

Structure, Electronic Properties, Morphology Evolution, and Photocatalytic Activity in PbMoO_4 and $\text{Pb}_{1-2x}\text{Ca}_x\text{Sr}_x\text{MoO}_4$ ($x = 0.1, 0.2, 0.3, 0.4$ and 0.5) Solid Solutions

E.O. Gomes¹, L. Gracia^{1,2}, A.A.G. Santiago³, R.L. Tranquilin⁴, F.V. Motta³, R.A.C. Amoresi^{1,5}, E. Longo⁴, M.R.D. Bomio³, J. Andres^{1*}.

¹*Department of Analytical and Physical Chemistry, Universitat Jaume I, 12071 Castelló, Spain*

²*Permanent address: Department of Physical Chemistry, University of Valencia (UV), 46100 Burjassot, Spain*

³*LSQM – Laboratory of Chemical Synthesis of Materials – Department of Materials Engineering, Federal University of Rio Grande do Norte – UFRN, P.O. Box 1524, Natal, RN, Brazil*

⁴*CDMF-LIEC, UFSCar, P.O. Box 676, 13565-905 São Carlos, SP, Brazil*

⁵*School of Engineering, Sao Paulo State University (UNESP), Guaratinguetá, São Paulo 12516-410, Brazil*

Supporting Information

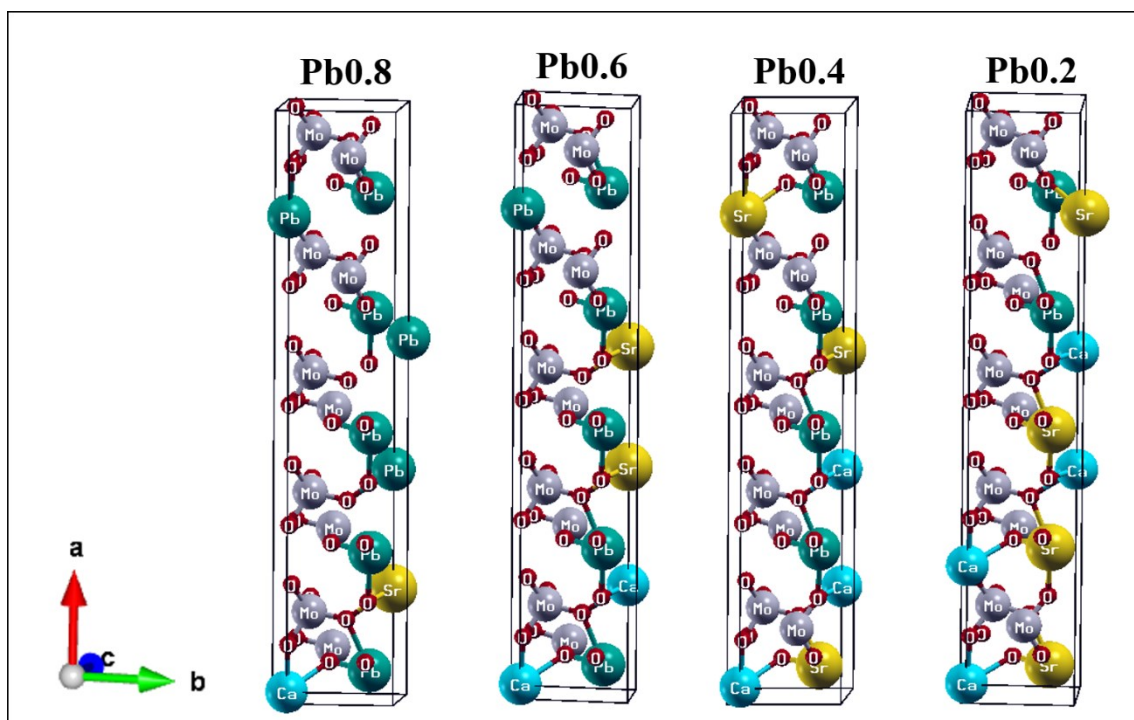


Figure S1. 5x1x1 supercell for $\text{Pb}_{1-2x}\text{Ca}_x\text{Sr}_x\text{MoO}_4$ (Pb=0.8, Pb0.6, Pb0.4, Pb0.2) solid solutions.

