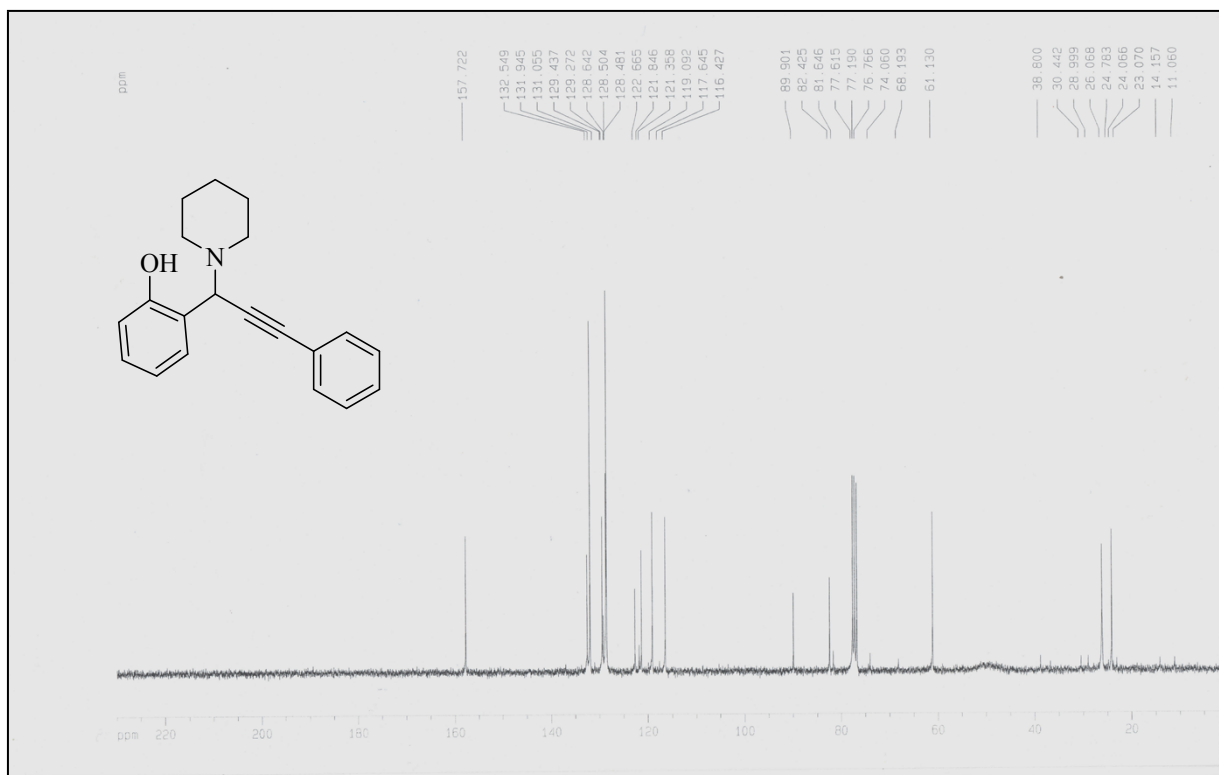
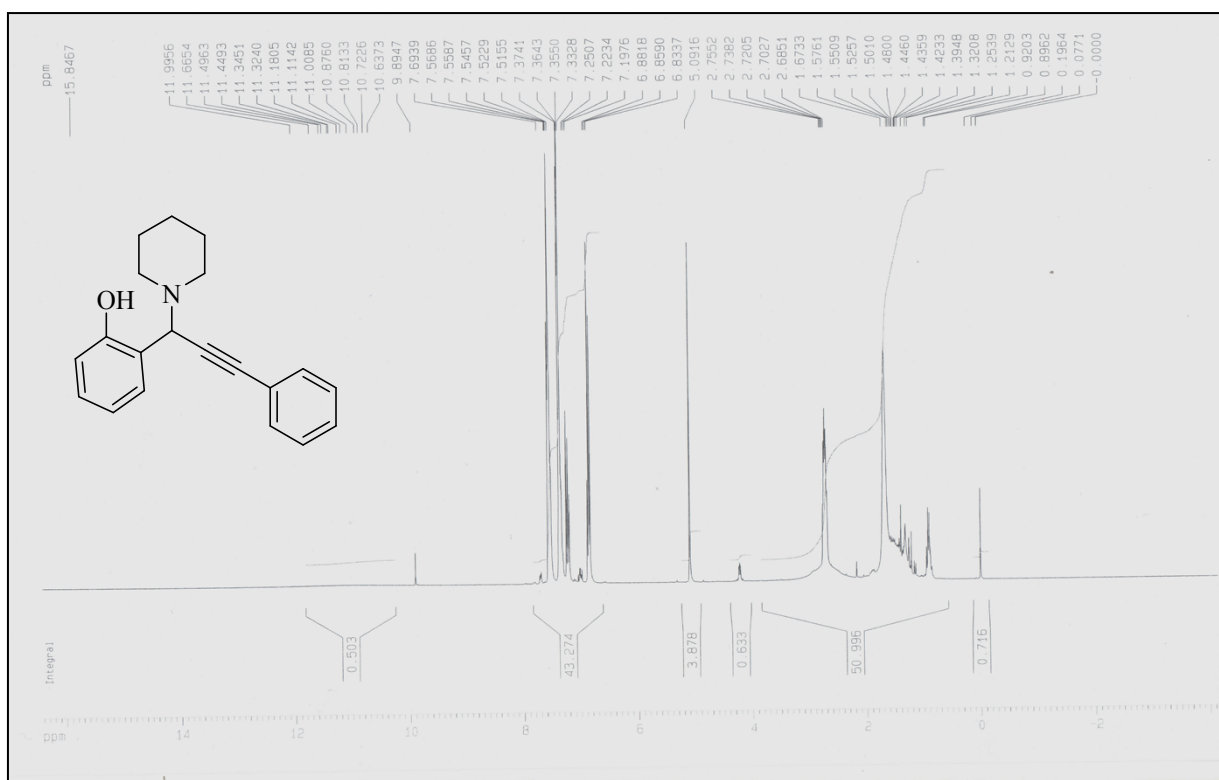


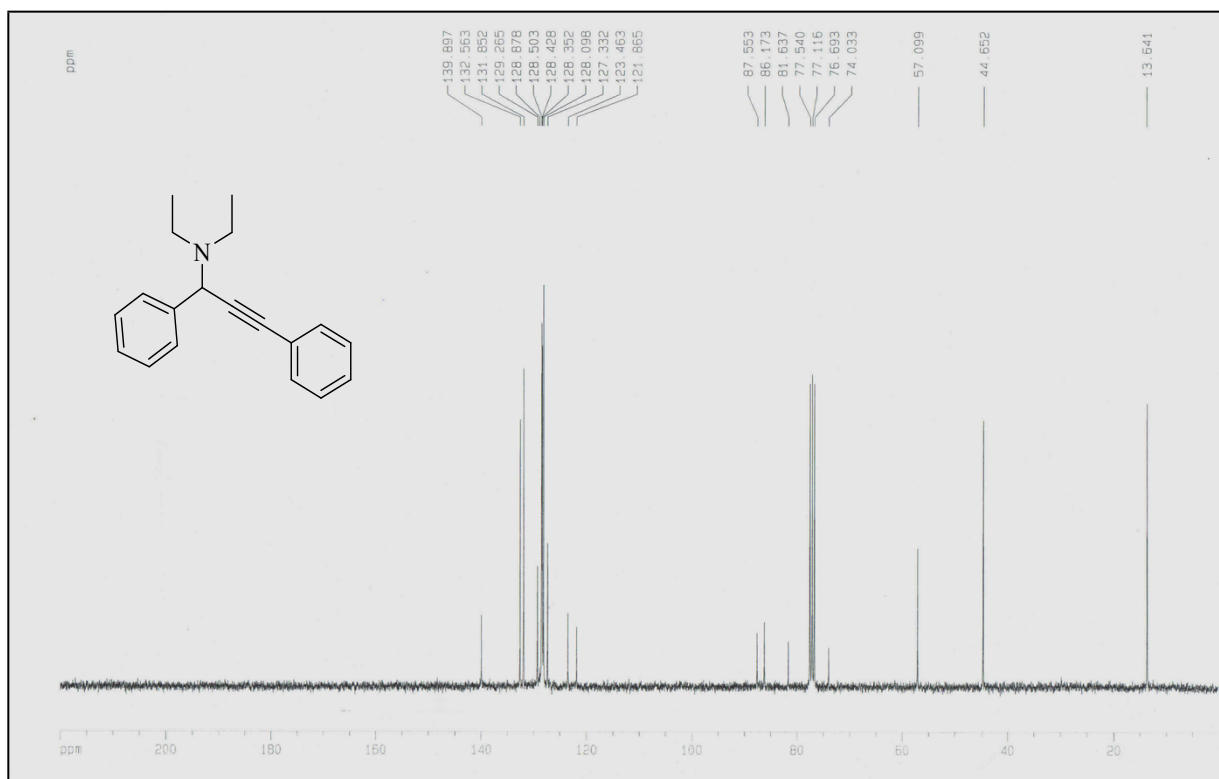
