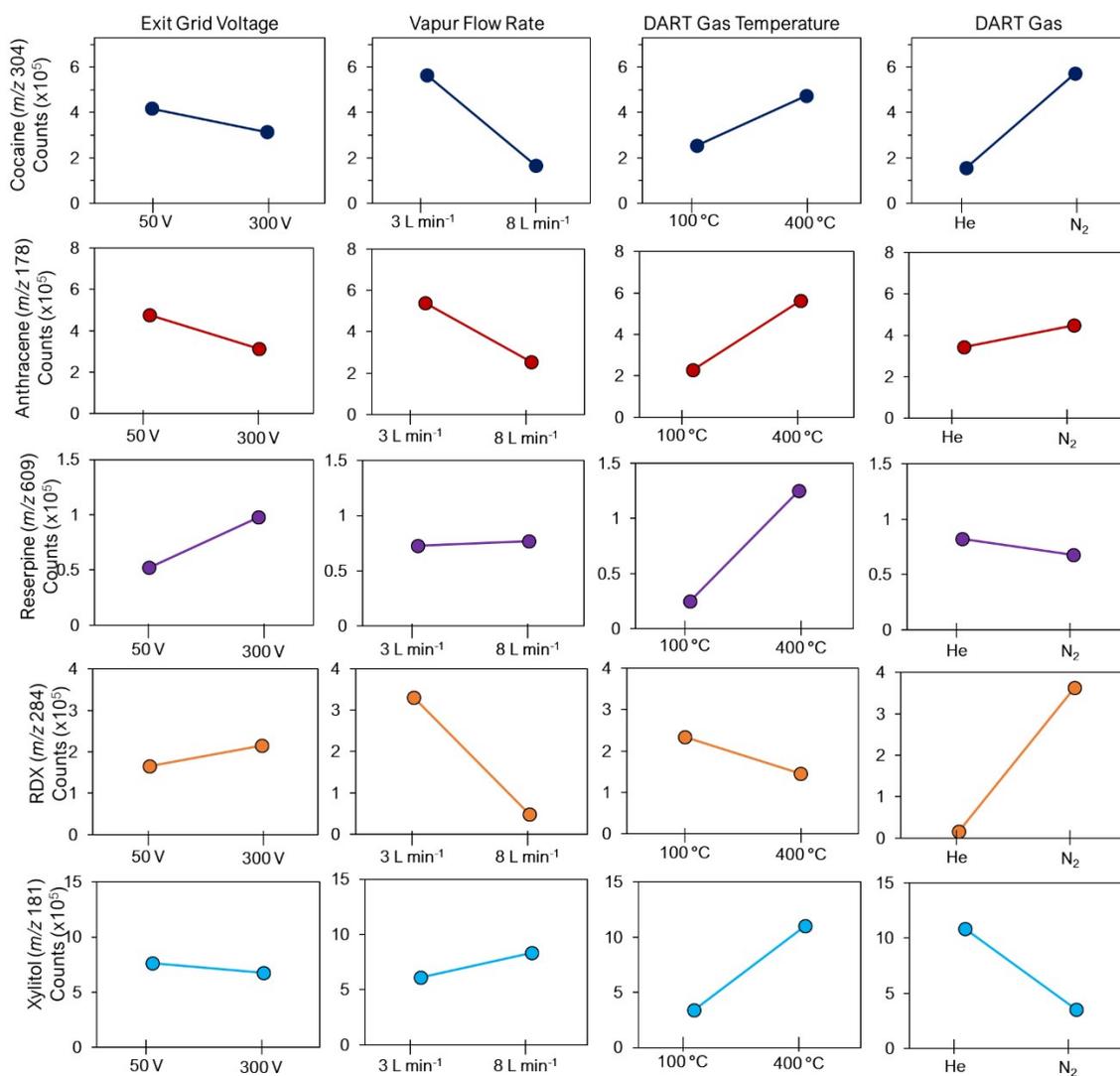


## Supplemental Information for: Optimization of Confined Direct Analysis in Real Time Mass Spectrometry (DART-MS)

Edward Sisco<sup>a</sup>, Matthew Staymates<sup>a</sup>, Thomas P. Forbes<sup>a</sup>