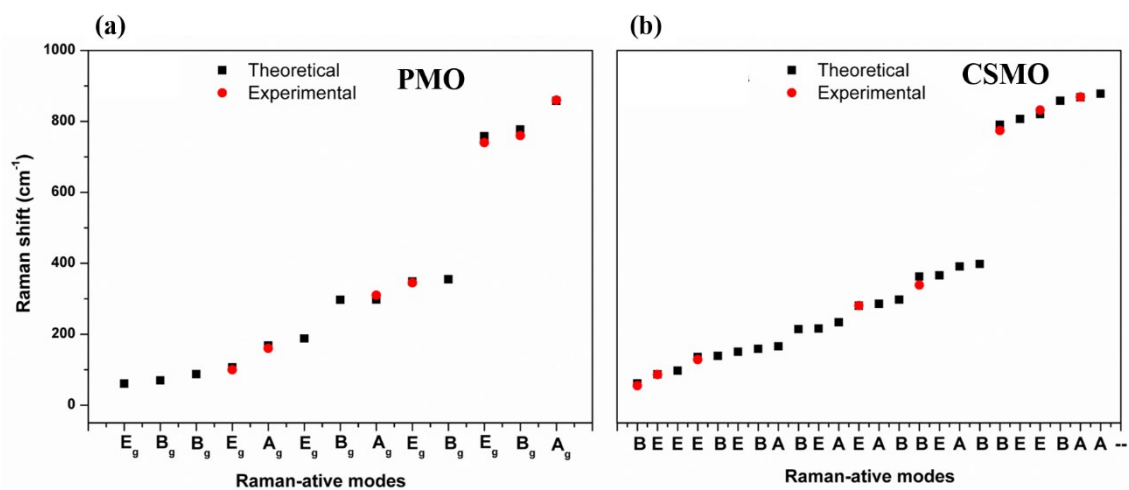


Figure S2. Theoretical and experimental values of the Raman modes for: (a) PMO, and (b) CSMO.

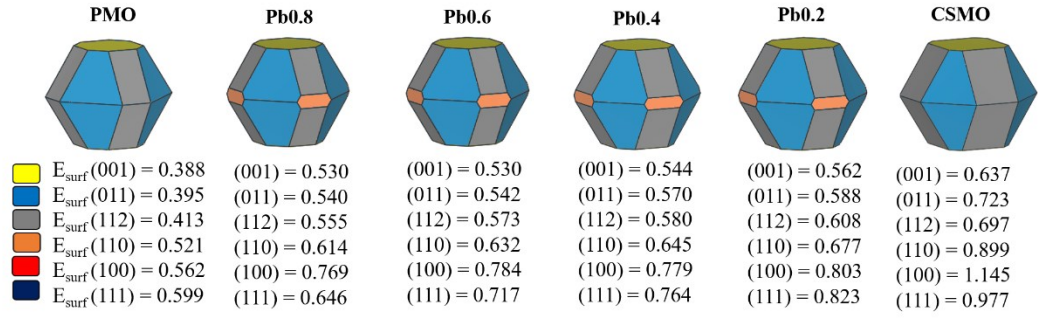


Figure S3. Values of E_{surf} in (J/m^2) and possible morphologies of PMO and $\text{Pb}_{1-2x}\text{Ca}_x\text{Sr}_x\text{MoO}_4$ solid solutions. PMO ($x=0$), Pb0.8 ($x=0.1$), Pb0.6 ($x=0.2$), Pb0.4 ($x=0.3$), Pb0.2 ($x=0.4$), CSMO ($x=0.5$).