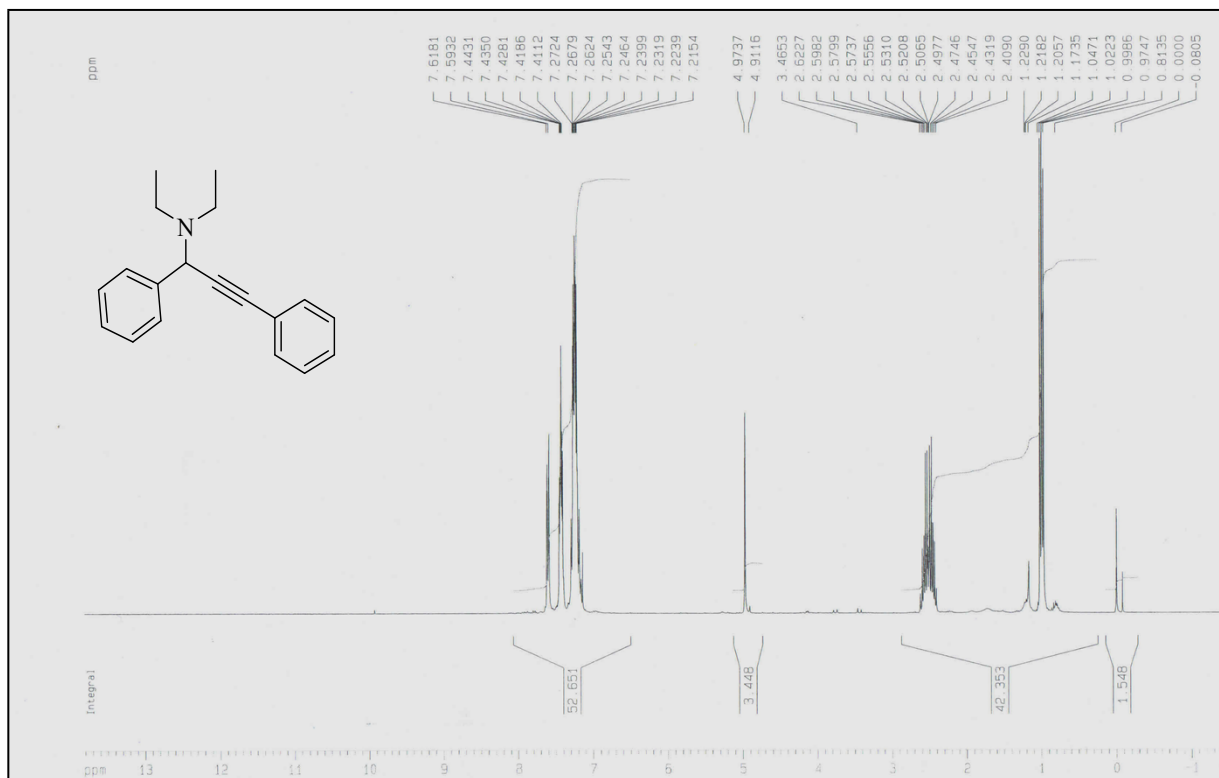


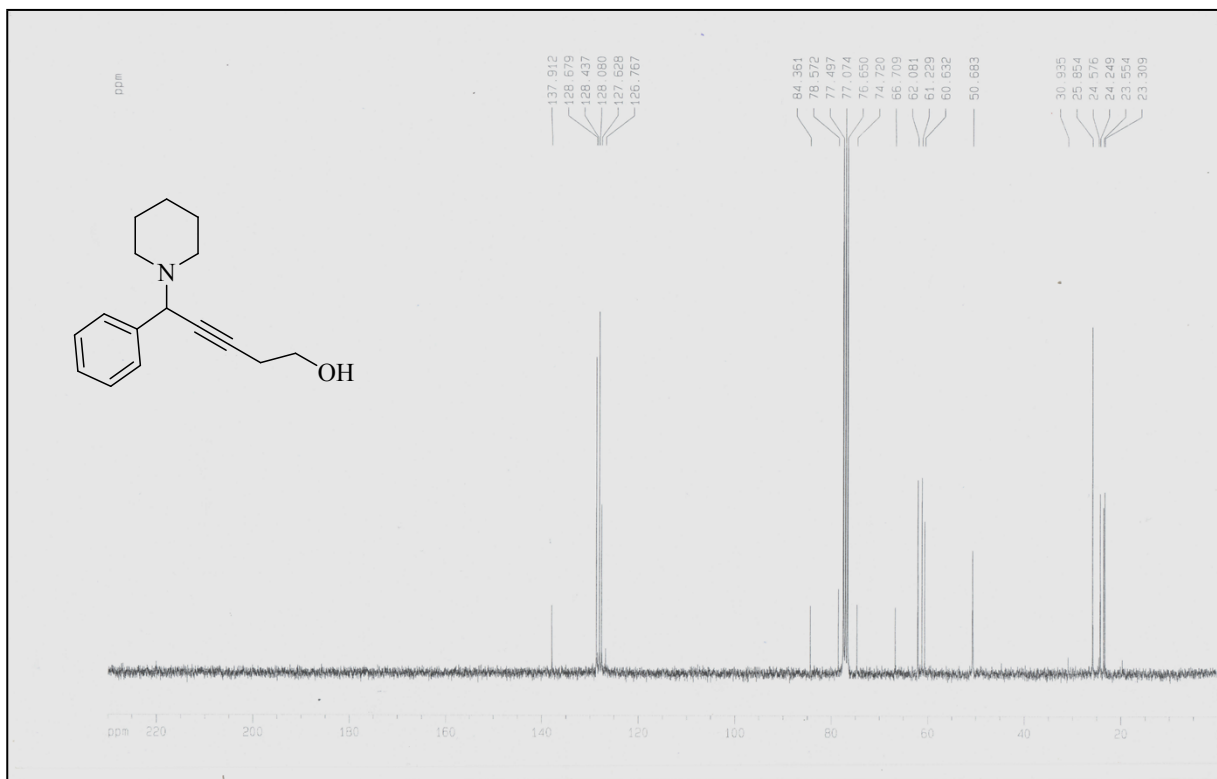