<sup>a</sup>National Institute of Standards and Technology

\*edward.sisco@nist.gov, 301-975-2093



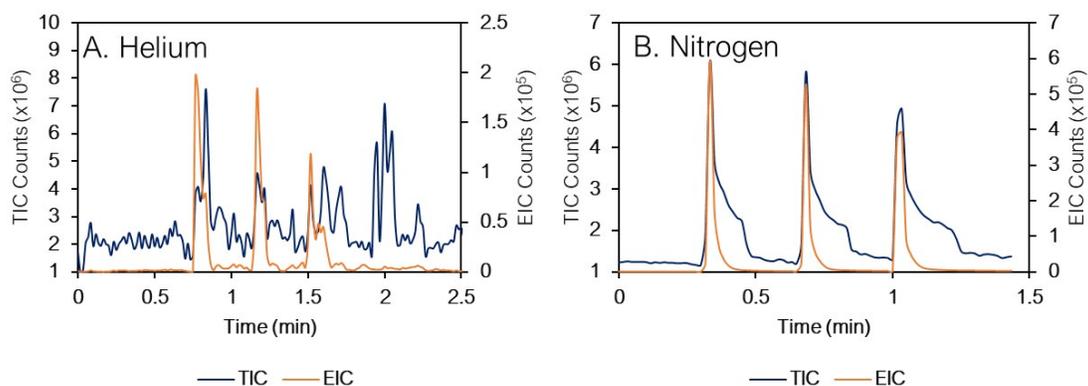
**Figure S1.** Main effects plots from the partial factorial design of experiments study for cocaine (navy blue), anthracene (red), reserpine (purple), RDX (orange), and xylitol (light blue).

**Table S1.** Parameters and results for each setting of the design of experiments study.

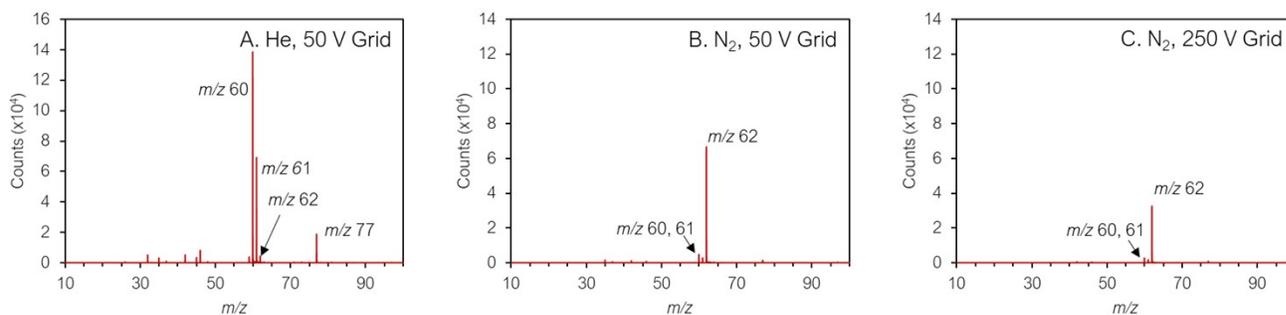
Run #	1	2	3	4
DART Exit Grid (V)	50	300	50	300
Vapur Flow (L min <sup>-1</sup> )	3	3	8	8
DART Gas Temp (°C)	100	100	100	100
DART Gas	He	N <sub>2</sub>	N <sub>2</sub>	He
<b>Average Analyte Response (counts)</b>				
Anthracene	15317	22315	23781	29661
Cocaine	109676	558434	238130	114943
Reserpine	17048	14829	41560	25995
RDX	38215	776070	111807	11353
Xylitol	280587	103244	297411	675226
Run #	5	6	7	8
DART Exit Grid (V)	50	300	50	300
Vapur Flow (L min <sup>-1</sup> )	3	3	8	8
DART Gas Temp (°C)	400	400	400	400
DART Gas	N <sub>2</sub>	He	He	N <sub>2</sub>
<b>Average Analyte Response (counts)</b>				
Anthracene	118735	58926	33074	14749
Cocaine	1251219	337861	66962	242630
Reserpine	61947	197617	87768	152391
RDX	502752	6604	6909	64468
Xylitol	576177	1476943	1911240	445836

**Table S2.** Valve dial positions for select Vapur flow rates used in this study.

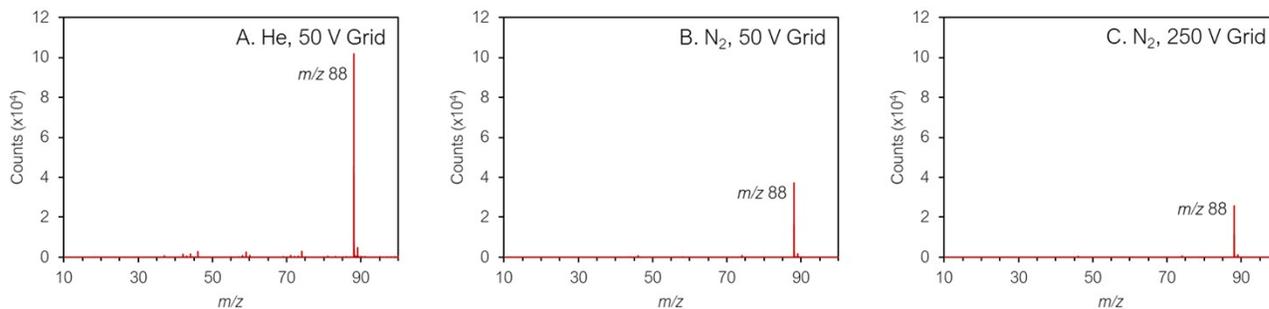
Vapur Flow Rate (L min <sup>-1</sup> )	2	3	4	5	6	7	8
Dial Position	2.5	3.5	5.0	7.0	8.5	0.0	1.0
	1 <sup>st</sup> Turn (Red Band)					2 <sup>nd</sup> Turn (Blue Band)	