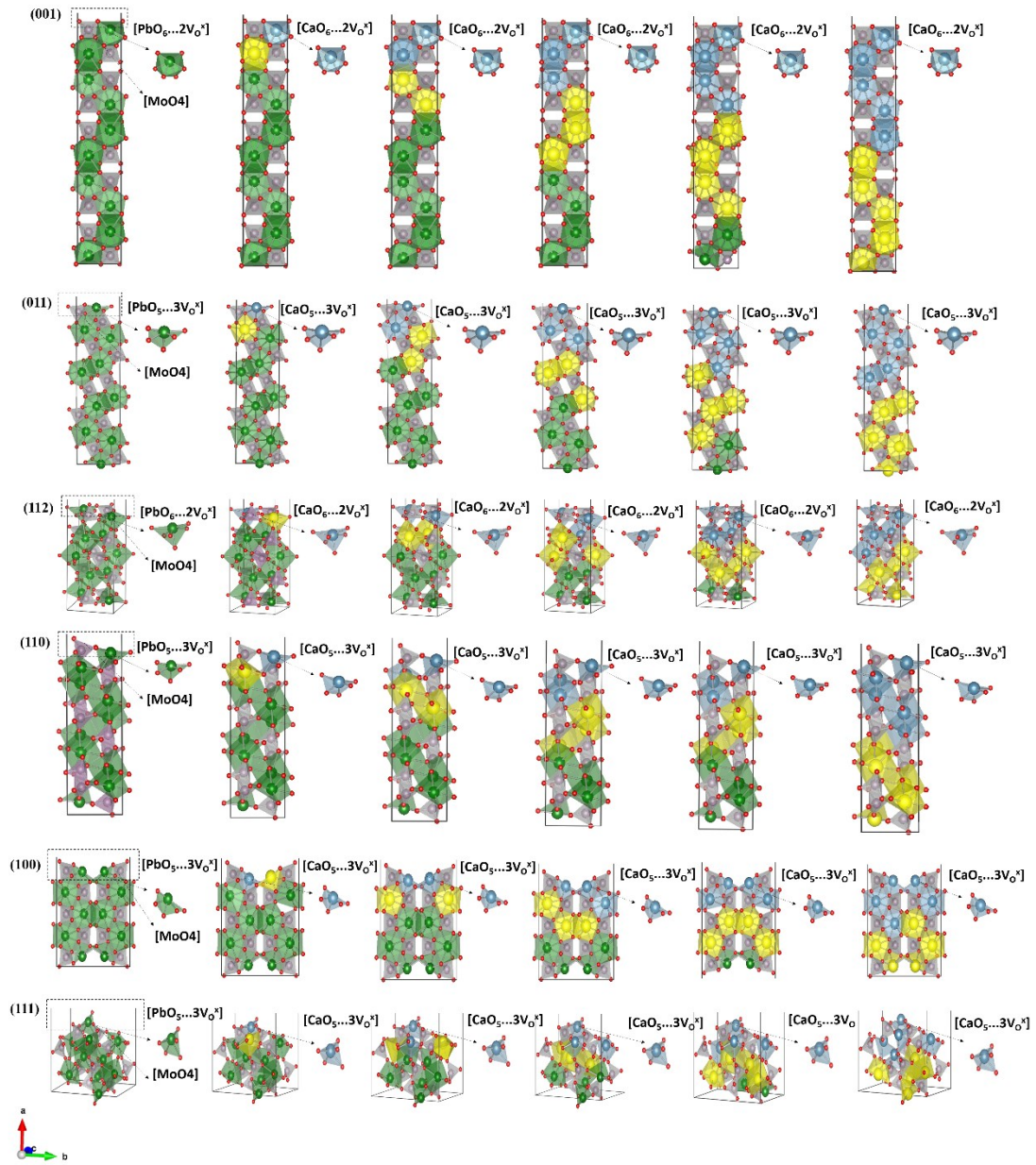


Figure S4. Schematic representations of surfaces: a) (001), b) (112), c) (100), d) (110), e) (111), and f) (011) for PMO and $\text{Pb}_{1-2x}\text{Ca}_x\text{Sr}_x\text{MoO}_4$ ($\text{Pb}=0.8, \text{Pb}0.6, \text{Pb}0.4, \text{Pb}0.2$ and CSMO) solid solutions.