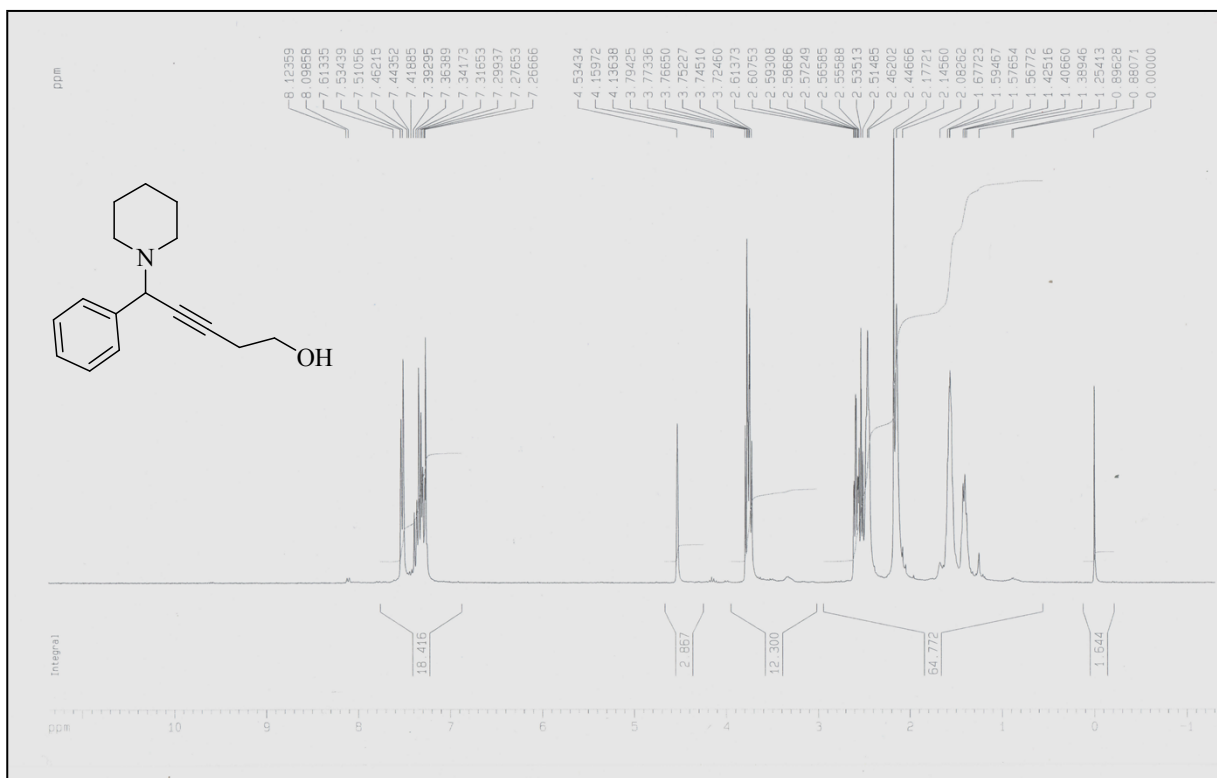


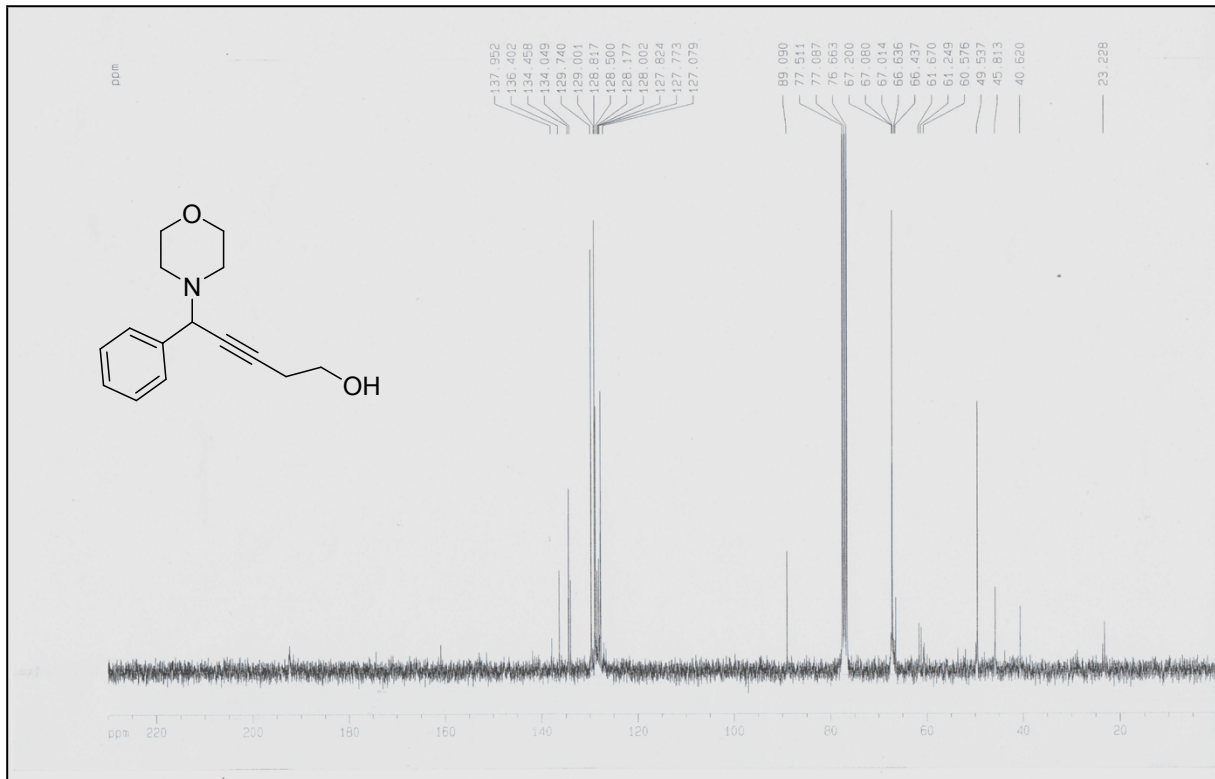
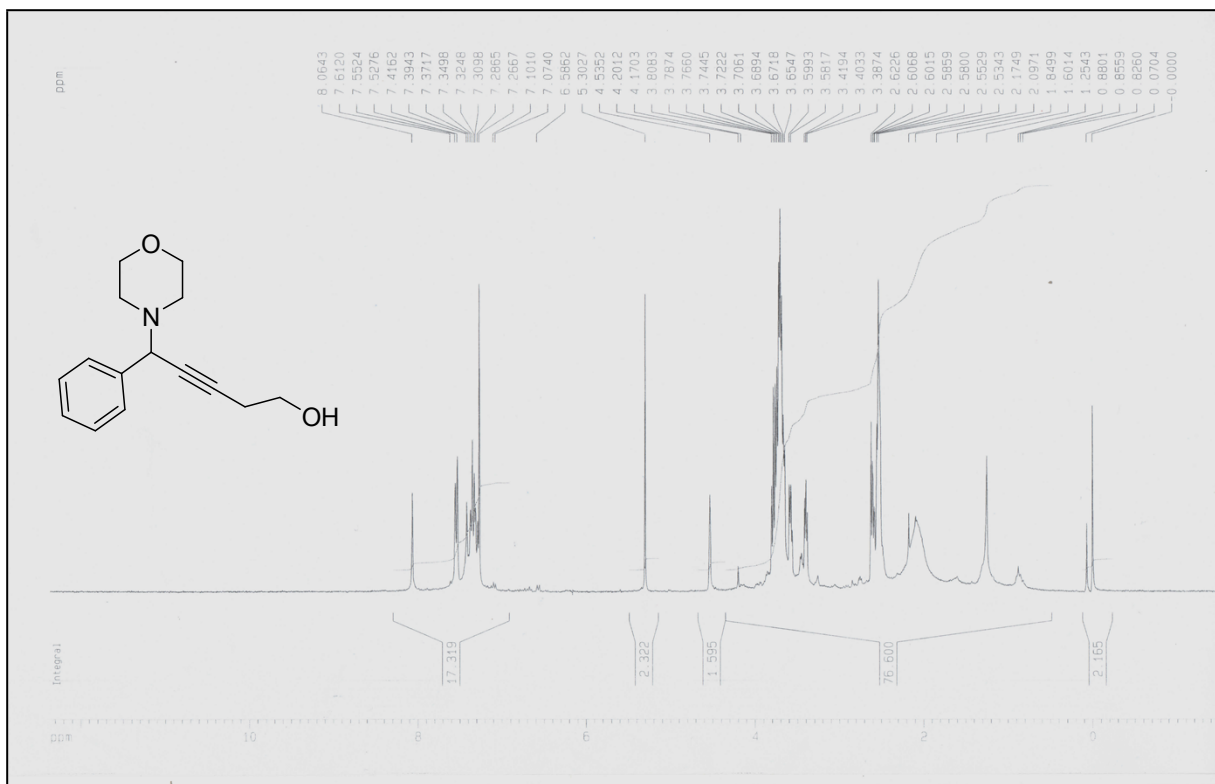