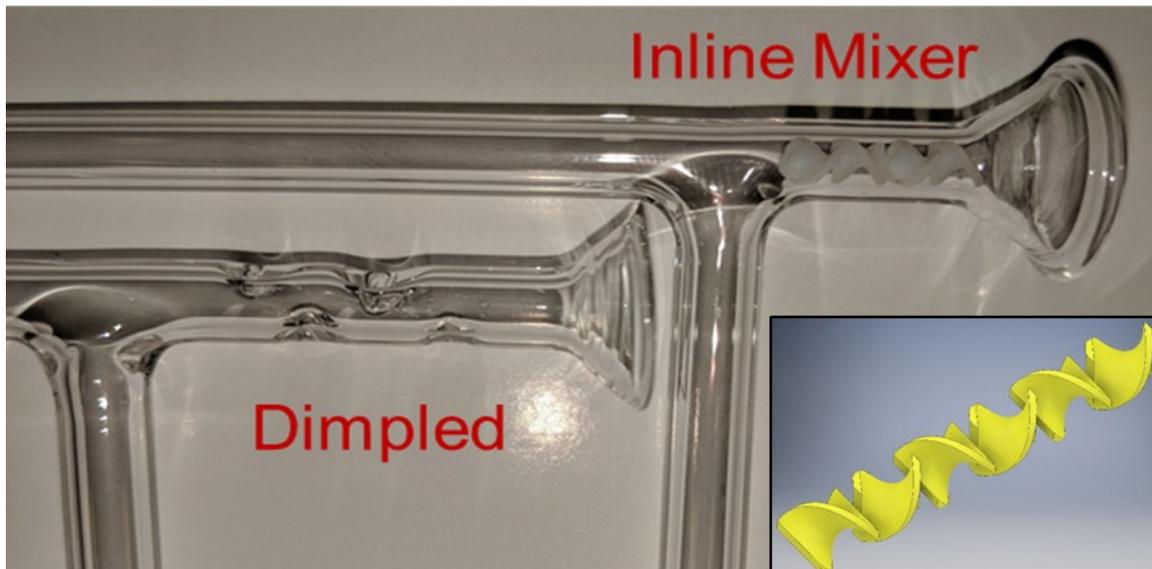
**Figure S2.** Example total ion chromatographs (TICs, blue trace) and corresponding extracted ion chromatographs (EICs, orange trace) of cocaine when using (A.) helium and (B.) nitrogen at a Vapur flow rate of 4.5 L min<sup>-1</sup>.



**Figure S3.** Negative mode background spectra when using helium (A.) and nitrogen (B. and C.) as the DART ionization gas. Also shown is background spectra using a -50 V DART exit grid voltage (B.) and a voltage of -250 V.



**Figure S4** Positive mode background spectra when using helium (A.) and nitrogen (B. and C.) as the DART ionization gas. Also shown is background spectra using a +50 V DART exit grid voltage (B.) and a voltage of +250 V.



**Figure S5.** Photo of inline mixer (top) and dimpled junction (bottom) modifications. A CAD drawing of the inline mixer is shown in the inset of the figure.

#### Supplemental Videos

**Video S1.** Schlieren imaging of the TD-DART front end at increasing Vapor flow rates when using nitrogen as the DART ionization gas.

**Video S2.** Schlieren imaging of the TD-DART front end at increasing Vapor flow rates when using helium as the DART ionization gas.

**Video S3.** High-speed video of the TD-DART junction when using helium and nitrogen at a Vapor flow rate of  $3.5 \text{ L min}^{-1}$ . Theatrical fog was used to simulate the analyte vapor.

**Video S4.** High-speed video of the TD-DART junction when using helium and nitrogen at a Vapor flow rate of  $8 \text{ L min}^{-1}$ . Theatrical fog was used to simulate the analyte vapor.

**Video S5.** High-speed video of the TD-DART junction containing the in-line mixer modification using helium and a Vapor flow rate of  $3.5 \text{ L min}^{-1}$ . Theatrical fog was used to simulate the analyte vapor.

**Video S6.** High-speed video of the TD-DART junction with the dimple modification when using helium and nitrogen at a Vapor flow rate of  $3.5 \text{ L min}^{-1}$ . Theatrical fog was used to simulate the analyte vapor.

**Video S7.** Schlieren imaging of the TD-DART front end, using the dimple modification, at increasing Vapor flow rates when using helium as the DART ionization gas.