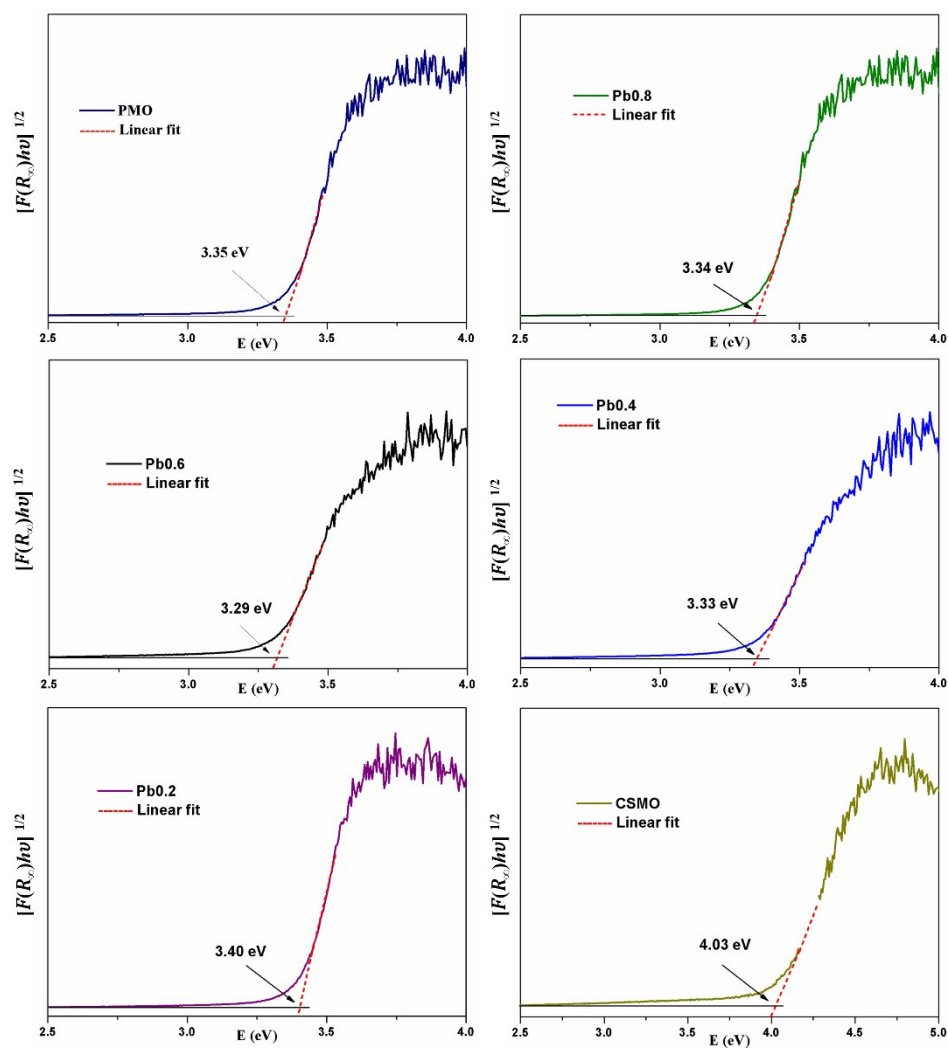


Figure S5. Calculated E_{gap} values by using the Wood and Tauc for PMO and $\text{Pb}_{1-2x}\text{Ca}_x\text{Sr}_x\text{MoO}_4$ (Pb=0.2, Pb0.4, Pb0.6, Pb0.8 and CSMO) solid solutions.

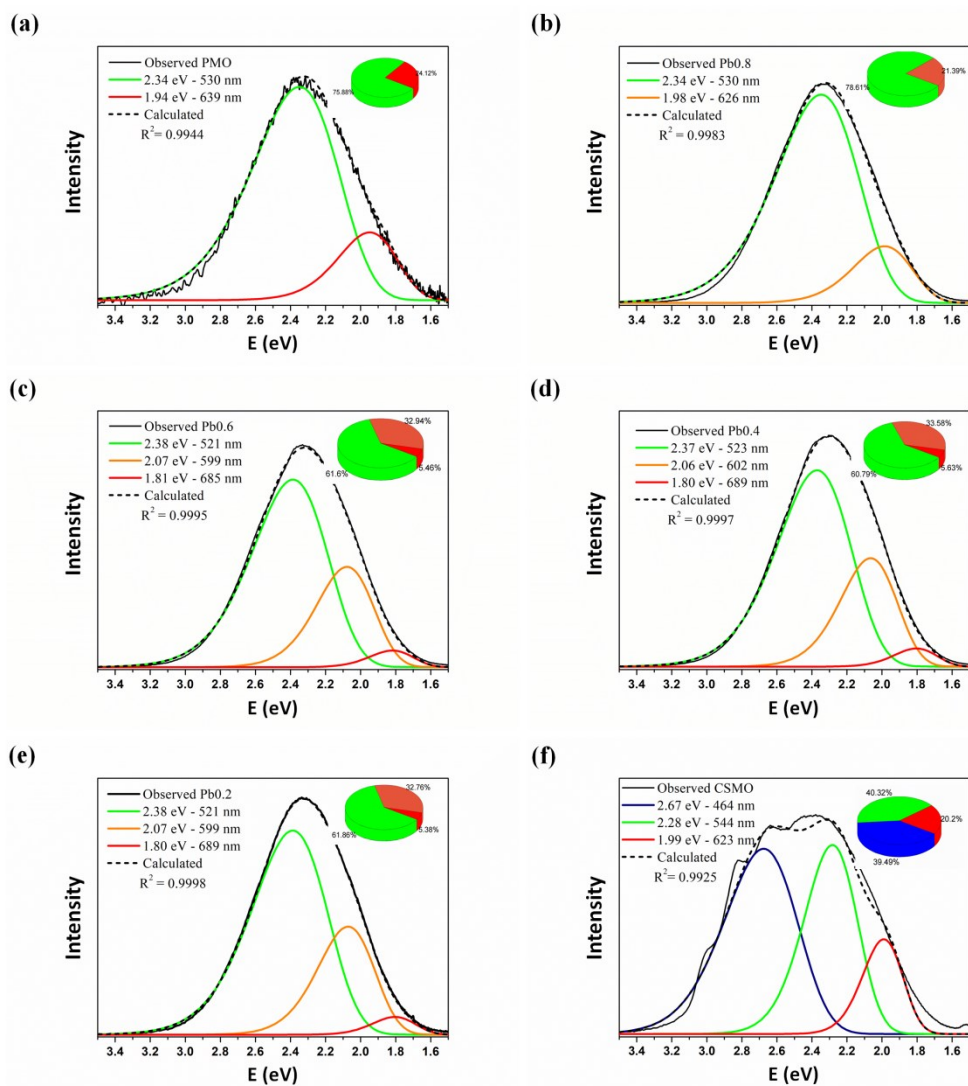


Figure S6. PL deconvolution curves: (a) PMO, (b) Pb0.8, (c) Pb0.6, (d) Pb0.4, (e) Pb0.2 and (f) CSMO.