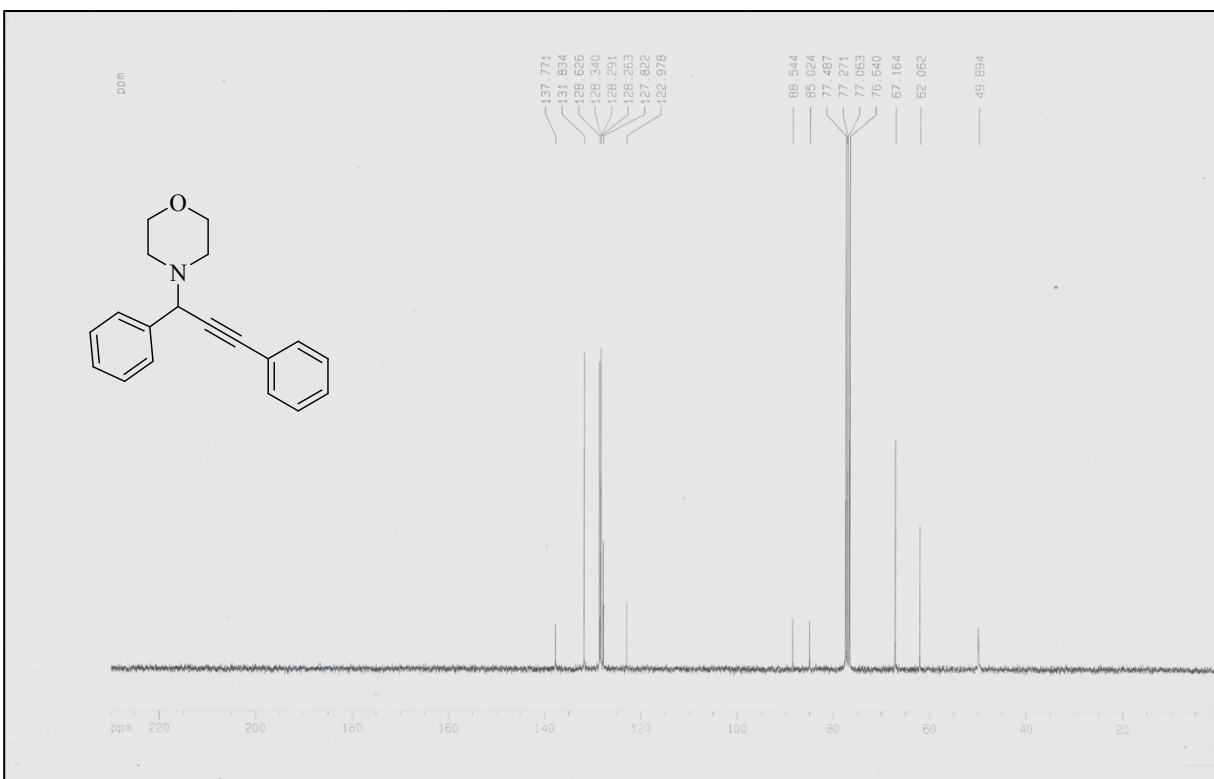
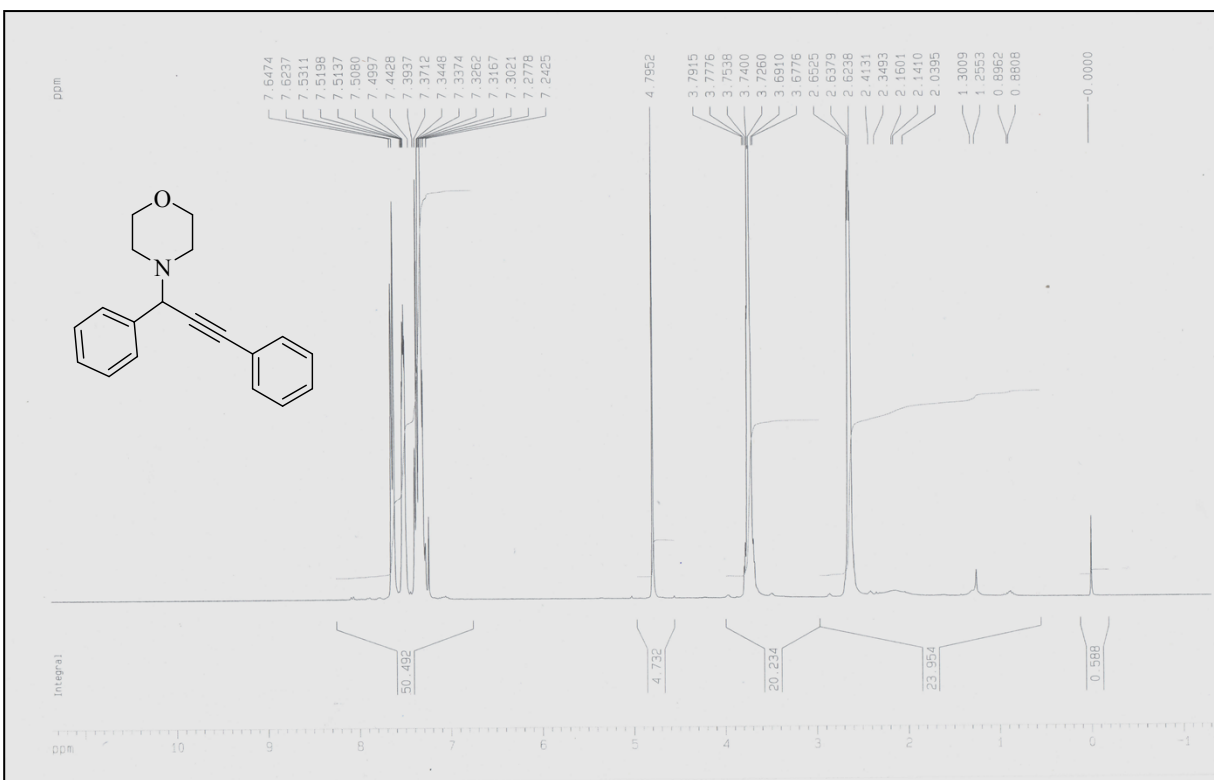