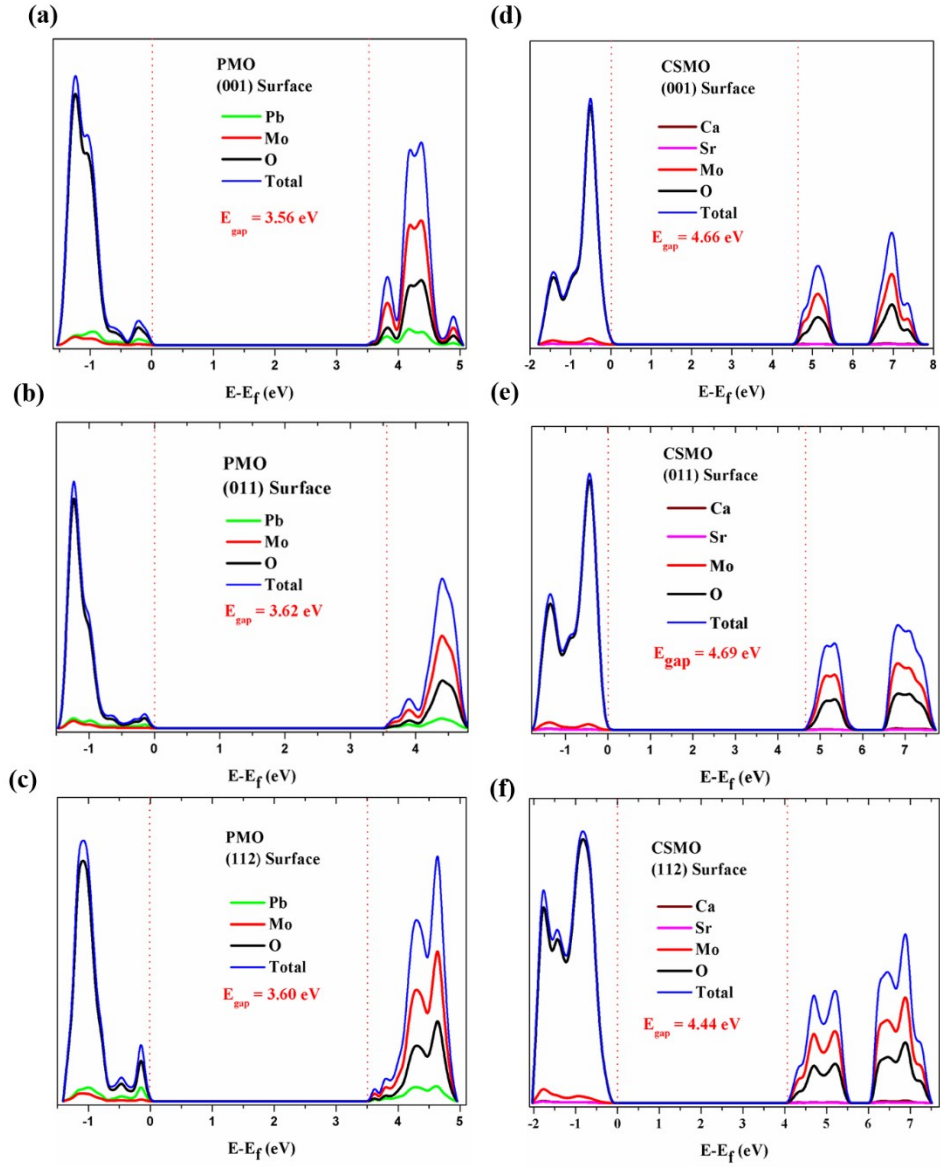


Figure S7. DOS of PMO surfaces (a) (001), (b) (011), (c) (112), and DOS of CSMO surfaces (d) (001), (e) (011), (f) (112).