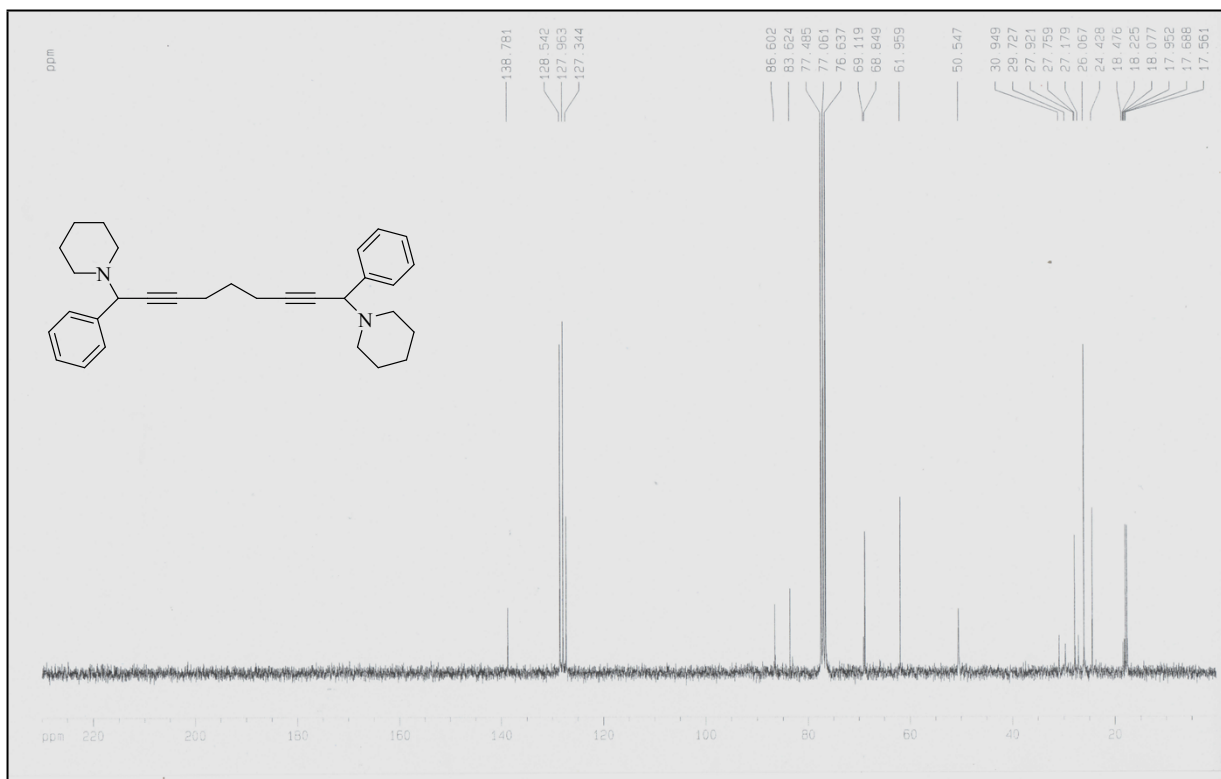
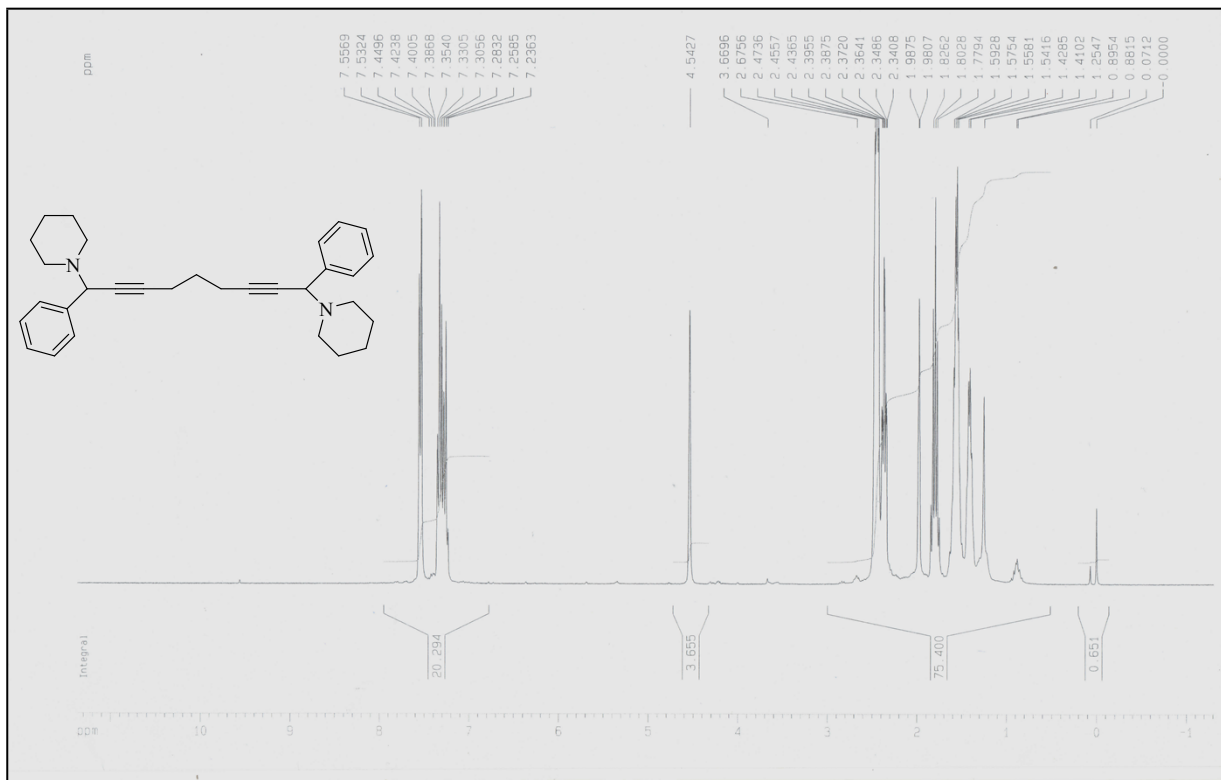












References

- [1] J. E. Gillott, Clay in Engineering Geology, **1968**, 1st ed. (Chapter 5), Elsevier, Amsterdam.
- [2] M. Borah, J. N. Ganguli, D. K. Dutta, *J. Colloid. Interface Sci.* **2001**, 233,171.
- [3] A. B. Searle, R. W. Grimshaw, The Chemistry and Physics of Clays and other Ceramic Materials, 3rd ed., Earnest Benn, London, **1960**, Chapter 5.