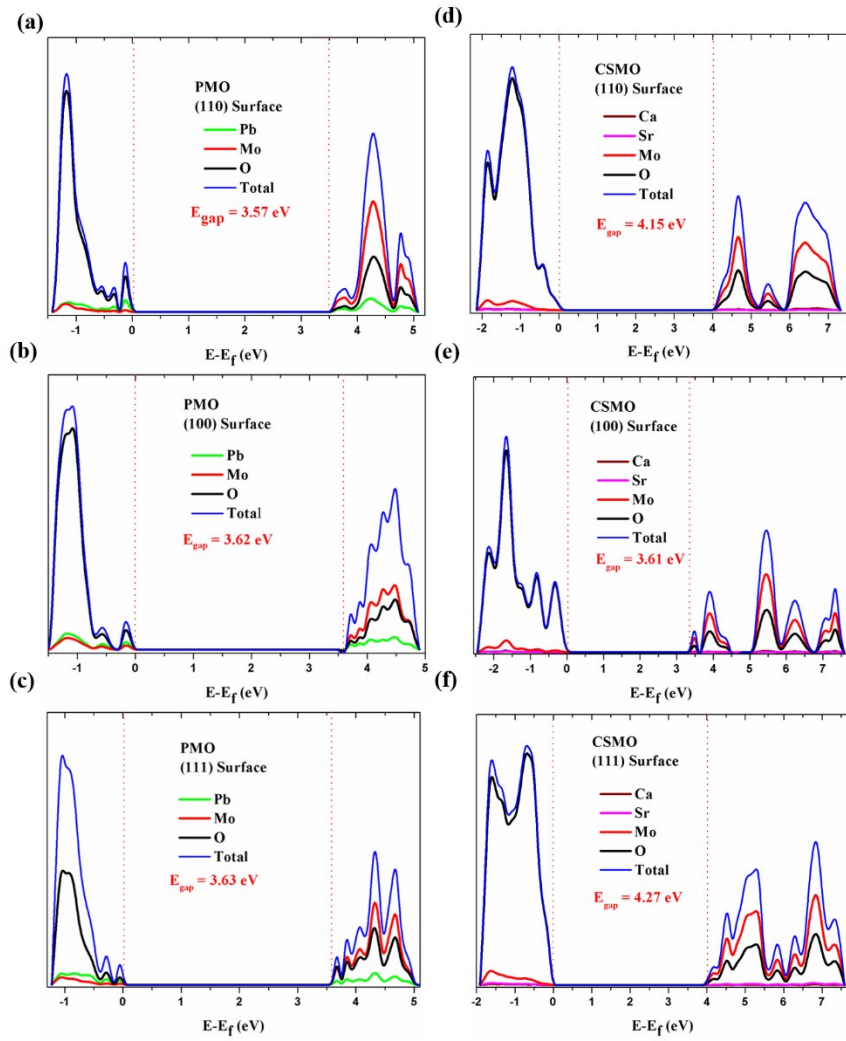


Figure S8. DOS of PMO surfaces (a) (110), (b) (100), (c) (111) and DOS of CSMO surfaces (d) (110), (e) (100), (f) (111).

Table S1. Theoretical values of the Pb-O, Ca-O, and Mo-O distances at the [PbO₆], [PbO₅], [CaO₆], [CaO₅] and [MoO₄] clusters at the exposed surfaces of the PMO and Pb_{1-2x}Ca_xSr_xMoO₄ (Pb= 0.2, Pb0.4, Pb0.6, Pb0.8 and CSMO) solid solutions in (Å) and the broken bond density (D_b) in (nm⁻²).

Surface Bonds (Å)												
(001) Surface												
Samples	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO
Bond (Å)	[PbO ₆]	[CaO ₆]	[CaO ₆]	[CaO ₆]	[CaO ₆]	[CaO ₆]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]
2	2.494	2.383	2.376	2.399	2.372	2.398	1.777	1.786	1.782	1.787	1.784	1.789
2	2.533	2.449	2.445	2.454	2.446	2.413	1.860	1.850	1.861	1.848	1.861	1.836
2	2.692	2.652	2.445	2.654	2.671	2.560						
Surface area (nm ²)	0.305	0.305	0.305	0.305	0.305	0.290						
Broken bond number (N _b)	2	2	2	2	2	2						
D _b (nm ⁻²)	6.557	6.557	6.557	6.557	6.557	6.896						
E _{relax}	0.134	0.046	0.103	0.193	0.271	0.095						
(011) Surface												
Samples	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO
Bond (Å)	[PbO ₅]	[CaO ₅]	[CaO ₅]	[CaO ₅]	[CaO ₅]	[CaO ₅]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]
1	2.338	2.277	2.268	2.267	2.269	2.302	1.779	1.798	1.844	1.795	1.791	1.795
2	2.403	2.347	2.345	2.348	2.348	2.320	1.789	1.803	1.822	1.800	1.798	1.796
3	2.498	2.389	2.382	2.396	2.379	2.336	1.823	1.813	1.798	1.816	1.821	1.821
4	2.568	2.419	2.432	2.419	2.428	2.400	1.853	1.836	1.791	1.841	1.845	1.831
5	2.588	2.471	2.529	2.492	2.527	2.433						
Surface area (nm ²)	0.371	0.371	0.371	0.371	0.371	0.380						
Broken bond number (N _b)	3	3	3	3	3	3						
D _b (nm ⁻²)	8.086	8.086	8.086	8.086	8.086	7.894						
E _{relax}	0.294	0.052	0.558	0.600	0.661	0.397						
(112) Surface												
Samples	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO
Bond (Å)	[PbO ₆]	[CaO ₆]	[CaO ₆]	[CaO ₆]	[CaO ₆]	[CaO ₆]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]
1	2.450	2.344	2.342	2.343	2.327	2.329	1.775	1.780	1.794	1.793	1.794	1.779
2	2.456	2.380	2.422	2.420	2.444	2.358	1.791	1.791	1.795	1.795	1.795	1.791
3	2.519	2.431	2.483	2.460	2.472	2.411	1.837	1.833	1.830	1.827	1.836	1.830
4	2.552	2.502	2.513	2.533	2.481	2.486	1.842	1.850	1.847	1.849	1.845	1.838
5	2.703	2.525	2.547	2.590	2.499	2.500						
6	2.736	2.711	2.691	2.670	2.766	2.530						
Surface area (nm ²)	0.567	0.567	0.567	0.567	0.567	0.538						
Broken bond number (N _b)	4	4	4	4	4	4						
D _b (nm ⁻²)	7.054	7.054	7.054	7.054	7.054	7.434						
E _{relax}	1.592	0.084	0.084	0.122	0.158	1.509						
(110) Surface												
Samples	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO

Bond (Å)	[PbO ₅]	[CaO ₅]	[CaO ₅]	[CaO ₅]	[CaO ₅]	[CaO ₅]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]
1	2.343	2.309	2.323	2.316	2.320	2.470	1.735	1.739	1.817	1.739	1.738	1.747
2	2.389	2.404	2.428	2.476	2.412	2.517	1.801	1.813	1.888	1.815	1.816	1.805
3	2.455	2.409	2.436	2.828	2.436	2.544	1.842	1.857	1.738	1.845	1.847	1.847
4	2.554	2.443	2.433	2.829	2.436	2.554	1.889	1.872	1.845	1.890	1.889	1.872
5	2.856	2.451	2.484	2.841	2.480	2.608						
Surface area (nm ²)	0.478	0.478	0.478	0.478	0.478	0.498						
Broken bond number (N _b)	4	4	4	4	4	4						
D _b (nm ⁻²)	8.368	8.368	8.368	8.368	8.368	8.032						
E _{relax}	1.861	0.103	0.121	0.157	0.199	0.330						

(100) Surface

Samples	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO
Bond (Å)	[PbO ₅]	[CaO ₅]	[CaO ₅]	[CaO ₅]	[CaO ₅]	[CaO ₅]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]
1	2.303	2.258	2.263	2.255	2.282	2.213	1.730	1.731	1.731	1.737	1.730	1.739
2	2.329	2.285	2.290	2.287	2.308	2.268	1.806	1.796	1.812	1.807	1.809	1.812
3	2.256	2.309	2.320	2.300	2.308	2.298	1.846	1.850	1.837	1.842	1.847	1.835
4	2.800	2.396	2.471	2.434	2.471	2.395	1.878	1.877	1.875	1.858	1.875	1.848
5	2.839	2.423	2.475	2.458	2.508	2.453						
Surface area (nm ²)	0.676	0.676	0.676	0.676	0.676	0.640						
Broken bond number (N _b)	6	6	6	6	6	6						
D _b (nm ⁻²)	8.875	8.875	8.875	8.875	8.875	9.375						
E _{relax}	0.573	0.028	0.166	0.183	0.246	0.746						

(111) Surface

Samples	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO
Bond (Å)	[PbO ₅]	[CaO ₅]	[CaO ₅]	[CaO ₅]	[CaO ₅]	[CaO ₅]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]	[MoO ₄]
1	2.351	2.283	2.365	2.283	2.322	2.492	1.763	1.790	1.732	1.799	1.795	1.785
2	2.364	2.292	2.368	2.307	2.377	2.496	1.808	1.806	1.856	1.801	1.806	1.793
3	2.422	2.445	2.435	2.431	2.457	2.521	1.820	1.816	1.861	1.808	1.815	1.817
4	2.528	2.457	2.450	2.446	2.465	2.538	1.836	1.822	1.831	1.832	1.817	1.825
5	2.854	2.497	2.481	2.471	2.531	2.667						
Surface area (nm ²)	1.004	1.004	1.004	1.004	1.004	0.994						
Broken bond number (N _b)	7	7	7	7	7	7						
D _b (nm ⁻²)	6.972	6.972	6.972	6.972	6.972	7.042						
E _{relax}	1.220	0.073	0.115	0.166	0.220	0.301						

Table S2. Theoretical and experimental values of the E_{gap} for PMO and $\text{Pb}_{1-2x}\text{Ca}_x\text{Sr}_x\text{MoO}_4$ (Pb=0.8, Pb0.6, Pb0.4, Pb0.2 and CSMO) solid solutions.

Samples	E_{gap} of the Bulk (eV)					
	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO
Experimental	3.35	3.34	3.29	3.33	3.40	4.03
Theoretical	3.61	3.65	3.75	3.89	3.92	4.74

Table S3. Calculated values of the E_{gap} for PMO and $\text{Pb}_{1-2x}\text{Ca}_x\text{Sr}_x\text{MoO}_4$ surfaces.

Surfaces	E_{gap} of the surfaces (eV)					
	PMO	Pb0.8	Pb0.6	Pb0.4	Pb0.2	CSMO
(001)	3.56	3.56	3.57	3.56	3.64	4.66
(100)	3.62	3.59	3.61	3.81	3.99	3.61
(011)	3.62	3.65	3.7	3.8	4.06	4.69
(110)	3.57	3.61	3.67	3.66	4.0	4.15
(111)	3.63	3.69	3.68	3.79	3.96	4.27
(112)	3.60	3.64	3.69	3.77	3.97	4.44

Table S4. Calculated values (at atoms and clusters) of atomic Mulliken charge for PMO and CSMO bulk, and surfaces.

Bulk				Surfaces			
PMO		CSMO		PMO (001)		CSMO (001)	
Species	Population	Species	Population	Species	Population	Species	Population
Pb	1.26	Ca	1.62	[PbO ₆]	-0.41	[CaO ₆]	-0.29
Mo	1.67	Sr	1.85	[MoO ₄]	0.42	[MoO ₄]	0.28
O5	-0.24	1Mo	1.73	(112)		(112)	
O6	-0.24	2Mo	1.71	[PbO ₆]	-0.36	[CaO ₆]	-0.20
O7	-0.24	O5	-0.29	[MoO ₄]	0.35	[MoO ₄]	0.21
O8	-0.24	O6	-0.29	(011)		(011)	
O9	-0.24	O7	-0.29	[PbO ₅]	-0.44	[CaO ₅]	-0.29
O10	-0.24	O8	-0.29	[MoO ₄]	0.35	[MoO ₄]	0.17
O11	-0.24	O9	-0.29	(110)		(110)	
O12	-0.24	O10	-0.29	[PbO ₅]	-0.18	[CaO ₅]	-0.41
[PbO ₈]	-0.69	O11	-0.29	[MoO ₄]	0.16	[MoO ₄]	0.12
[MoO ₄]	0.69	O12	-0.29	(100)		(100)	
		[CaO ₈]	-0.68	[PbO ₅]	-0.20	[CaO ₅]	-0.16
		[SrO ₈]	-0.45	[MoO ₄]	0.29	[MoO ₄]	0.09
		1[MoO ₄]	0.58	(111)		(111)	
		2[MoO ₄]	0.56	[PbO ₅]	-0.35	[CaO ₅]	-0.49
				[MoO ₄]	0.18	[MoO ₄]	0.20

Table S5. The comparison of the photocatalytic activity of the prepared nanoparticle, Pb0.6, with other samples of the literature.

Sample	Dye / Concentration	kinetic constant (min ⁻¹)	Irradiation	References
P25	Methyl Orange 1.5 × 10 ⁻³ mol/L	6.0 × 10 ⁻³	15 W, 365 nm, UV lamp	[1]
P25	RhB 1.0 × 10 ⁻⁵ mol/L	6.7 × 10 ⁻²	25 W, 254 nm, UV lamp	[2]
PMO	RhB 1.0 × 10 ⁻⁵ mol/L	7.1 × 10 ⁻³	18 W, UV lamp	[3]
Pb0.6	RhB 1.0 × 10 ⁻⁵ mol/L	2.0 × 10 ⁻²	15 W, 254 nm, TUV Philip	This work

References

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2. W. Dong, C. W. Lee, X. Lu, Y. Sun, W. Hua, G. Zhuang, S. Zhang, J. Chen, H. Hou and D. Zhao, *Appl. Catal., B*, 2010, **95**, 197–207.
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